

## KINGINOSIDE, A NEW ACYL GROUP CARRYING IRIDOID BIOSIDE FROM LONICERA MORROWII

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A new iridoid glycoside, kinginoside, was isolated from *Lonicera morrowii* (Caprifoliaceae). Its structure was deduced to be sweroside-6'-O-(4"-O-feruloyl)- $\alpha$ -L-rhamnoside mainly by spectroscopic analysis. The structure was confirmed by chemical synthesis of the corresponding acetate from sweroside of known structure.

KEYWORDS kinginoside; *Lonicera morrowii*; Caprifoliaceae; iridoid glycoside; structure determination; chemical correlation

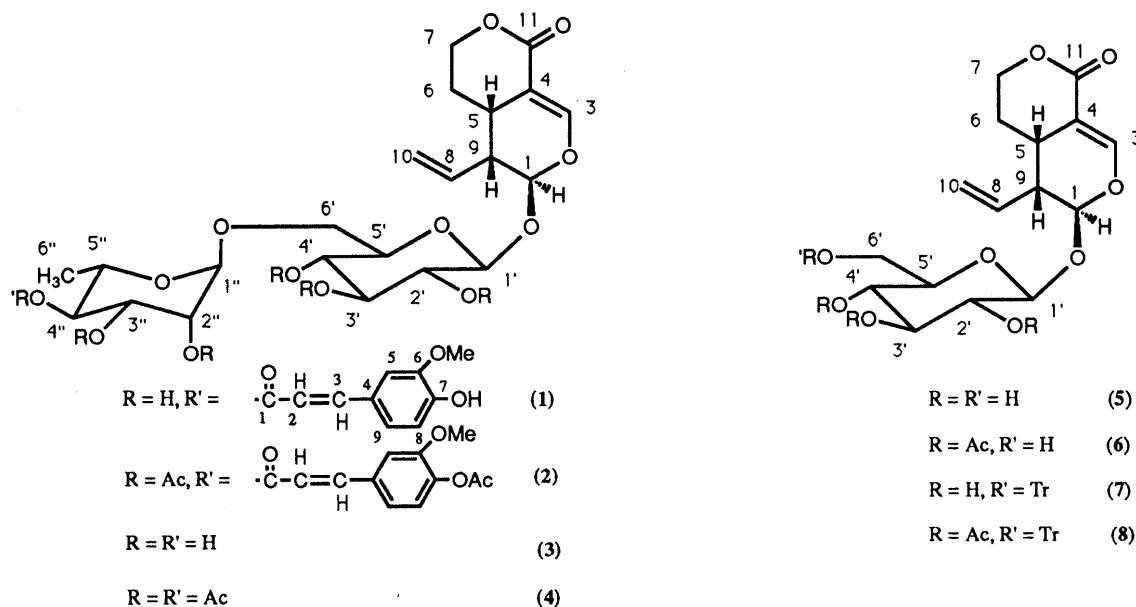
*Lonicera morrowii* is known as a source plant of secologanin, an iridoid glucoside of particular importance in biosynthesis of various types of monoterpenoid indole alkaloids. From this plant three other iridoid glycosides, morroniside, kingiside, and sweroside, were found.<sup>1)</sup> Recently the constituents of its fresh fruits, which are mentioned as toxic in authoritative books, have been thoroughly studied and six known iridoid glucosides and chlorogenic acid isolated.<sup>2)</sup> On our renewed study of the constituents of this plant we now found a new iridoid glycoside, kinginoside (1). This compound was proved to belong to a rare class of iridoids composed of seco-type iridoid aglycone, a biose unit, and an aroyl group. The structure of 1 was elucidated by use of spectral methods and regio- and stereoselective synthesis of kinginoside acetate (2).

The stem part of *Lonicera morrowii* was extracted with hot methanol, and the extract was partitioned between ether and water. The water layer was extracted with 1-butanol, and the butanol-soluble part was chromatographed over silica gel. The fraction eluted with CHCl<sub>3</sub>-MeOH-H<sub>2</sub>O was purified repeatedly by open column chromatography over silica gel and normal phase medium pressure liquid chromatography to give kinginoside (1)<sup>3)</sup> as an amorphous powder. High-resolution FAB-MS indicated the molecular formula to be C<sub>32</sub>H<sub>40</sub>O<sub>16</sub> (Observed m/z; 681.239 (M + H)<sup>+</sup>, Calcd. for C<sub>32</sub>H<sub>41</sub>O<sub>16</sub>, m/z; 681.239).

The UV spectrum of 1 showed the absorption maxima at 226, 230, 280, and 308(sh) nm and revealed the presence of a feruloyl residue as well as a  $\beta$ -alkoxy acrylic ester chromophore typical of iridoid moiety. This finding was chemically proved to be correct by the treatment of 1 with sodium methoxide, when ferulic acid and desferuloylkinginoside (3) were obtained. Compound 3 was shown to have the molecular formula C<sub>22</sub>H<sub>32</sub>O<sub>13</sub> by high-resolution FAB-MS (observed m/z 527.174 (100%); calcd. for C<sub>22</sub>H<sub>32</sub>O<sub>13</sub>Na, m/z 527.174), suggesting it was composed of an iridoid aglycone, a hexose and a pentose unit. At this stage kinginoside (1) was thought to be one analogue of pericymenoside as reported in the literature<sup>4)</sup>, C<sub>33</sub>H<sub>44</sub>O<sub>18</sub>, which had been isolated from a species of genus *Lonicera*. But the following structural work indicated that they belong to different structural types.

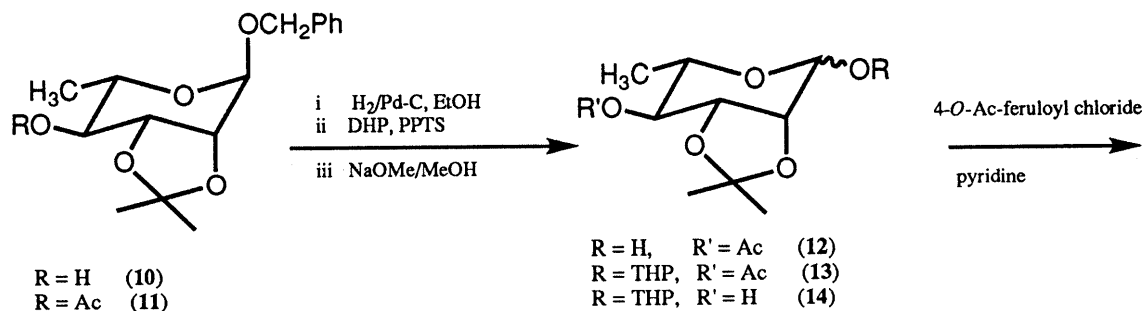
<sup>13</sup>C-NMR spectra of kinginoside and its derivatives showed the molecule to consist of sweroside aglycone, a rhamnoglucoside moiety, and a feruloyl group. Thus the <sup>13</sup>C-NMR signals of iridoid part of 1 are almost superimposable on those of the aglycone part of sweroside (5). The chemical shifts of the carbons C1'-C4' of the glucose part of 1 were also quite similar to those of sweroside (5) but C5' ( $\delta$  77.2) and C6' ( $\delta$  67.4) of 1 were shown to be highfield and lowfield shifts, respectively, compared to the corresponding signals of the glucose part of 5 (5; C5'  $\delta$  78.1; C6'  $\delta$  62.6). This observation

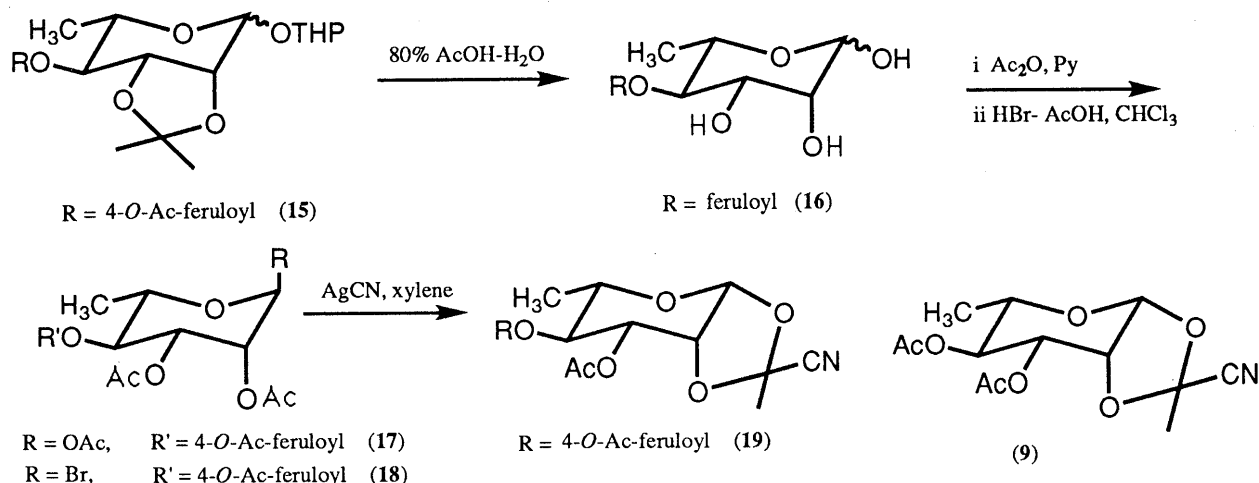
clearly indicated that the second sugar in **1** is linked at the C6' position of the glucose part of **5**. Careful inspection of  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR suggested that the second sugar was rhamnose and the glycoside linkage was  $\alpha$ -oriented. Furthermore, the feruloyl group was suggested to be at the C4'' position by the observed downfield shift of the  $^{13}\text{C}$  NMR signal due to C-4'' (C4'' in **1**;  $\delta$  75.6; C-4 in methyl  $\alpha$ -rhamnoside;  $\delta$  73.9). Thus the structure shown below was deduced to **1**:



To prove the correctness of this structure assignment, a synthetic study was undertaken. First planned was condensation between 2',3',4'-tri-*O*-acetylsweroside (**6**) and nitril (**9**). To prepare **6**, the primary hydroxyl group in **5** was tritylated to **7** and then **7** was acetylated to **8**. Removal of the trityl group of **8** with HCl (AcCl-MeOH), however, resulted in the formation of a 3:1 mixture of desired **6** and acyl migration products. Therefore direct condensation between **8** and **9** using Betanelli's method<sup>5</sup> was attempted. Thus **9**, prepared from L-rhamnose through 3 steps, was heated with **8** in methylene chloride in the presence of triphenylmethylium perchlorate and the expected condensation product (**4**) was obtained with the yield of 33 %.

Then the synthesis of a compound having a feruloyl moiety at the correct position was undertaken. Thus 1- $\alpha$ -*O*-benzyl-L-rhamnose-2,3-*O*-acetonide (**10**)<sup>6</sup> was acetylated to **11**. The resulting benzylether was converted to THP derivative (**13**) by use of dihydropyran (DHP) and pyridinium *p*-toluenesulfonate (PPTS). After introducing feruloyl group to C-4 position in acetonide (**14**), THP and isopropylidene groups were removed with 80 % AcOH-H<sub>2</sub>O at room temperature and the product (**16**) was acetylated to give **17**. Treatment of **17** with HBr followed by refluxing in xylene with AgCN afforded the desired nitrile (**19**). Condensation between **8** and **19** was performed by stirring the solution of the two compounds in dry





methylene chloride in the presence of triphenylmethylm perchlorate for 66 h. at room temperature. Final treatment of the resulting product with  $\text{Ac}_2\text{O}$ , Py. gave rise to the desired **2** (32%), which was shown to be identical (EI-MS, IR,  $^1\text{H-NMR}$ ) with kinginoside hexaacetate (**2**) derived from natural **1**.

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#### REFERENCES AND NOTES

- 1) a) I. Souzu, H. Mitsuhashi, *Tetrahedron Lett.*, **1969**, 2725; *idem, ibid.*, **1970**, 191.  
b) H. Inouye, S. Tobita, Y. Akiyama, K. Ito, T. Shingu, *Chem. Pharm. Bull.*, **21**, 846 (1973).
- 2) Y. Ikeshiro, Y. Toda, I. Mase, Y. Tomita, S. Tanaka, W. H. M. W. Herath, *Planta Medica*, **109**, 58 (1992).
- 3) A colorless amorphous powder,  $\text{C}_{32}\text{H}_{40}\text{O}_{16}$ , HR-FAB-MS,  $m/z$ ; Found, 681.239 ( $\text{M} + \text{H}$ ) $^+$ , Calcd. for  $\text{C}_{32}\text{H}_{41}\text{O}_{16}$ , 681.239. IR  $\nu_{\text{max}}$ (KBr): 3700 - 3100 (OH); 1710, 1695 (C=O); 1620 (C=C); 1515  $\text{cm}^{-1}$ .  $^1\text{H-NMR}$  (270 MHz,  $\text{CD}_3\text{OD}$ )  $\delta$ : 1.16 (3H, d,  $J = 6.4$  Hz, H<sub>3-6</sub>"), 1.55 - 1.75 (3H, m, H-5, 6a, 6b), 2.60 - 2.75 (m, H-9), 3.23 (dd,  $J = 7.9, 9.5$  Hz, H-2'), 3.23 (dd,  $J = 7.9, 7.9$  Hz, H-4'), 3.3 - 3.4 (overlapped with solvent peak, H-3', 5', 2", 3"), 3.75 (dd,  $J = 11.0, 4.9$  Hz, H-6'a), 3.89 (3H, s, H-fer-OMe), 3.95 (m, H-5"), 4.00 (dd,  $J = 11.0, 2.0$  Hz, H-6'b), 4.2 - 4.5 (2H, m, H-7a, 7b), 4.70 (d,  $J = 7.9$  Hz, H-1'), 4.79 (brs, H-1"), 5.04 (dd,  $J = 9.5, 9.5$  Hz, H-4"), 5.22 - 5.30 (2H, m, H-10a, 10b), 5.44 (d,  $J = 1.5$  Hz, H-1), 5.48 - 5.56 (m, H-8), 6.39 (d,  $J = 15.9$  Hz, H-fer-2), 6.82 (d,  $J = 8.2$  Hz, H-fer-8), 7.09 (dd,  $J = 8.2, 1.5$  Hz, H-fer-9), 7.20 (d,  $J = 1.5$  Hz, H-fer-5), 7.58 (d,  $J = 2.4$  Hz, H-3), and 7.65 (d,  $J = 15.9$  Hz, H-fer-3).  $^{13}\text{C-NMR}$  (67.8 MHz,  $\text{CD}_3\text{OD}$ )  $\delta$ : 18.3 (C-6"), 25.1 (C-6), 27.8 (C-5), 43.1 (C-9), 56.6 (C-fer-OMe), 67.4 (C-6'), 67.9 (C-7), 70.4 (C-5"), 71.2 (C-4'), 72.2 (C-2"), 74.7 (C-3"), 74.8 (C-2'), 75.6 (C-4"), 77.2 (C-5'), 78.3 (C-3'), 97.8 (C-1), 100.9 (C-1'), 102.3 (C-1"), 105.3 (C-4), 111.5 (C-fer-5), 116.8 (C-fer-2), 116.8 (C-fer-8), 120.2 (C-10), 126.5 (C-fer-9), 127.1 (C-fer-4), 132.6 (C-8), 145.7 (C-fer-3), 145.7 (C-fer-7), 151.2 (C-fer-6), 152.5 (C-3), 165.0 (C-11), 167.4 (C-fer-CO).
- 4) L. Calis, M. F. Lahloub, O. Sticher, *Helv. Chim. Acta*, **67**, 160 (1984).
- 5) V. I. Betaneli, M. V. Ovchinnikov, L. V. Backinowsky, N. Y. Kotchetkov, *Carbohydr. Res.*, **76**, 252 (1979).
- 6) J. S. Brimacombe, L. C. Tucker, *Carbohydr. Res.*, **5**, 36 (1967).

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