A NOVEL GLYCEROGLYCOLIPID FROM THE NITROGEN-FIXING CYANOBACTERIUM ANABAENA FLOS-AQUAE F. FLOS-AQUAE

Nobutoshi MURAKAMI,^a Hideaki SHIRAHASHI,^a Jinsaku SAKAKIBARA,*,^a and Yukiko TSUCHIDAb Faculty of Pharmaceutical Sciences, Nagoya City University,^a 3-1 Tanabe-dori, Mizuho-ku, Nagoya 467, Japan and Taisho Pharmaceutical Co.,Ltd.,^b 1-403, Yoshino-cho, Ohmiya, Saitama 330, Japan

A novel glyceroglycolipid, (2'S)-3',6-O-diacyl-glyceryl β -D-galactopyranoside (1), was isolated from the nitrogen-fixing cyanobacterium *Anabaena flos-aquae f. flos-aquae*. The chemical structure was elucidated on the basis of spectral properties, and the distribution of fatty acid residues was determined by enzymatic hydrolysis of 1 using lipase (from *Rhizopus arrhizus*).

KEYWORDS cyanobacterium; Anabaena flos-aquae f. flos-aquae; nitrogen-fixation; glycolipid

Some cyanobacteria are known to carry out nitrogen-fixation in a heterocyst possessing a thicker cell wall than that of the usual vegetative cell. Since glycolipids are shown to take part in several membrane functions in plants, the unique functions and structural features of heterocyst cell walls may be related to a specific glycolipid. This assumption prompted us to investigate glycolipids in the nitrogen-fixing cyanobacteria Anabaena flos-aquae f. flos-aquae. Here we describe the isolation and characterization of a novel glyceroglycolipid (1).

Cultures were grown for three weeks in CB-medium⁴⁾ at 25 °C illuminated continuously with cool-white fluorescent lights (1500 lux) and aerated vigorously with sterilized air through a 0.2 μ m membrane filter at the rate of 0.5-L per minute. The alga was collected by centrifugation at 20,000 × g and lyophilized to afford 8.02 g of the lyophilized cells from the 35-L culture medium. Extraction of the alga (8.02 g) with CHCl₃-MeOH (1:1) gave a residue (1.16 g), which was subjected to successive SiO₂ column chromatography (CHCl₃-MeOH=20:1 \rightarrow 10:1) and normal phase HPLC (CHCl₃:MeOH=96:4) to furnish 1 (6.0mg).⁵)

Compound 1, a white amorphous powder, $[\alpha]_D^{25}$ -2.1° (c=1.2, CHCl₃), exhibited a hydroxyl (3364 cm⁻¹) and two ester carbonyl absorption bands (1733, 1721 cm⁻¹) in its IR spectrum. The ¹H NMR spectrum of 1 indicated the presence of terminal methyl ($\delta 0.88$, t, 3H×2) and a number of methylene groups ($\delta 1.28$). It also showed a 12H signal covering the range from 4.14 to 4.90 ppm. The physicochemical features outlined above suggested that 1 was a glycolipid. Detailed analysis of the homonuclear decoupling spectra defined a sugar component as a galactose. In addition, observation of carbon signals

CH₂OR²
HO OO CH₂

$$CH_{2} = CH_{2}$$

$$CH_{2$$

© 1992 Pharmaceutical Society of Japan

Table I. ¹H and ¹³C NMR Data for 1 and 2 a)

	1b)		2
	$1_{ m H}$	13 _C	$1_{ m H}$
1	4.89 (1H, d, J=7.9)	105.5	4.91 (1H, d, J=7.9)
2	4.50 (1H, dd, J=7.9, 9.6)	72.0	4.51 (1H, dd, J=7.9, 9.8)
3	4.18 (1H, dd, J=3.4, 9.6)	74.7	4.15-4.20 (m) ^c)
4	4.41 (1H, d, J=3.4)	69.9	4.40 (1H, d, J=3.3)
5	4.17 (1H, dd, J=5.3, 7.5)	73.8	4.15-4.20 (m) ^c)
6	4.90 (1H, dd, J=7.5, 11.2)	64.5	4.88 (1H, dd, J=5.0, 10.9)
	4.80 (1H, dd, J=5.3, 11.2)		4.80 (1H, dd, J=5.0, 10.9)
1'	4.45 (1H, dd, J=5.3, 10.4)	72.0	4.51 (1H, dd, J=5.3, 9.7)
	4.14 (1H, dd, J=5.3, 10.4)		4.29 (1H, dd, J=4.5, 9.7)
2'	4.52 (1H, m)	68.8	4.45 (1H, quintet-like)
3'	4.62 (2H, m)	66.5	4.15-4.20 (m)¢)

a)The 1 H NMR spectra were measured in pyridine- d_{5} containing one drop of D₂O at 400 MHz, while the 13 C NMR spectrum was recorded in pyridine- d_{5} at 100 MHz. b)Assignments were made with the aid of the homonuclear decoupling and the C-H COSY spectra. c)These signals overlapped.

from two carbonyl (δ c173.6, C=O×2) and two terminal methyl groups (δ c14.3, CH₃×2) in the ¹³C NMR spectrum revealed the presence of two acyl residues. The glycolipid (1) is, however, different from a monogalactosyl diacylglycerol distributed widely in the plant membrane.

Intensive comparison of their ¹H NMR spectra disclosed the apparent difference that the proton signal due to C-6 methylene group in 1 was shown in a lower field and the signal assignable to 2'-H of glycerol portion in 1 appeared in a higher field than those of monogalactosyl diacylglycerol.⁶⁾ In the ¹³C NMR spectrum, the C-6 signal in 1 was observed downfield by 2.2 ppm and the neighboring C-5 signal was shifted upfield by 3.0 ppm. Each ¹³C signal of C-1', C-2', and C-3' were also shifted by +4.0, -2.2, and +3.2 ppm with respect to the glycerol moiety. These acylation shifts probably showed the two acyl residues to be present at both C-6 and C-3'.⁷⁾

The presumption was confirmed by the HMBC spectrum of 1: two carbonyl carbon signals showed the cross peak due to long-range coupling with the proton signals of C-6 and C-3' methylene groups. On treatment of 1 with NaOMe-MeOH, (2'R)-glyceryl β -D-galactopyranoside, $[\alpha]_D^{24}$ -8.4° (c=0.5, H₂O),⁸⁾ was quantitatively prepared along with a mixture of methyl esters of fatty acids. On the basis of the above findings, the structure of 1 was established as (2'S)-6,3 -O-diacyl-glyceryl β -D-galactopyranoside.

Since the glycolipid (1) contained several fatty acid residues,⁹⁾ the distributions of fatty acids were determined by enzymatic hydrolysis using lipase (from *Rhizopus arrhizus*).¹⁰⁾ The lipase catalyzed hydrolysis of 1 afforded (2'R)-6-*O*-acylglyceryl β-D-galactopyranoside (2), ¹¹⁾ and the fatty acid attached at C-3' in 30% yield, involving the recovery of 1 (52%).¹²⁾ The structure of 2 was confirmed by the physicochemical property that the proton signal ascribable to 3'-H₂ was observed in the range of 4.15-4.20 ppm. Treatment of 2 with NaOMe-MeOH gave a mixture of methyl esters of fatty acids liberated from C-6. Gas-liquid chromatography analysis of both fatty acid mixtures as their methyl esters showed the composition of two acyl residues linked to C-6 and C-3'.¹³⁾

January 1992 287

In conclusion, we have isolated and characterized the novel glycolipid attaching two acyl groups to C-6 and C-3' in (2'R)-glyceryl β -D-galactopyranoside from the nitrogen-fixing cyanobacterium. The physicochemical properties and biological activities of 1 may be of interest because of the low content of unsaturated fatty acids and the unprecedented positions of acyl groups.

ACKNOWLEDGEMENT We are grateful to Mr. M. Murata of Taisho Pharmaceutical Co., Ltd. for assisting with measurement of HMBC spectrum.

REFERENCES AND NOTES

- 1) C. V. Baalen, "The Cyanobacteria", ed by P. Fay and C. V. Baalen, Elsevier, Amsterdam, 1987, pp. 187-198.
- 2) I. Ishizuka and T. Yamakawa, "New Comprehensive Biochemistry", Vol.10, ed. by A. Neuberger, L. L. M. Van Deenen, and H. Wiegant, Elsevier, Amsterdam, 1985, pp. 101-198.
- 3) The strain of Anabaena flos-aquae f. flos-aquae (NIES-74) used for the present investigation was purchased from the National Institute for Environmental Studies of the Environment Agency.
- 4) CB-medium is composed of Ca(NO₃)₂·4H₂O 0.15; KNO₃ 0.1; β-Na₂ glycerophosphate 0.05; MgSO₄ 7H₂O 0.04; Bicine FeCl₃·6H₂O 0.196; MnCl₂·4H₂O 0.036; ZnSO₄·7H₂O 0.022; CoCl₂·6H₂O 0.004; Na₂MnO₄·2H₂O 0.0025; Na₂EDTA·2H₂O 1.0; in g/L.The trace elements solution is composed of vitamin B₁₂ 0.1; biotin 0.1; thiamine-HCl 10.0; in mg/L. The pH of the medium was adjusted to 9.0.
- 5) The algal extract was separated on the constituents, positive coloration to anthrone reagent in order to obtain glycolipids. The FAB-MS spectrum of 1 showed the most intense peak at m/z737, corresponding to a molecular formula of C₄₁H₇₈O₁₀Li (m/z737.5693, -6.2 mmu error).
- 6) N. Murakami, H. Imamura, J. Sakakibara, and N. Yamada, Chem. Pharm. Bull., 38, 3497(1990).
- 7) K. Yamasaki, R. Kasai, Y. Masaki, M. Okihara, and O. Tanaka, Tetrahedron Lett., 1977, 1231.
- 8) lit. $[\alpha]_D^{15}$ -9° (c=0.6, H₂O). H. Kikuchi, Y. Tsukitani, T. Manda, T. Fujii, H. Nakanishi, M. Kobayashi, and I. Kitagawa, *Chem. Pharm. Bull.*, 30, 3544(1982).
- 9) Since low solubility of the glycolipid (1) in CH₃CN and MeOH caused difficulty in applying the reversed phase HPLC, further separation of 1 to a compound having a single acyl composition has not been achieved yet.
- 10) N. Murakami, H. Imamura, T. Morimoto, T. Ueda, S. Nagai, J. Sakakibara, and N. Yamada, *Tetrahedron Lett.*, 32, 1331(1991).
- 11) A white amorphous powder. $[\alpha]_D^{25} + 6.0^{\circ} (c=0.4, CHCl_3)$. IR (KBr, cm⁻¹): 3406, 1737. FAB-MS(m/z): 499(M+Li)⁺.
- 12) In the hydrolysis of 1 using the lipase, deacylation of C-3' acyl group predominated without generation of a 3-monoacyl derivative and monogalactosyl glycerol at this stage.
- 13) The conditions for identification of fatty acid methyl esters: column, ULBON HR-SS-10 (0.25mm i.d. × 50m, Shinwa kako Co.Ltd.); column temperature, 150-220°C, 3°C/min; injection temperature, 250°C; carrier gas, N₂, 2.2kg/cm².

(Received November 5, 1991)