Synthesis of Saframycins. VIII. Synthesis of the ABC Ring of Safracins

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An efficient synthesis of the 1,5-imino-3-benzazocine derivative (18) from 1,4-diacetyl-2,5-piperazinedione (7) and 4-methoxy-3-methylbenzaldehyde (8) is described. The tricyclic lactam (18) is shown to be a useful intermediate for preparation of a phenol (5), which has the structure of the ABC ring system of safracins and ecteinascidins. This conversion is a promising route to total synthesis of (\pm) -safracin A (1a).

Keywords synthesis; safracin; ecteinascidin; 1,5-imino-3-benzazocine; biogenetic intermediate

Safracins A (1a) and B (1b) were first isolated by the Yoshitomi Laboratories group from Pseudomonas fluorescens A2-2 in 1983. The Squibb Laboratories group independently isolated 1b from Pseudomonas fluorescens SC 12695 in the same year.²⁾ Both safracins were active against mouse tumors, L1210 and P388 leukemias and B16 melanoma. The toxic and effective doses of 1b were much lower than those of 1a.3 The structure of the safracins with the absolute configuration was elucidated by X-ray crystallography of the 4-brominated derivative of safracin A (1c).4,5) The safracin structure is similar to that of saframycins (3); however, the pyruvamide side chain and one of the p-quinone parts of the saframycins are substituted, respectively, by an alaninamide chain and a monophenol ring in safracins. Ecteinascidins (4) with potent in vivo antitumor activity were independently isolated from the colonial tunicate Ecteinascidia turbinata by Rinehart et al. 6a,b) and Wright et al. 6c) in 1990, and the structures assigned to them were similar to those of safracins.

Saframycin Mxs-1 (2a) and -2 (2b) were isolated from *Myxococcus xanthus* Mx x 48 in 1988.⁷⁾

It is interesting that these monoquinone-type antibiotics. such as saframycins D (3d) and F (3e), safracins A (1a) and B (1b), and saframycin Mxs-1 (2a) and -2 (2b), have a quinone moiety at ring E and a highly substituted benzene ring A with a variety of oxidation levels. The possible biosynthetic pathway shown in Fig. 2 involves initial oxidation of a type I compound (such as safracins) to form a type II compound (such as saframycins A and B), into which a hydroxy group could be introduced at the C-5 position to give a type III compound, which in turn would be oxidized and reduced (redox process) to form a type IV compound (such as saframycins D and F). On the other hand, introduction of a methoxy group at the C-5 position of a type II compound would give a type V compound (such as saframycin C), which could then be reduced to a type VI compound (such as saframycin Mxs). We reported the total synthesis of (\pm) -saframycin B (3b), $^{8a)}$ the trans-

$$\begin{array}{c} \text{CH}_3 \\ \text{CH}_3 \\$$

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Chart 1

formation of (\pm) -saframycin B (3b) (type II) to (\pm) -saframycins C (3c) (type V) and D (3d) (type IV), show and recently the transformation of (-)-saframycin A (3a) (type II) to a (-)-saframycin Mx type compound (type VI) via the 5-methoxybisquinone (type V). We became interested in safracins as attractive synthetic targets because they were plausible and important biogenetic intermediates of saframycins and saframycin Mxs. We report here the first synthesis of the ABC ring system of safracins and ecteinascidins using the methods described in connection with our total synthesis of the ABC ring system of saframycins. 9)

0

5

disfavor

Our initial strategy for the synthesis of the ABC ring model compound (5) was based on the retrosynthetic analysis outlined in Chart 1. To prevent the formation of any unwanted tetrahydroisoquinoline isomer from 6, it was decided to introduce the hydroxy group at a late stage.

A mixture of commercially available 4-methoxy-3-methylbenzaldehyde (8) and 1,4-diacetyl-2,5-piperazinedione (7) in dimethylformamide (DMF) was treated with potassium tert-butoxide in tert-butyl alcohol at room temperature for 15 h to afford 3-arylidene-2,5-piperazinedione (9) in 74% yield (Chart 2). 4-Methoxybenzylation of 9 with 4-methoxybenzyl chloride in the presence of sodium hydride in DMF at room temperature for 1 h furnished 10 in quantitative yield, and successive treatment with hydrazine hydrate afforded 11 in 70% yield. Methylation of 11 with methyl iodide in the presence of sodium hydride in DMF at room temperature for 1 h furnished 12 in quantitative yield. Deprotection of 12 with trifluoroacetic acid (TFA) and concentrated H₂SO₄ at room

temperature for 14h gave 13 in 92% yield. Catalytic hydrogenation of 13 with hydrogen (3.5 atm) over 20% palladium on carbon in ethanol at 80 °C for 14 h furnished 14 in 76% yield. The piperazine ring of 14 was activated by introduction of an isopropoxycarbonyl group to give 15 in 78% yield. The chemoselective reduction of 15 with an excess of lithium tri-tert-butoxyaluminohydride in tetrahydrofuran (THF) afforded a diastereomeric mixture of the alcohol (6), which on treatment with formic acid afforded the desired cyclization product (16) in 85% overall yield. 10,11) The structure of 16 is supported by the proton nuclear magnetic resonance (¹H-NMR) spectrum, which displays the signals of H-1 as a singlet at δ 6.59 and H-4 as a singlet at δ 6.87. Deprotection of 16 with TFA and H₂SO₄ gave the secondary amine (17) in 81% yield. Methylation of 17 with formaldehyde and formic acid at 70 °C for 1 h gave **18** in 74% yield.

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The next stage of the investigation required establishing a method to introduce a hydroxy group at the C-1 position of compound 18 (Chart 3). Treatment of 18 with boron tribromide at $-20\,^{\circ}$ C for 1 h afforded the phenol (19) in 75% yield, 12) but numerous efforts to convert 19 to the o-quinone 20 using potassium nitrosodisulfonate (Fremy's salt), 13a) lead tetraacetate, 13b) and benzeneseleninic anhydride 13c) were all unsuccessful. 14)

These results indicated that the introduction of a hydroxy group required a different approach. Nitration of **18** with $HNO_3-H_2SO_4$ at 0 °C for 2 h afforded **21** in 80% yield. ¹⁵⁾ Hydrogenation of **21** with 10% palladium on carbon in ethanol gave **22** in 85% yield. Finally, the reaction of **22** with sodium nitrite in H_2SO_4 and water for 1 h gave the

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diazonium salt (23), which was subsequently treated with 20% $\rm H_2SO_4$ at 140 °C for 5 min to provide the desired phenol (5) in 5% yield. The major products were the reduction products 18 and 19, obtained in 22% and 16% yields, respectively. Assignment of 5 was made by $^1\rm H-NMR$ analysis. When H-4 (δ 6.49) was irradiated,

nuclear Overhauser enhancement (10%) of the methyl protons at δ 2.26 was observed.

In summary, we have succeeded in synthesizing the ABC ring system of safracins and ecteinascidins. The results described herein are being applied to the total synthesis of (\pm) -safracin A (1a) which is currently under way in our

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laboratories.

Experimental

All melting points were determined with a Yanagimoto micromelting point apparatus and are uncorrected. MS (electron impact (EI)) were obtained on a JEOL JMS-DX 302 spectrometer. IR spectra were recorded on Hitachi 260-10 and 270-30 spectrophotometers. UV spectra were determined in methanol with a Hitachi 340 spectrophotometer. ¹H-NMR spectra were obtained at 270 MHz with a JEOL EX 270 spectrometer. ¹³C-NMR spectra were measured at 67.5 MHz with a JEOL EX 270 spectrometer (multiplicity determined from off-resonance decoupled or insensitive nuclei enhanced by polarization transfer (INEPT) spectra). NMR spectra were taken in CDCl₃ and chemical shifts are reported in δ values in parts per million relative to tetramethylsilane as an internal standard. Column chromatography was performed with E. Merck Silica gel 60 (70-230 mesh). Elemental analyses were obtained by using a Perkin-Elmer Model 240B elemental analyzer. All reactions were conducted under an argon atmosphere. Dry solvents and reagents were obtained using standard procedures. Anhydrous sodium sulfate was used for drying organic solvent extracts; removal of the solvent was done with a rotary evaporator and finally, under high vacuum.

(Z)-1-Acetyl-3-(4-methoxy-3-methylphenylmethylene)-2,5-piperazinedione (9) A solution of potassium tert-butoxide (31.42 g, 0.28 mol) in tert-butyl alcohol (580 ml) was added to a stirred solution of 4-methoxy-3-methylbenzaldehyde 8 (42.0 g, 0.28 mol) and 1,4-diacetyl-2,5-piperazinedione 7 (55.5 g, 0.28 mol) in dry DMF (560 ml). After being stirred for 15 h at room temperature, the reaction mixture was poured into brine (21) and extracted with ethyl acetate (500 ml \times 3). The combined extracts were washed with water (500 ml), dried, and concentrated in vacuo to give the residue as a pale yellow solid, recrystallization of which from ethyl acetate gave 9 (59.4 g, 74%) as pale yellow prisms, mp 201.5—204.5 °C. Anal. Calcd for $C_{15}H_{16}N_2O_4$: C, 62.49; H, 5.59; N, 9.72. Found: C, 62.40; H, 5.56; N, 9.65. MS m/z (%): 288 (M⁺, 100), 247 (12), 246 (75), 231 (10), 161 (17), 147 (21). IR (KBr): 3240, 1710, 1695, 1638, 1615 cm⁻¹. UV λ_{max} nm (log ε): 237 (4.17), 338 (4.31). ¹H-NMR δ: 2.24 (3H, s, ArCH₃), 2.65 (3H, s, COCH₃), 3.88 (3H, s, OCH₃), 4.52 (2H, s, $6-H_2$), 6.88 (1H, d, J=8.6 Hz, 5'-H), 7.12 (1H, s, C=CH), 7.20 (1H, d, J=2.0 Hz, 2'-H), 7.25 (1H, dd, J=8.6, 2.0 Hz, 6'-H), 7.93 (1H, s, NH).

(Z)-3-(4-Methoxy-3-methylphenylmethylene)-4-(4-methoxyphenylmethyl)-2,5-piperazinedione (11) Sodium hydride (60% oil dispersion, washed with dry hexane three times, 1.62 g, 67.5 mmol) was added to a stirred solution of 9 (14.4 g, 50 mmol) in dry DMF (250 ml) under ice-cooling, and stirring was continued for 1 h at 0 °C. 4-Methoxybenzyl chloride (10.45 g, 66.7 mmol) in dry DMF (100 ml) was added during 10 min, and the reaction mixture was stirred for 1 h at room temperature. The reaction mixture was concentrated in vacuo, and the residue was diluted with water (100 ml) and extracted with benzene (100 ml × 3). The combined extracts were washed with water (100 ml), dried, and concentrated in vacuo to furnish 10 (20.4 g, 100%) as a pale yellow oil, which was used for the next step without further purification. An analytical sample was obtained by crystallization from ether to give pure 10 as pale yellow prisms, mp 155—156°C. Anal. Calcd for C₂₃H₂₄N₂O₅: C, 67.63; H, 5.92; N, 6.86. Found: C, 67.51; H, 5.95; N, 6.84. MS m/z (%): 408 (M⁺, 19), 122 (9), 121 (100). IR (KBr): 1715, 1690, 1625, 1610 cm⁻¹. UV λ_{max} nm (log ϵ): 229 (4.36), 248 sh (4.07), 278 sh (3.83), 284 sh (3.87), 333 (4.29). ¹H-NMR δ : 2.26 (3H, s, ArCH₃), 2.51 (3H, s, COCH₃), 3.75, 3.90 (each 3H, s, OCH₃), 4.50 (2H, s, 6-H₂), 4.65 (2H, s, NCH₂), 6.73 (2H, d, $J=8.6\,\mathrm{Hz},\ 2\times\mathrm{ArH}),\ 6.87\ (2\mathrm{H},\ \mathrm{d},\ J=8.6\,\mathrm{Hz},\ 2\times\mathrm{ArH}),\ 6.89\ (1\mathrm{H},\ \mathrm{d},\ \mathrm{d})$ J = 8.6 Hz, 5'-H), 7.20 (1H, s, C=CH), 7.22 (1H, d, J = 2.3 Hz, 2'-H), 7.29 (1H, dd, J=8.6, 2.3 Hz, 6'-H)

Hydrazine monohydrate (20 ml) was added to a stirred solution of the crude 10 (20.4 g) in dry DMF (250 ml), and the resulting solution was stirred for 1 h at room temperature. The reaction mixture was poured into water (500 ml), and extracted with chloroform (300 ml × 3). The combined extracts were washed with brine (300 ml), dried, and concentrated *in vacuo* to give a solid, recrystallization of which from ethyl acetate–ether gave 11 (12.81 g, 70% from 9) as pale yellow prisms, mp 174—177 °C. *Anal.* Calcd for C₂₁H₂₂N₂O₄·1/4H₂O: C, 68.26; H, 6.11; N, 7.55. Found: C, 68.26; H, 6.15; N, 7.51. MS m/z (%): 366 (M⁺, 28), 160 (4), 122 (10), 121 (100). IR (KBr); 3200, 1690, 1675, 1625, 1605 cm⁻¹. UV $\lambda_{\rm max}$ nm (log ε): 227 (4.35), 314 (4.29). ¹H-NMR δ: 2.24 (3H, s, ArCH₃), 3.73, 3.88 (each 3H, s, OCH₃), 4.05 (2H, d, J=2.0 Hz, 6-H₂), 4.61 (2H, s, NCH₂), 6.71 (2H, d, J=8.6 Hz, 2×ArH), 6.86 (1H, d, J=8.6 Hz, 5'-H), 6.89 (2H, d, J=8.6 Hz, 2×ArH), 7.13 (1H, s, C=CH), 7.16 (1H, d, J=2.3 Hz, 2'-H), 7.22 (1H, dd, J=8.6, 2.3 Hz, 6'-H), 7.54 (1H, br s, NH).

(Z)-1-Methyl-3-(4-methoxy-3-methylphenylmethylene)-2,5-piperazinedione (13) Sodium hydride (60% oil dispersion, washed with dry hexane three times, 864 mg, 26.0 mmol) was added to a stirred solution of 11 (10.98 g, 30 mmol) in dry DMF (80 ml) under ice-cooling, and stirring was continued for 30 min at 0 °C. Methyl iodide (2.24 ml, 36.0 mmol) in dry DMF (20 ml) was added during 30 min, and the reaction mixture was stirred for 1 h at room temperature. The reaction mixture was concentrated in vacuo, and the residue was diluted with water (100 ml) and extracted with benzene (100 ml × 3). The combined extracts were washed with water (100 ml), dried, and concentrated in vacuo to furnish 12 (11.4 g, 100%) as a pale yellow oil, which was used for the next step without further purification. An analytical sample was obtained by crystallization from ether to give pure 12 as pale yellow prisms, mp 73—75 °C. MS m/z (%): 380 (M⁺, 41), 122 (9), 121 (100). High-resolution MS Calcd for C₂₂H₂₄N₂O₄: 380.1736. Found: 380.1749. IR (KBr): 1670, 1650, $1610 \,\mathrm{cm}^{-1}$. UV λ_{max} nm (log ε): 229 (4.27), 284 sh (3.99), 314 (4.20). ¹H-NMR δ : 2.24 (3H, s, ArCH₃), 3.01 (3H, s, NCH₃), 3.73, 3.88 (each 3H, s, OCH₃), 4.05 (2H, s, 6-H₂), 4.64 (2H, s, NCH₂), 6.72 (2H, d, $J = 8.9 \text{ Hz}, 2 \times \text{ArH}$), 6.85 (1H, d, J = 8.6 Hz, 5'-H), 6.89 (2H, d, J = 8.9 Hz, $2 \times ArH$), 7.13 (1H, s, C=CH), 7.15 (1H, d, J=2.3 Hz, 2'-H), 7.21 (1H, dd, J = 8.6, 2.3 Hz, 6'-H).

Concentrated $\rm H_2SO_4$ (1.5 ml) was added to a stirred solution of the crude 12 (11.4 g) in TFA (50 ml), and stirring was continued for 14 h at room temperature. The reaction mixture was poured into brine (500 ml) and extracted with chloroform (200 ml × 3). The combined extracts were washed with 10% aqueous NH $_3$ (200 ml), dried, and concentrated *in vacuo* to give a solid, recrystallization of which from methanol afforded pure 13 (7.145 g, 92% from 11) as pale yellow needles, mp 192.5—194 °C. MS m/z (%): 262 (M $^+$, 11), 135 (100). High-resolution MS Calcd for C $_{14}$ H $_{16}$ N $_{2}$ O $_{3}$: 260.1161. Found: 260.1180. IR (KBr): 3600—3100, 1670, 1615, 1595 cm $^{-1}$. UV λ_{max} nm (log ε): 230 (4.16), 316 (4.21). ¹H-NMR δ : 2.23 (3H, s, ArCH $_{3}$), 3.96 (3H, s, NCH $_{3}$), 3.86 (3H, s, OCH $_{3}$), 4.16 (2H, s, 6-H $_{2}$), 6.86 (1H, d, J=8.3 Hz, 5'-H), 6.99 (1H, s, C=CH), 7.12 (1H, br s, 2'-H), 7.21 (1H, br d, J=8.3 Hz, 6'-H), 7.99 (1H, br s, NH).

3-(4-Methoxy-3-methylphenylmethyl)-1-methyl-2,5-piperazinedione (14) A solution of 13 (6.50 g, 25 mmol) in ethanol (50 ml) was shaken for 14 h at 80 °C under 3.5 atm of hydrogen in the presence of 20% palladium on carbon (1.2 g). The catalyst was removed by filtration and washed with ethanol (450 ml). The combined filtrates were evaporated and the residue was diluted with water (200 ml), and extracted with chloroform (200 ml \times 3). The combined extracts were washed with 5% NaHCO₃, dried, and concentrated in vacuo to give a solid, recrystallization of which from acetone afforded pure 14 (4.978 g, 76%) as colorless needles, mp 169-170.5°C. Anal. Calcd for C₁₄H₁₈N₂O₃: C, 64.10; H, 6.92; N, 10.68. Found: C, 63.97; H, 7.00; N, 10.62. MS m/z (%): 262 (M⁺, 11), 135 (100). IR (KBr): 3250, 1685, 1655 cm $^{-1}$. UV $\lambda_{\rm max}$ nm (log ε): 231 (3.94), 277 (3.19), 284 (3.05). $^{1}\text{H-NMR}$ δ : 2.19 (3H, s, ArCH₃), 2.85 (3H, s, NCH₃), 3.01 (1H, dd, J = 13.9, 6.6 Hz, ArCH), 3.08 (1H, dd, J = 13.9, 4.3 Hz, ArCH), 3.10 (1H, d, J=17.5 Hz, 6-H), 3.61 (1H, d, J=17.5 Hz, 6-H), 3.82 (3H, s, OCH_3), 4.21 (1H, m, 3-H), 6.14 (1H, br s, NH), 6.76 (1H, d, J=7.9 Hz, 5'-H), 6.95 (1H, d, J=2.0 Hz, 2'-H), 6.96 (1H, dd, J=7.9, 2.0 Hz, 6'-H).

3-(4-Methoxy-3-methylphenylmethyl)-1-methyl-4-isopropoxycarbonyl-2,5-piperazinedione (15) A solution of 14 (3.93 g, 15 mmol), triethylamine (4.18 ml, 30 mmol), and 4-dimethylaminopyridine (3.665 g, 30 mmol) in dry dichloromethane (200 ml) was cooled with ice-water, and isopropyl chloroformate (6.81 ml, 60 mmol) was added dropwise over 10 min. The solution was stirred for 1 h at room temperature. The organic layer was washed with 1 N HCl (50 ml), and then water (50 ml), dried, and concentrated in vacuo to give the residue (9.57 g). Chromatography on a silica gel (120 g) column with hexane-ethyl acetate (1:1) as the eluent gave 15 (4.08 g, 78%) as a colorless amorphous powder. MS m/z (%): 348 (M⁺, 17), 136 (10), 135 (100). High-resolution MS Calcd for $C_{18}H_{24}N_2O_5$: 348.1685. Found: 348.1673. IR (CHCl₃): 1775, 1725, 1625 cm⁻¹. UV λ_{max} nm (log ε): 228 (4.01), 277 (3.27), 284 (3.19). ¹H-NMR δ : 1.36, 1.39 (each 3H, d, J = 6.6 Hz, CHC $\underline{\text{H}}_3$), 2.17 (3H, s, ArCH₃), 2.31 (1H, d, J = 18.2 Hz, 6-H), 2.76 (3H, s, NCH₃), 3.12 (1H, dd, J = 14.2, 4.6 Hz, ArCH), 3.27 (1H, dd, J = 14.2, 3.6 Hz, ArCH), 3.40 (1H, d, J = 18.2 Hz, 6-H), 3.81 (3H, s, OCH_3), 4.95 (1H, dd, J=4.6, 3.6 Hz, 3-H), 5.15 (1H, sept, J=6.6 Hz, OCH), 6.72 (1H, d, J=8.9 Hz, 5'-H), 6.95 (1H, d, J=2.0 Hz, 2'-H), 6.96 (1H, dd, J=7.9, 2.0 Hz, 6'-H). ¹³C-NMR δ : 15.9 (q, ArCH₃), 21.7 (q, CHCH₃), 21.7 (q, CHCH₃), 32.8 (q, NCH₃), 38.4 (t, ArCH₂), 52.6 (t, 6-C), 55.3 (d, 3-C), 60.2 (q, OCH₃), 72.2 (d, OCH), 109.8 (d,), 125.9 (s), 127.2 (s), 128.7 (d), 132.2 (d), 151.1 (s, COO), 157.6 (s), 164.4 (s, CO), 166.0 (s, CO)

1,2,3,4,5,6-Hexahydro-1,5-imino-9-methoxy-3,8-dimethyl-4-oxo-3-

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benzazocine-11-carboxylic Acid Isopropyl Ester (16) A stirred solution of 15 (972 mg, 2.79 mmol) in dry THF (60 ml) was cooled with ice-water, and lithium tri-tert-butoxyaluminohydride (2.841 g, 11.2 mmol) was added over 15 min. After continued stirring at the same temperature for 1 h, the reaction mixture was quenched by addition of water (10 ml). The reaction mixture was filtered through a Celite pad, and filtrates were concentrated in vacuo. The unstable diastereomeric mixture of the alcohols 6 (1.05 g) obtained was used for the next step without isolation. A solution of 6 in formic acid (40 ml) was heated at 70 °C for 40 min. The reaction mixture was diluted with water (100 ml) and extracted with chloroform (50 ml \times 3). The combined extracts were washed with 5% NaHCO₃ (50 ml), dried, and concentrated in vacuo to give the residue (1.226 g). Chromatography on a silica gel (100 g) column with dichloromethane-methanol (100:1) as the eluent afforded a solid, recrystallization of which from ethyl acetate-ether gave 16 (787.0 mg, 85%) as colorless needles, mp 200.5-202 °C. Anal. Calcd for C₁₈H₂₄N₂O₄: C, 65.04; H, 7.28; N, 8.43. Found: C, 64.76; H, 7.25; N, 8.34. MS m/z (%): 332 (M⁺, 100), 260 (28), 247 (11), 246 (48), 245 (37), 219 (42), 218 (53), 175 (14), 43 (10). IR (KBr): 1710, 1690, 1640 cm $^{-1}$. UV $\lambda_{\rm max}$ nm (log ϵ): 226 (3.83), 234 sh (3.54), 280 (3.45), 288 (3.41). ¹H-NMR δ : 1.23, 1.27 (each 3H, d, J = 6.6 Hz, CHCH₃), 2.16 (3H, s, ArCH₃), 2.85 (3H, s, NCH₃), 2.99 (1H, d, J = 16.5 Hz, $6 - H\beta$), 3.12 (1H, dd, J = 16.5, 5.6 Hz, 6-H α), 3.23 (1H, d, J = 11.6 Hz, 2-H β), 3.81 (3H, s, OCH_3), 3.95 (1H, dd, J = 11.6, 4.3 Hz, 2-H α), 4.92—5.01 (2H, m, 5-H and OCH), 5.38 (1H, brd, 1-H), 6.59 (1H, s, 10-H), 6.87 (1H, brs, 7-H). ¹³C-NMR δ: 15.8 (q, ArCH₃), 22.1 (q, CHCH₃), 22.1 (q, CHCH₃), 31.1 (t, 6-C), 34.2 (q, NCH₃), 48.8 (d, 1-C), 53.3 (d, 5-C), 55.3 (q, OCH₃), 56.3 (t, 2-C), 69.4 (d, OCH), 107.5 (d, 10-C), 124.1 (s), 126.7 (s), 131.1 (d, 7-C), 132.6 (s), 153.5 (s, COO), 156.4 (s), 168.3 (s, 4-CO).

1,2,3,4,5,6-Hexahydro-1,5-imino-9-methoxy-3,8-dimethyl-4-oxo-3benzazocine (17) Concentrated H₂SO₄ (1.0 ml) was added to a stirred solution of 16 (1.04 g, 3.13 mmol) in TFA (20 ml), and stirring was continued for 15 h at room temperature. The reaction mixture was poured into water (100 ml), and extracted with chloroform (50 ml × 3). The combined extracts were washed with 10% aqueous NH₃ (100 ml), dried, and concentrated in vacuo to give a solid, recrystallization of which from ethyl acetate-ether afforded pure 17 (877.2 mg, 81%) as colorless prisms, mp 160—161 °C. Anal. Calcd for C₁₄H₁₈N₂O₂·1/3H₂O: C, 66.64; H, 7.46; N, 11.10. Found: C, 66.66; H, 7.31; N, 10.87. MS m/z (%): 246 (M⁺, 30), 175 (18), 174 (100). IR (KBr); 3295, 1640, 1625 cm⁻¹. UV λ_{max} nm $(\log \varepsilon)$: 226 (3.83), 234 sh (3.53), 282 (3.42), 288 (3.39). ¹H-NMR δ : 2.17 (3H, s, ArCH₃), 2.36 (1H, s, NH), 2.85 (3H, s, NCH₃), 2.95 (1H, dd, J = 16.8, 1.3 Hz, 6-H β), 3.11 (1H, dd, J = 16.8, 5.9 Hz, 6-H α), 3.22 (1H, dd, J=11.6, 1.0 Hz, 2-H β), 3.81 (3H, s, OCH₃), 3.91 (1H, ddd, J=5.9, 1.3, 1.0 Hz, 5-H), 3.98 (1H, dd, J=11.6, 4.6 Hz, 2-H α), 4.23 (1H, dd, J=4.6, 1.0 Hz, 1-H), 6.53 (1H, s, 10-H), 6.86 (1H, s, 7-H). ¹³C-NMR δ: 15.9 (q, ArCH₃), 32.1 (t, 6-C), 34.1 (q, NCH₃), 50.2 (d, 1-C), 53.6 (d, 5-C), 55.4 (q, OCH₃), 57.6 (t, 2-C), 107.5 (d, 10-C), 124.6 (s), 126.4 (s), 131.2 (d, 7-C), 134.3 (s), 156.2 (s), 170.2 (s, 4-CO).

1,2,3,4,5,6-Hexahydro-1,5-imino-9-methoxy-3,8,11-trimethyl-4-oxo-3benzazocine (18) Formaldehyde (37% solution in water, 14.3 ml) was added to a stirred solution of 17 (492 mg, 2.0 mmol) in formic acid (16.6 ml) at 60 °C. After being stirred at 70 °C for 1 h, the reaction mixture was poured into water (80 ml) and extracted with chloroform (40 ml \times 3). The combined extracts were washed with 5% NaHCO₃ (50 ml), dried, and concentrated in vacuo to give a solid, recrystallization of which from ethyl acetate gave 18 (386.7 mg, 74%) as colorless prisms, mp 149-150 °C. Anal. Calcd for C₁₅H₂₀N₂O₄·1/4H₂O: C, 68.03; H, 7.80; N, 10.58. Found: C, 68.06; H, 7.69; N, 10.33. MS m/z (%): 260 (M⁺, 27), 189 (18), 188 (100). IR (KBr): $1630 \,\mathrm{cm}^{-1}$. UV $\lambda_{\mathrm{max}} \,\mathrm{nm} \,(\log \varepsilon)$: $224 \,(3.81)$, $280 \,(3.37)$, $288 \,(3.33)$. ¹H-NMR δ : 2.17 (3H, s, ArCH₃), 2.50 (3H, s, NCH₃), 2.80 (1H, d, J=17.2 Hz, 6-H β), 2.84 (3H, s, NCH₃), 3.13 (1H, d, J=11.6 Hz, 2-H β), 3.15 (1H, dd, J=17.2, 6.6 Hz, 6-H α), 3.66 (1H, d, J=6.6 Hz, 5-H), 3.81 $(3H, s, OCH_3)$, 3.86 (1H, d, J = 5.0 Hz, 1-H), 3.99 (1H, dd, J = 11.6, 5.0 Hz, 1-H)2-H α), 6.52 (1H, s, 10-H), 6.86 (1H, s, 7-H). ¹³C-NMR δ : 15.9 (q, ArCH₃), 26.8 (t, 6-C), 33.9 (q, NCH₃), 39.9 (q, NCH₃), 55.2 (t, 2-C), 55.3 (q, OCH₃), 56.4 (d, 1-C), 59.7 (d, 5-C), 108.3 (d, 10-C), 123.9 (s), 126.3 (s), 130.9 (d, 7-C), 132.5 (s), 156.4 (s), 170.2 (s, 4-CO).

1,2,3,4,5,6-Hexahydro-9-hydroxy-1,5-imino-3,8,11-trimethyl-4-oxo-3-benzazocine (19) A dichloromethane solution of boron tribromide (1.0 m, 10 ml, 10 mmol) was added to a stirred solution of 18 (520.0 mg, 2.0 mmol) in dry dichloromethane (60 ml) at $-20\,^{\circ}\mathrm{C}$. After being stirred for 1 h at the same temperature, the reaction mixture was poured into ice-water (200 ml) and the phases were separated. The aqueous layer was brought to pH 7 with $10\%\,$ Na $_2\mathrm{CO}_3$, and extracted with dichloromethane (100 ml \times 3). The combined extracts were washed with water (100 ml), dried,

and concentrated *in vacuo* to give the residue (456.2 mg). Chromatography of the residue on a silica gel (60 g) column with chloroform—methanol (20:1) as the eluent afforded a solid, recrystallization of which from acetone—methanol gave **19** (371.2 mg, 75%) as colorless prisms, mp 249—250 °C. *Anal.* Calcd for $C_{14}H_{18}N_2O_2\cdot 1/5H_2O$: C, 67.28; H, 7.42; N, 11.20. Found: C, 67.21; H, 7.25; N, 11.30. MS m/z (%): 246 (M⁺, 28), 175 (18), 174 (100). IR (KBr): 1660, 1645, 1620 cm⁻¹. UV λ_{max} nm (log ε): 222 (3.81), 282 (3.43), 288 (3.42). ¹H-NMR δ : 2.18 (3H, s, ArCH₃), 2.47 (3H, s, NCH₃), 2.76 (1H, d, J=17.2 Hz, 6-H β), 2.83 (3H, s, NCH₃), 3.06 (1H, d, J=11.7 Hz, 2-H β), 3.15 (1H, dd, J=17.2, 6.6 Hz, 6-H α), 3.62 (1H, d, J=6.6 Hz, 5-H), 3.78 (1H, d, J=5.0 Hz, 1-H), 3.94 (1H, dd, J=11.7, 5.0 Hz, 2-H α), 6.50 (1H, s, 10-H), 6.82 (1H, s, 7-H). ¹³C-NMR δ : 15.8 (q, ArCH₃), 27.1 (t, 6-C), 34.1 (q, NCH₃), 39.8 (q, NCH₃), 55.0 (t, 2-C), 55.9 (d, 1-C), 59.6 (d, 5-C), 113.0 (d, 10-C), 123.1 (s), 124.6 (s), 131.1 (d, 7-C), 132.6 (s), 153.5 (s), 170.9 (s, 4-CO).

9-Acetoxy-1,2,3,4,5,6-hexahydro-1,5-imino-3,8,11-trimethyl-4-oxo-3benzazocine (19b) Acetic anhydride (0.5 ml) was added to a solution of 19 (49.2 mg, 0.2 mmol) in dry pyridine (2.0 ml) and the mixture was kept at room temperature for 1 h. After dilution with water (10 ml), the mixture was extracted with chloroform (10 ml × 3). The combined extracts were washed with 5% NaHCO₃, dried, and concentrated in vacuo. The residue (71.5 mg) was subjected to chromatography (silica gel, 8 g, methanol: dichloromethane = 1:50) to give a solid, recrystallization of which from ethyl acetate-methanol gave 19b (30.3 mg, 68%) as colorless prisms, mp 129—131 °C. Anal. Calcd for C₁₆H₂₀N₂O₃: C, 66.64; H, 6.99; N, 9.72. Found: C, 66.61; H, 6.98; N, 9.71. MS m/z (%): 288 (M+, 28), 217 (23), 216 (100), 174 (47). IR (KBr): 1760, 1650 cm $^{-1}$. UV λ_{max} nm (log ϵ): 216 (4.00), 272 (2.79), 278 (2.73). 1 H-NMR δ : 2.14 (3H, s, ArCH₃), 2.31 (3H, s, COCH₃), 2.49 (3H, s, NCH₃), 2.84 (3H, s, NCH₃), 2.86 (1H, d, $J = 17.5 \,\text{Hz}$, 6-H β), 3.09 (1H, dd, J = 11.6, 1.0 Hz, 2-H β), 3.19 (1H, dd, J=17.5, 6.6 Hz, 6-H α), 3.66 (1H, d, J=6.6 Hz, 5-H), 3.87 (1H, d, J=4.8 Hz, 1-H), 3.95 (1H, dd, J=11.6, 4.8 Hz, 2-H α), 6.79 (1H, s, 10-H), 6.98 (1H, s, 7-H). 13 C-NMR δ : 15.9 (q, ArCH₃), 20.8 (q, COCH₃), 27.3 (t, 6-C), 33.9 (q, NCH₃), 39.9 (q, NCH₃), 55.1 (t, 2-C), 55.9 (d, 1-C), 59.5 (d, 5-C), 120.3 (d, 10-C), 129.4 (s), 130.4 (s), 131.5 (d, 7-C), 133.5 (s), 147.8 (s), 169.3 (CO), 170.1 (s, 4-CO).

11-Formyl-1,2,3,4,5,6-hexahydro-1,5-imino-9-methoxy-3,8-dimethyl-4oxo-3-benzazocine (18a) Acetic anhydride (2.1 ml) was added to a stirred solution of 17 (492 mg, 2 mmol) in formic acid (6.3 ml) and the mixture was stirred for 1 h at room temperature. Removal of the solvent in vacuo afforded the residue, which was partitioned between chloroform and 5% NaHCO3 solution. The organic phase was washed with water, dried, and concentrated in vacuo to give a solid, recrystallization of which from methanol gave 18a (505.7 mg, 92.3%) as colorless needles, mp 267—270 °C. This product was a mixture of two rotational isomers (ratio, 2:1). Anal. Calcd for C₁₅H₁₈N₂O₃: C, 65.67; H, 6.61; N, 10.21. Found: C, 65.59; H, 6.60; N, 10.18. MS m/z (%): 274 (M⁺, 68), 231 (42), 204 (12), 203 (93), 202 (100), 174 (50), 159 (10), 131 (13), 130 (10). IR (KBr): 1650, 1640 cm⁻¹ UV λ_{max} nm (log ε): 222 (4.03), 280 (3.40), 288 (3.38). ¹H-NMR δ : 2.17 (3H, s, ArCH₃, both isomers), 2.86 ($2/3 \times 3H$, s, NCH₃, main isomer), $2.87 (1/3 \times 3H, s, NCH_3), 3.08 - 3.20 (2H, br, 6-H_2, both isomers), 3.24$ $(2/3H, d, J = 11.6 Hz, 2-H\beta, main isomer), 3.34 (1/3H, d, J = 11.6 Hz, 2-H\beta,$ minor isomer), 3.82 (3H, s, OCH₃, both isomers), 3.92 (2/3H, dd, J = 11.6, 4.6 Hz, $2\text{-H}\alpha$, main isomer), $3.93 (1/3 \text{H}, \text{dd}, J = 11.6, 4.6 \text{ Hz}, 2\text{-H}\alpha, \text{minor})$ isomer), 4.50 (2/3H, d, J = 6.3 Hz, 5-H, main isomer), 4.84 (1/3H, d, $J=4.6\,\mathrm{Hz}$, 1-H, minor isomer), 5.22 (1/3H, d, $J=6.3\,\mathrm{Hz}$, 5-H, minor isomer), 5.61 (2/3H, d, J = 4.6 Hz, 1-H, main isomer), 6.56 (1/3H, s, 10-H, minor isomer), 6.59 (2/3H, s, 10-H, main isomer), 6.88 (2/3H, s, 7-H, main isomer), 6.90 (1/3H, s, 10-H, minor isomer), 8.15 (2/3H, s, CHO, main isomer), 8.24 (1/3H, s, CHO, minor isomer).

11-Formyl-1,2,3,4,5,6-hexahydro-9-hydroxy-1,5-imino-3,8-dimethyl-4-oxo-3-benzazocine (19a) By the same procedure as used to prepare 19, *O*-demethylation of 18a (274 mg, 1 mmol) with a dichloromethane solution of boron tribromide (1.0 m, 5 ml, 5 mmol) in dry dichloromethane (30 ml) afforded 19a (200.7 mg, 77%) as colorless needles, mp 147—152 °C (chloroform–acetone). This product was a mixture of two rotational isomers (ratio, 5:2). *Anal.* Calcd for $C_{14}H_{16}N_2O_3 \cdot 2H_2O$: C, 57.53; H, 5.52; N, 9.59. Found: C, 57.74; H, 5.60; N, 9.25. MS m/z (%): 260 (M⁺, 82), 218 (11), 217 (76), 190 (12), 189 (90), 188 (100), 160 (64), 159 (10), 131 (13), 130 (10). IR (KBr): 3200, 1665, 1650, 1630, 1610 cm⁻¹. UV λ_{max} nm (log ε): 220 (3.98), 284 (3.38). ¹H-NMR δ: 2.18 (2/7 × 3H, s, ArCH₃, minor isomer), 2.21 (5/7 × 3H, s, ArCH₃, main isomer), 2.85 (5/7 × 3H, s, NCH₃, main isomer), 2.86 (2/7 × 3H, s, NCH₃, minor isomer), 2.87 (1/3 × 3H, s, NCH₃, both isomers), 3.01—3.15 (2H, br, 6-H₂, both isomers), 3.19 (5/7H, d, J=11.6 Hz,

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2-H β , minor isomer), 3.88 (2/7H, dd, J=11.9, 4.6 Hz, 2-H α , main isomer), 3.91 (1/3H, dd, J=11.6, 4.6 Hz, 2-H α , minor isomer), 4.50 (5/7H, d, J=5.9 Hz, 5-H, main isomer), 4.80 (2/7H, d, J=4.6 Hz, 1-H, minor isomer), 5.22 (2/7H, d, J=5.9 Hz, 5-H, minor isomer), 5.56 (5/7H, d, J=4.6 Hz, 1-H, main isomer), 6.60 (2/7H, s, 10-H, minor isomer), 6.62 (5/7H, s, 10-H, main isomer), 6.88 (5/7H, s, 7-H, main isomer), 6.89 (2/7H, s, 10-H, minor isomer), 8.15 (5/7H, s, CHO, main isomer), 8.23 (2/7H, s, CHO, minor isomer). 13C-NMR δ (CDCl₃-CD₃OD): main peaks; 15.4 (q, ArCH₃), 32.7 (t, 6-C), 34.0 (q, NCH₃), 45.7 (d, 1-C), 54.6 (d, 5-C), 55.9 (t, 2-C), 111.9 (d, 10-C), 121.9 (s), 125.4 (s), 130.9 (d, 7-C), 131.3 (s), 154.0 (s), 159.3 (s, CHO), 167.8 (s, CO), minor peaks; 15.4 (q, ArCH₃), 30.8 (t, 6-C), 34.0 (q, NCH₃), 49.8 (d, 5-C), 51.1 (d, 1-C), 57.4 (t, 2-C), 111.4 (d, 10-C), 122.6 (s), 125.5 (s), 130.8 (d, 7-C), 131.2 (s), 153.8 (s), 159.1 (s, CHO), 167.7 (s, CO)

1,2,3,4,5,6-Hexahydro-1,5-imino-9-methoxy-3,8,11-trimethyl-10-nitro-4oxo-3-benzazocine (21) The amine 18 (300.0 mg, 1.154 mmol) was added to a mixture of $\rm H_2SO_4\text{--}HNO_3$ (1:1, 2.0 ml) and the whole was stirred for 2 h under ice-cooling. The reaction mixture was diluted with water (50 ml), made alkaline with diluted aqueous NH₃, and extracted with chloroform $(30 \text{ ml} \times 3)$. The combined extracts were washed with water (30 ml), dried. and concentrated in vacuo to give the residue (360 mg). Chromatography of the residue on a silica gel (90 g) column with dichloromethane-methanol (50:1) as the eluent gave a solid, recrystallization of which from ethyl acetate-ether gave 21 (280.6 mg, 80%) as colorless prisms, mp 151-153 °C. Anal. Calcd for $C_{15}H_{19}N_3O_4$: C, 59.00; H, 6.27; N, 13.76. Found: C, 58.91; H, 6.28; N, 13.63. MS m/z (%): 305 (M⁺, 26), 234 (24), 233 (100), 186 (12). IR (KBr): 1660, 1530, 1400, 1380 cm⁻¹. UV λ_{max} nm (log ε): 216 (4.09), 272 (3.04), 306 (2.68). ¹H-NMR δ : 2.32 (3H, s, ArCH₃), 2.46 (3H, s, NCH₃), 2.87 (3H, s, NCH₃), 2.88 (1H, d, J = 17.2 Hz, 6-H β), 3.17 (1H, dd, J = 17.2, 6.6 Hz, 6-H α), 3.23 (1H, d, J = 12.2 Hz, 2-H β), 3.65 (1H, d, J=6.6 Hz, 5-H), 3.84 (3H, s, OCH₃), 3.88 (1H, d, J=12.2 Hz, 2-Hα), 3.88 (1H, s, 1-H), 7.09 (1H, s, 7-H). ¹³C-NMR δ: 15.9 (q, ArCH₃), 27.0 (t, 6-C), 33.8 (q, NCH₃), 40.0 (q, NCH₃), 51.9 (d, 1-C), 53.7 (t, 2-C), 58.5 (d, 5-C), 62.4 (q, OCH₃), 124.6 (s), 129.5 (s), 132.4 (s), 133.5 (d, 7-C), 145.3 (s), 148.4 (s), 169.5 (s, CO).

10-Amino-1,2,3,4,5,6-hexahydro-1,5-imino-9-methoxy-3,8,11-trimethyl-**4-oxo-3-benzazocine (22)** A solution of **21** (244.0 mg, 0.8 mmol) in ethanol (20 ml) was shaken for 7 h at room temperature under 1 atm of hydrogen in the presence of 10% palladium on carbon (70 mg). The catalyst was removed by filtration and washed with ethanol (100 ml). The combined filtrates were evaporated and the residue was diluted with water (50 ml), and extracted with chloroform (30 ml × 3). The combined extracts were washed with water (20 ml), dried, and concentrated in vacuo to give the residue (322.8 mg). Chromatography of the residue on a silica gel (20 g) column with dichloromethane-methanol (15:1) as the eluent gave a solid, recrystallization of which from ethyl acetate gave 22 (187.1 mg, 85%) as colorless prisms, mp 197—199 °C. Anal. Calcd for $C_{15}H_{21}N_3O_2 \cdot 1/6H_2O$: C, 64.72; H, 7.71; N, 15.10. Found: C, 64.71; H, 7.66; N, 15.15. MS m/z(%): 275 (M⁺, 28), 204 (28), 203 (100), 188 (21). IR (KBr): 3350, $1620 \,\mathrm{cm^{-1}}$. UV λ_{max} nm (log ε): 222 (4.05), 272 (3.75), 288 (3.27). ¹H-NMR δ : 2.29 (3H, s, $ArCH_3$), 2.50 (3H, s, NCH_3), 2.79 (1H, d, $J=17.2\,Hz$, $6-H\beta$), 2.86 (3H, s, NCH₃), 3.15 (1H, dd, J=17.2, 6.6 Hz, 6-H α), 3.19 (1H, d, J = 10.9 Hz, 2-H β), 3.61 (1H, d, J = 6.6 Hz, 5-H), 3.73 (3H, s, OCH₃), 3.88 (1H, d, $J = 5.0 \,\text{Hz}$, 1-H), 3.91 (1H, dd, J = 10.9, 5.0 Hz, 2-H α), 6.39 (1H, s, 7-H). 13 C-NMR δ : 15.8 (q, ArCH₃), 27.6 (t, 6-C), 34.0 (q, NCH₃), 40.2 (q, NCH₃), 51.7 (t, 2-C), 52.0 (d, 1-C), 59.2 (d, 5-C), 59.4 (q, OCH₃), 117.9 (s), 120.7 (d, 7-C), 128.2 (s), 129.7 (s), 136.2 (s), 144.2 (s), 170.2 (s, CO).

1,2,3,4,5,6-Hexahydro-10-hydroxy-1,5-imino-9-methoxy-3,8,11trimethyl-4-oxo-3-benzazocine (5) The amine 22 (55.0 mg, 0.2 mmol) was dissolved in concentrated H₂SO₄ (0.2 ml) and water (0.2 ml) and then cooled to below 5 °C. A solution of sodium nitrite (15.2 mg) in water (0.02 ml) was added dropwise under the surface of the ice-cooled solution with stirring at such a rate as to maintain the temperature at 0-5 °C. The solution was stirred for an additional 60 min. A solution of diazonium salt was added in portions to 20% H₂SO₄ (2 ml) at 140 °C over 5 min and after evolution of nitrogen gas ceased, the reaction mixture was diluted with water (10 ml), made alkaline with 5% NaHCO3, and extracted with chloroform ($20 \,\mathrm{ml} \times 3$). The combined extracts were washed with water (20 ml), dried, and concentrated in vacuo to give the residue (51.3 mg). Chromatography of the residue on a silica gel (6g) column with dichloromethane-methanol (30:1-10:1) as the eluent gave the residue (28.0 mg), which showed three major spots on TLC (dichloromethanemethanol (20:1)). This residues was subjected to chromatography on preparative layer silica gel plates (Merck 5715, solvent dichloromethane-methanol (200:15)) to afford **18** (11.1 mg, 22%), **19** (7.9 mg, 16%), and **5** (3.0 mg, 5%).

Compound 5: Colorless prisms from ethyl acetate, mp 217—218.5 °C. MS m/z (%): 276 (M⁺, 24), 204 (100), 189 (15). High-resolution MS Calcd for $C_{15}H_{20}N_2O_3$: 276.1474. Found: 276.1473. Calcd for $C_{12}H_{14}NO_2$: 204.1025. Found: 204.1011. IR (KBr): 3500—3050, 1645 cm⁻¹. UV λ_{max} nm (log ε): 226 (3.91), 272 (3.23), 278 (3.24). ¹H-NMR δ : 2.26 (3H, s, ArCH₃), 2.51 (3H, s, NCH₃), 2.85 (1H, d, J=16.2 Hz, 6-H β), 2.86 (3H, s, NCH₃), 3.16 (1H, dd, J=16.2, 6.6 Hz, 6-H α), 3.20 (1H, d, J=11.9 Hz, 2-H β), 3.65 (1H, d, J=6.6 Hz, 5-H), 3.78 (3H, s, OCH₃), 3.96 (1H, dd, J=11.9, 4.6 Hz, 2-H α), 4.22 (1H, d, J=4.6 Hz, 1-H), 5.93 (1H, br s, OH), 6.49 (1H, s, 7-H). ¹³C-NMR δ : 15.8 (q, ArCH₃), 27.3 (t, 6-C), 34.0 (q, NCH₃), 39.9 (q, NCH₃), 50.9 (q, 1-C), 52.6 (t, 2-C), 59.2 (d, 5-C), 60.8 (q, OCH₃), 118.8 (s, 10a-C), 122.0 (d, 7-C), 128.7 (s, 8-C), 129.5 (s, 6a-C), 143.4 (s, 9-C), 145.5 (s, 10-C), 169.8 (s, CO).

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References and Notes

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- 10) We previously reported that methanesulfonyl chloride and triethylamine in dichloromethane under reflux brought about mild and efficient dehydration/cyclization of the allylic alcohol to afford the cyclized compound in high yield.¹¹⁾ In this case, no cyclized compound was detected.
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steps (Chart 4). Oxidation of 19a with the above reagents was unsuccessful and afforded only a polar polymeric material.

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- 16) It is recognized that the standard method of preparing of phenols from diazonium salts is by heating the latter in a highly acidic aqueous solution, but the high temperature and acidity often cause a variety of unwanted side reactions. To convert diazonium salts to the corresponding phenols, Cohen et al., described a redox procedure using cuprous oxide. ¹⁷⁾ However, the conversion of 22 to the phenol (5) via the diazonium salt (23) gave only 2.7% yield.
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