## New Withanolides, Daturataturins A and B from Datura tatura L.1)

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Two new  $C_{28}$  steroidal glycosides called daturataturin A (1a) and daturataturin B (2a) were isolated from the methanolic extract of the fresh aerial parts of *Datura tatura* L. (Solanaceae), and their chemical structures were characterized as (22R)- $7\alpha$ ,27-dihydroxy-1-oxowitha-2,5,24-trienolide 27-O- $\beta$ -D-glucopyranoside and (22R)- $1\alpha$ ,3 $\beta$ ,7 $\alpha$ ,-27-tetrahydroxywitha-5,24-dienolide 3-O- $\beta$ -D-glucopyranoside by spectral analysis.

Keywords Datura tatura; Solanaceae; withanolide; daturataturin A; daturataturin B

As a part of our studies on the constituents of solanaceous plants, we previously reported the chemical structures of  $C_{28}$  steroidal lactones called daturametelins A, B, C, D, E, F and G-Ac from the methanolic extract of the fresh aerial parts of *Datura metel*  $L.^{2)}$ 

In continuing studies on the constituents of the same genus, we report here the isolation and structure elucidation of two new withanolide glucosides, daturataturins A (1a) and B (2a), from the fresh aerial parts of D. tatura L.

Daturataturin A (1a) C<sub>34</sub>H<sub>48</sub>O<sub>10</sub>, an amorphous powder, showed a  $[M + Na]^+$  peak at m/z 639, a  $[M - H_2O + H]^+$ peak at m/z 599 and a  $[M-H_2O-glc]^+$  peak at m/z 437 in the positive fast atom bombardment mass spectrum (FAB-MS). The infrared (IR) spectrum of 1a indicated the presence of an  $\alpha,\beta$ -unsaturated  $\delta$ -lactone (1698 cm<sup>-1</sup>) and an  $\alpha,\beta$ -unsaturated ketone (1663 cm<sup>-1</sup>). Acetylation of 1a with acetic anhydride and pyridine gave a pentaacetate (1b), which showed a  $[M-AcOH+H]^+$  peak at m/z 767 in the positive FAB-MS. By comparison of the proton nuclear magnetic resonance (<sup>1</sup>H-NMR) spectrum (Table I) of 1b with that of daturametelin A tetraacetate (3b), the following signals in 1b were assigned: four methyl groups [ $\delta$  0.73 (s,  $H_3$ -18), 1.04 (d, J = 6.6 Hz,  $H_3$ -21), 1.25 (s,  $H_3$ -19) and 2.06 (s,  $H_3$ -28)], five acetyl groups [ $\delta$  2.00, 2.01, 2.02, 2.03 and 2.09 (all s)], two methylene groups [ $\delta$  2.90 (dd, J=20.9 and 4.8 Hz, H-4), 3.35 (d, J = 20.9 Hz, H'-4), 4.47 (d, J = 11.0 Hz, H-27) and 4.59 (d,  $J=11.0 \,\text{Hz}$ , H'-27)], three olefinic protons [ $\delta$  5.92 (dd, J=9.9 and 2.2 Hz, H-2), 6.81 (ddd, J = 9.9, 5.0 and 2.6 Hz, H-3) and 5.83 (dd, J = 5.9 and 1.8 Hz, H-6)], and a characteristic H-22 proton [ $\delta$  4.41 (dt, J=13.2and 3.3 Hz)]. Furthermore, one methine proton signal ( $\delta$ 4.90, m) was found to be attributable to H-7 by its correlation with the olefinic proton at H-6 in the <sup>1</sup>H-<sup>1</sup>H correlated spectroscopy (COSY) spectrum. The chemical shift and the coupling pattern due to the H-22 signal indicated the presence of the typical C-17 side chain without a hydroxy group at C-20.3) As regards the configuration at C-22, the circular dichroism (CD) spectrum of 1a showed a positive Cotton effect at 255 nm, suggesting R configuration. Moreover, the <sup>1</sup>H-NMR spectrum of 1b revealed the existence of a 2,3,4,6-tetra-O-acetyl-β-Dglucopyranosyl moiety [ $\delta$  4.65 (d, J=8.1 Hz, H-1'), 4.94 (dd, J=9.6, 8.1 Hz, H-2'), 5.21 (t, J=9.6 Hz, H-3'), 5.08 (t, J=9.6 Hz, H-4', 3.69 (ddd, J=9.6, 4.4, 2.6 Hz, H-5'), 4.17 (dd, J=12.3, 2.6 Hz, H-6') and 4.23 (dd, J=12.3, 4.4 Hz,H'-6')]. From the evidence of the chemical shifts and coupling patterns of H<sub>3</sub>-28 and H<sub>2</sub>-27, the location of the glycosidic linkage was determined to be the 27-hydroxy group such as **3b**. The carbon-13 nuclear magnetic resonance ( $^{13}$ C-NMR) data for **1a** (Table II) were consistent with the assumed structure. Therefore, the structure of **1a** was concluded to be ( $^{22}R$ )- $^{7}\alpha$ ,27-dihydroxy-1-oxowitha-2,5,24-trienolide 27- $^{0}$ -D-glucopyranoside. The aglycone moiety of **1a** corresponds to compound IX, which was isolated from *Withania somnifera* Dun. by Kirson *et al.*<sup>4)</sup>

Daturataturin B (2a)  $C_{34}H_{52}O_{11}$ , an amorphous powder, showed a strong absorption band at  $3412\,\mathrm{cm}^{-1}$  (hydroxyl) and a characteristic band at  $1694\,\mathrm{cm}^{-1}$  ( $\alpha,\beta$ -unsaturated

Table I. <sup>1</sup>H-NMR Data (CDCl<sub>3</sub>) for Daturataturin A Pentaacetate (1b) and Daturataturin B Heptaacetate (2b)

Proton	1b	2b				
1		5.07 (br s)				
2	5.92 (dd)					
	$J=9.9, 2.2 \mathrm{Hz}$					
3	6.81 (ddd)	3.88 (m)				
	$J = 9.9, 5.0, 2.6 \mathrm{Hz}$					
4	2.90 (dd)					
	J = 20.9, 4.8  Hz					
	3.35 (d)					
	J = 20.9  Hz					
6	5.83 (dd)	5.77 (d)				
	$J = 5.9, 1.8 \mathrm{Hz}$	$J=4.0\mathrm{Hz}$				
7	$4.90 \ (m)^{a}$	$4.92 \text{ (m)}^{b)}$				
18	0.73 (s)	0.70 (s)				
19	1.25 (s)	1.26 (s)				
21	1.04 (d)	1.03 (d)				
	$J=6.6\mathrm{Hz}$	$J = 6.6 \mathrm{Hz}$				
22	4.41 (dt)	4.41 (br d)				
	J = 13.2, 3.3  Hz	$J=9.9\mathrm{Hz}$				
27	4.47 (d)	4.91 (d)				
	J = 11.0  Hz	$J = 11.5 \mathrm{Hz}$				
	4.59 (d)	4.86 (d)				
	J = 11.0  Hz	$J=11.5\mathrm{Hz}$				
28	2.06 (s)	$2.05 (s)^{c}$				
Glc-1'	4.65 (d)	4.58 (d)				
C1 &	J=8.1  Hz	$J=8.0\mathrm{Hz}$				
Glc-2'	4.94 (dd) <sup>a)</sup>	$4.94  (dd)^{b}$				
C1 2/	J=9.6, 8.1  Hz	J = 9.6, 8.0 Hz				
Glc-3'	5.21 (t)	5.18 (t)				
C1 4	J = 9.6 Hz	$J=9.6\mathrm{Hz}$				
Glc-4'	5.08 (t)	$5.05 (t)^{b}$				
C1 = 51	$J = 9.6 \mathrm{Hz}$	J = 9.6 Hz				
Glc-5'	3.69 (ddd)	3.68 (ddd)				
Cla C	J=9.6, 4.4, 2.6  Hz	J=9.6, 5.0, 2.5  Hz				
Glc-6'	4.17  (dd) J = 12.3, 2.6  Hz	4.10 (dd)				
	J = 12.3, 2.0  Hz 4.24 (dd)	J = 12.3, 2.5  Hz 4.23 (dd)				
	J = 12.3, 4.4  Hz	$J = 12.3, 5.0 \mathrm{Hz}$				
-OAc	J = 12.3, 4.4  Hz 2.00, 2.01, 2.02, 2.03,	J = 12.3, 3.0  Hz $2.00,^{c)}$ $2.02, 2.04,$				
-OAC	2.00, 2.01, 2.02, 2.03, 2.09	2.05, 2.08				
	ter (V /	2.05, 2.00				

a-c) Signals overlap each other in each vertical column.

TABLE II. <sup>13</sup>C-NMR Data for Daturataturin A (1a),<sup>a)</sup> Daturataturin B (2a),<sup>a)</sup> Daturametelin A (3a),<sup>a)</sup> Compound 1A (4),<sup>b)</sup> Pubescenin Ac (5)<sup>b)</sup> and Viscosalactone (6)<sup>b)</sup>

	1a	2a	3a	4	5	6
C-1	203.3	72.3	203.9	72.9	74.9	206
C-2	127.9	37.6	127.7	38.3	33.7	54.
C-3	145.6	73.8	145.8	66.3	74.9	55.
C-4	33.6	38.2	33.4	41.4	38.1	74
C-5	139.4	142.7	136.2	137.6	136.7	64
C-6	128.1	127.8	124.6	125.2	124.9	59
C-7	63.7	64.7	30.8	31.7	28.8	29
C-8	35.8	34.0	33.1	31.3	31.7	31
C-9	39.1	39.2	43.3	41.5	42.3	42
C-10	50.2	42.9	50.6	41.7	40.5	48
C-11	23.9	20.4	23.8	20.2	20.7	20
C-12	39.9	39.4	39.8	39.8	39.6	27
C-13	42.5	42.7	42.5	43.0	42.8	42
C-14	52.2	52.2	56.2	56.8	56.5	56
C-15	24.3	24.6	24.2	23.9	24.3	24
C-16	27.2	27.4	26.9	22.0	27.3	40
C-17	51.3	49.7	51.9	54.7	52.1	51
C-18	11.9	11.7	11.7	13.6	11.7	11
C-19	18.2	18.3	18.8	19.4	19.4	14
C-20	38.7	38.9	38.9	75.2	38.6	38
C-21	13.4	13.5	13.3	20.8	13.7	13
C-22	78.3	78.2	78.2	81.0	76.3	78
C-23	29.8	29.8	29.9	31.5	31.7	29
C-24	157.1	153.9	157.0	149.1	59.3	153
C-25	In solvent	127.3	122.8	122.0	62.6	125
C-26	166.1	166.4	166.0	166.2	170.6	167
C-27	63.4	56.2	63.2	12.5	13.0	57
C-28	20.6	20.1	20.4	20.5	18.0	20
Glc-1'	104.8	102.9	104.6		99.6	
Glc-2'	75.2	75.2	75.0		71.4	
Glc-3'	78.5	78.5	78.3		72.9	
Glc-4'	71.6	71.5	71.5		68.5	
Glc-5'	78.4	78.4	78.3		71.8	
Glc-6'	62.8	62.7	62.6		62.1	
–OAc					170.3	
					170.0	
					$169.4 (\times 2)$	
					$20.7(\times 4)$	
					21.1	

a) In pyridine- $d_5$ , b) in CDCl<sub>3</sub>.

 $\delta$ -lactone) in its IR spectrum. The positive FAB-MS of 2a showed a  $[M+Na]^+$  peak at m/z 659. Next, the <sup>1</sup>H- and <sup>13</sup>C-NMR spectral data (Tables I and II, respectively) of 2a and its heptaacetate (2b) were compared with those of compound 1A (4)<sup>5)</sup> and pubescenin-Ac (5)<sup>6)</sup> (for the A, B-ring moiety), and viscosalactone (6) $^{7}$  (for the side chain). The <sup>1</sup>H-NMR spectrum (Table I) of **2b** disclosed the occurrence of four methyl groups [ $\delta$  0.70 (H<sub>3</sub>-18), 1.03  $(H_3-21)$ , 1.26  $(H_3-19)$  and 2.05  $(H_3-28)$ ], one oxymethyl group [ $\delta$  4.91 and 4.86 (H<sub>2</sub>-27)] and two oxygenated methine protons  $[\delta 3.88 \text{ (H-3)}]$  and 4.41 (H-22)] together with the presence of the  $\beta$ -D-glucopyranosyl moiety. The chemical shifts of this oxymethyl signal in 2b indicated that the glucosyl moiety did not link to the C-27 hydroxymethyl group, in contrast to 1b. Furthermore, in the <sup>1</sup>H-NMR spectrum of 2b, the olefinic protons at C-2 and C-3 in ring A of 1b disappeared, and instead, two new oxygenated methine proton signals occurred at  $\delta$  3.88 (1H, m) and 5.07 (1H, brs). Thus, the steroidal 2-ene-1-one system in 1a no longer existed in 2a. These new methine protons were assigned to the 5-ene- $1\alpha$ ,  $3\beta$ -diol system, which is known as

the precursor of the 2-ene-1-one system in the biogenesis of the withanolides.<sup>3)</sup> The <sup>13</sup>C-NMR data of the side chain moiety (Table II) of the side chain part for **2a** were in good accordance with those of **6** except for C-25. Further, by comparison of the <sup>13</sup>C-NMR data of **2a** with those of **4** and **5**, the presence of a 5-ene- $1\alpha$ ,  $3\beta$ -diol structure (vide supra) and the  $\beta$ -D-glucopyranosyl group linked to the  $3\beta$ -hydroxy group in **2a** (and also **2b**) were reasonably clarified. Based on these spectral data and from the positive Cotton effect at 252 nm, the structure of **2a** having the (22R)-configuration was shown in the formula.

## Experimental

Optical rotations were measured on a JASCO DIP-360 automatic digital polarimeter and CD spectra on a JASCO J-50A spectropolarimeter. The IR spectra were recorded with a Hitachi IR spectrometer, model 270-30. The  $^1\mathrm{H}\text{-}$  and  $^{13}\mathrm{C}\text{-}\mathrm{NMR}$  spectra were measured with a JEOL JNM-GX 400 NMR spectrometer and chemical shifts are given on a  $\delta$  (ppm) scale with tetramethylsilane as an internal standard. The FAB-MS were taken in a glycerol matrix containing NaI. Thin layer chromatography was performed on precoated Kieselgel 60  $\mathrm{F}_{254}$  (Merck) and detection was achieved by spraying 10%  $\mathrm{H}_2\mathrm{SO}_4$  followed by heating. Column chromatography was carried out on Kieselgel (70—230 mesh and 230—400 mesh, Merck) and Sephadex LH-20 (Pharmacia Co.).

Extraction and Separation Frash aerial parts (1.78 kg) of Datura tatura L. (Solanaceae) harvested at the botanical garden in Kumamoto University in July 1988 were extracted with MeOH and the extract (40.3 g) was partitioned between n-BuOH and  $H_2O$ . The n-BuOH layer (16.3 g) was subjected to column chromatography repeatedly over silica gel using CHCl<sub>3</sub>-MeOH- $H_2O$ =1:0:1→8:2:0.1→0:1:0 and over Sephadex LH-20 using MeOH to give daturataturins A (1a, 1.70 g) and B (2a, 20.7 mg).

**Daturataturin A (1a)** An amorphous powder,  $[α]_{c}^{121} - 38.1^{\circ}$  (c = 0.72, pyridine). IR  $v_{max}^{KBr}$  cm<sup>-1</sup>: 3442, 1698, 1663, 1464, 1404, 1344, 1323. Positive FAB-MS m/z: 639 [M+Na]<sup>+</sup>, 599 [M-H<sub>2</sub>O+H]<sup>+</sup>, 437 [M-glc-H<sub>2</sub>O+H]<sup>+</sup>, 419 [M-glc-2H<sub>2</sub>O+H]<sup>+</sup>, 401 [M-glc-3H<sub>2</sub>O+H]<sup>+</sup>, 265, 171. CD (c = 0.94, MeOH) [ $\theta$ ] (nm): -25000 (325) (negative max.), +47000 (255) (positive max.).

Acetylation of Daturataturin A (1a) A solution of 1a (109 mg) in pyridine–acetic anhydride (1:1, v/v, 4 ml) was kept at room temperature overnight and the product was chromatographed on silica gel (n-hexane–EtOAc=1:1, v/v) to afford a pentaacetate (1b, 53.4 mg). An amorphous powder,  $[\alpha]_{3}^{33}$  -42.0° (c=0.59, CHCl<sub>3</sub>). Positive FAB-MS m/z: 767 [M-AcOH+H]<sup>+</sup>, 331.

**Daturataturin B (2a)** An amorphous powder,  $[\alpha]_{\rm b}^{15}$  -17.7° (c=0.52, pyridine). IR  $\nu_{\rm max}^{\rm KBr}$  cm<sup>-1</sup>: 3412, 1694, 1466, 1400, 1386, 1320, 1306. Positive FAB-MS m/z: 659 [M+Na]<sup>+</sup>, 619 [M-H<sub>2</sub>O+H]<sup>+</sup>, 439 [M-glc-2H<sub>2</sub>O+H]<sup>+</sup>, 289, 232, 176. CD (c=0.60, MeOH) [θ] (nm): -4770 (315) (negative max.), +48700 (252) (positive max.).

Acetylation of Daturataturin B (2a) Acetylation was performed with pyridine and acetic anhydride (2:1, v/v, 3 ml) overnight at room temperature. The product was purified on silica gel column chromatography (benzene-acetone=20:1, v/v), and provided a heptaacetate (2b, 11.1 mg). An amorphous powder,  $[\alpha]_D^{2.5} - 31.9^{\circ} (c=0.53, \text{CHCl}_3)$ . Positive FAB-MS m/z: 953  $[M+Na]^+$ , 871  $[M-AcOH+H]^+$ , 811  $[M-2AcOH+H]^+$ , 601  $[M-glc(Ac)+H]^+$ , 550, 522, 331, 289, 246.

## References and Notes

- This work is part XX in the series of studies on the constituents of solanaceous plants. Part XIX: K. Shingu, Y. Furusawa, N. Marubayashi, I. Ueda, S. Yahara and T. Nohara, Chem. Pharm. Bull., 38, 2866 (1990).
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