

Studies on Quinolizine Derivatives. XXIV.¹⁾ Catalytic Hydrogenation of Cycl[3.3.3]azines

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The catalytic hydrogenation of the cycl[3.3.3]azine derivative (6) or 1-azacycl[3.3.3]azine derivative (15) with PtO_2 in tetrahydrofuran at atmospheric pressure gave the dihydrocylazine derivative (10) or dodecahydro-1-azacycl[3.3.3]azine derivative (16), respectively. On the other hand, the catalytic hydrogenation of 6 or 19 in AcOH gave dodecahydrocylazine (11) or dodecahydro-2-hydroxymethyl-1-azacyclazine (21), respectively. The catalytic hydrogenation of the hydrochloride 18 in MeOH gave methyl dodecahydro-5-methyl-1-azacycl[3.3.3]azine-2-carboxylate (20). Compounds 11, 20 and 21 should be useful as intermediates for the preparation of natural products, such as coccinelline and ceruine.

Keywords hydrocyl[3.3.3]azine; hydroazacycl[3.3.3]azine; cycl[3.3.3]azine; azacycl[3.3.3]azine; antiaromatic character; coccinelline; ceruine

Previously we reported the synthesis and antiaromatic character of unstable 2-methyl-1-azacycl[3.3.3]azine (1), which contains a nitrogen-bridged peripheral 12π electron system.²⁾ The azacycl[3.3.3]azine derivatives (2—5) were rather stable owing to the effect of their electron-attracting groups, but some characteristic reactivities were noted. The methylthio group on the azacyclazine ring in 2, 3 and 4 was highly reactive with nucleophiles, such as amines or active methylene compounds.^{1,3)} Furthermore, the Diels-Alder reaction of azacyclazine (5) with a dienophile, methyl acetylenecarboxylate (MAC), gave the deaza compounds (6,7).⁴⁾ As part of our continuing studies on azacyclazines, we now wish to report that the catalytic reduction of cyclazines (6, 15, 18, 19) with PtO_2 catalyst readily gave hydrogenated cyclazines (10, 11, 16, 20, 21). Compounds 11, 20 and 21 should be useful as intermediates for the preparation of natural products, such as coccinelline⁵⁾ and

ceruine.⁶⁾

Farquhar *et al.*⁷⁾ reported that the catalytic reduction of diethyl cycl[3.3.3]azine-1,3-dicarboxylate (8) with PtO_2 catalyst in benzene gave the tetrahydrocylazine (9). In the present paper, we describe the catalytic reduction of 1,4-dimethyl 3-ethyl 8-methylcycl[3.3.3]azine-1,3,4-tricarboxylate (6) with PtO_2 catalyst in tetrahydrofuran (THF) at atmospheric pressure to give the dihydrocylazine (10), after absorption of 1 mol eq of hydrogen. The structure 10 was assigned to the dihydrocylazine on the basis of the proton nuclear magnetic resonance ($^1\text{H-NMR}$) spectrum, in which the signals due to $\text{C}_2\text{-H}$, $\text{C}_7\text{-H}$, $\text{C}_9\text{-H}$ were readily identifiable from their multiplicities and the methylene protons ($\text{C}_5\text{-H}$, $\text{C}_6\text{-H}$) gave a complex multiplet at higher field, δ (ppm): 2.7—3.1. On the other hand, when the reduction of 6 was carried out in AcOH, the dodecahydrocylazine (11) was obtained in quantitative yield, after

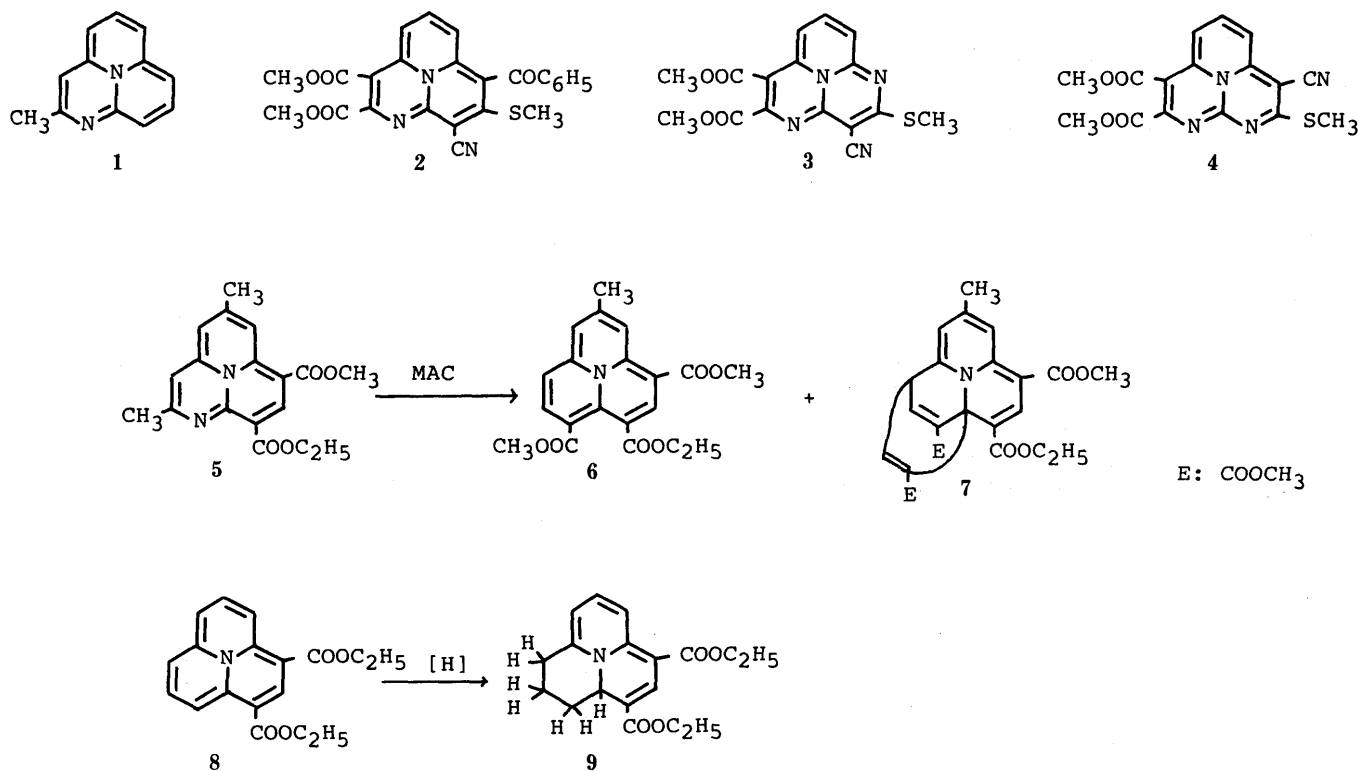
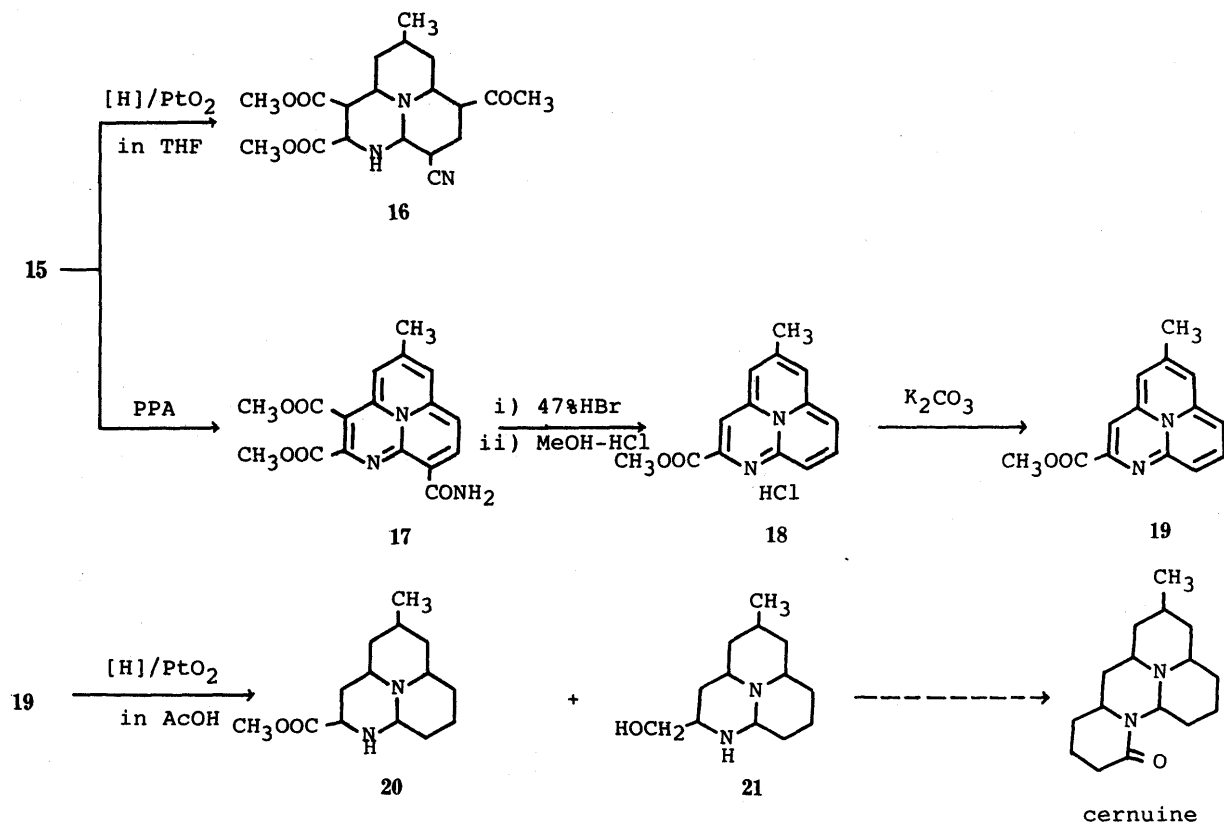
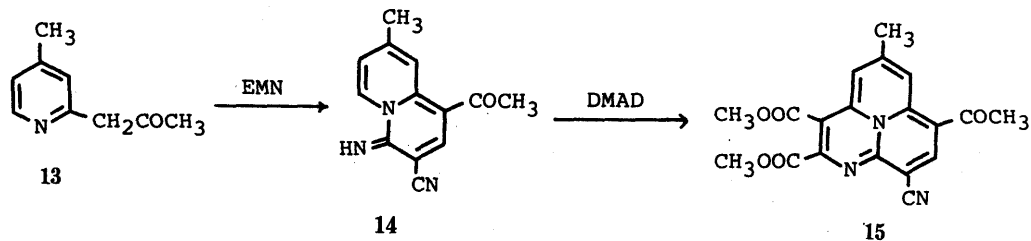
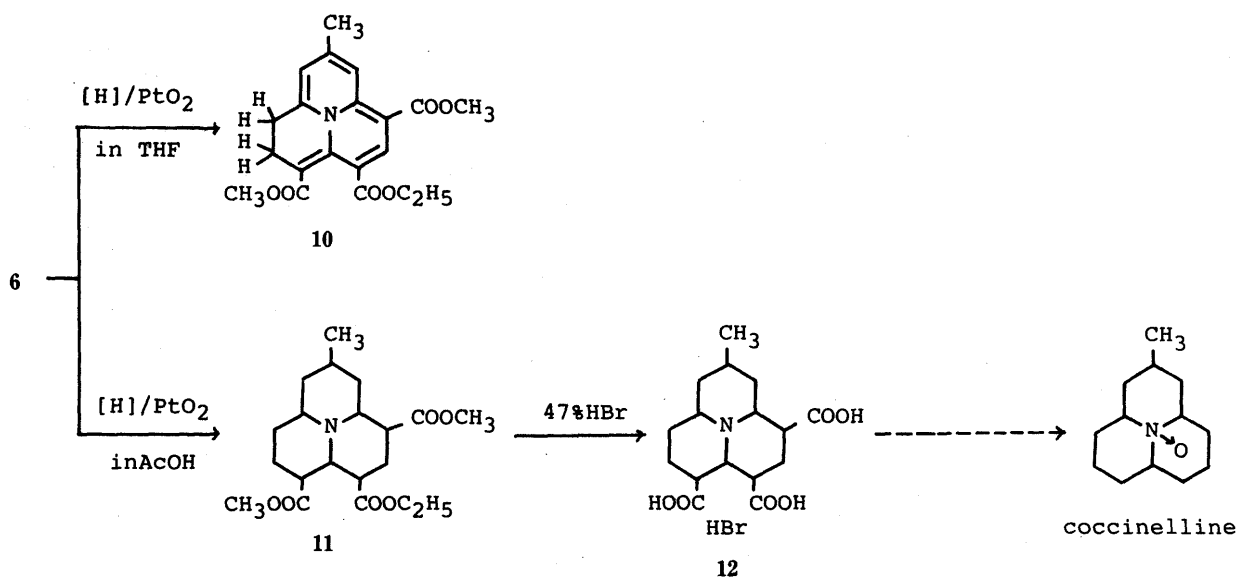


Chart 1



absorption of 6 mol eq of hydrogen. Then, a solution of 11 in 47% HBr was refluxed for 4h to give 1,2,3,3a,4,5,6,6a,7,8,9,9a-dodecahydro-8-methylcyclo-

[3.3.3]azine-1,3,4-tricarboxylic acid hydrobromide (12), as hygroscopic crystals.

Next, we examined the catalytic reduction of the azacy-

clazine derivative (**15**), which was prepared by the reaction of 1-acetyl-3-cyano-4-imino-8-methyl-4*H*-quinolizine (**14**) with dimethyl acetylenedicarboxylate (DMAD), by the method described in the previous paper.¹⁾ Thus, compound **15** was more readily reduced with PtO₂ catalyst in THF at atmospheric pressure to give the dodecahydroazacyclazine (**16**) than **6**. The ¹H-NMR (CDCl₃) spectrum of the reduced product **16** showed no vinyl protons and a sharp doublet at δ (ppm): 0.88 (3H, d, $J=6$ Hz, CH₃). In addition, the signals of two methoxycarbonyl groups (δ : 3.76) and one acetyl group (δ : 2.19) were observed. In order to obtain key intermediates (**20**, **21**) for the synthesis of the cernuine skeleton, we examined various conditions for removal of the methoxycarbonyl, acetyl, and cyano groups on **15**, and succeeded in the isolation of methyl 1-azacycl[3.3.3]azine-2-carboxylate (**19**) as an unstable compound. Thus, a mixture of **15** and polyphosphoric acid (PPA) was heated at 100 °C for 10 h to give the deacetylated compound, dimethyl 9-carbamoyl-5-methyl-1-azacycl[3.3.3]azine-2,3-dicarboxylate (**17**), in good yield. Then, a solution of **17** in 47% hydrobromic acid was refluxed for 4 h to give the hydrobromide, which was in turn converted to the hydrochloride (**18**) by esterification with MeOH-HCl. The free base, methyl 5-methyl-1-azacycl[3.3.3]azine-2-carboxylate (**19**), was obtained as unstable green crystals by treatment of **18** with potassium carbonate solution. The signals of all the ring protons of **19** appear at δ (ppm): 3.63—5.54, at comparatively high magnetic fields. Namely, compound **19** may have an antiaromatic character.²⁾

The catalytic reduction of **19** with PtO₂ in THF gave an unidentified decomposition product, but the reduction of **19** in AcOH for 8 h absorbed ca. 6 mol eq of hydrogen to give the methoxycarbonyl compound (**20**) and the hydroxymethyl compound (**21**). Furthermore, the hydrogenation of **19** in AcOH for 24 h gave only **21** in good yield. On the other hand, the hydrogenation of the hydrochloride **18** in MeOH readily gave **20** in quantitative yield. Further work on the synthesis of coccinelline and cernuine is in progress.

Experimental

Melting points were determined with a Mitamura Mel-Temp and are uncorrected. Infrared (IR) spectra were recorded in KBr discs on a JASCO IRA-2 spectrometer. Ultraviolet (UV) spectra were recorded on a Hitachi EP-S2 spectrometer in 95% ethanol. ¹H-NMR spectra were obtained on a JNM-FX-90 (90 MHz) spectrometer with tetramethylsilane as an internal standard. Chemical shifts of ¹H-NMR signal are given in δ values (ppm). Abbreviations: s=singlet, br=broad, d=doublet, dd=doublet of doublets, t=triplet, and m=multiplet.

1,4-Dimethyl 3-Ethyl 5,6-Dihydro-8-methylcyclo[3.3.3]azine-1,3,4-tricarboxylate (10) Compound **6** (0.001 mol)⁴⁾ was hydrogenated, at atmospheric pressure, in THF (50 ml) containing PtO₂ catalyst (50 mg). Hydrogen (1 mol eq) was absorbed during 8 h. The solution was then filtered and evaporated under reduced pressure and the residue was chromatographed on silica gel in benzene-CHCl₃ (1:1) to yield the dihydro compound **10** (76%), mp 210—212 °C. *Anal.* Calcd for C₂₀H₂₁N₃O₆: C, 64.68; H, 5.70; N, 3.77. Found: C, 64.59; H, 5.76; N, 3.85. UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 223 (4.25) sh, 254 (4.15), 299 (4.14), 338 (3.69) sh, 422 (4.19), 490 (3.82) sh. IR (KBr) cm⁻¹: 1700 (C=O), 1670 (C=O), 1650 (C=O). ¹H-NMR (CDCl₃) δ : 1.31 (3H, t, $J=7$ Hz, OCH₂CH₃), 2.34 (3H, s, CH₃), 2.70—3.10 (4H, m, CH₂CH₂), 3.67 (3H, s, OCH₃), 3.81 (3H, s, OCH₃), 4.23 (2H, q, $J=7$ Hz, OCH₂CH₃), 6.52 (1H, s, C₇-H), 8.28 (1H, s, C₂-H), 8.70 (1H, s, C₉-H).

1,2,3,3a,4,5,6,6a,7,8,9,9a-Dodecahydro-8-methylcyclo[3.3.3]azine-1,3,4-tricarboxylic Acid Hydrobromide (12) Compound **6** (0.001 mol) was hydrogenated, at atmospheric pressure, in AcOH (50 ml) containing PtO₂

catalyst (50 mg). Hydrogen (6 mol eq) was absorbed during 8 h. The solution was filtered and evaporated under reduced pressure and the residue was poured into ice-water (100 ml). The solution was made basic to litmus with K₂CO₃ and extracted with CHCl₃ (3 \times 50 ml). The extract was washed with water (50 ml), dried (Na₂SO₄), and evaporated under reduced pressure to give the oily dodecahydro compound (**11**) in quantitative yield. The ¹H-NMR spectrum of the crude product **11** was recorded. ¹H-NMR (CDCl₃) δ : 0.88 (3H, d, $J=6$ Hz, CH₃), 1.27 (3H, t, $J=7$ Hz, OCH₂CH₃), 3.66 (6H, s, OCH₃), 4.15 (2H, q, $J=7$ Hz, OCH₂CH₃). Without further purification, this oil was refluxed with 47% HBr (30 ml) for 4 h. The solution was evaporated under reduced pressure. The residue was recrystallized from MeOH-benzene to give **12** in 95% yield, mp 190 °C (dec.). *Anal.* Calcd for C₁₆H₂₄BrNO₆: C, 47.30; H, 5.95; N, 3.45. Found: C, 47.06; H, 3.67; N, 3.21. MS m/z : 325 (M⁺ - HBr). IR (KBr) cm⁻¹: 1720 (br) (C=O). ¹H-NMR (DMSO-*d*₆) δ : 0.90 (3H, d, $J=5$ Hz, CH₃).

Dimethyl 7-Acetyl-9-cyano-5-methyl-1-azacycl[3.3.3]azine-2,3-carboxylate (15) A solution of 1-(4-methyl-2-pyridinyl)-2-propanone (**13**) (0.01 mol), ethoxymethylenemalononitrile (EMN) (0.01 mol), and K₂CO₃ (2 g) in dimethylsulfoxide (DMSO) (20 ml) was stirred overnight at room temperature. The reaction mixture was poured into ice-water (300 ml). The solution was extracted with CHCl₃ (3 \times 50 ml), dried (Na₂SO₄) and evaporated under reduced pressure. The residue was recrystallized from CHCl₃-MeOH to give 1-acetyl-3-cyano-4-imino-8-methyl-4*H*-quinolizine (**14**) (95%), mp 236—237 °C. *Anal.* Calcd for C₁₃H₁₁N₃O: C, 69.32; H, 4.92; N, 18.66. Found: C, 69.53; H, 5.03; N, 18.62. UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 226 (4.40), 246 (4.08) sh, 273 (3.90) sh, 283 (3.94), 302 (4.05), 356 (4.14) sh, 374 (4.25), 438 (4.10). IR (KBr) cm⁻¹: 2200 (CN), 1655 (C=O). ¹H-NMR (CDCl₃) δ : 2.53 (3H, s, COCH₃), 2.56 (3H, s, CH₃), 7.17 (1H, dd, $J=2$, 7 Hz, C₇-H), 7.70 (1H, br, NH), 8.04 (1H, s, C₂-H), 9.37 (1H, d, $J=2$ Hz, C₉-H), 9.61 (1H, d, $J=7$ Hz, C₆-H). A solution of **14** (0.01 mol) and DMAD (0.03 mol) in *N,N*-dimethylformamide (10 ml) was heated at 130 °C for 3 h. The reaction mixture was poured into ice-water (300 ml). The solution was extracted with CHCl₃ (3 \times 50 ml). The extract was washed with H₂O (50 ml), dried (Na₂SO₄) and evaporated under reduced pressure. The residue was submitted to column chromatography on silica gel in benzene to yield **15** (37%), mp 232—233 °C (dec.). *Anal.* Calcd for C₁₉H₁₅N₃O₅: C, 62.46; H, 4.14; N, 11.50. Found: C, 62.22; H, 4.27; N, 11.32. IR (KBr) cm⁻¹: 2200 (CN), 1720 (C=O), 1655 (C=O), 1620 (C=O). UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 283 (4.43), 372 (4.13), 412 (3.98) sh, 437 (4.36), 462 (4.56). ¹H-NMR (CDCl₃) δ : 2.00 (3H, s, CH₃), 2.20 (3H, s, COCH₃), 3.71 (3H, s, OCH₃), 3.78 (3H, s, OCH₃), 6.52 (1H, dd, $J=0.5$, 2 Hz, C₄-H), 7.08 (1H, s, C₈-H), 7.81 (1H, dd, $J=0.5$, 2 Hz, C₆-H).

Dimethyl 7-Acetyl-9-cyano-1,2,3,3a,4,5,6,6a,7,8,9,9a-dodecahydro-5-methyl-1-azacycl[3.3.3]azine-2,3-carboxylate (16) Compound **15** (0.001 mol) was hydrogenated, at atmospheric pressure, in THF (50 ml) containing PtO₂ (50 mg). Hydrogen (6 mol eq) was absorbed during 12 h; the color of the solution changed from green to colorless. The solution was filtered and evaporated under reduced pressure. The residue was submitted to column chromatography on silica gel in benzene-CHCl₃ (2:1) to yield **16** (64%), mp 74—75 °C. *Anal.* Calcd for C₁₉H₁₇N₃O₅: C, 60.46; H, 7.21; N, 11.13. Found: C, 60.18; H, 7.46; N, 11.01. IR (KBr) cm⁻¹: 2180 (CN), 1740 (C=O), 1620 (C=O). ¹H-NMR (CDCl₃) δ : 0.88 (3H, d, $J=6$ Hz, CH₃), 2.19 (3H, s, COCH₃), 3.76 (6H, s, OCH₃) \times 2.

Dimethyl 9-Carbamoyl-5-methyl-1-azacycl[3.3.3]azine-2,3-dicarboxylate (17) A mixture of **15** (0.5 g) and an excess of PPA (10 g) was heated at 100 °C for 10 h. The reaction mixture was poured into ice-water (300 ml). The solution was made basic to litmus with K₂CO₃, and extracted with CHCl₃ (3 \times 50 ml). The extract was washed with water (50 ml), dried (Na₂SO₄), and evaporated under reduced pressure to give **17** as crude crystals (78%), which were recrystallized from CHCl₃-MeOH to give green needles, mp 269—271 °C (dec.). *Anal.* Calcd for C₁₇H₁₅N₃O₅: C, 59.82; H, 4.43; N, 12.31. Found: C, 59.68; H, 4.53; N, 12.25. IR (KBr) cm⁻¹: 1740 (C=O), 1700 (C=O), 1655 (C=O). UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 272 (4.43), 382 (4.14), 417 (4.34), 439 (4.49). ¹H-NMR (CDCl₃) δ : 1.77 (3H, s, CH₃), 3.59 (3H, s, OCH₃), 3.73 (3H, s, OCH₃), 5.29 (1H, d, $J=9$ Hz, C₇-H), 5.49 (1H, dd, $J=1$, 2 Hz, C₄-H or C₆-H), 6.29 (1H, dd, $J=1$, 2 Hz, C₄-H or C₆-H), 7.26 (1H, d, $J=9$ Hz, C₈-H), 5.36 (1H, br, NH), 8.71 (1H, br, NH).

Methyl 5-Methyl-1-azacycl[3.3.3]azine-2-carboxylate (19) A solution of **17** (0.5 g) in 47% HBr (20 ml) was refluxed for 3 h. The solution was evaporated under reduced pressure. A solution of the residue in MeOH-HCl (100 ml) was refluxed for 10 h. The solution was evaporated under reduced pressure to give the hydrochloride **18** (67%), mp 136 °C (dec.). A solution of **18** (0.5 g) in water (50 ml) was made basic to litmus with K₂CO₃ and extracted with CHCl₃ (3 \times 30 ml). The extract was dried

(Na_2SO_4) and evaporated under reduced pressure to give **19** as unstable crude crystals (97%), which were carefully recrystallized from CH_2Cl_2 -MeOH to give green needles, mp 168–170°C. *Anal.* Calcd for $\text{C}_{14}\text{H}_{12}\text{N}_2\text{O}_2$: C, 69.99; H, 5.03; N, 11.66. Found: C, 69.87; H, 5.11; N, 11.64. IR (KBr) cm^{-1} : 1710 (C=O). UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 227 (4.21), 256 (4.09) sh, 280 (4.33) sh, 291 (4.35), 306 (4.17) sh, 388 (4.12) sh, 409 (4.18) sh, 434 (4.22). $^1\text{H-NMR}$ (CDCl_3) δ : 1.17 (3H, s, CH_3), 3.59 (3H, s, OCH_3), 3.63 (1H, d, $J=2$ Hz, $\text{C}_4\text{-H}$ or $\text{C}_6\text{-H}$), 4.09 (1H, dd, $J=1, 8$ Hz, $\text{C}_7\text{-H}$ or $\text{C}_9\text{-H}$), 4.27 (1H, s, $\text{C}_3\text{-H}$), 4.32 (1H, dd, $J=1, 9$ Hz, $\text{C}_7\text{-H}$ or $\text{C}_9\text{-H}$), 4.33 (1H, d, $J=2$ Hz, $\text{C}_4\text{-H}$ or $\text{C}_6\text{-H}$), 5.54 (1H, dd, $J=8, 9$ Hz, $\text{C}_8\text{-H}$).

Hydrogenation of 18 and 19 a) Compound **19** (0.001 mol) was hydrogenated, at atmospheric pressure, in AcOH (50 ml) containing PtO_2 (50 mg). Hydrogen (ca. 6 mol eq) was absorbed during 8 h. The solution was then worked up as in the case of **11** to yield methyl 1,2,3,3a,4,5,6,6a,7,8,9,9a-dodecahydro-5-methyl-1-azacycl[3.3.3]azine-2-carboxylate (**20**) (62%) and 1,2,3,3a,4,5,6,6a,7,8,9,9a-dodecahydro-2-hydroxymethyl-5-methyl-1-azacycl[3.3.3]azine (**21**) (7%). b) Compound **19** (0.001 mol) was hydrogenated, at atmospheric pressure, in AcOH (50 ml) containing PtO_2 (50 mg) for 24 h. The solution was then worked up as in a) to yield **21** (98%), mp 141°C. *Anal.* Calcd for $\text{C}_{13}\text{H}_{24}\text{N}_2\text{O}$: C, 69.60; H, 10.78; N, 12.49. Found: C, 69.53; H, 10.85; N, 12.33. MS m/z : 224 (M^+). $^1\text{H-NMR}$ (CDCl_3) δ : 0.93 (3H, d, $J=6$ Hz, CH_3). c) Compound **18** (0.001 mol) was hydrogenated, at atmospheric pressure, in MeOH (100 ml) containing PtO_2 (50 mg) for 24 h. The solution was then worked up as in a) to yield the oily product **20** (98%). MS ($\text{C}_{14}\text{H}_{24}\text{N}_2\text{O}_2$) m/z : 252 (M^+). IR (KBr) cm^{-1} : 1730 (C=O). $^1\text{H-NMR}$ (CDCl_3) δ : 0.88 (3H, d, $J=5$ Hz, CH_3), 3.71 (3H, s, OCH_3).

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