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# Isoxazole Derivatives as Centrally Acting Muscle Relaxants. II.<sup>1)</sup> Synthesis and Structure-Activity Relationship of 3-AminoN-(3-phenyl-5-isoxazolyl)propanamides<sup>2)</sup>

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N-Substituted 3-amino-2-methyl-N-(3-phenyl-5-isoxazolyl)propanamide derivatives (5a—w) were prepared, and their muscle relaxant and anticonvulsant activities were evaluated. Among the compounds with high potency, 3-diethylamino-2,N-dimethyl-N-(3-phenyl-5-isoxazolyl)propanamide (5e) was found to possess selective muscle relaxant and anticonvulsant activities. Compounds with carbon side chains other than the 2-methylpropanamide moiety did not give increased muscle relaxant activity. Optical isomers of 5e were prepared from methyl (+)- and (-)-3-diethylamino-2-methylpropionates (15a, b), and the (+)-isomer (5x) was found to be twice as potent as the (-)-isomer (5y). The structure-activity relationship was studied with emphasis on the effects of the 3-amino moiety and of the substituent on the isoxazolyl 5-amino group. A good quadratic correlation equation was found between hydrophobicity and muscle relaxant activity.

**Keywords**—isoxazole derivative; muscle relaxant; propanamide; anticonvulsant; tolperisone; anemic decerebrate rigidity; hydrophobicity; optical resolution; structure–activity relationship

As described in the preceding paper, our studies to develop new types of muscle relaxant compounds led to the finding that 2-methyl-N-(3-phenyl-5-isoxazolyl)-3-pyrrolidinopropanamide (5c) exhibits activity as potent as that of tolperisone (1).<sup>1)</sup> In an attempt to strengthen the muscle relaxant activity and selectivity relative to general central nervous system (CNS) depressant activity, we introduced an alkyl group onto the 5-amino group of the isoxazole ring. In the present paper, we describe the synthesis of N-substituted 3-amino-N-(3-phenyl-5-isoxazolyl)propanamides (5, 12—14), and their pharmacological activity, together with the result of a QSAR study of the synthesized compounds.

Me COCHCH<sub>2</sub>N Ph Me NHCOCHCH<sub>2</sub>N tolperisone (1) 
$$5c$$

# Chemistry

5-Monoalkylamino-3-phenylisoxazoles (2b—f) were synthesized by the acylation of 5-amino-3-phenylisoxazole (2a), followed by reduction of the resulting amides (3b—f) with lithium aluminum hydride, because 2b—f were not obtained directly by the usual isoxazole formation. The aminoisoxazoles (2a—h) were then reacted with methacryloyl chloride under

basic conditions to give the 2-methyl-2-propenamides (4a—h), which were converted into the 3-amino-2-methylpropanamides (5a—w) by Michael addition reaction of various amines.

Compound 5i could not be prepared by this route, due to the low reactivity of 2,5-dimethylpyrrolidine with 4b. The magnesium salt of 2b, formed by treatment with methylmagnesium bromide in tetrahydrofuran (THF), was subsequently heated with methyl 3-(2,5-dimethylpyrrolidino)-2-methylpropionate (6) in the same solvent to give 5i.

As an alternative route to prepare N-methyl-2-propenamides, N-methylation using methyl iodide and potassium fluoride-alumina in acetonitrile<sup>3)</sup> was achieved in good yield. The 3-bromopropanamide  $(7)^{1)}$  was thus methylated with concomitant dehydrobromination to give N-methyl-2-propenamide (8), and similarly, the 2-butenamide  $(9)^{1)}$  gave N-methyl-2-butenamide (10).

2-Ethyl-N-methyl-2-propenamide (11) was synthesized by the acylation of 2b with 2-ethyl-2-propenoyl chloride.<sup>12)</sup>

The unsaturated amides (8, 10, 11) were converted into 3-aminopropanamides (12—14) by Michael addition of amines.

In order to prepare optical isomers of 5e, a resolution of methyl 3-diethylamino-2-methylpropionate (15) was performed using chiral  $\alpha$ -methylbenzylamine.  $D(+)-\alpha$ -methylbenzylamine was treated with ethylmagnesium bromide in THF and subsequently heated with 15 in the same solvent to form a diastereomeric mixture of amides (16a, b). Two triplet signals ( $\delta$  1.00 for 16a and  $\delta$  0.92 for 16b) due to methyls of the diethylamino groups were observed in the proton nuclear magnetic resonance ( $^1$ H-NMR) spectrum. The mixture of amides (16a, b) was separated by silica gel column chromatography, but due to the marginal separation and the occurrence of tailing, only pure 16a, which eluted first, was obtained.

$$\begin{array}{c} 2a \longrightarrow \begin{array}{c} \text{Ph} \\ \text{NO} \\ \text{NHCOCH}_2\text{CH}_2\text{Br} \\ \end{array} \\ 7 \\ 2a \longrightarrow \begin{array}{c} \text{NO} \\ \text{NHCOCH}_2\text{CH}_2\text{Br} \\ \end{array} \\ 7 \\ 8 \\ \end{array} \\ \begin{array}{c} \text{NO} \\ \text{NCOCH}=\text{CH}_2 \\ \text{NO} \\ \text{NHCOCH}=\text{CHMe} \\ \end{array} \\ \begin{array}{c} \text{NO} \\ \text{NCOCH}=\text{CHMe} \\ \end{array} \\ \begin{array}{c} \text{NO} \\ \end{array} \\ \begin{array}{c} \text{NO} \\ \text{NCOCH}=\text{CHMe} \\ \end{array} \\ \begin{array}{$$

Similarly, the magnesium salt of  $L(-)-\alpha$ -methylbenzylamine in THF was heated with 15 to give a diastereomeric mixture of 16c and 16d. Pure 16c was obtained by column chromatography on silica gel.

Compound 16a was smoothly reacted with boron trifluoride-methanol complex to give methyl (+)-3-diethylamino-2-methylpropionate (15a), which was heated with the magnesium salt of 2b to form the (+)-isomer (5x). The (-)-isomer (5y) was obtained from the amide (16c) similarly, via the (-)-ester (15b).

The yields, physical constants, and selected spectroscopic data of all the products (2b—h, 4b—h, 5a—w, 12a, b, 13a, b, 14a) are listed in Tables VIII—XI in the experimental section, and the pharmacological data are summarized in Tables I—V.

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Table I. Muscle Relaxant and Anticonvulsant Activities on 3-Amino-2-methyl-N-(3-phenyl-5-isoxazolyl)propanamides (5a—w)

Compd.		NR <sup>4</sup> R <sup>5</sup> -		R <sup>a)</sup>	$PTZ^{b)}$
No.	R <sup>1</sup>	NK'K' -	$I_{c)}$	$ED_{50}(i.v.)^{d)}$	$ED_{50} (i.p.)^{d)}$
5a	Н	NMe <sub>2</sub>	e)	38.9	50
5b	Н	$NPr_2$	5	12.0	79.3
5c	Н	$\bigcap^{N}$	e)	10.0	17.7
5d	Me	$NMe_2$	e)	13.2	e)
5e	Me	$NEt_2$	63	3.0	14.0
5f	Me	N(Et)Pr	59	4.3	13.2
5g	Me	$NPr_2$	76	3.5	28.1
5h	Me	$\bigcap_{N}$	66	4.0	11.1
5i	Me	N	e)	2.5	8.1
5j	Me	N	67	3.6	e)
5k	Me	N	95	2.0	e)
<b>5</b> l	Et	$\widetilde{NEt_2}$	78	3.6	12.5
5m	Et	$\bigcap^{N}$	73	3.8	22.3
5n	Pr	$NEt_2$	62	3.9	25—50
<b>5</b> 0	Pr	N	35	10.5	25—50
5p	$\mathbf{Bu^i}$	$NMe_2$	22	7.6	17.7
5q	$\mathbf{B}\mathbf{u}^{\mathrm{i}}$	$\bigcap$ N	58	4.2	28.1
5r	CH <sub>2</sub> CMe <sub>3</sub>	NHPr <sup>i</sup>	59	4.1	e)
5s	$CH_2^2CMe_3$	$NEt_2$	39	6.5	e)
5t	Ph	$NEt_2$	90	3.0	14.0
5u	CH <sub>2</sub> Ph	$NMe_2$	43	5.6	17.0
5v	$CH_2^2$ Ph	NEt <sub>2</sub>	45	5.4	35.4
5w	$CH_2Ph$	$\bigcirc_{N}$	42	6.0	25
Tolneri	isone (1)	•	50	5.0	31.4

Dose was calculated as free base. a) Anemic decerebrate rigidity in rats. b) Anticonvulsant activity was examined against tonic extensor convulsion induced by pentylenetetrazole (PTZ). c) Inhibition ratio  $\binom{6}{0}$  at 5 mg/kg, i.v. d) ED<sub>50</sub> is expressed as mg/kg. e) Not tested. Bu<sup>i</sup> = isobutyl. Pr<sup>i</sup> = isopropyl.

# **Pharmacology**

Pharmacological data of compounds 5a—w are shown in Table I. Muscle relaxant activity was tested in the anemic decerebrate rigidity model, and is shown as an inhibition ratio at 5 mg/kg i.v. and  $ED_{50}(i.v.)$ . Anticonvulsant activity against pentylenetetrazole (PTZ)-induced seizure is shown as  $ED_{50}(i.p.)$ . As regards  $R^1$ , replacement of hydrogen with a methyl group increased the potency for rigidity and anticonvulsion, but larger alkyl or aralkyl groups resulted in gradually decreased potency. Compounds 5e, i, k, t exhibited muscle relaxant

Table II. Muscle Relaxant Activity of 3-Amino-N-methyl-N-(3-phenyl-5-isoxazolyl)propanamides (12a, b, 13a, b, 14a)

Compd. No.	$\mathbb{R}^2$	$\mathbb{R}^3$	NR <sup>4</sup> R <sup>5</sup>	$\mathbf{R}^{a)}$ $\mathbf{I}^{a)}$
12a	Н	Н	NEt <sub>2</sub>	26
12b	Н	Н	N	4
13a	Н	Me	$NEt_2$	30
13b	Н	Me	N	18
14a	Et	Н	NEt <sub>2</sub>	$48^{b}$

a) See footnote in Table I. b) Inhibition ratio at 2.5 mg/kg, i.v.

TABLE III. Muscle Relaxant, Anticonvulsant, and Central Depressant Activities of the Isoxazole Derivatives and Muscle Relaxants

	$R^{a)}$	A	\	
Compound	i.v.	PTZ i.p.	MES i.p.	$\mathbf{D}^{c)}$ $i.p.$
5c	10.0	17.7	25.0	134
5e	3.0	14.0	25.0	56.1
Tolperisone (1)	5.0	31.4	61.5	63.0
Mephenesin	54.0	89.0	112	159
Baclofen	1.6	14.0	> 50	5.3

Activity is expressed as  $ED_{50}$ . a) See footnote in Table I. b) Anticonvulsant activity against pentylenetetrazole (PTZ), or maximal electroshock (MES). c) Depression of motor activity determined by the revolving wheel method.

activity approximately twice as potent as that of tolperisone (1).

In order to identify the optimal propanamide moiety, muscle relaxant activity of 12a, b, 13a, b, 14a was tested, and the results are shown in Table II. Compound 14a (with an ethyl group as R<sup>2</sup>) was twice as active as 1, but other compounds were less potent than the corresponding 2-methylpropanamide derivatives.

For comparing the isoxazole derivatives tested with known muscle relaxant drugs (tolperisone, mephenesin, and baclofen), activity ratios (ED<sub>50</sub> in depression of spontaneous motor activity/ED<sub>50</sub> in muscle relaxant or anticonvulsant activity) were calculated as the protective index (PI) according to Share and McFarlane.<sup>4)</sup> Depression of spontaneous motor activity was used as an index of general CNS depression. Detailed data for 5c, e, and the three muscle relaxant drugs are shown in Table III, and their PI values are given in Table IV. Baclofen showed the highest muscle relaxant activity among these five compounds, although its PI value was only three and PI values for anticonvulsant activity against PTZ and maximal electroshock (MES) were less than one. Compound 5c showed the same PI value for rigidity as tolperisone (1), and 5e showed a higher value than 1. The PI values of 5c, e for anticonvulsion were higher than those of the muscle relaxant drugs.

Common 4	$R^{b)}$	$\mathbf{A}^{b)}$		
Compound	<b>K</b> *′	PTZ	MES	
5c	13	7.6	5.4	
5e.	19	4.0	2.2	
Tolperisone (1)	13	2.0	1.0	
Mephenesin	2.9	1.8	1.4	
Baclofen	3.3	0.4	< 0.1	

TABLE IV. Protective Index (PI)a) of the Isoxazole Derivatives and Muscle Relaxants

Table V. Muscle Relaxant and Anticonvulsant Activities, and Acute
Toxicities of Optically Resolved Isomers

Compound No.	R <sup>a)</sup>	$PTZ^{a)}$			$LD_{50} (mg/kg)^{b}$		
	i.v.	<i>i.p.</i> (30 min) <sup>c)</sup>	p.o. (10 min)	p.o. (60 min)	i.v.	p.o.	
5e	3.4	8.5	42.2	46.5	31.0	367.7	
5x	2.9	7.3	45.9	44.1	22.3	341.0	
<b>5</b> y	6.9	14.7	42.2	34.2	50.0	367.7	

Dose was calculated as fumarate. a) See footnote in Table I. Activity is expressed as  $ED_{50}$  (mg/kg). b) The 50% lethal dose in mice. c) Test compound was administered intraperitoneally 30 min before the injection of PTZ.

The muscle relaxant and anticonvulsant activities of 5e, x, y are shown in Table V. The (+)-isomer (5x) was as potent as the racemic mixture (5e), and approximately twice as potent as the (-)-isomer (5y) in rigidity-reducing and anticonvulsant activities (i.p.), but when orally administered, these three showed identical anticonvulsant activity.

# Structure-Activity Relationship

The pED<sub>50</sub> values of compounds 5a-w are shown in Table VI together with the hydrophobic parameters<sup>5)</sup> of substituents, R<sup>1</sup> and NR<sup>4</sup>R<sup>5</sup>, used for calculation. Although  $\pi$  value is best established for aromatic substituents, it is available for a wide range of substituents and its additive nature is established, so the  $\pi$  value was used as a hydrophobic parameter. We considered  $\pi_1$ ,  $\pi_2$ , the sum of  $\pi_1$  and  $\pi_2$ , and quadratic terms of these three. Significant correlation equations are shown in Table VII.<sup>6)</sup>

Equations 1—6 showed that no one linear or quadratic term of each parameter or their sum could account for the activity. Equation 7 indicated that a quandratic equation using  $\pi_1 + \pi_2$  could give a reliable equation in terms of the F and t test, but the correlation coefficient was not satisfactory. With two more terms, an equation with improved r and identical F values was obtained as Eq. 8.

As this Eq. 8 was not easy to understand, it was plotted on a two-dimensional plane with  $\pi_1$  on the horizontal axis and  $\pi_2$  on the vertical axis, as shown in Fig. 1.<sup>7)</sup> The  $\pi$  values of 5a—w were also plotted on the same plane. This figure showed that potency is more dependent upon  $\pi_2$ , and is less sensitive to the increase of  $\pi_1$ . As the partition coefficient (log P) can be calculated by adding  $\pi$  values to the parent molecule, Eq. 7 showed a correlation between activity and partition coefficient of compounds 5a—w.

The preference for Eq. 8 over Eq. 7 indicated that the hydrophobicity of the terminal amino moiety (NR<sup>4</sup>R<sup>5</sup>) has a greater influence on the activity than that of the substituent (R<sup>1</sup>)

a)  $PI = ED_{50}$  in depression in motor activity/ $ED_{50}$  in each test. b) See footnote in Table III.

TABLE VI. Hydrophobic Parameters and Activity of Compounds 5a—w in the Anemic Decerebrate Rigidity Test, Used in Regression Analysis

Compound No.	$\pi_1^{\ a)}$	$\pi_2^{\ a\rangle}$	pED <sub>50</sub> (µ Observed	mol/kg) Calcd <sup>b)</sup>	Observed – calcd
5a	0	0.18	-2.16	-1.94	-0.22
5b	0	$2.3^{c)}$	-1.56	-1.30	-0.26
5c	0	$0.4^{c)}$	-1.52	-1.69	0.17
5d	0.56	0.18	-1.66	-1.65	-0.01
5e	0.56	1.18	-0.98	-1.01	0.03
5f	0.56	$1.7^{c)}$	-1.12	-1.00	-0.12
5g	0.56	$2.3^{c)}$	-1.01	-1.31	0.30
5h	0.56	$0.4^{c)}$	-1.11	-1.27	0.16
5i	0.56	$1.5^{c)}$	-0.87	-0.97	0.10
5j	0.56	0.85	-1.04	-1.12	0.08
5k	0.56	$1.3^{c)}$	-0.80	-0.96	0.16
51	1.02	1.18	-1.04	-0.97	-0.07
5m	1.02	0.85	-1.05	-1.03	-0.02
5n	1.55	1.18	-1.06	-0.98	-0.08
50	1.55	0.85	-1.47	-1.01	-0.46
5p	$2.1^{c)}$	0.18	-1.36	-1.28	-0.08
5q	2.1°)	$0.4^{c)}$	-1.07	-1.18	0.11
5r	2.5 <sup>c)</sup>	$0.6^{c)}$	-1.06	-1.16	0.10
5s	$2.5^{c)}$	1.18	-1.24	-1.18	-0.06
5t	1.96	1.18	-0.90	-1.05	0.15
5u	2.01	0.18	-1.19	-1.30	0.11
5v	2.01	1.18	-1.14	-1.07	-0.07
5w	2.01	$0.4^{c)}$	-1.19	-1.19	0

a)  $\pi_1$  and  $\pi_2$  are hydrophobic parameters for  $R^1$  and  $NR^4R^5$ , respectively. b) Calcd from Eq. 8 in Table VII. c) Estimated from  $\pi$  and f values in ref. 5.

Table VII. Correlation Equations between Muscle Relaxant Activity (pED<sub>50</sub>) and Hydrophobic Parameters of N-Isoxazolylpropanamides (5a—w)

Equation No.	pED <sub>50</sub> =	n	r	s	F
(1)	$0.105\pi_1 - 1.322$	23	0.287	0.300	$F_{21}^1 = 1.88$
(2)	$(\pm 0.159)^{a}$ $(\pm 0.226)$ $0.023(\pi_1)^2 - 1.246$	23	0.159	0.309	$F_{21}^1 = 0.55$
(3)	$(\pm 0.064)$ $(\pm 0.186)$ $0.179\pi_2 - 1.368$	23	0.367	0.291	$F_{21}^1 = 3.27$
(4)	$(\pm 0.206) (\pm 0.231)$ $0.040(\pi_2)^2 - 1.250$	23	0.194	0.307	$F_{21}^1 = 0.82$
(5)	$(\pm 0.091)$ $(\pm 0.176)$ $0.184(\sum \pi)^{b)} - 1.591$	23	0.537	0.264	$F_{21}^1 = 8.49$
(6)	$(\pm 0.131)$ $(\pm 0.301)$ $0.034(\sum \pi)^2 - 1.379$ (+0.037) $(+0.231)$	23	0.387	0.289	$F_{21}^1 = 3.69$
(7)	$\begin{array}{ll} (\pm 0.037) & (\pm 0.231) \\ -0.154(\sum \pi)^2 + 0.759(\sum \pi) - 2.001 \\ (\pm 0.105) & (\pm 0.409) & (\pm 0.380) \end{array}$	23	0.717	0.224	$F_{20}^2 = 10.55$
(8)	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	23	0.834	0.186	$F_{18}^4 = 10.32$
	$(\pm 0.338)$				

a) The figures in parenthesis are for the construction of 95% confidence limits of the coefficient in the t test. b)  $\sum \pi = \pi_1 + \pi_2$ .

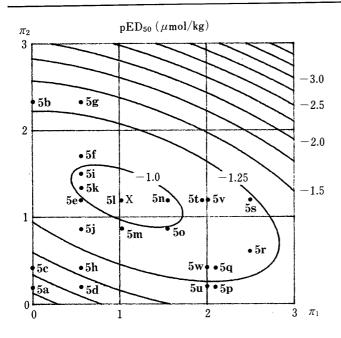


Fig. 1. Effect of Hydrophobic Parameters ( $\pi_1$ and  $\pi_2$ ) on Muscle Relaxant Activity of Propanamides (5a-w)

Muscle relaxant activity [pED50 (µmol/kg)] was calculated from Eq. 8 in Table VII.

on the isoxazolyl 5-amino group. Equation 8 indicated that the optimal  $\pi$  values of  $\mathbb{R}^1$  and  $NR^4R^5$  were 1.11 and 1.25, and the corresponding pED<sub>50</sub> was -0.96. It was plotted as X in Fig. 1. Although less than 70% of the variance of potency was explained by this equation, compounds of high activity (pED<sub>50</sub> > -1, 5e, i, k) were located close of the optimal position. In spite of its high potency, however, compound 5t was located far from the optimal position. This might be due to the influence of the aromatic phenyl group.

Calculated optimum pED<sub>50</sub> of -0.96 was realized in these four compounds, and this potency may be optimal in this series of alkylated derivatives of N-isoxazolylpropanamides 5.

The present study suggests that compound 5e may be more potent and selective than currently used muscle relaxant drugs.

### **Experimental**

Melting points were determined on a Sibata apparatus and are not corrected. Boiling points are uncorrected. Infrared (IR) spectra were recorded using a JASCO model IR-G, or a JASCO model A-202 instrument. Data are presented in reciprocal centimeters, and only the important diagnostic bands are reported. <sup>1</sup>H-NMR spectra were determined on a JEOL model JNM-PMX60 (60 MHz) spectrometer. Chemical shifts are reported in parts per million downfield from a tetramethylsilane internal standard ( $\delta$  scale) and coupling constants in Hz. Mass spectra (MS) were taken on a Shimadzu 7000 mass spectrometer operating at 70 eV; only major ion fragments are reported in the form of m/z (relative intensity). High-resolution MS were measured using a VG Analytical MM ZAB 2F-HF spectrometer. Optical rotations were recorded with a Perkin-Elmer 241 polarimeter. The structures and purities of all compounds are supported by the IR, 1H-NMR, and MS data. For column chromatography, Merck Kieselgel 60 and Sumitomo Active Alumina KCG-30 were used.

# Chemistry

5-Formylamino-3-phenylisoxazole (3b)<sup>8</sup> — A mixture of 5-amino-3-phenylisoxazole (2a, 48.0 g, 0.3 mol) and HCO<sub>2</sub>H (250 ml) was stirred at 100 °C for 1 h and poured into H<sub>2</sub>O (1 l). The resulting precipitate was dissolved in AcOEt and washed with H<sub>2</sub>O. The organic layer was dried over MgSO<sub>4</sub> and evaporated under reduced pressure. The residue was recrystallized from CHCl<sub>3</sub> to give 3b (38.4 g, 83%), mp 112 °C (lit., 8) mp 115—117 °C as the hemihydrate recrystallized from H<sub>2</sub>O).

N-(3-Phenyl-5-isoxazolyl)acetamide (3c) and N-(3-phenyl-5-isoxazolyl)propanamide (3d) were prepared in a usual manner from 2a, the corresponding acid anhydride, and pyridine. 2-Methyl-N-(3-phenyl-5-isoxazolyl)propanamide (3e) and 2,2-dimethyl-N-(3-phenyl-5-isoxazolyl)propanamide (3f) were prepared in a usual manner from 2a, the corresponding acid chloride, and Et<sub>3</sub>N in CHCl<sub>3</sub>. The products were used in the next step without purification.

5-Methylamino-3-phenylisoxazole (2b)—Compound 3b (145 g, 0.771 mol) was added portionwise under  $N_2$  to

TABLE VIII. 5-N-Substituted Amino-3-phenylisoxazoles (2b—h)

Compd. No.	R¹	Yield <sup>a)</sup>	mp (°C)	Recrystn. solvent	IR v (cm <sup>-1</sup> )	$^{1}\text{H-NMR}^{b)}$ $\delta$	MS m/z (M +)
2b	Me	71	110—112	PhH	3230, 1625 (Nujol)	5.28	174
2c	Et	40	115—116	Hexane-PhMe	3230, 1610 (Nujol)	5.27	188
2d	Pr	86	107—109	PhMe	3240, 1615 (Nujol)	5.27	202
2e 2f	Bu <sup>i</sup>	45	4647	Hexane-Et <sub>2</sub> O	3240, 1615 (KBr)	5.23	216
	CH <sub>2</sub> CMe <sub>3</sub>	47	102—104	PhH	3280, 3200, 1625 (KBr)	5.25	230
2g	Ph	35°)	$131 - 133^{d}$	MeOH	3280, 3190, 1650 (Nujol)	5.83	236
2h	CH <sub>2</sub> Ph	37	83—84 <sup>e)</sup>	EtOH-H <sub>2</sub> O	3430, 1610 (KBr)	5.30	250

a) Yield from 5-amino-3-phenylisoxazole (2a). b) Chemical shift of the C-4 proton of the isoxazole (CDCl<sub>3</sub>). c) Yield from 1-morpholino-1-phenylethylene. d) Lit.,  $^{9a}$  mp 143—144 °C (MeOH); Lit.,  $^{9b}$  mp 136—137 °C (EtOH). e) Lit.,  $^{10}$  91—92 °C (EtOH).

TABLE IX. 2-Methyl-N-(3-phenyl-5-isoxazolyl)-2-propenamides (4b—h)

Compd. No.	R¹	Yield	mp (°C)	Recrystn. solvent	IR (Nujol) v (cm <sup>-1</sup> )	$^{1}$ H-NMR $^{a)}$ $\delta$	MS m/z (M <sup>+</sup> )
<b>4b</b>	Me	70	60—61	Et <sub>2</sub> O	1675, 1615	6.50	242
4c	Et	88	Oil			6.39	256
<b>4</b> d	Pr	91	Oil	********		6.35	270
4e 4f	Bu <sup>i</sup>	87	43—44	Hexane-Et <sub>2</sub> O	1660, 1610	6.33	284
	$CH_2CMe_3$	72	75—77	Hexane-Et <sub>2</sub> O		6.25	298
4g	Ph	68	83—84	Et <sub>2</sub> O	1675	6.60	304
4h	CH <sub>2</sub> Ph	74	95—97	Et <sub>2</sub> O	1675, 1600	6.15	318

a) Chemical shift of the C-4 proton of the isoxazole (CDCl<sub>3</sub>).

a stirred suspension of LiAlH<sub>4</sub> (38.0 g, 1 mol) in anhydrous THF (1.4 l). The reaction muxture was stirred at room temperature for 4h, and excess LiAlH<sub>4</sub> was decomposed with H<sub>2</sub>O. The THF layer was dried over MgSO<sub>4</sub> and filtered. The filtrate was evaporated under reduced pressure, and the residue was recrystallized to afford **2b** (114 g, 85%). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 2.93 (3H, d, J=5Hz), 4.7 (1H, br s), 5.28 (1H, s), 7.3—8.0 (5H, m). *Anal.* Calcd for C<sub>10</sub>H<sub>10</sub>N<sub>2</sub>O: C, 68.95; H, 5.79; N, 16.08. Found: C, 68.67; H, 5.56; N, 15.84.

Compounds 2c-f were prepared in the same manner from 3c-f, respectively. Compounds  $2g^{9a}$  and  $2h^{10}$  were prepared according to the reported methods.

Typical Example of the Preparation of N-Substituted 2-Methyl-N-(3-phenyl-5-isoxazolyl)-2-propenamides (4b—h): 2,N-Dimethyl-N-(3-phenyl-5-isoxazolyl)-2-propenamide (4b)—A solution of methacryloyl chloride (19.2 g, 184 mmol) in CHCl<sub>3</sub> (20 ml) was added dropwise to a solution of 2b (16.0 g, 92 mmol) and Et<sub>3</sub>N (18.6 g, 184 mmol) in CHCl<sub>3</sub> (100 ml) below 5 °C. The reaction mixture was stirred at room temperature for 12 h, and quenched with H<sub>2</sub>O. After being washed with dilute HCl, the CHCl<sub>3</sub> layer was dried over MgSO<sub>4</sub> and evaporated under reduced pressure. The residue was purified on a column of silica gel (1.25 kg) using CHCl<sub>3</sub> as an eluent. The crude product from the CHCl<sub>3</sub> eluate was recrystallized from Et<sub>2</sub>O to give 4b (15.5 g). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 2.00 (3H, m), 3.47 (3H, s), 5.28 (2H, m), 6.50 (1H, s), 7.3—7.9 (5H, m). MS m/z (relative intensity): 242 (M<sup>+</sup>, 16), 158 (28), 117 (87), 77 (24), 69 (100), 41 (79).

Compounds 4c—h were prepared in the same manner as described above from 2c—h, respectively. Preparation of 2-methyl-N-(3-phenyl-5-isoxazolyl)-2-propenamide (4a) was described previously. 1)

Typical Example of the Preparation of N-Substituted 3-Amino-2-methyl-N-(3-phenyl-5-isoxazolyl)propanamides

TABLE X. 3-Amino-2-methyl-N-(3-phenyl-5-isoxazolyl)propanamides (5a—w)

Compd. Yield		$Salt^{b)}$	mp (°C)	Recrystn. solvent	Formula		alysis (	., .,
No.				sorvent		C	Н	N
5a	85	HCl	85—87	Et <sub>2</sub> O	$C_{15}H_{20}ClN_3O_2$	58.15	6.51	13.57
						(57.96	6.53	13.44)
5b	58	f	7982	CH <sub>3</sub> COCH <sub>3</sub>	$C_{23}H_{31}N_3O_6$	62.00	7.01	9.43
_						(62.14	6.75	9.50)
5c	88		107108	Hexane-AcOEt	$C_{17}H_{21}N_3O_2$	68.20	7.07	14.04
~ 1	0.2	c	100 100	CH COCH		(68.55	6.91	13.88)
5d	83	f	108—109	CH <sub>3</sub> COCH <sub>3</sub>	$C_{20}H_{25}N_3O_6$	59.54	6.25	10.42
<b>5</b> 0	72	f	112 114	CH COCH	CHNO	(59.62	6.11	10.72)
5e	12	I	113—114	CH <sub>3</sub> COCH <sub>3</sub>	$C_{22}H_{29}N_3O_6$	61.24	6.77	9.74
5f	34	f	79—81	CH <sub>3</sub> COCH <sub>3</sub>	СИМО	(60.88 62.00	6.70	9.71) 9.43
31	34	1	/901	CH <sub>3</sub> COCH <sub>3</sub>	$C_{23}H_{31}N_3O_6$	(61.83	7.01 6.94	9.43 9.59)
5g	27	t	5557	CH <sub>3</sub> COCH <sub>3</sub>	$C_{24}H_{35}N_3O_8$	58.40	7.15	9.39) 8.51
Jg	21	ι	3531	CH <sub>3</sub> COCH <sub>3</sub>	$C_{24}\Pi_{35}\Pi_{3}O_{8}$	(58.60	7.13	8.73)
5h	95	f	144—145	CH <sub>3</sub> COCH <sub>3</sub>	$C_{22}H_{27}N_3O_6$	61.52	6.34	9.79
	,,,	•	111 113	Chiquodhi	022112711306	(61.35	6.72	9.52)
5i	3 <sup>c)</sup>		Oil	·	$C_{20}H_{27}N_3O_2$	Calcd:	341.2	
	-				020112/11302		341.20	
5j	69	f	115117	CH <sub>3</sub> COCH <sub>3</sub>	$C_{23}H_{29}N_3O_6$	62.29	6.59	9.48
•				33	-23293-6	(62.31	6.44	9.56)
5k	45	f	144147	CH <sub>3</sub> COCH <sub>3</sub> -MeOH	$C_{24}H_{31}N_3O_6$	63.00	6.83	9.19
				3 3	- 24 31 3 - 0	(62.85	6.72	8.97)
51	56	f	99—101	CH <sub>3</sub> COCH <sub>3</sub>	$C_{23}H_{31}N_3O_6$	62.00	7.01	9.43
				J	20 01 0	(62.17	7.15	9.27)
5m	76	f	131—133	CH <sub>3</sub> COCH <sub>3</sub>	$C_{24}H_{31}N_3O_6$	63.00	6.83	9.19
						(62.83	6.68	8.83)
5n	52	f	115117	CH <sub>3</sub> COCH <sub>3</sub>	$C_{24}H_{33}N_3O_6$	62.72	7.24	9.14
						(62.65	7.05	8.94)
50	75	, <b>f</b>	123—125	$CH_3COCH_3$	$C_{25}H_{33}N_3O_6$	63.67	7.05	8.91
						(63.50	6.83	8.74)
<b>5</b> p	84	f	90—93	CH <sub>3</sub> COCH <sub>3</sub>	$C_{23}H_{31}N_3O_6$	62.00	7.01	9.43
_						(62.27	7.22	9.51)
<b>5</b> q	83	_	7576	Hexane–Et <sub>2</sub> O	$C_{21}H_{29}N_3O_2$	70.95	8.22	11.82
<b>F</b>	0.7		00 02	TT 77.0		(71.09	7.87	12.03)
5r	87		8082	Hexane-Et <sub>2</sub> O	$C_{21}H_{31}N_3O_2$	70.55	8.74	11.76
<b>E</b> o	60		90 93	Hawana Et O	CHNO	(70.68	8.64	11.88)
<b>5</b> s	68		8082	Hexane–Et <sub>2</sub> O	$C_{22}H_{33}N_3O_2$	71.12 (70.87	8.95 9.14	11.31 11.27)
5t	98	t	7678	Hexane	$C_{27}H_{33}N_3O_8$	61.47	6.31	7.97
J	70	ι	10-10	HOAMIC	C271133143 U8	(61.29	6.50	8.03)
5u	99	t	82—85	Hexane-Et <sub>2</sub> O	$C_{26}H_{31}N_3O_8$	60.81	6.08	8.18
~4	,,	٠	02 00	TIONAILO EtgO	26113111308	(60.95	5.99	8.35)
5v	41	t	54—55	Hexane-Et <sub>2</sub> O	$C_{28}H_{35}N_3O_8$	62.09	6.51	7.76
	. •	•		110 2020	-2833-13-8	(62.22	6.47	7.65)
5w	82	f	118—120	CH <sub>3</sub> COCH <sub>3</sub>	$C_{28}H_{31}N_3O_6$	66.52	6.18	8.31
				J 3	40 41 3 0			

a) Yield of free base from N-isoxazolylpropenamides (4a—h), except 5i. b) f, fumarate; t, tartrate. c) Yield of free base from methyl 3-(2,5-dimethyl-pyrrolidino)-2-methylpropionate (6). d) High MS Calcd for  $C_{20}H_{27}N_3O_2$ .

<sup>(5</sup>d—w): 2,N-Dimethyl-N-(3-phenyl-5-isoxazolyl)-3-pyrrolidinopropanamide (5h)—A mixture of 4b (4g, 16.5 mmol) and pyrrolidine (1.4g, 19.9 mmol) in PhH (80 ml) was refluxed for 24 h. The mixture was evaporated, and the residue was purified on a column of silica gel using CHCl<sub>3</sub>-MeOH (10:1) as an eluent to give 5h (4.91 g) as an oil. IR (neat):

1680,  $1615 \,\mathrm{cm}^{-1}$ . <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.20 (3H, d, J=7Hz), 1.75 (4H, m), 2.2—3.1 (7H, m), 3.43 (3H, s), 6.52 (1H, s), 7.3—7.9 (5H, m). MS m/z (relative intensity): 313 (M<sup>+</sup>, 2), 112 (1), 103 (1), 85 (13), 84 (100), 83 (1), 70 (2).

Compounds 5d—g, j—w were prepared in the same manner as described above from 4b—h. When the reaction was too slow, particularly in the case of the addition of diethylamine, a small amount of AcOH was used to accelerate the reaction. (11) Compounds 5a—c were prepared according to the reported method. (1)

Compound 5a was converted into the hydrochloride, compounds 5b, d—f, h, j—p, w were converted into the fumarates and compounds 5g, t—v were converted into the tartrates in a usual manner, and the resulting salts were recrystallized from an appropriate solvent.

**2,N-Dimethyl-3-(2,5-dimethylpyrrolidino)-***N***-(3-phenyl-5-isoxazolyl)propanamide (5i)**—A 1 M MeMgBr solution (21 ml, 21 mmol) in anhydrous THF was added dropwise to a solution of **2b** (3 g, 17.2 mmol) in anhydrous THF (30 ml) at room temperature. The mixture was stirred for 30 min, then methyl 3-(2,5-dimethylpyrrolidino)-2-methylpropionate [6, 3.43 g, 17.2 mmol, bp 107—109 °C (20 mmHg), 10

N-Methyl-N-(3-phenyl-5-isoxazolyl)-2-propenamide (8)— KF-Al<sub>2</sub>O<sub>3</sub> (Type B,<sup>3)</sup> 4g) and MeI (1.4g, 9.86 mmol) were added to a solution of 3-bromo-N-(3-phenyl-5-isoxazolyl)propanamide<sup>1)</sup> (7, 1.5g, 5.08 mmol) in CH<sub>3</sub>CN (20 ml), and the resulting reaction mixture was stirred at room temperature for 22 h. The mixture was filtered and the filtrate was evaporated. The residue was diluted with H<sub>2</sub>O and extracted with AcOEt. The organic layer was dried over MgSO<sub>4</sub> and evaporated, and the residue was purified on a column of Al<sub>2</sub>O<sub>3</sub> using hexane–CH<sub>2</sub>Cl<sub>2</sub> (1:1) as an eluent to give 8 (0.76g, 66%) as an oil. IR (neat): 1685, 1620 cm<sup>-1</sup>. <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 3.47 (3H, s), 5.77 (1H, t, J=6 Hz), 6.45 (1H, s), 6.50 (2H, d, J=6 Hz), 7.3—7.9 (5H, m). MS m/z (relative intensity): 228 (M<sup>+</sup>, 5), 144 (28), 117 (56), 102 (28), 77 (25), 55 (100). Anal. Calcd for C<sub>13</sub>H<sub>12</sub>N<sub>2</sub>O<sub>2</sub>: C, 68.41; H, 5.30; N, 12.27. Found: C, 68.52; H, 5.33; N, 12.06.

N-(3-Phenyl-5-isoxazoyl)-2-butenamide (9)<sup>1)</sup> was treated in the same manner to give N-methyl-N-(3-phenyl-5-isoxazolyl)-2-butenamide (10, 92%) as an oil. IR (neat): 1680, 1620 cm<sup>-1</sup>. <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.90 (3H, dd, J=7, 2 Hz), 3.43 (3H, s), 6.18 (1H, dq, J=15, 2 Hz), 6.43 (1H, s), 6.95 (1H, dd, J=15, 7 Hz), 7.3—7.9 (5H, m). Anal. Calcd for C<sub>14</sub>H<sub>14</sub>N<sub>2</sub>O<sub>2</sub>: C, 69.40; H, 5.83; N, 11.56. Found: C, 69.27; H, 5.73; N, 11.70.

Compound **2b** was treated with 2-ethyl-2-propenoyl chloride [bp 121—124 °C (lit., <sup>12)</sup> bp 117—119 °C)] and Et<sub>3</sub>N in CHCl<sub>3</sub> in the same manner as described for **4b** to afford 2-ethyl-*N*-methyl-*N*-(3-phenyl-5-isoxazolyl)-2-propenamide (**11**, 61%) as an oil. IR (neat): 1680, 1615 cm<sup>-1</sup>. <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.10 (3H, t, J=7 Hz), 2.35 (2H, dq, J=7, 1 Hz), 3.43 (3H, s), 5.25 (2H, m), 6.50 (1H, s), 7.3—7.9 (5H, m). MS m/z (relative intensity): 256 (M<sup>+</sup>, 3), 117 (61), 102 (30), 83 (94), 77 (18), 55 (100). *Anal*. Calcd for C<sub>15</sub>H<sub>16</sub>N<sub>2</sub>O<sub>2</sub>: C, 70.29; H, 6.29; N, 10.93. Found: C, 70.04; H, 6.52; N, 11.20.

Compounds 12a, b, compounds 13a, b, and compound 14a were prepared in the same manner as described for 5h from 8, 10, and 11, respectively. All of these products were converted into the fumarates in a usual manner.

(+)-3-Diethylamino-2-methyl-N-[(+)-1-phenylethyl]propanamide (16a)—A solution of D(+)-α-methyl-benzylamine (175 g, 1.44 mol) in anhydrous THF (270 ml) was added dropwise below 15 °C to a solution of EtMgBr freshly prepared from EtBr (188 g, 1.73 mol) and Mg (35 g, 1.44 mol) in anhydrous THF (1070 ml). After stirring at room temperature for 30 min, a solution of 15 [125 g, 0.72 mol, bp 75—76 °C (12 mmHg) [lit., <sup>13a)</sup> bp 77—79 °C (15 mmHg), <sup>13b)</sup> bp 98—99 °C (35 mmHg)]] in THF (140 ml) was added dropwise below 10 °C. The mixture was stirred at room temperature for 30 min and refluxed for 4.5 h. The reaction mixture was quenched with  $H_2O$  and extracted with PhMe. The organic layer was washed with  $H_2O$ , dried over MgSO<sub>4</sub>, and evaporated under reduced pressure to give a crude mixture of (+)- and (-)-3-diethylamino-2-methyl-N-[(+)-1-phenylethyl]propanamides (16a and 16b, 168 g), [ $\alpha$ ]<sub>0</sub><sup>25</sup> +27.9 ° (c=0.62, CHCl<sub>3</sub>), which was separated on a column of silica gel (8 kg) using CHCl<sub>3</sub>-EtOH-28% NH<sub>4</sub>OH (600:20:1) as an eluent to afford 16a (24.5 g, 13.0% yield from 15) as an oil. IR (neat): 3300, 1645 cm<sup>-1</sup>. H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.00 (6H, t, J=7 Hz), 1.10 (3H, d, J=7 Hz), 1.43 (3H, d, J=7 Hz), 1.8—3.1 (7H, m), 4.8—5.3 (1H, m), 7.25 (5H, s), 9.15 (1H, br s). MS m/z (relative intensity): 262 (M<sup>+</sup>, 0.8), 105 (7), 87 (7), 86 (100), 72 (6), 58 (7), [ $\alpha$ ]<sub>0</sub><sup>25</sup> +81.6 ° (c=0.711, CHCl<sub>3</sub>).

The purity was determined to be >99% by GLC analysis [column: 1.5% FFAP (Carbowax 20M terminated with 2-nitroterephthalic acid) on chromosorb WHP, 2 m, 210 °C, retention times of 3.8 min and 4.3 min for 16b and 16a, respectively].

 $L(-)-\alpha$ -Methylbenzylamine and 15 were treated in the same manner as described above to afford (-)-3-diethylamino-2-methyl-N-[(-)-1-phenylethyl]propanamide (16c, 24.6% yield from 15) as an oil (99% pure by gasliquid chromatography (GLC) analysis). [ $\alpha$ ]<sup>25</sup> -83.1° (c=0.603, CHCl<sub>3</sub>). IR, <sup>1</sup>H-NMR, and MS data were identical with those of 16a.

Methyl (+)-3-Diethylamino-2-methylpropionate (15a)—Boron trifluoride-methanol complex (BF<sub>3</sub>·2MeOH,

Compd. No.	Yield <sup>a)</sup>	$^{a)}$ mp (°C) $^{b)}$	Formula		Analysis (%) Calcd (Found)			
No.				С	Н	N		
12a	88	96—98	$C_{21}H_{27}N_3O_6$	60.42	6.52	10.07		
				(60.54	6.43	9.92)		
12b	70	144—145	$C_{21}H_{25}N_3O_6$	60.71	6.07	10.12		
				(60.80	5.82	10.33)		
13a	94	118—120	$C_{22}H_{29}N_3O_6$	61.24	6.77	9.74		
				(61.08	6.83	9.85)		
13b	86	146—148	$C_{22}H_{27}N_3O_6$	61.52	6.34	9.79		
				(61.47	6.25	9.55)		
14a	37	112—114	$C_{23}H_{31}N_3O_6$	62.00	7.01	9.43		
				(62.21	6.84	9.26)		

TABLE XI. 3-Amino-N-methyl-N-(3-phenyl-5-isoxazolyl)propanamides (12a, b, 13a, b, 14a)

128 ml) was slowly added dropwise to 16a (24 g, 91.5 mmol) under cooling in an ice bath. The mixture was stirred at  $100\,^{\circ}$ C for 3 h, and poured into ice-water. Conc. NH<sub>4</sub>OH was added to adjust the pH to 8 and the mixture was extracted with Et<sub>2</sub>O. The organic layer was dried over MgSO<sub>4</sub> and evaporated under reduced pressure, and the residue (21.2 g) was purified on a column of silica gel using CH<sub>2</sub>Cl<sub>2</sub>-MeOH (10:1) as an eluent to afford 15a (4.63 g, 29.2%). IR (neat):  $1740\,\text{cm}^{-1}$ . <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.00 (6H, t, J=7.Hz), 1.13 (3H, d, J=5 Hz), 2.2—3.0 (7H, m), 3.67 (3H, s). MS m/z (relative intensity): 173 (M<sup>+</sup>, 0.6), 144 (7), 87 (10), 86 (100), 58 (69), 43 (32). [ $\alpha$ ]<sub>D</sub><sup>25</sup> + 28.2  $^{\circ}$  (c= 0.621, CHCl<sub>3</sub>).

Methyl (-)-3-diethylamino-2-methylpropionate (15b) was obtained in the same manner in 49.2% yield.  $[\alpha]_D^{25}$  -28.4° (c = 0.828, CHCl<sub>3</sub>). IR, <sup>1</sup>H-NMR, and MS data were identical with those of 15a.

(+)-3-Diethylamino-2,N-dimethyl-N-(3-phenyl-5-isoxazolyl)propanamide (5x)—A solution of 2b (9.25 g, 53.1 mmol) in anhydrous THF (30 ml) was added dropwise to a solution of EtMgBr freshly prepared from EtBr (4.76 ml, 63.7 mmol) and Mg (1.29 g, 53.1 mmol) in anhydrous THF (60 ml). The mixture was stirred for 30 min, then 15a (4.60 g, 26.6 mmol) in anhydrous THF (30 ml) was added and the whole was refluxed for 3 h, quenched with  $H_2O$  and extracted with PhMe. The organic layer was extracted with 1 n HCl and the aqueous extract was basified with conc. NH<sub>4</sub>OH and extracted with PhMe. The organic layer was washed with  $H_2O$ , dried over MgSO<sub>4</sub>, and evaporated under reduced pressure to give 5x (6.60 g, 78.8% yield from 15a) as an oil. IR (neat): 1685 cm<sup>-1</sup>.  $^1$ H-NMR (CDCl<sub>3</sub>)  $\delta$ : 0.95 (6H, t, J=7 Hz), 1.13 (3H, d, J=7 Hz), 2.2—3.1 (7H, m), 2.44 (3H, s), 7.3—7.9 (5H, m). [ $\alpha$ ]<sup>25</sup>  $_D$  +63.4 ° (c=0.708, CHCl<sub>3</sub>).

Compound 5x was converted into the fumarate in a usual manner, mp 106.5-107.5 °C. [ $\alpha$ ]<sub>D</sub><sup>25</sup> +51.4 ° (c = 0.432, CH<sub>3</sub>COCH<sub>3</sub>).

(-)-3-Diethylamino-2, N-dimethyl-N-(3-phenyl-5-isoxazolyl) propanamide (5y) was prepared in the same manner in 82.9% yield from 15b. Compound 5y gave IR, <sup>1</sup>H-NMR, MS data identical with those of 5x. [ $\alpha$ ]<sub>D</sub><sup>25</sup> -64.6° (c = 0.608, CHCl<sub>3</sub>).

Compound 5y was converted into the fumarate in a usual manner, mp 107.5—108.5 °C.  $[\alpha]_D^{25}$  -48.1 °  $(c = 0.563, CH_3COCH_3)$ .

### Pharmacology

All pharmacological procedures described in this report were carried out in the manner previously described.<sup>1)</sup>

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# References and Notes

- 1) Part I: T. Tatee, S. Kurashige, A. Shiozawa, K. Narita, M. Takei, S. Ito, H. Miyazaki, H. Yamanaka, M. Mizugaki, T. Sakamoto, and H. Fukuda, *Chem. Pharm. Bull.*, 34, 1634 (1986).
- 2) This work was presented in part at the 3rd Symposium on Medicinal Chemistry, Osaka, November 1981.
- 3) J. Yamawaki, T. Ando, and T. Hanafusa, Chem. Lett., 1981, 1143 and references cited therein.
- 4) N. N. Share and C. S. McFarlane, Neuropharmacol., 14, 675 (1975).

a) Yield of free base from 8, 10, or 11. b) mp of fumarate (recrystallized from acetone).

- 5) C. Hansch and A. J. Leo, "Substituent Constants for Correlation Analysis in Chemistry and Biology," John Wiley and Sons, Inc., New York, 1979.
- 6) Regression analysis calculation was carried out on a FACOM M-170F computer.
- 7) pED<sub>50</sub> calculated from Eq. 8 was plotted on a two-dimensional plane using a SORD M243 computer.
- 8) Compound 3b was prepared according to the procedure of Shaw and Sugowdz with a slight modification; G. Shaw and G. Sugowdz, J. Chem. Soc., 1954, 665.
- 9) a) G. Griss and H. Machleidt, *Justus Liebigs Ann. Chem.*, **738**, 60 (1970); b) H. Fujita, R. Endo, A. Aoyama, and T. Ichii, *Bull. Chem. Soc. Jpn.*, **45**, 1846 (1972).
- 10) H. Kano and Y. Makisumi, Yakugaku Zasshi, 76, 1311 (1956).
- 11) R. B. Moffett, R. E. Strube, and L. Skaletzky, J. Med. Chem., 14, 1088 (1971).
- 12) Y. Iwakura, M. Sato, and Y. Matsuo, Nippon Kagaku Kaishi, 80, 502 (1959).
- 13) a) P. Bieber, Ann. Chim. (Paris), 9, 674 (1954); b) R. T. Coutts, J. W. Hubbard, K. K. Midha, and K. Prasad, J. Pharm. Sci., 60, 28 (1971).