Total Synthesis of (\pm) -Plumbazeylanone

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> The first total synthesis of plumbazeylanone (1), which is a trimer of naphthoquinone, was carried out successfully utilizing the unsymmetrical methylene-bridged dimer with the naphthoquinone unit and the naphthol unit, 11b as a key intermediate in 11 steps. This synthesis features a regioselective nucleophilic 1,2-addition reaction and dienone-phenol-type rearrangement.

Key words trimeric naphthoquinone; plumbazeylanone; total synthesis; 1,2-addition; rearrangement

The roots of Plumbago zeylanica (Plumbaginaceae), which is a perennial herb, have long been used in a variety of medicinal applications in many Asian countries. 1) Since 1971, several naphthoquinones^{1,2)} have been isolated from the plant including plumbazeylanone (1) and plumbagin (2a), the latter of which demonstrates in vitro immunosuppressive or cytotoxic activity against primary cell cultures of granulocytes.³⁾ However, the essential active ingredient that provides the biological activity in the medicinal applications of this plant has not yet been established. One suggestion for exploration of the active ingredient has been the absence of constituents such as alkaloids, saponins, and glycosides in the roots of this plant. Plumbazeylanone (1) was selected as the synthetic target due not only to its unique structure, but also in order to determine the active constituent in the medicinal applications of this plant.

Herein, we report the first total synthesis of (±)-plumbazeylanone (1) utilizing the unsymmetrical methylene-bridged dimer 11b as a key intermediate. We investigated the total synthesis of plumbazeylanone using the retrosynthetic plan as shown in Chart 1, in which (i) the condensation reaction between the naphthol 3a and paraformaldehyde followed by transformation to 11b as a key intermediate is performed, (ii) nucleophilic 1,2-addition reaction of the naphthyllithium reagent 4 to C-1 position on 11b followed by a dienone-phenol-type rearrangement of the naphthyl group on 12b to construct the trimer 13 is carried out, and (iii) finally demethylation of six methoxyls on 13, followed by air oxidation to lead to the target compound 1 is achieved.

Compounds 3a and 3b⁴⁾ were first synthesized from the corresponding naphthoquinones 2a and 2c through the reaction sequence shown in Chart 2, respectively. Subsequently, the treatment of 3a with ethylmagnesium bromide (EtMgBr) and then paraformaldehyde gave the symmetrical methylenebridged dimer 8a. The formation of 8a by this reaction can be postulated as shown in Chart 3. Next, selective monomethylation of naphtholic hydroxyl group of 8a using Me₂SO₄ in the presence of Bu₄N⁺HSO₄ gave the corresponding compound 8b in 71% yield. Moreover, the oxidation of the naphthol moiety on 8b using 10% FeCl₃ in MeCN gave the corresponding naphthoquinone 9 in 95% yield. Magnesium bromide hexahydrate (MgBr₂·6H₂O) effected selective demethylation of the methoxyl group at the C-5 position on 9 to afford the corresponding naphthol 11a in excellent yield. The mechanism for this demethylation can be postulated as follows. The selective nucleophilic attack of H₂O on the methyl group of the methoxyl group at the C-5 position on 10 derived from 9 with MgBr₂·6H₂O, in which the oxygen atom of the methoxyl group is chelated to the Mg atom with the carbonyl group at the C-4 position, may take place to form the demethylated compound 11a. The naphtholic hydroxyl group of 11a was protected by the bulky functional group using tert-butyldiphenylsilyl chloride (TBDPSCI) in the presence of 1,8-diazobicyclo[5.4.0]undec-7-ene (DBU) to afford the corresponding O-silylated compound 11b in 88% yield. The selective nucleophilic addition of 4 to the C-1 position on 11b in THF proceeded at -78 °C under a nitrogen atmosphere to give the expected 1,2-adduct

Chart 1

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Chart 3

12a in 78% yield. The deprotection of the TBDPS group of 12a with $Et_4N^+F^-\cdot xH_2O$ gave the corresponding naphthol 12b in quantitative yield. The dienone-phenol-type rearrangement of 12b with 2 N-NaOH in $EtOH^{5}$ stereoselectively gave a trimer 13 in 35% yield. The stereochemistry of 13 was established based on its NOESY experiments. The observation of correlation cross-peaks between the proton signal of C_2 -Me (δ 2.03) and the methylenic proton signals of C_3 -CH₂ (δ 2.85 and 3.25) indicated that C_2 -Me and C_3 -H in 13 have a trans-relative configuration. Finally, demethylation of 13 with AlI₃ in benzene followed by air oxidation afforded (\pm)-plumbazeylanone (1) in 65% yield, mp 245—248 °C. All physical data for the synthetic compound 1 was identical

with those of the natural products. $^{2a,b)}$ Thist first synthesis of (\pm) -1 consists of 11 steps from plumbagin (2a) and an overall yield of 5.9% is achieved.

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