

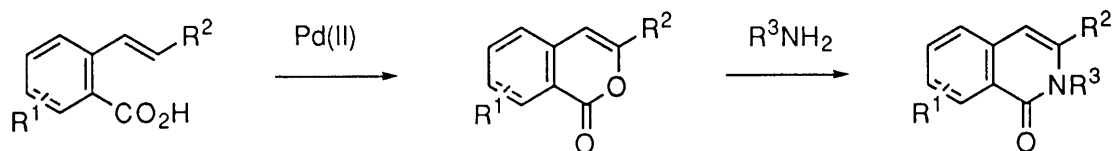
SYNTHESIS OF 3-SUBSTITUTED ISOCOUMARINS THROUGH ACYLOXYPALLADATION OF *o*-ALKENYLBENZOIC ACIDS

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Cyclization of *o*-alkenylbenzoic acids in the presence of Pd catalyst and benzoquinone led to 3-substituted isocoumarins in high yield. The isocoumarins obtained were converted to isoquinolones by treatment with primary amines.

KEYWORDS isocoumarin; acyloxypalladation; *o*-alkenylbenzoic acid; palladium(II); benzoquinone; isoquinolone

A wide variety of isocoumarins occur naturally, and the isocoumarin ring system also plays an important role as an intermediate for a synthesis of several natural products.¹⁾ Recently, several biologically interesting 3-substituted isocoumarins — for example, achlisocoumarin I²⁾ and thunberginol A³⁾ — were isolated. Various methods are known for the synthesis of isocoumarin-containing structures.⁴⁾ Palladium-catalyzed intramolecular addition of a carboxylic acid to an alkene becomes an efficient route toward lactones.⁵⁾ Although this method has been applied to a synthesis of 3-substituted isocoumarins from *o*-alkenylbenzoic acids,⁶⁾ the catalytic use of palladium reagent and the use of more highly functionalized benzoic acid derivatives have not yet been described. We wish to report in this paper Pd-catalyzed cyclization of *o*-alkenylbenzoic acid derivatives and a further synthesis of isoquinolone from the isocoumarine obtained.



Treatment of 4,5-dimethoxy-2-styrylbenzoic acid (**1a**)^{7, 8)} with 1 eq of PdCl₂(CH₃CN)₂ in the presence of triethylamine provided 3-phenylisocoumarin **2a** and benzylidenephthalide **3a** in 80% yield (**2a** : **3a** = 96 : 4). In order to improve this reaction to a catalytic one, the Pd(0) generated should be reverted back to Pd(II) by the addition of an oxidant. The reactions were carried out with 5 mol% of palladium reagent and an oxidant such as Cu(II) and O₂ at room temperature in THF for 12 h, giving **2a** and **3a**. However, neither the yield nor the cyclization selectivity were satisfied. Benzoquinone was found to be the best oxidant. The reaction with benzoquinone (1.1 eq) for 2 h

gave the isocoumarin **2a** in quantitative yield with high cyclization selectivity (Table I, entry 1).

The latter method was applied to cyclization of several *o*-alkenylbenzoic acids. The results are shown in Table I.⁹⁾ In the case of styryl compounds, the substituent on aromatic ring (R) affected cyclization selectivity. In the order of MeO > Me > CHO, the selectivity decreased according to electron-withdrawing ability (entry 2-4). On the other hand, *o*-hexenylbenzoic acid **1e** gave exclusively 3-butylisocoumarin **2e** in good yield (entry 5).

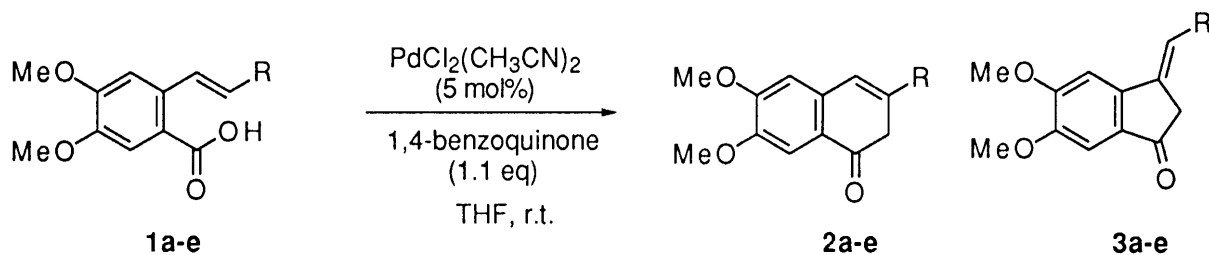
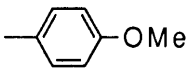
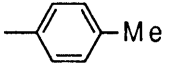
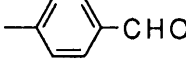
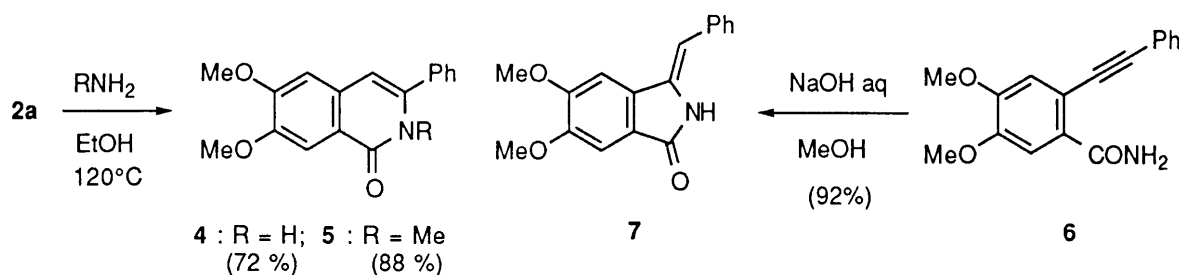


Table I. $\text{PdCl}_2(\text{CH}_3\text{CN})_2$ Catalyzed Cyclization of **1** in the Presence of Benzoquinone

Entry	1	R	Reaction time (h)	2 : 3	Yield (%)
1	1a	Ph	2	99 : 1	98
2	1b		2	93 : 7	90
3	1c		2	86 : 14	73
4	1d		2	74 : 26	82
5	1e	<i>n</i> -Bu	6	100 : 0	83

The reaction proceeded through acyloxypalladation followed by β -hydride elimination. Although the precise mechanism of 6-*endo* and 5-*exo* selectivity has not been clarified, the selectivity would be affected by the electron demand of carbon-carbon double bond.¹⁰⁾

The isocoumarins obtained were easily converted to the corresponding isoquinolones by treatment with primary amines. For example, treatment of **2a** with ammonia or methylamine in EtOH gave the isoquinolone **4** or **5**. This process will be very useful because direct cyclization of *o*-alkynylbenzamide **6** in basic medium gave only the benzylidenephthalimide **7**.¹¹⁾



Thus, we have identified a successful palladium-catalyzed cyclization method for the synthesis of isocoumarins.¹²⁾ The present method is easy to carry under mild conditions. This process is also applicable to the synthesis of isoquinolones, synthetic intermediates for benzo[c]phenanthridine alkaloids.¹³⁾

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REFERENCES AND NOTES

- 1) a) R. D. Barry, *Chem. Rev.*, 229 (1964); b) R. Livingstone, "Rodd's Chemistry of Carbon Compounds," Vol. IV, ed. by S. Caffey, Elsevier, New York, 1977, Part E, pp 290-297.
- 2) M. Mizuno, S. Yoshida, M. Iimura, T. Tanaka, F. A. Lang, K. Goto, *Chem. Pharm. Bull.*, **38**, 2075 (1990).
- 3) a) M. Yoshikawa, E. Uchida, N. Chatani, N. Murakami, J. Yamahara, *Chem. Pharm. Bull.*, **40**, 3121 (1992); b) M. Yoshikawa, E. Harada, H. Matsuda, T. Murakami, J. Yamahara, N. Murakami, *Chem. Pharm. Bull.*, **41**, 2069 (1993).
- 4) a) M. Yoshikawa, E. Harada, N. Yagi, Y. Okuno, O. Muraoka, H. Aoyama, N. Murakami, *Chem. Pharm. Bull.*, **42**, 721 (1994); b) S. Ohta, Y. Kamata, T. Inagaki, Y. Masuda, S. Yamamoto, M. Yamashita, I. Kawasaki, *Chem. Pharm. Bull.*, **41**, 1188 (1993); c) F. M. Hauser, V. M. Baghdanov, *J. Org. Chem.*, **53**, 4676 (1988) and references cited therein.
- 5) For a general review of the reactions of metal-activated alkenes with heteroatom nucleophiles, see: L. S. Hegedus, "Comprehensive Organic Synthesis," Vol. 4, ed. by B. M. Trost, Pergamon Press, Oxford, 1991, pp551-569.
- 6) a) D. E. Korte, L. S. Hegedus, R. K. Wirth, *J. Org. Chem.*, **42**, 1329 (1977); b) A. Kasahara, T. Izumi, K. Sato, M. Maemura, T. Hayasaka, *Bull. Chem. Soc. Jpn.*, **50**, 1899 (1977); c) R. C. Larock, S. Varaprath, H. H. Lau, C. A. Fellows, *J. Am. Chem. Soc.*, **106**, 5274 (1984).
- 7) The compounds **1a-1d** were prepared from methyl 2-iodo-4,5-dimethoxybenzoate and styrene derivatives by Heck reaction($\text{Pd}(\text{OAc})_2$, (*n*-Bu)₄NCl, DMF, 100°C / 12 h), followed by hydrolysis.
- 8) We chose 4,5-dimethoxy derivatives **1** because our final synthetic targets are benzo[c]-phenanthridine alkaloids,¹³⁾ most of which possess the same oxygenation pattern on one aromatic ring as **1**.
- 9) All new compounds were satisfied by elemental analysis and spectral data.
- 10) In order to clarify the mechanism, the reactions with various types of substituted 2-styrylbenzoic acids are in progress.
- 11) N. Nishiwaki, M. Komatsu, Y. Ohshiro, *Synthesis*, 41 (1991); N. Nishiwaki, S. Minakata, M. Komatsu, Y. Ohshiro, *Synlett*, 273 (1990).
- 12) Recently, $\text{Pd}(\text{OAc})_2$ -catalyzed cyclization of alkenoic acid was reported. R. C. Larock, T. R. Hightower, *J. Org. Chem.*, **58**, 5298 (1993). In this paper, *o*-propenylbenzoic acid was treated with 5 mol % of $\text{Pd}(\text{OAc})_2$ and NaOAc in DMSO under O₂ atmosphere at 80°C to give 3-methylisocoumarin in 80% yield. This condition was applied to cyclization of **1a**, giving **2a** and **3a** in a ratio of 99 : 1 (52% yield).
- 13) M. Hanaoka, C. Mukai, "Studies in Natural Products Chemistry," Vol. 14, ed. by Atta-ur-Rahman, Elsevier, Amsterdam, 1994, pp 769-803.

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