Communications to the Editor

Chem. Pharm. Bull. 34(5)2294—2297(1986)

ABSOLUTE CONFIGURATIONS OF REHMAIONOSIDES A, B, AND C AND REHMAPICROSIDE
THREE NEW IONONE GLUCOSIDES AND A NEW MONOTERPENE GLUCOSIDE
FROM REHMANNIAE RADIX

Masayuki Yoshikawa, Youichi Fukuda, Toshio Taniyama, Bae Cheon Cha, and Isao Kitagawa*

Faculty of Pharmaceutical Sciences, Osaka University, 1-6, Yamada-oka, Suita, Osaka 565, Japan

Three new ionone glucosides, rehmaionosides A (1), B (2), and C (3), and a new monoterpene glucoside rehmapicroside (4), were isolated from Chinese Rehmanniae Radix, the dried root of Rehmannia glutinosa Libosch. [kan-jiō (in Japanese) from China]. Their absolute configurations were determined on the basis of chemical and physicochemical evidence which included the application of the exciton chirality method for allylic benzoyl derivatives.

KEYWORDS — Rehmanniae Radix; Rehmannia glutinosa; Scrophulariaceae; rehmaionoside A; rehmaionoside B; rehmaionoside C; rehmapicroside; allylic benzoate exciton chirality method; ionone glucoside; monoterpene glucoside

During the course of chemical characterization studies of crude drug processing, 1) we have investigated the chemical constituents of Chinese Rehmanniae Radix, the dried root of Rehmannia glutinosa Libosch. (Scrophulariaceae) [kan-jiō in Japanese]. We have so far isolated from the Radix various new constituents such as four iridoids (rehmaglutins A, B, C, and D), one iridoid glucoside (glutinoside), three ionone glucosides (rehmaionosides A, B, and C), and one monoterpene glucoside (rehmapicroside), together with eight known glycosides. In our recent papers, we reported the absolute stereostructure of rehmaglutins A, 2) B, 2) C, 3) and D²⁾ and glutinoside. 3) This paper deals with the evidence for the absolute configurations of rehmaionosides A (1), B (2), and C (3) and rehmapicroside (4). 4,5)

Rehmaionoside A (1), $C_{19}H_{34}O_{8} \cdot 2H_{2}O$, $^{6)}$ [α] 20 $^{-49.3}$ ° (MeOH), IR (KBr) cm $^{-1}$:3400, 2924, 1633, 1075, SIMS (Xe $^{+}$, glycerol matrix) m/z: 391 (M+H) $^{+}$, 413 (M+Na) $^{+}$, 483 (M+H+glycerol) $^{+}$, rehmaionoside B (2), $C_{19}H_{34}O_{8} \cdot 2H_{2}O$, [α] 20 $^{-54.2}$ ° (MeOH), IR (KBr) cm $^{-1}$: 3400, 2926, 1631, 1074, SIMS m/z: 391 (M+H) $^{+}$, 413 (M+Na) $^{+}$, 483 (M+H+glycerol) $^{+}$, and rehmaionoside C (3), mp 217-218°C (prisms), $C_{19}H_{32}O_{8}$, [α] 24 $^{-59.4}$ ° (MeOH), IR (KBr) cm $^{-1}$: 3270, 1680, 1059, SIMS m/z: 389 (M+H) $^{+}$, 411 (M+Na) $^{+}$, 481 (M+H+glycerol) $^{+}$, were indicated as glucosides of ionone derivatives by their 1 H and 13 C NMR data (Table I). Oxidation of 1 2 and 2 2 with 20 2 with 20 3 selectively, whereas reduction of 3 3 with NaBH $_{4}$ in MeOH yielded 1 3 and 2 2 in a 1:1 ratio.

Rehmaionoside C (3) had an α,β -unsaturated carbonyl moiety and a tertiary hydroxyl group as indicated by the UV [λ max (MeOH): 232 nm (ϵ 10700)] and 1 H and

¹³C NMR (Table I) data. Acetylation of 3 with Ac₂O-pyridine gave the pentaacetate (3a), mp 183-184°C (needles), C₂₉H₄₂O₁₃, IR (CHCl₃) cm⁻¹: 2938, 1751, 1674, 1241, 1036, UV λmax (MeOH): 228 nm (ε 16600). The ¹H NMR spectrum (500 MHz, CDCl₃) showed signals assignable to four methyl groups (δ 0.81, 1.08, 1.18, 2.29; 3H each, all s), two olefinic protons (δ 6.19, 7.29; lH each, both d, J=16.5 Hz), and one β-anomeric proton (δ 4.70, d, J=7.9 Hz). Hydrolysis of 3 with β-glucosidase afforded an ionone derivative (5), mp 115-116°C (prisms), C₁₃H₂₂O₃, [α]_D -53.6° (EtOH), IR (CHCl₃) cm⁻¹: 3600, 3450, 1670, 1620. Detailed comparison of the physical data for 5 with those for its optical antipode (5') indicated the 1'R, 2'R configurations of 5. Furthermore, the glycosidation shift (between 3 and 5) and the acetylation shift (between 3 and 3a) observed in the ¹³C NMR data led us to formulate the stereostructure of rehmaionoside C (3) as shown. 8)

Methanolysis of rehmaionosides A (1) and B (2) under mild conditions $(4.5 \, ^{\circ} \, ^{\circ}$

Hydrolysis of rehmaionosides A (1) and B (2) with β -glucosidase afforded 9 (from 1), mp 88-90°C (prisms), $C_{13}^{H}_{24}^{Q}_{03}$, $[\alpha]_{D}^{28}$ -26.4° (MeOH), IR (CHCl₃) cm⁻¹:3610, 3445, 2933, 1600, and 10 (from 2), mp 111-113°C (prisms), $C_{13}^{H}_{24}^{Q}_{03}$, $[\alpha]_{D}^{28}$ -39.6° (MeOH), IR (CHCl₃) cm⁻¹: 3610, 3445, 2933, 1600, respectively. On the other hand, 9 and 10 were obtained in a 1:1 ratio from 5 by NaBH₄ reduction in MeOH. Thus, the structures of rehmaionosides A (1) and B (2) were elucidated except the C₂ configuration.

The C2 absolute configurations of rehmaionosides A (1) and B (2) were determined by the application of the allylic benzoate exciton chirality method. 10) NaBH, reduction of rehmaionoside C pentaacetate (3a) yielded la and 2a in 1:1 Respective benzoylation of la and 2a with benzoyl chloride-pyridine afforded $\underset{\sim}{\text{lb}}$ (from $\underset{\sim}{\text{la}}$), colorless oil, $\text{C}_{36}^{\text{H}}_{48}^{\text{O}}_{14}$, UV λ max (MeOH): 228 nm (ϵ 14100), IR (CHCl₃) cm⁻¹: 2935, 1753, 1716, 1600, 1272, 1034, 1 H NMR (500 MHz, CD₃OD) δ : 0.78, 0.90 [3H each, both s, $6'-(CH_3)_2$], 1.16 (3H s, $2'-CH_3$), 1.47 (3H d, J=6.4 Hz, 1-H₃), 5.65 (1H dq, J=6.1, 6.4 Hz, 2-H), 5.80 (1H dd, J=6.1, 15.9 Hz, 3-H), 6.34 (lH d, J=15.9 Hz, 4-H), and $\stackrel{2b}{\approx}$ (from $\stackrel{2a}{\approx}$), colorless oil, $C_{36}H_{48}O_{14}$, UV λ max (MeOH): 229 nm (ϵ 13000), IR (CHCl₃) cm⁻¹: 2935, 1753, 1728, 1600, 1272, 1040, ¹H NMR (500 MHz, CD₃OD) δ : 0.79, 0.94 [3H each, both s, 6'-(CH₃)₂], 1.19 (3H s, 2'-CH₃), 1.47 (3H d, J=6.4 Hz, $1-H_3$), 5.65 (1H dq, J=5.8, 6.4 Hz, 2-H), 5.85 (1H dd, J=5.8, 15.9 Hz, 3-H), 6.34 (1H d, J=15.9 Hz, 4-H). Deacylation of $\frac{1}{2}b$ and $\frac{2}{2}b$ with 1%NaOMe-MeOH regenerated parent 1 and 2. Thus, the structures of 1b and 2b were corroborated. The CD spectrum (MeOH) of \underline{lb} gave a positive first Cotton curve: [θ] $_{226}$ +13100, whereas that of $\frac{2b}{2}$ gave a negative first Cotton curve: [0] $_{226}$ -15200. Therefore, the 2S configuration in 1b and the 2R configuration in 2b have been substantiated.

Based on the above evidence, the absolute configurations of rehmaionosides A (1) and B (2) were determined as shown.

Table I. ^{13}C NMR Data for $\frac{1}{2}$, $\frac{2}{2}$, $\frac{3}{2}$, $\frac{3}{2}$, $\frac{4}{2}$, $\frac{4}{2}$, $\frac{4}{2}$, $\frac{5}{2}$, $\frac{9}{2}$, $\frac{10}{2}$, and $\frac{11}{2}$ (in d_5 -pyridine)

Carbon	1	2	3	3a	5_	2	10
1	24.8 (q) ^{a)}	24.7 (q)	25.8 (q)	25.0 (q)	26.9 (q)	24.9 (q)	25.0 (q)
2	68.1 (d)	68.1 (d)	198.3 (s)	198.3 (s)	197.7 (s)	68.3 (d)	68.4 (d)
3	130.8 (d)	130.9 (d)	131.9 (d)	132.0 (d)	131.1 (d)	130.9 (d)	130.9 (d)
4	135.5 (d)	135.1 (d)	151.9 (d)	149.9 (d)	151.4 (d)	135.9 (d)	135.9 (d)
1'	79.4 (s)	79.1 (s)	79.7 (s)	83.9 (s)	79.8 (s)	79.1 (s)	79.1 (s)
2'	82.9 (s)	82.7 (s)	82.6 (s)	79.2 (s)	74.5 (s)	74.8 (s)	74.8 (s)
3'	32.0 (t)	32.0 (t)	32.0 (t)	32.6 (t)	36.5 (t)	37.0 (t)	37.0 (t)
4'	18.3 (t)	18.2 (t)	18.2 (t)	18.1 (t)	18.4 (t)	18.9 (t)	18.9 (t)
5'	36.9 (t)	36.7 (t)	36.6 (t)	36.0 (t)	36.8 (t)	37.2 (t)	37.3 (t)
6'	38.9 (s)	38.7 (s)	39.4 (s)	39.1 (s)	39.0 (s)	38.7 (s)	38.8 (s)
2'-CH ₃	22.8 (q)	22.4 (q)	22.8 (q)	21.5 (q)	25.5 (q)	25.5 (q)	25.6 (q)
6'-СН	25.6 (q)	25.5 (q)	27.0 (q)	26.4 (q)	27.3 (q)	27.6 (q)	27.6 (q)
5 03	27.8 (q)	27.6 (q)	27.6 (q)	27.6 (q)	27.6 (q)	27.6 (q)	27.7 (q)

Carbon	<u>4</u>	4a ≈	4 <u>b</u>	빘
1	140.5 (s)	142.0 (s)	139.1 (s)	137.4 (s)
2	129.8 (s)	128.9 (s)	132.5 (s)	135.9 (s)
3	73.1 (d)	76.4 (d)	73.1 (d)	68.1 (d)
4	24.0 (t)	25.3 (t)	24.2 (t)	29.4 (t)
5	33.7 (t)	33.8 (ť)	34.0 (t)	35.2 (t)
6	32.7 (s)	33.5 (s)	33.3 (e)	33.8 (s)
1-COO-	171.7 (s)	170.2 (s)	170.2 (s)	170.7 (s)
2-CH ₃	17.9 (q)	18.5 (q)	18.2 (q)	18.3 (q)
6_CU	26.9 (q)	27.0 (q)	27.1 (q)	27.8 (q)
6-CH ₃	27.9 (q)	28.7 (q)	28.0 (q)	28.3 (q)

a) The characterization of each carbon signal was made by INEPT (Insensitive Nuclei Enhanced by Polarization) and the off-resonance experiments.

Rehmapicroside (4), mp 127-129°C (prisms), $C_{16}^{H}_{26}^{O}_{8}^{\circ}_{12}^{H}_{20}^{O}$, [α] $_{D}^{20}_{12}^{+8.5}^{\circ}$ (MeOH), SIMS m/z: 347 (M+H) $^{+}$, 369 (M+Na) $^{+}$, 439 (M+H+glycerol) $^{+}$, IR (KBr) cm $^{-1}$: 3405, 1691, 1637, 1073, was shown to be a glucoside having one olefinic methyl group, two tertiary methyl groups, one tetrasubstituted olefin moiety, and one α , β -unsaturated carboxyl group by the 1 H and 13 C NMR (Table I) data. Ordinary acetylation of 4 yielded the tetraacetate (4a), mp 146-148°C (prisms), $C_{24}^{H}_{34}^{O}_{12}^{O}$, IR (CHCl $_{3}^{O}$) cm $^{-1}$: 1746, 1698, 1239, 1030, while CH $_{2}^{N}_{2}$ treatment of 4 gave the methyl ester (4b), hygroscopic powder, 11 SIMS m/z: 361 (M+H) $^{+}$, 383 (M+Na) $^{+}$, 453 (M+H+glycerol) $^{+}$, IR (KBr) cm $^{-1}$: 3400, 1722, 1071. Hydrolysis of 4b with β -glucosidase liberated 11,

colorless oil, $C_{11}H_{18}O_3$, $[\alpha]_D^{20}$ +53.8° (CHCl₃), IR (CCl₄) cm⁻¹: 3420, 1718, 1216, 1059.

Detailed comparison of the ^{13}C NMR data (Table I) for 4 , 4 a, 4 b, and 11 has led us to conclude that rehmapicroside (4) is the allylic β -glucoside of 11-acid. Treatment of $\frac{4}{\infty}$ with 9% HCl-dry MeOH (reflux for 2 h) liberated $\frac{12}{\infty}$ (racemic) and Oxidation of 11 with CrO_3 -pyridine gave the known enone (13). methyl glucoside. Thus, the structure 11 was substantiated.

Finally, the allylic benzoate exciton chirality method was applied to determine the C_3 absolute configuration of rehmapicroside (4). Benzoylation of 11 with benzoyl chloride-pyridine furnished $\frac{1}{100}$, colorless oil, $c_{18}^{\rm H}_{22}^{\rm O}_4$, UV λ max (EtOH): 229 nm (ε 14600), IR (CC1₄) cm⁻¹: 1717, 1595, 1268, 1101. The CD spectrum of 11a gave a positive first Cotton curve: $[\theta]_{229}$ +36500. configuration of lla was determined.

The absolute stereostructure $\frac{4}{2}$ of rehmapicroside was further confirmed by its partial synthesis from α -ionone. The enone methyl ester (13), which was prepared from α -ionone, 13) was treated with NaBH₄ in MeOH (0°C, 20 min) to afford $(\pm)-11$. Glycosidation of $(\pm)-11$ with 1-bromo-2,3,4,6-tetra-0acetylglucose and Hg(CN) 2 and subsequent treatment of the reaction product with 0.5% NaOMe-MeOH (r.t., 15 min) followed by HPLC purification provided 4b (30%) and the diastereoisomer (14, 31%). Hydrolysis of 4b with 10% KOH-aq. MeOH (60°C, 12 h) furnished rehmapicroside (4, 69%).

Based on the above evidence, the absolute configuration of rehmapicroside (4) was determined as shown.

REFERENCES AND NOTES

- 1) I. Kitagawa, Z. L. Chen, M. Yoshihara, and M. Yoshikawa, Yakugaku Zasshi, 104 867 (1984), and the preceding papers.
- 2) I. Kitagawa, Y. Fukuda, T. Taniyama, and M. Yoshikawa, Chem. Pharm. Bull., in the press.
- 3) M. Yoshikawa, Y. Fukuda, T. Taniyama, and I. Kitagawa, Chem. Pharm. Bull., in the press.
- 4) M. Yoshikawa, T. Taniyama, Y. Fukuda, B. C. Cha, and I. Kitagawa, presented at the Annual Meeting of the Japanese Society of Pharmacognosy, Okayama, Oct. 1985, Abstract Papers, p. 34.
- 5) Rehmaionosides A (1), B (2), and C (3) and rehmapicroside (4) were isolated from the Radix in 0.002%, 0.006%, 0.001%, and 0.002% yields, respectively. Rehmaionosides A and B were obtained as hygroscopic white powders.
- 6) The molecular composition of the compound given with the chemical formula was
- determined by elemental analysis or by high resolution mass spectrometry.

 7) W. Eschenmoser, P. Uebelhart, and C. H. Eugster, Helv. Chim. Acta, 64, 2681 (1981).
- 8) a) The plane structure of rehmaionoside C (3) is identical with that proposed for hydroxy-β-ionone glucoside which was isolated from Aeginetia indica L. var. gracilis Nakai^{8b}; b) T. Endo, H. Taguchi, H. Sasaki, and I. Yosioka, Chem. Pharm. Bull., 27, 2807 (1979).
- 9) Detailed comparison of the physical data for 7 and 8 has indicated that both compounds are epimeric at the C₂ configuration. However, their absolute configurations have not yet been determined.

 10) N. Harada and K. Nakanishi, "Circular Dichroic Spectroscopy-Exciton Coupling in Organic Stereochemistry-," Tokyo Kagaku Dojin, Tokyo, 1982.

 11) Due to its intense hygroscopic nature, satisfactory analytical data for this compound could not be obtained

- compound could not be obtained.
 12) J. B. Heather, R. S. D. Mittal, and C. J. Sih, J. Am. Chem. Soc., 98, 3611 (1976).
- 13) D. W. Brooks, H. S. Bevinakatti, E. Kennedy, and J. Hathaway, J. Org. Chem., 50, 628 (1985).

(Received March 17, 1986)