## Structures of Six New Eremophilenolides from the Rhizomes of *Petasites japonicus* MAXIM.<sup>1)</sup>

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Six new eremophilenolides,  $3\beta$ -hydroxyeremophil-7(11)-en-12, $8\beta$ -olide (1),  $3\beta$ -hydroxy- $6\beta$ -methoxyeremophil-7(11)-en-12, $8\beta$ -olide (2),  $3\beta$ -hydroxy- $6\beta$ , $8\alpha$ -dimethoxyeremophil-7(11)-en-12, $8\beta$ -olide (3), the mixture of  $3\beta$ , $8\alpha$ -dihydroxy- $6\beta$ -tigloyloxyeremophil-7(11)-en-12, $8\beta$ -olide (4) and  $3\beta$ , $8\beta$ -dihydroxy- $6\beta$ -tigloyloxyeremophil-7(11)-en-12, $8\alpha$ -olide (5), and  $6\beta$ -angeloyloxy- $8\beta$ -hydroxy-3-oxoeremophil-7(11)-en-12, $8\alpha$ -olide (6), were isolated from the dried rhizomes of *Petasites japonicus* Maxim. (Compositae). The structures of these compounds were elucidated on the basis of spectroscopic evidence.

Keywords Petasites japonicus; Compositae; sesquiterpenoid; eremophilenolide

The rhizomes of Petasites japonicus MAXIM. (Compositae) have been used for the treatment of tonsillitis, contusion and poisonous snake bite in China.2) In previous papers, we reported on the structure elucidation of eremophilenolides, 3) triterpenoids, 4) anthraquinones, 4) and phenolic compounds.<sup>5)</sup> The present paper describes the further isolation and structure elucidation of six new eremophilenolides:  $3\beta$ -hydroxyeremophil-7(11)-en-12,8 $\beta$ olide (1),  $3\beta$ -hydroxy- $6\beta$ -methoxyeremophil-7(11)-en-12,8 $\beta$ -olide (2), 3 $\beta$ -hydroxy-6 $\beta$ ,8 $\alpha$ -dimethoxyeremophil-7(11)-12,8 $\beta$ -olide (3), the mixture of  $3\beta$ ,8 $\alpha$ -dihydroxy-6 $\beta$ tigloyloxyeremophil-7(11)-en-12,8 $\beta$ -olide (4) and 3 $\beta$ ,8 $\beta$ dihydroxy-6β-tigloyloxyeremophil-7(11)-en-12,8α-olide (5), and  $6\beta$ -angeloyloxy- $8\beta$ -hydroxy-3-oxoeremophil-7(11)-en-12,8α-olide (6). Extraction and isolation were carried out as described in the Experimental section.

Compound 1,  $C_{15}H_{22}O_3$ ,  $[\alpha]_D - 136.0^\circ$ , was isolated as colorless needles, mp 168—169 °C. The IR spectrum suggested the presence of a hydroxyl group (3471 cm<sup>-1</sup>), an  $\alpha,\beta$ -unsaturated- $\gamma$ -lactone (1741 cm<sup>-1</sup>) and a double bond (1684 cm<sup>-1</sup>). The UV spectrum also suggested the presence of an  $\alpha,\beta$ -unsaturated- $\gamma$ -lactone ( $\lambda_{max}$ : 223 nm). The <sup>1</sup>H- (Table I) and <sup>13</sup>C-NMR (Table II) spectra were

similar to those of the eremophilenolides previously isolated from the rhizomes of P. japonicus<sup>3)</sup> and showed signals due to a tertiary methyl group [ $\delta_{\rm H}$  0.85 (s, H-15),  $\delta_{\rm C}$  25.0 (C-15)], a secondary methyl group [ $\delta_{\rm H}$  0.98 (d, J=7.3 Hz, H-14),  $\delta_{\rm C}$  7.5 (C-14)], an olefinic methyl group  $[\delta_{\rm H} \ 1.80 \ ({\rm dd}, \ J = 1.5, \ 1.5 \,{\rm Hz}, \ {\rm H}\text{-}13), \ \delta_{\rm C} \ 8.1 \ ({\rm C}\text{-}13)], \ {\rm an}$ AB-type methylene [ $\delta_{\rm H}$  2.20 (d,  $J = 13.9 \, {\rm Hz}$ , H-6 $\beta$ ), 2.75 (dd, J=13.9, 1.5 Hz, H-6 $\alpha$ ),  $\delta_{\rm C}$  33.7 (C-6)], a hydroxybearing methine [ $\delta_{\rm H}$  4.13 (ddd, J=11.4, 4.4, 4.4 Hz, H-3 $\alpha$ ),  $\delta_{\rm C}$  68.9 (C-3)], an oxygenated methine [ $\delta_{\rm H}$  4.80 (m, H-8 $\alpha$ ),  $\delta_{\rm C}$  77.7 (C-8)] and  $\alpha,\beta$ -unsaturated- $\gamma$ -lactone [ $\delta_{\rm C}$  122.3 (C-11), 161.8 (C-7), 174.7 (C-12)], establishing that the lactone belongs to a sesquiterpene of the eremophilanetype. Naya et al.6) reported that for 8α-methoxyeremophilenolide derivatives, the chemical shifts due to the secondary methyl group (H-14) are downfield from those due to the tertiary methyl group (H-15), whereas this relationship is reversed in the  $8\beta$ -series. These variations in the chemical shifts may be explained similarly in terms of the effect due to the alteration in geometry of the skeleton observed in the steroid field: by bending rings away from the angular methyl group, or by blocking the angular methyl's view over the remaining skeleton the

Table I. <sup>1</sup>H-NMR Chemical Shifts (CDCl<sub>3</sub>, 400 MHz)

Proton	1	2	3	4a	5a	6
3α	4.13 ddd	4.07 ddd	4.01 ddd	4.97 ddd	4.99 ddd	
	(J=11.4, 4.4, 4.4)	(J=11.4, 4.4, 4.4)	(J=11.4, 4.4, 4.4)	(J=11.7, 4.4, 4.4)	(J=3.0, 3.0, 2.9)	
6α	2.75  dd  (J = 13.9, 1.5)	4.35 d (J=1.5)	4.28 d (J=1.5)	6.19 d (J=1.5)	5.76 s	5.61 s
6β	2.20 d (J=13.9)	,				
8α	4.80 m	4.77 m				
9α	2.17 ddd	2.17 ddd				
	(J=12.8, 6.6, 2.2)	(J=12.8, 6.6, 2.2)				
13	1.80  dd  (J=1.5, 1.5)	1.97 dd $(J=1.5, 1.5)$	2.01 d (J=1.5)	$1.83 \mathrm{d}  (J = 1.5)$	2.03 s	2.02 s
14	$0.98  d  (\hat{J} = 7.3)$	$0.95 \mathrm{d} (J = 7.3)$	0.94 d (J=7.3)	$0.96 \mathrm{d}  (J = 7.3)$	$0.95 \mathrm{d}  (J = 7.0)$	$1.02 \mathrm{d} (J = 6.6)$
15	0.85 s	0.83 s	0.83 s	0.97 s	1.26 s	1.00 s
3′				6.97  qq  (J=7.0, 1.5)	$6.84 \mathrm{qq} (J=7.0,1.5)$	$6.20 \mathrm{qq} (J = 7.3,  1.5)$
4′				$1.87 \mathrm{dq} (J = 7.0,  1.1)$	1.81 dq $(J=7.0, 1.1)$	$2.04 \mathrm{dq} (J = 7.3,  1.5)$
5′				$1.93 \mathrm{dq} (J = 1.5,  1.1)$	$1.84 \mathrm{dq} (J = 1.5,  1.1)$	1.92 dq $(J=1.5, 1.5)$
6-OCH <sub>3</sub>		3.49 s	3.47 s			
8-OCH <sub>3</sub>			3.20 s			
COCH <sub>3</sub>				2.00 s	1.89 s	
COCH <sub>3</sub>				2.10 s	2.08 s	

Coupling constants (J in Hz) are given in parentheses

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TABLE II. 13C-NMR Chemical Shifts (CDCl<sub>3</sub>, 100 MHz)

Carbon	1	2	3	4a	5a	6				
1	27.3	27.3	26.9	26.3	21.1	26.3				
2 3	29.3	28.9	28.7	25.3	24.6	38.7				
	68.9	68.7	68.8	71.5	73.1	209.9				
4	44.7	38.2	38.8	$35.5^{a}$	32.3	45.1				
5	41.0	46.6	47.2	45.6	41.9	46.6				
6	33.7	80.5	79.8	70.2	70.2	70.6				
7	161.8	160.9	157.2	154.2	149.7	151.1				
8	77.7	77.4	106.9	104.6	103.4	103.5				
9	34.6	35.3	38.1	$35.6^{a}$	38.0	36.7				
10	35.2	35.1	34.9	35.1	34.6	34.7				
11	122.3	121.9	126.2	126.2	129.8	129.4				
12	174.7	174.6	171.4	170.8	$170.3^{c}$	170.5				
13	8.1	8.5	8.4	$8.5^{b}$	8.9	9.0				
14	7.5	7.4	7.6	$8.6^{b}$	12.3	8.0				
15	25.0	19.5	18.9	19.7	18.3	18.9				
1'				166.4	166.5	166.8				
2'				127.9	127.9	126.7				
3'				139.0	137.5	141.1				
4′				14.7	14.4	16.0				
5'				12.2	12.1	20.6				
6-OCH <sub>3</sub>		60.0	59.6							
8-OCH <sub>3</sub>			50.1							
COCH <sub>3</sub>				21.3	21.3					
COCH₃				22.2	21.6					
COCH <sub>3</sub>				168.1	168.3					
COCH₃				170.0	$170.3^{c)}$					

a-b) Assignments may be reversed. c) Signals were overlapped.

methyl signal may cause a downfield shift. 6) Naya and his colleagues also reported that the homoallylic coupling  $(J=1.0-1.5 \,\mathrm{Hz})$  between the olefinic methyl group (H-13) and H-6 $\alpha$  found in the 8 $\alpha$ -series, is absent in the 8 $\beta$ -series. The value of the optical rotation of the  $8\beta$ -series, which had a steroidal conformation, was positive, and that of the 8α-series, which had a non-steroidal conformation, was negative. 6) The <sup>1</sup>H-NMR spectrum of 1 showed a singlet of the tertiary methyl group (H-15) at  $\delta$  0.85 and a doublet of the secondary methyl group (H-14) at  $\delta$  0.98 (J = 7.3 Hz), as well as the homoallylic coupling  $(J=1.5 \,\mathrm{Hz})$  of the olefinic methyl group (H-13) with H-6α. Furthermore, the value of the optical rotation was negative. These data indicated that 1 exists in a non-steroidal conformation. The position of the hydroxyl group was determined to be at the C-3 $\beta$  by comparing the chemical shift, coupling pattern and constants of the hydroxy-bearing methine proton of 1 with those of  $3\beta$ -hydroxy- $6\beta$ -acyleremophil-7(11)-en-12,8 $\beta$ -olides.<sup>3)</sup> On the basis of the above evidence, the structure of 1 was determined to be  $3\beta$ -hydroxyeremophil-7(11)-en-12,8 $\beta$ -olide. Compound 1 was isolated from a natural source for the first time, although 1 has already been synthesized by Naya et al.<sup>7)</sup>

Compound 2,  $C_{16}H_{24}O_4$ ,  $[\alpha]_D - 189.6^\circ$ , was isolated as colorless needles, mp 180—181 °C. The IR spectrum suggested the presence of a hydroxyl group (3474 cm<sup>-1</sup>), an  $\alpha,\beta$ -unsaturated- $\gamma$ -lactone (1746 cm<sup>-1</sup>) and a double bond (1678 cm<sup>-1</sup>). The UV spectrum also suggested the presence of an  $\alpha,\beta$ -unsaturated- $\gamma$ -lactone ( $\lambda_{max}$ : 219 nm). The <sup>1</sup>H- and <sup>13</sup>C-NMR spectra of 2 were virtually identical to those of 1 except for the presence of a methoxyl group [ $\delta_H$  3.49 (s),  $\delta_C$  60.0]. The position of the methoxyl group was determined by <sup>1</sup>H-detected heteronuclear mul-

tiple bond correlation (HMBC) and nuclear Overhauser effect correlation spectroscopy (NOESY). In the HMBC spectrum, a cross peak was observed between the methoxyl group at  $\delta$  3.49 and the C-6 at  $\delta$  80.5, so that the methoxyl group is attached at the C-6. In the NOESY spectrum, each signal of H-6 $\alpha$  and H-8 $\alpha$  showed a correlation, so that the methoxyl group is  $\beta$ -oriented. On the basis of this evidence, the structure of 2 was determined to be  $3\beta$ -hydroxy- $6\beta$ -methoxyeremophil-7(11)-en- $12.8\beta$ -olide.

Compound 3,  $C_{17}H_{26}O_5$ ,  $[\alpha]_D - 86.7^\circ$ , was isolated as colorless oil. The IR spectrum suggested the presence of a hydroxyl group (3509 cm<sup>-1</sup>), an  $\alpha,\beta$ -unsaturated- $\gamma$ -lactone (1757 cm<sup>-1</sup>) and a double bond (1680 cm<sup>-1</sup>). The UV spectrum also suggested the presence of an  $\alpha,\beta$ unsaturated- $\gamma$ -lactone ( $\lambda_{max}$ : 225 nm). The <sup>1</sup>H- and <sup>13</sup>C-NMR spectra of 3 were virtually identical to those of 2 except for the presence of one more methoxyl group [ $\delta_{\rm H}$  3.20 (s),  $\delta_{\rm C}$  50.1], the position of which was determined as follows. In the HMBC spectrum, a cross peak was observed between the methoxyl group at  $\delta$  3.20 and the C-8 at  $\delta$  106.9, so that this methoxyl group is attached at the C-8. In the NOESY spectrum, each signal of the methoxyl proton and H-6α showed a correlation, so that this methoxyl group is  $\alpha$ -oriented. Based on this evidence, the structure of 3 was determined to be  $3\beta$ -hydroxy- $6\beta$ ,8 $\alpha$ dimethoxyeremophil-7(11)-en-12,8 $\beta$ -olide.

A mixture of compounds **4** and **5** was obtained as colorless oil. These compounds could not be separated by silica gel column chromatography or HPLC. The <sup>1</sup>H-NMR spectrum was virtually identical to that of C-8-epimers of  $6\beta$ -angeloyloxy- $3\beta$ ,8-dihydroxyeremophilenolides<sup>8)</sup> except for the presence of a tigloyloxyl group in place of an angeloyloxyl group. The mixture was treated with acetic anhydride-pyridine to afford an epimeric mixture of diacetate (**4a** and **5a**), which was then separated by silica gel column chromatography. Compound **4a**,  $C_{24}H_{32}O_8$ ,  $[\alpha]_D - 66.7^\circ$ , was isolated as a coloress oil. The IR spectrum suggested the presence of an  $\alpha$ , $\beta$ -unsaturated- $\gamma$ -lactone (1776 cm<sup>-1</sup>), an  $\alpha$ , $\beta$ -unsaturated ester (1729 cm<sup>-1</sup>) and a double bond (1650 cm<sup>-1</sup>). The UV spectrum also suggested the presence of an  $\alpha$ , $\beta$ -

unsaturated- $\gamma$ -lactone ( $\lambda_{max}$ : 222 nm). The <sup>1</sup>H- and <sup>13</sup>C-NMR spectra of **4a** were virtually identical to those of  $3\beta$ ,8 $\alpha$ -diacetoxy- $6\beta$ -angeloyloxyeremophilenolide<sup>8)</sup> except for the presence of a tigloyloxy group [ $\delta_{\rm H}$  1.87 (dq, J=7.0, 1.1 Hz, H-4'), 1.93 (dq, J=1.5, 1.1 Hz, H-5'), 6.97 (qq, J=7.0, 1.5 Hz, H-3'),  $\delta_{\rm C}$  12.2 (C-5'), 14.7 (C-4'), 127.9 (C-2'), 139.0 (C-3'), 166.4 (C-1')]<sup>3)</sup> in place of an angeloyloxyl group. The position of the tigloyloxyl group was confirmed by the HMBC spectrum, in which a cross peak was observed between the H-6 $\alpha$  at  $\delta$  6.19 and the C-1' at  $\delta$  166.4, confirming that a tigloyloxyl group was attached at the C-6 $\beta$ . The structure of **4a** was then determined to be  $3\beta$ ,8 $\alpha$ -diacetoxy- $6\beta$ -tigloyloxyeremophil-7(11)-en-12,8 $\beta$ -olide. Thus, the structure of **4** is  $3\beta$ ,8 $\alpha$ -dihydroxy- $6\beta$ -tigloyloxyeremophil-7(11)-en-12,8 $\beta$ -olide.

Compound 5a,  $C_{24}H_{32}O_8$ ,  $[\alpha]_D + 55.6^\circ$ , was isolated as colorless oil. The IR spectrum suggested the presence of an  $\alpha, \beta$ -unsaturated- $\gamma$ -lactone (1775 cm<sup>-1</sup>), an  $\alpha,\beta$ -unsaturated ester (1733 cm<sup>-1</sup>) and a double bond (1650 cm<sup>-1</sup>). The UV spectrum also suggested the presence of an  $\alpha,\beta$ -unsaturated- $\gamma$ -lactone ( $\lambda_{max}$ : 225 nm). The <sup>1</sup>H- and <sup>13</sup>C-NMR spectra of **5a** were virtually identical to those of  $3\beta$ ,  $8\beta$ -diacetoxy- $6\beta$ -angeloyloxyeremophilenolide8) except for the presence of a tigloyloxyl group  $[\delta_{\rm H} 1.81 (dq, J=7.0, 1.1 \, {\rm Hz}, {\rm H}\text{-}4'), 1.84 (dq, J=1.5, 1.1 \, {\rm Hz},$ H-5'), 6.84 (qq, J=7.0, 1.5 Hz, H-3'),  $\delta_{\rm C}$  12.1 (C-5'), 14.4 (C-4'), 127.9 (C-2'), 137.5 (C-3'), 166.5 (C-1')]<sup>3)</sup> in place of an angeloyloxyl group. The position of a tigloyloxyl group was confirmed by the HMBC spectrum, in which a cross peak was observed between the H-6 $\alpha$  at  $\delta$  5.76 and the C-1' at  $\delta$  166.5, showing that a tigloyloxyl group was attached at the C-6 $\beta$ . The structure of **5a** was then determined to be  $3\beta$ ,  $8\beta$ -diacetoxy- $6\beta$ -tigloyloxyeremophil-7(11)-en-12,8 $\alpha$ -olide. Thus, the structure of 5 is  $3\beta$ ,8 $\beta$ dihydroxy- $6\beta$ -tigloyloxyeremophil-7(11)-en- $12,8\alpha$ -olide.

Compound 6,  $C_{20}H_{26}O_6$ ,  $[\alpha]_D + 38.1^\circ$ , was isolated as colorless oil. The IR spectrum suggested the presence of a hydroxyl group (3600—3200 cm<sup>-1</sup>), an  $\alpha,\beta$ -unsaturated- $\gamma$ -lactone (1759 cm<sup>-1</sup>), a six-membered ring ketone and an  $\alpha,\beta$ -unsaturated ester (1713 cm<sup>-1</sup>), and a double bond (1648 cm<sup>-1</sup>). The UV spectrum also suggested the presence of an  $\alpha,\beta$ -unsaturated- $\gamma$ -lactone ( $\lambda_{max}$ : 220 nm). The <sup>1</sup>Hand <sup>13</sup>C-NMR spectra showed signals due to a tertiary methyl group [ $\delta_{\rm H}$  1.00 (s, H-15),  $\delta_{\rm C}$  18.9 (C-15)], a secondary methyl group [ $\delta_{\rm H}$  1.02 (d, J=6.6 Hz, H-14),  $\delta_{\rm C}$  8.0 (C-14)], an olefinic methyl group [ $\delta_{\rm H}$  2.02 (s, H-13),  $\delta_{\rm C}$ 9.0 (C-13)], an angeloyloxyl group  $[\delta_H 1.92 \text{ (dq, } J=1.5,$ 1.5 Hz, H-5'), 2.04 (dq, J=7.3, 1.5 Hz, H-4'), 6.20 (qq, J=7.3, 1.5 Hz, H-3'),  $\delta_{\rm C}$  16.0 (C-4'), 20.6 (C-5'), 126.7 (C-2'), 141.1 (C-3'), 166.8 (C-1')], <sup>3)</sup> an angeloyloxy-bearing methine [ $\delta_{\rm H}$  5.61 (s, H-6 $\alpha$ ),  $\delta_{\rm C}$  70.6 (C-6)], an  $\alpha,\beta$ -unsaturated- $\gamma$ -lactone [ $\delta_{\rm C}$  129.4 (C-11), 151.1 (C-7), 170.5 (C-12)], a hemi-ketal carbon [ $\delta_{\rm C}$  103.5 (C-8)] and a carbonyl carbon [ $\delta_{\rm C}$  209.9 (C-3)], which is correlated with the secondary methyl group (C-14) at  $\delta$  1.02 in the HMBC spectrum. Thus, the structure of 6 was estimated to be  $6\beta$ -angeloyloxy-8( $\alpha$  or  $\beta$ )-hydroxy-3-oxoeremophilenolide. In the <sup>1</sup>H-NMR spectrum, the homoallylic coupling between the olefinic methyl group (H-13) and H-6α is lacking. In the NOESY spectrum, each signal of the secondary methyl group (H-14) and H-6a showed a correlation. These data indicated that  $\mathbf{6}$  exists in a steroidal conformation. On the basis of the above evidence, the structure of  $\mathbf{6}$  was determined to be  $6\beta$ -angeloyloxy- $8\beta$ -hydroxy-3-oxoeremophil-7(11)-en- $12,8\alpha$ -olide.

## Experimental

General Procedures Melting points were determined with a Yanagimoto micromelting apparatus and are uncorrected. Optical rotations were determined with a JASCO DIP-360 digital polarimeter. IR spectra were recorded with a Perkin-Elmer FT-IR 1725X infrared spectrophotometer and UV spectra with a Beckman DU-64 spectrophotometer. <sup>1</sup>H- and <sup>13</sup>C-NMR spectra were recorded with a JEOL JNM-GSX 400 (400 and 100 MHz, respectively) spectrometer. Chemical shifts were given on a  $\delta$  (ppm) scale with tetramethylsilane as an internal standard (s, singlet; d, doublet; dd, double doublet; ddd, double double doublet; dq, double quartet; qq, quartet quartet; m, multiplet). The electron ionization mass spectrum (EI-MS) and high resolution mass spectrum (HR-MS) were recorded on a JEOL JMS-DX 303 mass spectrometer. Column chromatography was carried out on Kieselgel 60 (Merck; 230-400 mesh). Preparative HPLC was carried out on a Tosoh HPLC system (pump, CCPD; detector, UV-8011) using a TSK gel ODS-120T column (Tosoh) and TSK gel OH-120 column (Tosoh).

**Plant Material** The dried rhizomes of *Petasites japonicus* were purchased from Tochimoto Tenkaido Co., Ltd. in 1990.

Extraction and Isolation The dried rhizomes of Petasites japonicus (3.0 kg) were extracted three times with MeOH at room temperature for 2 weeks. The MeOH extract was concentrated under reduced pressure and the residue was suspended in a small excess of water. This suspension was extracted, successively, with CHCl<sub>3</sub>, Et<sub>2</sub>O, AcOEt and n-BuOH. The CHCl3-soluble fraction was concentrated under reduced pressure to afford a residue (112.5 g). This residue (60.0 g) was chromatographed on a silica gel column using benzene-AcOEt (9:1, 8:2, 7:3) and CHCl<sub>3</sub>-MeOH (8:2), and the eluate was separated into 4 fractions (frs. 1-4). Fraction 4 was rechromatographed on a silica gel column using benzene-AcOEt (6:4, 5:5, 4:6, 3:7) and CHCl<sub>3</sub>-MeOH (9:1, 8:2), and the eluate was separated into 4 fractions (frs. 1'-4'). Fraction 2' was rechromatographed on a silica gel column using n-hexane-acetone (5:4, 5:5, 4:5, 3:6) and acetone, and the eluate was separated into 8 fractions (frs. 1"-8"). Fraction 3" was rechromatographed on a silica gel column using benzene-AcOEt (3:2, 2:2), and the eluate was separated into 7 fractions (frs. 1"'-7"'). Fraction 6" was purified by preparative HPLC (Column, TSK gel ODS-120T, 21.5 mm i.d. × 30 cm; mobile phase, MeOH-H<sub>2</sub>O (1:1); flow rate, 3.0 ml/min; UV detector, 220 nm) to give 3 (20.5 mg), a mixture of 4 and 5 (16.8 mg) and 6 (5.3 mg). Fraction 7" was separated by preparative HPLC (Column, TSK gel OH-120, 7.8 mm i.d.  $\times$  30 cm; mobile phase, *n*-hexane–EtOH (5:1); flow rate, 0.85 ml/min; UV detector, 220 nm) to give a mixture of 1 and 2. The mixture of 1 and 2 was purified by preparative HPLC (Column, TSK gel ODS-120T, 7.8 mm i.d.  $\times$  30 cm; mobile phase, MeOH-H<sub>2</sub>O (1:1); flow rate, 1.2 ml/min; UV detector, 220 nm) to give 1 (2.7 mg) and 2 (8.9 mg).

3β-Hydroxyeremophil-7(11)-en-12,8β-olide (1) Colorless needles (from Et<sub>2</sub>O–AcOEt), mp 168—169 °C.  $[\alpha]_{D}^{22}$  – 136.0° (c=0.3, CHCl<sub>3</sub>). IR  $\nu_{\max}^{\text{CHCl}_3}$  cm<sup>-1</sup>: 3471, 1741, 1684. UV  $\lambda_{\max}^{\text{MeOH}}$  nm (log ε): 223 (4.2). HR-MS m/z: 250.1577 (M<sup>+</sup>, Calcd for C<sub>15</sub>H<sub>22</sub>O<sub>3</sub>; 250.1569). ¹H-NMR: see Table II. ¹³C-NMR: see Table II.

3β-Hydroxy-6β-methoxyeremophil-7(11)-en-12,8β-olide (2) Colorless needles (from Et<sub>2</sub>O–AcOEt), mp 180—181 °C  $[\alpha]_D^{23}$  –189.6°  $(c=0.9, \text{CHCl}_3)$ . IR  $\nu_{\text{max}}^{\text{CHCl}_3}$  cm $^{-1}$ : 3474, 1746, 1678. UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm (log ε): 219 (4.2). HR-MS m/z: 280.1662 (M $^+$ , Calcd for  $C_{16}H_{24}O_4$ ; 280.1675).  $^1$ H-NMR: see Table II.  $^1$ 3C-NMR: see Table II.

3β-Hydroxy-6β,8α-dimethoxyeremophil-7(11)-en-12,8β-olide (3) Colorless oil.  $[\alpha]_{\rm max}^{26}$  -86.7° (c=1.1, CHCl $_3$ ). IR  $\nu_{\rm max}^{\rm CHCl}$  cm $^{-1}$ : 3509, 1757, 1680. UV  $\lambda_{\rm max}^{\rm MCOH}$  nm (log  $\varepsilon$ ): 225 (3.8). HR-MS m/z: 310.1807 (M $^+$ , Calcd for C $_{17}$ H $_{26}$ O $_5$ ; 310.1780).  $^1$ H-NMR: see Table I.  $^{13}$ C-NMR: see Table II.

The Mixture of 3 $\beta$ ,8 $\alpha$ -Dihydroxy-6 $\beta$ -tigloyloxyeremophil-7(11)-en-12,8 $\beta$ -olide (4) and 3 $\beta$ ,8 $\beta$ -dihydroxy-6 $\beta$ -tigloyloxyeremophil-7(11)-en-12,8 $\alpha$ -olide (5) Colorless oil. <sup>1</sup>H-NMR (CDCl<sub>3</sub>) 4 δ: 0.97 (3H, d, J=7.3 Hz, H-14), 0.97 (3H, s, H-15), 7.00 (1H, qq, J=7.3, 1.3 Hz, H-3'). 5 δ: 1.06 (3H, d, J=7.0 Hz, H-14), 1.32 (3H, s, H-15), 6.91 (1H, qq, J=7.3, 1.3 Hz, H-3').

Acetylation of the Mixture of 4 and 5 Acetylation of 11 mg of the

mixture with acetic anhydride-pyridine for 2 d at room temperature followed by the usual work up and purification by silica gel column chromatography [benzene-AcOEt (19:1)] gave two diacetates [4a (5.4 mg) and 5a (0.9 mg)].

3β,8α-Diacetoxy-6β-tigloyloxyeremophil-7(11)-en-12,8β-olide (4a) Colorless oil.  $[\alpha]_D^{20} - 66.7^\circ$  (c = 0.5, CHCl<sub>3</sub>). IR  $\nu_{\text{max}}^{\text{HaCl}_3}$  cm  $^{-1}$ : 1776, 1729, 1650. UV  $\lambda_{\text{max}}^{\text{MeO}}$  nm (log ε): 222 (4.2). HR-MS m/z: 448.2169 (M<sup>+</sup>, Calcd for C<sub>24</sub>H<sub>32</sub>O<sub>8</sub>; 448.2148). <sup>1</sup>H-NMR: see Table II. <sup>13</sup>C-NMR: see Table II.

3β,8β-Diacetoxy-6β-tigloyloxyeremophil-7(11)-en-12,8α-olide (5a) Colorless oil.  $[\alpha]_{\rm D}^{21}$  + 55.6° (c = 0.1, CHCl<sub>3</sub>). IR  $\nu_{\rm max}^{\rm CHCl}$  cm  $^{-1}$ : 1775, 1733, 1650. UV  $\lambda_{\rm max}^{\rm MeOH}$  nm (log  $\varepsilon$ ): 225 (4.9). HR-MS m/z: 448.2133 (M $^+$ , Calcd for C<sub>24</sub>H<sub>32</sub>O<sub>8</sub>; 448.2148).  $^{1}$ H-NMR: see Table II.  $^{13}$ C-NMR: see Table III.

6β-Angeloyloxy-8β-hydroxy-3-oxoeremophil-7(11)-en-12,8α-olide (6) Coloress oil.  $[\alpha]_D^{19} + 38.1^\circ (c=0.5, \text{CHCl}_3)$ . IR  $\nu_{\max}^{\text{CHCl}_3} \text{cm}^{-1}$ : 3600—3200, 1759, 1713, 1648. UV  $\lambda_{\max}^{\text{MeOH}} \text{nm} (\log \varepsilon)$ : 220 (4.3). HR-MS m/z: 362.1706 (M<sup>+</sup>, Calcd for  $C_{20}H_{26}O_6$ ; 362.1729). <sup>1</sup>H-NMR: see Table II. <sup>13</sup>C-NMR: see Table II.

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## References and Notes

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