

Novel α -Mannoside Synthesis Promoted by the Combination of Trimethylsilyl Chloride and Zinc Triflate

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α -Mannosides were obtained in good yields and with good selectivity directly from benzyl-protected mannopyranosyl *p*-nitrobenzoate or acetate using a trimethylsilyl chloride–zinc triflate catalyst system.

Keywords glycosidation; α -mannoside; mannosyl ester; trimethylsilyl chloride; zinc triflate; solvent effect

The glycosidation reaction is one of the most important in carbohydrate chemistry.¹⁾ During the course of our studies on synthesis of human serum albumin modified with D-mannose through octamethylene spacer group *via* ether and amide bonds, it was necessary to develop a new method for the preparation of benzyl-protected α -mannopyranoside (**4a**)²⁾ in good yield and with good selectivity. Usually, α -mannosyl bonds are synthesized with 2-acyl-protected mannosyl donors utilizing the neighboring-group effect,³⁾ but recently, synthesis of α -mannosides from mannosyl fluoride⁴⁾ or chloride,⁵⁾ without the neighboring-group effect, has been reported. These mannosyl halides are prepared from mannosyl esters. In this paper, we wish to describe a selective synthesis, promoted with trimethylsilyl chloride (TMSCl) and zinc triflate (Zn(OTf)₂), of α -mannosides from mannosyl esters without the neighboring-group effect (Chart 1).

Results and Discussion

In a previous paper, we reported a novel method for the direct activation of stable glycosyl esters with TMSCl and Zn(OTf)₂.⁶⁾ We applied the method to the reaction of

benzyl-protected mannopyranosyl *p*-nitrobenzoate (**1**)^{7,8)} to extend the utility of our new glycosidation method. Reaction of **1** with 8-methoxycarboxyloctanol (**3a**) (2 eq)⁹⁾ in dichloromethane in the presence of TMSCl (1.5 eq) and Zn(OTf)₂ (1.5 eq) gave the known substances, **4a**²⁾ and its β -isomer (**4a** β),²⁾ at an 86 : 14 ratio in 89% combined yield (Table I, run 1).

We examined the solvent effect by reaction of **1** and **3a** in the presence of TMSCl and Zn(OTf)₂ as promoters. The results (runs 1–4) made it clear that acetonitrile (run 2) was the best solvent in terms of yields and α -selectivity.

To optimize the reaction conditions, some other silyl halides, such as trimethylsilyl bromide (TMSBr, run 5), *tert*-butyldimethylsilyl chloride (TBDMSCl, run 6), and triphenylsilyl chloride (run 7), were examined using the above-mentioned reaction of **1** and **3a** in the presence of Zn(OTf)₂ in acetonitrile. High yields and α -selectivity were attained when TMSCl was employed as the promoter. Next, we tried using other metal triflates, such as copper triflate (Cu(OTf)₂, run 8) and tin triflate (Sn(OTf)₂, run 9) in combination with TMSCl in acetonitrile, resulting in 72% and 70% yields, and 80 : 20 and 98 : 2 α : β selectivity. These

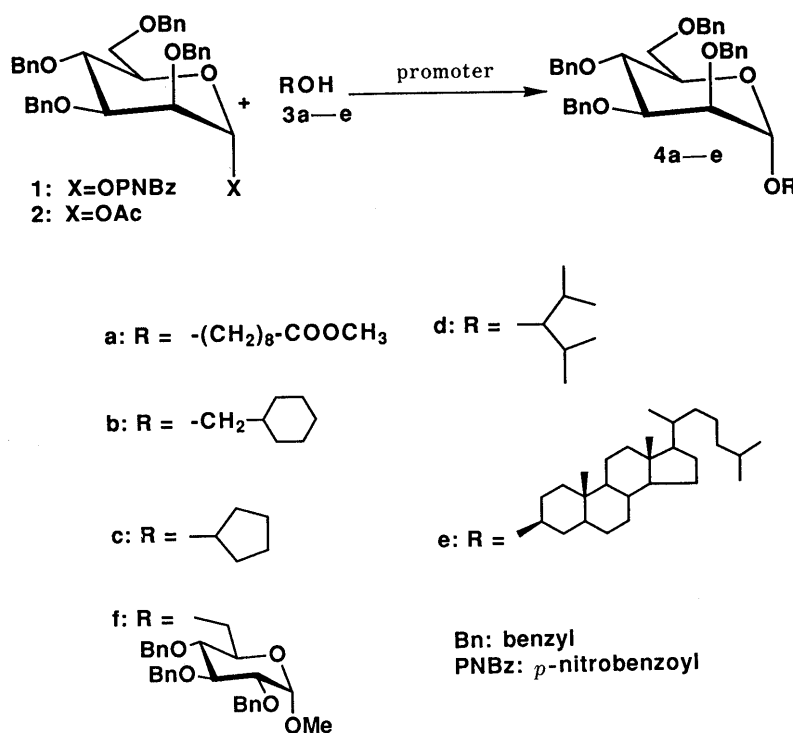


Chart 1

TABLE I. α -Selective Mannoside Synthesis

Run	Donor	Acceptor ^{a)}	Promoter ^{b)}	Solvent ^{c)}	Time ^{d)} (h)	Yield ^{e)} (%)	α : β ^{f)}
1	1	3a	TMSCl-Zn(OTf) ₂	CH ₂ Cl ₂	5.5	89	86:14
2	1	3a	TMSCl-Zn(OTf) ₂	MeCN	4	96	99:1
3	1	3a	TMSCl-Zn(OTf) ₂	PhCH ₃	7.5	73	94:6
4	1	3a	TMSCl-Zn(OTf) ₂	DME	23	73	64:36
5	1	3a	TMSBr-Zn(OTf) ₂	MeCN	1.5	85	99:1
6	1	3a	TBDMSCl-Zn(OTf) ₂	MeCN	3	88	94:6
7	1	3a	Ph ₃ SiCl-Zn(OTf) ₂	MeCN	5.5	89	99:1
8	1	3a	TMSCl-Cu(OTf) ₂	MeCN	5	72	80:20
9	1	3a	TMSCl-Sn(OTf) ₂	MeCN	3	70	98:2
10	2	3a	TMSCl-Zn(OTf) ₂	MeCN	4	88	99:1
11	1	3b	TMSCl-Zn(OTf) ₂	MeCN	5	95	98:2
12	2	3b	TMSCl-Zn(OTf) ₂	MeCN	4	92	97:3
13	1	3c	TMSCl-Zn(OTf) ₂	MeCN	4	89	99:1
14	2	3c	TMSCl-Zn(OTf) ₂	MeCN	5	83	99:1
15	1	3d	TMSCl-Zn(OTf) ₂	MeCN	6	57	96:4
16	1	3d	TMSCl-Zn(OTf) ₂ ^{g)}	MeCN	6	82	95:5
17	1	3e	TMSBr-Zn(OTf) ₂	CH ₂ Cl ₂ ^{h)}	15	91	93:7
18	1	3f	TMSCl-Zn(OTf) ₂	MeCN	1	80	97:3

a) Two amounts of acceptor were used in each case. b) Each promoter was used in an amount of 1.5 eq. c) DME; 1,2-dimethoxyethane. d) All reactions were carried out with ice cooling. e) Isolated total yield. f) The α : β ratios were determined by ¹H-NMR analysis. g) Addition of powdered molecular sieves AW-300 (twice the weight of 1). h) Five equivalents of diglyme was used as an additive.

results showed that the metal cation influenced selectivity. Good yields and α -selectivity were achieved, when Zn(OTf)₂ was employed as the promoter. The best result was attained when the reaction was carried out in acetonitrile using TMSCl and Zn(OTf)₂ as promoters. After stirring with **3a** (2 eq), TMSCl (1.5 eq) and Zn(OTf)₂ (1.5 eq) in acetonitrile with ice cooling for 4 h, **4a β** was isomerized to an epimeric mixture (**4a α** :**4a β** =92:8). This demonstrated that thermodynamic control contributed to the α -selectivity.

In a similar manner, the reaction of benzyl-protected mannopyranosyl acetate (**2**)¹⁰ with **3a** gave **4a α** in good yield and with good selectivity (run 10). We then investigated the reaction of mannosyl esters (**1** and **2**) with other alcohols (**3b**–**f**). All the reactions resulted in the formation of the corresponding α -glycosides (**4b α** –**4f α**) with good selectivity (runs 11–18). When the alcohol (**3d**) was sterically hindered, the yield was low, because of the formation of a hydrolysis product (2,3,4,6-tetrabenzylmannopyranose), which was detected by thin-layer chromatography in the reaction mixture (run 15). In this case, addition of powdered molecular sieves AW-300 improved the yield (run 16). Since β -cholestanol (**3e**) showed poor solubility in acetonitrile, the reaction of **1** and **3e** (run 17) was carried out in dichloromethane containing diglyme, a condition⁶⁾ that we had used for the synthesis of steroidal glycosides. The reaction of **1** with glycosyl alcohol (**3f**) gave the disaccharide **4f α** and its β -isomer (**4f β**) at a 97:3 ratio in an 80% combined yield (run 18); and when **1** and **3f** were reacted using trimethylsilyl triflate⁸⁾ instead of TMSCl and Zn(OTf)₂ as the promoter, with ice cooling for 1 h, the yield was decreased (70%) but the α -selectivity was the same (97:3).

The stereochemical assignment of the glycosidic bond in **4a**–**f** was verified by comparison of the specific rotations of the minor and major isomers based on Hudson's rule of isorotation.¹¹⁾ The major isomers of reaction products were dextrorotatory, in contrast with the minor isomers, indicating the stereochemistry of the glycosidic bond that had formed to be α in the major isomers and β in the minor isomers.

In summary, we have demonstrated that combination of TMSCl and Zn(OTf)₂ can serve as a potential activator in α -mannosidation using mannosyl esters as mannosyl donors.

Experimental

Melting points were determined on a Yanagimoto melting point apparatus and are uncorrected. Proton nuclear magnetic resonance (¹H-NMR) spectra were obtained on a Varian VXR-500 spectrometer (500 MHz) in deuteriochloroform using tetramethylsilane as an internal standard. Optical rotations were measured in chloroform with a Perkin-Elmer 241 polarimeter. IR spectra were obtained on a Hitachi 270-30 IR spectrophotometer. Mass spectra (MS) and high-resolution mass spectra (HRMS) were obtained on a JEOL JMS-HX110 mass spectrometer. Column chromatography was performed with Merck Silica gel 60 (230–430 mesh). Glycosyl alcohol **3f** was purchased from Junsei Chemical Co., Ltd.

Typical Procedure for the α -Selective Glycosidation (Table I, Run 1). TMSCl (0.038 ml, 0.3 mmol) was added to a mixture of the mannosyl ester **1** (138 mg, 0.2 mmol), the alcohol **3a** (75 mg, 0.4 mmol) and Zn(OTf)₂ (109 mg, 0.3 mmol) in dichloromethane (2 ml) was added TMSCl (0.038 ml, 0.3 mmol) under ice-cooling, and the mixture was stirred for 5.5 h. The reaction mixture was diluted with ethyl acetate and washed with water, saturated NaHCO₃ solution and water, then dried over MgSO₄ and concentrated under reduced pressure. The residue was chromatographed on silica gel (70 g) with toluene–ethyl acetate (19:1) to give a 86:14 epimeric mixture (126 mg, 89%) of **4a α** and **4a β** as a colorless oil. Analytical samples of **4a α** and **4a β** were prepared by repeated silica gel column chromatography.

4a α ²⁾: A colorless oil. $[\alpha]_D^{25} +21.1^\circ$ ($c=1.48$). ¹H-NMR δ : 1.27 (8H, s), 1.51–1.52 (2H, m, CH₂–CH₂–COOCH₃), 1.59–1.62 (2H, m, CH₂–CH₂–O), 2.29 (2H, t, $J=7.6$ Hz, CH₂–COOCH₃), 3.34 (1H, dt, $J=9.8, 6.6$ Hz, CH₂–O), 3.37 (1H, dt, $J=9.5, 6.8$ Hz, CH₂–O), 3.66 (3H, s, OCH₃), 3.67–3.80 (4H, m, H-2, 5, 6, 6'), 3.90 (1H, dd, $J_{3,2}=2.9, J_{3,4}=9.3$ Hz, H-3), 3.98 (1H, dd $J_{4,3}=9.3, J_{4,5}=9.0$ Hz, H-4), 4.50 (1H, d, $J=10.7$ Hz, ArCH₂), 4.55 (1H, d, $J=12.2$ Hz, ArCH₂), 4.63 (2H, s, ArCH₂), 4.66 (1H, d, $J=12.2$ Hz, ArCH₂), 4.72 (1H, d, $J=12.7$ Hz, ArCH₂), 4.76 (1H, d, $J=12.7$ Hz, ArCH₂), 4.85 (1H, d, $J_{1,2}=1.7$ Hz, H-1), 4.87 (1H, d, $J=10.7$ Hz, ArCH₂), 7.15–7.17 (2H, m, ArH), 7.23–7.38 (18H, m, ArH). IR (KBr): 2936, 2864, 1736, 1496, 1454, 1364 cm⁻¹. MS m/z : 710 (M⁺), 619 (M⁺–C₇H₇). HRMS Calcd for C₄₄H₅₄O₈–C₇H₇ 619.3271. Found 619.3257.

4a β ²⁾: A colorless solid. mp 54–55 °C. $[\alpha]_D^{25} -46.5^\circ$ ($c=0.10$). ¹H-NMR δ : 1.25–1.39 (8H, m), 1.47–1.53 (2H, m, CH₂–CH₂–COOCH₃), 1.59–1.65 (2H, m, CH₂–CH₂–O), 2.30 (2H, t, $J=7.4$ Hz, CH₂–COOCH₃), 3.40 (1H, dt, $J=9.3, 6.8$ Hz, CH₂–O), 3.45 (1H, ddd, $J_{5,4}=9.5, J_{5,6}=6.1,$

$J_{5,6} = 2.0$ Hz, H-5), 3.50 (1H, dd, $J_{3,2} = 2.9$, $J_{3,4} = 9.5$ Hz, H-3), 3.66 (3H, s, OCH₃), 3.74 (1H, dd, $J_{6,5} = 6.1$, $J_{6,6} = 10.8$ Hz, H-6), 3.81 (1H, dd, $J_{6,5} = 2.0$, $J_{6,6} = 10.8$ Hz, H-6'), 3.86 (1H, dd, $J_{4,3} = 9.5$, $J_{4,5} = 9.5$ Hz, H-4), 3.90 (1H, d, $J_{2,3} = 2.9$ Hz, H-2), 3.97 (1H, dt, $J = 9.3$, 6.4 Hz, CH₂-O), 4.37 (1H, s, H-1), 4.44 (1H, d, $J = 11.7$ Hz, ArCH₂), 4.51 (1H, d, $J = 11.7$ Hz, ArCH₂), 4.53 (2H, d, $J = 10.7$ Hz, ArCH₂), 4.59 (1H, d, $J = 12.0$ Hz, ArCH₂), 4.63 (1H, d, $J = 12.0$ Hz, ArCH₂), 4.87 (1H, d, $J = 12.5$ Hz, ArCH₂), 4.90 (1H, d, $J = 10.7$ Hz, ArCH₂), 5.00 (1H, d, $J = 12.5$ Hz, ArCH₂), 7.17–7.19 (2H, m, ArH), 7.22–7.35 (16H, m, ArH), 7.46–7.47 (2H, m, ArH). IR (KBr): 2932, 2860, 1732, 1497, 1454, 1362 cm⁻¹. MS m/z : 710 (M⁺), 619 (M⁺ - C₇H₇). HRMS Calcd for C₄₄H₅₄O₈ - C₇H₇ 619.3271. Found 619.3241.

Physical and spectral data for glycosides (**4b**–**f**) are as follows.¹²⁾

4ba: A colorless oil. $[\alpha]_D + 28.7^\circ$ ($c = 1.09$). ¹H-NMR δ : 0.84–0.91 (2H, m, cyclohexyl-H), 1.12–1.26 (3H, m, cyclohexyl-H), 1.47–1.70 (6H, m, cyclohexyl-H), 3.15 (1H, dd, $J = 6.1$, 9.3 Hz, O-CH₂), 3.45 (1H, dd, $J = 6.8$, 9.3 Hz, O-CH₂), 3.71–3.79 (4H, m, H-2, 5, 6, 6'), 3.89 (1H, dd, $J_{2,3} = 3.2$, $J_{3,4} = 9.3$ Hz, H-3), 3.97 (1H, dd, $J_{4,3} = 9.3$, $J_{4,5} = 9.0$ Hz, H-4), 4.51 (1H, d, $J = 10.7$ Hz, ArCH₂), 4.55 (1H, d, $J = 12.2$ Hz, ArCH₂), 4.62 (1H, d, $J = 12.0$ Hz, ArCH₂), 4.65 (1H, d, $J = 11.7$ Hz, ArCH₂), 4.66 (1H, d, $J = 12.0$ Hz, ArCH₂), 4.70 (1H, d, $J = 12.5$ Hz, ArCH₂), 4.75 (1H, d, $J = 12.5$ Hz, ArCH₂), 4.81 (1H, d, $J_{1,2} = 1.7$ Hz, H-1), 4.87 (1H, d, $J = 10.5$ Hz, ArCH₂), 7.16–7.18 (2H, m, ArH), 7.23–7.38 (18H, m, ArH). IR (KBr): 2928, 2860, 1498, 1454, 1364 cm⁻¹. MS m/z : 636 (M⁺), 545 (M⁺ - C₇H₇). HRMS Calcd for C₄₁H₄₈O₆ - C₇H₇ 545.2903. Found 545.2883.

4bb: A colorless solid. mp 63–64°C. *Anal.* Calcd for C₄₁H₄₈O₆ · 1/4 H₂O: C, 76.79; H, 7.62. Found: C, 76.64; H, 7.90. $[\alpha]_D - 46.6^\circ$ ($c = 1.26$). ¹H-NMR δ : 0.96–1.01 (2H, m, cyclohexyl-H), 1.16–1.29 (3H, m, cyclohexyl-H), 1.63–1.81 (6H, m, cyclohexyl-H), 3.21 (1H, dd, $J = 6.6$, 9.3 Hz, O-CH₂), 3.45 (1H, dd, $J_{5,4} = 9.8$, $J_{5,6} = 6.1$, $J_{5,6'} = 2.0$ Hz, H-5), 3.50 (1H, dd, $J_{3,4} = 9.6$, $J_{3,2} = 2.9$ Hz, H-3), 3.74 (1H, dd, $J_{6,5} = 6.1$, $J_{6,6'} = 11.0$ Hz, H-6), 3.80–3.83 (2H, m, O-CH₂, H-6'), 3.85 (1H, dd, $J_{4,3} = 9.6$, $J_{4,5} = 9.8$ Hz, H-4), 3.91 (1H, d, $J_{2,3} = 2.9$ Hz, H-2), 4.34 (1H, s, H-1), 4.44 (1H, d, $J = 11.7$ Hz, ArCH₂), 4.51 (1H, d, $J = 11.7$ Hz, ArCH₂), 4.53 (1H, d, $J = 10.7$ Hz, ArCH₂), 4.60 (1H, d, $J = 12.2$ Hz, ArCH₂), 4.63 (1H, d, $J = 12.2$ Hz, ArCH₂), 4.87 (1H, d, $J = 12.7$ Hz, ArCH₂), 4.90 (1H, d, $J = 10.7$ Hz, ArCH₂), 5.03 (1H, d, $J = 12.7$ Hz, ArCH₂), 7.16–7.21 (2H, m, ArH), 7.23–7.36 (16H, m, ArH), 7.46–7.48 (2H, m, ArH). IR (KBr): 2924, 2856, 1496, 1454, 1366 cm⁻¹. MS m/z : 636 (M⁺), 545 (M⁺ - C₇H₇). HRMS Calcd for C₄₁H₄₈O₆ - C₇H₇ 545.2903. Found 545.2911.

4bc: Colorless oil. $[\alpha]_D + 36.1^\circ$ ($c = 1.01$). ¹H-NMR δ : 1.45–1.73 (8H, m), 3.69 (1H, dd, $J_{2,1} = 2.0$, $J_{2,3} = 3.2$ Hz, H-2), 3.72 (1H, dd, $J_{6,5} = 3.9$, $J_{6,6'} = 8.8$ Hz, H-6), 3.78–3.82 (2H, m, H-5, H-6'), 3.90 (1H, dd, $J_{3,2} = 3.2$, $J_{3,4} = 9.3$ Hz, H-3), 3.98 (1H, dd, $J_{4,3} = 9.3$, $J_{4,5} = 9.0$ Hz, H-4), 4.16–4.18 (1H, m, O-CH), 4.35 (1H, d, $J = 10.7$ Hz, ArCH₂), 4.54 (1H, d, $J = 12.0$ Hz, ArCH₂), 4.63 (2H, s, ArCH₂), 4.67 (1H, d, $J = 12.5$ Hz, ArCH₂), 4.70 (1H, d, $J = 12.7$ Hz, ArCH₂), 4.82 (1H, d, $J = 12.5$ Hz, ArCH₂), 4.87 (1H, d, $J = 10.7$ Hz, ArCH₂), 4.90 (1H, d, $J_{1,2} = 2.0$ Hz, H-1), 7.16–7.19 (2H, m, ArH), 7.24–7.38 (18H, m, ArH). IR (KBr): 2940, 2876, 1498, 1456, 1364 cm⁻¹. MS m/z : 608 (M⁺), 517 (M⁺ - C₇H₇). HRMS Calcd for C₃₉H₄₄O₆ - C₇H₇ 517.2590. Found 517.2615.

4cb: Colorless solid. mp 94–95°C. *Anal.* Calcd for C₃₉H₄₄O₆ · 1/4 H₂O: C, 76.38; H, 7.31. Found: C, 76.36; H, 7.47. $[\alpha]_D - 51.1^\circ$ ($c = 1.23$). ¹H-NMR δ : 1.46–1.91 (8H, m), 3.45 (1H, ddd, $J_{5,4} = 9.6$, $J_{5,6} = 6.1$, $J_{5,6'} = 1.7$ Hz, H-5), 3.50 (1H, dd, $J_{3,2} = 2.9$, $J_{3,4} = 9.5$ Hz, H-3), 3.74 (1H, dd, $J_{6,5} = 6.1$, $J_{6,6'} = 10.8$ Hz, H-6), 3.80–3.85 (3H, m, H-2, H-4, H-6'), 4.32–4.34 (1H, m, O-CH), 4.42 (1H, s, H-1), 4.43 (1H, d, $J = 11.7$ Hz, ArCH₂), 4.51 (1H, d, $J = 11.7$ Hz, ArCH₂), 4.54 (1H, d, $J = 10.8$ Hz, ArCH₂), 4.60 (1H, d, $J = 12.0$ Hz, ArCH₂), 4.63 (1H, d, $J = 12.0$ Hz, ArCH₂), 4.63 (1H, d, $J = 12.0$ Hz, ArCH₂), 4.88 (1H, d, $J = 12.7$ Hz, ArCH₂), 4.90 (1H, d, $J = 10.8$ Hz, ArCH₂), 4.98 (1H, d, $J = 12.7$ Hz, ArCH₂), 7.18–7.35 (18H, m, ArH), 7.46–7.47 (2H, m, ArH). IR (KBr): 2960, 2916, 2860, 1498, 1456, 1362 cm⁻¹. MS m/z : 608 (M⁺), 517 (M⁺ - C₇H₇). HRMS Calcd for C₃₉H₄₄O₆ - C₇H₇ 517.2590. Found 517.2615.

4dc: Colorless oil. $[\alpha]_D + 39.6^\circ$ ($c = 0.51$). ¹H-NMR δ : 0.72 (3H, d, $J = 6.8$ Hz, CH₃), 0.75 (3H, d, $J = 6.8$ Hz, CH₃), 0.86 (3H, d, $J = 6.6$ Hz, CH₃), 0.90 (3H, d, $J = 6.8$ Hz, CH₃), 1.68–1.81 (2H, m, CH(CH₃)₂), 3.01 (1H, t, $J = 4.9$ Hz, O-CH), 3.69 (1H, dd, $J_{6,5} = 1.5$, $J_{6,6'} = 10.5$ Hz, H-6), 3.76 (1H, dd, $J_{2,1} = 1.5$, $J_{2,3} = 2.9$ Hz, H-2), 3.81 (1H, dd, $J_{6,6'} = 10.5$, $J_{6,5} = 4.8$ Hz, H-6'), 3.88 (1H, dd, $J_{3,2} = 2.9$, $J_{3,4} = 9.3$ Hz, H-3), 3.92 (1H, ddd, $J_{5,4} = 9.8$, $J_{5,6} = 1.5$, $J_{5,6'} = 4.8$ Hz, H-5), 4.05 (1H, dd, $J_{4,3} = 9.3$, $J_{4,5} = 9.8$ Hz, H-4), 4.50 (1H, d, $J = 12.2$ Hz, ArCH₂), 4.52 (1H, d, $J = 10.7$ Hz, ArCH₂), 4.62 (1H, d, $J = 11.7$ Hz, ArCH₂), 4.67 (1H, d, $J = 12.5$ Hz,

ArCH₂), 4.68 (1H, d, $J = 11.7$ Hz, ArCH₂), 4.69 (1H, d, $J = 12.2$ Hz, ArCH₂), 4.73 (1H, d, $J = 12.5$ Hz, ArCH₂), 4.83 (1H, d, $J_{1,2} = 1.5$ Hz, H-1), 4.88 (1H, d, $J = 10.7$ Hz, ArCH₂), 7.15–7.46 (20H, m, ArH). IR (KBr): 2964, 2936, 2880, 1498, 1456, 1366 cm⁻¹. MS m/z : 638 (M⁺), 547 (M⁺ - C₇H₇). HRMS Calcd for C₄₁H₅₀O₆ - C₇H₇ 547.3059. Found 547.3087.

4db: Colorless oil. $[\alpha]_D - 52.7^\circ$ ($c = 1.39$). ¹H-NMR δ : 0.89 (3H, d, $J = 6.6$ Hz, CH₃), 0.90 (3H, d, $J = 6.8$ Hz, CH₃), 0.99 (3H, d, $J = 6.8$ Hz, CH₃), 1.00 (3H, d, $J = 6.8$ Hz, CH₃), 1.85–1.89 (2H, m, CH(CH₃)₂), 3.13 (1H, t, $J = 5.4$ Hz, O-CH), 3.39 (1H, dt, $J_{5,4} = 3.9$, $J_{5,6} = 9.5$ Hz, H-5), 3.48 (1H, dd, $J_{3,4} = 9.5$, $J_{3,2} = 2.9$ Hz, H-3), 3.78 (2H, d, $J_{6,5} = 9.5$ Hz, H-6), 3.88 (1H, dd, $J_{4,3} = 9.5$, $J_{4,5} = 3.9$ Hz, H-4), 3.93 (1H, d, $J_{2,3} = 2.9$ Hz, H-2), 4.38 (1H, s, H-1), 4.48 (1H, d, $J = 11.7$ Hz, ArCH₂), 4.53 (1H, d, $J = 11.7$ Hz, ArCH₂), 4.57 (1H, d, $J = 12.0$ Hz, ArCH₂), 4.58 (1H, d, $J = 10.7$ Hz, ArCH₂), 4.69 (1H, d, $J = 12.0$ Hz, ArCH₂), 4.85 (1H, d, $J = 12.5$ Hz, ArCH₂), 4.91 (1H, d, $J = 10.7$ Hz, ArCH₂), 5.02 (1H, d, $J = 12.5$ Hz, ArCH₂), 7.22–7.34 (18H, m, ArH), 7.46–7.47 (2H, m, ArH). IR (KBr): 2968, 2880, 1498, 1456, 1366 cm⁻¹. MS m/z : 638 (M⁺), 547 (M⁺ - C₇H₇). HRMS Calcd for C₄₁H₅₀O₆ - C₇H₇ 547.3059. Found 547.3060.

4dc: Colorless solid. mp 109–100°C. *Anal.* Calcd for C₆₁H₈₂O₆: C, 80.39; H, 9.07. Found: C, 80.28; H, 9.22. $[\alpha]_D + 45.4^\circ$ ($c = 0.61$). ¹H-NMR δ : 0.63 (3H, s, CH₃), 0.76 (3H, s, CH₃), 0.86 (3H, d, $J = 6.6$ Hz, CH₃), 0.86 (3H, d, $J = 6.6$ Hz, CH₃), 0.89 (3H, d, $J = 6.6$ Hz, CH₃), 3.73 (1H, dd, $J_{6,5} = 1.7$, $J_{6,6'} = 10.7$ Hz, H-6), 3.75 (1H, dd, $J_{2,1} = 1.5$, $J_{2,3} = 2.9$ Hz, H-2), 3.79 (1H, dd, $J_{6,5} = 5.1$, $J_{6,6'} = 10.7$ Hz, H-6'), 3.86 (1H, ddd, $J_{5,4} = 9.5$, $J_{5,6} = 1.7$, $J_{5,6'} = 5.1$ Hz, H-5), 3.92 (1H, dd, $J_{3,2} = 2.9$, $J_{3,4} = 9.3$ Hz, H-3), 3.97 (1H, dd, $J_{4,3} = 9.3$, $J_{4,5} = 9.5$ Hz, H-4), 4.50 (1H, d, $J = 10.7$ Hz, ArCH₂), 4.54 (1H, d, $J = 12.0$ Hz, ArCH₂), 4.63 (2H, s, ArCH₂), 4.66 (1H, d, $J = 12.0$ Hz, ArCH₂), 4.71 (1H, d, $J = 12.7$ Hz, ArCH₂), 4.76 (1H, d, $J = 12.7$ Hz, ArCH₂), 4.88 (1H, d, $J = 10.7$ Hz, ArCH₂), 5.02 (1H, d, $J_{1,2} = 1.5$ Hz, H-1), 7.16–7.20 (2H, m, ArH), 7.24–7.49 (18H, m, ArH). IR (KBr): 2936, 2872, 1498, 1456, 1368 cm⁻¹. MS m/z : 910 (M⁺), 819 (M⁺ - C₇H₇). HRMS Calcd for C₆₁H₈₂O₆ - C₇H₇ 819.5563. Found 819.5559.

4db: Colorless solid. mp 95–96°C. *Anal.* Calcd for C₆₁H₈₂O₆: C, 80.39; H, 9.07. Found: C, 80.45; H, 9.35. $[\alpha]_D - 25.1^\circ$ ($c = 1.38$). ¹H-NMR δ : 0.65 (3H, s, CH₃), 0.82 (3H, s, CH₃), 0.86 (3H, d, $J = 6.6$ Hz, CH₃), 0.87 (3H, d, $J = 6.6$ Hz, CH₃), 0.90 (3H, d, $J = 6.6$ Hz, CH₃), 3.44 (1H, ddd, $J_{5,4} = 9.5$, $J_{5,6} = 6.4$, $J_{5,6'} = 1.7$ Hz, H-5), 3.49 (1H, dd, $J_{3,2} = 3.2$, $J_{3,4} = 9.3$ Hz, H-3), 3.62–3.69 (1H, m, O-CH), 3.72 (1H, dd, $J_{5,6} = 6.4$, $J_{6,6'} = 10.7$ Hz, H-6), 3.81–3.85 (3H, m, H-2, 4, 6'), 4.43 (1H, d, $J = 11.7$ Hz, ArCH₂), 4.50 (1H, d, $J = 11.7$ Hz, ArCH₂), 4.51 (1H, s, H-1), 4.54 (1H, d, $J = 10.8$ Hz, ArCH₂), 4.59 (1H, d, $J = 12.0$ Hz, ArCH₂), 4.62 (1H, d, $J = 12.7$ Hz, ArCH₂), 4.88 (1H, d, $J = 12.5$ Hz, ArCH₂), 4.91 (1H, d, $J = 10.8$ Hz, ArCH₂), 4.99 (1H, d, $J = 12.5$ Hz, ArCH₂), 7.19–7.21 (2H, m, ArH), 7.27–7.49 (16H, m, ArH), 7.47–7.49 (2H, m, ArH). IR (KBr): 2936, 2872, 1498, 1456, 1368 cm⁻¹. MS m/z : 910 (M⁺), 819 (M⁺ - C₇H₇). HRMS Calcd for C₆₁H₈₂O₆ - C₇H₇ 819.5563. Found 819.5559.

4dc: A colorless oil. $[\alpha]_D + 35.2^\circ$ ($c = 1.24$). ¹H-NMR δ : 3.30 (3H, s, OCH₃), 3.38 (1H, dd, $J_{4,3} = 9.3$, $J_{4,5} = 9.0$ Hz, H(Glc)-4), 3.46 (1H, dd, $J_{2,1} = 3.4$, $J_{2,3} = 9.5$ Hz, H(Glc)-2), 3.58–3.71 (5H, m, H(Glc)-5, 6, H(Man)-6), 3.78 (1H, d, $J_{2,1} = 1.7$ Hz, H(Man)-2), 3.81–3.85 (2H, m, H(Man)-3, 5), 3.95–4.00 (2H, m, H(Glc)-3, H(Man)-4), 4.44 (1H, d, $J = 12.2$ Hz, ArCH₂), 4.48 (1H, d, $J = 10.7$ Hz, ArCH₂), 4.48 (1H, d, $J = 11.2$ Hz, ArCH₂), 4.56 (1H, d, $J = 3.7$ Hz, H(Glc)-1), 4.59 (1H, d, $J = 12.0$ Hz, ArCH₂), 4.61 (2H, s, ArCH₂), 4.67 (1H, d, $J = 12.2$ Hz, ArCH₂), 4.68 (1H, d, $J = 12.5$ Hz, ArCH₂), 4.72 (1H, d, $J = 12.7$ Hz, ArCH₂), 4.77 (2H, d, $J = 12.0$ Hz, ArCH₂), 4.79 (1H, d, $J = 10.7$ Hz, ArCH₂), 4.85 (1H, d, $J = 10.3$ Hz, ArCH₂), 4.87 (1H, d, $J = 10.5$ Hz, ArCH₂), 4.96 (1H, d, $J = 1.7$ Hz, H(Man)-1), 4.97 (1H, d, $J = 10.7$ Hz, ArCH₂), 7.13–7.37 (35H, m, ArH). IR (KBr): 1498, 1456, 1362 cm⁻¹. MS m/z 906 (M⁺), 895 (M⁺ - C₇H₇).

4db: A colorless solid. mp 123–125°C. *Anal.* Calcd for C₆₂H₆₆O₁₁ · 1/2 H₂O: C, 74.75; H, 6.78. Found: C, 74.81; H, 6.85. $[\alpha]_D - 5.6^\circ$ ($c = 1.01$). ¹H-NMR δ : 3.32 (3H, s, OCH₃), 3.36–3.47 (4H, m), 3.51 (1H, dd, $J = 3.7$, 9.8 Hz), 3.69–3.84 (5H, m), 4.01 (1H, dd, $J = 9.3$, 9.3 Hz), 4.16 (1H, dd, $J = 10.5$, 1.7 Hz), 4.47 (1H, d, $J = 12.0$ Hz, ArCH₂), 4.49–4.54 (4H, m, ArCH₂), 4.56 (1H, d, $J = 12.0$ Hz, ArCH₂), 4.57 (1H, d, $J = 4.4$ Hz, H(Glc)-1), 4.58 (1H, d, $J = 12.0$ Hz, ArCH₂), 4.66 (1H, d, $J = 12.0$ Hz, ArCH₂), 4.77 (1H, s, H(Man)-1), 4.72 (1H, d, $J = 12.7$ Hz, ArCH₂), 4.77 (2H, d, $J = 12.0$ Hz, ArCH₂), 4.78 (1H, d, $J = 12.5$ Hz, ArCH₂), 4.82 (1H, d, $J = 11.0$ Hz, ArCH₂), 4.88 (1H, d, $J = 10.7$ Hz, ArCH₂), 4.93 (1H, d, $J = 12.5$ Hz, ArCH₂), 5.01 (1H, d, $J = 11.0$ Hz, ArCH₂), 7.17–7.42 (35H, m, ArH). IR (KBr): 1497, 1454, 1362 cm⁻¹. MS m/z : 906 (M⁺), 895 (M⁺ - C₇H₇).

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