SYNTHESIS OF (-)-ARISTEROMYCIN FROM D-GLUCOSE

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(-)-Aristeromycin, a carbocyclic nucleoside with various biological activities, was synthesized from D-glucose using a Michael-type addition reaction of N^6 -benzoyladenine with a nitrocyclopentene derivative as its key step.

KEYWORDS (-)-aristeromycin; carbocyclic nucleoside antibiotic; <u>pseudo</u>-nucleoside; nitro-cyclopentene; Michael-type addition reaction; <u>pseudo</u>-aminosugar; D-glucose; D-allofuranose

(-)-Aristeromycin (1), a carbocyclic analog of adenosine, exhibits a variety of interesting biological activities such as antibacterial, antifungal, antiviral, and antitumor properties. 1,2) Aristeromycin was first synthesized in a racemic form named C-Ado 3) and subsequently, (-)-aristeromycin (1) was isolated from Streptomyces citricolor nov. sp. as a natural product. 1,4) Later, it was reported that the (+)-enantiomer of 1 was totally inactive in the inhibition of tumor-cell growth and virus replication. 2) Following a pioneering synthesis of (±)-aristeromycin, 3) this class of compounds have been given much attention from the synthetic and biological viewpoints, 5) and in recent years, several enantioselective syntheses of (-)-aristeromycin (1) have been reported. 6)

By use of a stereoselective deacetoxyhydrogenation and a cyclitol formation from nitrofuranose derivatives as key reactions, we have recently developed versatile methods for converting carbohydrates to optically active pseudo-hexopyranoses, pseudo-pentofuranoses, and pseudo-aminosugar. Furthermore, we have devised a new method for synthesizing optically active pseudo-glycosides using a Michael-type addition reaction with a nitro-cyclohexene. As an extension of this study of synthesizing pseudo-glycosides, we have successfully synthesized several optically active carbocyclic analogs of nucleosides such as (+)-cyclaradine and (+)-9-pseudo- β -L-xylofuranosyladenine, which are commonly called pseudo-nucleosides. This paper deals with a synthesis of (-)-aristeromycin (1) from D-glucose.

Swern oxidation 10) of 1,2:5,6-di-O-isopropylidene- $^{\alpha}$ -D-glucofuranose (2) gave a ketone (3, 92%) 11) which was treated with NaBH₄ 11 and benzylated to yield a D-allofuranose derivative (4, 86%), colorless oil, $[\alpha]_D^{23}$ +107° (CHCl₃), $C_{19}H_{26}O_6$, 12 IR $_{\rm max}$ 3 cm⁻¹: 2995, 2940, 1386. Treatment of 4 with 80% aq. AcOH and subsequent selective benzoylation of the product furnished 5 (90%), colorless oil, $[\alpha]_D^{21}$ +64° (CHCl₃), $C_{23}H_{26}O_7$, IR $_{\rm max}$ 3 cm⁻¹: 3524, 2993, 1711, EI-MS (m/z): 414 (M⁺). Swern oxidation of 5 gave an unstable ketone 6 which was immediately treated with CH₃NO₂ in DMF in the presence of KF and 18-crown-6 to provide 7 (90%, a mixture of the 5-epimers), colorless oil, IR $_{\rm max}$ 3 cm⁻¹: 3552, 2990, 1723, 1552, 1375, EI-MS (m/z): 458 (M⁺-CH₃). The nitrofuranose derivatives (7), without separation, were acetylated and subjected to deacetoxy-hydrogenation with NaBH₄ to furnish 8 (67%), colorless oil, IR $_{\rm max}$ 3 cm⁻¹: 1721, 1555, 1378, EI-MS (m/z): 442 (M⁺-CH₃). Removal of the isopropylidene group of 8 with 80% aq. AcOH gave 9 (82%), colorless oil, IR $_{\rm max}$ cm⁻¹: 3449, 1730, 1550, 1374, EI-MS (m/z): 417 (M⁺), which was oxidized with Pb(OAC)₄ to provide an epimeric mixture of aldehydic formates (10). Intramolecular condensation of 10 with KF in DMF in the presence of 18-crown-6 yielded the cyclization products (11). Dehydration of 11 with Ac₂O and p-TsOH·H₂O provided two nitro-cyclopentenes, 12 (31% from 9), colorless oil, $[\alpha]_D^{20}$ +83° (CHCl₃), $C_{21}H_{19}NO_7$, IR $_{\rm max}$ CHCl₃ cm⁻¹: 1728, 1556, 1523, 1345, and 13 (22% from 9), colorless oil, $[\alpha]_D^{24}$ +31° (CHCl₃), $C_{21}H_{19}NO_7$, IR $_{\rm max}$ CHCl₃ cm⁻¹: 1720, 1557, 1522, 1367.

(a) (COC1) $_2$ / DMSO / Et $_3$ N / CH $_2$ Cl $_2$ (-78 °C, 1 h) (b) NaBH $_4$ / 95% aq. EtOH (2 °C, 1.5 h); BnCl / NaH / DMF (25 °C, 1.5 h) (c) 80% aq. AcOH (25 °C, 24 h); BzCl / pyridine / CH $_2$ Cl $_2$ (2 °C, 1 h) (d) CH $_3$ NO $_2$ / KF / 18-crown-6 / DMF (25 °C, 3 h) (e) Ac $_2$ O / p-TsOH·H $_2$ O (25 °C, 3 h); NaBH $_4$ / 95% aq. EtOH (25 °C, 2 h) (f) 80% aq. AcOH (80 °C, 15 h) (g) Pb(OAc) $_4$ / benzene (25 °C, 40 min) (h) KF / 18-crown-6 / DMF (2 °C, 3.5 h) (i) Ac $_2$ O / p-TsOH·H $_2$ O (25 °C, 1.5 h) (j) 28% aq. NH $_4$ OH / 95% aq. EtOH (25 °C, 2 h) [or liq. NH $_3$ / THF (-78 °C, 2 h)]; Ac $_2$ O / p-TsOH·H $_2$ O (25 °C, 2 h) (k) n-Bu $_3$ SnH / AIBN / benzene (80 °C, 3 h) (1) KF / 18-crown-6 / THF (2 °C, 1 h) (m) 28% aq. NH $_4$ OH / 95% aq. EtOH (25 °C, 30 min); \wedge O \wedge / CSA / CH $_2$ Cl $_2$ (25 °C, 1 h) : n-Bu $_3$ SnH / AIBN / toluene (110 °C, 20 min); 10% aq. AcOH / acetone (25 °C, 14 h) (n) 5% NaOMe-MeOH (25 °C, 2 h); Na / 1iq. NH $_3$ / THF (-78 °C, 30 min)

1723, 1696, 1559, 1367, and 15 [31% (or 28%)], colorless oil, [α] $_{D}^{24}$ +14° (CHCl $_{3}$), $C_{23}H_{24}N_{2}O_{8}$, IR ν CHCl $_{3}$ cm $^{-1}$: 1724, 1693, 1557, 1367. The 1 H NMR decoupling experiments in detail (500 MHz) of 14 and 15 resulted in the following assignments (J in Hz): 14 (in $C_{6}D_{6}$), δ 3.60 (dd, J = 4, 4, 2-H), 3.73 (m, 4-H), 3.85 (dd, J = 8, 11), 4.13 (dd, J = 5, 11)(6-H $_{2}$), 4.44 (dd, J = 7, 9, 5-H), 4.54 (dd, J = 4, 10, 3-H), 4.77 (ddd, J = 4, 9, 10, 1-H),

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5.50 (d, J = 10, >NH), 7.44 (s, -OCHO); 15 (in CDC1₃), δ 3.32 (m, 4-H), 4.28 (dd, J = 8, 12), 4.63 (dd, J = 5, $12)(6-H_2)$, 4.48 (dd, J=4, 5, 2-H), 5.00 (dd, J=5, 10, 5-H), 5.11 (ddd, J=5, 5, 8, 1-H), 5.31 (dd, J=4, 11, 3-H), 6.13 (d, J=8, >NH), 8.12 (s, -OCHO). The NOE's appeared for example between the following pairs of protons $^{13)}$: $_{14}^{14}$, $_{1\alpha-H}^{1\alpha-H}$ & $_{4\alpha-H}^{14}$ (4%), $_{1\alpha-H}^{1\alpha-H}$ & $_{5\alpha-H}^{14}$ (11%), $_{5\alpha-H}^{14}$ & $_{4\alpha-H}^{14}$ (9%), $_{5\alpha-H}^{14}$ & $_{1\alpha-H}^{14}$ (11%); $_{15}^{15}$, 1α -H & 4α -H (4%), 3β -H & 5β -H (3%), 4α -H & 1α -H (4%), 5β -H & 3β -H (3%). Denitrohydrogenation of 14 and 15 with n-Bu₃SnH in benzene in the presence of α , α' -azobis-iso-butyronitrile (AIBN) gave $\frac{16}{100}$ (63% from $\frac{14}{100}$, 62% from 15), colorless oil, $[\alpha]_D^{24}$ +16° (CHCl₃), $C_{23}H_{25}NO_6$, $IR \vee \frac{CHCl_3}{max}$ cm⁻¹: 1717, 1684, $\frac{1}{1}H$ NMR (500 MHz, CDCl₂): δ 1.92 (ddd, J = 7, 8, 15), 2.03 (ddd, J = 6, 6, 15)(5-H₂), 2.79 (m, 4-H), 4.10 (dd, J = 4, 5, 2-H), 4.32, 4.35 (both dd, J = 6, 11, $6-H_2$), 4.54 (m, 1-H), 5.13 (dd, J = 4, 7, 3-H), 6.05 (d, J = 6, >NH), 8.11 (s, -OCHO). On the other hand, treatment of 13 with 28% aq. NH₄OH (or liq. NH₃) and subsequent acetylation, furnished 17 [62% (or 60%)], colorless oil, $[\alpha]_D^{24}$ -8° (CHCl₃), $C_{23}H_{24}N_2O_8$, IR $v_{max}^{CHCl_3}$ 3 cm⁻¹: 1706, 1667, 1555, 1379, v_{max}^{1} 4 NMR (500 MHz, CDCl₃): δ 3.21 (m, 4-H), 4.33 (dd, J = 4, 6, 2-H), 4.46, 4.51 (both dd, J = 5, 10, 6- H_2), 4.88 (dd, J = 7, 7, 5-H), 5.00 (ddd, J = 6, 7, 7, 1-H), 5.69 (dd, J = 4, 6, 3-H), 6.27 (d, J = 7, >NH), 8.16 (s, -OCHO), NOE (e.g.): 1β -H & 2β -H (9%), 1β -H & 4β -H (2%), 2β -H & 1β -H (13%), 2β -H & 4β -H (11%), 4β -H & 2B-H (2%), 5α -H & 1α -NH (6%). Therefore, it was inferred that the Michael-type addition reactions for nitro-cyclopentenes (12, 13) proceeded to provide thermodynamically favored addition products in which the introduced $1-NH_2$ group had the same orientation as that of the 4-benzoyloxymethyl group.

Next, the Michael-type addition of N^6 -benzoyladenine for 12 was examined. Treatment of 12 with N^6 benzoyladenine in THF in the presence of KF and 18-crown-6 provided 18 (78%), colorless oil, $\left[\alpha\right]_{D}^{25}$ -102 ° (CHCl₃), $C_{33}H_{28}N_{6}O_{8}$, UV λ_{max}^{MeOH} nm (ϵ): 230 (18500), 281 (12100), IR $\nu_{max}^{CHCl_3}$ cm⁻¹: 1724, 1606, 1583, 1551, 1371, ¹H NMR (500 MHz, CDCl₃): δ 3.23 (m, 4-H), 4.76, 4.81 (both dd, J = 4, 12, 6-H₂), 4.87 (dd, J = 5, 10, 2-H), 5.37 (dd, J=10, 10, 1-H), 5.63 (d, J=5, 3-H), 5.94 (dd, J=6, 10, 5-H), 7.97 (s, 8'-H), 8.16 (s, -OCHO), 8.23 (s, 2'-H), NOE (e.g.): $\underline{1\alpha}$ & $\underline{4\alpha}$ -H (3%), $\underline{2\beta}$ -H & 3β -H (5%), $\underline{2\beta}$ -H & 5β -H (6%), $\underline{3\beta}$ -H & 5β -H (2%), 5β-H & 2β-H (4%). Deformylation of 18 with aq. NH_ΔOH - EtOH followed by ethoxyethylation and denitrohydrogenation, yielded the denitro-product from which the ethoxyethyl group was removed with 10% aq. AcOH to furnish 19 (48%), colorless oil, $[\alpha]_D^{23}$ -75 ° (CHCl₃), $C_{32}H_{29}N_5O_5$, UV λ_{max}^{MeOH} nm (ϵ): 281 (11600), IR $\nu_{max}^{CHCl_3}$ cm⁻¹: 1706, 1690, 1580, 1 H NMR (500 MHz, CDCl₃): $\delta 2.40$ (ddd, J = 4, 9, 13), 2.50 (ddd, J = 9, 9, 13)(5-H₂), 2.61 (m, 4-H), 4.28 (dd, J = 3, 5, 3-H), 4.48 (dd, J = 5, 11), 4.52 (dd, J = 3, 11)(6-H₂), 4.68 (dd, J = 5, 9, 2-H), 4.86 (ddd, J=9, 9, 9, 1-H), 7.85 (s, 8'-H), 8.49 (s, 2'-H). Finally, removal of the benzoyl and benzyl groups furnished (-)-aristeromycin (1, 93%), mp 213-215 °C, $[\alpha]_D^{24}$ -51 ° (DMF), which was identified by comparing its physical data (1 H and 13 C NMR, etc) with those reported. 1 , 6 a, 6 b)

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- 12) The molecular composition of the compound given with the chemical formula was determined either by elemental analysis or by high resolution mass spectrometry.
- 13) The magnitude of NOE (%) given in the parenthesis was observed when the underlined proton was irradiated.

(Received July 7, 1989)