Dioxopyrrolines. LIII.¹⁾ Stereochemical Pathway of [2+2] Photocycloaddition Reaction of 4-Ethoxycarbonyl-5-phenyl-1*H*-pyrrole-2,3-dione to Cyclic Enol Trimethylsilyl Ethers

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The photocycloaddition reaction of the dioxopyrroline 1 to cyclic trimethylsilyloxy olefins was examined. The addition of 3-trimethylsilyloxy-1H-indene and 4-trimethylsilyloxy-1,2-dihydronaphthalene occurred in a regio- and stereoselective manner to give the *cis-anti-cis* cyclobutanes 3 and 4, respectively. The stereochemical result supported the stereo-selection rule which states that the photocycloaddition of a very polar dioxopyrroline-olefin pair proceeds in an s+s manner *via* a favored *O-endo-n*-complex transition. On the other hand, the addition of 1-trimethylsilyloxy-cyclopentene and 1-trimethylsilyloxycyclohexene occurred less stereoselectively, proceeding predominantly in an s+a manner to give the adducts 5a, b, 6a, b, 7a, b, and 12. The results are consistent with the prediction of the stereo-selection rule obtained by assuming that these dioxopyrroline-olefin pairs are polar. The latter cases demonstrate that an increased steric hindrance around the olefin moiety affects the polarity of enone-olefin pair in the transition state.

Keywords photocycloaddition; dioxopyrroline; stereochemistry; donor-acceptor interaction; cyclobutane; X-ray analysis

The stereo-selection rule²⁾ for the enone-olefin photocycloaddition seems to work reliably for rationalizing the stereochemical results observed in the photocycloaddition reactions of dioxopyrroline-olefin pairs.³⁾ The reported examples confirm that the polarity (the magnitude of donor-acceptor interaction) of the pair in the excited π -complex plays an important role in determining the stereochemical pathway of the reaction. The donor-acceptor interaction could be evaluated in terms of the electronic properties of the two addends. In addition, it was demonstrated that the steric factor of the addends is also important in evaluating the polarity of the transition state: that is, increase of steric interaction between donor and acceptor decreases the magnitude of the donor-acceptor interaction (the polarity of the pair), which affects the stereochemical pathway of the [2+2] photocycloaddition.3d)

The photocycloaddition reaction of 4-ethoxycarbonyl-5-phenyl-2,3-dioxo-2,3-dihydro-1H-pyrrole (1, dioxopyrroline) with acyclic enol trimethylsilyl ethers proceeded in a highly regio-selective and stereoselective manner to give the corresponding cyclobutane 2 with O-endo stereochemistry. This stereochemical result is consistent with the prediction (2s+2s) of the stereo-selection rule obtained by assuming that the dioxopyrroline-olefin pair is very polar, since the enol silyl ether is electron rich, and the addition proceeds via the O-endo- π -complex as a favored transition state,

although the experimental evidence for this assumption is lacking. In this paper we treat the photocycloaddition reaction of the dioxopyrroline to cyclic olefins with a trimethylsilyloxy group, which is expected to provide the experimental evidence revealing the stereochemical pathway of the dioxopyrroline-very polar olefin pair as well as the steric effect of the olefinic component.

Results and Discussion

Irradiation of a solution of 1 and 3-trimethylsilyloxy-1*H*-indene in dimethoxyethane (DME) with a high-pressure mercury lamp gave the cyclobutane 3 as a single adduct in 49% yield. The structure of the photoadduct, except for the stereochemistry of the ring juncture, was deduced from their spectral similarity to the cyclobutanes prepared by the photocycloaddition of 1 to olefins. The structure and stereochemistry of 3 were unambiguously established by an X-ray crystallographic analysis as the cyclobutane

EtOOC OOTMS 6
$$\frac{4}{7}$$
 $\frac{H}{8}$ $\frac{COOEt}{30}$ $\frac{2}{9}$ $\frac{1}{9}$ $\frac{1}{$

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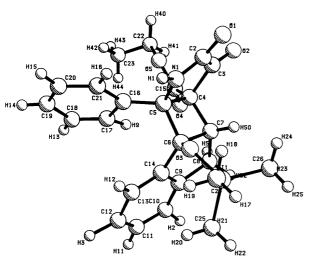


Fig. 1. Cyclobutane 3

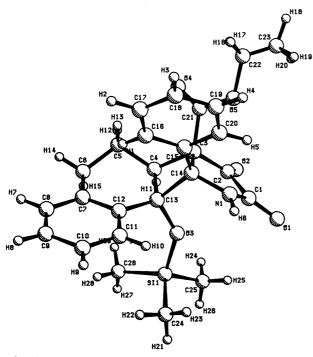


Fig. 2. Cyclobutane 4

with cis-anti-cis configuration (Fig. 1).

4-Trimethylsilyloxy-1,2-dihydronaphthalene, on similar irradiation, gave the cyclobutane 4 as a single adduct in 23% yield. In contrast, 3-trimethylsilyloxy-1,2-dihydronaphthalene did not give any adduct on irradiation. In the latter case, apparent increase of the steric hindrance around the olefin may block the addition. The structure and stereochemistry of 4 were also unambiguously determined by an X-ray analysis as the cyclobutane with cis-anti-cis configuration (Fig. 2).

The results clearly demonstrated that the photocyclo-addition of 1 to 3-trimethylsilyloxy-1H-indene or 4-trimethylsilyloxy-1,2-dihydronaphthalene proceeded stereoselectively in an s+s manner via the O-endo- π -complex of head to tail orientation. This stereochemical result is consistent with the prediction (s+s) of the stereoselection rule obtained by assuming that these pairs are very polar for a favored transition state. This assumption that the

O-endo- π -complex should be most favored among stereochemically different complexes, can be rationalized since this orientation gains the maximum donor-acceptor interaction by accessing the electron-poor dioxopyrroline ring and the electron-rich enol ether moiety.

Next, we carried out photocycloaddition reactions of enol trimethylsilyl ethers of cyclopentanone and cyclohexanone. Irradiation of a solution of 1 and 1-trimethylsilyloxy-1-cyclopentene in DME for 1 h gave four adducts, 5a (4%), 6a (12%), 7a (5%), and 8a (4%). No cyclobutane derivative was isolated in this photocycloaddition.

The adduct 5a has the molecular formula C₂₁H₂₇NO₅Si, indicating that the product is a 1:1 adduct of the addends. However, the spectral data are not consistent with the expected cyclobutane structure. The adduct exhibited a strong ultraviolet (UV) absorption band (280 nm, ε 20000) and the infrared (IR) spectra showed no signals due to an NH group. In addition to these data, the signal due to a methine proton (s, δ 5.01) in the ¹H-nuclear magnetic resonance (1H-NMR) spectrum and the methine sp3 carbon signal in the 13 C-NMR spectrum (δ 86.0) were not consistent with the cyclobutane structure. Instead, the spectral features of this adduct were compatible with the spectral data reported for the 2,4-dioxa-10-azatricyclo-[6.3.0.0^{3,7}]undecane 9, which was generated by the photocycloaddition of 1 to dihydrofuran. The cis-syn-cisstereochemistry of 9 has been unambiguously determined by an X-ray crystallographic analysis. 3b) Thus, the adduct 5a was assigned as a 2-oxa-5-azatricyclo[6.3.0.0^{3,7}]undecane derivative.

The major adduct 6a has the molecular formula $C_{21}H_{29}NO_6Si$ which corresponds to a 1:1 adduct of addends plus H_2O . The 1H - and ^{13}C -NMR spectral data were very similar to those of 5a except for the signals attributable to the acylimino moiety, thus indicating that the adduct 6a was a hydrate of 5. In fact, the hydrate 6a on heating at 200 °C under reduced pressure underwent dehydration to yield 5a. Similarly, 5a readily added methanol at room temperature with retention of the trimethylsilyloxy group to give the methanol adduct 10, which was identical with the compound obtained by recrystallization of 6a from methanol. Treatment of 5a with acidic water caused hydration with concomitant ring

ROH

EIOOC O TOTMS

Ph 6 H 9 10
$$\Delta$$

Ph 6 H 9 10 Δ

Ph 6 H 9 10 Δ

Ph 6 H 9 10 Δ

Ph 7 O TMS

Ph 8 H 9 0 O TMS

Ph 8 H 9 0 O TMS

Ph 9 10 Δ

Ph 9 10 Δ

Ph 10

Chart 4

7b

opening by desilylation, to give the keto-hydrate 11, which was identical with the product derived from 6a with a similar acidic treatment.

6b

66

The observed highly stereoselective addition of water or methanol to the acylimino moiety can be reasonably explained from the assigned stereochemistry of 5a (cissyn-cis), since convex face addition of ROH to 5a leads to the adduct 6a with the RO group stereochemistry shown. If 5a had different stereochemistry, such as cis-anti-cis, the addition of ROH would not be expected to occur stereoselectively.

The structure of **7a** was readily deduced from the spectral data. This compound has a molecular formula corresponding to a 1:1 adduct minus CO. The presence of the dihydropyridone moiety was revealed by the UV (285 nm) and IR spectra (1640 cm⁻¹). These were compatible with the reported data for the dihydropyridones.^{3c)}

The adduct 8a has a molecular formula corresponding to a 1:1 adduct minus the (TMS) group and exhibited a fairly strong UV absorption band at 309 nm (ε6100). The IR (3200 cm⁻¹) and ¹H-NMR (δ 5.64, disappeared on addition of D₂O) spectra indicated the presence of an OH group. In addition to these data, the ¹H-NMR and ¹³C-NMR data clearly indicated the presence of a cyclopentanone moiety. The structure 8a, corresponding to an aldol-type condensation product to the 3-keto group of the dioxopyrroline, was obtained for the spectral similarity

of the trimethylsilyloxycyclohexene adduct **8b** (see below), whose structure was established by an X-ray analysis (Fig. 4)

8**b**

Similar irradiation of 1 and 1-trimethylsilyloxy-1-cyclohexene afforded five adducts, the cyclobutane 12 (6%), the acylimino derivative 5b (4%), the hydrate derivative 6b (3%), the dihydropyridone 7b (1%), and the aldol-type adduct 8b (1%).

The structure of 12 was established as the cyclobutane with *cis-syn-trans* stereochemistry by an X-ray analysis (Fig. 3) and the structure of 8b was also determined by an X-ray analysis (Fig. 4). The structural assignment of the other adducts 5b, 6b, and 7b was achieved by spectral comparison with the corresponding adducts obtained from the trimethylsilyloxycyclopentene.

The formation of **5** and **7** can be rationalized as discussed in connection with the photocycloaddition of **1** to dihydrofuran or dihydropyran. The products **5** should be produced through the cyclobutane of *cis-anti-trans* configuration **13**, which was formed *via* the *O-endo-* π -complex in an s+a manner. This highly strained intermediate would decompose into **15** *via* the lactim form **14**. The resulting zwitter-ion **15** cyclizes, with rotation of the C_6 - C_7 bond, followed by ketonization to give the thermodynamically more stable *cis*-fused furano derivative **5**. This mechanism explains well the stereochemistry at the four chiral centers newly formed in this reaction. On the

January 1993 67

other hand, 7 may be produced through the cyclobutane of cis-syn-cis configuration 16 formed via the O-exo- π complex in an s+s manner. This intermediate has favorable stereochemistry for 1,3-rearrangement through a lactim form 17. Subsequent cheletropic loss of CO from 18 gives the dihydropyridones (7).

The stereochemical pathway of the photocycloaddition reactions can be summarized as shown in Table I. In the case of the trimethylsilyloxycyclopentene, the product distribution appears to be consistent with the stereoselection rule if this pair is polar, which predicts that a [2s+2a] addition and a [2s+2s] addition are preferred in the favored transition (the *O-endo-\pi*-complex) and in the less favored one (the *O-exo-\pi*-complex), respectively. The

decreased polarity of this pair, when compared to the dioxopyrroline-trimethylsilyloxyindene pair (very polar pair), may be attributable to the increased steric effect due to the more puckered olefin geometry. A similar steric effect affecting the polarity of dioxopyrroline-olefin pairs and, therefore, the stereochemical pathway of their photocycloadditions was previously observed. 3c,d)

In the case of the trimethylsilyloxycyclohexene, the cyclobutane of cis-syn-trans configuration 12 was isolated. The stereochemistry clearly shows that this adduct is formed in an s+a manner through the O-exo- π -complex. It is reasonable to consider that the product 5b, like 5a, is formed in an s+a manner via the O-endo- π -complex as described in the case of the trimethylsilyloxycyclopentene.

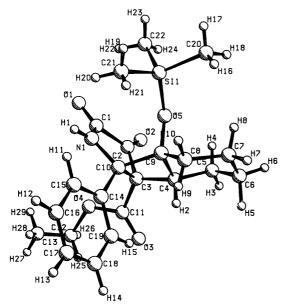


Fig. 3. Cyclobutane 12

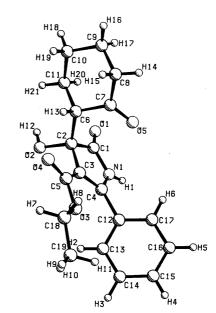


Fig. 4. Paterio-Büchi Product 8b

68 Vol. 41, No. 1

Table I. The Stereochemical Pathways of the Photocycloaddition Reaction of 1 to Trimethylsilyloxy (OTMS) Cyclic Olefins

OTMS cyclic olefins	via O-endo-π- complex		via O-exo-π- complex		Paterio- Büchi
	s + s	s + a	s+s	s+a	
3-OTMS-indene	49 (3)			_	_
4-OTMS-dihydronaphthalene	23 (4)	_	_	_	
1-OTMS-cyclopentene		16 (5a + 6a)	5 (7a)		4 (8a)
1-OTMS-cyclohexene	_	7(5b+6b)	1 (7b)	6 (12)	1 (8b)

The stereochemical results appear to be inconsistent with the stereo-selection rule, since both routes give the products of a [2s+2a] addition. The product ratio of the O-endo-pathway/the O-exo-pathway is roughly one, and this indicates that the transition states should be similar with respect to their transition energy. Thus, both complexes might give a [2s+2a] adduct with the stereochemistry predicted from the polar pair.

The products $\bf 8$ are, in formal sense, adducts of aldol-condensation of $\bf 1$ to the silyl enol ethers. However, the formation is rationalizable in terms of the Paterio-Büchi reaction⁴⁾ rather than the aldol condensation. The [2+2] addition between the ketone of dioxopyrroline and the olefin forms the oxetane $\bf 19$ and hydrolysis of the OTMS group followed by spontaneous opening of the oxetane ring gives the product $\bf 8$. The *erythro* configuration of $\bf 8$ is consistent with the assumption that the [2+2] addition occurs stereospecifically, probably in an $\bf s+\bf s$ manner, as shown in the Chart $\bf 6$. This Paterio-Büchi reaction is the first to be observed in the photocycloaddition of $\bf 1$ to olefins.

In conclusion, the results obtained by the addition of 3-trimethylsilyloxy-1*H*-indene and 4-trimethylsilyloxy-1,2-dihydronaphthalene to 1 confirm that the photocycloaddition of the very polar pair proceeds in an s+s manner *via* a favored transition as predicted by the stereo-selection rule. On the other hand, the photocycloaddition reactions of 1-trimethylsilyloxy-1-cyclopentene and 1-trimethylsilyloxy-1-cyclohexene to 1 gave examples showing that the steric effect due to the bulkiness around the olefin moiety lowers the polarity of the dioxopyrroline-olefin pair as described in the previous paper.^{3d)}

Experimental

Unless otherwise noted, the following procedures were adopted. All melting points are uncorrected. IR spectra were measured as Nujol mulls and are given in cm⁻¹. NMR spectra were taken on a JEOL JNM-FX 100 (1 H-NMR, 100 MHz; 13 C-NMR, 25 MHz) spectrometer in CDCl₃ with tetramethylsilane as an internal standard and the chemical shifts are given in δ values. The following abbreviations are used; s=singlet,

TABLE II. Crystal Data for 3, 4, 8b, and 12

Compound	3	4	8b	12
Crystal forms	Monoclinic	Monoclinic	Triclinic	Monoclinic
a (Å)	12.833 (0)	14.309 (3)	9.977 (2)	15.100 (2)
b (Å)	19.546 (0)	12.593 (2)	12.374 (2)	10.626 (3)
c (Å)	20.605 (0)	15.337 (2)	7.672 (2)	15.517 (2)
α (degree)			10594 (2)	
β (degree)	107.50 (0)	116.24 (1)	96.19 (2)	116.24 (1)
γ (degree)			68.19 (1)	
$V(Å^3)$	4929 (0)	2478.9 (8)	845.5 (3)	2318.6 (8)
$D_c (g/cm^3)$	1.21	1.242	1.349	1.190
Space group	$P2_1/n$	$P2_1/n$	$P\overline{1}$	$P2_1/n$
Z value	8	4	2	4
Reflection observed	5018	6208	4096	5842
Reflection used for calc.	2070	2650	1746	2442
R value	0.094	0.050	0.041	0.050

d=doublet, t=triplet, q=quartet, m=multiplet, and br=broad. Highresolution mass spectra (HRMS) were determined with a JEOL JMS-D 300 spectrometer at 30 eV by using a direct inlet system. UV spectra were measured in EtOH and data are given in $\lambda_{\rm max}$ nm (s). Preparative thin layer chromatography (PTLC) was performed with precoated silica gel plates Merck 60 F254 (0.5 mm thick). Column chromatography was carried out with silica gel (Wakogel C-200). Medium pressure liquid chromatography (MPLC) was performed on a Kusano CIG prepacked silica gel column. The photolysis was done by internal irradiation using a 300 W high-pressure mercury lamp (Eikosha Halos PIH 300) with a Pyrex filter.

Photocycloaddition of 1 to 3-Trimethylsilyloxy-1H-indene A solution of 1 (2 g, 8 mmol) and 3-trimethylsilyloxy-1H-indene (8.4 g, 40 mmol) in DME (300 ml) was irradiated at 0 °C for 1 h. After removal of the solvent by evaporation in vacuo, the residue in benzene was chromatographed over SiO₂ (Mallinckrodt, CC-7). Elution with benzene-CH₂Cl₂ (1:1) and crystallization of the product from CH₂Cl₂-Et₂O gave dl-(3aR*,3bS*,8bS*,8cS*)-3a-ethoxycarbonyl-8c-phenyl-2,3-dioxo-8btrimethylsilyloxy-1,2,3,3a,3b,4,8b,8c-octahydro-1H-indeno[2',1':3,4]cyclobuta[1,2-b]pyrrole (3, 1.8 g, 49%) as colorless prisms, mp 206-207 °C. IR: 1760, 1720. ¹H-NMR: 0.13 (9H, s, OTMS), 0.94 (3H, t, J=7 Hz, COOCH₂C $\underline{\text{H}}_3$), 3.42 (1H, d, J=9 Hz, H-4), 3.70 (1H, dd, J=9, 17 Hz, H-4), 3.81—4.21 (2H, m, COOC $\underline{\text{H}}_2\text{CH}_3$), 4.69 (1H, d, J=9 Hz, H-3b), 6.77—7.61 (9H, m, ArH). $^{13}\text{C-NMR}$: 1.1 (3C, q, OTMS), 13.2 (q, COOCH₂CH₃), 31.1 (t, C4), 50.0 (d, C3b), 56.7 (s, C3a), 61.2 (t, COOCH₂CH₃), 71.9 (s, C8c), 91.6 (s, C8b), 124.9 (d, Ar), 126.3 (d, Ar), 127.6 (d, Ar), 127.8 (d, Ar), 126.6 (2C, d, Ar), 127.3 (2c, d, Ar), 129.8 (d, Ar), 134.3 (s, Ar), 140.0, 144.2 (each s, C4a, C8a), 161.3 (s, C4), 165.7 (s, COOCH₂CH₃), 195.7 (s, C3).

Photocycloaddition of 1 to 4-Trimethylsilyloxy-1,2-dihydronaphthalene A solution of 1 (2 g, 8 mmol) and 4-trimethylsilyloxy-1,2-dihydronaphthalene (8.9 g, 40 mmol) in DME (300 ml) was irradiated at 0 °C for 1 h. After removal of the solvent by evaporation *in vacuo*, the residue in benzene was chromatographed over SiO₂(Mallinckrodt, CC-7). Elution with benzene—CH₂Cl₂ (1:1) and crystallization of the product from CH₂Cl₂—Et₂O gave *dl*-(3a *R**,3b *S**,9b *S**,9c *S**)-3a-ethoxycarbonyl-9c-phenyl-2,3-dioxo-9b-trimethylsilyloxy-1,2,3,3a,3b,4,5,9b,9c-octahydro-1*H*-naphtho[2',1':3,4]cyclobuta[1,2-*b*]pyrrole (4, 880 mg, 23%) as colorless prisms, mp 241—242 °C. IR: 1770, 1740. ¹H-NMR: 0.15 (9H, s, OTMS), 1.12 (3H, t, *J*=7 Hz, COOCH₂CH₃), 2.79—3.23 (5H, m, H-3b, 4, 5), 4.16 (2H, q, *J*=7 Hz, COOCH₂CH₃), 7.12—7.58 (9H, m, ArH). ¹³C-NMR: 1.3 (3C, q, OTMS), 13.4 (q, COOCH₂CH₃), 25.4 (t, C4 or C5), 27.9 (t, C4 or C5), 50.0 (d, C3b), 60.1 (s, C3a), 61.3 (t,

H(3)

H(4)

H(5)

H(6)

-0.0262

0.2762

0.5452

0.8026

0.4570

0.0671

0.1369

-0.0629

0.8577

0.7359

0.7055

0.6872

7.3

4.6

5.1

7.3

 $B_{\rm eq}$ 6.8 5.8 6.8 6.4 7.5 6.5 7.7 5.8 7.4 6.1 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 6.9 5.7 7.1 8.9 7.8 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 3.9 6.0 6.0 6.5 6.5

Table III. Positional Parameters and $B_{\rm eq}$ for the Cyclobutane 3

TABLE III. (continued)

TABLE III.	Positional Para	B_{eq} imeters and B_{eq} i	or the Cyclobi	itane 3	TABLE III.	(continued)		
Atom	x	у	z	$B_{ m eq}$	Atom	x	у	Z
Si(1)	0.1240 (5)	0.1753 (3)	0.9204 (3)	7.6 (3)	H(7)	0.5005	-0.2030	0.6220
Si(2)	0.3451 (5)	0.0707 (4)	0.5661 (3)	8.7 (4)	H(8)	0.3994	-0.1207	0.6432
O(1)	0.144 (1)	0.0562 (6)	0.7026 (7)	6.9 (7)	H(9)	-0.1270	0.3111	0.6689
O(2)	-0.0925 (9) 0.1049 (8)	0.0549 (6)	0.6509 (6)	5.6 (6)	H(10),	0.2823	-0.0239	0.7979
O(3) O(4)	-0.264 (1)	0.2041 (5) 0.2051 (7)	0.8418 (6)	4.2 (5)	H(11)	-0.2138	0.4198	0.8496
O(4)	-0.142 (1)	0.2031 (7)	0.6355 (6) 0.5824 (7)	7.3 (8) 7.3 (8)	H(12) H(13)	0.0733 -0.0947	0.3534	0.8339
O(6)	0.3403 (9)	0.1832 (6)	0.7833 (7)	6.1 (6)	H(14)	0.0713	0.4248 0.4570	0.6331 0.6163
O(7)	0.5750 (8)	0.1903 (6)	0.8291 (6)	5.5 (6)	H(15)	0.2146	0.3797	0.6388
O(8)	0.3794 (8)	0.0379 (6)	0.6397 (5)	5.1 (6)	H(16)	0.1954	0.2729	0.6866
O(9)	0.753 (1)	0.0392 (6)	0.8427 (7)	7.3 (7)	H(17)	0.2869	0.1598	1.0029
O(10)	0.626 (1)	0.0479 (7)	0.8989 (7)	5.5 (6)	H(18)	0.3009	0.1460	0.9319
N(1)	0.113 (1)	0.1699 (7)	0.7228 (7)	4.7 (7)	H(19)	0.2957	0.2206	0.9565
N(2)	0.372 (1)	0.0719 (7)	0.7615 (6)	3.5 (6)	H(20)	0.0988	0.2780	0.9708
C(2)	0.085 (1)	0.108 (1)	0.701 (1)	5 (1)	H(21)	-0.0106	0.2380	0.9484
C(3) C(4)	-0.040 (1) -0.080 (1)	0.103 (1) 0.1719 (8)	0.6757 (8)	4.0 (9)	H(22)	0.0795	0.2174	1.0144
C(5)	0.025 (1)	0.1719 (8)	0.696 (1) 0.7203 (8)	4.0 (8) 3.6 (8)	H(23) H(24)	-0.0093 0.0997	0.0946	0.9019
C(6)	0.009 (1)	0.2196 (9)	0.7203 (8)	4.3 (9)	H(25)	0.0997	0.0611 0.0795	0.9006 0.9696
C(7)	-0.089 (1)	0.1653 (8)	0.7673 (8)	3.3 (8)	H(26)	0.7104	-0.1765	0.6504
C(8)	-0.187 (1)	0.199 (1)	0.7858 (9)	5 (1)	H(27)	0.6055	-0.0718	0.8032
C(9)	-0.155 (1)	0.269 (1)	0.8055 (8)	4.2 (9)	H(28)	0.6001	-0.1751	0.8595
C(10)	-0.218 (1)	0.323 (1)	0.821 (1)	5 (1)	H(29)	0.4404	-0.2059	0.8809
C(11)	-0.171 (2)	0.385 (1)	0.839 (1)	6 (1)	H(30)	0.2943	-0.1366	0.8537
C(12)	-0.066 (2)	0.400 (1)	0.844 (1)	6 (1)	H(31)	0.4889	0.0611	0.5271
C(13)	-0.002 (1)	0.345 (1)	0.829 (1)	5 (1)	H(32)	0.4206	0.1253	0.4965
C(14) C(15)	-0.046 (1) -0.174 (2)	0.2830 (8) 0.1950 (9)	0.8081 (8)	3.7 (8)	H(33)	0.4983	0.1270	0.5705
C(15)	-0.174 (2) 0.038 (1)	0.1930 (9)	0.635 (1) 0.6867 (8)	4 (1) 3.2 (7)	H(34) H(35)	0.1966 0.2352	0.0206	0.5313
C(17)	-0.048 (1)	0.332 (1)	0.670 (1)	5 (1)	H(36)	0.2332	$0.0490 \\ -0.0168$	0.4720 0.5101
C(18)	-0.035 (2)	0.394 (1)	0.644 (1)	6 (1)	H(37)	0.3582	0.1791	0.6123
C(19)	0.063 (2)	0.4125 (9)	0.6335 (9)	5 (1)	H(38)	0.2810	0.1804	0.5381
C(20)	0.148 (2)	0.368 (1)	0.647 (1)	6 (1)	H(39)	0.2391	0.1528	0.5963
C(21)	0.135 (1)	0.304 (1)	0.6754 (8)	5 (1)	H(40)	-0.2082	0.1772	0.4776
C(22)	-0.229 (2)	0.202 (1)	0.511 (1)	10.7 (7)	H(41)	-0.2986	0.1866	0.5126
C(23)	-0.230 (2)	0.271 (1)	0.499 (1)	10.9 (7)	H(42)	-0.1592	0.2859	0.5001
C(24)	0.269 (2) 0.066 (2)	0.176 (1)	0.957 (1)	9.1 (6)	H(43)	-0.2804	0.2810	0.4557
C(25) C(26)	()	0.234 (1) 0.093 (1)	0.969 (1)	10.7 (7)	H(44)	-0.2521	0.2950	0.5332
C(32)	0.067 (2) 0.400 (1)	0.093 (1)	0.923 (1) 0.7826 (8)	9.0 (6) 4 (1)	H(45) H(46)	0.7366 0.7615	-0.0160	0.9523
C(33)	0.523 (1)	0.1401 (9)	0.7820 (8)	4.1 (9)	H(47)	0.7813	0.0596 0.0617	0.9744 1.0115
C(34)	0.566 (1)	0.0756 (8)	0.7863 (8)	3.0 (8)	H(48)	0.5972	-0.017	0.9882
C(35)	0.463 (1)	0.0265 (8)	0.7621 (8)	3.3 (8)	H(49)	0.7038	0.0032	1.0457
C(36)	0.477 (1)	0.0235 (7)	0.6889 (8)	2.9 (8)	H(50)		0.1210	0.7869
C(37)	0.571 (1)	0.0781 (8)	0.709 (1)	5 (1)	H(51)	-0.2509	0.1983	0.7473
C(38)	0.670 (1)	0.0508 (9)	0.693 (1)	6 (1)	H(52)	-0.2018	0.1753	0.8224
C(39)	0.641 (2)	-0.0274(9)	0.6793 (8)	4.1 (9)	H(53)	0.6796	0.0727	0.6543
C(40)	0.711 (1)	-0.078 (1)	0.671 (1)	6 (1)	H(54)	0.7342	0.0568	0.7308
C(41) C(42)	0.666 (2) 0.557 (2)	-0.140 (1)	0.656 (1)	6 (1)				
C(42) C(43)	0.337 (2)	-0.153 (1) -0.103 (1)	0.6496 (9) 0.6588 (9)	6 (1)	COOCH	L) 73.5 (s. C9	c), 81.5 (s, C9b),	126.5 (d. Ar
C(44)	0.534 (1)	-0.103 (1) -0.040 (1)	0.6738 (8)	4.2 (9) 4 (1)	Ar), 127.6 (d	i, Ar), 127.8 (2	2C, d, Ar), 128.2	2 (2C d Ar)
C(45)	0.660 (2)	0.054 (1)	0.844 (1)	5 (1)	134.1 (s, Ar),	, 134.5 (s, C5a	or C9a), 138.2 (s	6. C5a or C9a
C(46)	0.542 (1)	-0.084 (1)	0.8150 (9)	5 (1)		OCH ₂ CH ₃), 19		,
C(47)	0.538 (2)	-0.146 (1)	0.847 (1)	6 (1)	Photocyclo	addition of 1 to	1-Trimethylsily	loxy-1-cyclope
C(48)	0.445 (2)	-0.163 (1)	0.860 (1)	7 (1).	tion of 1 (2	2g, 8mmol) a	nd 1-trimethylsi	lyloxy-1-cycle
C(49)	0.359 (2)	-0.122 (1)	0.845 (1)	7 (1)	40 mmol) in	DME (300 mi) was irradiated	d at 0°C for
C(50)	0.361 (1)	-0.056 (1)	0.815 (1)	5 (1)	removal of the	ne solvent by	evaporation in v	acuo, the resi
C(51)	0.454 (2)	-0.0408(9)	0.7997 (8)	4.0 (9)	was chromat	ographed over	SiO ₂ . The elua	ites with ben
C(52) C(53)	0.706 (2) 0.655 (2)	0.026 (1)	0.960 (1)	9 (1)	(1:4 and 1:) and CH ₂ Cl ₂ '	were purified by (194 mg, 4%), 7	MPLC [with
C(54)	0.655 (2)	0.019 (1) 0.099 (1)	1.004 (1) 0.537 (1)	10.4 (7)		a (100 mg, 4%)		a (145 mg, 5%
C(55)	0.430 (2)	0.039 (1)	0.515 (1)	10.8 (7) 13.7 (8)			Ethoxycarbonyl-	4-0x0-6-phen
C(56)	0.302 (2)	0.156 (1)	0.513 (1)	14.5 (8)	silyloxy-2-oxa	a-5-azatricyclo	6.3.0.0 ^{3,7} Jundec	-5-ene (5a): Pa
H(1)	0.2073	0.1857	0.7451	5.7	IR: 1760, 17	740. UV (diox	ane): 280 (2000	0). ¹ H-NMR
H(2)	-0.3097	0.3144	0.8264	6.7	OTMS), 1.07	7 (3H, t, $J = 7$)	Hz, COOCH₂C <u>I</u>	(\underline{H}_3) , 1.44—2.0
H(3)	-0.0262	0.4570	0.8577	7.3	10), 2.0—2.2	8 (2H, m. H-1	1), 3.36 (1H t	1-8H2 H2

ar), 127.5 (2C, d, r), 129.6 (d, Ar), a), 164.5 (s, C2),

pentene A solulopentene (6.4 g, or 45 min. After sidue in benzene enzene, benzeneh EtOAc-hexane 5%), **6a** (400 mg,

enyl-1-trimethyl-Pale yellow gum. R: 0.17 (9H, s, J=7 Hz, COOCH₂CH₃), 5.01 (1H, s, H-3), 7.27—7.65 (3H, m, ArH), 8.01—8.05 (2H, m, ArH). ¹³C-NMR: 0.9 (3C, q, OTMS), 13.6 (q, COOCH₂CH₃), 24.0 (t, C9 or C10), 26.7 (t, C9 or C10), 36.9 (t, C11), 53.1 (d, C8), 62.1 (t, COOCH₂CH₃), 67.9 (s, C7), 86.0 (d, C3), 124.5 (s,

Table IV. Positional Parameters and B_{eq} for the Cyclobutane 4

 $B_{\rm eq}$ Atom 0.03940 (7) 3.94 (4) 0.77518 (8) 0.34163 (8) Si(1) 1.0909 (2) 0.3890 (2) 4.7 (1) 0.0765 (2) O(1)0.2683(2) 6.6 (1) O(2)1.1220 0.3180 (2) (2)0.4256 0.1058 (1)3.64(9)0.8682 O(3)(2) (2)0.4480 12.0 (2) O(4) 1.1365 0.5299 (4) (2) (2) 1.2108 0.5619 0.3535(2) 6.9 (1) (3) O(5)(2)0.1196 (2) 3.2 (1) 0.5351 N(1)1.0242 (2) (2) 1.0682 0.4392 (3) 0.1327 (2) 3.6 (1) C(1)(2) 0.2341 (2) 4.1 (2) 1.0858 0.4025 C(2)(3) (3) 0.4885 0.2702 (2) 3.5 (1) C(3)1.0460 (2)(3) 0.9407 0.4574 0.2789 (2) 3.7 (1) C(4)(3) (3) 0.5083 0.3529 (3) 4.7 (2) 0.9108 (4) C(5)(3) (2) 0.3121 (3) 5.2 C(6)0.7923 (3) 0.5047 (4) 0.7397 0.5645 0.2176(2) 4.3 (1) C(7)(3) (3) (2) (2) 0.1927 (4)5.6 0.64710.6196 (4) C(8)(3) 6.4 0.1079C(9)0.6015 (4)0.6759 (4)(4) 0.6456 0.6785 0.0442 (4) 5.8 (2) C(10)(3) (4) 0.0672 4.4 (2) (3)0.7374 0.6236 C(11)(3)(3)3.5 (1) C(12)0.7842 (2)0.5667 (3)0.1526(2) 0.1756 C(13) 0.8825 0.5051 (2) (2) 3.1 (1) (2) C(14)0.9886 (2)0.5662(3) 0.1911 (2) 3.1 (1) 1.0041 (2) 0.68510.2094 (2) 3.4 (1) C(15)(3) 0.9469 (3) 0.7479 (3) 0.2417 (3) 4.4 (2) C(16)5.5 0.2577 (2) C(17)0.9661 (3) 0.8551 (3) (3)C(18) 1.0427 (4) 0.9027 (4) 0.2416 (3) 5.6 (2)0.2125 (4) 6.6 (2) 1.1031 0.8417 (4) C(19)(4) C(20)1.0841 (3) 0.7344(3) 0.1969 (3)5.5 (2) C(21) 1.1344 0.5289 (4) 0.3714 (3) 5.4 (2)(3) 0.6147 0.4316 13.6 (4) C(22)1.2987 (4) (7) (5)C(23)1.3822 0.60660.4346 (7)16.5 (5) (5) (7) C(24) 0.6992 0.3989 -0.0818 (3)6.7 (2)(4) (3)6.5 (2) 0.0316 0.2227 (3)C(25)0.8467 (3) (3) 0.6890 0.3082 0.0959 (3) 6.0 (2) C(26)(3) (3) 3.7 0.894 0.718 0.252 (2) (8) (2) H(1)(2) 0.900 0.278 (3) 6 (1) H(2)0.925 (3) (3)1.058 0.974 0.250 (2)4.4 (8) H(3)(2) (3) 0.871 0.199 (3) 7 (1) H(4)1.157 (3)(3)8 0.176 (3) H(5)1.127 (3)0.692 (3)(1) 0.998 0.560 0.065 (2) 2.8 (7) H(6)(2)(2)0.238 (3) 7 (1) H(7)0.616 (3) 0.618 (3)9 0.094 0.539 (3) 0.717 (3) (3) (1) H(8)-0.013(3) 0.617 (3) 0.718 (3) (1) H(9) 3.5 (7) 0.022 (2) H(10)0.767 (2) 0.625 (2)H(11)0.929 (2) 0.382 (2) 0.275 (2) 3.1 (7) 0.412 0.943 (2) 0.472 (3) (2) 4.6 (8) H(12)6 (1) H(13)0.936 (3) 0.589 (3) 0.367 (2) 5.5 0.771 0.536 (3) 0.357 (2) (9) H(14)(3) 0.306 0.428 6 (1) H(15)0.772 (3) (3) 0.4406 8.0 0.6889 H(16) 1.2851 H(17)1.3122 0.5835 0.4989 8.0 0.6385 0.4828 8.0 H(18)1.4455 H(19)1.3802 0.6445 0.3709 8.0 0.5353 0.4229 8.0 H(20)1.4034 -0.12927.9 0.6604 0.3472 H(21)7.9 H(22)0.6467 0.4501 -0.0804-0.10437.9 0.7408 0.4386 H(23)0.0958 9.4 0.1912 H(24)0.8847 H(25)0.8964 0.2406 0.0085 9.4 -0.00949.4 0.8005 0.1704 H(26)0.2406 0.0743 8.8 H(27)0.6572 0.6352 0.3604 0.0799 8.8 H(28)H(29)0.7276 0.3061 0.1655

C1), 128.8 (2C, d, Ph), 130.4 (2C, d, Ph), 130.6 (s, Ph), 134.8 (d, Ph), 169.4 (s, $COOCH_2CH_3$), 190.1 (s, C6), 192.1 (s, C4). HRMS m/z: Calcd for $C_{21}H_{27}NO_5Si(M^+)$: 401.1658. Found: 401.1671.

dl- $(1R^*,3S^*,6S^*,7S^*,8S^*)$ -7-Ethoxycarbonyl-6-hydroxy-4-oxo-6-phenyl-1-trimethylsilyloxy-2-oxa-5-azatricyclo[6.3.0.0^{3,7}]undecane (**6a**): Colorless prisms from CH₂Cl₂-Et₂O, mp 189—191 °C. IR: 3400, 3200,

Table V. Positional Parameters and B_{eq} for the Paterio-Büchi Product 8b

Atom	x	У	Z	$B_{ m eq}$
O(1)	0.1743 (2)	0.5173 (2)	0.4285 (2)	3.7 (1)
O(2)	0.1319(2)	0.3404(2)	0.6002(3)	3.6(1)
O(3)	0.5541 (2)	0.1324(1)	0.7595 (2)	3.5 (1)
O(4)	0.3571(2)	0.2503(2)	0.9235 (3)	5.9 (1)
O(5)	0.4127 (2)	0.5353 (2)	0.7692 (3)	4.3 (1)
N(1)	0.3923 (2)	0.3608 (2)	0.4102 (3)	3.6 (1)
C(1)	0.2623 (3)	0.4350(2)	0.4855 (3)	2.9(1)
C(2)	0.2430 (3)	0.3907(2)	0.6471 (3)	2.7(1)
C(3)	0.3864 (2)	0.2871 (2)	0.6465 (3)	2.6 (1)
C(4)	0.4662 (3)	0.2744 (2)	0.5061 (3)	2.8 (1)
C(5)	0.4271 (3)	0.2239 (2)	0.7895 (4)	3.0 (1)
C(6)	0.2107 (3)	0.4900 (2)	0.8263 (4)	2.9 (1)
C(7)	0.3193 (3)	0.5514 (2)	0.8736 (4)	3.4 (1)
C(8)	0.2994 (4)	0.6385 (3)	1.0579 (5)	5.6 (2)
C(9)	0.1469 (4)	0.7337 (3)	1.0729 (5)	4.9 (2)
C(10)	0.0351 (3)	0.6761 (3)	1.0268 (4)	4.4 (2)
C(11)	0.0593 (3)	0.5892 (3)	0.8396 (4)	3.8 (2)
C(12)	0.6136 (3)	0.1925 (2)	0.4443 (3)	2.8 (1)
C(13)	0.6502 (3)	0.0685 (2)	0.3917 (4)	3.7 (1)
C(14)	0.7879 (3)	-0.0056(3)	0.3341 (4)	4.5 (2)
C(15)	0.8891 (3)	0.0438 (3)	0.3276 (4)	5.0 (2)
C(16)	0.8551 (3)	0.1658 (3)	0.3799 (4)	4.8 (2)
C(17)	0.7163 (3)	0.2404 (3)	0.4352 (4)	3.9 (2)
C(18)	0.6084 (4)	0.0681 (3)	0.8992 (4)	4.8 (2)
C(19)	0.7290 (5)	-0.0448(4)	0.8224 (6)	6.5 (2
H(1)	0.429 (3)	0.368 (2)	0.331 (3)	3.8 (7)
H(2)	0.580 (3)	0.037 (2)	0.398 (3)	4.4 (7)
H(3)	0.813 (3)	-0.094(2)	0.297 (4)	5.4 (7
H(4)	0.981 (3)	-0.007 (2)	0.291 (4)	6.0 (8
H(5)	0.929 (3)	0.199 (3)	0.379 (4)	7.3 (9
H(6)	0.693 (3)	0.322 (2)	0.477 (3)	4.5 (7
H(7)	0.527 (3)	0.054 (3)	0.935 (4)	7.1 (9
H(8)	0.636 (3)	0.126 (3)	1.006 (5)	8 (1
H(9)	0.767 (4)	-0.095 (3)	0.906 (5)	8 (1
H(10)	0.706 (4)	-0.092(3)	0.719 (5)	10 (1
H(11)	0.809 (4)	-0.029 (4)	0.789 (6)	11 (1
H(12)	0.056 (3)	0.397 (3)	0.594 (4)	6.3 (9
H(13)	0.218 (2)	0.452 (2)	0.919 (3)	2.9 (5
H(14)	0.372 (3)	0.672 (3)	1.068 (5)	8 (1
H(15)	0.307 (4)	0.596 (3)	1.155 (5)	9 (1
H(16)	0.135 (3)	0.782 (2)	1.192 (4)	5.0 (7
H(17)	0.133 (3)	0.786 (2)	0.986 (4)	4.9 (7
H(18)	-0.060 (3)	0.741 (2)	1.027 (3)	4.6 (7
H(19)	0.039 (3)	0.635 (3)	1.122 (4)	6.0 (8
H(20)	0.052 (3)	0.633 (2)	0.747 (3)	4.1 (6
H(21)	-0.016 (3)	0.554 (2)	0.815 (3)	4.9 (7

1740, 1730, 1720, 1700. $^1\text{H-NMR}$: 0.15 (9H, s, OTMS), 1.53 (3H, t, J=7Hz, COOCH₂CH₃), 0.67—2.05 (6H, m, H-9, 10, 11), 2.90 (1H, t, J=8Hz, H-8), 4.23—4.59 (2H, m, COOCH₂CH₃), 5.33 (1H, brs, OH), 5.59 (1H, s, H-3), 7.31—7.59 (3H, m, ArH), 7.71—7.99 (2H, m, ArH), 9.30 (1H, br s, NH). $^{13}\text{C-NMR}$: 0.9 (3C, q, OTMS), 14.1 (q, COOCH₂CH₃), 23.1 (t, C9 or C10), 25.1 (t, C9 or C10), 35.3 (t, C11), 55.7 (d, C8), 61.7 (t, COOCH₂CH₃), 68.0 (s, C7), 82.1 (d, C3), 87.8 (s, C6), 120.8 (s, C1), 126.7 (2C, d, Ph), 127.9 (2C, d, Ph), 128.7 (d, Ph), 140.4 (s, Ph), 170.6 (s, COOCH₂CH₃), 175.1 (s, C4). HRMS m/z: Calcd for $C_{21}H_{27}\text{NO}_{5}\text{Si}(\text{M}^+ - \text{H}_{2}\text{O})$: 401.1658. Found: 401.1710.

dl-(4a R^* ,7a S^*)-4-Ethoxycarbonyl-1-oxo-3-phenyl-7a-trimethylsilyloxy-1,4a,5,6,7,7a-hexahydro-2H-cyclopenta[1,2-c]pyridine (7a): Colorless prisms from hexane–Et₂O, mp 138—140 °C. IR: 1690, 1670, 1640. UV: 285 (10000). ¹H-NMR: 0.14 (9H, s, OTMS), 0.94 (3H, t, J=7 Hz, COOCH₂CH₃), 1.20—1.98 (4H, m, H-5, 6), 2.04—2.72 (2H, m, H-7), 3.23 (1H, t, J=8 Hz, H-4a), 3.95 (2H, q, J=7 Hz, COOCH₂CH₃), 7.1 (1H, brs, NH), 7.26—7.44 (5H, m, ArH). ¹³C-NMR: 1.5 (3C, q, OTMS), 13.7 (q, COOCH₂CH₃), 20.4 (t, C5 or C6), 29.6 (t, C5 or C6), 35.7 (t, C7), 48.2 (d, C4a), 59.9 (t, COOCH₂CH₃), 80.6 (s, C7a), 109.2 (s, C4), 127.8 (2C, d, Ph), 128.3 (2C, d, Ph), 129.1 (d, Ph), 135.9 (s, Ph), 142.8 (s, C3), 167.0 (s, COOCH₂CH₃), 171.1 (s, C1). HRMS m/z: Calcd

TABLE VI. Positional Parameters and B_{eq} for the Cyclobutane 12

Atom	X	y	z	$B_{ m eq}$
			<u> </u>	
Si(1)	0.53708 (7)	0.4074 (1)	0.75896 (8)	4.51 (5)
O(1)	0.5738 (2)	0.3580 (2)	1.0328 (2)	4.4 (1)
O(2)	0.7648 (2)	0.2941 (3)	1.0378 (2)	5.1 (1)
O(3)	0.9238 (2)	0.5910 (4)	1.1118 (2)	7.8 (2)
O(4)	0.8019 (2)	0.5609 (3)	1.1572 (2)	6.0 (1)
O(5)	0.6467 (2)	0.4398 (2)	0.8318 (2)	3.8 (1)
N(1)	0.6080 (2)	0.5430 (3)	0.9747 (2)	3.2 (1)
C(1)	0.6262 (2)	0.4271 (3)	1.0098 (2)	3.4 (1)
C(2)	0.7285 (2)	0.3957 (4)	1.0192 (2)	3.6 (1)
C(3)	0.7735 (2)	0.5171 (3)	1.0036 (2)	3.5 (1)
C(4)	0.8008 (2)	0.5249 (4)	0.9179 (3)	4.0 (2)
C(5)	0.8467 (3)	0.4344 (5)	0.8729 (3)	5.3 (2)
C(6)	0.8461 (4)	0.5054 (5)	0.7866 (4)	6.6 (3)
C(7)	0.7494 (4)	0.5590 (5)	0.7258 (3)	6.2 (2)
C(8)	0.6944 (3)	0.6336 (4)	0.7769 (3)	4.8 (2)
C(9)	0.6968 (2)	0.5551 (3)	0.8586 (2)	3.7 (1)
C(10)	0.6840 (2)	0.6024 (3)	0.9528 (2)	3.1 (1)
C(11)	0.8442 (3)	0.5605 (4)	1.0965 (3) 1.2492 (3)	4.6 (2)
C(12)	0.8559 (5)	0.6091 (7)	` '	9.9 (3) 24 (1)
C(13)	0.8025 (6) 0.6863 (3)	0.649 (1) 0.7432 (3)	1.2896 (6) 0.9656 (2)	24 (1) 3.4 (1)
C(14) C(15)	0.6863 (3) 0.6011 (3)	0.8085 (4)	0.9424 (3)	5.0 (2)
C(15)	0.5991 (4)	0.9377 (4)	0.9492 (4)	6.3 (2)
C(10)	0.6818 (5)	1.0051 (5)	0.9794 (3)	6.4 (3)
C(17)	0.7662 (4)	0.9435 (5)	1.0012 (3)	5.9 (2)
C(19)	0.7691 (3)	0.8137 (4)	0.9947 (3)	4.9 (2)
C(20)	0.5345 (5)	0.3775 (7)	0.6404 (4)	6.7 (3)
C(21)	0.4533 (4)	0.5364 (7)	0.7544 (5)	7.1 (3)
C(22)	0.5038 (5)	0.2605 (6)	0.8033 (5)	6.7 (3)
H(1)	0.554 (2)	0.577 (3)	0.964 (2)	4.0 (8)
H(2)	0.834 (2)	0.600 (3)	0.926 (2)	3.0 (7)
H(3)	0.915 (3)	0.406 (4)	0.915 (3)	7 (1)
H(4)	0.811 (2)	0.356 (3)	0.859 (2)	4.5 (9)
H(5)	0.901 (3)	0.572 (4)	0.814 (3)	8 (1)
H(6)	0.867 (2)	0.452 (3)	0.748 (2)	4.7 (9)
H(7)	0.750 (3)	0.612 (4)	0.672 (3)	8 (1)
H(8)	0.710 (3)	0.488 (4)	0.693 (3)	6 (1)
H(9)	0.726 (2)	0.711 (3)	0.804 (2)	4.1 (8)
H(10)	0.628 (2)	0.653 (3)	0.733 (2)	3.8 (8)
H(11)	0.540 (2)	0.764 (3)	0.917 (2)	4.5 (8)
H(12)	0.536 (3)	0.975 (4)	0.930 (3)	8 (1) 7 (1)
H(13)	0.677 (3)	1.095 (4)	0.983 (3)	7 (1) 7 (1)
H(14)	0.825 (3) 0.828 (2)	0.989 (4) 0.771 (3)	1.020 (3) 1.009 (2)	3.4 (8)
H(15)	0.828 (2) 0.539 (3)	0.771 (3) 0.456 (5)	0.608 (3)	8 (1)
H(16) H(17)	0.339 (3)	0.430 (3)	0.609 (3)	9 (2)
H(18)	0.582 (3)	0.312 (5)	0.639 (3)	9 (1)
H(19)	0.402 (4)	0.525 (5)	0.706 (4)	9 (2)
H(20)	0.441 (4)	0.557 (6)	0.806 (4)	11 (2)
H(21)	0.464 (5)	0.609 (6)	0.733 (5)	13 (3)
H(22)	0.489 (3)	0.273 (5)	0.860 (3)	9 (2)
H(23)	0.449 (4)	0.229 (6)	0.762 (4)	12 (2)
H(24)	0.550 (4)	0.205 (5)	0.820 (4)	10 (2)
H(25)	0.8951	0.6763	1.2438	11.9
H(26)	0.8946	0.5435	1.2852	11.9
H(27)	0.8403	0.6792	1.3495	28.8
H(28)	0.7641	0.7158	1.2546	28.8
H(29)	0.7628	0.5828	1.2953	28.8

for C₂₀H₂₇NO₄Si(M⁺): 373.1708. Found: 373.1743.

dl-(3R*,1'R*)-4-Ethoxycarbonyl-3-hydroxy-2-oxo-3-2'-oxocyclopentyl-5-phenyl-2,3-dihydro-1H-pyrrole (8a): Colorless prisms from CH₂Cl₂−Et₂O, mp 138—141 °C. IR: 3200, 1760, 1740, 1690, 1670. UV: 309 (6100). ¹H-NMR: 1.08 (3H, t, J=7 Hz, COOCH₂CH₃), 1.60—2.24 (4H, m), 2.24—2.48 (2H, m), 2.71—2.88 (1H, m), 4.06 (2H, q, J=7 Hz, COOCH₂CH₃), 5.64 (1H, br s, OH), 7.21—7.61 (5H, m, ArH), 8.10 (1H, br s, NH). ¹³C-NMR: 13.8 (q, COOCH₂CH₃), 20.4 (t, C4' or C5'), 24.8 (t, C4' or C5'), 39.1 (t, C3'), 50.8 (d, C1'), 60.2 (t, COOCH₂CH₃), 80.3 (s, C3), 109.2 (s, C4), 128.1 (2C, d, Ph), 128.6 (2C, d, Ph), 129.3 (s, Ph),

130.8 (d, Ph), 153.2 (s, C5), 163.2 (s, $QOOCH_2CH_3$), 177.2 (s, C2), 220.5 (s, C2'). *Anal.* Calcd for $C_{18}H_{19}NO_5$: C, 65.64; H, 5.82; N, 4.25. Found: C, 65.39; H, 5.66; N, 4.02. HRMS m/z: Calcd for M^+ : 329.1264. Found: 329.1286.

Pyrrolysis of 6a Compound **6a** $(50 \, \text{mg})$ was heated neat at $205 \, ^{\circ}\text{C}$ under reduced pressure $(5 \, \text{mmHg})$ to give **5a** as a pale yellow oil in quantitative yield.

Methanol Adduct 10 1) Compound 5a (50 mg) in MeOH was heated under reflux for 5 min. Evaporation of MeOH gave dl-(1 R^* ,3 S^* ,6 S^* ,7 R^* , 8 S^*)-7-ethoxycarbonyl-6-methoxy-4-oxo-6-phenyl-1-trimethylsilyloxy-2-oxa-5-azatricyclo[6.3.0.0^{3.7}]undec-5-ene (10) in a quantitative yield, as colorless prisms from hexane–Et₂O, mp 218—224 °C. IR (KBr): 3332, 3200, 3100, 1719. The ¹H- and ¹³C-NMR spectra of 10 in CDCl₃ exhibited complex signals which probably correspond to a mixture of 5a, 10, and methanol. The material recovered from the NMR measurement was crystallized from MeOH to give pure crystals of 10. HRMS m/z: Calcd for $C_{21}H_{27}NO_5Si(M^+-MeOH)$: 401.1632. Found: 401.1634.

2) Compound **6a** (10 mg) in MeOH (0.1 ml) was heated under reflux for 1 min. Evaporation of the solvent gave **10**.

Hydrolysis of 5a with HCl Compound 5a (50 mg) in acetone (10 ml) and 0.01% HCl (10 ml) was stirred at room temperature for 2.5 h. After removal of the solvent in vacuo, the residue was crystallized from acetone to give dl-(3R*,4R*,5R*,1R*)-4-ethoxycarbonyl-3,5-dihydroxy-2-oxo-4-2'-oxocyclopentyl-5-phenyltetrahydropyrrole (11, 25 mg) as colorless prisms, mp 206—207 °C. IR: 3400, 3170, 3080, 1740, 1722, 1705.

1H-NMR (dioxane- d_6): 1.31 (3H, t, J=7 Hz, COOCH₂CH₃), 1.70—2.0 (7H, m), 4.1—4.3 (2H, m, COOCH₂CH₃), 5.15 (1H, br s), 7.3—7.4 (3H, m, Ph), 7.6—7.7 (2H, m, Ph). Anal. Calcd for $C_{18}H_{21}NO_6$: C, 62.24; H, 6.10; N, 4.03. Found: C, 62.10, H. 6.04; N, 3.95. MS: 329 (M + H_2 O), 246 (base peak), 200 (base peak)

Hydrolysis of 6a with HCl Compound 6a (50 mg) in acctone (10 ml) was treated with HCl as described above to give 11 (30 mg) as colorless prisms, mp 209—211 °C.

Photocycloaddition of 1 to 1-Trimethylsilyloxy-1-cyclohexene A solution of 1 (3 g, 12 mmol) and 1-trimethylsilyloxy-1-cyclohexene (10.4 g, 60 mmol) in DME (300 ml) was irradiated at 0 °C for 45 min. After removal of the solvent by evaporation in vacuo, the residue in benzene was chromatographed over SiO_2 . The eluates with benzene, benzene-CH₂Cl₂ (1:1) and CH₂Cl₂ were purified by MPLC [with EtOAc-hexane (1:4 and 1:1)] and PTLC [with EtOAc-hexane (1:1)] to give 12 (320 mg, 6%), 5b (186 mg, 4%), 6b (120 mg, 3%), 7b (40 mg, 1%), and 8b (50 mg, 1%).

dl-(1R*,2R*,6R*,7R*)-6-Ethoxycarbonyl-4,5-dioxo-2-phenyl-1-trimethylsilyloxy-3-azatricyclo[5.4.0.0².6]undecane (12): Colorless prisms from CH₂Cl₂-Et₂O, mp 198—199 °C. IR: 3200, 1780, 1720. ¹H-NMR: 0.19 (9H, s, OTMS), 0.94 (3H, t, J=7 Hz, COOCH₂CH₃), 1.22—1.70 (6H, m, H-8, 9, 10), 1.70—1.83 (2H, m, H-11), 2.80 (1H, t, J=8 Hz, H-7), 4.02 (2H, q, J=7 Hz, COOCH₂CH₃), 7.42 (5H, br s, ArH). ¹³C-NMR: 2.1 (3C, q, OTMS), 13.7 (q, COOCH₂CH₃), 19.7 (t, C9 or C10), 21.7 (t, C9 or C10), 25.2 (t, C8 or C11), 32.0 (t, C8 or C10), 47.6 (d, C7), 61.8 (t, COOCH₂CH₃), 64.5 (s, C6), 74.3 (s, C2), 84.9 (s, C1), 125.1 (2C, d, Ph), 128.3 (d, Ph), 129.2 (2C, d, Ph), 133.5 (s, Ph), 165.3 (s, C4), 166.9 (s, COOCH₂CH₃), 191.7 (s, C5). HRMS m/z: Calcd for C₂₂H₂₂9NO₅Si(M⁺): 415.1815. Found: 415.1852.

dl-(1R*,3R*,7S*,8R*)-7-Ethoxycarbonyl-4-oxo-6-phenyl-1-trimethylsilyloxy-2-oxa-5-azatricyclo[6.4.0.0^{3,7}]dodec-5-ene (**5b**): Pale yellow gum. IR: 1740. UV (dioxane): 281 (13000). 1 H-NMR: 0.17 (9H, s, OTMS), 1.06 (3H, t, J=7 Hz, COOCH₂CH₃), 1.18—1.92 (6H, m, H-9, 10, 11), 2.14—2.38 (1H, m, H-12), 2.48—2.76 (1H, m, H-12), 3.39 (1H, dd, J=6, 10 Hz, H-8), 4.10 (2H, q, J=7 Hz, COOCH₂CH₃), 5.15 (1H, s, H-3), 7.28—7.65 (3H, m, ArH), 8.13—8.22 (2H, m, ArH). 13 C-NMR: 1.5 (3C, q, OTMS), 13.6 (q, COOCH₂CH₃), 22.5, 23.5, 26.3 (each t, C9, C10, or C11), 35.2 (t, C12), 48.6 (d, C8), 62.0 (t, COOCH₂CH₃), 72.0 (s, C7), 83.3 (d, C3), 128.8 (2C, d, Ph), 130.8 (2C, d, Ph), 131.3 (s, Ph), 134.8 (d, Ph), 169.6 (s, COOCH₂CH₃), 190.0 (s, C6), 192.4 (s, C4). HRMS m/z: Calcd for C₂₂H₂₉NO₅Si(M*): 415.1813. Found: 415.1802.

dl-(1R*,3S*,6S*,7R*,8S*)-7-Ethoxycarbonyl-6-hydroxy-4-oxo-6-phenyl-1-trimethylsilyloxy-2-oxa-5-azatricyclo[6.4.0.0^{3.7}]dodecane (**6b**): Colorless prisms from CH₂Cl₂-Et₂O, mp 176—177.5 °C. IR: 3400, 3200, 1720, 1700. ¹H-NMR: 0.16 (9H, s, OTMS), 1.05 (3H, t, J=7 Hz, COOCH₂CH₃), 1.2—1.95 (7H, m, H-9, 10, 11, 12), 2.2—2.4 (1H, m, H-12), 3.22 (1H, dd, J=6, 13 Hz, H-8), 4.10 (2H, q, J=7 Hz, COOCH₂CH₃), 5.16 (1H, s, H-3), 7.3—7.6 (3H, m, ArH), 8.19 (2H, m, ArH), 9.19 (1H, br s, NH). ¹³C-NMR: 0.9 (3C, q, OTMS), 14.0 (q, COOCH₂CH₃), 22.5, 24.2, 25.0 (each t, C9, C10 or C11), 35.8 (t, C12),

50.8 (d, C8), 61.1 (t, COOCH₂CH₃), 69.4 (s, C7), 80.6 (d, C3), 87.1 (s, C6), 109.6 (s, C1), 126.6 (2C, d, Ph), 127.6 (2C, d, Ph), 128.1 (d, Ph), 141.4 (s, Ph), 174.1 (s, COOCH₂CH₃), 174.8 (s, C4). HRMS m/z: Calcd for $C_{19}H_{29}NO_{5}(M^{+}-TMS)$: 343.1420. Found: 343.1446.

dl-(4a R^* ,8a R^*)-4-Ethoxycarbonyl-1-oxo-3-phenyl-8a-trimethylsilyloxy1,2,4a,5,6,7,8,8a-octahydroisoquinoline (7b): Colorless prisms from hexane–Et₂O, mp 203—205 °C. IR: 1680, 1640. UV: 286 (10000). ¹H-NMR: 0.13 (9H, s, OTMS), 0.90 (3H, t, J=7 Hz, COOCH₂CH₃), 1.20—1.84 (6H, m, H-6, 7, 8), 1.92—2.16 (1H, m, H-5), 2.52—2.70 (1H, m, H-5), 2.90 (1H, dd, J=4, 11 Hz, H-4a), 3.93 (2H, q, J=7 Hz, COOCH₂CH₃), 7.07 (1H, br s, NH), 7.27—7.51 (5H, m, ArH). HRMS m/z: M⁺ Calcd for C₂₁H₂₉NO₄Si: 387.1864. Found: 387.1856.

dl-(3R*,1′R*)-4-Ethoxycarbonyl-3-hydroxy-2-oxo-3-2′-oxocyclohexyl-5-phenyl-2,3-dihydro-1H-pyrrole (8b): Colorless prisms, mp 166—168 °C from CH₂Cl₂-Et₂O. IR: 3350, 1720, 1700, 1680. UV: 236 (12000), 300 (8000). ¹H-NMR: 1.03 (3H, t, J=7 Hz, COOCH₂CH₃), 1.44—2.60 (8H, m), 3.50 (1H, dd, J=5.5, 12 Hz, H-1′), 3.90 (1H, s, OH), 4.03 (2H, q, J=7 Hz, COOCH₂CH₃), 7.36—7.59 (5H, m, ArH), 8.02 (1H, br s, NH). ¹³C-NMR: 13.8 (q, COOCH₂CH₃), 25.0, 27.2 27.4 (each t, C4′, C5′ or C6′), 42.5 (t, C3′), 56.2 (d, C1′), 60.0 (t, COOCH₂CH₃), 77.4 (s, C3), 109.3 (s, C4), 127.9 (2C, d, Ph), 128.6 (2C, d, Ph), 130.0 (d, Ph), 130.5 (s, Ph), 153.6 (s, C5), 163.8 (s, COOCH₂CH₃), 177.7 (s, C2), 211.7 (s, C2′). Anal. Calcd for C₁₉H₂₁NO₅: C, 66.46; H, 6.16; N, 4.08. Found: C, 66.15; H, 6.06; N, 4.14. HRMS m/z: Calcd for M*: 343.1420. Found: 343.1455.

X-Ray Crystallographic Analyses The reflection data were collected on a Rigaku AFC-5 four-circle diffractometer using graphite-monochromated MoK_{α} radiation in the ω - 2θ scan mode at a 2θ scan speed of

 4° /min for $3^{\circ} < 2\theta < 55^{\circ}$. Of the reflections collected, those above the $3\sigma(I)$ level were used for the calculation. The structures of 3, 4, 8b, and 12 were solved by the direct method using MITHRIL⁵) and refined by the full matrix least-squares procedure with anisotropic temperature factors for the non-hydrogen atoms and isotropic temperature factors for the hydrogen atoms. Some hydrogen atoms were located at calculated positions. The crystal data for 3, 4, 8b, and 12 are collected in Table II and the atomic parameters of these compounds are listed in Tables III, IV, V, and VI.

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