## SYNTHESIS OF BIOLOGICALLY ACTIVE DERIVATIVES OF D-GLUCOSAMINE-4-PHOSPHATE AND 1-THYMINYL-D-GLUCOSAMINE-4,6-DISULFATE

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New 1-thyminy1-D-glucosamine-4-phosphate and 1-thyminy1-D-glucosamine-4,6-disulfate derivatives were synthesized. The 1-thyminy1-D-glucosamine-4,6-disulfate derivative showed antiviral activity against HIV.

**KEYWORDS** 1-thyminyl-D-glucosamine-4-phosphate; 1-thyminyl-D-glucosamine-4,6-disulfate; lipid A; antitumor activity; antiviral activity; HIV

Lipid A is of considerable biological and pharmacological interest, because it is responsible for the expression of many biological activities of the lipopolysaccharide (LPS) of Gram-negative bacteria, e.g., endotoxicity, adjuvanticity, antitumor activity and so on. Previously, we reported the synthesis of various derivatives of acyloxyacylglucosamine-4-phosphate as the nonreducing sugar moiety of lipid A. These compounds exhibited mitogenic activity, antitumor activity, and lethal toxicity. From our study concerning chemical modification of D-glucosamine-4-phosphate to yield more effective antitumor and immunopotent substances, we report here the synthesis of the 1-thyminyl-D-glucosamine-4-phosphate (1) and 1-thyminyl-D-glucosamine-4,6-disulfate derivatives (3), (4). As our synthesis strategy, to prepare nucleosides containing glucosamine-4-phosphate, we designed a suitably functionalized key intermediate (2) carrying one amino and one hydroxyl group at the C-2 and C-3 positions of glucosamine skeleton, respectively.

First, we describe a synthesis sequence for the 1-thyminyl-D-glucosamine-4-phosphate (1). The key intermediate (2) was prepared starting from 1,3,4,6-tetra-0-acetyl-2-chloroacetamido-2-deoxy-D-glucopyranose (5) in 4 steps. Condensation of 5 with the disilylated thymine (6) in the presence of  $SnCl_4$  and Molecular Sieves 4A (MS4A) in  $ClCH_2CH_2Cl$  afforded the N-glycoside (7) (64% mp 132-135°C,  $[\alpha]_D$  -8.00°), as a single anomer, the glycosidic linkage of which was assigned the  $\alpha$ -configuration from the anomeric proton (J=10Hz) at 6.04 ppm in the  $^1$ H NMR spectrum of 7. N-Glycoside (7) was 0-deacetylated by NaOMe in MeOH to give the triol (8) (76%, mp 143-147°C,  $[\alpha]_D$  +9.50°), and treated with 2,2-dimethoxypropane in DMF in the presence of a catalytic amount of p-toluenesulfonic acid to give 9 (78%, mp 219-221°C,  $[\alpha]_D$  -16.6°). The chloroacetyl group of 9 was removed with thiourea and diisopropylethylamine in THF to give the key intermediate (2) (85%, mp 98-102°C,  $[\alpha]_D$  +19.6°).

Reagents: a) 6 (1.2 eq), SnCl<sub>4</sub>(1.5 eq), MS4A, ClCH<sub>2</sub>CH<sub>2</sub>Cl, 0°C, 1h then rt, 15 h; b) 0.1 M NaOMe, MeOH, 0°C, 1h then rt, 15h; c) 2,2-dimethoxypropane (5 eq), p-toluenesulfonic acid (0.2 eq), DMF, rt, 20 h; d) (NH<sub>2</sub>)<sub>2</sub>C=S (2 eq), (i-Pr)<sub>2</sub>NEt (2 eq), MS4A, THF, reflux, 3h. Chart 1

The key intermediate (2) thus obtained was used to synthesize of 1 as follows. We selected the tetradecanoyloxytetradecanoyl group at N-2 and the tetradecanoyl group at 0-3 of the glucosamine skeleton of the GLA-27 type as the monosaccharide analogue of lipid  $A.^{5}$ ) For further conversion of 2 into 1, several attempts to condense the free amino group of 2 with the fatty acid residue in the presence of either dicyclohexylcarbodiimide (DCC) and 1-hydroxybenzotriazole (HOBt) or DCC and N-hydroxysuccinimide (NSC) or the chloride as the activator of the carbonyl group led only to the recovery of 2. The effective result was obtained as follows. The free hydroxyl group of  $\mathbf{9}$  was acylated with tetradecanoyl chloride and pyridine in the presence of dimethylaminopyridine (DMAP) in  $\text{CH}_2\text{Cl}_2$ -DMF (10:1) to give 10 (82%, mp 164-167°C,  $[\alpha]_D$  -28.3°), then the isopropylidene group of 10 was removed by hydrolysis with aqueous 90% AcOH to give the diol (11) (78%, mp 157-160°C,  $[\alpha]_{\overline{D}}$  -19.6°). The primary hydroxyl group of 11 was selectively protected with tert-butyldimethylsilyl chloride (TBDMS-C1) and triethylamine in  $\text{CH}_2\text{Cl}_2$  to give the 6-0-TBDMS ether (12) (83%, mp 111-115°C,  $[\alpha]_D$  -52.2°). The chloroacetyl group of 12 was removed with thiourea in THF to give 13 (78%, syrup,  $[\alpha]_D$  +1.37°). Subsequently, the free amino group of 13 was acylated with optically pure (R)-3-tetradecanoyloxytetradecanoic acid, DCC, and DMAP in dry  $\mathrm{CH}_2\mathrm{Cl}_2$  to afford the diacylate (14) (67%, syrup,  $[\alpha]_D$  -13.1°). The free hydroxyl group of 14 was efficiently phosphorylated with bis(2,2,2-trichloroethyl)diisopropylaminophoshine<sup>6)</sup> and lH-tetrazole in  $CH_2Cl_2$ , then oxidized with  $t-BuO_2H^{7}$  in  $CH_2Cl_2$  to give the 4-O-phosphate (15) (78%, syrup,  $[\alpha]_D$  -5.88°). <sup>1</sup>H NMR spectrum of 15 showed two doublets (J=6.5Hz) (two protons each) at 4.60 and 4.58 ppm which could be assigned as the methylene protons of the trichloroethyl groups. Finaly, deprotection of the silyl group and the bis(2,2,2-trichloroethyl) group of 15 were simultaneously cleaved with Zn dust in aqueous 80% AcOH to yield the desired product (1) (85%, syrup,  $[\alpha]_n$  -4.23°).

B: thyminyl;  $C_{14}$ -:  $CH_3(CH_2)_{12}C(O)$ -;  $C_{14}OC_{14}$ -:  $CH_3(CH_2)_{10}CHCH_2C(O)$ -  $CH_3(CH_2)_{12}C(O)O$ 

**Reagents:** a)  $C_{14}OC1$  (1.2 eq), pyridine (2 eq), DMAP (0.5eq),  $CH_2Cl_2$ -DMF (10 : 1), 0°C, 1h then rt, 15 h; b) AcOH-H<sub>2</sub>O (9 : 1), 90-95°C, 1h; c) t-Butyl(Me)<sub>2</sub>SiCl (3 eq), NEt<sub>3</sub> (5 eq), DMAP (0.5eq),  $CH_2Cl_2$ , 0°C, 1h then rt, 15 h; d)  $(NH_2)_2C=S$  (5 eq), MS4A, THF, 50-60°C, 2.5 h; e)  $C_{14}OC_{14}OH$  (1.5 eq), DCC (1.8 eq),  $CH_2Cl_2$ , 0°C, 1 h then rt, 20 h; f) 1)  $(TCEO)_2PN(i-Pr)_2$  (4 eq), 1H-Tetrazole (6 eq),  $CH_2Cl_2$ , rt, 1 day; 2) t-BuO<sub>2</sub>H (1.5 eq),  $CH_2Cl_2$ , - 40°C, 1h then 0°C, 20 h; g) Zn dust, AcOH-H<sub>2</sub>O (8 : 2), 50-60°C, 5h.

Chart 2

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Next, for the synthesis of 1-thyminyl-D-glucosamine-4,6-disulfate derivatives, the amino and hydroxyl groups of 2 were simultaneously acylated with either tetradecanoic acid or (R)-3-tetradecanoyloxy-tetradecanoic acid in the presence of DCC and DMAP in  $\mathrm{CH_2Cl_2}$ -DMF (10:1) to give the corresponding diacylate (16), (17) in 36% and 8% yield, respectively. The isopropylidene group of 16, 17 was subsequently deprotected with 90% AcOH to give the diol (18), (19) in 54% and 39% yield, respectively. The 4,6-hydroxyl groups of 18, 19 were sulfated with  $\mathrm{Me_3NSO_3}$  complex in DMF, then treated with  $\mathrm{CF_3CO_2H}$  in  $\mathrm{CH_2Cl_2}$  to afford the disulfate (3), (4) in 36% and 67% yield, respectively, after gelfiltration with Sephadex LH-20 (CHCl $_3$ -MeOH=2:1).

Reagents: a)  $C_{14}OH$  or  $C_{14}OC_{14}OH$  (1.5 eq), DCC (1.5 eq), DMAP (1.5 eq),  $CH_2Cl_2$ -DMF (5 : 2), 0°C, 1h then rt, 15 h; b) AcOH-H<sub>2</sub>O (9:1), 90-95°C, 1h; c) 1)  $Me_3NSO_3$  (3.5 eq), DMF, 50°C, 20 h; 2)  $CF_3CO_2H$  (1.5 eq),  $CH_2Cl_2$ , rt, 1h.

## Chart 3

Preliminary examination of the biological activity of three synthetic compounds revealed that  ${\bf 1}$  showed weak antitumor activity and  ${\bf 4}$  exhibited significant antiviral activity against HIV (Humman Immunodeficiency Virus). In addition,  ${\bf 4}$  had little or no cytotoxicity.

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