Synthesis and Analysis of Positive Inotropic Effects of 3-Substituted-2*H*-cyclohepta[*b*]furan-2-one Derivatives

Masayuki Yokota,*,^a Takashi Yanagisawa,^a Kazuhiro Kosakai,^a Shuichi Wakabayashi,^a Tsuyoshi Томіуама,^a and Masafumi Yasunami^b

Kotobuki Research Laboratories, Kotobuki Seiyaku Co., Ltd., 6351 Sakaki-machi, Hanishina-gun, Nagano 389–06, Japan and Department of Chemistry, Faculty of Science, Tohoku University, Aramaki-aza-Aoba, Sendai 980, Japan. Received June 7, 1993; accepted November 12, 1993

Several 3-substituted-2H-cyclohepta[b]furan-2-one derivatives were prepared and tested *in vitro* for positive inotropic character. Introduction of an isopropyl group at the 5-position of compound 8a caused an increase of PIC_{50} (negative logarithm of the dosage which increases the contractile force by 50%) from 4.48 to 5.10. Among the 5-isopropyl-8-alkoxy compounds, the isopropoxy compound 12f had the most potent activity with a PIC_{50} value of 5.99. Conversion of the ester group at the 3-position to a methylene group and of the alkoxy group at the 8-position to a substituted amino group caused a decrease in activity. The most active compound, 12f, was also found to have a weaker heart rate (HR)-increasing effect compared to milrinone and amrinone.

Keywords positive inotropic effect; 2*H*-cyclohepta[*b*]furan-2-one; phosphodiesterase

Digitalis glycosides such as digoxin (1) and digitoxin (2) have been known for centuries as positive inotropic agents that increase heart muscle contraction. The positive inotropic effect of digitalis is beneficial for most patients with congestive heart failure but the ratio between the full therapeutic dose and the toxic dose of digitalis is small, and up to 20% of hospitalized patients receiving digitalis may develop some signs of intoxication. 1) The development of safe and effective positive inotropic agents for treatment of congestive heart failure is required.2) Amrinone (3), 3) milrinone $(4)^{4)}$ and imazodan $(5)^{5)}$ have been reported to improve congestive heart failure (CHF), but side effects such as leukopenia were observed in 20% of patients⁶⁾ and more effective and safer cardiotonic agents for CHF are still required. We reported 3-ethyl-7isopropyl-1-azulenesulfonate (KT1-32)7 and 3-ethyl-6isopropyl-1-azulenesulfonate (KT1-785)8) as potent and chemically stable anti-ulceratives. The precursor of the synthetic azulene, 2H-cyclohepta[b]furan-2-one (6), has a lactone in its molecular structure and a lactone is considered to be essential to the positive inotropic activities of digitalis glycosides.9) We examined the role of lactone

$$\begin{array}{c} & & & & \\ & & & \\ & & & \\ & & & \\ &$$

 $2\textit{H}\text{-}\mathsf{cyclohepta}[\,b\,]\mathsf{furan}\text{-}2\text{-}\mathsf{one}\,(\,6\,)$

as bioisoster of lactam in the molecules of amrinone and milrinone (having a pyridone moiety) or in that of imazodan (having a pyridazinone moiety). We then synthesized and examined a series of 3-substituted-2*H*-cyclohepta[*b*]furan-2-one derivatives. We describe herein the synthesis and biological characterization of this new class of compounds.

Chemistry

The synthesis of these compounds proceeded from the known 5-substituted-3-methoxycarbonyl-2H-cyclohepta-[b]furan-2-ones 8 and 5-substituted-8-hydroxy-3-ethoxycarbonyl-2*H*-cyclohepta[b]furan-2-ones 11^{10} (Chart 1). The reaction of the known tosyl tropolones 7 with dimethyl malonate and sodium methoxide yielded a mixture of compounds 8 and 9 in a ratio of 90:8 (route a). The 8-hydroxy compounds 11 were obtained more efficiently by route b. The reaction of 7 with diethyl malonate and sodium ethoxide yielded a mixture of 5-substituted-3ethoxycarbonyl-2H-cyclohepta[b]furan-2-one 10 and 5substituted-8-hydroxy-3-ethoxycarbonyl-2H-cyclohepta-[b] furan-2-one 11 in a ratio of 8:80. In route a, the arylsulfonyloxy group is eliminated via a protonated intermediate to give a tropone-type intermediate, 10) which subsequently cyclizes to the 2H-cyclohepta[b]furan-2-one 8 as the major product. On the other hand, in route b the arylsulfonyl group of the intermediate is eliminated as arenesulfinate ion with the aid of a strong base (NaOEt) to produce a tropolonate type intermediate, 10) which subsequently cyclizes to the 8-hydroxy-2H-cyclohepta[b]furan-2-one 11 as the major product. After recrystallization, 8 and 11 were obtained in pure form. Alkylation and subsequent amination of the 8-hydroxy group of 11 are shown in Chart 2. Compounds 11 were alkylated with diazomethane in ether or alkyl halide with potassium carbonate to yield compounds 12, which were aminated with appropriate amines, yielding the amino compounds 13. The ester 8 was demethoxycarbonylated with 75% sulfonic acid, followed by formylation via the Vilsmeier-

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synthesis of 2H-cyclohepta[b]furan-2-one

Chart 1

 $R_1 = H$ or isopropyl

11

B)
$$R_{1} = \text{H or isopropyl}$$

- (a) CH_2N_2 /ether; (b) alkyl halide, K_2CO_3 /toluene, r. t,; (c) NHR_2 /EtOH, reflux;
- $\text{(d) } 75\% \; H_2SO_4, \\ 100^{\circ}C \; ; \; \text{(e) } POCl_3/DMF, \\ 0^{\circ}C ; \; \text{(f) } NaBH_4, \\ MeOH, \\ r.t. \; ; \; \text{(g) } SOCl_2/C_6H_6, \\ reflux \; ; \; \text{(f) } POCl_2/C_6H_6, \\ reflux \;$
- (h) NHR₂/MeOH, r. t.; (i) P(EtO)₃, 150°C

Chart 2

Haack reaction to give the corresponding aldehyde 14. Reduction of 14 with NaBH₄ in methanol, followed by chlorination of the resulting 3-hydroxymethyl compound 15 with thionyl chloride yielded 16. Amination of 16 with appropriate amines yielded 17. Phosphorylation of 15 with triethylphosphate yielded the phosphoryl esters 18.

Results and Discussion

The compounds described herein were examined for positive inotropic activity. They were tested for their effects on isolated guinea pig atria and papillary muscle according to the method previously reported. 11) The contractile activities of these compounds were expressed as the negative logarithm of the dosage which increases the contractile force by 50% (PIC₅₀) and the maximal change of contractile force (CF) and percentage increase in the heart rate (HR) were also recorded. The results for the 3-alkoxycarbonyl compounds (8a, b, 12a—i) are presented in Table I. The introduction of the isopropyl group at the C-5 position of compound 8a increased the PIC₅₀ value from 4.48 to 5.10. The compound having an 8-methoxy group (12a) also showed an increased the PIC₅₀ value, from 4.48 (8a) to 4.75 (12a). The introduction of an isopropyl group at the 5-position of compound 12a resulted in an increase of the PIC₅₀ value from 4.75 to 5.70. The effect of a 5-isopropyl group on the PIC_{50} value was also seen in compound 8b as compared to the unsubstituted 8a. In the 5-isopropyl compounds, the 8hydroxy group of compound 12b was replaced with OMe (12c) up to O-n-butyl (12h) groups. The maximal activity

Table I. 8-Unsubstituted and 8-Substituted-3-alkoxycarbonyl-2H-cyclohepta[b]furan-2-ones

Compd.	R ₁	R ₂	R ₃	mp (°C) ^{a)}	Formula ⁶⁾	Analysis (%) Calcd (Found)		Atria			
						C	Н	Inotropic PIC ₅₀ c)	Max change CF (%) ^{d)}	Max change HR (%) ^{e)}	PMI ^f)
8a	Н	Me	Н	172—173	$C_{11}H_8O_4$	64.71	3.95	4.48	93	26	
01					** ° *	(64.38	3.90)	. 4.40	93	26	10
8b	iso-Pr	Me	H	119—120	$C_{14}H_{14}O_{4}$	68.28	5.73	5.10	70	16	$NE^{g)}$
12a	Н	E.	014			(68.39	5.70)		, ,	10	1415-
124	п	Et	OMe	139—140	$C_{13}H_{12}O_{5}$	62.90	4.87	4.75	124	13	NE
12b	iso-Pr	Et	ОН	120 122	C 11 0	(62.89	4.66)				
	130 11	Lt	OH	130—132	$C_{15}H_{16}O_{5}$	65.21	5.84	3.58	53	10	NE
12c	iso-Pr	Et	OMe	170—171	$C_{16}H_{18}O_5$	(64.90	6.18)				
			01110	170-171	$C_{16}\Pi_{18}O_5$	66.20 (66.19	6.25	5.70	113	45	20
12d	iso-Pr	Et	OEt	182—184	$C_{17}H_{20}O_{5}$	67.09	6.23) 6.62	6.20	1.10		
					01/11/2005	(66.74	6.77)	6.20	142	19	16
12e	iso-Pr	Et	O-n-Pr	130-132	$C_{18}H_{22}O_{5}$	67.91	6.96	5.60	140	1.4	<i>~</i> 1
100					10 22 3	(68.30	7.04)	5.00	140	-14	51
12f	iso-Pr	Et	O-iso-Pr	119—120	$C_{18}H_{22}O_5$	67.91	6.96	5.99	173	17.5	95
120	Н	TC4	0 5			(68.17	7.07)		1,5	17.3	93
12g	н	Et	On-Bu	127—129	$C_{16}H_{18}O_{5}$	66.20	6.25	5.13	111	3	NE
12h	iso-Pr	Et	O-n-Bu	101 100	G ** *	(66.60	6.29)			-	. 12
	130-11	Lit	0- <i>n</i> -ви	121—123	$C_{19}H_{24}O_{5}$	68.66	7.28	5.30	48	0	NE
12i	Н	Et	OAc	134—136	C II C	(68.70	7.29)				
			OAC	134130	$C_{14}H_{12}O_6$	60.87	4.38	3.60	60	9	NE
						(60.87	4.36)				

a) Melting points were measured without correction. b) Analytical values for C, H were within 0.4% of calculated values. c) Negative logarithm of the dose which increases contractile force of isolated guinea pig left atria by 50%. d) Maximum change (%) of contractile force of isolated guinea pig left atria. e) Maximum no effect.

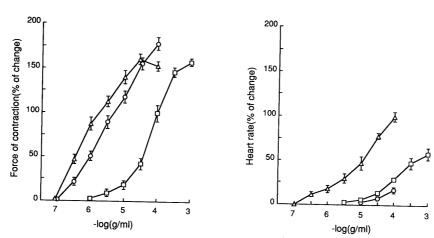


Fig. 1. Cumulative Concentration–Response Curves for the Effects of Compound 12f (—○—), Amrinone (—□—) and Milrinone (—△—) on Force of Contraction (Left Panel) and Heart Rate (Right Panel) in Isolated Guinea Pig Atria

was obtained with the isopropyl compound 12f, with PIC₅₀ of 5.99 and CF of 173%. Further elongation of the alkyl chain to *n*-butyl gave no further increase in activity. The activity of the 8-acetyl compound 12i was weaker than that of the 8-alkylated compound 12a. Among a series of 2*H*-cyclohepta[*b*] furan-2-one compounds, 12f had the most potent activity. The results for 8-amino-substituted compounds are shown in Table II. In this series, the diethylamino compound 13a had the maximal effect, with

a PIC₅₀ of 5.30. The introduction of a diamino group instead of an amino group at the 8-position of compound **13a** (compound **13g**) caused a decrease of PIC₅₀ activity from 5.30 to 4.07. The results for the 3-substituted methyl compounds **17a**—**g** and **18** are presented in Table III. Replacement of the carbonyl group (=CO) at the 3-position with methylene (=CH₂) caused loss of PIC₅₀ and CF activity, so the ester carbonyl group at the 3-position of 2H-cyclohepta[b]furan-2-one is considered

TABLE II. 8-Amino Substituted 3-Ethoxycarbonyl-2H-cyclohepta[b]furan-2-ones

		R_2	mp (°C) ^{a)}	Formula ^{b)}	Analysis (%)			Atria			
Compd.	R_1				Calcd (Found)			Inotropic ^{c)}	Max change	Max change	C)
	ιτη				С	Н	N	PIC ₅₀	CF ^{d)} (%)	HR ^{e)} (%)	PMI ^f)
13a	Н	NEt ₂	78—79	$C_{16}H_{19}N_1O_4$	66.42	6.62	4.84	5.30	129	6	46
		\sim	215 216	CHNO	(66.45 66.89	6.59 5.96	4.88) 4.88	4.97	160	0	26
13b	Н	N	215—216	$C_{16}H_{17}N_1O_4$	(66.63	5.89	4.91)	1.57	- • •		
13c	iso-Pr	N	121—123	$C_{20}H_{25}N_1O_4$	69.95	7.34	4.08	4.30	65	2	25
100					(69.75	7.38	4.46)				
13d	iso-Pr	N O	156—158	$C_{19}H_{23}N_1O_5$	65.98	6.71	4.06	4.75	85	2	8
					(66.07	6.49	4.20)				
13e	iso-Pr	HN—	165—167	$C_{21}H_{27}N_1O_4$	70.56	7.61	3.96	4.50	54	-3	$NE^{g)}$
150	100 11				(70.57	7.65	4.14)		18	-16	NE
13f	Н	$HN(CH_2)_2NMe_2$	143 (dec.)		59.61 (59.52	6.88 6.79	8.69 8.80)		10		
	**	HN(CH ₂) ₂ NEt ₂	197 (dec.)	$^{\cdot}\mathrm{H_{2}O}$ $\mathrm{C_{18}H_{24}N_{2}O_{4}}$	61.70	7.48	7.99	4.07	53	6	NE
13g	13g H	$HN(Cn_2)_2NLl_2$	177 (dec.)	·H ₂ O	(61.63	7.59	7.91)			4.5	NE
13h	iso-Pr	$HN(CH_2)_2NMe_2$	228 (dec.)	$C_{19}H_{26}N_2O_4$	65.92	7.54	8.05	4.88	38	45	NE
		~					,		-63	-22	NE
13i	Н	$HN(CH_2)_2-N$	210 (dec.)						03		
13h 13i	iso-Pr H	$HN(CH_2)_2NMe_2$ $HN(CH_2)_2-N$		$C_{19}H_{26}N_2O_4$ $C_{18}H_{22}N_2O_4$ $\cdot H_2O$	65.92 (65.87 62.06 (62.25	7.54 7.56 6.94 6.79	8.05 8.09) 8.04 7.66)	4.88	-63		

Footnotes, see Table I.

to be essential for the activities. The introduction of pyridazinylmethyl, a constituent of imazodan, at the 3-position (compound 17g) increased PIC $_{50}$ and CF%, but the activities are not superior to those of 12f. The activities of amrinone and milrinone as reference compounds are also shown in Table III. From these findings, compound 12f is the most active compound in this series. As shown in Fig. 1, compound 12f increased the contractile force of guinea-pig atria from 3×10^{-7} to 3×10^{-5} g/ml in a dose-dependent manner with the maximal change of 173%. The positive inotropic effect of compound 12f was three times less potent than that of milrinone and 30 times stronger than that of amrinone. In guinea-pig papillary muscle compound 12f at 3×10^{-5} g/ml increased the contractile force by 95%. The effect of compound 12f was comparable to those of amrinone and milrinone at the same dose. Positive chronotropic activity in cardiotonic agents is considered undesirable, 12) since an ideal positive inotropic agent should not induce tachycardia, which can lead to cardiac arrhythmias and an increase of myocardial oxygen consumption. 13) Compound 12f was evaluated for effect on HR. Compared to milrinone, 12f caused a relatively minor HR increase (17.5%, 12f v.s. 155%, milrinone) and had a more favorable profile, as shown in Fig. 1. To determine the mechanism responsible for the inotropy exhibited by 12f, the phosphodiesterase III (PDE III)-inhibitory activity of 12f was tested according to the method of Schuhkmacker et al. 13) as shown in Table IV. The IC50 of 12f $(7.67 \times 10^{-6} \,\mathrm{M})$ was the same as that of milrinone $(IC_{50} = 7 \times 10^{-6} \text{ M})$. Thus, the action of 12f may be due

to PDE III inhibition, but other possible explanations include Na+, K+-ATPase-inhibitory activity or Ca2+sensitizing effect, a mechanism first described by Herzig et al. 14) The potential interest of this effect is that it does not require an increase in calcium concentration for cardiac contraction, thereby avoiding the risk of calcium overload and arrhythmia. 15) The inotropic effect of digitalis glycosides having lactones in the molecule is due to Na+,K+-ATPase-inhibitory effects.9) Since 12f also has lactone in the molecule, it may have the same mechanism of inhibitory effect as digitalis glycosides. Another possibility is that 12f has an additional undefined mechanism of action. These possibilities are currently under investigation. In conclusion, we have demonstrated that in the 3-substituted-2H-cyclohepta[b]furan-2-one series, ethyl 5-isopropyl-8-isopropyloxy-2-oxo-2H-cyclohepta[b]furan-3-carboxylate (12f) showed positive inotropic activity. Furthermore, the chronotropic effect of (12f) was less than those of the reference compounds. Therefore, 12f represents a new structural lead compound.

Experimental

Melting points were determined on a Yanagimoto melting point apparatus and are uncorrected. Infrared (IR) spectra were recorded on a Hitachi 270-30 instrument. Proton nuclear magnetic resonance (1H-NMR) spectra were measured at 90 MHz on a Hitachi R-90H Fourier-transform NMR spectrometer, and date are summarized in Table V. Chemical shifts are quoted in parts per million (ppm) with tetramethylsilane as an internal standard. Coupling constants (J) are given in hertz (Hz). The following abbreviations are used: s, singlet; d, doublet; t, triplet; q, quartet; br s, broad singlet; sept, septet; m, multiplet. Mass spectra (MS) were taken on a Hitachi M-80B spectrometer. For column chromatography, silica gel (Merck, Kieselgel 60, 70-230 mesh)

Table III. 3-Aminomethyl Substituted 3-Diethylphosphinomethyl 2*H*-cyclohepta[*b*]furan-2-ones

Compd.	R,	R ₂	mp (°C) ^{a)}	Formula ^{b)}	Analysis (%) Calcd (Found)			Atria			
	1				C	Н	N	Inotropic ^{c)} PIC ₅₀	Max change CF ^d (%)	Max change HR ^{e)} (%)	PMI ^f)
17a	iso-Pr	NMe_2	210—212	$C_{15}H_{19}N_1O_2$	68.42	8.04	5.32		40	-38	NE ^{g)}
17b	Н	NO	233—235	${}^{\cdot}\mathrm{H_{2}O}$ $\mathrm{C_{14}H_{15}N_{1}O_{3}}$ ${}^{\cdot}\mathrm{H_{2}O}$	(68.52 63.86 (63.83	7.77 6.51 6.21	5.31) 5.32		3	-2	NE
17c	iso-Pr	NO	153—155	$C_{17}H_{21}N_1O_3$ · H_2O	66.86	7.59 7.43	5.10) 4.59 4.82)	_	45	-37	NE
17d	Н	NMe	239—241	$C_{21}H_{21}N_{1}O_{2} \cdot H_{2}O$	75.11 (74.75	6.20 6.27	4.23 4.15)		26	35	81
17e	Н	HN-CH ₂ -OMe OMe OMe		$C_{19}H_{19}N_1O_4$	70.14 (70.13	5.87 5.78	4.31 4.29)	· <u> </u>	-34	-66	NE
17f	Н 1	OH CH ₃		$C_{24}H_{28}N_2O_5$	67.91 (68.06	6.65 6.65	6.60 6.39)	4.02	51	-31	NE
17g	Н	HN-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-	164—165	$C_{21}H_{17}N_3O_3$	69.88 (70.18	4.70 4.77	11.92 11.70)	5.25	59	38.5	65
18	Н	$ \begin{array}{c} O \\ -P \cdot (OEt)_2 \end{array} $	7071	$C_{14}H_{17}O_5P_1$	56.76 (56.75	5.78 5.75)			44	6	-38
Amrinon Milrinon					•	- /		4.43	151	51 .	95
Footpot	es, see Ta	abla I						6.43	155	97	97

Footnotes, see Table I.

Table IV. IC_{50} Values for Bovine Myocardiac Phosphodiesterase (Type III)

Compound	IC_{50} (M)
8b	4.27×10^{-5}
12a	$> 10^{-4}$
12f	7.67×10^{-6}
12g	1.30×10^{-5}
Amrinone	1.16×10^{-6}
Milrinone	7×10^{-6a}
Papaverine	1.16×10^{-6}

a) Data from N. Komas, C. Lugnier, A. L. Bec, C. S-L. Gal, G. Barthelemy, J. C. Stoclet, J. Cardiovasc. Pharmacol., 14, 213 (1989).

was used.

Route a: Methyl 2-Oxo-2*H*-cyclohepta[*b*]furan-3-carboxylate (8a) Sodium methoxide, prepared from sodium metal (1.8 g, 0.8 mol) and anhydrous MeOH (80 ml) was, gradually added to a stirred solution of 2-troponyl *p*-toluenesulfonate 7 (11.0 g, 0.4 mol) and dimethyl malonate (10.5 g, 0.8 mol) in anhydrous MeOH (80 ml) at 0 °C. After being stirred for an additional 6 h at 0 °C, the reaction mixture was poured into H_2O . Crystals were collected, washed with H_2O , dried *in vacuo* and recrystallized from EtOH, giving 8a (7.3 g, 89%) as yellow needles (mp 172—173 °C, lit. ¹⁶⁾ 172.5—173.8 °C). The aqueous filtrate was acidified with 6 N HCl and crystals were collected and recrystallized from EtOH, giving 9 (0.75 g, 8.5%) as yellow needles.

Route b: Ethyl 8-Hydroxy-2-oxo-2*H*-cyclohepta[*b*]furan-3-carboxylate (11) Similar treatment of a solution of 7 (13.8 g, 0.05 mol) and diethyl malonate (16.0 g, 0.1 mol) in anhydrous EtOH (100 ml) with 0.1 m NaOEt gave 10 (1.1 g, 8.3%) as yellow needles (mp 129—130 °C) and 11 (11 g,

79%) as yellow needles (mp 223-225°C).

Ethyl 8-Methoxy-2-oxo-2*H*-cyclohepta[*b*]furan-3-carboxylate (12a) A suspension of 11 (10.0 g, 0.043 mol) in AcOEt 100 ml was treated with freshly prepared diazomethane in Et₂O (100 ml) with stirring under ice-cooling. After additional stirring for 2 h, the precipitate formed was collected by filtration and recrystallized from MeOH to give 12a (6.78 g, 63.8%). MS m/z: 248 (M⁺). IR (KBr): 2960, 1748 cm⁻¹. ¹H-NMR (CDCl₃) δ : 1.42 (3H, t, J=7.0 Hz, -CH₂CH₃), 4.18 (3H, s, -OCH₃), 4.40 (2H, q, J=7.0 Hz, -CH₂CH₃), 7.20—7.50 (3H, m, C₅-, C₆-, C₇-H), 8.90 (1H, d, J=10 Hz, C₄-H).

Ethyl 8-Diethylamino-2-oxo-2*H*-cyclohepta[*b*]furan-3-carboxylate (13a) A mixture of 12a (2.5 g, 0.01 mol) and diethylamine (2.2 g, 0.03 mol) in EtOH 10 ml was heated under reflux for 3 h. After removal of the solvent *in vacuo*, the residue was extracted with AcOEt, then the extract was washed with brine, dried and evaporated. The product was recrystallized from EtOH, giving 13a (1.85 g, 64%) as yellow needles. MS m/z: 289 (M⁺). IR (KBr): 3400, 2970, 1740 cm⁻¹. ¹H-NMR (CDCl₃) δ : 1.40 (6H, t, J=7.0 Hz, $-N(CH_2CH_3)_2$), 1.42 (3H, t, J=7.0 Hz, $-CH_2CH_3$), 3.69 (4H, q, J=7.0 Hz, $-N(CH_2CH_3)_2$), 4.38 (2H, q, J=7.0 Hz, $-CH_2CH_3$), 7.10—7.30 (3H, m, C_5 -, C_6 -, C_7 -H), 8.85 (1H, d, J=10 Hz, C_4 -H).

2-Oxo-2H-cyclohepta[b]**furan-3-carbaldehyde (14)** A mixture of **8a** (23 g, 0.11 mol) and 75% $\rm H_2SO_4$ (230 ml) was warmed at about 100 °C for 30 min and poured into ice-water. The precipitates were collected by filtration, and recrystallization from EtOH gave 2*H*-cyclohepta[b]furan-2-one (11.2 g, 68.8%) as yellow prisms (mp 72 °C). The obtained 2*H*-cyclohepta[b]furan (3 g, 0.021 mol) was added to a mixture of POCl₃ (22 ml) and dry *N*,*N*-dimethylformamide (DMF) (25 ml) at 0 °C. The mixture was stirred at room temperature for 2h and then poured into ice-water (100 ml). The aqueous layer was adjusted to pH 8 with 10% NaOH and extracted with CHCl₃. The organic layer was washed with water, dried, and then evaporated. The product was recrystallized from

TABLE V. Spectral Data for the 2H-Cyclohepta[b]furan-2-ones

Product	MS m/z (M +)	1 H-NMR (CDCl $_{3}$) δ (ppm) J (Hz)
8b	246	1.38 (6H, d, $J = 6.8$), 3.14 (1H, sept, $J = 6.8$), 4.00 (3H, s), 7.40—7.80 (3H, m), 9.10 (1H, s)
12a	248	$1.42.(211 + 1-7.0) \cdot 4.18.(3H + s) \cdot 4.40.(2H + 0.1=7.0) \cdot 7.20-7.50.(3H, m), 8.90.(1H, 0.1=10.0)$
12c	290	1.42 (SH, t, $J = 7.0$), 4.16 (SH, 3), 4.16 (SH, 4), 4.17 (SH, s), 4.40 (2H, q, $J = 7.0$), 7.20—7.32 (2H, 1.32 (6H, d, $J = 6.8$), 1.40 (3H, t, $J = 7.0$), 3.00 (1H, sept, $J = 6.8$), 4.17 (3H, s), 4.40 (2H, q, $J = 7.0$), 7.20—7.32 (2H,
		m), 8.92 (1H, s)
12d	304	m), 8.92 (1H, 8) 1.32 (6H, d, $J=6.8$), 1.40 (3H, t, $J=7.0$), 1.48 (3H, t, $J=7.0$), 3.02 (1H, sept, $J=6.8$), 4.00 (2H, q, $J=7.0$), 4.20 (2H,
		q, J=7.0, 7.20-7.40 (2H, m), 8.95 (1H, s)
12e	318	(4, J = 7.0), 7.20 = 7.40 (211, m), 6.75 (111, 6), 1.42 (3H, t, $J = 7.0$), 1.70 (2H, m), 3.02 (1H, sept, $J = 6.8$), 1.07 (3H, t, $J = 7.0$), 1.33 (6H, d, $J = 6.8$), 1.42 (3H, t, $J = 7.0$), 1.70 (2H, m), 3.02 (1H, sept, $J = 6.8$),
		4.20—4.60 (2H, m), 4.38 (2H, q, $J=7.0$), 7.20—7.50 (2H, m), 8.96 (1H, s)
12f	318	4.20—4.00 (2H, III), 4.36 (2H, q, $J=7.0$), 7.20 (2H, III), 4.31 (6H, d, $J=6.8$), 1.41 (6H, d, $J=7.0$), 1.42 (3H, t, $J=7.0$), 3.00 (1H, sept, $J=6.8$), 4.41 (2H, q, $J=7.0$), 5.01
		(1H, sept, $J=7.0$), $7.20-7.50$ (2H, m), 8.95 (1H, s)
12g	290	1.00 (3H, t, $J=7.0$), 1.40 (3H, t, $J=7.0$), 1.71—2.00 (4H, m), 4.40 (2H, q, $J=7.0$), 4.20—4.60 (2H, m), 7.20—7.60
		(3H, m), 8.95 (1H, d, $J=10.0$) 0.98 (3H, t, $J=7.0$), 1.32 (6H, d, $J=6.8$), 1.42 (3H, t, $J=7.0$), 1.30—2.00 (4H, m), 3.00 (1H, sept, $J=6.8$), 4.38 (2H,
12h	332	0.98 (3H, t, J = 7.0), 1.32 (6H, d, J = 0.8), 1.42 (3H, t, J = 7.0), 1.30 - 2.00 (4H, m), 5.00
		q, $J=7.0$), $4.20-4.60$ (2H, m), $7.20-7.40$ (2H, m), 8.95 (1H, s) 1.41 (3H, t, $J=7.0$), 2.42 (3H, s), 4.41 (2H, q, $J=7.0$), $7.20-7.70$ (3H, m), 8.88 (1H, d, $J=10.0$)
12i	276	1.41 (3H, t, $J = 7.0$), 2.42 (3H, 8), 4.41 (2H, q, $J = 7.0$), 7.20 7.70 (3H, m), 6.85 (2H, q), 1.40 (3H, t, $J = 7.0$), 1.90—2.20 (4H, m), 3.80—4.00 (4H, m), 4.38 (2H, q, $J = 7.0$), 7.00—7.30 (3H, m), 8.85 (2H, d,
13b	287	
	2.42	J=10.0) 1.29 (6H, d, $J=6.8$), 1.42 (3H, t, $J=7.0$), 1.50—1.90 (6H, m), 3.02 (1H, sept, $J=6.8$), 3.40—3.70 (4H, m), 4.80 (2H,
13c	343	7.70) 7.20, 7.40 (2H m) 8.85 (1H s)
12.1	345	q, $J = 7.0$), $7.20 = 7.40$ (211, iii), 8.65 (111, 3) 1.31 (6H, d, $J = 6.8$), 1.41 (3H, t, $J = 7.0$), 3.02 (1H, sept, $J = 6.8$), 3.50—3.70 (4H, m), 3.80—4.00 (4H, m), 4.40 (2H,
13d	343	7.70\ 7.20\ 7.40\(\text{(2H m)}\)\(\text{ 8.92}\(\text{(1H s)}\)
13e	357	q, $J = 7.0$), $7.20 = 7.40$ (211, m), 8.22 (111, 3) 1.30 (6H, d, $J = 6.8$), 1.42 (3H, t, $J = 7.0$), 1.51 = 2.20 (10H, m), 3.0 (1H, sept, $J = 6.8$), 3.30 = 3.70 (1H, m), 4.39 (2H,
136	331	7.70) 5.79 (1H brs) 7.10—7.30 (2H m) 8.86 (1H s)
13f	304	1.20 (211 + 1.70) 3.03 (6H s) 3.30 - 3.70 (5H m) 4.30 (6H, q, $J = 7.0$), $7.30 - 7.80 (3H, m)$, $8.30 (1H, q, J = 10.0)$
13g	332	1.39 (3H, t, $J=7.0$), 3.65 (6H, s), 3.50 (5H, m), 4.30 (6H, q, $J=7.0$), 7.20—7.80 (3H, m), 8.65 (1H, d, 1.39 (3H, t, $J=7.0$), 1.48 (6H, t, $J=7.0$), 3.20—3.70 (5H, m), 4.30 (6H, q, $J=7.0$), 7.20—7.80 (3H, m), 8.65 (1H, d, m)
136	222	r 10.0)
13h	346	J = 10.01 1.32 (6H, d, $J = 6.8$), 1.38 (3H, t, $J = 7.0$), 3.00 (1H, sept, $J = 6.8$), 3.04 (6H, s), 3.30—3.68 (5H, m), 4.32 (2H, q,
10		r 70) 720 725 (2H m) 8 00 (1H s)
13i	330	J = 7.0), $7.20 - 7.33$ (2H, m), 6.50 (1H, s) 1.38 (3H, t, $J = 7.0$), $2.00 - 2.30$ (4H, m), $3.00 - 3.40$ (5H, m), $3.80 - 4.00$ (4H, m), 4.20 (2H, q, $J = 7.0$), $7.30 - 7.70$
		(2H m) = 9.48 (1H d. I - 10.0)
17b	245	1.58 (4H, br s), 2.80 (4H, br s), 4.00 (1H, s), 7.40—7.70 (4H, m), 8.50 (1H, d, J=10.0)
17c	287	1.36 (4H, br s), 2.80 (4H, br s), 3.03 (1H, m), 8.20 (1H, s) 1.36 (6H, d, J=6.8), 1.60 (4H, br s), 2.80 (4H, br s), 3.03 (1H, m), 8.20 (1H, s)
17d	319	1.09—3.05 (7H, m), 3.28 (3H, s), 4.20 (2H, br s), 7.20—8.20 (9H, m)
17e	325	2.18 (6H s) 2.95 (2H hrs) 3.45 (3H, m), 6.30—7.30 (8H, m)
17f	424	2.30—2.90 (11H, m), 3.49 (2H, s), 3.86 (3H, s), 3.89 (3H, s), 4.66 (1H, m), 6.60—7.70 (8H, m)
17g	359	2.15 (3H, s), 3.30 (1H, br s), 4.10—4.20 (3H, m), 6.40—7.80 (10H, m)

EtOH, giving 14 (2.1 g, 63.3%) as yellow needles. MS m/z: 174 (M⁺). mp 219 °C. IR (KBr): 1760, 1740, 1700 cm⁻¹. ¹H-NMR (CDCl₃) δ: 7.30—7.90 (4H, m, C₅-, C₆-, C₇-H), 9.00 (1H, d, J=10 Hz, C₄-H), 10.05 (1H, s, CHO).

3-Chloromethyl-2*H*-cyclohepta[*b*] furan-2-one (16) NaBH₄ (8 g. 0.21 mol) was added to a stirred solution of 14 (25 g, 0.14 mol) in MeOH (30 ml) under ice-cooling. The mixture was stirred for 2 h at room temperature, then the solvent was evaporated *in vacuo*. H₂O (30 ml) was added to the residue and the mixture was extracted with CHCl₃. The organic layer was washed with water, dried, and then evaporated. The product was recrystallized from EtOH, giving 3-hydroxymethyl-2*H*-cyclohepta[*b*] furan-2-one 15 (23 g, 91%). MS *m/z*: 176 (M⁺). mp 143 °C (dec.). IR (KBr): 3450, 1720, 1700 cm⁻¹. A solution of 15 (18 g, 1.01 mol) in benzene (100 ml) was treated with SOCl₂ (17 g, 0.15 mol) and refluxed for 2.5 h. The reaction mixture was concentrated *in vacuo*. The residue was washed with Et₂O to afford 16 (14 g, 72%). MS *m/z*: 195 (M⁺). mp 208—210 °C (dec.). IR (KBr): 1730, 1700, 1600 cm⁻¹.

3-Dimethylamimomethyl-5-isopropyl-2*H*-cyclohepta[*b*]furan-2-one (17a) A mixture of 5-isopropyl-3-chloromethyl-2*H*-cyclohepta[*b*]-furan-2-one (1.5 g, 6.3 mmol) and dimethylamine (0.61 g, 13.3 mmol) in MeOH (30 ml) was stirred at room temperature for 3 h and evaporated *in vacuo*. The residue was dissolved in CHCl₃ and this solution was washed with H₂O, dried, and then evaporated. The residue was recrystallized from EtOH, giving 17a (1.13 g, 72.6%). MS m/z: 245 (M⁺). IR (KBr): 3450, 3400, 1740 cm⁻¹. ¹H-NMR (CDCl₃) δ : 1.38 (6H, d, J=6.8 Hz, iso-Pr-CH₃), 2.86 (6H, s, -(CH₃)₂), 3.15 (1H, sept, J=6.8 Hz, iso-Pr-CH), 4.10 (2H, br s, -CH₂-), 7.10—7.40 (4H, m, C₆-, C₇-, C₈-H), 8.10 (1H, s, C₄-H).

Diethyl 2-Oxo-2*H*-cyclohepta[*b*] furan-3-ylmethylphosphonate (18) A mixture of 15 (176 mg, 0.1 mol) and P(EtO)₃ (70 mg, 0.1 mmol) was stirred for 20 min at 150 °C. After cooling, the reaction mixture was poured into

water and extracted with ether. The extract was washed with water, dried, and then evaporated. The residue was recrystallized from ether–hexane to give **18** (200 mg, 71%). MS m/z: 296 (M⁺). IR (KBr): 2950, 1750, 1250 cm⁻¹. ¹H-NMR (CDCl₃) δ : 1.30 (6H, t, J=7.0 Hz, –(CH₂CH₃)₂), 3.05 (2H, d, J=22 Hz, –CH₂–), 4.11 (4H, q, J=7.0 Hz, –(CH₂CH₃)₂), 6.60—7.60 (5H, m, Ar-H).

Positive Inotropic Action In this study, atrial muscle of male Hartley guinea-pigs was used. The preparations were dissected and mounted vertically in a 20 ml tissue bath containing Krebs–Henseleit buffer solution of the following composition (millimolar concentrations): NaCl, 118.4; KCl, 4.7; MgSO₄·7H₂O, 2.4; KH₂PO₄, 1.2; CaCl₂·H₂O, 1.9; NaHCO₃, 25.0 and glucose 5.6; the solution was bubbled with a gas mixture of 95% O₂ and 5% CO₂ and maintained at 32 °C, pH 7.4. One end of the tissue was connected to a force displacement transducer by a silk ligature while the other was secured to a tissue holder. Tension was recorded isometrically through a force displacement transducer.

Cyclic AMP Phosphodiesterase Assay The reaction mixture for the assay of cyclic AMP phosphodiesterase was composed of (in final concentrations) 250 mM Tris–HCl (pH 7.5), 15 mM (CH₃COO)₂ Mg, 10 mM 5'-AMP, 1.25 nCi of [¹⁴C]cyclic AMP, test drug and bovine heart cyclic AMP phosphodiesterase in a volume of 0.25 ml. The mixture excluding [¹⁴C]cyclic AMP was preincubated for 10 min at 37 °C, then the reaction was started by addition of [¹⁴C]cyclic AMP. Incubation was continued for 30 min at 37 °C, and was stopped by the addition of 200 μ l of ZnSO₄ (170 mM) and Ba(OH)₂ (150 mM). After centrifugation (2000 × g, 10 min), 300 μ l aliquots of the clear supernatant solution were transferred to counting vials, and 7 ml of scintillation fluid was added.

References

 "AMA Drug Evaluations," 4th ed. by Am. Med. Assoc., John Wiley & Sons, Inc., U.S.A., 1980, p. 498.

- 2) For a review, see: P. W. Erhardt, J. Med. Chem., 30, 231 (1987).
- R. E. Weishaar, M. H. Cain, J. A. Bristol, J. Med. Chem., 28, 537 (1985).
- 4) A. A. Alousi, J. M. Canter, M. J. Montenaro, D. J. Fort, R. A. Ferrari, J. Cardiovasc. Pharmacol., 5, 792 (1983).
- J. A. Bristol, J. Sircar, W. H. Moss, D. B. Evans, R. E. Weishaar, J. Med. Chem., 27, 1099 (1984).
- A. G. Gillman, L. S. Goodman, T. W. Rall, F. Murad, "Goodman and Gillman's The Pharmacological Basis of Therapeutics," 7th ed., Macmillan Publishing Company, N.Y., 1985, Chapter 36, p. 744.
- 7) T. Yanagisawa, S. Wakabayashi, T. Tomiyama, M. Yasunami, K. Takase, *Chem. Pharm. Bull.*, **36**, 641 (1988).
- 8) T. Yanagisawa, K. Kosakai, T. Tomiyama, M. Yasunami, K. Takase, *Chem. Pharm. Bull.*, **38**, 3355 (1990).
- 9) R. Thomas, J. Boutagy, A. Gelbart, J. Pharm. Sci., 63, 1699

- (1984).
- T. Nozoe, K. Takase, M. Kato, T. Nogi, Tetrahedron, 27, 6023 (1971).
- 11) a) K. Takagi, I. Takayanagi, Jpn. J. Pharmacol., 20, 92 (1970); b) G. Bertaccini, G. Coruzz, Br. J. Pharmacol., 72, 197 (1981).
- J. Bagli, T. Bogri, B. Palameta, S. Rakhit, S. Peseclcis, J. McQuillan,
 D. K. H. Lee, J. Med. Chem., 31, 814 (1988).
- P. Schuhmacker, K. Gree, E. A. Noack, Eur. J. Pharmacol., 95, 71 (1983).
- 14) J. W. Herzig, K. Feile, J. C. Ruegg, Arzneim. Forsh., 31, 188 (1981).
- 15) M-C. Forest, P. Lahomate, M. Martin, G. Nadler, M. J. Quimion, R. G. Zimmermann, J. Med. Chem., 35, 163 (1992).
- T. Sato, Bull. Chem. Res. Inst. Non-aqueous Solutions, Tohoku Univ., 8, 47 (1959).