Triterpene Alcohols from Camellia and Sasanqua Oils and Their Anti-inflammatory Effects

Toshihiro Akihisa,*,a Ken Yasukawa,b Yumiko Kimura,b Sei-ichi Takase,c Sakae Yamanouchi,b and Toshitake Tamura

College of Science and Technology, Nihon University, I-8, Kanda Surugadai, Chiyoda-ku, Tokyo 101, Japan, College of Pharmacy, Nihon University, 7-7-1 Narashinodai, Funabashi-shi, Chiba 274, Japan, and College of Science and Technology (Funabashi Campus), Nihon University, 7-24-1, Narashinodai, Funabashi-shi, Chiba 274, Japan. Received June 27, 1997; accepted August 17, 1997

The nonsaponifiable lipids of camellia and sasanqua oils from the seeds of Camellia japonica L. and C. sasanqua Thunb., respectively, were investigated for their triterpene alcohol constituents. This led to the isolation of twenty-seven triterpene alcohols of which seven were novel naturally occurring compounds, tirucalla-5,7,24-trien-3 β -ol (1), lemmaphylla-7,21-dien-3 β -ol (2), isoeuphol (3), isotirucallol (4), (24R)-24,25-epoxybutyrospermol (5) and its 24S-epimer (6), and isoaglaiol (7). The structures were determined by spectroscopic and chemical methods. The inhibitory effects of 3, 4, a mixture of 5 and 6, a mixture of 7 and its 24S-epimer (aglaiol), and eight known triterpene alcohols isolated in this study were evaluated in ear inflammation in mice induced by 12-O-tetradecanoylphorbol-13-acetate (TPA). The 50% inhibitory dose of these triterpenes for TPA-induced inflammation (1 μ g per ear) was 0.2—0.9 mg/ear.

Key words Camellia japonica; Camellia sasanqua; seed oil; triterpene alcohol; antioedema; 12-O-tetradecanoylphorbol-13-acetate-induced ear oedema

The triterpene alcohol fractions of Theaceae seed oils contain a significant amount of euphane/tirucallane-type compounds, viz., butyrospermol (9) and tirucalla-7,24-dien-3β-ol (11). We undertook detailed re-investigation of the constituents of the triterpene alcohol fractions separated from the nonsaponifiable lipids (NSL) of camellia (Camellia japonica L.) and sasanqua (Camellia sasanqua Thunb.) seed oils and isolated seven novel compounds, 1—7, along with twenty known compounds. This paper deals with the structure elucidation of the seven novel compounds and the anti-inflammatory activity of fourteen triterpene alcohols, isolated in this study, on 12-O-tetradecanoylphorbol-13-acetate (TPA)-induced inflammation in mice.

Results and Discussion

Column chromatography on silica-gel of the NSL obtained by alkaline hydrolysis of camellia and sasanqua oils yielded triterpene alcohol fractions. These were acetylated and the resulting acetate fractions were subjected to argentation TLC and HPLC, which enabled the isolation of twenty-seven compounds including seven novel ones. 1-7, as their acetyl derivatives (1a-7a, respectively). Table 1 shows the compositions of triterpene alcohol fractions from the NSL of camellia and sasanqua oils along with their chromatographic data. Among the twenty known compounds 8—27, aglaiol (8), 3,4) δ -amyrin (21). 5 and 17-epilupeol (26)⁶⁾ were identified as their acetyl derivatives by spectral comparison with literature data, whereas the others were identified by direct comparison of chromatographic and spectral data with reference compounds.

All the novel compounds, **1a**—**7a**, have a secondary acetoxyl group [v_{max} 1729—1733, 1241—1250 cm⁻¹; δ_{H} 2.05—2.07 (3H, s)] associated with an adjacent methine [δ_{H} ca. 4.5—4.6 (1H, dd), J=ca. 4—6, 10—11 Hz]. The shift and coupling constants of the methine ¹H signal

suggested that the acetoxyl group of the seven triterpene acetates is oriented equatorially (β) at C-3.⁷⁾

The molecular formula of 1a was determined as C₃₂H₅₀O₂ on the basis of the high-resolution mass spectrum (HR-MS) (M $^+$, m/z 466.3789). The compound had a $\Delta^{5,7}$ -conjugated diene [λ_{max} 273 (log ϵ 4.04), 281 (4.00) nm; v_{max} 1646, 837, 820 cm⁻¹; δ_{H} 5.55 (1H, d, J=2.7, 5.5 Hz) and 5.89 (1H, d, J=5.5 Hz)],⁸⁾ a terminal isopropylidene group $[\delta_H \ 1.61 \ (s), \ 1.68 \ (s)]$, and one secondary and five tertiary methyl groups. These data, in combination with fragment ions having m/z 451 (M⁺ – Me), 391 (M⁺ – Me – HOAc), 295 [loss of side-chain (s.c.; C_8H_{15}) and HOAc], 253 (295-42), 239 (295-42-CH₂)⁹⁾ and 69 $[CH_2CH = C(Me)_2^+]$, suggested that 1a had a 3β -acetoxy tetracyclic triterpene skeleton possessing a $\Delta^{5,7}$ -diene system and a C₈-side-chain containing an isopropylidene group. The diagnostic fragment ion at m/z171 (C₁₃H₁₅⁺) formed by the loss of ring D plus sidechain due to cleavage of the C-8-C-14 and C-9-C-11 bonds with concomitant HOAc loss supported the $\Delta^{5,7}$ unsaturation. 7,10) Furthermore, analysis of the ¹H-¹H and ¹³C-¹H correlation spectroscopies (COSYs) and heteronuclear multiple-bond correlation (HMBC) spectra enabled the structure of 1a to be formulated as 4,4,14trimethyl- $\Delta^{5,7,24}$ -statrien-3 β -yl acetate with an as-vet-tobe-determined stereochemistry. In order to establish the stereostructure of 1a, a difference nuclear Overhauser effect (NOE) experiment was undertaken. Thus, 1a showed significant NOE correlations between [H-29(4β-Me)-H- $19(10\beta\text{-Me})$ -H-30(14 β -Me)-H-17 β -H-21] on the β -face. $[H-3\alpha-H-28(4\alpha-Me)]$ and $[H-9\alpha-H-18(13\alpha-Me)-H-20]$ on the α -face and [H-12 α -H-21] of the molecule. These NOE correlations were consistent with those of two tirucallanetype triterpenes, tirucalla-5,24-dien-3 β -yl acetate and 11a (the acetate of 11).11) Compound 9a (the acetate of 9), a euphane-type triterpene, on the other hand, did not exhibit the NOE correlation between [H-12 α -H-21]. Further-

* To whom correspondence should be addressed.

© 1997 Pharmaceutical Society of Japan

December 1997 2017

Table 1. Composition (%) and Chromatographic Data of Acetates of Triterpene Alcohols from the Saponified Neutral Fraction of Camellia and Sasanqua Oils

C 1			Acetate, Rt _R	Composition (%) ^{c)}			
Code	Compound ^{a)}	GLC	HPLC I	HPLC II	Camellia	Sasanqua	
1	Tirucalla-5,7,24-trienol ^{d)}	1.42	0.67	0.41	0.2	0.2	
2	Lemmaphylla-7,21-dienol ^{d)}	1.86	0.86	0.52	1.6	1.2	
3	Isoeuphol $[(20R)$ -dammara- $13(17)$,24-dienol] ^{d)}	1.21	0.61	0.44	0.1	0.1	
4	Isotirucallol $\lceil (20S) - \text{dammara} - 13(17), 24 - \text{dienol} \rceil^d $	1.28	0.69	0.38	1.7	2.1	
5	(24R)-24,25-Epoxybutyrospermol $[(24R)$ -24,25-epoxyeuph-7-enol] ^{d)}	2.53	0.18	0.28	0.1	0.1	
6	(24S)-24,25-Epoxybutyrospermol $[(24S)$ -24,25-epoxyeuph-7-enol] ^{d)}	2.53	0.18	0.28	0.1	0.1	
7	Isoaglaiol $\lceil (24R)-24,25$ -epoxydammar-20-enol \rceil^{d}	2.53	0.22	0.16	0.1	0.1	
8	Aglaiol [(24S)-24,25-epoxydammar-20-enol]	2.53	0.22	0.16	0.1	0.1	
9	Butyrospermol (eupha-7,24-dienol)	1.66	0.80	0.47	16.3	16.9	
10	Euphol (eupha-8,24-dienol)	1.30	0.76	0.44	0.5	0.4	
11	Tirucalla-7,24-dienol	1.89	0.83	0.58	25.6	22.4	
12	Tirucallol (tirucalla-8,24-dienol)	1.47	0.80	0.53	0.4	0.1	
13	Dammaradienol (dammara-20,24-dienol)	1.66	0.61	0.39	6.0	6.9	
14	24-Methylenedammarenol [24-methyldammara-20,24(24¹)-dienol]	1.79	0.68	0.41	0.1	0.1	
15	Cycloartenol (cycloart-24-enol)	1.83	0.98	0.82	0.6	0.9	
16	24-Methylenecycloartanol [24-methylcycloart-24(24 ¹)-enol]	2.00	1.07	0.90	0.4	2.0	
17	24-Methyllanosta-8,24(24¹)-dienol	1.71	0.98	0.87	0.3	0.4	
18	24-Methyllanosta-9(11),24(24 ¹)-dienol	1.95	0.93	0.67	0.1	0.1	
19	Bacchara-12,21-dienol	2.10	0.93	0.76	2.8	3.3	
20	β-Amyrin (olean-12-enol)	1.70	0.92	0.60	24.4	24.6	
21	δ-Amyrin [olean-13(18)-enol]	1.71	0.95	0.62	3.0	3.3	
22	Germanicol (olean-18-enol)	1.70	0.88	0.88	1.4	0.9	
23	α-Amyrin (urs-12-enol)		1.01	0.66	1.0	1.8	
24	Taraxerol (tarax-14-enol)	1.91 1.63	0.88	0.56	3.6	4.9	
25	Lupeol [lup-20(29)-enol]	2.00	0.74	0.44	4.4	2.9	
26	17-Epilupeol (17-epilup-20(29)-enol)	2.07	0.86	0.50	0.3	0.4	
27	ψ-Taraxasterol (taraxast-20-enol)	2.49	1.06	0.72	1.5	0.7	
	Others, unidentified				3.3	3.0	

a) All compounds have a hydroxyl group at C-3 β . All compounds, with the exception of C-5 unsaturated ones, are 5α -compounds. b) Standard: cholesteryl acetate (Rt_R: 1.00). c) Percentage composition of the triterpene alcohol fraction determined based on HPLC and GLC data. d) New compounds reported in this paper.

more, although the side-chain ¹³C-NMR signals (C-20-C-27) of 1a (Table 2) were fully consistent with those of two tirucallanes, tirucalla-5,24-dien-3 β -yl acetate and 11a, 11) small but significant differences for some sidechain carbons (C-21-C-23) were observed between 1a and euphane 9a, 11) viz., C-21 ($\delta_{1a} - \delta_{9a} = -0.3$), C-22 (1.0), and C-23 (-0.4). We concluded that **1a** is tirucalla-5,7,24trien-3 β -yl acetate. The most stable conformation of **1a** with minimum steric energy, ¹² simulated by using the CAChe and MM2 program, ¹³ orients C-22 in a "righthanded" conformation (C-22 trans-oriented with respect to C-13), as shown in Fig. 1, similar to that of 11a¹¹⁾ and the crystal structure of another tirucallane, tirucallyl acetate (12a; the acetate of 12).14) This conformation of 1a was fairly consistent with results from the NOE experiment carried out in solution. Alkaline hydrolysis of 1a yielded tirucalla-5,7,24-trien-3 β -ol (1; C₃₀H₄₈O; M⁺, m/z 424.3712).

Compound **2a**, which showed M⁺ at m/z 468.3984 (C₃₂H₅₂O₂) in the HR-MS, possesses two trisubstituted double bonds [ν_{max} 817, 800 cm⁻¹; δ_{H} 5.07 (1H, tt, J=1.4, 7.4 Hz) and 5.37 (1H, dd, J=3.1, 3.6 Hz)], a terminal isopropylidene group [δ_{H} 1.59 (s) and 1.68 (s)], and six tertiary methyl groups. These data, in combination with mass fragmentations at m/z 453 (M⁺ – Me), 393 (M⁺ – Me–HOAc), 325 [loss of s.c.(C₆H₁₁) and HOAc], and 69 [CH₂CH=C(Me)₂⁺], suggested that **2a** was a 3 β -acetoxy triterpene with a six-membered tetracyclic skeleton

possessing one double bond and a C₆-side-chain containing an isopropylidene functionality. 15) Further fragment ions at m/z 255 ($C_{19}H_{27}^+$; loss of ring D plus s.c., due to cleavage of C-13-C-18 and C-14-C-15, with concomitant loss of HOAc), and 203 $(C_{15}H_{23}^{+})$ and 189 (C₁₄H₂₁⁺) formed by the loss of ring D plus side-chain due to cleavage of the C-8-C-14 and C-12-C-13 bonds, and C-8-C-14 and C-11-C-12 bonds, respectively, with concomitant HOAc loss suggested 2a had the skeleton of a migrated baccharane having a Δ^7 - or a $\Delta^{9(11)}$ -double bond. 15) The 1H-NMR signals of 2a were, with the exception of those arising from the ring A substituents, in agreement with the corresponding ^{1}H signals for lemmaphylla-7,24-diene but not for its $\Delta^{9(11)}$ -isomer $^{15)}$ and, thus, 2a was proposed to have the structure lemmaphylla-7,24-dien-3 β -yl acetate (D: C-friedo-18,19seco-lupa-7,19-dien-3 β -yl acetate). The proposed structure and stereochemistry of 2a were confirmed by analysis of its two-dimensional (2D) NMR (¹H-¹H and ¹³C-¹H COSYs, and HMBC) spectra and difference NOE spectra. 16) Compound 2a showed a significant NOE correlation between [H-24(4 β -Me)–H-25(10 β -Me)–H- $26(14\beta-Me)-H-16\beta$, $H-18\beta-H-28(17\beta-Me)$] on the β -face and $[H-3\alpha-H-23(4\alpha-Me)-H-5\alpha-H-9\alpha-H-27(13\alpha-Me)]$ on the α -face of the molecule. The most stable conformation of 2a with minimum steric energy, 17) simulated by using the CAChe and MM2 program¹³⁾ is shown in Fig. 1. This conformation of 2a was fairly consistent with results from

2018 Vol. 45, No. 12

Chart 1. Structures of Seven Novel Triterpene Alcohols from Camellia and Sasanqua Oils

the NOE experiment carried out in solution. Alkaline hydrolysis of **2a** yielded lemmaphylla-7,24-dien-3 β -ol (**2**; $C_{30}H_{50}O$; M^+ , m/z 426.3872).

Compound 3a, which showed M^+ at m/z 468.3944 (C₃₂H₅₂O₂) in the HR-MS, possesses a trisubstituted $[v_{\text{max}} 828 \,\text{cm}^{-1}; \, \delta_{\text{H}} 5.13 \,(1\text{H}, \,\text{tt}, \, J = 1.2, \, 7.1 \,\text{Hz})]$ and a tetrasubstituted [$\delta_{\rm C}$ 134.7 (s) and 138.9 (s)] double bonds, a terminal isopropylidene group [$\delta_{\rm H}$ 1.59 (s) and 1.68 (s)], and one secondary and five tertiary methyl groups. This, in combination with a mass fragment at m/z 399 [loss of part of s.c. by cleavage of C-22-C-23], 357 (loss of s.c.), 355 (357-2H), and 297 (357-HOAc), suggested that **3a** was a 3β -acetoxy tetracyclic triterpene possessing a monounsaturated ring system with a tetrasubstituted double bond and a Δ^{24} -unsaturated C_8 -side-chain. Diagnostic fragment ions at m/z 263 (A, B rings formed by cleavages of C-8-C-14 and C-11-C-12), and 249 (A, B rings formed by cleavages of C-8-C-14 and C-9-C-11) further suggested that 3a had a dammara- $\Delta^{13(17),24}$ -dien- 3β -yl acetate structure¹⁸⁾ with an undetermined stereochemistry. The structure containing the stereostructure of 3a was determined by direct comparison with an authentic synthetic compound as follows. Thus, 9a, upon treatment with BF₃-etherate in ether, 18) yielded (20R)-dammara13(17),24-dien-3 β -yl acetate (**3a**; isoeuphyl acetate) in almost quantitative yield. The synthetic **3a** and its alkaline-hydrolysis product, isoeuphol (**3**; C₃₀H₅₀O; M⁺, m/z 426.3830), were identical following chromatographic and spectral comparison with natural **3a** and its hydrolysis product, **3**, respectively, and hence, natural **3** was determined as isoeuphol.

The HR-MS of compound 4a showed M^+ at m/z 468.3974 ($C_{32}H_{52}O_2$), and its fragmentation pattern (see Experimental section) was essentially identical with that of 3a. Furthermore, the 1H - and ^{13}C -NMR (Tables 2, 3), and IR spectral properties of 4a were very similar to those of 3a suggesting that 4a had the same structure as that of 3a except for stereochemistry. Furthermore, 4a and its alkaline hydrolysis product, 4 ($C_{30}H_{50}O$; M^+ , m/z 426.3840), were identical following chromatographic and spectral comparison with an authentic specimen of (20S)-dammara-13(17),24-dien- 3β -yl acetate (4a; isotirucallyl acetate), prepared from tirucalla-7,24-dien- 3β -yl acetate (11a) by treatment with BF_3 -etherate in ether, and its hydrolysis product 4, respectively. Hence, natural 4 was characterized as isotirucallol.

Compounds 5a and 6a were a homogeneous mixture in reverse-phase HPLC and GLC. The mixture 5a/6a,

December 1997 2019

Table 2. ¹H-NMR Data (δ/ppm, 400 MHz, CDCl₃) of the Acetates of Seven Novel Triterpene Alcohols from Camellia and Sasanqua Oils^a)

Proton	1a	2a	3a	4a	5a	6a	7a
1	1.40 (α), 1.71 (β)	1.20 (α), 1.64 (β)	1.08 (α), 1.74 (β)	1.10 (α), 1.75 (β)	1.25 (α), 1.69 (β)	1.22 (α), 1.68 (β)	1.06 (α), 1.72 (β)
2	$1.75 (\alpha), 1.69 (\beta)$	1.66 (2H)	1.65 (2H)	1.65 (2H)	1.68 (2H)	1.66 (2H)	1.64 (2H)
3	4.61 (dd, 5.0, 10.7)	4.51 (dd, 4.4, 10.7)	4.50 (dd, 6.2, 10.3)	4.50 (dd, 5.9, 11.0)	4.52 (dd, 4.0, 10.6)	4.52 (dd, 4.0, 11.0)	4.48 (dd, 5.8, 10.4
OAc-3	2.07 (s)	2.05 (s)	2.05 (s)	2.05 (s)	2.06 (s)	2.05 (s)	2.05 (s)
5		1.41 (dd, 6.2, 12.1)	0.85	0.85	1.42	1.42	0.85
6	5.55 (dd, 2.7, 5.5)	$2.15 (\alpha), 1.98 (\beta)$	1.53 (α), 1.38 (β)	1.54 (α), 1.37 (β)	$2.13 (\alpha), 1.95 (\beta)$	$2.13 (\alpha), 1.98 (\beta)$	$1.53 (\alpha), 1.42 (\beta)$
7	5.89 (d, 5.5)	5.37 (dd, 3.1, 3.6)	1.48 (α), 1.36 (β)	$1.50 (\alpha), 1.38 (\beta)$	5.26 (dd, 3.0, 3.6)	5.26 (dd, 3.0, 3.6)	$1.57 (\alpha), 1.30 (\beta)$
9	2.36	2.36	1.42	1.42	2.23	2.23	1.34
11	1.56 (2H)	$1.45 (\alpha), 1.52 (\beta)$	1.51 (α), 1.20 (β)	$1.52 (\alpha), 1.20 (\beta)$	1.50 (2H)	1.53 (2H)	$1.52 (\alpha), 1.24 (\beta)$
12 13	1.70 (α), 1.86 (β)	1.33 (α), 1.40 (β)	1.82 (α), 2.37 (β)	1.83 (α), 2.37 (β)	$1.65 (\alpha), 1.84 (\beta)$	$1.70 \ (\alpha), \ 1.82 \ (\beta)$	1.10 (α), 1.58 (β) 1.68
15	$1.60 (\alpha), 1.42 (\beta)$	$1.58 (\alpha), 1.46 (\beta)$	$1.23 (\alpha), 1.88 (\beta)$	$1.25 (\alpha), 1.89 (\beta)$	1.49 (2H)	1.46 (2H)	1.12 (α), 1.62 (β)
16	$1.30 (\alpha), 1.97 (\beta)$	$1.54 (\alpha), 1.43 (\beta)$	$2.24 (\alpha), 2.06 (\beta)$	2.14 (2H)	1.30 (α), 1.94 (β)	1.30 (α), 1.93 (β)	$1.92 (\alpha), 1.42 (\beta)$
17	1.53				1.52	1.50	2.20
18	0.70 (s)	$1.30 (\alpha), 1.15 (\beta)$	1.06 (s)	1.08 (s)	0.82 (s)	0.83 (s)	0.87 (s)
19	0.87 (s)	1.14, 1.62	0.87 (s)	0.87 (s)	0.77 (s)	0.77 (s)	0.87 (s)
20	1.38	1.88, 1.94	2.46	2.47	1.47	1.47	. ,
21	0.89 (d, 6.6)	5.07 (tt, 1.4, 7.4)	0.92 (d, 6.6)	0.96 (d, 7.0)	0.86 (d, 6.6)	0.86 (d, 7.0)	4.72 (d, 1.5), 4.78 (br s)
22	1.04, 1.39		1.32 (2H)	1.32 (2H)	1.20, 1.67	1.06, 1.78	2.06, 2.18
23	1.84, 2.06	0.84 (s)	1.85, 1.96	1.74, 1.80	1.38, 1.65	1.40, 1.58	1.68 (2H)
24	5.10 (tt, 1.4, 7.1)	0.93 (s)	5.13 (br t, 7.0)	5.07 (tt, 1.5, 7.3)	2.70 (t, 6.0)	2.70 (t, 6.2)	2.75 (t, 6.2)
25		0.77 (s)			,	,	` ' '
26	1.68 (s)	0.95 (s)	1.68 (s)	1.67 (s)	1.32 (s)	1.31 (s)	1.32 (s)
27	1.61 (s)	1.06 (s)	1.59 (s)	1.56 (s)	1.27 (s)	1.27 (s)	1.28 (s)
28	1.10 (s)	0.86 (s)	0.87 (s)	0.86 (s)	0.85 (s)	0.85 (s)	0.86 (s)
29	1.13 (s)	1.68 (s)	0.85 (s)	0.84 (s)	0.94 (s)	0.93 (s)	0.85 (s)
30	1.06 (s)	1.59 (s)	0.85 (s)	0.81 (s)	0.98 (s)	0.98 (s)	0.98 (s)

a) Figures in parentheses denote J values (Hz). J values not included in the Table were not determined.

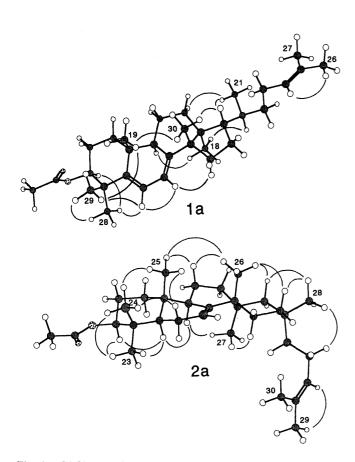


Fig. 1. CAChe Drawings and Some Representative NOE Correlations (—) for Tirucalla-5,7,24-trien-3 β -yl Acetate (1a) and Lemmaphylla-7,21-dien-3 β -yl Acetate (2a)

which possesses a trisubstituted double bond (v_{max} 824, 799 cm⁻¹), showed M⁺ at m/z 484.3900 (C₃₂H₅₂O₃) accompanied with fragment ions at m/z 409 (M⁺ - Me -HOAc), 355 [M⁺ - s.c.(C₈H₁₅O)-2H], 315 (M⁺ - s.c. -42), 255 (315-HOAc), and 241 (255-CH₂).⁹⁾ This indicated that 5a/6a was a mixture of 3β -acetoxy tetracyclic triterpenes containing a monounsaturated ring system and a mono-oxygenated C₈-side-chain. The ring system ¹H signals (see Experimental section) were consistent with the corresponding signals of 9a and 11a,11) whereas the side-chain ¹H signals were very close to those of 24,25epoxylanosteryl acetate (24,25-epoxylanosta-8,24-dien- 3β -yl acetate). ¹⁹⁾ Thus, **5a/6a** were supposed to be either 24,25-epoxylated euph-7-en-3 β -yl (20R) or tirucall-7-en- 3β -yl (20S) acetate. Chromatographic and spectral coincidence of 5a/6a with an authentic specimen of (24R/S)-24,25-epoxybutyrospermyl acetate possessing a 20Rchirality, prepared from 9a by treatment with m-chloroperbenzoic acid (m-CPBA) in dichloromethane, 20) revealed that 5a/6a has a euphane (20R) group. Separation of 5a and 6a from the mixture was achieved by normalphase HPLC in which the less-polar component was designated as 5a and the more-polar one 6a. The stereochemical assignment at C-24 of 5a and 6a was undertaken based on the 13C-NMR chemical shift differences in the side-chain ¹³C signals. Thus, 24R- and 24S-epimers of 24,25-epoxylanosteryl acetate, which possess the same chirality at C-20 (20R) as that of 5a and 6a, exhibited ¹³C-NMR chemical shift differences $(\delta_R - \delta_S)$ in the side-chain 13 C signals as C-20 ($\delta_{\rm R} - \delta_{\rm S} = -0.1$), C-21 (0.1), C-22 (-0.2), C-23 (-0.3), C-24 (-0.2), C-25 (0.3), C-26

2020 Vol. 45, No. 12

Table 3. ¹³C-NMR Data (δ/ppm, 100.6 MHz, CDCl₃) of the Acetates of Seven Novel Triterpene Alcohols from Camellia and Sasanqua Oils

¹³ C	1a	2a	3a	4a	5a	6a	7a	¹³ C	1a	2a	3a	4a	5a	6a	7a
1	36.9	36.5	38.7	38.7	36.8	36.8	38.8	17	52.6	32.7	134.7	134.6	53.0	53.3	47.6
2	23.9	24.2	23.7	23.7	24.2	24.2	23.7	18	21.0	46.6	22.8	23.1	22.1	22.1	15.9
3	79.2	81.1	80.9	81.0	81.1	81.1	80.9	19	15.6	39.8	16.5	16.5	13.1	13.1	16.3
4	39.8	37.8	37.9	37.9	37.8	37.8	37.9	20	35.9	23.0	31.9	31.6	35.8	35.9	151.9
5	150.5	50.8	56.0	56.0	50.8	50.7	56.0	21	18.3	125.3	19.9	20.1	18.7	18.6	107.9
6	119.2	24.0	18.1	18.2	23.8	23.7	18.2	22	36.2	130.7	35.6	35.7	31.5	31.8	31.0
7	115.3	116.1	35.4	35.5	117.7	117.7	35.3	23	25.0	27.5	26.7	26.4	26.0	26.4	27.8
8	147.6	144.7	41.0	41.3	145.9	145.8	40.5	24	125.2	15.7	124.9	125.0	64.7	64.8	64.2
9	46.0	48.0	51.5	51.6	48.8	48.8	50.8	25	131.0	12.9	131.1	131.0	58.4	58.1	58.4
10	37.1	35.2	37.3	37.3	34.8	34.8	37.1	26	25.7	23.8	25.7	25.7	24.9	24.9	25.0
11	16.6	16.4	21.8	22.0	18.1	18.1	21.4	27	17.7	26.8	17.6	17.6	18.8	18.7	18.8
12	33.2	34.8	22.8	23.0	33.9	33.9	24.9	28	26.5	32.1	28.0	28.1	27.6	27.6	28.0
13	43.6	40.2	138.9	139.1	43.6	43.5	45.4	29	25.7	25.7	16.6	16.6	15.9	15.9	16.5
14	50.9	34.4	56.5	56.4	51.3	51.3	49.5	30	24.3	17.6	17.2	16.7	27.3	27.3	15.7
15	32.9	29.5	30.6	30.8	33.8	33.9	31.3	OCOMe	21.4	21.3	21.3	21.3	21.3	21.3	21.3
16	28.1	35.2	29.0	29.1	28.4	28.4	29.0	ОСОМе	170.9	171.0	171.0	171.0	171.0	171.0	171.0

(0.0), and C-27 (0.1).¹⁹⁾ Almost the same differences as those of the lanostanes were observed in the side-chain ¹³C signals between **5a** and **6a**, *viz.*, C-20 ($\delta_{5a} - \delta_{6a} = -0.1$), C-21 (0.1), C-22 (-0.3), C-23 (-0.4), C-24 (-0.1), C-25 (0.3), C-26 (0.0), and C-27 (0.1), as calculated from the ¹³C-NMR data in Table 3, and hence, **5a** and **6a** were attributed to the 24*R*- and 24*S*-epimers of 24,25-epoxy-butyrospermyl acetate, respectively. On alkaline hydrolysis, **5a** and **6a** yielded (24*R*)- (**5**) and (24*S*)-24,25-epoxy-butyrospermol (**6**), respectively.

Compounds 7a and 8a, which were a homogeneous mixture in reverse-phase HPLC and GLC, showed the presence of a terminal methylene group (v_{max} 3072, 1642, $876 \,\mathrm{cm}^{-1}$) in the IR spectrum and M⁺ at m/z 484.3887 (C₃₂H₅₂O₃) in the HR-MS accompanied with fragment ions at m/z 409 (M⁺-Me-HOAc), 357 [M⁺-s.c. $(C_8H_{13}O)-2H$], and 297 (357-HOAc). This suggested that the mixture 7a/8a was a 3β -acetoxy saturated tetracyclic triterpene containing a mono-oxygenated C₈-sidechain with a terminal methylene group. The ¹H-NMR signals (see Experimental section) of 7a/8a were consistent with the corresponding signals of aglaive acetate $\Gamma(24S)$ -24,25-epoxydammar-20-en-3 β -yl acetate; (24S)-24,25epoxydammaradienyl acetate]3,4) and an authentic specimen of (24R/S)-24,25-epoxydammaradienyl acetate prepared from the acetate of dammaradienol (13) by epoxidation.²⁰⁾ Thus, 7a/8a were supposed to be a mixture of (24R/S)-24,25-epoxydammar-20-en-3 β -yl acetate. Normal-phase HPLC enabled the mixture to be separated into less-polar 7a and more-polar 8a. The more-polar 8a had a melting point (mp 167—169 °C) close to the known 24S-epimer, aglaiyl acetate (lit.: mp 161—162 °C, 3) mp 159—164 °C⁴⁾), while the less-polar 7a had a distinctly higher melting point (mp 185—187°C) than 8a. We, therefore, concluded that 7a was a 24R-epimer, which we named isoaglaiyl acetate. Compounds 7a and 8a, upon alkaline hydrolysis, yielded isoaglaiol (7) and aglaiol (8), respectively.

Tables 2 and 3 show the assigned ¹H- and ¹³C-NMR data, respectively, of seven compounds, **1a**—**7a**. Signal assignments were aided by ¹³C distortionless enhancement by polarization transfer (DEPT), ¹H-¹H and ¹³C-¹H

COSYs, HMBC, and difference NOE experiments.

Several naturally occurring sterols with a $\Delta^{5,7}$ -conjugated diene system, represented by ergosterol $\lceil (22E) \rceil$ ergosta-5,7,22-trien-3 β -ol], have been reported.²⁾ However, the tirucallane-type triterpene 1 is the first example of a triterpene alcohol with a $\Delta^{5,7}$ -diene system as a natural product. The natural occurrence of lemmaphyllane (D: C-friedo-18,19-seco-lupane)-type triterpene is extremely rare and the only previously known compound of this type is lemmaphylla-7,21-diene from a fern Lemmaphyllum micropyllum var. obovatum. 15) As for the (20R)-dammar-13(17)-ene (isoeuphane)-type triterpene, the only compound of this type previously reported as a natural product is (20R)-dammara-13(17),24-diene isolated from Polypodium ferns. 18) Aglaiol (8), isolated in this study along with its 24R-epimer 7, has so far been isolated only from the leaves of Aglaia odorata (Meliaceae). 3,4) Among the known compounds described in this paper, bacchara-12,21-dien-3 β -ol (19) has previously been isolated only from the seeds of Glycine max (Leguminosae) and some other Leguminosae seeds, 21) and 17-epilupeol (26), as the acetyl derivative, only from the whole herb of Ixeris chinensis (Compositae).⁶⁾

Triterpene alcohols and plant sterols have recently been demonstrated to possess inhibitory effects on TPA-induced inflammation in mice. 22-24) Thus, the fourteen triterpene alcohols, 3, 4, 5/6 (examined as a C-24 epimeric mixture), 7/8 (C-24 epimeric mixture), 9, 10, 12, 14, 18, 19, 21, and 22, isolated from the Theaceae oils in this study were examined for their inhibitory effects.²⁵⁾ These are shown in Table 4 along with those of nine other triterpene alcohols, which have already been evaluated for their anti-inflammatory activity.²³⁾ These consist of Theaceae triterpene alcohols and the reference compounds, 23) quercetin (3,3',4',5,7-pentahydroxyflavone), a known inhibitor of TPA-induced inflammation in mice, and two commercially available anti-inflammatory drugs, indomethacin and hydrocortisone. All the triterpene alcohols examined here inhibited the TPA-induced inflammation with 0.2-0.9 mg/ear of the 50% inhibitory dose. Although the inhibitory effects of these triterpenes were weaker than that of hydrocortisone, some of them inhibitDecember 1997 2021

Table 4. Inhibitory Effect of Triterpene Alcohols Isolated from Camellia and Sasanqua Oils, and Reference Compounds on TPA-Induced Inflammation in Mice^{a)}

Code	Compound	ID ₅₀ ^{b)} (mg/ear)
3	Isoeuphol	0.3
4	Isotirucallol	0.3
5/6	(24R/S)-24,25-Epoxybutyrospermol ^{c)}	0.5
7/8	(24R/S)-24,25-Epoxydammaradienol ^{c)}	0.5
9	Butyrospermol	0.6
10	Euphol	0.2
12	Tirucallol	0.4
14	24-Methylenedammarenol	0.5
18	24-Methyllanosta-9(11),24(24 ¹)-dienol	0.4
19	Bacchara-12,21-dienol	0.8
21	δ -Amyrin	0.3
22	Germanicol	0.9
11	Tirucalla-7,24-dienol ^{d)}	0.8
13	Dammaradienol ^{d)}	0.8
15	Cycloartenol ^{d)}	0.3
16	24-Methylenecycloartanol ^{d)}	0.2
20	β -Amyrin ^{d)}	0.4
23	α -Amyrin d)	0.2
24	Taraxerol ^{d)}	0.3
25	Lupeol ^{d)}	0.6
27	ψ -Taraxasterol ^{d)}	0.4
	Quercetin ^{d)}	1.6
	Indomethacin ^{d)}	0.3
	Hydrocortisone ^{d)}	0.03

a) Compounds dissolved in CHCl₃-MeOH (1:1, v/v) were applied 30 min before TPA treatment. Ear thickness was determined 8 h after TPA treatment. b) 50% Inhibitory dose. c) C-24 Epimeric mixture in almost equal proportion of each epimer. d) Data taken from refs. 22 and 23b.

ed at a grade almost corresponding to that of indomethacin, and were far more potent than quercetin. As far as the euphane/tirucallane-type compounds were concerned, the euphanes (9, 10) were more potent than the tirucallanes (11, 12), while the Δ^{8} -(10, 12) were more potent than the Δ^7 -(9, 11) isomers. There was an effect of double bond isomerism, moreover, with three pentacyclic oleanaes 20—22, in which 21, possessing a tetrasubstituted $\Delta^{13(18)}$ -double bond, was most inhibitory. 24-Methylenation of tetracyclic triterpenes increased the activity as has been shown with the dammaranes (13, 14) and cycloartanes (15, 16). Inhibitors of TPA-induced inflammation have been demonstrated to have an almost parallel inhibitory effect against tumor promotion. 23a,24) and further study is necessary to investigate the correlation between the structural features of triterpenes and their anti-inflammatory and anti-tumor promoting activities.

Experimental

Crystallizations were performed from acetone–MeOH. Melting points measured were uncorrected. Reverse-phase HPLC was carried out on octadecyl silica columns (25 cm × 10 mm i.d.), on a Superiorex ODS S-5 μ m column (Shiseido Co., Ltd., Tokyo) (HPLC I) and on a TSK ODS-120A 5 μ m column (Toso Co., Tokyo) (HPLC II), with MeOH (4 ml/min) as mobile phase. Normal-phase HPLC was done on a Senshu Pak Silica-4251-N column (25 cm × 10 mm i.d.; Senshu Scientific Co., Tokyo) with n-hexane–EtOAc (97:3, v/v; 4 ml/min) as mobile phase. GLC was performed using a DB-17 fused-silica capillary column (30 m × 0.3 mm i.d., column temp. 275 °C). In both HPLC and GLC, cholesteryl (cholest-5-en-3 β -yl) acetate was the standard for the determination of R t_R of triterpene acetate. Electron-impact MS and HR-MS were recorded at 70 eV. NMR spectra were recorded at 400 MHz (1 H-

NMR) and 100.6 MHz (13 C-NMR) in CDCl₃ with tetramethylsilane (TMS) (1 H-NMR) and CDCl₃ at δ 77.0 (13 C-NMR) as internal standard, and chemical shifts were recorded in δ values. IR and UV spectra were recorded in KBr and EtOH, respectively. Instrumental details and general procedures were the same as described previously. 26 Crude camellia and sasanqua oils were donated by Takada Oil Manufacturing Co. (Ohshima, Tokyo) and Nikko Fine Products Co. (Tokyo), respectively. Sources of seventeen reference triterpene alcohols (9—20, 22—25, 27) were described in our previous article. $^{1.23b,26,27}$

Isolation Procedures Alkaline hydrolysis (5% KOH in MeOH, reflux, 3 h) of crude Theaceae oils (camellia oil, 3.0 kg; sasanqua oil, 5.0 kg) followed by diisopropyl ether extraction yielded neutral NSL (10.4 g; 19.2 g). Column chromatography of the NSL over silica-gel afforded triterpene alcohol fractions (3.3 g; 6.7 g) which, upon acetylation, gave acetylated fractions (2.9 g; 5.2 g). Isolation of individual components from the acetylated fractions was performed by argentic TLC followed by HPLC.

Tirucalla-5,7,24-trien-3 β -yl Acetate (1a) and Tirucalla-5,7,24-trien-3 β ol (1) 1a: mp 123—124 °C. IR v_{max} cm⁻¹: 1733, 1646, 1241, 837, 820. UY λ_{max} nm: 273 (log ε 4.04), 281 (4.00). MS m/z (%): 466 (M⁺, 31), 451 (4), 406 (5), 391 (53), 337 (6), 295 (1), 253 (3), 239 (4), 201 (20), 187 (17), 186 (11), 185 (16), 171 (21), 157 (18), 145 (16), 69 (100). HR-MS: m/z 466.3789 [Calcd for $C_{32}H_{50}O_2$ (M⁺): 466.3807]; 451.3569 [Calcd for $C_{31}H_{47}O_2$: 451.3573]; 391.3309 [Calcd for $C_{29}H_{43}$: 391.3362]; 295.2431 [Calcd for C₂₂H₃₁: 295.2424]; 253.1915 [Calcd for C₁₉H₂₅: 23.1954]; 239.1820 [Calcd for $C_{18}H_{23}$: 239.1798]; 171.1181 [Calcd for $C_{13}H_{15}$: 171.1173]; 69.0704 [Calcd for C_6H_9 : 69.0704]. Alkaline hydrolysis of 1a yielded a free alcohol 1. 1: Amorphous gum. IR $v_{\rm max}$ cm⁻¹: 3430, 1650, 830, 820. MS m/z (%): 424 (M⁺, 33), 409 (6), 391 (28), 337 (7), 311 (6), 271 (5), 253 (5), 239 (3), 201 (6), 187 (10), 186 (6), 185 (10), 171 (13), 157 (13), 149 (16), 69 (100). HR-MS: m/z 424.3712 [Calcd for $C_{30}H_{48}O(M^+)$: 424.3703]. ¹³C- and ¹H-NMR: C-1 [δ_C 37.4; $\delta_{\rm H}$ 1.43(α), 1.71(β)], C-2 [27.5; 1.71(α), 1.67(β)], C-3 [77.2; 3.39, dd, J = 6.2, 9.2 Hz, C-4 [41.1], C-5 [151.5], C-6 [119.0; 5.91, d, J = 5.5 Hz], C-7 [115.4; 5.55, dd, J = 2.9, 5.5 Hz], C-8 [147.0], C-9 [46.1; 2.35; ddd, J=3.3, 8.1, 11.8 Hz], C-10 [37.3], C-11 [16.7; 1.58 (2H)], C-12 [33.3; $1.66(\alpha)$, $1.85(\beta)$], C-13 [43.6], C-14 [50.8], C-15 [32.9; $1.58(\alpha)$, $1.41(\beta)$], C-16 [28.1; $1.30(\alpha)$, $1.99(\beta)$], C-17 [52.6; 1.53], C-18 [21.1; 0.71, s], C-19 [15.5; 0.85, s], C-20 [35.9; 1.38], C-21 [18.3; 0.89, d, $J = 6.6 \,\mathrm{Hz}$], C-22 [36.2; 1.05, 1.42], C-23 [25.0; 1.86, 2.04], C-24 [125.2; 5.10, brt, J=7.0 Hz, C-25 [130.9], C-26 [25.7; 1.69, s], C-27 [17.7; 1.61, s], C-28 [26.5; 1.22, s], C-29 [24.3; 1.07, s], C-30 [24.4; 1.06, s].

Lemmaphylla-7,21-dien-3β-yl Acetate (2a) and Lemmaphylla-7,21-dien-**3** β -ol (2) **2a**: Amorphous gum. IR v_{max} cm⁻¹: 1731, 1249, 817, 800. MS m/z (%): 468 (M⁺, 16), 453 (29), 408 (1), 393 (18), 371 (1), 325 (1), 311 (5), 289 (2), 271 (3), 257 (3), 255 (3), 241 (4), 229 (11), 215 (6), 203 (9), 201 (7), 189 (10), 187 (7), 69 (76), 43 (100). HR-MS: *m/z* 468.3984 [Calcd for C₃₂H₅₂O₂ (M⁺): 468.3965]; 453.3721 [Calcd for C₃₁H₄₉O₂: 453.3729]; 393.3512 [Calcd for $C_{29}H_{45}$: 393.3518]; 325.2933 [Calcd for $C_{24}H_{37}$: 325.2894]; 255.2115 [Calcd for $C_{19}H_{27}$: 255.2112]; 203.1790 [Calcd for $C_{15}H_{23}$: 203.1798]; 189.1648 [Calcd for $C_{14}H_{21}$: 189.1642]; 69.0698 [Calcd for C₅H₉: 69.0704]. Alkaline hydrolysis of 2a yielded a free alcohol 2. 2: mp 142—143 °C. IR v_{max} cm⁻¹: 3380, 840, 817, 800. MS m/z (%): 426 (M⁺, 33), 411 (72), 393 (26), 343 (1), 325 (8), 311 (6), 271 (4), 255 (4), 247 (7), 241 (5), 229 (20), 215 (10), 203 (11), 201 (11), 189 (17), 187 (10), 69 (100). HR-MS: m/z 426.3872 [Calcd for $C_{30}H_{50}O$ (M $^+$): 426.3859]. $^{13}\text{C-}$ and $^{1}\text{H-NMR:}$ C-1 [$\delta_{\rm C}$ 36.8; $\delta_{\rm H}$ 1.14(α), 1.60(β)], C-2 [27.7; 1.60 (2H)], C-3 [79.3; 3.24, dd, J=4.0, 11.4 Hz], C-4 [38.9], C-5 [50.7; 1.32, dd, J = 5.5, 11.7 Hz], C-6 [24.1; 2.16(α), 1.99(β)], C-7 [116.3; 5.38, dd, J=3.1, 3.7 Hz], C-8 [144.6], C-9 [48.1; 2.34], C-10 [35.3], C-11 [16.4; 1.42(α), 1.56(β)], C-12 [34.8; 1.37 (2H)], C-13 [40.2], C-14 [34.4], C-15 [29.5; 1.56(α), 1.48(β)], C-16 [35.2; 1.52(α), 1.42(β)], C-17 [32.7], C-18 [46.6; 1.30(α), 1.14, d, $J = 14.3 \text{ Hz}(\beta)$], C-19 [39.8; 1.16, 1.64], C-20 [23.0; 1.92 (2H)], C-21 [125.3; 5.07, brt, J=7.0 Hz], C-22 [130.7], C-23 [27.5; 0.96, s], C-24 [14.6; 0.85, s], C-25 [12.9; 0.75, s], C-26 [23.8; 0.95, s], C-27 [26.9; 1.06, s], C-28 [32.1; 0.86, s], C-29 [25.7; 1.67, s], C-30 [17.6; 1.59, s].

Isoeuphyl Acetate (3a) and Isoeuphol (3) 3a: mp 92—93 °C. IR $\nu_{\rm max}$ cm $^{-1}$: 1733, 1250, 828. MS m/z (%): 468 (M $^+$, 26), 453 (2), 399 (12), 384 (5), 357 (11), 355 (9), 342 (5), 339 (5), 297 (3), 276 (3), 263 (2), 249 (4), 229 (4), 218 (12), 205 (32), 203 (23), 189 (32), 149 (43), 43 (100). HR-MS: m/z 468.3944 [Calcd for $C_{32}H_{52}O_2$ (M $^+$): 468.3965]; 399.3265 [Calcd for $C_{27}H_{43}O_2$: 399.3261]; 357.2807 [Calcd for $C_{24}H_{37}O_2$: 357.2792]; 355.2597 [Calcd for $C_{24}H_{35}O_2$: 355.2635]; 297.2572 [Calcd

for $C_{22}H_{33}$: 297.2580]; 263.2050 [Calcd for $C_{17}H_{27}O_2$: 263.2009]; 249.1851 [Calcd for $C_{16}H_{25}O_2$: 249.1851]. Alkaline hydrolysis of 3agave a free alcohol 3. 3: Amorphous gum. $\bar{IR} v_{max}$ cm⁻¹: 3387, 828. MS m/z (%): 426 (M⁺, 42), 411 (5), 393 (1), 357 (31), 342 (7), 339 (5), 315 (20), 313 (19), 300 (10), 297 (4), 234 (4), 229 (6), 221 (7), 218 (8), 207 (45), 205 (29), 203 (25), 189 (30), 149 (77), 55 (100). HR-MS: m/z 426.3830 [Calcd for $C_{30}H_{50}O$ (M⁺): 426.3859]. ¹³C- and ¹H-NMR: C-1 [δ_C 39.0; $\delta_{\rm H}$ 1.01(α), 1.75(β)], C-2 [27.4; 1,56(α), 1.63(β)], C-3 [79.0; 3.22, dd, J = 5.1, 11.4 Hz, C-4 [38.9], C-5 [55.9; 0.74], C-6 [18.2; 1.54(α), 1.36(β)], C-7 [35.4; 1.48(α), 1.38(β)], C-8 [41.0], C-9 [51.6; 1.40], C-10 [37.4], C-11 [21.8; 1.52(α), 1.20(β)], C-12 [22.9; 1.84(α), 2.37(β)], C-13 [139.0], C-14 [56.5], C-15 [30.7; 1.22(α), 1.88(β)], C-16 [29.1; 2.24(α), 2.06(β)], C-17 [134.7], C-18 [22.8; 1.06, s], C-19 [16.4; 0.85, s], C-20 [31.9; 2.47], C-21 [19.9; 0.92, d, $J = 7.0 \,\text{Hz}$], C-22 [35.6; 1.31 (2H)], C-23 [26.7; 1.84, 1.98], C-24 [125.0; 5.13, br t, $J = 7.0 \,\text{Hz}$], C-25 [131.1], C-26 [25.7; 1.69, s], C-27 [17.6; 1.59, s], C-28 [28.1; 0.99, s], C-29 [15.5; 0.77, s], C-30

Preparation of Isoeuphyl Acetate (3a) from Butyrospermyl Acetate (9a) 9a (20 mg) in 20% BF₃-etherate in dry diethyl ether (Et₂O; 4 ml) was stirred for 24 h at room temperature. The reaction mixture, after usual work-up followed by reverse-phase HPLC, yielded 3a (11 mg) which was identical with the natural 3a by HPLC, GLC, ¹H-NMR, and MS.

Isotirucallyl Acetate (4a) and Isotirucallol (4) 4a: mp 85-87 °C. IR v_{max} cm⁻¹: 1730, 1248, 829. MS m/z (%): 468 (M⁺, 13), 453 (1), 399 (13), 384 (2), 357 (5), 355 (4), 339 (4), 297 (3), 249 (4), 229 (2), 205 (22), 203 (16), 189 (23), 43 (100). HR-MS: m/z 468.3974 [Calcd for $C_{32}H_{52}O_2$ (M⁺): 468.3965]. Alkaline hydrolysis of **4a** afforded a free alcohol **4**. **4**: mp 155—156 °C. IR ν_{max} cm⁻¹: 3393, 832, 820. MS m/z (%): 426 (M⁺, 36), 411 (2), 357 (45), 342 (4), 339 (6), 315 (17), 313 (11), 300 (7), 297 (4), 234 (2), 229 (5), 221 (6), 218 (7), 207 (42), 205 (26), 203 (23), 189 (26), 175 (16), 161 (39), 149 (69), 69 (100). HR-MS: m/z 426.3840 [Calcd for $C_{30}H_{50}O$ (M⁺): 426.3859]. ¹³C- and ¹H-NMR: C-1 [$\delta_{\rm C}$ 39.0; $\delta_{\rm H}$ $1.00(\alpha)$, $1.74(\beta)$], C-2 [27.4; 1,56(α), 1.64(β)], C-3 [79.0; 3.22, dd, J = 5.1, 11.4 Hz], C-4 [38.9], C-5 [55.9; 0.74], C-6 [18.2; 1.54(α), 1.40(β)], C-7 [35.3; $1.47(\alpha)$, $1.38(\beta)$], C-8 [41.3], C-9 [51.7; 1.41], C-10 [37.4], C-11 $[22.0; 1.52(\alpha), 1.20(\beta)], C-12[23.0; 1.83(\alpha), 2.37(\beta)], C-13[139.1], C-14$ [56.4], C-15 [30.7; 1.25(α), 1.90(β)], C-16 [29.1; 2.12(α), 2.16(β)], C-17 [134.6], C-18 [23.1; 1.08, s], C-19 [16.4; 0.85, s], C-20 [31.6; 2.47], C-21 [20.1; 0.96, d, $J = 7.0 \,\text{Hz}$], C-22 [35.7; 1.30 (2H)], C-23 [26.4; 1.81 (2H)], C-24 [125.0; 5.07, tt, J = 1.5, 7.0 Hz], C-25 [131.0], C-26 [25.7; 1.69, d, J = 1.1 Hz], C-27 [17.6; 1.56, s], C-28 [28.1; 0.99, s], C-29 [15.5; 0.77, s], C-30 [16.7; 0.82, s].

Preparation of Isotirucallyl Acetate (4a) from Tirucalla-7,24-dien-3β-yl Acetate (11a) 11a (26 mg) in 20% BF₃-etherate in dry Et₂O (5 ml) was stirred for 24 h at room temperature, and after usual work-up and reverse-phase HPLC, yielded 4a (17 mg). The synthetic 4a was identical with the natural 4a by HPLC, GLC, ¹H-NMR, and MS.

A Mixture of (24R)- (5a) and (24S)-24,25-Epoxybutyrospermyl Acetate (6a) and a Mixture of (24R)- (5) and (24S)-24,25-Epoxybutyrospermol (6) 5a/6a: Amorphous gum. IR v_{max} cm⁻¹: 1729, 1247, 869, 824, 799. MS m/z (%): 484 (M⁺, 6), 469 (12), 451 (4), 409 (16), 391 (2), 355 (1), 341 (1), 315 (1), 273 (2), 255 (2), 241 (2), 227 (3), 215 (3), 43 (100). HR-MS: m/z 484.3900 [Calcd for $C_{32}H_{52}O_3$ (M⁺): 484.3913]; 409.3449 [Calcd for $C_{29}H_{45}O$: 409.3468]; 355.2633 [Calcd for $C_{24}H_{35}O_2$: 355.2635]; 315.2287 [Calcd for C₂₁H₃₁O₂: 315.2322]; 255.2148 [Calcd for $C_{19}H_{27}$: 255.2111]; 241.1988 [Calcd for $C_{18}H_{25}$: 241.1955]. ¹H-NMR: δ 0.77 (3H, s, H-19), 0.82, 0.83 (each 3H, s, H-18), 0.85 (3H, s, H-28), 0.86 (3H, d, J = 6.0 Hz, H-21), 0.93 (3H, s, H-29), 0.98 (3H, s, H-30), 1.27 (3H, s, H-27), 1.31, 1.32 (each 3H, s, H-26), 2.05 (3H, s, OAc-3), 2.70 (1H, t, J=6.0 Hz, H-24), 4.52 (1H, dd, J=4.7, 11.5 Hz, H-3), 5.25 (1H, dd, J = 3.0, 3.6 Hz, H-7). Alkaline hydrolysis of a portion of the mixture 5a/6a yielded a mixture of free alcohols 5/6. 5/6: mp 57—60 °C. IR v_{max} cm⁻¹: 3485, 865, 824, 799. MS m/z (%): 442 (M⁺, 20), 427 (47), 409 (52), 391 (6), 327 (4), 313 (7), 297 (3), 295 (2), 287 (6), 273 (12), 255 (7), 241 (6), 43 (100). HR-MS: m/z 442.3807 [Calcd for C₃₀H₅₀O₂ (M⁺): 442.3807]. Normal-phase HPLC of the other portion of the acetate mixture 5a/6a yielded isolated less-polar 5a ($Rt_R = 6.09$ on HPLC) and more-polar **6a** (R t_R = 6.42 on HPLC).

(24*R*)-24,25-Epoxybutyrospermyl Acetate (5a) and (24*R*)-24,25-Epoxybutyrospermol (5) 5a: mp 144—146 °C. MS: m/z 484 (M⁺). Alkaline hydrolysis of 5a gave a free alcohol 5. 5: Amorphous gum. MS: m/z 442 (M⁺). ¹H-NMR: δ 0.75 (3H, s, H-19), 0.82 (3H, s, H-18), 0.86 (3H, d, J=6.0 Hz, H-21), 0.86, 0.97 (each 3H, s, H-28, H-29), 0.98 (3H, s, H-30), 1.27 (3H, s, H-27), 1.32 (3H, s, H-26), 2.70 (1H, t, J=6.3 Hz, H-24),

3.24 (1H, dd, J=4.1, 11.0 Hz, H-3), 5.26 (1H, dd, J=3.0, 3.6 Hz, H-7). (24*S*)-24,25-Epoxybutyrospermyl Acetate (6a) and (24*S*)-24,25-Epoxybutyrospermol (6) 6a: mp 169—171 °C. MS: m/z 484 (M⁺). Alkaline hydrolysis of 6a gave a free alcohol 6. 6: Amorphous gum. MS: m/z 442 (M⁺). ¹H-NMR: δ 0.75 (3H, s, H-19), 0.83 (3H, s, H-18), 0.86 (3H, d, J=6.0 Hz, H-21), 0.86, 0.97 (each 3H, s, H-28, H-29), 0.98 (3H, s, H-30), 1.27 (3H, s, H-27), 1.31 (3H, s, H-26), 2.70 (1H, t, J=6.3 Hz, H-24), 3.24 (1H, dd, J=4.1, 11.0 Hz, H-3), 5.26 (1H, dd, J=3.0, 3.6 Hz, H-7).

Preparation of a Mixture of (24R)- (5a) and (24S)-24,25-Epoxybutyrospermyl Acetate (6a) from Butyrospermyl Acetate (9a) To a solution of 9a (50 mg) in dry CH₂Cl₂ (6 ml) was added m-CPBA (20 mg). After stirring overnight at room temperature, the mixture was extracted with Et₂O and, after washing with NaHCO₃ aq. soln. and usual workup (47 mg) followed by reverse-phase HPLC, yielded 5a/6a (30 mg). The synthetic 5a/6a was identical with the natural 5a/6a by HPLC, GLC, ¹H-NMR, and MS.

A Mixture of (24R)- (7a) and (24S)-24,25-Epoxydammaradienyl Acetate (8a) and a Mixture of (24R)- (7) and (24S)-24,25-Epoxydammaradienol **(8)** 7a/8a: mp 175—176 °C. IR v_{max} cm⁻¹: 3072, 1729, 1642, 1245, 876, 865. MS *m/z* (%): 484 (M⁺, 2), 469 (1), 466 (2), 424 (2), 409 (1), 357 (1), 353 (1), 299 (3), 297 (1), 289 (6), 273 (2), 249 (5), 229 (7), 203 (8), 189 (26), 43 (100). HR-MS: m/z 484.3887 [Calcd for $C_{32}H_{52}O_3$ (M⁺): 484.3913]; 409.3453 [Calcd for $C_{29}H_{45}O$: 409.3468]; 357.2844 [Calcd for $C_{24}H_{37}O_2$: 357.2792]; 297.2567 [Calcd for $C_{22}H_{33}$: 297.2580]. ¹H-NMR: δ 0.85 (3H, s, H-29), 0.86 (3H, s, H-28), 0.87 (6H, s, H-18, H-19), 0.98 (3H, s, H-30), 1.28 (3H, s, H-27), 1.32 (3H, s, H-26), 2.05 (3H, s, OAc-3), 2.74, 2.75 (each 1H, d, J=6.0 Hz, H-24), 4.48 (1H, dd, J = 5.8, 10.4 Hz, H-3), 4.72 (1H, t, J = 1.4 Hz), 4.78 (1H, br s) (H-21). Alkaline hydrolysis of a portion of the mixture 7a/8a yielded a mixture of free alcohols 7/8. 7/8: mp 113—115 °C. IR ν_{max} cm $^{-1}$: 3412, 3080, 1640, 885, 860. MS m/z (%): 442 (M⁺, 8), 427 (4), 424 (6), 409 (3), 371 (1), 355 (1), 317 (3), 315 (5), 299 (5), 297 (3), 275 (3), 257 (3), 255 (2), 247 (38), 234 (7), 229 (18), 207 (90), 189 (43), 43 (100). HR-MS: m/z 442.3839 [Calcd for $\rm C_{30}H_{50}O_{2}~(M^{+})\!:$ 442.3807]. Normal-phase HPLC of the other portion of the acetate mixture 7a/8a yielded less-polar 7a $(Rt_R = 4.60 \text{ on HPLC})$ and more-polar 8a $(Rt_R = 4.69 \text{ on HPLC})$.

(24*R*)-24,25-Epoxydammaradienyl (Isoaglaiyi) Acetate (7a) and (24*R*)-24,25-Epoxydammaradienol (Isoaglaiol) (7) 7a: mp 185—187 °C. MS: m/z 484 (M⁺). Alkaline hydrolysis of 7a gave a free alcohol 7. 7: mp 94—96 °C. MS: m/z 442 (M⁺). ¹H-NMR: δ 0.78, 0.98 (each 3H, s, H-28, H-29), 0.85 (3H, s, H-19), 0.87 (3H, s, H-18), 0.98 (3H, s, H-30), 1.28 (3H, s, H-27), 1.32 (3H, s, H-26), 2.75 (1H, t, J=6.3 Hz, H-24), 3.20 (1H, dd, J=5.0, 11.3 Hz, H-3), 4.72 (1H, d, J=1.4 Hz) and 4.78 (1H, br s) (H-21).

(24S)-24,25-Epoxydammaradienyl (Aglaiyl) Acetate (8a) and (24S)-24,25-Epoxydammaradienol (Aglaiol) (8) 8a: mp 167—169 °C (lit.: mp 161—162 °C, $^{3)}$ mp 159—164 °C $^{4)}$). MS: m/z 484 (M $^{+}$). 1 H-NMR: δ 0.85 (3H, s, H-29), 0.86 (3H, s, H-28), 0.87 (6H, s, H-18, H-19), 0.98 (3H, s, H-30), 1.28 (3H, s, H-27), 1.32 (3H, s, H-26), 2.05 (3H, s, OAc-3), 2.75 (1H, t, J = 6.0 Hz, H-24), 4.48 (1H, dd, J = 5.8, 10.4 Hz, H-3), 4.72, 4.78 (each 1H, br s, H-21). Alkaline hydrolysis of 8a gave a free alcohol 8. 8: mp 101—105 °C (lit.: mp 113—115 °C 3). MS: m/z 442 (M $^{+}$).

Preparation of a Mixture of (24R)- (7a) and (24S)-24,25-Epoxy-dammaradienyl Acetates (8a) from Dammaradienyl Acetate (13a) To a solution of 13a (50 mg) in dry $\mathrm{CH}_2\mathrm{Cl}_2$ (6 ml) was added m-CPBA (20 mg). After stirring overnight at room temperature, the mixture was extracted with $\mathrm{Et}_2\mathrm{O}$ and, after washing with NaHCO₃ aq. soln. and usual work-up (48 mg) followed by reverse-phase HPLC, yielded 7a/8a (27 mg; mp 176—178 °C). The synthetic 7a/8a was identical with the natural 7a/8a by HPLC, GLC, $^1\mathrm{H}\text{-NMR}$, and MS.

δ-Amyrin Acetate (21a) and δ-Amyrin (21) 21a: mp 197—200 °C (lit.: mp 208 °C^{5a}); mp 197—198 °C^{5b}) HR-MS: m/z 468.3959 [Calcd for $C_{32}H_{52}O_2$ (M⁺): 468.3964]. The ¹H-NMR data of 21a were consistent with those reported for 21a.^{5a}) Alkaline hydrolysis of 21a gave a free alcohol 21. 21: mp 204—207 °C (lit.: mp 191 °C^{5a}); mp 212—213 °C^{5b}). HR-MS: m/z 426.3875 [Calcd for $C_{30}H_{50}O$ (M⁺): 426.3859]. The MS data of 21 were identical with those reported for 21.^{5c})

17-Epilupeyl Acetate (26a) mp 206—209 °C (lit.: mp 219—221 °C⁶)). HR-MS: m/z 468.3916 [Calcd for $\rm C_{32}H_{52}O_2$ (M $^+$): 468.3964]. The MS, 1 H-NMR, and 13 C-NMR data of 26a were identical with those reported for 26a.⁶)

Assay of TPA-Induced Inflammation in Mice $\,$ The assay procedures were the same as those described in our previous article. $^{23,24)}$

Acknowledgements The authors are indebted to Takada Oil Manufacturing Co. and Nikko Fine Products Co. for a supply of camellia and sasanqua oils, respectively.

References and Notes

- Itoh T., Tamura T., Matsumoto T., Lipids, 9, 173—184 (1974); idem, ibid., 11, 434—441 (1976); Itoh T., Uetsuki T., Tamura T., Matsumoto T., ibid., 15, 407—411 (1980).
- Akihisa T., Kokke W. C. M. C., Tamura T., "Physiology and Biochemistry of Sterols," ed. by Patterson G. W., Nes W. D., American Oil Chemists' Society, Champaign, Illinois, 1991, pp. 172—228
- Shiengthong D., Verasarn A., NaNonggai-Suwanrath P., Warnhoff E. W., Tetrahedron, 21, 917—924 (1965).
- Boar R. B., Damps K., J. Chem. Soc., Perkin Trans. 1, 1977, 510—512.
- a) Bohlmann F., Jacob J., Grenz M., Chem. Ber., 108, 433—436 (1975);
 b) Misra T. N., Singh R. S., Upadhyay J., Srivastava R., J. Nat. Prod., 47, 368—372 (1984);
 c) Bhutani K. K., Kapoor R., Atal C. K., Phytochemistry, 23, 407—410 (1984).
- Shiojima K., Suzuki H., Kodera N., Kubota K., Tsushima S., Ageta H., Chang H.-C., Chen Y.-P., Chem. Pharm. Bull., 42, 2193—2195 (1994).
- Goad L. J., Akihisa T., "Analysis of Sterols," Blackie Academic & Professional, London, 1997.
- Schroepfer G. J., Jr., Parish E. J., Tsuda M., Kandutsch A. A., *Chem. Phys. Lipids*, 47, 187—207 (1988).
- Rahier A., Benveniste P., "Analysis of Sterols and Biologically Significant Steroids," ed. by Nes W. D., Parish E. J., Academic Press, New York, 1989, pp. 223—250.
- 10) Galli G., Maroni S., Steroids, 10, 189-197 (1967).
- Akihisa T., Kimura Y., Kokke W. C. M. C., Takase S., Yasukawa K., Tamura T., J. Chem. Soc., Perkin Trans. 1, 1996, 2379—2384.
- 12) The minimum steric energy of **1a** calculated was 55.55 kcal/mol.
- 13) CAChe with extended MM2 parameters (CAChe Scientific Inc., Beaverton, Oregon, U.S.A.). The conformation with minimum steric energy was obtained from the potential energy map using the "Sequential Search" option. Drawings were performed by the

- Chem3D program (Cambridge Scientific Computing Inc., Cambridge, Massachusetts, U.S.A.).
- Nes W. D., Wong R. Y., Benson M., Landrey J. R., Nes W. R., Proc. Natl. Acad. Sci. U.S.A., 81, 5896—5900 (1984).
- Masuda K., Shiojima K., Ageta H., Chem. Pharm. Bull., 31, 2530—2533 (1983).
- 16) Assignments for the 1 H signals of H-26, H-27, and H-28 of lemmaphylla-7,24-diene¹⁵⁾ should be revised as δ 0.953, 1.068, and 0.860, respectively.
- 17) The minimum steric energy of 2a calculated was 60.40 kcal/mol.
- Arai Y., Masuda K., Ageta H., Chem. Pharm. Bull., 30, 4219—4221 (1982).
- Emmons G. T., Wilson W. K., Schroepfer G. J., Jr., J. Lipid Res., 30, 133—138 (1989).
- 20) Knapp F. F., Schroepfer G. J., Jr., Steroids, 26, 339-357 (1975).
- Akihisa T., Kimura Y., Tamura T., Phytochemistry, 37, 1413—1415 (1994).
- 22) Yasukawa K., Akihisa T., Hyomen, 33, 627-632 (1995).
- 23) a) Kasahara Y., Kumaki K., Katagiri S., Yasukawa K., Yamanouchi S., Takido M., Akihisa T., Tamura T., Phytother. Res., 8, 327—331 (1994); b) Akihisa T., Yasukawa K., Oinuma H., Kasahara Y., Yamanouchi S., Takido M., Kumaki K., Tamura T., Phytochemistry, 43, 1255—1260 (1996).
- 24) Yasukawa K., Takido M., Matsumoto T., Takeuchi T., Nakagawa S., Oncology, 48, 72—76 (1991); Yasukawa K., Akihisa T., Oinuma H., Kaminaga T., Kanno H., Kasahara Y., Tamura T., Kumaki K., Yamanouchi S., Takido M., *ibid.*, 53, 341—344 (1996).
- 25) Preliminary evaluation of the NSL and the triterpene alcohol fractions separated from camellia oil for the inhibitory effect on TPA-induced inflammation in mice revealed that these possess marked inhibitory effects. Their inhibition ratios were 67% and 88%, respectively, at $1.0 \,\text{mg/ear}$ (p < 0.01 by Student's *t*-test as compared to the control group).
- Akihisa T., Kimura Y., Kokke W. C. M. C., Itoh T., Tamura T., Chem. Pharm. Bull., 44, 1202—1207 (1996).
- Akihisa T., Kimura Y., Tamura T., Phytochemistry, 35, 1413—1415 (1994).