

TOTAL SYNTHESIS OF PHOTOSYNTHETIC PIGMENT FUcoxANTHIN BY USE OF OXO-METALLIC CATALYST

Yumiko YAMANO and Masayoshi ITO*

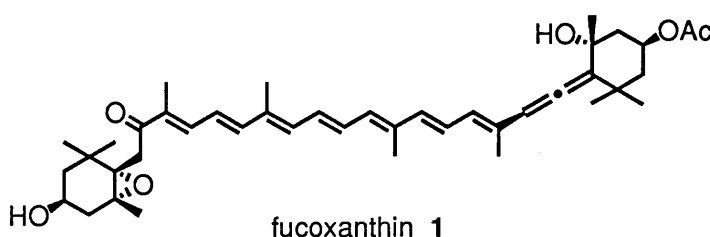
Kobe Women's College of Pharmacy, Motoyamakita, Higashinada, Kobe 658, Japan

The first total synthesis of optically active fucoxanthin **1** has been accomplished *via* the 8-oxo-compound **7**, efficiently prepared by rearrangement of the α -acetylenic alcohol **2** using oxo-metallic catalyst and subsequent iodine catalyzed double bond-shift.

KEYWORDS fucoxanthin; carotenoid; rearrangement; oxo-metallic catalyst

The allenic carotenoid fucoxanthin (**1**) is known to be widely distributed in brown algae and to function as a light harvesting pigment²⁾ for photosynthesis in the sea. In order to elucidate the fucoxanthin-protein interaction in algal photosynthetic pigment systems and to clarify the antenna function by chemical methods, development of a synthetic method for fucoxanthin molecule has been strongly desired for a long time.

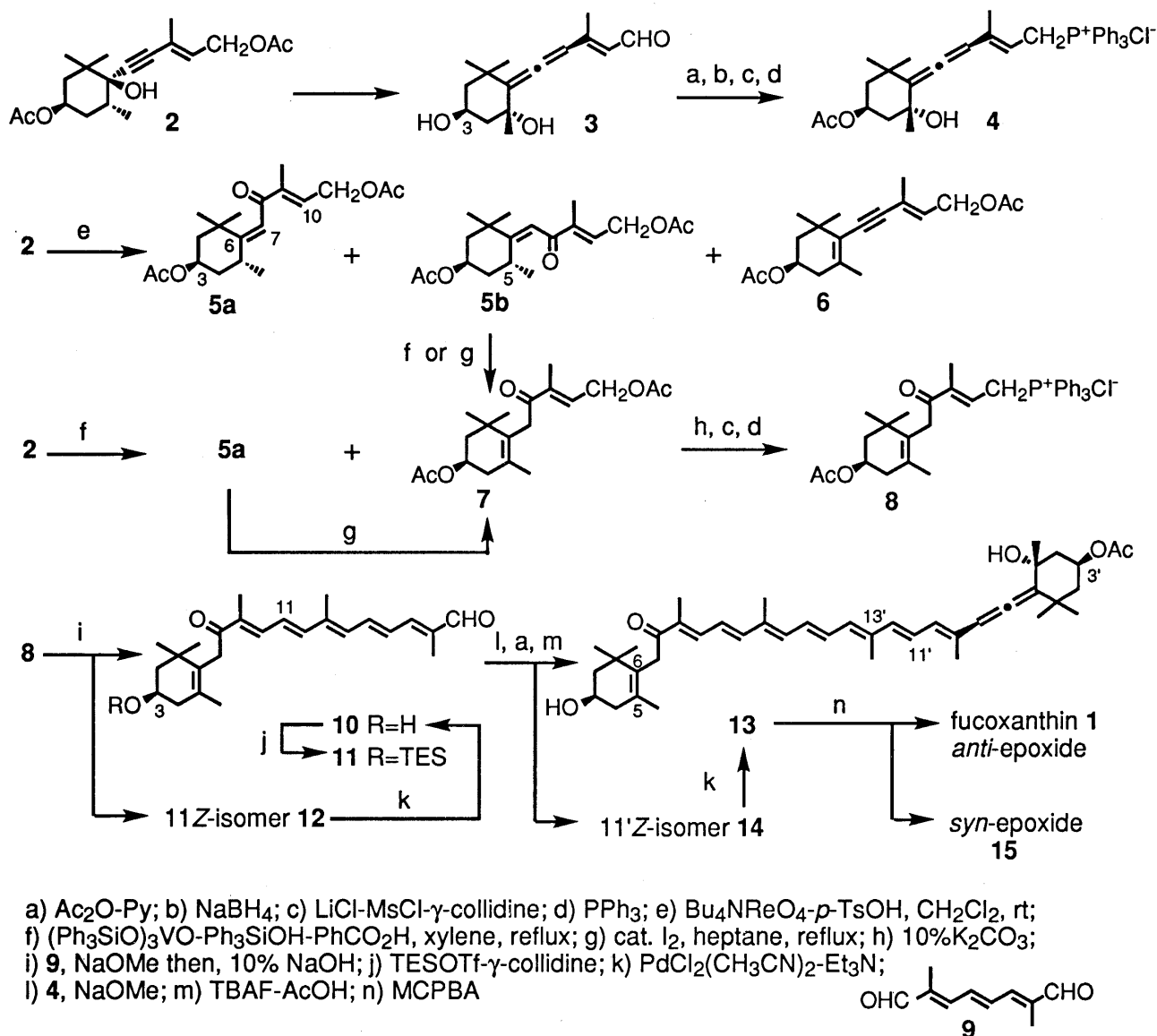
Moreover, it has recently been found³⁾ that **1** has effective antiproliferative and antitumor promoting activities. Here we wish to describe the first total synthesis of optically active **1**.



As shown in the Scheme, epoxidation of 5,6-double bond in the fucoxanthin skeletal compound **13** was employed at the final step because of an extreme alkali-lability⁴⁾ of β,γ -epoxy-keto-moiety in **1**. The compound **13** (C₄₀) was constructed by the Wittig reaction of C₁₀-dialdehyde **9** with two kinds of C₁₅-Wittig salts **4** and **8**, which were synthesized from the previously prepared⁵⁾ common intermediate **2** in an optically active form (97% ee) starting from the readily available (4*R*,6*R*)-4-hydroxy-2,2,6-trimethylcyclohexanone.

The allenic Wittig salt **4** was synthesized in 4 steps from the allenic aldehyde **3**, whose preparation from **2** was reported.⁵⁾ The 8-oxo-Wittig salt **8** was constructed by the application of the key reaction, i.e., the rearrangement of α -acetylenic alcohols to α,β -unsaturated carbonyl compounds by oxo-metallic catalysts⁶⁾ and subsequent iodine catalyzed double bond-shift. Reaction of the α -acetylenic alcohol **2** with catalytic amount of tetrabutylammonium perrhenate and *p*-toluenesulfonic acid^{6a)} at room temperature afforded the rearranged α,β -unsaturated ketones **5a** (6*Z*-isomer) (32%) and **5b** (6*E*-isomer) (50%) accompanied by the dehydrated product **6** (14%). On the other hand, treatment of **2** with tris(triphenylsilyl)vanadate catalyst^{6b)} in refluxing xylene gave α,β - and β,γ -unsaturated ketones **5a** (35%) and **7** (58%).⁷⁾ Under the reaction conditions, **5b** was converted to the β,γ -unsaturated ketone **7** (81%); nevertheless **5a** was not changed. Thus, **7** was assumed to be derived from the 6*E*-isomer **5b** by intramolecular hydrogen shift (C₅ to carbonyl oxygen). In addition, transformation of the

6*Z*-isomer **5a** to **7** was achieved in 80% yield by treatment with iodine in refluxing heptane. This reaction was found to proceed through the intermediate 6*E*-isomer **5b**, which was isolated in the course of the conversion. The structures of **5a,b** and **7** were determined on the basis of the IR and ¹H-NMR data⁸⁾ including NOE experiments. The 8-oxo-compound **7** was transformed in 3 steps into the Wittig salt **8** in 60% yield.



The Wittig condensation of **8** with C₁₀-dialdehyde **9** in the presence of NaOMe as a base and followed by hydrolysis afforded a mixture of the all-*E*-8-oxo-apocarotenal **10** (32%) and the 11*Z*-isomer **12** (29%). The latter was isomerized to the former in 94% yield by treatment⁹⁾ with palladium catalyst. After protection (79%) of the hydroxyl group of **10**, the product **11** was treated with the allenic Wittig salt **4** with NaOMe as a base to give a mixture of the condensed products which was acetylated and desilylated by the combined use of tetrabutylammonium fluoride (TBAF) and acetic acid to provide the all-*E*-fucoxanthin skeletal compound **13** (25%) and its 11'*Z*-isomer **14** (31%). These structures were characterized by spectral data.⁸⁾ Isomerization of the 11'*Z*-isomer

14 using palladium catalyst⁹⁾ afforded the all-*E*-isomer **13** in 45% yield. Finally, epoxidation of **13** with MCPBA followed by HPLC purification furnished a mixture (36%) of the *syn*-epoxide **15** and the *anti*-one **1** with the recovery (27%) of **13**. Separation of the epoxide mixture by preparative HPLC using a chiral column (CHIRALCEL OD; DAICEL) gave **15** (28%) and **1** (8%) in pure form, respectively. Spectral data (IR, UV-VIS, ¹H-NMR¹⁰⁾ and MS), including CD data of synthetic fucoxanthin **1**, were identical with those of natural specimen.

This is the first total synthesis of optically active fucoxanthin. Thus, this route has general applicability to the synthesis of a variety of fucoxanthin analogues.

ACKNOWLEDGEMENTS The authors are indebted to Professor Y. Koyama, Kwansei Gakuin University, and Professor K. Tsujimoto, Japan Advanced Institute of Science and Technology, Hokuriku, for their invaluable gift of natural fucoxanthin. This work was supported in part by research grants from the Ministry of Education, Science, and Culture (Japan). We also appreciate the financial and chemical support of Kuraray Co., Ltd. Japan.

REFERENCES AND NOTES

- 1) a) R. Bonnett, A. K. Mallams, A. A. Spark, J. L. Tee, B. C. L. Weedon, A. McCormick, *J. Chem. Soc. C*, **1969**, 429; b) K. Bernhard, G. P. Moss, Gy. Tóth, B. C. L. Weedon, *Tetrahedron Lett.*, **1976**, 115.
- 2) F. T. Haxo, "Comparative Biochemistry of Photoreactive Systems," ed. by M. B. Allen, Academic Press, New York, 1960, pp. 339-360.
- 3) J. Okuzumi, H. Nishino, M. Murakoshi, A. Iwashima, Y. Tanaka, T. Yamane, Y. Fujita, T. Takahashi, *Cancer Lett.*, **55**, 75 (1990).
- 4) S. Liaaen-Jensen, *Pure Appl. Chem.*, **63**, 1 (1991).
- 5) Y. Yamano, M. Ito, *J. Chem. Soc., Perkin Trans. 1*, **1993**, 1599.
- 6) a) K. Narasaka, H. Kusama, Y. Hayashi, *Chem. Lett.*, **1991**, 1413; b) H. Pauling, D. A. Andrews, N. C. Hindley, *Helv. Chim. Acta*, **59**, 1233 (1976); c) M. B. Erman, I. S. Aul'chenko, L. A. Kheifits, V. G. Dulova, J. N. Novikov, M. E. Vol'pin, *Tetrahedron Lett.*, **1976**, 2981; d) P. Chabardes, *Tetrahedron Lett.*, **29**, 6253 (1988).
- 7) In this reaction, only a small amount of 6*E*-isomer **5b** was detected by HPLC.
- 8) Characteristic ¹H-NMR data (in CDCl₃) for compounds **5a,b**, **7**, **13** and **14** are as follows:
5a; δ(500 MHz): 1.12 (3H, d, *J* 6.5, 5-Me), 2.78 (1H, m, 5-H), 5.77 (1H, d-like, *J* 1.5, 7-H).
5b; δ(500 MHz): 1.31 (3H, d, *J* 7.5, 5-Me), 3.57 (1H, qdd, *J* 7.5, 6, 1.5, 5-H), 6.38 (1H, s, 7-H).
7; δ(200 MHz): 1.45 (3H, s, 5-Me), 3.43 (2H, s, 7-H₂). **13**; δ(500 MHz): 1.99 (3H, s, 13'-Me), 6.13 (1H, dd-like, *J* 11.5, 1, 10'-H), 6.35 (1H, d, *J* 15, 12'-H), 6.59 (1H, dd, *J* 15, 11.5, 11'-H).
14; δ(500 MHz): 2.12 (3H, s, 13'-Me), 5.98 (1H, d, *J* 12, 12'-H), 6.27 (1H, t, *J* 12, 11'-H), 6.63 (1H, br d, *J* 12, 10'-H).
- 9) A. Fischli, H. Mayer, W. Simon, H. -J. Stoller, *Helv. Chim. Acta*, **59**, 397 (1976).
- 10) G. Englert, T. Bjørnland, S. Liaaen-Jensen, *Magn. Reson. Chem.*, **28**, 519 (1990).

(Received December 10, 1993; accepted January 16, 1994)