Saponins from Talinum triangulare

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Talinum triangulare WILLD. have been used as a traditional treatment for tonic in Indonesia. From roots of this plant, two saponins were isolated. One was identified as chikusetsusaponin IVa, a known saponin of oleanolic acid from rhizomes of *Panax* spp. The other was a new saponin, and was elucidated as β -D-glucopyranosyl methyl spergulagenate 3-O- β -D-glucoropyranoside.

Keywords Talinum triangulare; Portulacaceae; saponin; oleanane triterpene; methyl spergulagenate

In continuing our search for chemical constituents of Indonesian medicinal plants, we have studied *Talinum triangulare* WILLD. *Talinum triangulare* is known locally as Ginseng Bugis and is used as a traditional treatment for tonic in Indonesia.

The present paper deals with the isolation and structure elucidation of the saponins.

The dried roots (1.0 kg) of *Talinum triangulare* (Portulacaceae) were extracted with MeOH and the concentrated residue was separated with Et₂O and 1-BuOH, successively. The 1-BuOH extract (7.0 g) was subjected to column chromatography on silica gel by eluting with CHCl₃-MeOH-H₂O (30:12:1, 15:6:1, 10:6:1) to give two fractions A and B. Fraction B was separated by chromatography on silica gel (CHCl₃: MeOH: H₂O = 30:8:1, 15:6:1) followed by repeated column chromatography on a Lichroprep RP-18 (65% MeOH) affording a new saponin named talinumoside I (1) and one known saponin (2).

Talinumoside I (1) showed an $[M-H]^-$ ion at m/z 837 on the negative fast atom bombardment mass spectrum (FAB-MS), indicating its molecular weight to be 838. On acid hydrolysis, 1 afforded glucose and glucuronic acid, while on enzymatic hydrolysis with crude hesperidinase, 1 yielded an aglycone (3) which was identified as a known triterpene, methyl spergulagenate. 1) The 13C-nuclear magnetic resonance (13C-NMR) spectrum of 1 revealed the presence of an oxygenated carbon (δ 89.3), a pair of olefinic carbons (δ 123.5, 143.8), two ester carbons (δ 176.1, 177.0) and two anomeric carbons (δ 95.8, 106.6). Comparison of the ¹³C-NMR spectrum of 1 with that of 3 indicated the presence of glycosylation shifts^{2,3)} for the signals due to 2-, 3-, and 28-C of the aglycone moiety, disclosing that 1 must be a 3,28-bisdesmoside of 3, On saponification, 1 yielded a prosapogenin (4) and 1,6anhydroglucose, which is characteristic of β -glucosyl ester of di- and triterpenes.4) Compound 4 showed one anomeric proton signal at δ 6.28 (1H, d, $J=7.9\,\mathrm{Hz}$) and one anomeric carbon signal. The prosapogenin afforded

Chart 1

glucuronic acid on acid hydrolysis. Based on these results, 1 can be formulated as shown in Chart 1.

Saponin 2 was identified as chikusetsusaponin IVa which has been isolated from rhizomes of *Panax japonicus* C. A. MEYER and many other *Panax* species.⁵⁾

TABLE I. ¹³C-NMR Chemical Shifts in C₅D₅N

Carbon No.	1	2	3	4
1	38.7	38.7	38.7	38.7
2	26.3	26.3	28.1	26.3
3	89.3	89.2	78.1	89.3
4	39.5	39.5	39.5	39.5
5	55.9	55.9	55.9	55.9
6	18.5	18.5	18.8	18.5
7	33.2	32.6	33.2	33.2
8	39.9	39.9	39.9	39.9
9	48.0	48.0	48.0	48.0
10	37.0	37.0	37.0	37.0
11	23.8	23.4	23.8	23.8
12	123.5	123.5	123.5	123.5
13	143.8	144.1	143.8	143.8
14	42.0	42.2	42.0	42.0
15	28.4	28.3	28.8	28.4
16	23.6	23.8	23.6	23.6
17	46.5	47.0	46.5	46.5
18	43.2	41.8	43.2	43.2
19	42.5	46.3	42.5	43.2
20	44.0	30.8	44.0	44.0
21	30.6	34.1	30.6	31.2
22	34.0	32.6	34.0	34.8
23	28.4	28.3	28.4	28.4
24	17.1	17.1	17.1	17.1
25	15.6	15.6	15.6	15.6
26	17.5	17.5	17.5	17.5
27	26.2	26.2	26.2	26.2
28	177.0	176.5	179.9	180.0
29	28.4	33.2	28.4	29.1
30	176.1	23.6	177.2	179.5
OMe	51.7		51.7	
C-28 G1	95.8	95.8		
G2	74.1	74.1		
G3 G4	79.4	79.3		
	71.0	71.1		
G5 G6	78.9	78.9		
C-3 GA1	61.9 106.6	62.3		1066
GA2	75.3	106.6		106.6
GA2 GA3	78.3	75.3		75.3
GA3 GA4	78.3 73.6	78.4		78.3
GA5	75.6 76.7	73.6 76.6		73.6
GA5 GA6	176.1	76.6 176.2		76.7
UAU	1 / 0. 1	1/0.2		176.1

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Experimental

Optical rotations were measured with a Union PM-101 automatic digital polarimeter. NMR spectra were recorded on a JEOL GX-400 instrument in ${\rm C_5D_5N}$, using tetramethylsilane (TMS) as an internal standard. For gas liquid chromatography (GLC), a Shimadzu GC-6A apparatus was used. Mass spectra (MS) was taken on a JEOL JMS-01-SG-2 spectrometer by the direct inlet method; ionization volatage 75 eV. For column chromatography, Kieselgel 60H (Merck, art. 7736) and Lichroprep RP-18 (25—40 μ m, reversed-phase, Merck) were used.

Extraction and Separation of Saponins The roots of Talinum triangulare were extracted with hot MeOH. The concentrated MeOH extract was separated Et₂O and 1-BuOH, successively. The 1-BuOH extract was subjected to column chromatography on silica gel by eluting with CHCl₃-MeOH-H₂O (30:12:1, 15:6:1, 10:6:1) to give two fractions A and B.

Fration B was chromatographed on silica gel with CHCl₃–MeOH–H₂O (30:8:1, 15:6:1) followed by repeated column chromatography on Lichroprep RP-18 (65% MeOH) affording two saponins, 1 and 2.

Compound 1: A white powder, $[\alpha]_D + 57.5^\circ$ (c = 0.8, C_5H_5N). Anal. Calcd for $C_{43}H_{66}O_{16} \cdot 5H_2O$: C, 54.53; H, 8.30. Found: C, 54.53; H, 8.16. Compound 2: Identification was achieved by comparison of the optical rotation, 1H - and ${}^{13}C$ -NMR spectra with those of an authentic sample.

Acid Hydrolysis of 1 A solution of 1 (2 mg) in 10% $\rm H_2SO_4$ was refluxed for 4h. The reaction mixture was diluted with $\rm H_2O$ and then extracted with $\rm Et_2O$. The $\rm H_2O$ layer was neutralized with Amberlite MB-3 ion exchange resin and evaporated to dryness. The resulting monosaccharides were trimethylsilylated with N-trimethylsilylimidazole and identified by GLC comparison with authentic samples.

Compound 1 afforded D-glucose and D-glucuronic acid, while a genuine aglycone of this saponin was not obtained owing to the acid-catalyzed modification.

A solution of 1 (10 mg) and crude hesperidinase (Tanabe Pharm. Ind. Co. Ltd., Osaka, 10 mg) in $\rm H_2O$ was incubated at 40 °C for 40 h. After heating at 100 °C for a few minutes, the reaction mixture was extracted with Et₂O. The Et₂O layer afforded an aglycone. This product was identified as methyl spergulagenate from the $^1\rm H$ - and $^{13}\rm C$ -NMR spectra and other physical constants.

A suspension of 1 (50 mg) in aqueous 5% KOH was heated on a boiling water bath for 2h. The reaction mixture was neutralized with Amberlite MB-3 resin and then filtered. The filtrate afforded 1,6-anhydroglucose and compound 4 (25 mg). Acid hydrolysis of 4 was carried out by the above mentioned method. Compound 4 yielded D-glucuronic acid.

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References

- I. Kitagawa, K. Kitazawa, K. Aoyama, M. Asanuma, and I. Yoshioka, *Tetrahedron*, 28, 923 (1972).
- 2) R. Kasai, M. Suzuo, and O. Tanaka, Tetrahedron Lett., 1977, 175.
- 3) R. Kasai, M. Okihara, J. Asakawa, K. Mizutani, and O. Tanaka, *Tetrahedron*, 35, 1427 (1979).
- 4) H. Kohda, R. Kasai, K. Yamasaki, K. Murakami, and O. Tanaka, *Phytochemistry*, **15**, 981 (1976).
- T. D. Lin, N. Kondo, and J. Shoji, Chem. Pharm. Bull., 24, 253 (1976).