Synthesis of 6-C-Substituted 9-Tetrahydrofuranylpurine Derivatives

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6-Dicyanomethylene-9-tetrahydrofuranylpurine (4), which was obtained by the reaction of 9H-1,6-dihydropurine- $4^{6,\alpha}$ -propanedinitrile (3) with 2,3-dihydrofuran, has been catalytically hydrogenated to the α -(aminomethylene)-9-(tetrahydrofuran-2-yl)-9H-purine-6-acetonitrile (5) in good yield using N,N-dimethylformamide-benzene as a solvent over Pd-C under medium pressure. Substitution of 5 with amines gave the corresponding alkylaminomethylene purines (6 and 7). Reaction of 5 with hydrazine gave the pyrazole derivative (8).

Keywords 9-(tetrahydrofuran-2-yl)-9H-1,6-dihydropurine- $\Delta^{6,\alpha}$ -propanedinitrile; α-(aminomethylene)-9-(tetrahydrofuran-2-yl)-9H-purine-6-acetonitrile; N-substituted α-(aminomethylene)-9-(tetrahydrofuran-2-yl)-9H-purine-6-acetonitrile; N-substituted α-(aminomethylene)-9-(tetrahydrofuran-2-yl)-9N-purine-6-acetonitrile; N-substituted α-(aminomethylene)-9-(aminome

6-N-Substituted purine derivatives and 9-alkylpurine derivatives exhibit a number of interesting biological²⁾ and antiviral activities.³⁾ Therefore, in recent years much attention has been focused on 6-C-substituted purine derivatives.

As part of our studies on the synthesis of 6-C-substituted purine derivatives, we have recently reported the reduction of 9H-1,6-dihydropurine- $\Delta^{6,\alpha}$ -propanedinitrile (3) and 9-alkyl 1,6-dihydropurine- $6-\Delta^{6,\alpha}$ -propanedinitrile to give 6-enaminonitrile purine derivatives.⁴⁾ Also, it was found that 9- β -D-ribofuranosyl-9H-1,6-dihydropurine- $\Delta^{6,\alpha}$ -propanedinitrile (1) and α -(aminomethylene)-9- β -D-ribofuranosyl-9H-purine-6-acetonitrile (2) exhibited antitumor activity.⁵⁾ In this paper we describe the synthesis of α -(aminomethylene)-9-(tetrahydrofuran-2-yl)-9H-purine-6-acetonitrile (5) via 6-dicyanomethylene-9-tetrahydrofuranylpurine (4) using catalytic hydrogenation in an N,N-dimethylformamide (DMF)-benzene solvent system, and synthesis of purine derivatives via 5.

Reaction of 3 with 2,3-dihydrofuran in the presence of boron trifluoride etherate gave 9-(tetrahydrofuran-2-yl)-9H-1,6-dihydropurine- $\Delta^{6,\alpha}$ -propanedinitrile (4) in 68% yield. The ultraviolet (UV) spectrum of 4 was identical with that of 9-(methoxymethyl)-9*H*-1,6-dihydropurine- $\Delta^{6,\alpha}$,-propanedinitrile prepared by the substitution of 6-chloro-9-(methoxymethyl)purine with malononitrile.⁴⁾ Therefore, the tetrahydrofuranyl group was concluded to be at position 9 of the purine ring. Hydrogenation of 4 over Pd-C in DMF-benzene (1:1) under medium pressure (4 atm) gave 5 in 62% yield. The proton nuclear magnetic resonance (1H-NMR) spectrum of 5 in deuteriodimethyl sulfoxide (DMSO- d_6) showed two sets of vinyl proton signals [δ 7.69 (E) and δ 9.32 (Z)] as a doublet of doublets which collapsed to a singlet with deuterium oxide. Generally, it is known that the signal of the vinyl proton of an enamine moiety

appears at δ 7—8.⁶⁾ However, the downfield shift of the vinyl proton of the Z-form of 5 is due to the anisotropic effect of the purine ring. Therefore, the geometrical isomers (E/Z) of 5 exist in an enamino-nitrile tautomeric equilibrium rather than an imino-nitrile equilibrium. The ratio of geometrical isomers (E/Z) was found to be 15:85 by comparing the area of the vinyl proton signal of each isomer in the ¹H-NMR spectrum.

Substitution of **5** with furfurylamine under heating gave α -(*N*-furfurylaminomethylene)-9-(tetrahydrofuran-2-yl)-9*H*-purine-6-acetonitrile (**6**) in 91% yield. The ¹H-NMR spectrum of **6** in DMSO- d_6 showed the presence of two sets of vinyl protons at δ 7.66 (*E*) and at δ 9.33 (*Z*). The ratio of geometrical isomers (E/Z) in DMSO- d_6 was 29:71. Substitution of **5** with piperidine gave (Z)- α -(piperidinomethylene)-9-(tetrahydrofuran-2-yl)-9*H*-purine-6-acetonitrile (**7**) in 88% yield. The ¹H-NMR spectrum of **7** showed the vinyl proton signal at δ 9.25 (1H, s). This indicates that the enamine moiety was in the *Z* form only. Reaction of **5** with hydrazine gave the pyrazole (**8**) in 70% yield.⁷⁾

Experimental

All melting points were determined on a Yamato capillary melting point apparatus, MP-21, and are uncorrected. Infrared (IR) spectra were taken on a JASCO A-102 spectrometer. UV spectra were measured using a Hitachi EPS-3T spectrometer. ¹H-NMR spectra were recorded on JEOL JNM-FX100 and JEOL GX spectrometers using tetramethylsilane as an internal standard. Mass spectra (MS) were measured with a JEOL JMS-D300 spectrometer using a direct inlet system; ionizing potential, 70 eV. The following abbreviations are used: s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet; br, broad; sh, shoulder; and dd, doublet of doublets.

9-(Tetrahydrofuran-2-yl)-9*H*-1,6-dihydropurine- $\Lambda^{6.x}$ -propanedinitrile (4) A cold (0—5 °C) stirred solution of 3 (0.500 g, 2.7 mmol) and 2,3-dihydrofuran (3.80 g, 54 mmol) in dry DMF (10 ml) was treated with BF₃O(C₂H₅)₂ (0.08 ml), and the solution was stirred for 15 h at room temperature. The solution was adjusted to pH 7 with concentrated NH₄OH, and evaporated *in vacuo*. The residue was diluted with water (10 ml), and allowed to stand overnight in an ice-bath. The precipitate was collected by filtration, and recrystallized from DMF-ether to give 4 (0.469 g, 68%) as colorless microcrystals, mp 250—253 °C (dec.). IR (KBr): 3225 (NH), 2210 (CN), 2190 (CN) cm⁻¹. UV $\lambda_{\rm max}^{\rm MeOH}$ nm (log ε): 236 (3.94), 335 (4.48). ¹H-NMR (DMSO- $d_{\rm G}$) δ : 2.06—2.39 (4H, m), 3.90—4.19 (2H, m), 6.28 (1H, t, J=4.4Hz, C₁·H), 8.24 (1H, s, C₂H), 8.46 (1H, s, C₈H). MS m/z: 254 (M⁺), 184 (B+1). *Anal*. Calcd for C₁₂H₁₀N₆: C, 56.68; H, 3.96; N, 33.06. Found: C, 56.52; H, 3.85; N, 33.32.

α-(Aminomethylene)-9-(tetrahydrofuran-2-yl)-9*H*-purine-6-acetonitrile (5) A solution of 4 (0.500 g, 1.9 mmol) and 5% (w/w) methanolic ammonia (5 ml) in DMF-benzene (1:1 (v/v), 240 ml) was hydrogenated over 5% Pd-C (250 mg) at room temperature for 40 h under 4 atm pressure. The catalyst was filtered off, and the filtrate was evaporated *in vacuo*. The residue was purified by column chromatography on neutral alumina (20 g) with 3% MeOH in CHCl₃ as the eluent and the eluate was evaporated *in*

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Chart 1

vacuo. The residue was recrystallized from MeOH–hexane to give **5** (313 mg, 62%) as slightly yellow needles, mp 180—183 °C (dec.). IR (KBr): 3253 (NH), 3175 (NH), 2200 (CN) cm $^{-1}$. UV λ_{ms}^{MoOH} nm (log ε): 229 (4.21), 329 (4.44), 340 (4.38 sh). ¹H-NMR (DMSO-d₆) δ: 2.02—2.40 (4H, m), 3.82—4.29 (2H, m), 6.31 (1H, m, C₁·H), 7.69 (0.15H, dd, J=16, 8 Hz, C=CH, E), 7.97—8.42 (1.7H, m, NH, Z, exchangeable with D₂O), 8.51 (0.85H, s, C₈H, Z), 8.57 (0.15H, s, C₈H, E), 8.59 (0.85H, s, C₂H, Z), 8.68 (0.15H, s, C₂H, E), 9.32 (0.85H, dd, J=16, 8 Hz, C=CH, Z), 10.68 (0.15H, m, NH, E, exchangeable with D₂O). MS m/z: 256 (M $^+$), 186. Anal. Calcd for C₁₂H₁₂N₆O: C, 56.24; H, 4.72; N, 32.80. Found: C, 56.23; H, 4.81; N, 32.89.

α-(N-Furfurylaminomethylene)-9-(tetrahydrofuran-2-yl)-9H-purine-6acetonitrile (6) A solution of 5 (0.100 g, 0.39 mmol) and furfurylamine (0.114 g, 1.17 mmol) in EtOH (4 ml) was refluxed for 3.5 h, then allowed to cool. The solvent was evaporated in vacuo and the residue was purified by column chromatography on silica gel (5 g) with CH₂Cl₂ as the eluent and the eluate was evaporated in vacuo. The residue was recrystallized from MeOH-hexane to give 6 (0.120 g, 91%) as colorless needles, mp 146—147 °C. IR (KBr): 3340 (NH), 2200 (CN), 1635 cm⁻¹. UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ε): 232 (4.23), 340 (4.49), 350 (4.48, sh), 376 (3.85). $^{1}\text{H-NMR}$ (DMSO- d_6) δ : 2.09—2.47 (4H, m), 3.83—4.29 (2H, m), 4.56—4.74 (2H, m, CH₂), 6.44 (2H, d, J = 1.5 Hz, C₁·H), 7.66 (1H, br s), 7.91 (0.29H, d, J = 14 Hz, C=CH, E), 8.51 (0.71H, s, C₈H, Z), 8.59 (1H, s, C₂H, Z and C_8H , E), 8.68 (0.29H, s, C_2H , E), 9.33 (0.71H, br s, C=CH, Z), 11.35 (0.29H, m, NH, E, exchangeable with D_2O). MS m/z: 336 (M⁺), 266 (B+1). Anal. Calcd for $C_{17}H_{16}N_6O_2$: C, 60.70; H, 4.80; N, 24.99. Found: C, 60.51; H, 4.67; N, 25.18.

(Z)-α-(Piperidinomethylene)-9-(tetrahydrofuran-2-yl)-9H-purine-6-acetonitrile (7) A solution of 5 (0.100 g, 0.39 mmol) and piperidine (0.116 g, 1.17 mmol) in EtOH (3 ml) was refluxed for 2 h, then allowed to cool. The solvent was evaporated *in vacuo*. The residue was purified by column chromatography on silica gel (5 g) with 2% MeOH in CHCl₃ as the eluent and the eluate was evaporated *in vacuo*. The residue was recrystallized from CHCl₃-hexane to give 7 (0.110 g, 88%) as colorless needles, mp 164-165 °C. IR (KBr): 3105 (NH), 2200 (CN), 1617 cm⁻¹. UV $\lambda_{\rm max}^{\rm MeOH}$ nm

(log ε): 237 (4.18), 260 (3.94), 270 (3.87, sh), 342 (4.56), 350 (4.55, sh).

¹H-NMR (DMSO- d_6) δ : 1.69 (6H, br s), 2.02—2.40 (4H, m), 3.40 (4H, m), 3.70—4.26 (2H, m), 6.34 (1H, t, J=5 Hz, C_1 ·H), 8.52 (1H, s, C_8 H), 8.59 (1H, s, C_2 H), 9.25 (1H, s, C=CH). MS m/z: 324 (M⁺), 254 (B+1).
Anal. Calcd for $C_{17}H_{20}N_6O$: C, 62.94; H, 6.21; N, 25.91. Found: C, 62.85; H, 6.23; N, 26.08.

6-(3-Aminopyrazol-4-yl)-9-(tetrahydrofuran-2-yl)-9*H*-**purine (8)** A solution of **5** (0.200 g, 0.78 mmol) and hydrazine hydrate (0.120 g, 2.34 mmol) in EtOH (8 ml) was refluxed for 5 h, then allowed to cool. The solution was evaporated *in vacuo*. The residue was purified by silica gel column chromatography on silica gel (10 g) with 2% MeOH in CH₂Cl₂ as eluent, and the eluate was evaporated *in vacuo*. The residue was recrystallized from AcOEt-hexane to give **8** (0.15 g, 70%) as colorless needles, mp 133—136 °C (dec.). IR (KBr): 3380 (NH), 3255 (NH), 1590 cm⁻¹. UV $\lambda_{\rm max}^{\rm MeOH}$ nm (log ε): 220 (4.03, sh), 265 (3.57), 230 (4.09). ¹H-NMR (DMSO- d_6) δ: 2.09—2.32 (4H, m), 3.83—4.28 (2H, m), 6.35 (1H, t, J=5 Hz, C₁-H), 6.77 (2H, br s, NH₂, exchangeable with D₂O), 8.38 (1H, s, pyrazole-H), 8.53 (1H, s, C₈H), 8.67 (1H, s, C₂H), 12.38 (1H, br s, NH, exchangeable with D₂O). MS m/z: 271 (M⁺). *Anal*. Calcd for C₁₂H₁₃N₇O: C, 53.13; H, 4.83; N, 36.15. Found: C, 53.55; H, 4.97; N, 36.38.

References and Notes

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