

Tannins of Tamaricaceous Plants. V.¹⁾ New Dimeric, Trimeric and Tetrameric Ellagitannins from *Reaumuria hirtella*

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A new hydrolyzable tannin dimer [hirtellin G (1)], a trimer [hirtellin T₁ (5)] and a tetramer [hirtellin Q₁ (6)] have been isolated from the 70% acetone extract of the leaves of *Reaumuria hirtella* JAUB. et SP. (Tamaricaceae). These tannins were characterized as oligomers composed of two, three or four molecules of tellimagrandin II which are similarly linked with each other through one, two or three dehydrodigalloyl (*m*-GOG) units. Their structures including the orientations of the *m*-GOG units have been elucidated based on chemical degradation and spectroscopic analyses.

Keywords ellagitannin; oligomeric hydrolyzable tannin; *Reaumuria hirtella*; Tamaricaceae; hirtellin T₁; hirtellin Q₁

We previously reported the isolation of new hydrolyzable tannin monomers (remurins A and B) and dimers (hirtellins A–F and tamarixinins A–C)^{1,2)} having a GOG³⁾ (dehydrodigalloyl=DHDG=*m*-GOG and/or iso-DHDG=*p*-GOG) group in each molecule, from Tamaricaceous plants, *Reaumuria hirtella* JAUB. et SP. and *Tamarix pakistanica* QUAISER (Tamaricaceae). Some of them exhibited significant biological activities, such as promotion of iodination in leucocytes,⁴⁾ and host-mediated antitumor activity against sarcoma-180 in mice.⁵⁾ Further investigation of the tannin constituents of *R. hirtella* has resulted in the isolation of three additional new oligomeric hydrolyzable tannins, named hirtellin G, hirtellin T₁ and hirtellin Q₁. This paper deals with the isolation and structural elucidation of these new oligomers.

Repeated column chromatographies of the EtOAc-soluble portion of the aqueous acetone homogenate of the dried leaves^{2c)} over Toyopearl HW-40 and MCI-gel CHP-20P gave hirtellin G (1). Hirtellin T₁ (5) and hirtellin

Q₁ (6) were obtained from the *n*-BuOH extract as described in Experimental.

These new compounds showed positive coloration with the FeCl₃ and NaNO₂-AcOH reagents⁶⁾ on a TLC plate, and were characterized as ellagitannin oligomers structurally related to hirtellin A,^{2b)} as described below.

Structure of Hirtellin G (1) Hirtellin G (1) showed the [M+Na]⁺ ion peak at *m/z* 1595 in the FAB-MS, which is 302 mass units [corresponding to a hexahydroxydiphenyl (HHDP) group] less than that of hirtellin A (3). Methylation of 1 with diazomethane followed by methanolysis with sodium methoxide gave methyl tri-*O*-methylgallate (7), dimethyl hexamethoxydiphenate (8) and dimethyl penta-*O*-methyldehydrodigallate (9). The ¹H-NMR spectrum of 1 showed two sets of seven-spin system signals due to *C*1 glucopyranose as revealed by ¹H-¹H shift correlation spectroscopy (COSY) (Table I). The presence of a *m*-GOG³⁾ unit and an HHDP and four galloyl groups as constituent phenolic units in 1 was

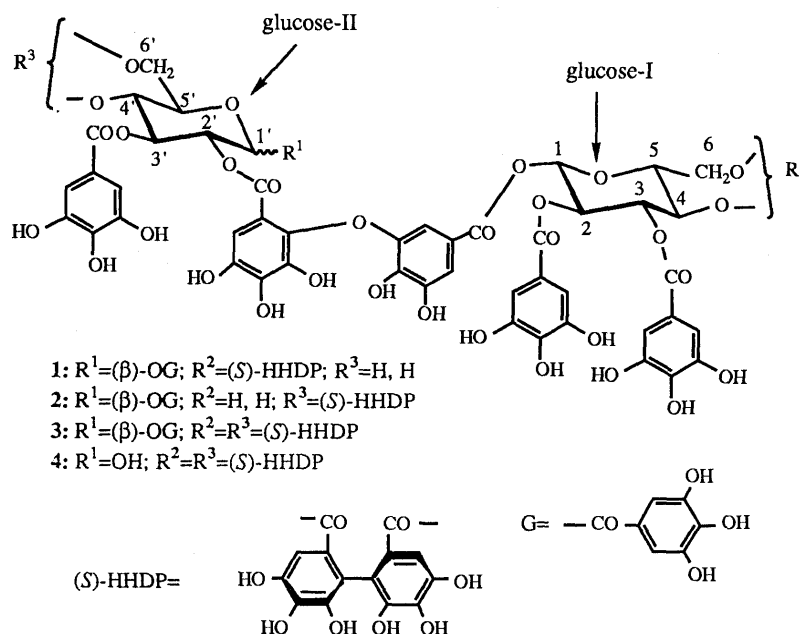


Chart 1

indicated by three 1H-singlets, two meta-coupled 1H doublets, two 2H-singlets and one 4H-singlet; in the aromatic region from δ 6.39 to 7.25 ppm. Upon $^1\text{H-NMR}$ spectral comparison of **1** and **3** (Table I), the chemical shifts and coupling pattern of the proton signals of a glucose core (glucose-I) in these compounds were similar to each other, while the H-4'—6' signals (δ 3.55—3.86) of the other glucose core (glucose-II) in **1** were shifted significantly upfield ($\Delta\delta$ ca. 1.4) relative to those of **3**. These upfield shifts implied the absence of an HHDP group at O-4' and O-6' of glucose-II in **1**. Hirtellin G (**1**) was thus assumed to be an analog of hirtellin A (**3**) lacking an HHDP group on the glucose-II. The structure **1** for hirtellin G was chemically supported by partial acid hydrolysis of hirtellin A (**3**), which afforded dimeric hydrolyzates **1** and **2**, besides monomeric hydrolyzates **11**, **14** and **17**.^{2b)} Hydrolyzate **2** was characterized as a regioisomer of **1** concerning the position of HHDP group, as follows. The $^1\text{H-NMR}$ spectrum of **2** showed the presence of two *C1* glucopyranose cores (Table I), a *m*-GOG unit, an HHDP and four galloyl groups as found in **1**. The chemical shifts of glucose-II proton signals in **2** were almost identical with those of hirtellin A (**3**), whereas the H-4—H-6 signals (δ 3.81—4.02) of glucose-I showed large upfield shifts relative to those of **3**, indicating the absence of an HHDP group at O-4 and O-6 of glucose-I.

The characteristic upfield shift of the anomeric proton of glucose-II in **1** (δ 5.24) relative to that of 1,2,3-tri-*O*-galloyl- β -D-glucose⁷⁾ (δ 6.02) showed that the binding mode (orientation) of the *m*-GOG unit in **1** is the same as that of hirtellin A (**3**).^{2b)} Based on these data, the structure of hirtellin G is represented as **1**.

Structure of Hirtellin T₁ (5) Hirtellin T₁ (**5**) was established as a trimer composed of hirtellin A (**3**) and tellimagrandin II (**10**), which are linked with each other forming a *m*-GOG unit, as follows. The trimeric nature of hirtellin T₁ was shown by stronger adsorption on Toyopearl HW-40 gel as compared with **3** and other dimers, and by its retention time, which was similar to that of trimers,⁸⁾ in normal-phase HPLC. Methylation of **5** with dimethyl sulfate and potassium carbonate in dry acetone afforded the permethylated derivative **5a**. Upon methanolysis with sodium methoxide, **5a** gave **7**, **8** and **9**, which were identical with those obtained by similar treatment of hirtellin A (**3**).^{2b)} The sugar liberated upon complete acid hydrolysis of **5** was identified as glucose by GLC after trimethylsilylation. The ^1H and $^{13}\text{C-NMR}$ spectra of **5** indicated the presence of three glucose residues in the molecule. The chemical shifts and coupling patterns of the glucose proton signals, which were assigned as shown in Table II by $^1\text{H-}^1\text{H}$ COSY, suggested that all the three glucose cores adopt the *C1* conformation, and are fully acylated. The $^1\text{H-NMR}$ spectrum also displayed five 2H singlets at δ 7.02, 7.00, 6.95, 6.95 and 6.91, eight 1H singlets and two pairs of 1H doublets ($J=2$ Hz) in the aromatic region. These data, together with the fifteen ester carbonyl carbon signals (δ 163.71—168.31) in the $^{13}\text{C-NMR}$ spectrum, indicated that **5** is a trimeric ellagitannin composed of five galloyl, two *m*-GOG and three HHDP groups, and three glucose cores, and that its molecular formula corresponds to $\text{C}_{123}\text{H}_{86}\text{O}_{78}$. The

presence of the HHDP group at O-4 and O-6 of each glucopyranose core was evident from the large difference among the chemical shifts of each C-6 methylene proton⁹⁾ ($\Delta\delta$ ca. 1.4 ppm), as also seen in **3**. The absolute configuration at the HHDP groups of **5** was determined to be the same as that of hirtellin A (**3**)^{2b)} (*S*-configuration) on the basis of the levorotatory optical activity of **8**, and the strong positive Cotton effect at 235 nm ($[\theta] + 34.7 \times 10^4$) in the CD spectrum.¹⁰⁾ Based on these data, coupled with the fact that the chemical shifts of glucose carbon signals in the $^{13}\text{C-NMR}$ spectrum of **5** are closely similar to those of **10**¹¹⁾ and **3** (Table III), hirtellin T₁ (**5**) was regarded as a trimer, biogenetically produced by intermolecular C-O oxidative coupling between **10** and **3** forming a *m*-GOG unit. The orientation of the two *m*-GOG units, which link the three glucose moieties in **5**, was deduced from the following observations. The hydrolyzable tannins possessing a *m*-GOG unit at O-2 of *C1* glucopyranose generally show a remarkable upfield shift of the anomeric proton due to the anisotropic effect of the aromatic ring of the *m*-GOG unit.^{2b)} This characteristic anomaly is useful for determining the position of the *m*-GOG unit. In the $^1\text{H-NMR}$ spectrum of **5**, the proton signals of a glucose core (glucose-I) were in agreement with those of **10** or of glucose-I in **3** (Table I), while those of the other two glucoses (II and III), including the characteristic upfield shifts (δ 5.77, 5.50) of the anomeric protons (H-1', H-1'') were closely similar to those of glucose-II of **3**. Hirtellin T₁ (**5**) was thus concluded to have two *m*-GOG units with the same binding mode as that of **3**, *i.e.*, the *m*-GOG units bind three glucoses in the following sequence; O-1—O-2' and O-1'—O-2''. Further chemical evidence in support of this linking mode was obtained by partial degradation of **5** in hot water, which afforded **11**, **14** and **17** together with a new compound **18**, as major hydrolyzates (Chart 5). The $^1\text{H-NMR}$ spectrum of the hydrolyzate **18** showed two sets of fully acylated glucose proton signals, of which one (H-1—6) was similar to that of remurin A (**14**).^{2b)} The other set of signals (H-1'—6') was similar to that of glucose-II of hirtellin A (**3**)^{2b)} (Table I). The presence of two *m*-GOG, two HHDP and three galloyl groups in **18** was shown by three 2H singlets, six 1H singlets and two pairs of 1H doublets ($J=2$ Hz) in the aromatic region (δ 6.40—7.19). The significant upfield shifts of the anomeric protons H-1 (δ 5.51—5.55) and H-1' (δ 5.38) in **18** are consistent with the presence of the *m*-GOG unit at O-2 of each glucose. This hydrolyzate was thus characterized as a dimer represented by formula **18**. Based on these data, the structure **5** was assigned to hirtellin T₁.

Structure of Hirtellin Q₁ (6) Hirtellin Q₁ (**6**) showed a longer retention time than the trimer **5** in normal-phase HPLC,⁸⁾ suggesting its tetrameric nature. Methylation of **6** with diazomethane followed by methanolysis with sodium methoxide gave **7**, **8** and **9**, which are the same compounds as those obtained from hirtellin A (**3**) and hirtellin T₁ (**5**). The sugar component liberated upon acid hydrolysis was identified as glucose. The $^1\text{H-NMR}$ spectrum (Table II) of **6** showed the presence of six galloyl, four HHDP groups and three *m*-GOG units, as well as four glucose cores, which were supported by the $^{13}\text{C-NMR}$

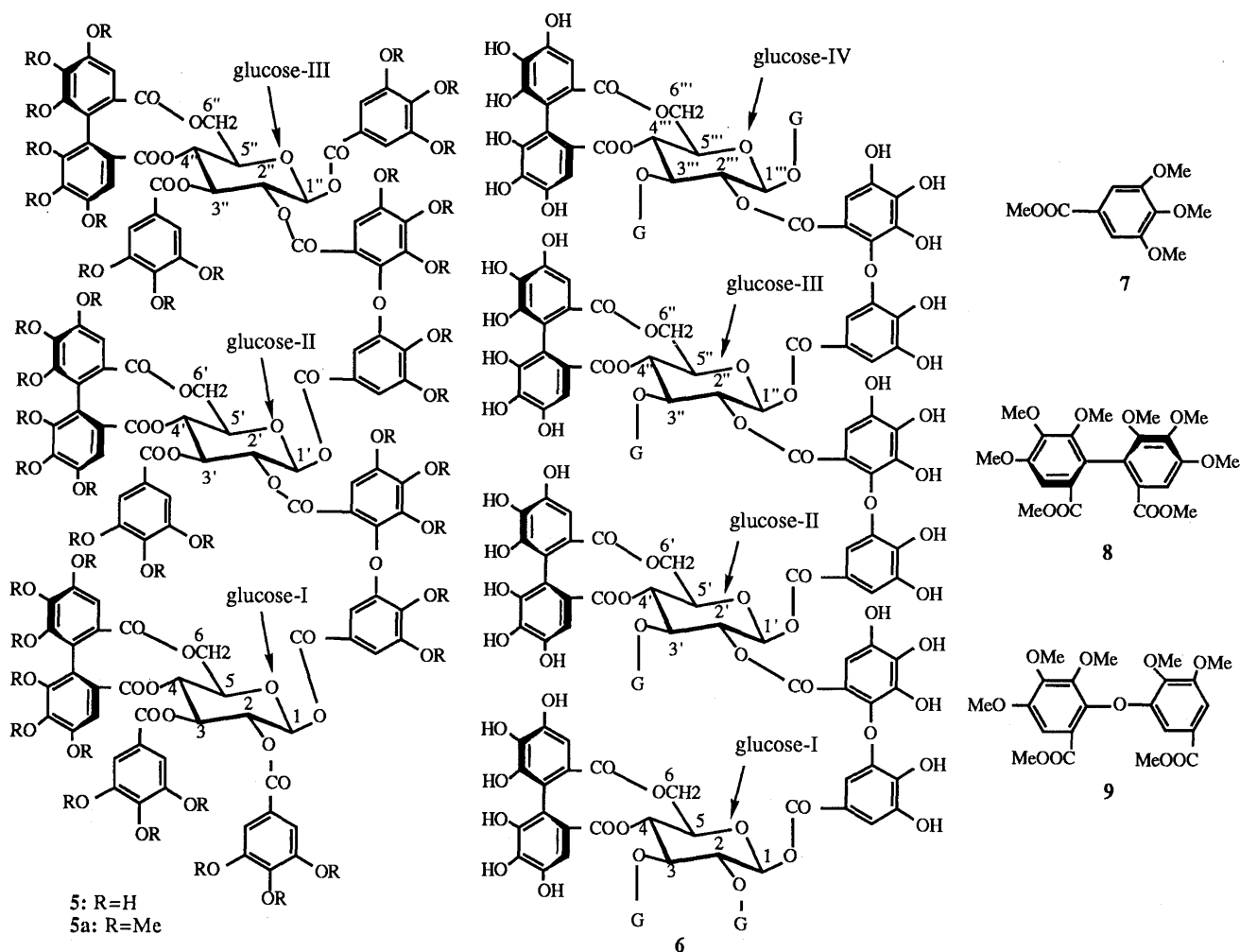


Chart 2

spectrum exhibiting four sets of glucose carbon signals closely similar to those of hirtellin A (3) and hirtellin T₁ (5) (Table III), and twenty ester carbonyl carbon signals at δ 163.71–168.31. Moreover, the sugar proton signals in the ¹H-NMR spectrum of 6 were in agreement with the sum of those of 5 and glucose-II of 3 (Tables I and II). The orientations of the *m*-GOG units in 6 were assumed to be the same as those in 3 or 5 on the basis of characteristic upfield shifts ($\Delta\delta$ 0.37–0.76) of the three anomeric protons (H-1'–H-1'''). The CD spectrum of 6 showed a strong positive Cotton effect at 233 nm ($[\theta] + 47.2 \times 10^4$), of the same sign as that of 10⁹) but with an amplitude about 4 times larger, indicating the presence of four (*S*)-HHDP groups in the molecule.¹⁰) The partial hydrolysis of 6 with hot water gave seven monomeric (11, 12, 13, 14, 15, 16 and 17), two dimeric (4 and 18) and a trimeric (19) hydrolyzates (Chart 5). The monomeric hydrolyzates, except for 16, were identified by comparison with authentic samples. Hydrolyzate 16 exhibited, in the aromatic region of the ¹H-NMR spectrum, two 2H singlets, two 1H doublets ($J=2$ Hz) and one 1H singlet attributable to two galloyl groups and a *m*-GOG unit. Upon the ¹H-NMR spectral comparison of the glucose signals between 16 and 14, the shifts of H-1–3 signals were analogous, whereas the H-4–6 signals in the former

were shifted markedly upfield from those of the latter. Therefore, 16 was considered as an analog of remurin A (14) lacking the HHDP group, and this correlation was confirmed by partial degradation of 14 in hot water, yielding 16. Hydrolyzate 4 showed in the FAB-MS the ($M+Na$)⁺ ion peak at m/z 1745, which is 152 (galloyl group) mass units lower than that of hirtellin A (3). The presence of three galloyl groups in 4 was shown by the ¹H-NMR signals at δ 6.99, 6.98, 6.95 and 6.94 (each s, 5H in total), although these signals were complicated by duplication of each proton signal due to the formation of a mixture of α - and β -anomers. A series of singlets (5H in total) at δ 6.45–6.64 and δ 7.09–7.08, and two pairs of *meta*-coupled doublets at δ 7.23, 7.21, 6.68, and 6.67 (each d, $J=2$ Hz, 2H in total) are attributable to two HHDP groups and a *m*-GOG unit. The chemical shifts of the anomeric protons of 4 [δ 6.11, 6.10 (α - and β -anomer), 5.36 (α -anomer) and 4.45 (β -anomer)] (Table I) clearly indicate that one pair of the anomeric centers is acylated and the other is free. As the signals of the fully acylated glucose core (glucose-I) in 4 are closely similar to the corresponding signals of 3, the hydrolyzate 4 is regarded as a degalloylated congener of 3, and this structure was confirmed by enzymatic hydrolysis of 3 with tannase, yielding compound 4. The other dimeric hydrolyzate 18

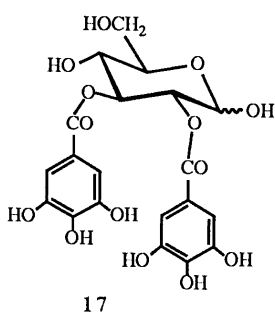
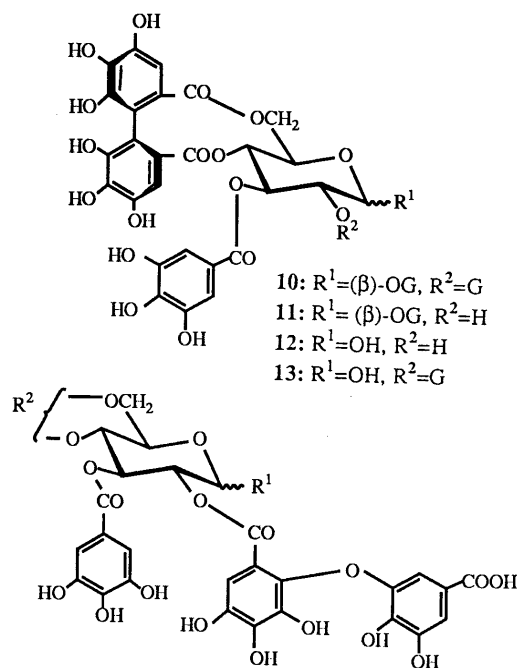


Chart 3

was identical with the product obtained by partial hydrolysis of **5**, mentioned above. Compound **19** was suggested to be a trimeric hydrolyzate on the basis of its behavior on normal-phase HPLC, and the following $^1\text{H-NMR}$ spectral data. The presence of three *m*-GOG units, three HHDP and four galloyl groups was shown by four 2H singlets, nine 1H singlets and three pairs of 1H doublets ($J=2\text{ Hz}$) in the aromatic region ($\delta\ 6.45\text{--}7.40$). Three sets of fully acylated glucose proton signals were exhibited in the aliphatic region, and their chemical shifts and coupling patterns were almost identical with those of remurin A (**14**). The characteristic upfield shifts of the three anomeric protons, H-1 ($\delta\ 5.45\text{--}5.54$), H-1' ($\delta\ 5.81$) and H-1'' ($\delta\ 5.31$), indicated the presence of three *m*-GOG units at O-2 of each glucose core. Compound **19** was thus characterized as a trimer produced by removal of the terminal tellimagrandin I moiety from **6**. The tetrameric ellagitannin structure of hirtellin Q₁, as represented by **6**, was thus established. Hirtellin Q₁ (**6**) is regarded as a tetramer biogenetically producible by C–O oxidative

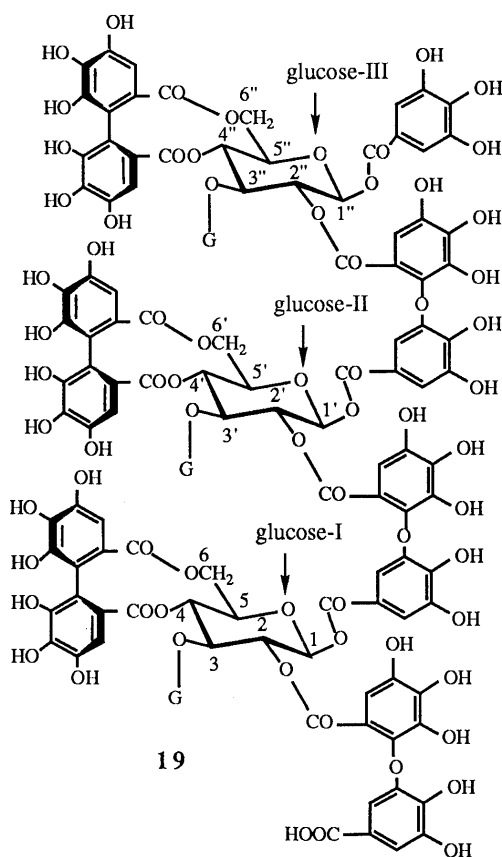
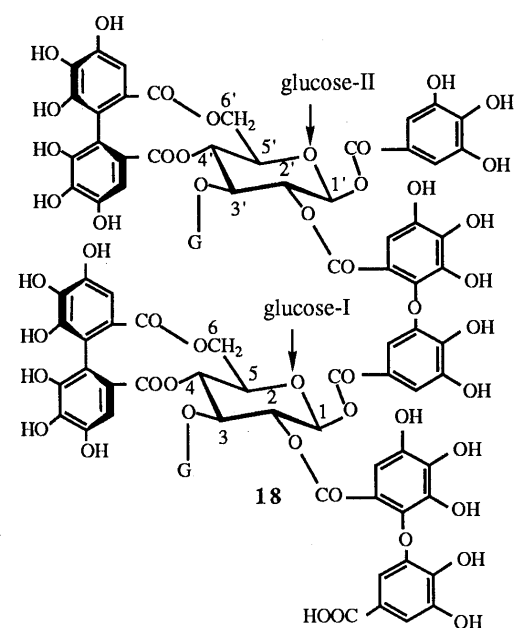
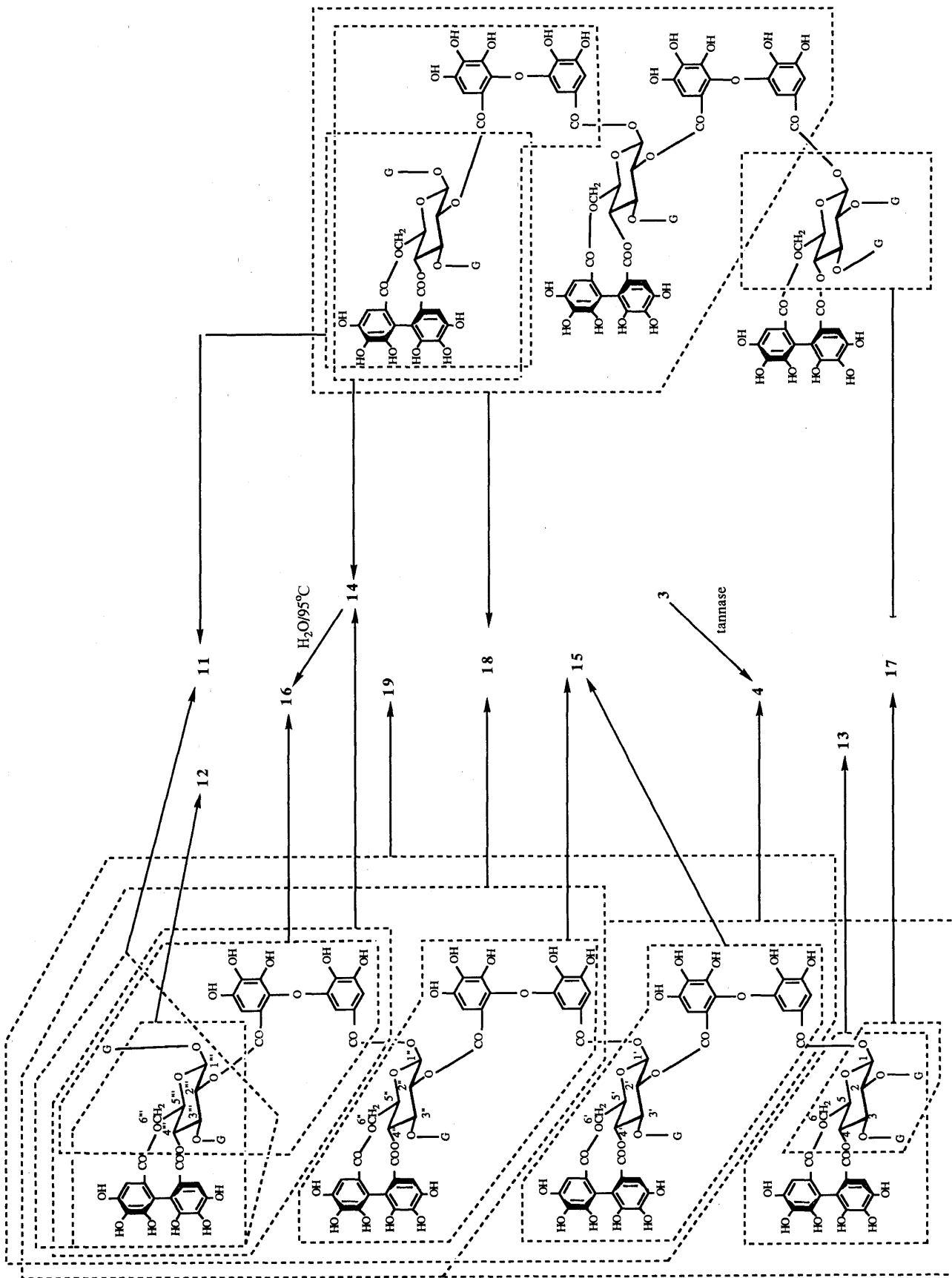


Chart 4

coupling between the galloyl groups at glucose O-1'' of **5** and O-2 of **10**.

It is noteworthy that the dimers (hirtellins and tamarixinins)^{1,2)} previously isolated from Tamaricaceae plants, and the oligomers presented in this paper, have common structural features, with the *m*-GOG and/or *p*-GOG units³⁾ between monomeric constituents at O-1 of a C1 glucopyranose core and O-2 of the other C1 glucopyranose. These oligomers may be useful as chemo-



5

Chart 5

6

TABLE I. ¹H-NMR Spectral Data for Glucose Moieties of **1**, **2**, **3**, **4**, **10**, **14** and **18** (500 MHz, Acetone-*d*₆+D₂O, *J* in Hz)

H	1	2	3	4		10	14	18
				α-Anomer	β-Anomer			
Glucose-I								
1	6.08 d (<i>J</i> =8)	5.93 d (<i>J</i> =8)	6.07 d (<i>J</i> =8)	6.11, 6.10 each d (<i>J</i> =8)		6.20 d (<i>J</i> =8)	5.49—5.54 Overlapped	5.51—5.55 Overlapped
2	5.59 dd (<i>J</i> =8, 10)	5.36 dd (<i>J</i> =8, 10)	5.58 dd (<i>J</i> =8, 10)	5.55, 5.54 each dd (<i>J</i> =8, 10)		5.58 dd (<i>J</i> =8, 9.5)		
3	5.76 t (<i>J</i> =10)	5.54 t (<i>J</i> =10)	5.77 t (<i>J</i> =10)	5.79 t (<i>J</i> =10)		5.75 t (<i>J</i> =9.5)		
4	5.22 t (<i>J</i> =10)	3.91 t (<i>J</i> =10)	5.22 t (<i>J</i> =10)	5.17 t (<i>J</i> =10)		5.14 t (<i>J</i> =9.5)	5.14 t (<i>J</i> =10)	5.14 t (<i>J</i> =10)
5	4.50 dd (<i>J</i> =6.5, 10)	3.81 ddd (<i>J</i> =2, 6.5, 10)	4.53 dd (<i>J</i> =6, 10)	4.48 dd (<i>J</i> =6.5, 10)		4.41 dd (<i>J</i> =5.5, 9.5)	4.22 dd (<i>J</i> =6.5, 10)	4.23 dd (<i>J</i> =6.5, 10)
6	5.33 dd (<i>J</i> =6.5, 13.5)	3.85 dd (<i>J</i> =6.5, 12)	5.33 dd (<i>J</i> =6, 13)	5.29 dd (<i>J</i> =6.5, 13.5)		5.33 dd (<i>J</i> =5.5, 13.5)	5.31 dd (<i>J</i> =6.5, 13.5)	5.30 dd (<i>J</i> =6.5, 13.5)
	4.00 d (<i>J</i> =13.5)	4.02 dd (<i>J</i> =2, 12)	3.98 d (<i>J</i> =13)	3.85, 3.84 each d (<i>J</i> =13.5)		3.80 d (<i>J</i> =13.5)	3.81 d (<i>J</i> =13.5)	3.82 d (<i>J</i> =13.5)
Glucose-II								
1'	5.24 d (<i>J</i> =8)	5.45 d (<i>J</i> =8.5)	5.35 d (<i>J</i> =8)	5.36 d (<i>J</i> =4)	4.45 d (<i>J</i> =8)			5.38 d (<i>J</i> =8.5)
2'	5.30 dd (<i>J</i> =8, 10)	5.52 dd (<i>J</i> =8.5, 10)	5.55 dd (<i>J</i> =8, 10)	4.99 dd (<i>J</i> =4, 10)	5.14 dd (<i>J</i> =8, 10)			5.44 dd (<i>J</i> =8.5, 10)
3'	5.42 t (<i>J</i> =10)	5.63 t (<i>J</i> =10)	5.64 t (<i>J</i> =10)	5.81 t (<i>J</i> =10)	5.53 t (<i>J</i> =10)			5.67 t (<i>J</i> =10)
4'	3.87 t (<i>J</i> =10)	5.13 t (<i>J</i> =10)	5.13 t (<i>J</i> =10)	5.06 t (<i>J</i> =10)	5.05 t (<i>J</i> =10)			5.14 t (<i>J</i> =10)
5'	3.52, overlapped	4.25 ddd (<i>J</i> =1, 6.5, 10)	4.22 dd (<i>J</i> =6.5, 10)	4.60 dd (<i>J</i> =6.5, 10)	4.05 dd (<i>J</i> =6.5, 10)			4.24 dd (<i>J</i> =6.5, 10)
6'	3.74 dd (<i>J</i> =4.5, 12.5)	5.29 dd (<i>J</i> =6.5, 13.5)	5.29 dd (<i>J</i> =6.5, 13)	5.32 dd (<i>J</i> =6.5, 13)	5.25 dd (<i>J</i> =6.5, 13.5)			5.28 dd (<i>J</i> =6.5, 13)
	3.86 dd (<i>J</i> =2, 12.5)	3.81, overlapped	3.82 d (<i>J</i> =13)	3.73 d (<i>J</i> =13)	3.80 d (<i>J</i> =13.5)			3.92 d (<i>J</i> =13)

taxonomic markers of Tamaricaceous plants.

Experimental

General The material (*R. hirtella*), instruments and chromatographic methods in this work were the same as those described in the preceding paper.^{2c)} Preparative HPLC (R) was carried out on YMC A312 (ODS) (6 mm i.d. × 150 mm) and YMC A324 (ODS) (10 mm i.d. × 300 mm) (Yamamura Kagaku, Japan) at 40°C, using 0.01 M KH₂PO₄–0.01 M H₃PO₄–EtOH–EtOAc (41.5:41.5:12:5) as a mobile phase at a flow rate of 1.3 ml/min (detection, Union-Giken MCPD-350 at 280 or 300 nm). Normal-phase HPLC was carried out on a Superspher Si 60 cartridge column (4 mm i.d. × 125 mm) (Merck) at room temperature using the following solvent systems: (A) *n*-hexane–EtOAc (2:1); (B) *n*-hexane–MeOH–THF–HCOOH (60:45:15:1) and oxalic acid (450 mg/l), at a flow rate of 1.0 ml/min (detection: Shimadzu SPD-6A UV detector at 280 nm).

Isolation of Tannins An EtOAc-soluble portion (28.0 g) obtained from the 70% acetone extract of *R. hirtella* leaves was chromatographed over Toyopearl HW-40 (coarse) using 70% MeOH → MeOH–H₂O–acetone (7:2:1 → 6:2:2) as the eluant, as described in the previous paper.^{2c)} The earlier fractions eluted with MeOH–H₂O–acetone (6:2:2) gave a mixture of two minor dimeric tannins which were separated by rechromatography over MCI-gel CHP-20P with H₂O containing increasing amounts of MeOH to yield hirtellin F^{2c)} and a new tannin, hirtellin G (**1**) (12.1 mg).

A part (35 g) of the *n*-BuOH extract was subjected to column chromatography over Toyopearl HW-40 (coarse),^{1c)} using aqueous MeOH (70% → 80% MeOH) and MeOH–H₂O–acetone (7:2:1 → 6:2:2 → 5:2:3) as eluants to afford remurin B (**15**), and hirtellins D, E, B and A. A mixture of trimers (4.3 g) and a crude tetramer (1.4 g) were obtained from the MeOH–H₂O–acetone (5:2:3) eluate. A part of the former (660 mg) was subjected to repeated column chromatography over MCI-gel CHP-20P and/or preparative HPLC (R) to yield hirtellin T₁ (**5**) (250 mg) as the major trimer. The crude tetramer was also purified by column chromatography over MCI-gel CHP-20P followed by preparative HPLC (R) to afford hirtellin Q₁ (**6**) (166 mg).

Hirtellin G (1) An off-white amorphous powder, FAB-MS *m/z*: 1595 (M+Na)⁺, [α]_D+57° (*c*=1, MeOH). UV λ_{max}^{MeOH} nm (log ε): 216 (5.11), 276 (4.75). CD (MeOH) [θ] (nm): +16 × 10⁴ (228), –5.9 × 10⁴ (264), +2.5 × 10⁴ (302). ¹H-NMR (acetone-*d*₆+D₂O): δ 7.05, 6.95 (2H each,

s, Gal × 2), 6.97 (4H, s, Gal × 2), 6.67, 6.48 (1H each, s, HHDP), 7.25, 6.39 (1H each, d, *J*=2 Hz, *m*-GOG), 7.03 (1H, s, *m*-GOG); glucose protons, see Table I.

Methylation of 1 Followed by Methanolysis A solution of **1** (1.3 mg) in MeOH (0.5 ml) was methylated with ethereal CH₂N₂ at room temperature for 5 h. The residue after removal of the solvent was directly methanolized with 1% NaOMe (50 μl) in MeOH (0.2 ml) overnight at room temperature. The reaction mixture was worked up in a similar way to that described for methanolysis of hirtellin D,^{1c)} to give **7**, **8** and **9**, which were shown to be identical with authentic samples by TLC and normal-phase HPLC (solvent A). The identity of **9** was further confirmed by EI-MS spectral comparison (*M*⁺, *m/z* 436) with authentic material.

Partial Hydrolysis of Hirtellin A (3) An aqueous solution of **3** (100 mg) was heated at 95°C, and the reaction mixture was submitted to column chromatography over MCI-gel CHP 20P, developing with water containing increasing amounts of MeOH. The fractions eluted with 30% and 40% MeOH gave **1** (1.5 mg) and **2** (1.1 mg), respectively. The product **1** was identified as hirtellin G by co-chromatography with the natural product using normal-phase HPLC (solvent B) and reversed-phase HPLC.^{2c)}

2: An off-white amorphous powder, ¹H-NMR (acetone-*d*₆+D₂O): δ 7.04, 7.00 [2H each, s, galloyl (Gal) × 2], 6.94 (4H, s, Gal × 2), 6.59, 6.48 (1H each, s, HHDP), 7.14, 6.44 (1H each, d, *J*=2 Hz, *m*-GOG), 7.04 (1H, s, *m*-GOG); glucose protons, see Table I.

Hirtellin T₁ (5) An off-white amorphous powder, [α]_D+51° (*c*=1.0, MeOH). Anal. Calcd for C₁₂₃H₈₆O₇₈·13H₂O: C, 48.21; H, 3.66. Found: C, 48.27; H, 3.86. UV λ_{max}^{MeOH} nm (log ε): 223 (5.37), 275 (5.08). CD (MeOH) [θ] (nm): +35 × 10⁴ (235), –8.4 × 10⁴ (260), +8.4 × 10⁴ (285), –0.8 × 10⁴ (333). Normal phase HPLC (solvent B), *t*_R 17.07 min. ¹H-NMR (acetone-*d*₆+D₂O): δ 7.02, 7.00, 6.91 (2H each, s, Gal × 3), 6.67, 6.65, 6.62, 6.49, 6.46, 6.45 (1H each, s, HHDP × 3), 7.25, 7.23, 6.54, 6.45 (1H each, d, *J*=2 Hz), 7.07, 7.05 (1H each, s) (*m*-GOG × 2); glucose protons, see Table II. ¹³C-NMR (acetone-*d*₆+D₂O): δ 107.22 (*m*-GOG C-2), 107.79 (4C) (HHDP C-3, 3', *m*-GOG C-2), 107.96, 108.19, 108.23 (HHDP C-3, C-3'), 110.08 (6C), 110.20 (4C) (Gal C-2, C-6), 110.40 (2C) (*m*-GOG C-6'), 112.60 (*m*-GOG C-6), 113.30 (2C) (*m*-GOG C-1, C-6), 113.52 (*m*-GOG C-1), 115.54 (2C), 115.58, 115.69, 115.73, 115.77 (HHDP C-1, C-1'), 119.55, 119.60 (*m*-GOG C-1'), 119.89, 120.12 (2C), 120.21, 120.29 (Gal C-1), 125.60, 125.71 (2C), 126.26, 126.35 (2C) (HHDP C-2, C-2'), 136.30 (4C), 136.53 (3C), 136.96 (HHDP C-5, C-5', *m*-GOG C-2'), 139.10 (2C), 139.28 (2C), 139.50 (Gal C-4), 140.00, 140.21

TABLE II. ¹H-NMR Spectral Data^{a)} for Glucose Moieties of **5**, **6** and **19** (500 MHz, Acetone-*d*₆ + D₂O, *J* in Hz)

H	5	6	19
Glucose-I			
1	6.12 d (<i>J</i> =8.5)	6.12 d (<i>J</i> =8)	5.45—5.54 Overlapped
2	5.53 dd (<i>J</i> =8.5, 10)	5.53 dd (<i>J</i> =8, 10)	
3	5.78 t (<i>J</i> =10)	5.77 t (<i>J</i> =10)	
4	5.18 t (<i>J</i> =10)	5.19 t (<i>J</i> =10)	5.15 t (<i>J</i> =10)
5	4.48 dd (<i>J</i> =6.5, 10)	4.50 dd (<i>J</i> =7, 10)	4.26 dd (<i>J</i> =6.5, 10)
6	5.30 dd (<i>J</i> =6.5, 13.5)	5.32 dd (<i>J</i> =7, 13)	5.29 dd (<i>J</i> =6.5, 12.5)
	3.94 d (<i>J</i> =13.5)	3.93 d (<i>J</i> =13)	3.83 d (<i>J</i> =12.5)
Glucose-II			
1'	5.77 d (<i>J</i> =8.5)	5.83 d (<i>J</i> =7.5)	5.81 d (<i>J</i> =8)
2'	5.51 dd (<i>J</i> =8.5, 10)	5.51 dd (<i>J</i> =7.5, 10)	5.45—5.54
3'	5.63 t (<i>J</i> =10)	5.63 t (<i>J</i> =10)	5.45—5.54
4'	5.13 t (<i>J</i> =10)	5.13 t (<i>J</i> =10)	5.10 t (<i>J</i> =10)
5'	4.32 ddd (<i>J</i> =7.5, 10)	4.33 dd (<i>J</i> =7, 10)	4.34 dd (<i>J</i> =6.5, 10)
6'	5.29 dd (<i>J</i> =7.5, 13.5)	5.27 dd (<i>J</i> =7, 13)	5.24 dd (<i>J</i> =6.5, 13)
	3.83 dd (<i>J</i> =13.5)	3.94 d (<i>J</i> =13)	3.83 d (<i>J</i> =13)
Glucose-III			
1''	5.50 d (<i>J</i> =8)	5.75 d (<i>J</i> =7.5)	5.31 d (<i>J</i> =8)
2''	5.47 dd (<i>J</i> =8, 10)	5.54 dd (<i>J</i> =7.5, 10)	5.45—5.54
3''	5.52 t (<i>J</i> =10)	5.46 t (<i>J</i> =10)	5.63 t (<i>J</i> =10)
4''	5.14 t (<i>J</i> =10)	5.13 t (<i>J</i> =10)	5.14 t (<i>J</i> =10)
5''	4.25 dd (<i>J</i> =6.5, 10)	4.26 dd (<i>J</i> =7, 10)	4.17 dd (<i>J</i> =6.5, 10)
6''	5.30 dd (<i>J</i> =6.5, 13)	5.30 dd (<i>J</i> =7, 13)	5.31 dd (<i>J</i> =6.5, 13)
	3.89 d (<i>J</i> =13)	3.84 d (<i>J</i> =13)	3.93 d (<i>J</i> =13)
Glucose-IV			
1'''		5.44 d (<i>J</i> =8)	
2'''		5.45 dd (<i>J</i> =8, 10)	
3'''		5.52 t (<i>J</i> =10)	
4'''		5.09 t (<i>J</i> =10)	
5'''		4.29 dd (<i>J</i> =7, 10)	
6'''		5.25 dd (<i>J</i> =7, 13)	
		3.85 d (<i>J</i> =13)	

a) Assignments of **5**, **6** and **19** were based on the COSY, and correspondences with known compounds **1**, **10** and **14**

(2C), 140.30, 140.60 (2C) (*m*-GOG C-4, C-3', C-4'), 143.30, 143.50 (*m*-GOG C-5'), 144.26 (3C), 144.34 (3C) (HHDP C-6, C-6'), 145.10 (6C) (HHDP C-4, C-4'), 145.70 (6C), 145.82 (4C) (Gal C-3), 147.68 (2C), 147.76 (2C) (*m*-GOG C-3, C-5), 163.71, 164.20, 164.42 (2C), 164.71, 165.63, 166.39, 166.41, 166.63, 167.55, 167.63, 167.65, 168.09, 168.13. 168.31 (ester carbonyl carbons); glucose carbons, see Table III.

Acid Hydrolysis of Hirtellin T₁ (5) A solution of **5** (2 mg) in 1 N H₂SO₄ (1 ml) in a sealed ampule was heated on a boiling-water bath for 5 h. After cooling, the solution was extracted with EtOAc. The aqueous layer was neutralized with Amberlite IR-410 (OH-form) and evaporated to give a syrup. The GLC analysis after trimethylsilylation of the syrup showed the sugar component to be glucose.

Methylation of 5 A mixture of **5** (100 mg), anhydrous K₂CO₃ (700 mg) and dimethyl sulfate (1 ml) in dry acetone (20 ml) was stirred overnight at room temperature, and refluxed for 8 h. After removal of the inorganic material by filtration, the filtrate was concentrated *in vacuo*, and subjected to preparative TLC (Kieselgel PF₂₅₄, benzene-acetone 5:1) to give the permethylated derivative of hirtellin T₁ (**5a**) (60.8 mg).

5a: A yellowish-white amorphous powder. ¹H-NMR (acetone-*d*₆): δ 7.28 (1H, d, *J*=2 Hz), 7.26 (2H, s), 7.22 (3H, s), 7.19 (4H, s), 7.18 (2H

TABLE III. ¹³C-NMR Data for the Glucose Moieties of **10**, **3**, **5** and **6**

	C-1	C-2	C-3	C-4	C-5	C-6
	C-1'	C-2'	C-3'	C-4'	C-5'	C-6'
	C-1''	C-2''	C-3''	C-4''	C-5''	C-6''
	C-1'''	C-2'''	C-3'''	C-4'''	C-5'''	C-6'''
10	93.8	71.8	73.3	70.8	73.1	63.1
3	93.5 (2C)	71.5	73.4	70.3	72.9 (2C)	62.9
		71.8	73.2	70.6		63.1
5	93.5	71.4	73.2	70.4	73.0 (3C)	63.0 (3C)
	93.6	71.5	73.3	70.5		
	93.7	71.7	73.4	70.6		
6	93.5 (2C)	71.7	73.3 (3C)	70.5 (3C)	72.9 (4C)	63.1 (2C)
	93.7	71.5	73.5	70.6		63.0 (2C)
	93.6	71.4 (2C)				

overlapped signals), 7.16 (2H, s), 6.94 (1H, s), 6.90 (2H, s), 6.80 (1H, s), 6.76 (1H, s), 6.75 (3H, br d), 6.26 [1H, d, *J*=8.5 Hz, glucose (Glc) H-1], 5.62 (1H, t, *J*=8.5 Hz, Glc H-2), 5.87 (1H, t, *J*=10 Hz, Glc H-3), 5.30 (1H, t, *J*=10 Hz, Glc H-4 or H-4''), 4.61 (1H, dd, *J*=6.5, 10 Hz, Glc H-5), 5.31 (1H, dd, *J*=6.5, 13.5 Hz, Glc H-6), the other H-6 is overlapped by OMe signals, 6.21 (1H, d, *J*=7 Hz, Glc H-1'), 5.66 (1H, t, *J*=7 Hz, Glc H-2'), 5.82 (1H, t, *J*=10 Hz, Glc H-3'), 5.44 (1H, t, *J*=10 Hz, Glc H-4'), 4.47 (1H, m, Glc H-5'), 5.21 (1H, dd, *J*=6, 13.5 Hz, Glc H-6'), the other H-6' is overlapped by OMe signals, 6.10 (1H, d, *J*=7 Hz, Glc H-1''), 5.58 (1H, t, *J*=7 Hz, Glc H-2''), 5.62 (1H, t, *J*=9.5 Hz, Glc H-3''), 5.29 (1H, t, *J*=10 Hz, Glc H-4'' or H-4), 4.29 (1H, dd, *J*=6.5, 10 Hz, Glc H-5''), 5.17 (1H, dd, *J*=6.5, 13.5 Hz, Glc H-6''), the other H-6'' is overlapped by OMe signals, 3.97 (6H, s, OMe × 2), 3.90, 3.88 (each 3H, s, OMe × 2), 3.85, 3.84 (18H each, s, OMe × 12), 3.83, 3.81, 3.80, 3.79 (3H each, s, OMe × 4), 3.75 (15H, s, OMe × 5), 3.73 (6H, s, OMe × 2), 3.72—3.71 (15H, overlapped signals, OMe × 5), 3.69 (3H, s, OMe × 1), 3.67, 3.65 (6H each, s, OMe × 4), 3.64, 3.63 (3H each, s, OMe × 2), 3.62 (6H, s, OMe × 2), 3.53, 3.51 (each 3H, s, OMe × 2).

Methanolysis of 5a A solution of **5a** (20 mg) in 1% NaOMe (0.5 ml) in absolute MeOH (1 ml) was left standing overnight at room temperature. After neutralization with AcOH, the solvent was evaporated off under an N₂ stream, and the residue was treated with an excess of ethereal CH₂N₂ for 6 h. The residue obtained after removal of the solvent was subjected to preparative TLC (Kieselgel PF₂₅₄, benzene-acetone, 15:1) to yield **7** (3 mg), **8** (5 mg), [α]_D²⁰ -34° (*c*=1, acetone) and **9** (4 mg), which were identified by co-chromatography and by EI-MS and ¹H-NMR spectral comparison with authentic samples.

Partial Hydrolysis of Hirtellin T₁ (5) A solution of **5** (150 mg) in H₂O (50 ml) was heated at 80 °C for 17 h. After cooling, the precipitate formed was removed by centrifugation, and the concentrated supernatant was subjected to column chromatography over MCI-gel CHP-20P (1.1 cm i.d. × 20 cm) with H₂O → 10% MeOH → 15% MeOH → 20% MeOH → 30% MeOH → 40% MeOH in a stepwise gradient mode. The H₂O eluate gave gallic acid (3.6 mg) and the 10% MeOH eluate gave 2,3-di-*O*-galloyl-*D*-glucose (**17**) (2.9 mg). Mixture eluted by 30% MeOH was subjected to further chromatographic separation on MCI-gel CHP-20P column and preparative HPLC (R) to afford 1,3-di-*O*-galloyl-4,6-*O*-(*S*)-hexahydroxydiphenoyl-β-*D*-glucose (**11**) (5.1 mg) and remurin A (**14**) (3 mg). Their identities were confirmed by co-chromatography (normal-phase and reversed-phase HPLC) and by ¹H-NMR spectral comparison with authentic samples. The fraction eluted with 40% MeOH gave a crude dimeric hydrolyzate and unreacted hirtellin T₁ (37.2 mg). Further purification of the crude material by preparative HPLC (R) afforded the hydrolyzate **18** (2.1 mg).

Hydrolyzate **18**: An off-white amorphous powder. ¹H-NMR (acetone-*d*₆ + D₂O): δ 7.00, 6.96, 6.92 (2H each, s, Gal × 3), 6.65, 6.61, 6.49, 6.48 (1H each, s, HHDP × 2), 7.19, 7.12, 6.44, 6.40 (1H each, d, *J*=2 Hz), 7.09, 7.07 (1H each, s) (*m*-GOG × 2); glucose protons, see Table I.

Hirtellin Q₁ (6) An off-white amorphous powder, [α]_D²⁰ +60° (*c*=1.0, MeOH). *Anal.* Calcd for C₁₆₄H₁₁₄O₁₀₄·21H₂O: C, 47.69; H, 3.81. Found: C, 47.40; H, 3.64. UV λ_{max}^{MeOH} nm (log ε): 217 (5.39), 275 (5.17). CD (MeOH) [θ] (nm): +47.2 × 10⁴ (235), -9.7 × 10⁴ (262) and +12.7 × 10⁴ (283) and normal-phase HPLC (solvent B), t_R 22.23 (N2). ¹H-NMR (acetone-*d*₆ + D₂O): δ 7.02, 6.99, 6.96, 6.95, 6.91, 6.90 (2H each, s, Gal × 6), 6.64, 6.47 (2H each, s), 6.67, 6.61, 6.51, 6.48 (1H each, s)

(HHDP \times 4), 7.25 (2H, overlapped d, $J=2$ Hz), 7.24, 6.51, 6.50, 6.44 (1H each, d, $J=2$ Hz), 7.9, 7.07, 7.06 (1H each, s) (*m*-GOG \times 3); glucose protons, see Table II. ^{13}C -NMR (acetone- $d_6 + \text{D}_2\text{O}$): δ 107.16, 107.68 (2C) (*m*-GOG C-2), 107.76 (5C), 108.06, 108.20 (2C) (HHDP C-3, C-3'), 110.02 (4C), 110.14 (8C) (Gal C-2, C-6), 110.31, 110.36 (2C) (*m*-GOG C-6'), 112.69, 112.99, 113.16 (*m*-GOG C-6), 113.20, 113.33 (2C) (*m*-GOG C-1), 115.52, 115.55 (2C), 115.59, 115.78 (2C), 115.84 (2C) (HHDP C-1, C-1'), 119.41 (3C) (*m*-GOG C-1'), 119.78, 119.89, 119.96 (2C), 120.07 (2C) (Gal C-1), 125.43, 125.50, 125.52, 125.54, 126.10, 126.16 (3C) (HHDP C-2, C-2'), 136.22 (2C), 136.26 (2C), 136.44, 136.49, 136.54 (3C), 136.90, 136.93 (HHDP C-5, C-5', *m*-GOG C-2'), 139.07, 139.14, 139.17, 139.28, 139.35, 139.66 (Gal C-4), 139.88, 140.15 (2C), 140.18 (2C), 140.26, 140.54, 140.58, 140.67 (*m*-GOG C-4, C-3', C-4'), 143.20, 143.22, 143.40 (*m*-GOG C-5'), 144.17, 144.21 (4C), 144.26, 144.31, 144.34 (HHDP C-6, C-6'), 145.02, 145.07 (2C), 145.09 (3C), 145.18 (HHDP C-4, C-4'), 145.54 (2C), 145.59 (2C), 145.68 (2C), 145.72 (2C), 145.80 (4C) (Gal C-3), 145.86, 145.93, 146.05, 147.57, 147.70, 147.77 (*m*-GOG C-3, C-5), 163.82, 163.93, 164.40, 164.50, 164.59, 164.65, 164.72, 165.79, 166.44, 166.56 (2C), 166.73, 167.62 (2C), 167.70, 167.75, 168.24 (2C), 168.37, 168.46 (ester carbonyl carbons); glucose carbons, see Table III.

Acid Hydrolysis of Hirtellin Q₁ (6) A solution of **6** (1 mg) in 2N HCl (1 ml) in a sealed ampule was heated on a boiling-water bath for 10 h. After cooling, the solution was extracted with EtOAc. HCl in the aqueous layer was removed by adding Ag₂O, followed by filtration. The TLC analysis of the concentrated filtrate showed the presence of glucose. TLC: Kieselgel PF₂₅₄, *n*-BuOH-AcOH-H₂O (4:1:2); detection: 2-amino-phenylhydrogen oxalate reagent.

Methylation of Hirtellin Q₁ (6) Followed by Methanolysis A solution of **6** (4.5 mg) in MeOH (1 ml) was methylated with ethereal CH₂N₂ overnight at room temperature. The residue after removal of the solvent was treated with 1% NaOMe (0.15 ml) in MeOH (1 ml), and the reaction mixture was purified by preparative TLC to give **7** (0.7 mg) (EI-MS: M^+ , m/z 226), **8** (0.9 mg) (EI-MS: M^+ , m/z 450) and **9** (0.8 mg) (EI-MS: M^+ , m/z 436) which were identified by co-chromatography with authentic samples.

Partial Hydrolysis of Hirtellin Q₁ (6) A solution of **6** (140 mg) in H₂O (50 ml) was heated at 80 °C for 15 h. After cooling, the precipitate was collected by centrifugation and identified as ellagic acid. The concentrated supernatant was subjected to column chromatography over Toyopearl HW-40 (fine) (1.1 cm i.d. \times 40 cm) with 70% MeOH \rightarrow MeOH-H₂O-acetone [(7:2:1) \rightarrow (6:2:2) \rightarrow (5:2:3)] in a stepwise gradient mode. The 70% MeOH eluate afforded gallic acid (2 mg), 1,3-di-*O*-galloyl-4,6-*O*-(*S*)-hexahydroxydiphenoyl- β -D-glucose (**11**), gemin D (**12**) (1.3 mg), tellimagrandin I (**13**) (4.2 mg), remurin B (**15**) (1 mg), compound **16** (1.1 mg) and 2,3-di-*O*-galloyl-D-glucose (**17**) (1.8 mg). The fractions eluted with MeOH-H₂O-acetone (7:2:1) afforded remurin A (**14**) (4.4 mg) in addition to a mixture of two dimers, which were separated by preparative HPLC (R) to give compounds **4** (1.2 mg) and **18** (3.2 mg). The fraction eluted with MeOH-H₂O-acetone (6:2:2) gave a crude trimeric hydrolyzate which was purified by preparative HPLC to give compound **19** (2.5 mg) and unreacted hirtellin Q₁ (53.7 mg).

Compound **4**: An off-white amorphous powder, FAB-MS: m/z 1745 ($M + \text{Na}$)⁺. ^1H -NMR (acetone- $d_6 + \text{D}_2\text{O}$): δ 6.99, 6.98, 6.95, 6.94 (each, s, 6H in total, Gal), 6.64, 6.63, 6.62, 6.46, 6.45 (each, s, 4H in total, HHDP), 7.23, 7.21, 6.68, 6.67 (each, d, $J=2$ Hz, 2H in total, *m*-GOG), 7.09, 7.08 (each, s, 1H in total, *m*-GOG), 6.11, 6.11 (each, d, $J=8$ Hz, 1H in total, Glc H-1), 5.55, 5.54 (each, dd, $J=8$, 10 Hz, 1H in total, Glc H-2), 5.79 (1H, t, $J=10$ Hz, Glc H-3), 5.17 (1H, t, $J=10$ Hz, Glc H-4), 4.48 (1H, dd, $J=6.5$, 10 Hz, Glc H-5), 5.29 (1H, dd, $J=6.5$, 13.5 Hz, Glc H-6), 3.85, 3.84 (each, d, $J=13.5$ Hz, 1H in total, Glc H-6). α - and β -anomer; 5.36 (d, $J=4$ Hz, Glc H-1'), 4.99 (dd, $J=4$, 10 Hz, Glc H-2'),

5.81 (t, $J=10$ Hz, Glc H-3'), 5.06 (t, $J=10$ Hz, Glc H-4'), 4.60 (dd, $J=6.5$, 10 Hz, Glc H-5'), 5.23 (dd, $J=6.5$, 13.5 Hz, Glc H-6'), 3.73 (dd, $J=13$ Hz, Glc H-6') (α -anomer); 4.45 (d, $J=8$ Hz, Glc H-1'), 5.14 (dd, $J=8$, 10 Hz, Glc H-2'), 5.53 (t, $J=10$ Hz, Glc H-3'), 5.05 (t, $J=10$ Hz, Glc H-4'), 4.05 (dd, $J=6.5$, 10 Hz, Glc H-5'), 5.25 (dd, $J=6.5$, 13.5 Hz, Glc H-6'), 3.80 (d, $J=13.5$ Hz, Glc H-6') (β -anomer).

Compound **16**: An off-white amorphous powder. ^1H -NMR (acetone- $d_6 + \text{D}_2\text{O}$): δ 7.06, 6.94 (2H each, s, Gal), 7.21, 6.48 (1H each, d, $J=2$ Hz, *m*-GOG), 6.92 (1H, s, *m*-GOG), 5.35 (1H, d, $J=8$ Hz, Glc H-1), 5.32 (1H, dd, $J=8$, 9 Hz, Glc H-2), 5.41 (1H, t, $J=9$ Hz, Glc H-3), 3.89 (1H, t, $J=10$ Hz, Glc H-4), 3.57 (1H, ddd, $J=2.5$, 4.5, 10 Hz, Glc H-5), 3.73 (1H, dd, $J=4.5$, 12.5 Hz, Glc H-6), 3.84 (1H, dd, $J=2.5$, 12.5 Hz, Glc H-6).

Compound **19**: An off-white powder, ^1H -NMR (acetone- $d_6 + \text{D}_2\text{O}$): δ 7.02, 6.95, 6.93, 6.91 (each 2H, s, Gal), 6.65, 6.64, 6.62, 6.50, 6.48, 6.47 (each 1H, s, HHDP), 7.40, 7.21, 7.09, 6.48, 6.453, 6.450 (1H each, d, $J=2$ Hz, *m*-GOG), 7.11, 7.06, 6.95 (each 1H, s, *m*-GOG); glucose protons, see Table II.

Degalloylation of Hirtellin A (3) to Compound 4 A solution of hirtellin A (**3**) (21 mg) in H₂O (2 ml) was incubated with tannase (15 drops) at 37 °C for 12.5 h. The reaction was terminated by acidification with two drops of 1N HCl, and the reaction mixture was applied to a BondElut RP-18 cartridge. Elution was conducted with H₂O (2 ml \times 5) and then MeOH. The H₂O eluate afforded gallic acid (1.8 mg), and the MeOH eluate gave compound **4** (18 mg), which was identified as the degalloylated derivative of hirtellin A^(2b) by co-chromatography on HPLC and ^1H -NMR spectral comparison with an authentic sample.

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