

Communications to the Editor

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(+)-13 β -HYDROXYMAMANINE, A NEW LUPIN ALKALOID FROM
MAACKIA AMURENSIS VAR. BUERGERI¹⁾

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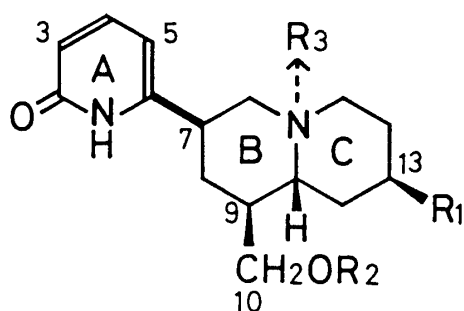
A new lupin alkaloid, (+)-13 β -hydroxymamanine, was isolated from the stems of Maackia amurensis var. buengeri (Leguminosae). Its structure was determined by spectroscopic analysis and by chemical transformation to its acetate and N-oxide.

KEYWORDS — lupin alkaloid; quinolizidine alkaloid; (+)-13 β -hydroxymamanine; 10,13-diacetyl-13 β -hydroxymamanine; 13 β -hydroxymamanine N-oxide; (-)-baptifoline; (+)-mamanine; biogenesis; Maackia amurensis; Leguminosae

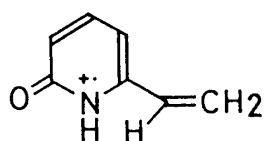
As a result of screening plants belonging to the Leguminosae for lupin alkaloids, a new alkaloid, (+)-13 β -hydroxymamanine 1, was isolated from Maackia amurensis Rupr. et Maxim. var. buengeri C. K. Schn. (Leguminosae), which is a deciduous tree widely distributed in Eastern Asia. In the present communication, we report the structure elucidation of 1 by spectroscopic data of this compound, its acetate 2 and N-oxide 3.

The new alkaloid 1 was isolated from the fresh stems of M. amurensis by repeated silica gel chromatography in a yield of 0.007% of the fresh weight as a colorless amorphous solid, $[\alpha]_D^{17} +31.2^\circ$ ($c=0.107$, CH_3OH).²⁾ The molecular formula, $\text{C}_{15}\text{H}_{22}\text{N}_2\text{O}_3$ (M^+ , m/z 278.1651, calc. 278.1629), was established by high resolution electron impact mass spectrometry (HREIMS). The IR spectrum of 1 showed the bands 3100-3500, 2800-3000, 1650, 1610 and 1550 cm^{-1} and the UV spectrum revealed absorption maxima at 304 ($\log \epsilon=3.82$) and 226 nm ($\log \epsilon=3.83$). Both of these indicate the presence of a 2-pyridone moiety in the molecule.^{3,4)}

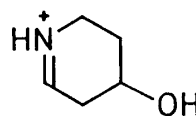
In the HREIMS, fragment peaks were observed at m/z 261 (7%), 260 (6), 247(9) and 229 (13). These peaks were assigned to M^+-OH , $\text{M}^+-\text{H}_2\text{O}$, $\text{M}^+-\text{CH}_2\text{OH}$ and $\text{M}^+-\text{CH}_2\text{OH}-\text{H}_2\text{O}$, respectively, suggesting the presence of hydroxy and hydroxymethyl groups in the molecule. The predominant fragment peaks at m/z 121 (64%) and 100 (base peak) were assigned by HREIMS to 6 ($\text{C}_7\text{H}_7\text{NO}$) and 7 ($\text{C}_5\text{H}_{10}\text{NO}$). These data indicate that 1 has a 2-pyridone moiety and quinolizidine ring substituted with hydroxy and hydroxymethyl groups. Thus, compound 1 is assumed to be a hydroxy



- 1: R₁=OH, R₂=H, R₃=lone pair
2: R₁=OCOCH₃, R₂=COCH₃, R₃=lone pair
3: R₁=OH, R₂=H, R₃=O
4: R₁=H, R₂=H, R₃=lone pair
5: R₁=H, R₂=H, R₃=O



6: m/z 121 (C₇H₇NO)



7: m/z 100 (C₅H₁₀NO)

Table. ¹³C-NMR Data of the Compounds 1 to 5 (67.8 MHz)

Carbon number	Chemical shift (ppm)		Difference of chemical shift Δ(1-4)	Chemical shift (ppm)		
	<u>1</u> ^{a)}	<u>4</u> ^{b)}		<u>2</u> ^{b)}	<u>3</u> ^{a)}	<u>5</u> ^{c)}
2	166.5 (s)	165.3 (s)	+1.2	165.6 (s)	166.4 (s)	164.8 (s)
3	118.2 (d)	117.5 (d)	+0.7	117.8 (d)	118.9 (d)	118.9 (d)
4	143.7 (d)	141.8 (d)	+1.9	141.7 (d)	143.7 (d)	141.8 (d)
5	105.2 (d)	104.2 (d)	+1.0	103.6 (d)	105.8 (d)	104.3 (d)
6	152.7 (s)	151.1 (s)	+1.6	150.9 (s)	150.6 (s)	148.2 (s)
7	40.5 (d)	39.9 (d)	+0.6	39.5 (d)	36.0 (d)	34.6 (d)
8	32.8 (t)	32.4 (t)	+0.4	29.4 (t)	31.6 (t)	31.5 (t)
9	44.6 (d)	43.4 (d)	+1.2	40.7 (d)	39.2 (d)	37.9 (d)
10	63.9 (t)	63.8 (t)	+0.1	65.4 (t)	63.1 (t)	62.5 (t)
11	58.6 (d)	63.0 (d)	-4.4	57.9 (d)	68.1 (d)	73.0 (d)
12	37.0 (t)	29.5 (t)	+7.5	33.8 (t)	33.9 (t)	22.9 (t)
13	64.7 (d)	24.4 (t)	+40.3	67.6 (d)	63.0 (d)	23.2 (t)
14	34.8 (t)	25.7 (t)	+9.1	33.1 (t)	28.2 (t)	20.1 (t)
15	51.0 (t)	56.6 (t)	-5.6	50.4 (t)	64.4 (t)	69.2 (t)
17	60.9 (t)	60.6 (t)	+0.3	60.2 (t)	71.0 (t)	70.9 (t)
-O-CO-	-	-	-	170.2 (s)	-	-
-CH ₃	-	-	-	170.9 (s)	-	-
	-	-	-	20.9 (q)	-	-
	-	-	-	21.3 (q)	-	-

a) In CD₃OD.

b) In CDCl₃.

c) In 5% CD₃OD - 95% CDCl₃, data from ref. 6.

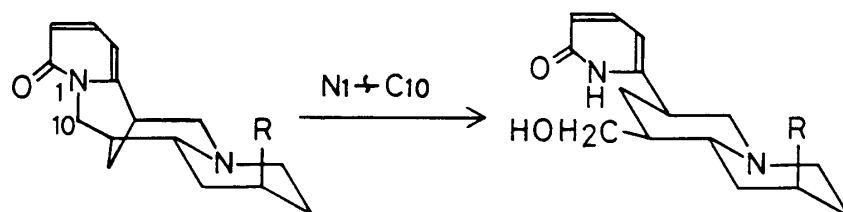
derivative of (+)-mamanine 4, which had been isolated from *Sophora chrysophylla*^{5,6)} and *S. flavescens*.⁷⁾

Signals appearing in ¹³C- and ¹H-NMR spectra were assigned by selective proton decoupling and homonuclear decoupling experiments. In the ¹³C-NMR spectrum of 1, the signals corresponding to the carbons of ring A and B coincided with those of 4 (Table). This indicates that the relative stereochemistry of the substituents on ring B in 1 is the same as in 4. The remaining five signals were assigned to C-11 to C-15 by considering the substitution effects of a hydroxy group. The position and configuration of the hydroxy group was determined to be 13β (axial) on the basis of the shift effect of a hydroxy residue on decaline⁸⁾ and 13-hydroxylupanine.⁹⁾ The alpha effects of an axial and an equatorial hydroxy group are reported to be low-field shifts of ca. 40 ppm and ca. 44 ppm, respectively.^{8,9)} The low-field shift of 40.3 ppm of C-13 supported the existence of an axial hydroxy group. The beta and gamma effects also showed good agreement with those reported for an axial configuration. Therefore, the hydroxy group of 1 was determined to be in the 13β (axial) position.

The structure of 1 was confirmed by the ¹H-NMR spectrum (270 MHz, CD₃OD). The olefinic protons of C-3, C-4 and C-5 resonated at δ 6.38 ppm (1H, d, J=9.5), 7.52 (1H, dd, J=9.5 and 7) and 6.25 (1H, d, J=7), respectively. The signal at δ 4.09 (1H, quintet, J=3) was assigned to an equatorial proton on C-13. The two protons on C-10 exhibited the AB parts of the ABX pattern (J_{A,B}=11, J_{A,X}=2.5 and J_{B,X}=5) centered at δ 3.52 and 3.62. The signal at δ 2.84 (1H, tt, J=11 and 4) was assigned to an axial proton at C-7.

For further confirmation of the structure of 1, the 10,13-diacetate 2¹⁰⁾ and the N-oxide 3¹¹⁾ were synthesized from 1. The substituent effects on the ¹³C-chemical shifts of 2 induced by 13β-acetoxy group showed good agreement with those reported in 13(axial)-acetoxy sparteine.⁹⁾ In the ¹³C-NMR spectrum of 3, the low-field shifts of C-11, C-15, C-17 in the range of 9 to 14 ppm and the up-field shifts of C-7, C-9, C-12, C-14 in the range of 3 to 7 ppm were induced by N-oxidation. These substituent effects of N-oxide were also observed in the spectrum of maminine N-oxide 5.⁶⁾ From these several lines of evidence, the structure of the new base was determined to be (+)-13β-hydroxymamanine 1 or its enantiomer.

The structure of 1 corresponds to an oxidative product derived from the N₁-C₁₀ cleavage of (-)-baptifoline 8 coexisting in *M. amurensis*. Similar relationships in the structures involving oxidative bond cleavage have been found between (+)-mamanine 4 and (-)-anagryne 9, (+)-kuraramine and (-)-N-methylcytisine, and (-)-pohakuline and (-)-lupanine.^{6,7)} Biosynthetic relations are suggested between these bridged quinolizidine alkaloids and the unbridged bases.



8: R=OH; (-)-baptifoline

9: R=H; (-)-anagryne

1: R=OH

4: R=H

Further studies of the absolute configuration and the biosynthetic relations of these alkaloids are being undertaken in our laboratories.

REFERENCES AND NOTES

- 1) Studies on Plant Constituents of Genus Maackia. Part I.
- 2) 1 was isolated from 75% EtOH extracts of the fresh stems of M. amurensis harvested in September 1984 at Medicinal Plant Gardens, Chiba University, together with eight known alkaloids, (-)-cytisine (main base), (-)-anagyrine, (-)-N-methylcytisine, (-)-lupanine, (-)-baptifoline, N-formylcytisine, ammodendrine, and camoensidine.
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- 9) F. Bohlmann and R. Zeisberg, Chem. Ber., 108, 1043 (1975).
- 10) 2 was obtained by acetylation of 1 with Ac₂O/pyridine. Colorless amorphous solid, MS (70 eV): m/z (rel.%) 362 (5, M⁺), 302 (24, M⁺-CH₃COOH), 243 (100), 229 (7), 202 (44), 121 (39), 96 (34), 82 (45), 43 (34). IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 2750-2850 (trans-quinolizidine), 1730, 1650, 1620, 1250. ¹³C-NMR in Table. ¹H-NMR (270 MHz, CDCl₃) δ : 1.48 (2H, m), 1.70-2.00 (4H, m), 2.05 (3H, s, -COCH₃), 2.08 (3H, s, -COCH₃), 2.09-2.50 (4H, m), 2.69 (1H, dt, J=12 and 5), 2.88-3.07 (2H, m), 4.01 (1H, dd, J=11 and 6.5, H-10), 4.09 (1H, dd, J=11 and 4, H-10), 5.11 (1H, m, W_{1/2}=7.5, H-13), 6.07 (1H, d, J=6.5, H-5), 6.43 (1H, d, J=9, H-3), 7.39 (1H, dd, J=9 and 6.5, H-4), 12.3 (1H, br, H-1).
- 11) The oxidation of 1 with m-chloroperbenzoic acid in CH₂Cl₂-MeOH gave 3. Colorless needle, mp 207-210° (benzene-EtOH). MS (20 eV): m/z (rel.%) 294 (3, M⁺), 278 (62, M⁺-O), 277 (23, M⁺-OH), 276 (77, M⁺-H₂O), 220 (43), 121 (64), 100 (100). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 1650, 1610, 1000. ¹³C-NMR in Table. ¹H-NMR (270 MHz, CD₃OD) δ : 1.61-2.20 (4H, m), 2.52 (1H, br t, J=13.5), 3.09 (1H, br d, J=11), 3.25 (1H, br d, J=13.5), 3.38-3.80 (5H, m), 4.10 (1H, m, W_{1/2}=7.5, H-13), 6.29 (1H, d, J=7, H-5), 6.42 (1H, d, J=9.5, H-3), 7.53 (1H, dd, J=9.5 and 7, H-4).

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