THE ABSOLUTE STEREOSTRUCTURE OF PANAXYTRIOL, A BIOLOGICALLY ACTIVE DIACETYLENIC ACETOGENIN, FROM GINSENG RADIX RUBRA

Motomasa KOBAYASHI,* Taifo MAHMUD, Takashi UMEZOME, and Isao KITAGAWA*,1)

Faculty of Pharmaceutical Sciences, Osaka University, Yamada-oka 1-6, Suita, Osaka 565, Japan

The absolute stereostructure of panaxytriol (1), a diacetylenic constituent of Ginseng Radix Rubra, was determined by applying the modified Mosher method and the CD exciton chirality method to be expressed as (3R, 9R, 10R)-heptadec-1-ene-4,6-diyne-3,9,10-triol.

KEY WORDS Panaxytriol; Ginseng Radix Rubra; Mosher's method modified; CD exciton chirality method

Panaxytriol (1) was first isolated in 1983 as a characteristic diacetylenic constituent of Ginseng Radix Rubra (red ginseng, processed ginseng root) and the plain structure was elucidated.²⁾ Since then, the biological activity of 1 has been extensively investigated and recently it has been received attention as a potential new type of antitumor agent.³⁾ As a continuation of our chemical studies on crude drug processing, *i.e.*, on the characteristic constituents of Ginseng Radix Rubra, ^{2,4)} we have investigated the absolute stereostructure of panaxytriol (1), which is essential from the biological activity viewpoint.

In regard to the absolute configuration at C-3 of panaxynol (2), another diacetylenic constituent of white ginseng (Fig.1), Shim et al. applied the CD exciton chirality method to the p-bromobenzoate 3 and concluded that panaxynol (2) possessed a 3S configuration.⁵⁾ On the other hand, by means of the modified Mosher method, Bernard et al.⁶⁾ recently defined the 3S configuration for (+)-farcarinol, which is a known enantiomer of panaxynol (2), and found that the $[\alpha]_D$ value of (+)-farcarinol as well as the CD spectrum of its p-bromobenzoate showed opposite signs to those of panaxynol (2) and its p-bromobenzoate 3. Therefore Bernard et al. claimed that panaxynol (2) must possess a 3R configuration and that the CD exciton chirality method applied to secondary allylic alcohols was not applicable to secondary alcohols flanked by two unsaturated chromophores as seen in 2.

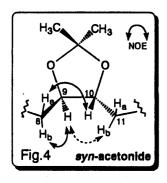
Bearing in mind the panaxynol problem, in order to determine the absolute configuration at C-3 of panaxytriol (1), which showed levorotatory ($[\alpha]_D - 19.0^\circ$, c = 1.0, CHCl₃), both the R- and S-MTPA esters (5 and 6) of 1 were first prepared after prior acetonide protection of the 9,10-diol moiety in 1.⁷⁾ Examination of the proton NMR signals of 5 and 6, of which the $\Delta\delta$ values ($\delta_S - \delta_R$) are illustrated in Fig. 2, has led us to assign a 3R configuration to 5 and 6. For comparison purposes, the p-bromobenzoate 7 of 1 was next prepared and was found to show a negative CD maximum [249 nm ($\Delta \varepsilon = -7.9$)], which was similar to that of 3.

1596 Vol. 43, No. 9

In order to confirm our results further and to shed light on the above conflicting reports on the C-3 configuration of panaxynol (2) as well, 5,6) we then prepared the simple p-bromobenzoate derivative 11,8) to which the CD exciton chirality method is unequivocally applicable. Thus, partial hydrogenation of 4 using Lindlar catalyst 9) furnished three products, 8, 9, and 10, among which 8 was further converted to p-bromobenzoate 11.

The CD spectrum of 11 showed a negative maximum at 245 nm ($\Delta \epsilon = -9.1$), and thus the absolute configuration of C-3 in panaxytriol (1) has been reconfirmed to be R^{10} (Fig. 3).

MTPA-O H H H H
$$^{+22}$$
 H $^{-2}$ H $^{+3}$ $^{-1.5}$



Next, we investigated the absolute stereochemistry of the vicinal glycol moiety at C-9 and C-10 of panaxytriol (1). In the relative configuration, the *syn* structure was defined by a NOESY experiment of acetonide 4. Thus, the NOE correlations in 4 were observed between 8-H_b and 9-H; 8-H_a and 10-H; and 9-H and 11-H_b, as illustrated in Fig. 4. Furthermore, the coupling constant (8 Hz) between 9-H and 10-H as well as the isopropylidene methyl proton signals observed at δ 1.40 (6H, s) supported the *syn*-acetonide structure in 4.¹¹, ¹²)

The absolute configurations of C-9 and C-10 were then defined as based on the CD analysis of the di-p-bromobenzoate 12,¹³⁾ which showed a negative exciton split [255 nm ($\Delta \varepsilon = -6.5$) and 239 nm ($\Delta \varepsilon = +10.6$)] to indicate 9R, 10R stereochemistry. ¹⁴⁾

In order to eliminate the effect of the 4,6-diyne chromophore, we further prepared a saturated di-p-bromobenzoate derivative 14. Here again, the CD spectrum of 14 showed a clear exciton split with similar amplitude [first Cotton at 255 nm ($\Delta \varepsilon = -6.5$); second Cotton at 239 nm ($\Delta \varepsilon = +7.1$)], and thus the 9R and 10R configurations were confirmed unambiguously.

Reagents and Conditions: a) i) Ac_2O , pyridine, r.t., ii) 80% aq. AcOH, $60^{\circ}C$, iii) ρ -BrC₆H₄COCl, pyridine, $60^{\circ}C$, 3 steps 60%; b) i) H₂, Pd-CaCO₃, Pb(OAc)₂, cyclohexane, 90%, ii) Ac_2O , pyridine, r.t., 85%; c) i) 80% aq. AcOH, $60^{\circ}C$, ii) ρ -BrC₆H₄COCl, pyridine, $60^{\circ}C$, 2 steps 72%.

From the above-mentioned evidence, the absolute stereostructure of panaxytriol has been determined to be (3R, 9R, 10R)-heptadec-1-ene-4,6-diyne-3,9,10-triol (1).

REFERENCES AND NOTES

- 1) Present address: Faculty of Pharmaceutical Sciences, Kinki University, Kowakae 3-4-1, Higashiosaka, Osaka 577, Japan.
- 2) Kitagawa I., Yoshikawa M., Yoshihara M., Hayashi T., Taniyama T., Yakugaku Zasshi, 103, 612-622 (1983).
- 3) For examples: a) Sata T., Matsunaga H., Yamamoto H., Nagumo F., Fujito H., Mori M., Katano M., Biol. Pharm. Bull., 17, 798-802 (1994); b) Matsunaga H., Katano M., Yamamoto H., Mori M., Takata K., Chem. Pharm. Bull., 37, 1279-1281 (1989).
- 4) a) Kitagawa I., Taniyama T., Yoshikawa M., Ikenishi Y., Nakagawa Y., Chem. Pharm. Bull., 37, 2961-2970 (1989); b) Kitagawa I., Taniyama T., Shibuya H., Noda T., Yoshikawa M., Yakugaku Zasshi, 107, 495-505 (1987); c) Kitagawa I., Taniyama T., Hayashi T., Yoshikawa M., Chem. Pharm. Bull., 31, 3353-3356 (1983).
- 5) Shim S. C., Koh H. Y., Chang S-K., Tetrahedron Lett., 26, 5775-5776 (1985).
- 6) Bernard M. W., Hallock Y. F., Cardellina II J. H., Boyd M. R., Tetrahedron Lett., 35, 993-994 (1994).
- 7) 4: $[\alpha]_D$ 22.5 ° (c= 1.2, acetone, 25°C). FAB-MS $m \angle z$: 341 (M+Na)+, ¹H-NMR (500 MHz, CDCl₃) δz : 5.94 (ddd, J=17, 10, 5 Hz, 2-H), 5.47 (d, J=17, 1-H_b), 5.25 (d, J=10, 1-H_a), 4.92 (dd, J=7, 5, 3-H), 3.80 (td, J=8, 4, 10-H), 3.73 (dt, J=8, 5, 9-H), 2.63 (dd, J=17, 5, 8-H_b), 2.59 (dd, J=17, 5, 8-H_a), 1.87 (d, J=7, 3-OH), 1.58 (m, 11-H₂), 1.40 (s, 19-H₃ and 20-H₃), 0.89 (t, J=7, 17-H₃).
- 8) 11: FAB-MS $m \not = 531$ [M+Na]⁺, ¹H-NMR (270 MHz, CDCl₃) $\delta = 5.69$ (dt, J=9, 7 Hz, 3-H), 5.64 (dt, J=11, 7, 5-H), 5.42 (dd, J=9, 11, 4-H), 2.24 (m, 6-H₂), 1.79 (m, 2-H_b), 1.69 (m, 2-H_a), 0.95 (t, J=7, 1-H₃), 0.87 (t, J=7, 17-H₃).
- 9) Marvell E. N., Tashiro J., J. Org. Chem., 30, 3991-3993 (1965).
- 10) Gonnella N. C., Nakanishi K., J. Am. Chem. Soc., 104, 3775-3776 (1982).
- 11) Chuche J., Dana G., Monot M-R., Bull. Soc. Chim. France, 9, 3300-3307 (1967).
- 12) Oikawa H., Matsuda I., Kagawa T., Ichihara A., Kohmoto K., Tetrahedron, 50, 13347-13368 (1994).
- 13) 12: FAB-MS m/z: 707 [M+Na]⁺, ¹H-NMR (500 MHz, CDCl₃) δ : 5.87 (d, J=6 Hz, 3-H), 5.84 (ddd, J=16, 10, 6, 2-H), 5.49 (d, J=16, 1-H_b), 5.49 (m, 10-H), 5.38 (m, 9-H), 5.33 (d, J=10, 1-H_a), 2.83 (dd, J=18, 6, 8-H_b), 2.76 (dd, J=18, 6, 8-H_a), 2.10 (s, 3-OCOCH₃), 1.74 (m, 11-H₂), 0.84 (t, J=3, 17-H₃).
- 14) a) Harada N., Nakanishi K., J. Am. Chem. Soc., 91, 3989-3991 (1969); b) Harada N., Saito A., Ono H., Uda H., Gawronski J., Gawronska K., Symposium Papers, 33rd Symposium on the Chemistry of Natural Products, Osaka, Japan, Oct. 1991, pp. 464-471.

(Received June 13, 1995; accepted July 15, 1995)