Studies on Antiulcer Drugs. II.¹⁾ Synthesis and Antiulcer Activities of Imidazo[1,2-a]pyridinyl-2-alkylaminobenzoxazoles and 5,6,7,8-Tetrahydroimidazo[1,2-a]pyridinyl Derivatives

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A series of imidazo[1,2-a]pyridinylbenzoxazoles (4) and 5,6,7,8-tetrahydroimidazo[1,2-a]pyridinylbenzoxazoles (5) were synthesized and tested for anti-stress ulcer activity in rats. Several compounds were found to be more active than the reference compounds, sucralfate, cimetidine and ranitidine. Some of them exhibited potent protective activity against ethanol-induced gastric lesion. The synthesis and structure-activity relationships of these compounds are discussed.

Keywords anti-stress ulcer activity; ethanol-induced gastric lesion; cytoprotective activity; imidazo[1,2-a]pyridinylbenzoxazole; 5,6,7,8-tetrahydroimidazo[1,2-a]pyridinylbenzoxazole; sucralfate; cimetidine; structure–activity relationship

In a previous paper¹⁾ we reported the synthesis and pharmacological activities of a series of imidazo[1,2-a]-pyridinyl-2-oxobenz(o)-azolidines (1), imidazo[1,2-a]pyridinyl-3-oxo-2H-1,4-benz(o)-azines (2) and 5,6,7,8-tetra-hydroimidazo[1,2-a]pyridine analogues (3) as a novel class (i.e., two bicyclic hetero-ring system) of antiulcer agents. Some derivatives of these compounds showed good antiulcer activities in restraint and water-immersion stressed rats and cytoprotective activities against ethanol-induced gastric lesion in rats, and the potencies were superior to that of the prototype compound, zolimidine.

A previously obtained result in which annulation with heterocycles onto the phenyl ring improved the pharmacological activities prompted us to investigate a different type of benzo-fused heterocyclic moiety instead of the oxobenz(o)-azolidines and oxobenz(o)-azines.

In this paper, we describe the synthesis and pharmacological evaluation of imidazo[1,2-a]pyridinyl-2-alkylaminobenzoxazoles and 5,6,7,8-tetrahydroimidazo[1,2-a]pyridinyl analogues.

Chemistry The desired benzoxazoles (4 and 5) listed in Tables I and II were synthesized by the routes outlined in Charts 2—7.

The imidazo[1,2-a]pyridinylbenzoxazoles (4) were prepared by three different pathways. The first pathway is shown in Chart 2. Reduction of 6b, followed by treatment with alkylisocyanate gave o-hydroxyphenylureas (8). Cyclization of 8 with polyphosphate ester (PPE)²⁾ as the

zolimidine
$$\begin{array}{c} 1: \ Y = O, \ S \\ 2: \ Y = OCH(R^4), \ SCH(R^4), \ A = aromatic \\ 2: \ Y = OCH(R^4), \ SCH(R^4), \ A = aromatic \\ 3: \ Y = O, \ OCH(R^4), \ A = aromatic \\ 3: \ Y = O, \ OCH(R^4), \ A = aromatic \\ 3: \ Y = O, \ OCH(R^4), \ A = aromatic \\ 5: \ A = saturation \\ \hline \\ Chart \ 1 \\ \hline \\ C_2H_5CO \longrightarrow N_{PV} \longrightarrow NHR^3 \\ \hline \\ PPE \longrightarrow OH \\ \hline \\ A_{COH} \longrightarrow NHR^3 \longrightarrow PPE \longrightarrow OHR^3 \\ \hline \\ PPE \longrightarrow OH \\ \hline \\ A_{COH} \longrightarrow NHR^3 \longrightarrow PPE \longrightarrow OHR^3 \\ \hline \\ A_{COH} \longrightarrow NHR^3 \longrightarrow OHR^3 \longrightarrow OHR^3 \\ \hline \\ A_{COH} \longrightarrow OH \\ \hline \\ A_{COH} \longrightarrow OHR^3 \longrightarrow OHR^3 \longrightarrow OHR^3 \\ \hline \\ A_{COH} \longrightarrow OHR^3 \longrightarrow OHR^3 \\ \hline \\ A_{COH} \longrightarrow OHR^3 \longrightarrow OHR^3 \longrightarrow OHR^3 \\ \hline \\ A_{COH} \longrightarrow OHR^3 \longrightarrow OHR^3 \longrightarrow OHR^3 \\ \hline \\ A_{COH} \longrightarrow OHR^3 \longrightarrow OHR^3 \longrightarrow OHR^3 \\ \hline \\ A_{COH} \longrightarrow OHR^3 \longrightarrow OHR^3 \longrightarrow OHR^3 \\ \hline \\ A_{COH} \longrightarrow OHR^3 \longrightarrow OHR^3 \longrightarrow OHR^3 \longrightarrow OHR^3 \\ \hline \\ A_{COH} \longrightarrow OHR^3 \longrightarrow OHR^3 \longrightarrow OHR^3 \longrightarrow OHR^3 \\ \hline \\ A_{COH} \longrightarrow OHR^3 \longrightarrow OHR^3 \longrightarrow OHR^3 \longrightarrow OHR^3 \\ \hline \\ A_{COH} \longrightarrow OHR^3 \longrightarrow OHR^3 \longrightarrow OHR^3 \longrightarrow OHR^3 \longrightarrow OHR^3 \\ \hline \\ A_{COH} \longrightarrow OHR^3 \longrightarrow OH$$

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TABLE I. Physical Data of Imidazo[1,2-a]pyridinylbenzoxazoles (4)

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Compd.	R^1	R ²	\mathbb{R}^3	Method	Yield (%)	mp (°C) (Recryst.	Formula	Analysis (%) Calcd (Found)		
140.					(70)	solvent) ^{a)}		С	Н	
4a	Н	CH ₃	CH ₃	Α	22	217—219	C ₁₆ H ₁₄ N ₄ O	69.05	5.07	
4b	Н	CH_3	C_2H_5	Α	24	(A-H) 198—200	$C_{17}H_{16}N_4O$	(68.84 69.85	4.99 5.52	19.16
4c	6-CH ₃	Н	C_2H_5	B_2	56	(C-E) 150—151	$C_{17}H_{15}N_4O$	(69.60 69.85	5.50 5.52	19.16
4 d	6-CH ₃	CH_3	C_2H_5	Α	34	(AN) 238—241	$C_{18}H_{18}N_4O$	(69.74 70.57	5.71 5.92	
4e	7-CH ₃	Н	C_2H_5	$\mathbf{B_1}$	38	(M–I) 234—236	$C_{17}H_{16}N_4O$	(70.61 69.85	5.92 5.52	19.16
4f	7-CH ₃	CH ₃	Н	Α	41	(EA-M-T) 260261	$C_{16}H_{16}N_4O$	(69.72 69.05	5.61 5.07	20.13
4 g	7-CH ₃	CH ₃	CH ₃	Α	49	(EA) 245—246	$C_{17}H_{16}N_4O$	(68.77 69.85	5.02 5.52	19.16
4h	7-CH ₃	CH ₃	C_2H_5	\mathbf{B}_1	50	(M–I) 215—217	$C_{18}H_{18}N_4O$	(69.61 70.57	5.51 5.92	18.29
4 i	7-CH ₃	CH ₃	<i>n</i> -C ₃ H ₇	\mathbf{B}_{1}	44	(C) 203—204	$C_{19}H_{20}N_4O$	(70.32 71.23	5.93 6.29	17.49
4 j	7-CH ₃	CH ₃	iso-C ₃ H ₇	B_1	31	(A-H) 192—194	$C_{19}H_{20}N_4O$	(71.30 71.23	6.28 6.29	17.49
4k	7-CH ₃	CH ₃	$CH_2CH = CH_2$	$\mathbf{B_1}$	23	(EA) 168—169	C ₁₉ H ₁₈ N ₄ O	(70.84 70.87	6.39 5.76	17.40
41	7-CH ₃	CH ₃	$c-C_6H_{11}$	$\mathbf{B_1}$	31	(A–I) 209—212		(70.98 73.31	5.82 6.71	15.54
4m	7-CH ₃	Br	C_2H_5	C	79	(I–M) 230––231	$C_{17}H_{15}BrN_4O$	(73.53 55.00	6.62 4.07	15.10
4n	7-CH ₃	СНО	C_2H_5	B_3	36	(I–M–T) 220––221 (I–M–T)	$C_{18}H_{16}N_4O_2$	(54.77 67.49	4.04 5.03 4.95	17.49
40	$7-C_2H_5$	Н	C_2H_5	B_2	30	234—236 (A–EA)	$C_{18}H_{18}N_4O$	(67.33 70.57 (70.42	5.92 6.08	18.29
4p	$7-C_2H_5$	CH ₃	C_2H_5	Α	47	162—164 (A–EA)	$C_{19}H_{20}N_4O$	71.23	6.29	17.49
4 q	7 -iso- C_3H_7	Н	C_2H_5	\mathbf{B}_{2}	35	232—234 (I–M)	$C_{19}H_{20}N_4O$	(70.98 71.23 (71.15	6.25 6.29 6.41	17.49
4r	7-iso-C ₃ H ₇	CH ₃	C_2H_5	Α	68	193—195 (I–M)	$C_{20}H_{22}N_4O$	71.83 (71.55	6.63 6.74	16.75
4 s	8-CH ₃	Н	C_2H_5	B_2	71	173—175 (AN)	$C_{17}H_{16}N_4O$	69.85 (69.73	5.52 5.62	19.16
4t	8-CH ₃	CH ₃	C_2H_5	Α	69	212—214 (A–EA)	$C_{18}H_{18}N_4O$	70.57 (70.81	5.92 5.97	18.29

a) A, EtOH; AN, MeCN; C, CH₂Cl₂; EA, AcOEt; H, n-hexane; I, diisopropyl ether; M, MeOH; T, tetrahydrofuran.

dehydrating reagent furnished 2-alkylaminobenzoxazoles (9). Bromination of 9 with pyridinium bromide perbromide and subsequent condensation with the appropriate substituted 2-aminopyridines in the usual way³⁾ gave the desired compounds (4). The 2-unsubstituted aminobenzoxazole moiety was constructed from o-aminophenol derivative (7) by cyclization with cyanogen bromide (BrCN).⁴⁾ The resulting 2-aminobenzoxazole (9a) was brominated and condensed with 2-amino-4-methylpyridine to afford 4f (method A). The second pathway is shown in Chart 3. The 4-(substituted imidazo[1,2-a]pyridin-2-yl)-2-aminophenols (13), obtained by the manner described in the previous paper,¹⁾ were allowed to react with alkyl isocyanate to give the corresponding ureas (14). Cyclization of the ureas (14) with PPE provided the final products

(4) (method B₁). In the third pathway, shown in Chart 4, the urea derivatives (14) were also formed by the reaction of the acetophenone (15) with pyridinium bromide perbromide and subsequent condensation with 2-aminopyridines. The o-hydroxyphenylureas were similarly cyclized with PPE to afford the required products (4) (method B₂). Introduction of formyl group at the 3-position on the imidazo[1,2-a]pyridine ring was performed through the route given in Chart 5. Formylation of 19, prepared by the condensation of 18⁵⁾ with 2-amino-4-methylpyridine, with phosphorous oxychloride in N,N-dimethylformamide (DMF) produced 3-formylimidazo[1,2-a]pyridine derivative (20). Reduction of the nitro group and simultaneous debenzylation by catalytic hydrogenation over Pd-C at room temperature under atmospheric pressure of hydrogen

TABLE II. Physical Data of 5,6,7,8-Tetrahydroimidazo[1,2-a]pyridinylbenzoxazoles (5)

Compd. No.	R ¹ H H	R ²	\mathbb{R}^3	Method	Yield	mp (°C)			nalysis (% lcd (Foun	
	Н				(%)	(Recryst.	Formula		ica (i can	
5a		Н			(70)	solvent) ^{a)}		C	Н	N
	Н	**	C ₂ H ₅	Е	16	176—179 (AN)	C ₁₆ H ₁₈ N ₄ O	68.07 (67.76	6.43 6.38	19.84 19.75)
5b		CH ₃	C_2H_5	D_1	73	217—218 (AN)	$C_{17}H_{20}N_4O$	68.90 (69.13	6.80 6.82	18.90 18.65)
5c	6-CH ₃	Н	C_2H_5	D_1	33	194—195 (I–M)	$C_{17}H_{20}N_4O$	68.90 (69.28	6.80 7.01	18.90 18.88)
5d	6-CH ₃	CH_3	C_2H_5	D_1	24	228—231 (I–M)	$C_{18}H_{22}N_4O$	69.65 (69.47	7.14 7.28	18.05 17.83
5e	7-CH ₃	Н	Н	F	55	248—250 (A–I)	$C_{15}H_{16}N_4O$	64.96 (64.90	6.18 6.10	20.20 19.87
5f	7-CH ₃	Н	CH ₃	E	24	221—222 (AN)	$C_{16}H_{18}N_4O$ ·1/3 H_2O	66.65	6.52 6.56	19.43 19.59
5g	7-CH ₃	Н	C_2H_5	E	18	184—186 (A-EA)	$C_{17}H_{20}N_4O$	68.90 (68.98	6.80 6.94	18.90 18.91
5h	7-CH ₃	Н	n - C_3H_7	E	39	169—171 (AN)	$C_{18}H_{22}N_4O$	69.65 (69.50	7.14 7.12	18.05 17.96
5i	7-CH ₃	Н	iso-C ₃ H ₇	E	34	188—192 (A-AN)	$C_{18}H_{22}N_4O$	69.65 (69.35	7.14 7.42	18.05 18.02
5j	7-CH ₃	Н	<i>n</i> -C ₄ H ₉	Е	36	169—172 (I–M)	$C_{19}H_{24}N_4O$	70.34 (70.17	7.46 7.53	17.27 17.21
5k	7-CH ₃	Н	$CH_2CH = CH_2$	E	19	186—190 (I–M)	$C_{18}H_{20}N_4O$	70.11 (70.32	6.54 6.61	18.17 18.14
51	7-CH ₃	CH ₃	Н	$rac{D_2}{F}$	55 23	270—272 (EA-T)	$C_{16}H_{18}N_4O$	68.07 (67.79	6.43 6.60	19.84 19.98
5m	7-CH ₃	CH ₃	CH ₃	$_{\mathrm{E}}^{\mathrm{D_{2}}}$	68 15	250—252 (EA)	$C_{17}H_{20}N_4O$	68.90 (68.69	6.80 6.82	18.9 18.5
5n	7-CH ₃	CH ₃	C_2H_5	Е	24	224—227 (A–I)	$C_{18}H_{22}N_4O_2$	69.65 (69.52	7.14 7.31	18.0 17.8
50	7-CH ₃	CH ₃	<i>n</i> -C ₃ H ₇	D_1	62	191—193 (EA-T)	$C_{19}H_{24}N_4O$	70.34 (70.26	7.46 7.54	17.2 17.2
5p	7-CH ₃	CH ₃	iso-C ₃ H ₇	E	40	221—224 (A–I)	$C_{19}H_{24}N_4O$	70.34 (70.47	7.46 7.58	17.2 17.0
5q	7-CH ₃	CH ₃	<i>n</i> -C ₄ H ₉	E	40	1 3 7—138 (A–H)	$C_{20}H_{26}N_4O$	70.98 (70.63	7.74 8.06	16.5 16.3
5r	7-CH ₃	CH ₃	$CH_2CH = CH_2$		32	213—216 (A–AN)	$C_{19}H_{22}N_4O$	70.78 (71.05	6.88 6.95	17.3 17.3
5s	7-CH ₃	Br	C_2H_5	С	62	226—229 (A)	C ₁₇ H ₁₉ BrN ₄ O	54.41 (54.66	5.10 5.08	14.9 14.7
5t	$7-C_2H_5$	Н	C_2H_5	D_2	35	222—223 (I–M)	$C_{18}H_{22}N_4O$	69.65 (69.37	7.14 7.24	18.0 17.7
5u	$7-C_2H_5$	CH ₃	C_2H_5	D_2	66	218—222 (EA-C)	$C_{19}H_{24}N_4O$	70.34 (70.56	7.46 7.38	17.2 17.5
5v	7 -iso- C_3H_7	Н	C_2H_5	D_2	37	168—170 (EA-C)	$C_{19}H_{24}N_4O$	70.34 (70.08	7.46 7.48	17.2 17.3
5w	7 -iso- C_3H_7	CH ₃	C_2H_5	D_2	25	229—230 (EA-C)	$C_{20}H_{26}N_4O$	70.98 (70.72	7.74 7.88	16.5 16.3
5x	8-CH ₃	Н	C_2H_5	D_1	23	186—189 (A-H)	$C_{17}H_{20}N_4O$	68.90 (69.05	6.80 6.97	18.9 19.1
5 y	8-CH ₃	CH ₃	C_2H_5	D_2	25	185—186 (I–M)	$C_{18}H_{22}N_4O$	69.65 (69.41	7.14 7.32	18.0 17.9

a) A, EtOH; AN, MeCN; C, CH2Cl2; EA, AcOEt; H, n-hexane; I, diisopropyl ether; M, MeOH; T, tetrahydrofuran.

gave o-aminophenol derivative (21). Treatment of 21 with ethyl isocyanate afforded urea derivative (14h), which was cyclized with PPE to provide 4n (method B_3). The 3-bromo derivative (4m) of the imidazo[1,2-a]pyridine ring was obtained by the treatment of 4e with pyridinium bromide perbromide (Chart 6) (method C).

5,6,7,8-Tetrahydroimidazo[1,2-a]pyridinylbenzoxazoles

(5) were prepared from imidazo[1,2-a]pyridine derivatives (4) by catalytic reduction over Pd-C or PtO $_2$ at 3—9 atmospheric pressure of hydrogen (method D $_1$ or D $_2$). Alternatively, 5 could be obtained starting from the 4-(substituted imidazo[1,2-a]pyridin-2-yl)-2-nitrophenols (12). Catalytic reduction of 12 over Pd-C (3—5 atm), followed by treatment with alkyl isocyanate, and finally,

$$\begin{array}{c} R^2 C H_2 C C \\ C H_3 \\ C H_4 \\ C H_5 \\ C H_6 \\ C H_6 \\ C H_7 \\ C H_8 \\ C H_8 \\ C H_7 \\ C H_8 \\ C H_8 \\ C H_8 \\ C H_7 \\ C H_8 \\ C H_8$$

Chart 5

cyclization in the presence of PPE gave the desired with BrCN afforded 2-aminobenzoxazole derivatives (5e products (5) (method E). Cyclization of o-aminophenols (22b, $R^1 = 7$ -CH₃ $R^2 = H$ and 22c, $R^1 = 7$ -CH₃ $R^2 = CH_3$)

and 51), respectively (method F) (Chart 7).

Pharmacological Results and Discussion

The imidazo[1,2-a]pyridinylbenzoxazoles (4) and 5,6,7,8-tetrahydro derivatives (5) obtained in this study were evaluated for antiulcer activity at a dose of 32 mg/kg p.o. in restraint and water-immersion stressed rats (stress ulcer). The results were compared with those of reference compounds, sucralfate, cimetidine and ranitidine, and are summarized in Table III.

In the series of the imidazo [1,2-a] pyridines (4), it could be seen that 3-methyl, 3,7-dimethyl and 3,8-dimethyl groups on this nucleus produced enhanced activity. Of these, the compounds 4b, 4h and 4t possessing an ethyl group and 4k containing an allyl group at the 2-position on the benzoxazole moiety exhibited remarkable antiulcer activity. Compounds 4a, 4g, 4i and 4j, produced by replacing the ethylamino group in 4b and 4h with a methyl- or a propylamino group, showed fairly good activity. Introduction of a more bulky substituent as in 41, however, caused disappearance of activity. Less potent activity was observed for the compound 4f without substitution at the amino group. Concerning the substitution effect on the imidazo[1,2-a]pyridine ring, conversion of the 7-methyl group of 4h to an ethyl group (4p) showed moderate activity which was comparable to that of cimetidine. On the other hand, 3-bromo (4m), 3-formyl (4n) and 3-unsubstituted (4e) derivatives on the imidazo-[1,2-a]pyridine nucleus resulted in a marked fall in activity. Compound 4d, a positional isomer of the methyl group on the pyridine ring of 4h, also had weak activity.

In the series of 5,6,7,8-tetrahydroimidazo[1,2-a]pyridines (5), substitution at the 7-position on the tetrahydropyridine ring was advantageous for the expression of good activity. 7-Methyl (5e—g), 7-ethyl (5t and 5u), 7-isopropyl (5v) and 3,7-dimethyl (5l—p and 5r) derivatives revealed excellent potency. The activities of unsubstituted (5b) and 8-methyl (5y) derivatives on the tetrahydropyridine ring were relatively low. In particular, compounds 5a and 5x, which lack the methyl group on the 3-position, revealed completely loss of the activity. Replacement of the 3-

Table III. Anti-ulcer Activity of Benzoxazoles (4 and 5) against Restraint and Water-Immersion Stressed Rats (n=5)

Compd. No.	Anti-stress ulcer activity (% inhibition) 32 mg/kg p.o.	Compd. No.	Anti-stress ulcer activity (% inhibition) 32 mg/kg p.o.
4a	75.7°)	5e	98.8 ^{a)}
4b	$95.8^{b)}$	5f	98.1 ^{a)}
4c	NT	5g	97.9^{a}
4d	24.3	5h	$69.8^{b)}$
4e	14.6	5i	42.9
4f	46.0	5j	$56.3^{b)}$
4g	72.6	5k	67.6^{b}
4h	97.1 ^{a)}	51	$91.8^{b)}$
4i	79.7°)	5m	100^{a}
4j	$80.9^{a)}$	5n	$97.8^{b)}$
4k	$89.4^{a)}$	50	97.1 ^{b)}
41	2.2	5p	96.6^{a}
4m	8.9	5q	59.3
4n	39.6	5r	95.7^{a}
40	NT	5s	$59.2^{b)}$
4 p	68.2	5t	96.2^{a}
4q	NT	5u	100^{a}
4r	NT	5v	92.0a)
4s	NT	5w	76.1
4t	86.0	5x	-6.0
5a	-12.5	5у	$64.4^{b)}$
5b	$65.8^{b)}$	Sucralfate	45.6^{d}
5e	$61.5^{c)}$	Cimetidine	$69.4^{b)}$
5d	19.1	Ranitidine	81.8

a) p < 0.001. b) p < 0.01. c) p < 0.05. d) 100 mg/kg. NT = Not tested.

methyl group in **5n** with a bromo group (**5s**) caused a reduction of the activity. As a substituent on the 2-amino group of benzoxazole moiety, introduction of a bulky group (**5i**, **5j** and **5q**) resulted in a reduced activity like the former series (**4**), while unsubstituted amino derivatives (**5e** and **5l**) showed good activity, contrary to **4**. These results appear to suggest that the activity of tetrahydro derivatives (**5**) were generally equivalent or superior to that of the corresponding unsaturated compounds (**4**) except

in the cases of 5b and 5y versus 4b and 4t, respectively.

Next, dose–response activities of several compounds were examined on stress ulcer and cytoprotection against ethanol-induced gastric lesion (ethanol ulcer). The ED_{50} values and 95% confidence limits calculated in each test are shown in Table IV. All of the listed compounds were more efficacious than the reference compounds sucralfate, cimetidine and ranitidine on ethanol ulcer, and the activities of most of their derivatives on stress ulcer were superior to that of sucralfate and cimetidine.

In conclusion, we obtained several 2-aminobenzoxazole derivatives as novel structural antiulcer drugs. Although the reference drugs demonstrated a difference in their degree of activities on two ulcer models (stress ulcer and ethanol ulcer), these compounds synthesized showed strong inhibitory potency on both of the ulcer models.

Experimental

Melting points were determined on a Thomas-Hoover capillary

Table IV. ED₅₀ Values of Some Benzoxazole Derivatives on Anti-stress Ulcer and Cytopotective Activity

Compound No.	ED ₅₀ value (mg/kg, p.o.) (95% confidence limits)							
110.	Anti-stress ulcer activity	Cytoprotective activity						
4b	20.1 (5.3—76.7)	14.2 (9.9—20.3)						
4h	10.6 (6.3—17.8)	0.05^{b}						
4 j	15.1 (7.1—32.1)	0.10 (0.0063—2.17)						
4k	16.2 (11.6—22.8)	4.4 (0.04—9.6)						
5e	3.9 (1.3—11.7)	1.2 (0.18—8.5)						
5f	7.6 (4.8—12.0)	1.6 (0.45—5.9)						
5g	1.7 (0.32—8.8)	1.6 (0.51—5.0)						
51	9.1 (3.2—26.1)	6.3 (2.6—11.3)						
5 0	9.5 (6.7—13.5)	0.80 (0.20-3.5)						
5p	$0.13^{a)}$	2.64 (1.67—4.17)						
5r	9.4 (2.6—34.3)	9.5 (5.4—19.0)						
5t	1.3 (0.27—6.4)	4.2 (0.84—7.4)						
5u	7.3 (3.7—14.2)	12.5 (6.0—26.0)						
5v	4.5 (1.2—16.7)	5.1 (2.5—10.8)						
Sucralfate	156 (45.5—535)	55.7 (8.6—363)						
Cimetidine	18.8 (5.1—69.1)	$> 320^{c}$						
Ranitidine	1.5 (0.18—11.6)	29.6 (14.7—71.1)						

a) No dose-related response was observed. b) Inhibitory rate: 84.2% at $3.2\,\text{mg/kg}$, 85.0% at $10\,\text{mg/kg}$, 95.4% at $32\,\text{mg/kg}$. c) Inhibitory rate: 40.3% at $100\,\text{mg/kg}$, 23.7% at $320\,\text{mg/kg}$.

melting point apparatus and are uncorrected. Infrared (IR) spectra were taken with a Hitachi 260-10 spectrometer. Proton nuclear magnetic resonance (¹H-NMR) spectra were recorded with a JNM-PMX 60 or a Varian EM-390 spectrometer using tetramethylsilane as an internal standard. Mass spectral measurements (MS) were made on a Hitachi M-80 or a JEOL JMS D-300 mass spectrometer.

3'-(3-Ethylureido)-4'-hydroxypropiophenone (8b) A solution of 3'-amino-4'-hydroxypropiophenone (7) (16.5 g, 100 mmol) and ethyl isocyanate (8.5 g, 120 mmol) in EtOH (80 ml)-THF (100 ml) was refluxed for 8 h with stirring. After the mixture was cooled, the resulting precipitate was isolated by filtration to afford 8b (8.6 g, 36%). The filtrate was concentrated to a half volume in vacuo and the precipitate formed was collected by filtration to recover a second crop of 8b (9.1 g, 39%). An analytical sample was obtained by recrystallization from MeOH-THF, mp 194—195°C. Anal. Calcd for $C_{12}H_{16}N_2O_3$: C, 61.00; H, 6.83; N, 11.86. Found: C, 61.39; H, 7.08; N, 11.79. IR (Nujol): 3400, 1660 cm⁻¹. ¹H-NMR (DMSO- d_6) δ : 1.10 (6H, t, J=7 Hz), 2.90 (2H, q, J=7 Hz), 2.83—3.47 (2H, m), 6.83 (1H, t, J=5 Hz), 6.87 (1H, d, J=8 Hz), 7.45 (1H, dd, J=2, 8 Hz), 7.93 (1H, s), 8.58 (1H, d, J=2 Hz), 10.70 (1H, s). MS m/z: 236 (M⁺).

2-Amino-5-propionylbenzoxazole (9a) A solution of 7 (12.4 g, 75 mmol) and BrCN (8.8 g, 83 mmol) in EtOH (150 ml) was stirred for 2 h at room temperature. The resulting precipitate was collected by filtration and dissolved in $\rm H_2O$ (120 ml). After being adjusted to pH 8 with 20% aqueous $\rm K_2CO_3$, the aqueous solution was extracted with AcOEt. The extract was washed $\rm H_2O$, dried over MgSO₄ and evaporated *in vacuo*. The residue was triturated with Et₂O to afford **9a** (5.7 g, 40%). An analytical sample was obtained by recrystallization from MeOH. *Anal.* Calcd for $\rm C_{10}H_{10}N_2O_2$: C, 63.15; H, 5.30; N, 14.73. Found: C, 63.18; H, 5.30; N, 14.69. IR (Nujol): 3375, 3300, 1690, 1670 cm⁻¹. MS m/z: 190 (M⁺).

2-Ethylamino-5-propionylbenzoxazole (9c) A mixture of **8b** (8.4 g, 36 mmol) and PPE (67 g) was stirred at 120 °C for 30 min. The reaction mixture was poured into H_2O (70 ml)–AcOEt (70 ml), and the aqueous layer was separated. The aqueous solution was adjusted to pH 8 with 20% aqueous K_2CO_3 and extracted with AcOEt. The extract was washed with H_2O , dried over MgSO₄ and evaporated *in vacuo*. The residue was purified by silica gel column chromatography eluting with AcOEt–diisopropyl ether (IPE) (2:8) to afford **9c** (3.6 g, 44%). An analytical sample was obtained by recrystallization from MeCN. *Anal.* Calcd for $C_{12}H_{14}N_2O_2$: C, 66.04; H, 6.47; N, 12.84. Found: C, 66.43; H, 6.48; C0, 12.87. IR (Nujol): 3350, 1655 cm⁻¹. MS C0 m/z: 218 (C0.

5-(2-Bromopropionyl)-2-ethylaminobenzoxazole (10c) A solution of 9c (1.6 g, 7.3 mmol), pyridinium bromide perbromide (2.8 g, 8.8 mmol) and 30% HBr–AcOH (5 ml) in AcOH (20 ml) was stirred at room temperature for 1 h. After the mixture was poured into H_2O , the solution was adjusted to pH 8 with 20% aqueous K_2CO_3 and extracted with AcOEt. The extract was washed with H_2O , dried over MgSO₄ and evaporated in vacuo. The residue was recrystallized from AcOEt–IPE to afford 10c (1.8 g, 83%). Anal. Calcd for $C_{12}H_{13}BrN_2O_2$: C, 48.50; H, 4.41; N, 9.43. Found: C, 48.56; H, 4.42; N, 9.32. IR (Nujol): 1700, 1675 cm⁻¹. MS m/z: 298 (M^+-1).

2-Ethylamino-5-(3-methylimidazo[1,2-a]pyridin-2-yl)benzoxazole (4b)

TABLE V. Physical Data of 2-Substituted Amino-5-acylbenzoxazoles (9 and 10)

Compd. No.	X	R³	Yield (%)	mp (°C)	¹ H-NMR (DMSO- d_6 , δ : J =Hz)
9a	Н	Н	40	225—227	1.09 (3H, t, J=7), 3.05 (2H, q, J=7), 7.43 (1H, d, J=8), 7.63 (2H, s), 7.67 (1H, dd, J=2, 8), 7.78 (1H, d, J=2)
9b	Н	CH ₃	45	153—155	1.10 (3H, t, $J=7$), 2.95 (3H, d, $J=5$), 3.05 (2H, q, $J=7$), 7.43 (1H, d, $J=9$), 7.72 (1H, dd, $J=2$), 9), 7.85 (1H, d, $J=2$), 8.00 (1H, q, $J=5$)
9c	Н	C_2H_5	44	130—132	1.17 (3H, t, $J=7$), 1.25 (3H, t, $J=7$), 3.07 (2H, q, $J=7$), 3.13—3.67 (2H, m), 7.43 (1H, d, $J=9$), 7.70 (1H, dd, $J=2$, 9), 7.85 (1H, d, $J=2$), 8.10 (1H, t, $J=5$)
10a	Br	Н	90	138140	1.83 (3H, d, $J=7$), 5.87 (1H, q, $J=7$), 7.33—8.50 (5H, m)
10b	Br	CH ₃	67	131—133	1.78 (3H, d, $J = 7$), 2.93 (3H, d, $J = 5$), 5.87 (1H, q, $J = 7$), 7.50 (1H, d, $J = 9$), 7.82 (1H, dd, $J = 2$, 9), 7.97 (1H, d, $J = 2$), 8.07 (1H, q, $J = 5$)
10c	Br	C_2H_5	83	142—144	1.22 (3H, t, $J=7$), 1.79 (3H, t, $J=7$), 3.18—3.53 (2H, m), 5.84 (1H, q, $J=7$), 7.44 (1H, d, $J=9$), 7.73 (1H, dd, $J=2$, 9), 7.90 (1H, d, $J=2$), 8.13 (1H, t, $J=5$)

TABLE VI. Physical Data of 2-[3-(3-Alkylureido)-4-hydroxyphenyl]imidazo[1,2-a]pyridines (14)

Compd.	\mathbb{R}^1	R ²	R³	Yield	(90)	IR (Nujol)		¹H-NHR
No.	K.	K*	K ^o	(%)	mp (°C)	$cm^{-1} (C=O)$	Solvent	δ (<i>J</i> = Hz)
14a	6-CH ₃	Н	C ₂ H ₅	72	> 300	1630	CF₃COOH	1.35 (3H, t, <i>J</i> = 7), 2.57 (3H, s), 3.53 (2H, q, <i>J</i> = 7),
14b	7-CH ₃	Н	C_2H_5	84	> 300	1630	CF ₃ COOH	7.23 (1H, d, J =9), 7.64 (1H, d, J =9), 7.80 (2H, s), 8.03 (2H, s), 8.35 (1H, s) 1.37 (3H, t, J =7), 2.70 (3H, s), 3.57 (2H, q, J =7), 7.07—7.67 (4H, m), 8.00—8.07 (2H, m), 8.47 (1H, d, J =7)
14c	7-CH ₃	CH ₃	C_2H_5	77	> 300	1660	CF ₃ COOH	1.37 (3H, t, <i>J</i> = 7), 2.72 (3H, s), 2.75 (3H, s), 3.57 (2H, q, <i>J</i> = 7), 7.20—8.23 (5H, m), 8.35 (1H, d, <i>J</i> = 7)
14d	7-CH ₃	CH ₃	<i>n</i> -C ₃ H ₇	91	233—236	1650	DMSO-d ₆	0.90 (3H, t, <i>J</i> = 7), 1.40—1.75 (2H, m), 2.36 (3H, s), 2.63 (3H, s), 3.21 (2H, t, <i>J</i> = 7), 6.55—6.95 (2H, m), 7.05—7.31 (2H, m), 7.87 (1H, s), 8.10 (1H, d, <i>J</i> = 7), 8.30 (1H, d, <i>J</i> = 2)
14e	7-CH ₃	CH ₃	iso-C ₃ H ₇	78	231	1640	DMSO-d ₆	1.12 (6H, d, <i>J</i> =7), 2.36 (3H, s), 2.59 (3H, s), 3.47—3.86 (1H, m), 6.60—6.94 (2H, m), 7.08— 7.35 (2H, m), 7.83 (1H, s), 8.13 (1H, d, <i>J</i> =6), 8.35 (1H, d, <i>J</i> =2), 9.94 (1H, br s)
14f	7-CH ₃	CH ₃	$CH_2CH = CH_2$	91	226—228	1630	DMSO-d ₆	2.37 (3H, s), 2.61 (3H, s), 3.76 (2H, t, <i>J</i> =5), 4.94—5.39 (2H, m), 5.59—6.18 (1H, m), 6.73 (1H, d, <i>J</i> =7), 6.86 (1H, d, <i>J</i> =8), 7.21 (1H, dd, <i>J</i> =2, 8), 7.27 (1H, s), 8.07 (1H, s), 8.14 (1H, d, <i>J</i> =7), 8.40 (1H, d, <i>J</i> =2)
14g	7-CH ₃	CH ₃	c-C ₆ H ₁₁	93	215—218	1655	DMSO-d ₆	1.08—1.79 (10H, m), 2.36 (3H, s), 2.58 (3H, s), 3.41—3.56 (1H, m), 6.76 (1H, dd, J =2, 7), 6.82 (1H, d, J =6), 6.85 (1H, d, J =8)7.20 (1H, dd, J =2, 8), 7.28 (1H, d, J =2), 7.89 (1H, s), 8.16 (1H, d, J =7), 8.40 (1H, d, J =2), 10.02 (1H, br s)
14h	7-CH ₃	СНО	C_2H_5	79	213—214	1635	DMSO-d ₆	(111, 41, 47 - 7), $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 41, 47 - 2)$, $(111, 4$
14i	$7-C_2H_5$	Н	C_2H_5	59	168—169	1640	CF ₃ COOH	1.38 (3H, t, $J=7$), 1.48 (3H, t, $J=7$), 3.00 (2H, q, $J=7$), 3.57 (2H, q, $J=7$), 7.20—7.77 (3H, m),
14j	7-iso-C ₃ H ₇	Н	C_2H_5	55	233	1635	DMSO-d ₆	7.73 (1H, s), 7.90—8.10 (2H, m), 8.46 (1H, d, J =7) 1.11 (3H, t, J =7), 1.28 (6H, d, J =7), 2.80—3.50 (3H, m), 6.70—7.00 (3H, m), 7.32 (1H, d, J =2), 7.39 (1H, dd, J =2, 7), 8.02 (1H, d, J =7), 8.06 (1H, s), 8.40 (1H, d, J =7), 8.51 (1H, d, J =2)
14k	8-CH ₃	Н	C_2H_5	71	216—218	1630	CF ₃ COOH	1.36 (3H, t, J =7), 2.78 (3H, s), 3.54 (2H, q, J =7), 7.24 (1H, d, J =8), 7.39—7.96 (3H, m), 8.02—8.23 (2H, m), 8.44 (1H, d, J =7)

TABLE VII. Physical Data of 2-(3-Amino-4-hydroxyphenyl)-5,6,7,8-tetrahydroimidazo[1,2-a]pyridines (22)

Compd. No.	R ¹	R ²	Yield (%)	mp (°C)	¹ H-NMR (DMSO- d_6 , δ : J =Hz)
22a	Н	Н	72	182—187	1.58—2.05 (4H, m), 2.54—2.85 (2H, m), 3.73—3.96 (2H, m), 4.43 (2H, br s), 6.56 (1H, d, $J=8$), 6.74 (1H, dd, $J=2$, 8), 6.97 (1H, d, $J=2$), 7.04 (1H, s)
22b	7-CH ₃	Н	64	252—257	1.06 (3H, dd, $J=2$, 8), 6.97 (1H, d, $J=2$), 7.04 (1H, 8) 1.06 (3H, d, $J=6$), 1.47—2.41 (4H, m), 2.67—3.11 (1H, m), 3.75—4.33 (2H, m), 6.57 (1H, d, $J=8$), 6.79 (1H, dd, $J=2$, 8), 7.04 (1H, s), 7.08 (1H, d, $J=2$)
22c	7-CH ₃	CH ₃	82	238—242	1.07 (3H, d, $J = 2$, 3), 7.04 (11, 3), 7.06 (11, d, $J = 2$) 1.07 (3H, d, $J = 7$), 1.37—2.40 (4H, m), 2.26 (3H, s), 2.64—3.07 (1H, m), 3.56—4.04 (2H, m), 6.64 (2H, s), 6.91 (1H, s)

Table VIII. Physical Data of 2-[3-(3-Alkylureido)-4-hydroxyphenyl]-5,6,7,8-tetrahydroimidazo[1,2-a]pyridines (23)

Compd.	\mathbb{R}^1	R ²	\mathbb{R}^3	Yield	mp (°C)	IR (Nujol)		¹ H-NMR
No.	K	K	K	(%)	mp (C)	$cm^{-1}(C=O)$	Solvent	$\delta~(J\!=\!{ m Hz})$
23a	Н	Н	C ₂ H ₅	92	197—199	1660	DMSO-d ₆	1.06 (3H, t, <i>J</i> =7), 1.63—2.10 (4H, m), 2.36—2.86 (2H, m), 2.94—3.45 (2H, m), 3.78—4.10 (2H, m), 6.60—6.93 (1H, m), 6.70 (1H, d, <i>J</i> =8), 7.07 (1H, dd, <i>J</i> =2, 8),
23b	7-CH ₃	Н	CH ₃	100	226—228	1630	CF ₃ COOH	7.10 (1H, s), 7.83 (1H, s), 8.13 (1H, s) 1.91 (3H, d, <i>J</i> = 5), 1.77—2.65 (4H, m), 2.70—3.00 (1H, m), 3.07 (3H, s), 4.13—4.51 (2H, m), 7.17 (1H, d, <i>J</i> = 8), 7.35 (1H, s), 7.46 (1H, dd, <i>J</i> = 2, 8), 7.70 (1H, d, <i>J</i> = 2)
23c	7-CH ₃	Н	C_2H_5	97	209—211	1680	DMSO-d ₆	1.07 (3H, t, J =7), 1.07 (3H, d, J =6), 1.46—2.60 (4H, m), 2.67—2.98 (1H, m), 2.94—3.41 (2H, m), 3.73—4.26 (2H, m), 6.56—6.86 (1H, m), 6.69 (1H, d, J =8), 7.0 (1H, dd, J =2, 8), 7.10 (1H, s), 7.81 (1H, s), 8.11 (1H,
23d	7-CH ₃	Н	<i>n</i> -C ₃ H ₇	85	224—226	1640	DMSO-d ₆	d, <i>J</i> = 2), 9.73 (1H, s) 0.92 (3H, t, <i>J</i> = 7), 1.08 (3H, d, <i>J</i> = 5), 1.25—2.40 (6H, m), 2.66—3.42 (3H, m), 3.72—4.38 (2H, m), 6.55—6.90 (1H, m), 6.72 (1H, d, <i>J</i> = 8), 7.09 (1H, dd, <i>J</i> = 2, 8), 7.13 (1H, s), 7.86 (1H, s), 8.17 (1H, d, <i>J</i> = 2), 9.75 (1H, br s)
23e	7-CH ₃	Н	iso-C ₃ H ₇	96	218—220	1670	DMSO-d ₆	0.78—1.34 (9H, m), 1.55—2.66 (4H, m), 2.67—3.11 (1H, m), 3.57—4.30 (3H, m), 6.52—6.86 (2H, m), 7.08 (1H, dd, <i>J</i> =2, 8), 7.14 (1H, s), 7.75 (1H, s), 8.18 (1H, d, <i>J</i> =2), 9.73 (1H, br s)
23f	7-CH ₃	Н	<i>n</i> -C ₄ H ₉	95	221—223	1640	CF₃COOH	1.10 (3H, t, <i>J</i> =7), 1.17—2.60 (8H, m), 1.31 (3H, d, <i>J</i> =5), 2.67—2.97 (1H, s), 3.30—3.67 (2H, m), 3.90—4.50 (2H, m), 7.16 (1H, d, <i>J</i> =9), 7.35 (1H, s),
23g	7-CH ₃	Н	$CH_2CH = CH_2$	97	221—223	1640	DMSO-d ₆	7.45 (1H, dd, J =2, 9), 7.79 (1H, d, J =2) 1.06 (3H, d, J =6), 1.38—2.40 (4H, m), 2.70—3.06 (1H, m), 3.51—4.14 (4H, m), 4.96—5.37 (2H, m), 5.65—6.17 (1H, m), 6.73 (1H, d, J =8), 6.88 (1H, t, J =6), 7.10 (1H, dd, J =2, 8), 7.12 (1H, s), 8.95 (1H, s), 8.21 (1H, d, J =2), 9.74 (1H, s)
23h	7-CH ₃	CH ₃	CH ₃	95	207—208	1640	CF ₃ COOH	1.31 (3H, d, J = 5), 1.68—2.63 (4H, m), 2.47 (3H, s), 2.63—2.91 (1H, m), 3.08 (3H, s), 3.82—4.38 (2H, m) 7.20 (1H, d, J = 8), 7.38 (1H, dd, J = 2, 8), 7.60 (1H, d, J = 2)
23i	7-CH ₃	CH ₃	C ₂ H ₅	87	225—226	1635	DMSO-d ₆	1.06 (3H, t, <i>J</i> = 7), 1.07 (3H, d, <i>J</i> = 6), 1.49—2.41 (4H, m), 2.26 (3H, s), 2.65—2.93 (1H, m), 2.93—3.44 (2H m), 3.57—3.94 (2H, m), 6.56—6.88 (2H, m), 6.98 (1H, d, <i>J</i> = 2, 8), 7.83 (1H, s), 8.04 (1H, d, <i>J</i> = 2), 9.76 (1H, br s)
23j	7-CH ₃	CH ₃	iso-C ₃ H ₇	78	204—206	1655	CDCl ₃ +CD ₃ OD	1.17 (3H, d, <i>J</i> = 6), 1.20 (6H, d, <i>J</i> = 6), 1.56—2.70 (4H, m), 2.32 (3H, s), 2.86—3.16 (1H, m), 3.56—4.16 (3H, m), 6.92 (1H, d, <i>J</i> = 8), 7.15 (1H, dd, <i>J</i> = 2, 8), 7.37 (1H,
23k	7-CH ₃	CH ₃	<i>n</i> -C ₄ H ₉	59	209—211	1680	CF ₃ COOH	d, $J=2$) 1.00 (3H, t, $J=6$), 1.18—2.56 (8H, m), 1.31 (3H, d, $J=5$), 2.44 (3H, s), 2.63—2.93 (1H, m), 3.33—3.71 (2H m), 3.90—4.39 (2H, m), 7.21 (1H, d, $J=8$), 7.63 (1H s)
231	7-CH ₃	CH ₃	$CH_2CH = CH_2$	57	210—212	1630	CDCl ₃ +CD ₃ OD	d, <i>J</i> = 8), 7.63 (1H, s) 1.15 (3H, d, <i>J</i> = 5), 1.46—2.71 (4H, m), 2.30 (3H, s), 2.87—3.17 (1H, s), 3.48—4.20 (4H, m), 5.04—5.40 (2H m), 5.88—6.15 (1H, m), 6.90 (1H, d, <i>J</i> = 9), 7.14 (1H dd, <i>J</i> = 2, 9), 7.46 (1H, d, <i>J</i> = 2)

Method A: A solution of **10c** (1.9 g, 6.4 mmol) and 2-aminopyridine (1.8 g, 19 mmol) in MeCN (40 ml) was refluxed for 7 h with stirring. The resulting precipitate was collected by filtration and dissolved in AcOEt (20 ml)–5% HCl (20 ml). The aqueous layer was separated out, adjusted to pH 8 with 20% aqueous K_2CO_3 and extracted with CHCl₃. The extract was washed with H_2O , dried over MgSO₄ and evaporated *in vacuo*. The residue was recrystallized from CH_2Cl_2 – Et_2O to afford **4b** (0.45 g, 24%). IR (Nujol): $1675 \, \mathrm{cm}^{-1}$. 1H -NMR (DMSO- d_6) δ : 1.20 (3H, t, J=7 Hz), 2.63 (3H, s), 3.17—3.53 (2H, m), 6.90 (1H, dt, J=2, 7 Hz), 7.20—7.63 (5H, m), 7.90 (1H, t, J=5 Hz), 8.25 (1H, dd, J=2, 7 Hz).

2-Bromo-3'-(3-ethylureido)-4'-hydroxyacetophenone (16) Pyridinium bromide perbromide (114 g, 356 mmol) was added in portions to a solution of 3'-(3-ethylureido)-4'-hydroxyacetophenone (**15**) (72 g, 320 mmol) and 25% HBr-AcOH (140 ml) in AcOH (580 ml). After being stirred at room temperature for 4h, the mixture was poured into H₂O. The resulting precipitate was collected by filtration and washed with water to afford **16** (69 g, 70%), which was not recrystallized because of its relative instability, mp 180—181 °C. *Anal.* Calcd for $C_{11}H_{13}BrN_2O_3$: C, 43.87; H, 4.35; N, 9.30. Found: C, 43.65; H, 4.22; N, 9.04. IR (Nujol): 3400, 1680, 1650 cm⁻¹. ¹H-NMR (DMSO- d_6) δ : 1.05 (3H, t,

 $J=7 \,\mathrm{Hz}$), 3.11 (2H, q, $J=7 \,\mathrm{Hz}$), 4.72 (2H, s), 6.89 (1H, d, $J=8 \,\mathrm{Hz}$), 7.51 (1H, dd, J=2, 8 Hz), 8.00 (1H, br s), 8.64 (1H, d, $J=2 \,\mathrm{Hz}$).

2-[3-(3-Ethylureido)-4-hydroxyphenyl]-6-methylimidazo[1,2-a]pyridine (14a) A mixture of 16 (10.0 g, 33 mmol) and 2-amino-5-methylpyridine (9.0 g, 83 mmol) in EtOH (100 ml) was refluxed for 3.5 h with stirring. After cooling in an ice bath, the resulting precipitate was collected by filtration and added to H_2O (100 ml). The pH of the mixture was adjusted to 8 with 10% aqueous K_2CO_3 and the product separated out was collected by filtration to afford 14a (7.4 g, 72%). An analytical sample was obtained by recrystallization from IPE–MeOH–THF. Anal. Calcd for $C_{17}H_{18}N_4O_2 \cdot 4/5H_2O$: C, 62.82; H, 6.07; N, 17.24. Found: C, 62.74; H, 6.02; N, 16.97. IR (Nujol): 3300, 1630 cm⁻¹. MS m/z: 310 (M⁺).

2-[3-(3-Ethylureido)-4-hydroxyphenyl]-7-methylimidazo[1,2-a]pyridine (14b) A mixture of 2-(3-amino-4-hydroxyphenyl)-7-methylimidazo[1,2-a]pyridine (13b)¹⁾ (12.8 g, 49 mmol) and ethyl isocyanate (4.2 g, 59 mmol) in MeOH (50 ml)-THF (100 ml) was refluxed with stirring for 3 h. After the reaction mixture was cooled, the resulting precipitate was collected by filtration and washed with MeOH to afford 14b (12.8 g, 84%). An analytical sample was obtained by recrystallization from IPE-MeOH-THF. Anal. Calcd for $C_{17}H_{18}N_4O_2$: C, 65.79; H, 5.85; N, 18.05. Found: C, 65.76; H, 5.93; N, 17.80. IR (Nujol): 3300, 1630 cm⁻¹. MS m/z: 310 (M⁺).

2-Ethylamino-5-(7-methylimidazo[1,2-a]pyridin-2-yl)benzoxazole (4e) Methods B₁ and B₂: A mixture of **14b** (10.5 g, 34 mmol) and PPE (100 g) was stirred at 120 °C for 1.5 h and the reaction mixture was poured into H₂O (100 ml). The solution was washed with AcOEt (100 ml), adjusted to pH 8 with 20% aqueous K_2CO_3 and extracted with AcOEt. The extract was washed with H₂O, dried over MgSO₄ and evaporated *in vacuo*. The residue was subjected to silica gel chromatography by eluting with CHCl₃–MeOH (20:1), and the product obtained was recrystallized from AcOEt–MeOH–THF to afford **4e** (3.7 g, 38%). IR (Nujol): 1695, 1645 cm⁻¹. ¹H-NMR (DMSO- d_6) δ : 1.22 (3H, t, J=7 Hz), 2.36 (3H, s), 3.18—3.55 (2H, m), 6.72 (1H, dd, J=2, 7 Hz), 7.32 (1H, d, J=8 Hz), 7.34 (1H, d, J=2 Hz), 7.57 (1H, dd, J=2, 8 Hz), 7.78 (1H, d, J=2 Hz), 7.90 (1H, t, J=5 Hz), 8.23 (1H, s), 8.34 (1H, d, J=7 Hz).

2-(4-Benzyloxy-3-nitrophenyl)-7-methylimidazo[1,2-a]pyridine (19) This compound was prepared from 4'-benzyloxy-3'-nitro-2-bromoacetophenone (**18**)⁵⁾ in the similar manner as described for **4b** from **10c**. Yield 65%, mp 162—164 °C (DMF–H₂O). *Anal*. Calcd for $C_{21}H_{17}N_3O_3$: C, 70.58; H, 4.77; N, 11.69. Found: C, 70.21; H, 4.92; N, 11.54. IR (Nujol): 1640, 1620, 1525, 1350 cm⁻¹. ¹H-NMR (DMSO- d_6) δ : 2.35 (3H, s), 5.35 (2H, s), 6.75 (1H, dd, J=2, 7 Hz), 7.35—7.53 (7H, m), 8.18 (1H, s), 8.37 (1H, s), 8.41 (1H, d, J=7 Hz), 8.43 (1H, d, J=2 Hz). MS m/z: 359 (M⁺).

2-(4-Benzyloxy-3-nitrophenyl)-3-formyl-7-methylimidazo[1,2-a]pyridine (20) POCl₃ (24.7 g, 161 mmol) was added dropwise to a solution of 19 (29 g, 81 mmol) in DMF (300 ml) at room temperature with stirring. After being stirred for 8 h at 65 °C, the mixture was poured into H₂O (1.2 l) and adjusted to pH 8 with 20% aqueous K_2CO_3 . The resulting precipitate was collected by filtration and washed with H₂O to afford 20 (30 g, 95%). An analytical sample was obtained by recrystallization from DMF-H₂O, mp 162—164 °C. *Anal.* Calcd for $C_{22}H_{17}N_3O_4$: C, 68.21; H, 4.42; N, 10.85. Found: C, 68.05; H, 4.51; N, 10.73. IR (Nujol): 1640, 1535, 1360 cm⁻¹. ¹H-NMR (DMSO- d_6) δ : 2.47 (3H, s), 5.41 (2H, s), 7.18 (1H, dd, J=2, 7Hz), 7.33—7.68 (8H, m), 8.18 (1H, d, J=2 Hz), 9.40 (1H, d, J=7 Hz), 9.99 (1H, s).

2-(3-Amino-4-hydroxyphenyl)-3-formyl-7-methylimidazo[1,2-a]pyridine (21) A suspension of 20 (10.0 g, 26 mmol) in MeOH (200 ml)-THF (200 ml) was hydrogenated under atmospheric pressure of $\rm H_2$ over 10% Pd-C (1.5 g) at room temperature. After the catalyst was filtered off, the filtrate was evaporated *in vacuo*. The residue was triturated with Et₂O to afford 21 (3.2 g, 47%). An analytical sample was obtained by recrystallization from MeOH-IPE, mp 265—270 °C. *Anal.* Calcd for $\rm C_{15}H_{13}N_3O_2$: C, 67.40; H, 4.90; N, 15.72. Found: C, 67.51; H, 4.86; N, 15.68. IR (Nujol): 3450, 3360, 1630 cm⁻¹. 11 H-NMR (DMSO- 11) 12 : 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 11 $^$

2-[3-(3-Ethylureido)-4-hydroxyphenyl]-3-formyl-7-methylimidazo[1,2-a]pyridine (14h) This compound was prepared from **21** in the similar manner described for **14b** from **13b**. An analytical sample was obtained by recrystallization from DMF-H₂O. *Anal*. Calcd for $C_{18}H_{18}N_4O_3$: C, 63.89; H, 5.36; N, 16.56. Found: C, 63.60; H, 5.40; N, 16.32. IR (Nujol): 3330, 1635 cm⁻¹. MS m/z: 338 (M⁺).

2-Ethylamino-5-(3-formyl-7-methylimidazo[1,2-a]pyridin-2-yl)benzox-azole (4n) Method B₃: A mixture of 14h (2.9 g, 8.6 mmol) and PPE

(20 g) was stirred for 1 h at 120 °C. The reaction mixture was poured into $\rm H_2O$ (30 ml), adjusted to pH 8 with 20% aqueous $\rm K_2CO_3$ and extracted with AcOEt–THF. The extract was washed with $\rm H_2O$, dried over MgSO₄ and evaporated *in vacuo*. The residue was chromatographed on alumina by eluting with CHCl₃–AcOEt (4:1), and the product obtained was recrystallized from IPE–MeOH–THF to afford $\rm 4n$ (1.0 g, 36%). IR (Nujol): 3370, 1700, 1640 cm⁻¹. ¹H-NMR (DMSO- $\rm d_6$) $\rm \delta$: 1.23 (3H, t, $\rm J=7Hz$), 2.47 (3H, s), 3.21–3.55 (2H, m), 7.14 (1H, dd, $\rm J=2$, 7Hz), 7.47 (2H, s), 7.66 (1H, s), 7.68 (1H, d, $\rm J=2Hz$), 8.03 (1H, t, $\rm J=5Hz$), 9.40 (1H, d, $\rm J=7Hz$), 9.93 (1H, s).

5-(3-Bromo-7-methylimidazo[1,2-a]pyridin-2-yl)-2-ethylaminobenzoxazole (4m) Method C: Pyridinium bromide perbromide (1.3 g, 4.1 mmol) was added in portions to a solution of **4e** (1.0 g, 3.4 mmol) and 25% HBr–AcOH (0.2 ml) in AcOH (10 ml) at room temperature with stirring. After being stirred for 1 h, the mixture was poured into H_2O (100 ml). The solution was adjusted to pH 8 with 20% aqueous K_2CO_3 and extracted with AcOEt. The extract was washed with H_2O , dried over MgSO₄ and evaporated *in vacuo*. The residue was recrystallized from IPE–MeOH–THF to afford **4m** (1.0 g, 79%). IR (Nujol): $1675 \, \text{cm}^{-1}$. 1^4 H-NMR (DMSO- d_6) δ : 1.23 (3H, t, J=7 Hz), 2.40 (3H, s), 3.20—3.56 (2H, m), 6.91 (1H, dd, J=2, 7Hz), 7.41 (1H, d, J=8 Hz), 7.95 (1H, t, J=2 Hz), 7.71 (1H, dd, J=2, 8 Hz), 7.90 (1H, d, J=2 Hz), 7.95 (1H, t, J=5 Hz), 8.23 (1H, d, J=7 Hz).

5-(3,7-Dimethyl-5,6,7,8-tetrahydroimidazo[1,2-a]pyridin-2-yl)-2-n-propylaminobenzoxazole (50) Method D₁: A mixture of **4k** (0.65 g, 2.0 mmol) in EtOH (200 ml) was hydrogenated over 10% Pd–C (0.1 g) under 3—5 atmospheric pressure of H₂ for 3 h at 60—70 °C. The catalyst was removed by filtration and the filtrate was evaporated *in vacuo*. The residue was recrystallized from AcOEt–THF to afford **5o** (0.41 g, 62%). IR (Nujol): 1670 cm⁻¹. ¹H-NMR (DMSO- d_6) δ : 0.93 (3H, t, J=7 Hz), 1.07 (3H, d, J=6 Hz), 1.31—2.39 (6H, m), 2.29 (3H, s), 2.66—3.01 (1H, m), 3.03—3.42 (2H, m), 3.45—4.08 (2H, m), 7.16 (1H, d, J=8 Hz), 7.32 (1H, dd, J=2, 8 Hz), 7.48 (1H, d, J=2 Hz), 7.83 (1H, t, J=6 Hz).

5-(3,7-Dimethyl-5,6,7,8-tetrahydroimidazo[1,2-a]pyridin-2-yl)-2-methyl-aminobenzoxazole (5m) Method D_2 : A mixture of 4g (1.4 g, 4.8 mmol) in MeOH (100 ml)—THF (40 ml)—AcOH (10 ml) was hydrogenated over PtO₂ (0.15 g) under 3—3.5 atmospheric pressure of H_2 for 8 h at room temperature. The catalyst was filtered off and the filtrate was concentrated to dryness. The residue was dissolved in H_2 O. The solution was adjusted to pH 8 with 20% aqueous K_2 CO₃ and extracted with AcOEt. The extract was washed with H_2 O, dried over MgSO₄ and evaporated in vacuo. The residue was recrystallized from AcOEt to afford 5m (0.97 g, 68%). IR (Nujol): $1660 \, \text{cm}^{-1}$. $^{1}\text{H-NMR}$ (DMSO- d_6) δ : 1.15 (3H, d, J=7 Hz), 1.47—2.63 (4H, m), 2.34 (3H, s), 2.90—3.23 (1H, m), 3.08 (3H, s), 3.50—4.16 (2H, m), 5.70 (1H, br s), 7.18 (1H, d, J=8 Hz), 7.38 (1H, dd, J=2, 8 Hz), 7.51 (1H, d, J=2 Hz).

2-(3-Amino-4-hydroxyphenyl)-7-methyl-5,6,7,8-tetrahydroimidazo[1,2-*a***]pyridine (22b)** A solution of 2-(4-hydroxy-3-nitrophenyl)-7-methylimidazo[1,2-*a*]pyridine (**12c**) (26 g, 97 mmol) in EtOH (500 ml) was hydrogenated over 10% Pd–C (2.5 g) under 3—5 atmospheric pressure of $\rm H_2$ for 3 h at 60—70 °C. The catalyst was removed by filtration and the filtrate was evaporated *in vacuo*. The residue was recrystallized from EtOH to afford **22b** (15 g, 64%). *Anal*. Calcd for $\rm C_{14}\rm H_{17}N_3O\cdot 1/5H_2O$: C, 68.10; H, 7.10; N, 17.02. Found: C, 68.24; N, 7.19; N, 16.83. IR (Nujol): $1610\,\rm cm^{-1}$. $^1\rm H-NMR$ (DMSO- d_6) δ : 1.06 (3H, d, J=6 Hz), 1.47—2.41 (4H, m), 3.75—4.33 (2H, m), 6.57 (1H, d, J=8 Hz), 6.79 (1H, dd, J=2, 8 Hz), 7.04 (1H, d, J=2 Hz), 7.08 (1H, s). MS m/z: 243 (M $^+$).

2-[4-Hydroxy-3-(3-methylureido)phenyl]-7-methyl-5,6,7,8-tetrahydroimidazo[1,2-a]pyridine (23b) A mixture of **22b** (2.6 g, 11 mmol) and methyl isocyanate (0.8 ml, 14 mmol) in MeOH (13 ml)–THF (26 ml) was stirred for 2 h at room temperature. The reaction mixture was concentrated *in vacuo* to about half volume and the resulting precipitate was collected by filtration to afford **23b** (3.2 g, 100%). An analytical sample was obtained by recrystallization from DMF–H₂O. *Anal.* Calcd for $C_{16}H_{20}N_4O_2 \cdot 3/4H_2O$: C, 61.23; H, 6.90; N, 17.85. Found: C, 61.38; H, 7.06; N, 18.03. IR (Nujol): 3320, 1630 cm⁻¹. MS m/z: 300 (M⁺).

5-(7-Methyl-5,6,7,8-tetrahydroimidazo[1,2-a]pyridin-2-yl)-2-methylaminobenzoxazole (5f) Method E: A mixture of 23b (3.0 g, 10 mmol) and PPE (30 g) was stirred at 120 °C for 30 min. The reaction mixture was dissolved in AcOEt (15 ml)-THF (15 ml)-H₂O (30 ml), and the resulting solution was adjusted to pH 8 with 20% aqueous K₂CO₃. The separated organic layer was washed with H₂O, dried over MgSO₄ and evaporated *in vacuo*. The residue was chromatographed on alumina by eluting with CHCl₃-MeOH (39:1) to give crystals. Recrystallization from MeCN to afford 5f (0.67 g, 24%). IR (Nujol): 1670, 1645 cm⁻¹.

¹H-NMR (DMSO- d_6) δ: 1.17 (3H, d, J=6 Hz), 1.40—2.39 (4H, m), 2.70—3.04 (1H, m), 2.90 (3H, d, J=5 Hz), 3.65—4.19 (2H, m), 7.19 (1H, d, J=8 Hz), 7.33 (1H, s), 7.35 (1H, dd, J=2, 8 Hz), 7.51 (1H, d, J=2 Hz), 7.65 (1H, q, J=5 Hz).

2-Amino-5-(7-methyl-5,6,7,8-tetrahydroimidazo[1,2-a]pyridin-2-yl)benzoxazole (5e) Method F: A mixture of **22b** (3.0 g, 12 mmol) and BrCN (1.6 g, 15 mmol) in EtOH (30 ml) was stirred for 2 h at room temperature. The resulting precipitate was collected by filtration, and dissolved in AcOEt (30 ml)– $\rm H_2O$ (30 ml). The solution was adjusted to pH 8 with 20% aqueous $\rm K_2CO_3$. A solid separated was collected, washed with AcOEt– $\rm H_2O$ and recrystallized from EtOH–IPE to afford **5e** (1.8 g, 55%). IR (Nujol): 1655, 1625 cm⁻¹. $^{\rm 1}\rm H$ -NMR (DMSO- $^{\rm 4}\rm G_0$) $^{\rm 6}\rm H_2$: 1.24—2.44 (4H, m), 2.68—3.09 (1H, m), 3.68—4.25 (2H, m), 7.23 (1H, d, $^{\rm 2}\rm H_2$), 7.33 (2H, s), 7.42 (1H, dd, $^{\rm 2}\rm H_2$), 7.54 (1H, d, $^{\rm 2}\rm H_2$).

Biological Test Restraint and water-immersed stress ulcer (stress ulcer) and ethanol-induced gastric lesion (ethanol ulcer; cytoprotective activity) were evaluated by the methods described in the literature, $^{6.7)}$ and in our previous paper. $^{1)}$ The ED $_{50}$ values, the dose required for 50% inhibition of the ulcer index, were estimated according to the method of Litchfield and Wilcoxon. $^{8)}$

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