Synthesis of (R)-(-)-3-Methoxymethyl-3-propyl-3,4-dihydrocoumarin from a Chiral Michael Adduct: Absolute Configurations of the Allylated Products of Enantioselective Radical-Mediated Reactions

Masatoshi Murakata,* Yusuke Mizuno, Hiromi Yamaguchi, and Osamu Hoshino*

Faculty of Pharmaceutical Sciences, Science University of Tokyo, Shinjuku-ku, Tokyo 162–0826, Japan. Received April 16, 1999; accepted July 16, 1999

(*R*)-3-Methoxymethyl-3-propyl-3,4-dihydrocoumarin was synthesized, starting from a chiral Michael adduct [(S)-methyl 2,3-dihydro-1-oxo-2-(3-oxobutyl)-1*H*-indene-2-carboxylate], in order to determine the absolute configurations of the products obtained by enantioselective radical-mediated allylation. Aldol cyclization of the Michael adduct proceeded smoothly with suppression of the retro Michael reaction to afford an optically active cyclized product. The Baeyer–Villiger reaction of (*R*)-2-methoxymethyl-2-propylindanone in the presence of BF₃·Et₂O afforded the desired dihydrocoumarin.

Key words absolute configuration; 3-methoxymethyl-3-propyldihydrocoumarin; aldol reaction; Baeyer–Villiger reaction

In the field of asymmetric synthesis, construction of chiral quaternary carbon centers has received wide attention.^{1,2)} Asymmetric induction in radical-mediated reactions is also a current topic in synthetic organic chemistry.^{2,3)} We have reported the efficient creation of chiral quaternary carbon centers by enantioselective radical-mediated reactions catalyzed by a chiral Lewis acid. For example, the reactions of 3-alkyl-3-iododihydrocoumarins 1 with allyltributyltin in the presence of a chiral aluminum reagent proceeded enantioselectively to afford optically active 3-alkyl-3-allyldihydrocoumarins 2 as shown in Chart 1.²⁾ The absolute configurations of the allylated products 2 were determined by circular dichroism measurements of their derivatives.^{2a)} To confirm the result of such an empirical method for determining the absolute configurations of 2, a direct chemical correlation of 2 with a configurationally defined compound seemed necessary. The present paper deals with the synthesis of (R)-3methoxymethyl-3-propyldihydrocoumarin starting from a chiral Michael adduct, [(S)-methyl 2,3-dihydro-1-oxo-2-(3oxobutyl)-1H-indene-2-carboxylate (3)], in order to determine the absolute configurations of 2 obtained by enantioselective radical-mediated allylation.

In our synthetic strategy, the methoxycarbonyl and 3oxobutyl groups of a Michael adduct (S)-3, which is prepared by the asymmetric reaction reported by Wynberg and other workers,⁴⁾ are converted into methoxymethyl and propyl groups, respectively. In this study, the crucial steps are as follows: an aldol cyclization of (S)-3 with suppression of the retro Michael reaction, and the Baeyer–Villiger reaction of an indanone, having a chiral quaternary carbon center, to afford an optically active dihydrocoumarin (Chart 2).

To protect the ketone moiety, conversion of (*S*)-**3** into enone **4** was considered. At first, aldol cyclization of a chiral Michael adduct (*S*)-**3** was carried out as described in the literature.^{4b)} Namely, treatment of (*S*)-**3** with NaOMe in MeOH provided enone **4**, the specific rotation of which, however, had a lower value { $[\alpha]_{577}$ +75.3° (*c*=2.58, benzene)} than that reported in the literature.⁵ This result suggested that partial racemization, which is attributed to a retro Michael reaction and recombination of the side chain under basic conditions, occurred easily through the cyclization reaction. Indeed, cleavage of the side chain of the **3**, 3-oxobutyl group, took place quantitatively on treatment with *tert*-BuOK in *tert*-BuOH to give the carbomethoxyindanone. To improve the optical purity of the enone, we examined the cyclization under mild conditions. The aldol cyclization of (S)-3 in benzene with pyrrolidine and acetic acid was found to give 4 in 61% isolated yield. The degree of specific rotation of this product was $[\alpha]_{577} + 279.3^{\circ}$ (c=2.44, benzene).

With the optically active 4 available, we next focused our attention on the preparation of indanone 13. Protection of the carbonyl group in the enone moiety in 4 provided ketal 5 in 53% yield. For conversion of the ester moiety, reduction of 5 with $LiAlH_4$ was carried out to give alcohol 6 in 92% yield, and methylation of this with MeI and NaH gave 7 in 93% yield. Deprotection of the ketal group in 7, using $SiO_2-10\%$ aqueous oxalic acid, gave enone 8 in 97% yield.⁶⁾ Oxidative cleavage of the enone moiety in 8, by ozone-H₂O₂, gave acid 9 to regenerate the indanone structure and remove one carbon atom. Selective reduction of the carboxyl group was carried out using borane-tetrahydrofuran complex (BH₃·THF) to give indanone 10 having an alcohol moiety, in 96% yield. Chlorination of alcohol 10 gave 11, but reduction with tributyltin hydride did not produce a propyl group. Therefore, 11 was converted into iodide 12 in 88% yield. Reduction of 12 with tributyltin hydride successfully afforded (R)-2-methoxymethyl-2-propylindanone 13.

Next, conversion of indanone **13** into dihydrocoumarin **14** by the Baeyer–Villiger oxidation was examined.^{7,8)} It is known that the reaction of indanone with CF_3CO_3H – CF_3CO_2H , perbenzoic acid or AcOH–H₂O₂, gives dihydrocoumarin.^{7*a*–*c*)} It is also reported that reaction of 2-methylindanones with CF_3CO_3H –Na₂HPO₄ gives the corresponding dihydrocoumarins.^{7*d*} Recently, we reported that oxidation



© 1999 Pharmaceutical Society of Japan



of 2-methoxymethylindanone using m-chloroperoxybenzoic acid (*m*-CPBA) afforded 3-methoxymethyldihydrocoumarin.^{7e)} However, indanone 13 under these conditions did not afford the desired product. Other agents, such as m-CPBA-CF₃CO₂H, *m*-CPBA-Na₂HPO₄, AcOOH, AcOOH-Na₂HPO₄, K₂S₂O₈-H₂SO₄, ceric ammonium nitrate or O₂-PhCHO- Fe_2O_3 , were also found to be unsuitable for the oxidation of 13.⁸⁾ These results are understandable because steric hindrance of the quaternary carbon adjacent to the carbonyl group, the reactivity of which is reduced due to conjugation to the benzene ring, influences the approach of the peracid.⁹⁾ Therefore, reaction in the presence of a Lewis acid, which would coordinate to the carbonyl group of 13 so that the energy of the transition state would be lower, was considered. As expected, the best result was obtained when the reaction was carried out in the presence of BF₃·Et₂O.^{8a)} Reaction of 13 with *m*-CPBA and $BF_3 \cdot Et_2O$ in CH_2Cl_2 at room temperature afforded (R)-14 in 52% yield. The enantiomeric excess of (R)-14 was determined by HPLC using a chiral column and found to be 92%, reflecting the enantiomeric purity of 4.

On the other hand, hydrogenation of **2a** afforded **14**, the specific rotation of which was identical to that of (*R*)-**14** prepared from (*S*)-**3** (Chart 2). Thus, the absolute configuration of all the allylated products obtained by our enantioselective radical-mediated reactions was found to be *R*, because **2b**—**d** had already been correlated with **2a** by a chemical method.^{2a}

In conclusion, the absolute configuration of 2a was unequivocally established. The result is consistent with the previously assigned configuration using circular dichroism measurements. It was also demonstrated that a chiral Michael adduct was useful as a synthon for the synthesis of optically active indanones and dihydrocoumarins. These findings should provide an efficient approach to the syntheses of fiveand six-membered optically active cyclic compounds bearing quaternary carbon centers.

Experimental

General All melting points were measured on a Yanagimoto (hot plate) melting point apparatus and are uncorrected. IR spectra were obtained with a Hitachi 260-10 or a Horiba FT-210 spectrophotometer. ¹H-NMR (270 MHz) and ¹³C-NMR (67.5 MHz) spectra were recorded with a JEOL EX-270 spectrometer in CDCl₃ solution using tetramethylsilane as an internal standard. Mass spectra were measured on a Hitachi M-80, a JEOL JMS D-300 or a JEOL JMS-SX102A spectrometer. The enantiomeric excess (ee) of **14** was determined by HPLC using chiral columns (DAICEL). Column chromatography was performed on silica gel.

A chiral Michael adduct (*S*)-**3** was prepared according to the method reported by Wynberg and his co-workers.^{4*a*,*b*} After recrystallization from benzene–Et₂O, conversion of (*S*)-**3** was carried out.

Methyl (S)-1,2,9,9a-Tetrahydro-3-oxo-3H-fluorene-9a-carboxylate (4)^{4b} To a solution of (S)-3 {10.0 g, 38.5 mmol, $[\alpha]_{577}^{rt}$ -70.4° (c=2.16, benzene)} in benzene (175 ml) were added pyrrolidine (13 ml, 156 mmol) and AcOH (20 ml, 349 mmol). The mixture was stirred for 20 h at room temperature. After addition of water, the resulting mixture was extracted with benzene (50 ml×2). The extracts were washed successively with 10% aqueous HCl, saturated NaHCO₃ and saturated NaCl, and dried over MgSO₄. The solvent was removed under reduced pressure, and the crude product was purified by column chromatography (Et_2O : hexane=1:1) followed by rinsing with Et₂O-hexane to afford (S)-4 (5.7 g, 61%). mp 114–115 °C. $[\alpha]_{577}^{rt}$ $+279.3^{\circ}$ (c=2.44, benzene), $[\alpha]_{546}^{27}$ +354.0° (c=1.99, benzene). IR (Nujol) cm⁻¹: 2960, 2943, 2937, 2912, 2858, 1724, 1662, 1632, 1601, 1259, 1201, 1182. ¹H-NMR δ : 2.10—2.22 (1H, m), 2.51—2.56 (2H, m), 2.73—2.80 (1H, m), 3.05, 3.57 (each 1H, d, J=16.7 Hz), 3.63 (3H, s), 6.37 (1H, s), 7.29–7.63 (4H, m). ¹³C-NMR δ : 32.9, 35.1, 43.0, 52.8, 54.4, 119.1 122.8, 125.3, 127.7, 131.9, 138.0, 145.6, 164.4, 173.8, 198.6.

(S)-3,3-Ethylenedioxy-9a-methoxycarbonyl-1,2,9,9a-tetrahydro-3H**fluorene (5)** To a solution of (S)-4 (10.0 g, 41.0 mmol) in benzene (150 ml) were added ethylene glycol (38 ml, 680.0 mmol) and pyridinium p-toluenesulfonate (PPTS) (2 g, 11.6 mmol). The resulting mixture was refluxed for 4d in a flask equipped with a Dean-Stark trap. After addition of saturated NaHCO₃, the resulting mixture was extracted with benzene (50 ml \times 2). The extracts were washed successively with water and saturated NaCl, and dried over K₂CO₃. The solvent was removed under reduced pressure, and the crude product was purified by column chromatography (benzene: acetone= 8:1) followed by rinsing with Et₂O-hexane to afford (S)-5 (6.23 g, 53%). mp 169—172 °C (iso-Pr₂O). $[\alpha]_{546}^{30^{-}}$ +178.0° (*c*=2.02, benzene). IR (Nujol) cm⁻¹: 2954, 2881, 1716, 1670, 1603, 1441, 1360, 1267, 1196, 1120, 1072. ¹H-NMR δ : 1.90—2.03 (3H, m), 2.45—2.57 (1H, m), 2.93, 3.38 (each 1H, d, J=15.8 Hz), 3.58 (3H, s), 3.63-3.97 (1H, m), 4.01-4.10 (3H, m), 5.90 (1H, d, J=1.0 Hz), 7.18—7.25 (3H, m), 7.33—7.49 (1H, m). ¹³C-NMR δ : 32.1, 32.5, 43.1, 52.5, 54.4, 64.4, 65.0, 106.6, 118.2, 121.1, 124.8, 127.0, 128.9, 139.7, 142.8, 147.5, 175.6. MS m/z: 286 (M⁺). HR-MS Calcd for C₁₇H₁₈O₄ 286.1203, Found 286.1200. Anal. Calcd for C₁₇H₁₈O₄: C, 71.31; H, 6.34. Found: C, 71.24; H, 6.30.

(S)-3,3-Ethylenedioxy-9a-hydroxymethyl-1,2,9,9a-tetrahydro-3*H*-fluorene (6) To a suspension of LiAlH₄ (1.2 g, 31.6 mmol) in tetrahydrofuran (THF) (100 ml) was added a solution of (S)-5 (5.61 g, 19.6 mmol) in THF (80 ml) at 0 °C. After being stirred at 0 °C for 30 min, the reaction mixture was quenched with water (1.2 ml), followed by 15% aqueous NaOH, and

then more water (3.6 ml) at 0 °C. The resulting mixture was dried over K_2CO_3 . The solvent was removed under reduced pressure, and the crude product was purified by rinsing with Et_2O -hexane to afford (*S*)-**6** (4.74 g, 92%). mp 143—144 °C (iso-Pr₂O). $[\alpha]_{546}^{29}$ +136.1° (*c*=1.02, benzene). IR (Nujol) cm⁻¹: 3435, 1672, 1604, 1462, 1356, 1292, 1161, 1120, 1088. ¹H-NMR δ : 1.65 (1H, t, *J*=4.6 Hz), 1.76—2.00 (2H, m), 2.14—2.31 (2H, m), 2.62, 3.07 (each 1H, d, *J*=15.8 Hz), 3.30, 3.64 (each 1H, dd, *J*=4.6, 10.9 Hz), 3.89—4.10 (4H, m), 5.85 (1H, d, *J*=1.7 Hz), 7.15—7.41 (4H, m). ¹³C-NMR δ : 28.7, 31.1, 41.1, 48.0, 64.4, 64.9, 65.6, 106.8, 118.0, 121.3, 125.6, 126.8, 129.0, 139.3, 144.2, 150.2. MS *m/z*: 258 (M⁺). HR-MS Calcd for C₁₆H₁₈O₃ 258.1255, Found 258.1262. *Anal.* Calcd for C₁₆H₁₈O₃: C, 74.40; H, 7.02. Found: C, 74.29; H, 7.01.

(S)-3,3-Ethylenedioxy-9a-methoxymethyl-1,2,9,9a-tetrahydro-3H-fuluorene (7) To a suspension of NaH (60% in oil, 1.45 g, 36.2 mmol) in dimethylformamide (DMF) (15 ml) was added MeI (5.8 ml, 91 mmol) at 0 °C. A solution of (S)-6 (4.67 g, 18.1 mmol) in DMF (25 ml) was added at 0 °C, the mixture was stirred at room temperature for 3 h. After addition of water (100 ml) at 0 °C, the resulting mixture was diluted with benzene (60 ml) and AcOEt (120 ml). The organic layer was separated and washed with water, saturated NaCl, and dried over MgSO4. The solvent was removed under reduced pressure, and the crude product was purified by column chromatography (Et₂O:hexane=2:1) followed by distillation to afford (S)-7 (4.6 g, 93%) as an oil. bp 225 °C (2 mmHg) (bulb-to-bulb distillation). $[\alpha]_{546}^{30}$ $+126.0^{\circ}$ (c=1.01, benzene). IR (NaCl) cm⁻¹: 2947, 2877, 1668, 1470, 1464, 1126, 1111, 1092. ¹H-NMR δ : 1.70–2.35 (4H, m), 2.60 (1H, d, J= 15.8 Hz), 2.96 (1H, d, J=9.9 Hz), 3.18 (1H, d, J=15.8 Hz), 3.28 (3H, s), 3.43 (1H, d, J=9.9 Hz), 3.91-4.10 (4H, m), 5.82 (1H, d, J=1.3 Hz), 7.14-7.42 (4H, m). ¹³C-NMR δ: 28.6, 30.9, 41.3, 47.2, 59.1, 64.4, 64.8, 74.6, 106.8, 117.6, 121.2, 125.6 126.6, 128.9, 139.4, 144.4, 150.6. MS m/z: 272 (M⁺). HR-MS Calcd for $\rm C_{17}H_{20}O_3$ 272.1411, Found 272.1415.

(S)-9a-Methoxymethyl-1,2,9,9a-tetrahydro-3-oxo-3H-fuluorene (8) To a suspension of silica gel (40 g; Silica gel 60 , Merck, 70-230 mesh) in CH₂Cl₂ (60 ml) was added a solution of 10% aqueous oxalic acid (4.1 g) at room temperature, and the mixture was stirred for 10 min. A solution of (S)-7 (4.54 g, 16.7 mmol) in CH₂Cl₂ (20 ml) was added dropwise at room temperature, followed by stirring for 30 min. After addition of NaHCO₃ (powder, 1.4 g), the mixture was stirred for 5 min. The solid phase was separated by filtration and rinsed with CH₂Cl₂. Removal of the solvent under reduced pressure afforded (S)-8 (3.7 g, 97%). mp 96—97 °C (Et₂O). $[\alpha]_{546}^{29} + 329.5^{\circ}$ (c=1.03, benzene). IR (Nujol) cm⁻¹: 2924, 2854, 1659, 1601, 1464, 1105. ¹H-NMR δ : 1.91–2.01 (1H, m), 2.44–2.72 (3H, m), 2.74 (1H, d, J=16.5 Hz), 3.13 (1H, d, J=9.2 Hz), 3.27 (1H, d, J=16.5 Hz), 3.32 (3H, s), 3.46 (1H, d, J=9.2 Hz), 6.32 (1H, s), 7.30–7.58 (4H, m). ¹³C-NMR δ : 29.8, 34.1, 41.7, 48.2, 59.3, 75.7, 118.7, 123.0, 126.0, 127.3, 131.8, 137.8, 147.1, 168.4, 199.3. MS m/z: 228 (M⁺). HR-MS Calcd for C₁₅H₁₆O₂ 228.1128, Found 228.1147. Anal. Calcd for C15H16O2: C, 78.92; H, 7.06. Found: C, 79.03: H. 7.05.

(R)-3-(2-Methoxymethyl-1-oxoindan-2-yl)propanoic Acid (9) A solution of (S)-8 (1.85 g, 8.1 mmol) in MeOH (90 ml)-CH₂Cl₂ (90 ml) was cooled to -78 °C. Ozone was introduced at -78 °C over a period of 5 h in order to keep the color of the solution blue. The solution was then purged with argon, allowed to warm to room temperature. To the resulting solution were added AcOH (3 ml), water (3 ml), 36% aqueous HCl (1 ml) and 30% aqueous H₂O₂ (12 ml), in that order. After being stirred for 20 h at room temperature, 10% Pd-C (1 g) was added, and the mixture was stirred for 17 h. After filtration, the solution was concentrated to a volume of ca. 5 ml. The resulting mixture was diluted with Et₂O, and then treated with saturated aqueous Na2SO3 to make the aqueous phase basic. The aqueous layer was separated, and then acidified with 10% aqueous HCl. The mixture was extracted with Et₂O (100 ml \times 2) and the extracts were washed with saturated NaCl and dried over MgSO4. The solvent was removed under reduced pressure, and the crude product was purified by rinsing with Et₂O-hexane to afford (*R*)-9 (1.71 g, 85%). mp 149—150 °C (benzene). $[\alpha]_{546}^{28}$ -3.7° (*c*= 1.06, CHCl₃). IR (KBr) cm⁻¹: 3053, 2926, 1713, 1687, 1606, 1435, 1107. ¹H-NMR δ: 1.87–2.06 (2H, m), 2.16–2.37 (2H, m), 2.93 (1H, d, J=17.5 Hz), 3.26 (3H, s), 3.37 (1H, d, J=17.5 Hz), 3.45, 3.57 (each 1H, d, J=8.9 Hz), 7.33—7.75 (4H, m). ¹³C-NMR δ: 28.8, 29.2, 35.9, 52.9, 59.3, 77.0, 124.1, 126.5, 127.5, 135.2, 136.5, 153.2, 178.7, 208.3. MS m/z: 248 (M⁺). HR-MS Calcd for C14H16O4 248.1046, Found 248.1044. Anal. Calcd for C₁₄H₁₆O₄: C, 67.73; H, 6.50. Found: C, 67.82; H, 6.50.

(*R*)-2-(3-Hydroxypropyl)-2-methoxymethyl-1-indanone (10) To a solution of (*R*)-9 (1.65 g, 6.7 mmol) in THF (30 ml) was added a 0.98 M solution of BH₃·THF complex (6.7 ml, 6.6 mmol) over a period of 30 min at $-14 \,^{\circ}$ C, and the mixture was stirred for 1 h. The mixture was warmed to

0 °C and then stirred for 1 h. Additional BH₃·THF complex (1 ml, 0.98 mmol) was added, and the mixture was stirred for 5 h at 0 °C. After addition of water at 0 °C, the resulting mixture was extracted with Et₂O (100 ml×2). The extracts were washed with water, saturated NaCl, and dried over MgSO₄. The solvent was removed under reduced pressure, and the crude product was purified by distillation to afford (*R*)-**10** (1.52 g, 96%) as an oil. bp 200 °C (2 mmHg) (bulb-to-bulb distillation). $[\alpha]_{546}^{30}$ -16.1° (*c*=0.75, CHCl₃). IR (NaCl) cm⁻¹: 3406, 2927, 2877, 1709, 1608, 1466, 1109. ¹H-NMR δ : 1.25—1.76 (4H, m), 2.99 (1H, d, *J*=17.3 Hz), 3.24 (3H, s), 3.33 (1H, d, *J*=17.3 Hz), 3.43 (1H, d, *J*=8.6 Hz), 7.31—7.73 (4H, m). ¹³C-NMR δ : 27.1, 30.8, 35.6, 53.4, 59.3, 62.7, 77.1, 123.8, 126.4, 127.2, 134.9, 136.8, 153.7, 209.5. MS *m/z*: 234 (M⁺). HR-MS Calcd for C₁₄H₁₈O₃ 234.1254, Found 234.1246.

(R)-2-(3-Chloropropyl)-2-methoxymethyl-1-indanone (11) A mixture of (R)-10 (1.46 g, 6.23 mmol) and SOCl₂ (10 ml) was stirred for 4 h at room temperature. To this solution was added pyridine (1 ml) and the resulting mixture was stirred for 2 h at room temperature, then refluxed for 6 h. The mixture was poured into ice-water, and then extracted with Et_2O (50 ml×2). The extracts were washed successively with 10% aqueous HCl, saturated NaHCO3 and saturated NaCl, and dried over MgSO4. Removal of the solvent under reduced pressure afforded (R)-11 (970 mg, 62%) as an oil. bp 200-210 °C (3 mmHg) (bulb-to-bulb distillation). $[\alpha]_{546}^{25}$ -0.7° (c=1.08, benzene). IR (NaCl) cm⁻¹: 2926, 1709, 1608, 1463, 1275, 1105, 744. ¹H-NMR δ: 1.54—1.79 (4H, m), 2.97 (1H, d, J=17.5 Hz), 3.26 (3H, s), 3.37 (1H, d, J=17.5 Hz), 3.41-3.46 (2H, m,), 3.44, 3.59 (each 1H, d, J=8.7 Hz), 7.32-7.75 (4H, m). ¹³C-NMR δ: 27.2, 32.0, 35.7, 45.0, 53.2, 59.3, 77.0, 123.9, 126.4, 127.3, 135.0, 136.7, 153.4, 208.8. MS m/z: 252, 254 (M⁺). HR-MS Calcd for C₁₄H₁₇³⁵ClO₂ 252.0915, Found 252.0909, Calcd for C₁₄H₁₇³⁷ClO₂ 254.0887, Found 254.0904.

(*R*)-2-(3-Iodopropyl)-2-methoxymethyl-1-indanone (12) To a solution of (*R*)-11 (866.3 mg, 3.43 mmol) in acetone (10 ml) was added NaI (5.14 g, 3.43 mmol). After being refluxed for 7 h, the solvent was removed under reduced pressure. After addition of water, the resulting mixture was extracted with CHCl₃ (50 ml×2) and the extracts were washed with 10% Na₂S₂O₃ and saturated NaCl, and dried over MgSO₄. Removal of the solvent under reduced pressure afforded (*R*)-12 (1.04 g, 88%) as an oil. bp 200—210 °C (2 mmHg) (bulb-to-bulb distillation). $[\alpha]_{546}^{25}$ –0.6° (*c*=1.00, benzene). IR (NaCl) cm⁻¹: 2926, 2848, 1709, 1608, 1464, 1296, 1277, 1200, 1111, 741. ¹H-NMR δ : 1.58—1.76 (4H, m), 2.96 (1H, d, *J*=17.5 Hz), 3.05—3.11 (2H, m), 3.26 (3H, s), 3.35 (1H, d, *J*=17.5 Hz), 3.42, 3.58 (each 1H, d, *J*=8.6 Hz), 7.33—7.76 (4H, m). ¹³C-NMR δ : 6.4, 28.1, 35.6, 35.8, 53.5, 59.4, 77.0, 124.0, 126.5, 127.4, 135.1, 136.7, 153.5, 208.8. MS *m/z*: 344 (M⁺). HR-MS Calcd for C₁₄H₁₇IO₂ 344.0273, Found 344.0268.

(*R*)-2-Methoymethyl-2-propyl-1-indanone (13) Under an argon atmosphere, to a solution of (*R*)-12 (944.9 mg, 2.75 mmol) in toluene (10 ml) were added Bu₃SnH (0.62 ml, 3.0 mmol) and 2.2'-azobisisobutyronitrile (AIBN) (10 mg, 0.04 mmol). After being refluxed for 5 h, the solvent was removed under reduced pressure, and the crude product was purified by column chromatography (Et₂O:acetone=10:1) to afford (*R*)-13 (385.1 mg, 64%) as an oil. bp 150 °C (2 mmHg) (bulb-to-bulb distillation). [α]₅₄₆²⁴ + 6.6° (*c*=1.07, benzene). IR (NaCl) cm⁻¹: 2958, 2929, 2873, 1713, 1610, 1466, 1113. ¹H-NMR δ : 0.83 (3H, t, *J*=7.1 Hz), 0.98—1.29 (2H, m), 1.47—1.67 (2H, m), 2.98 (1H, d, *J*=17.2 Hz), 3.24 (3H, s), 3.32 (1H, d, *J*=17.2 Hz), 3.42, 3.61 (each 1H, d, *J*=8.7 Hz), 7.30—7.99 (4H, m). ¹³C-NMR δ : 14.5, 17.2, 35.6, 37.1, 54.0, 59.3, 77.2, 123.8, 126.3, 127.1, 134.7, 137.0, 153.7, 209.5. MS *m/z*: 218 (M⁺). HR-MS Calcd for C₁₄H₁₈O₂ 218.1306, Found 218.1307.

(*R*)-3-Methoxymethyl-3-propyl-3,4-dihydrocoumarin (14) Conversion of **2a** to **14**: A solution of **2a** (74% ee, 300.1 mg, 1.29 mmol), which was prepared by an enantioselective radical reaction according to Chart 1,^{2a} in Et₂O (10 ml) was hydrogenated over 10% Pd–C (30 mg) under H₂ at room temperature for 6 h. After filtration, the solvent was removed under reduced pressure, and the crude product was purified by distillation to afford (*R*)-**14** (290.9 mg, 96%) as an oil. bp 130–140 °C (1 mmHg) (bulb-to-bulb distillation). [α]_D²⁷ – 12.0° (*c*=1.01, acetone). IR (NaCl) cm⁻¹: 2960, 2931, 2873, 1767, 1591, 1167, 1109. ¹H-NMR δ : 0.83 (3H, t, *J*=7.3 Hz), 1.16–1.45 (2H, m), 1.48–1.56 (2H, m), 2.80, 3.22 (each 1H, d, *J*=16.3 Hz), 3.38 (3H, s), 3.51 (1H, d, *J*=9.2 Hz), 3.68 (1H, d, *J*=9.2 Hz), 6.98–7.26 (4H, m). ¹³C-NMR δ : 14.3, 17.3, 32.2, 35.0, 45.4, 59.5, 74.6, 116.0, 121.8, 124.4, 127.9, 128.6, 150.9, 170.6. MS *m*/*z*: 234 (M⁺). HR-MS Calcd for C₁₄H₁₈O₃ 234.1254, Found 234.1252.

Conversion of **13** to **14**: To a solution of (*R*)-**13** (150 mg, 0.69 mmol) in CH_2Cl_2 (18 ml) were added 70% *m*-CPBA (1.83 g, 7.42 mmol) and BF_3 . Et₇O (1.5 ml, 11.8 mmol), and the resulting mixture was stirred for 42 h at

room temperature. After addition of hexane, the mixture was filtered. The filtrate was washed with saturated NaHCO₃, saturated NaCl, and dried over MgSO₄. The solvent was removed under reduced pressure, and the crude product was purified by column chromatography (benzene) followed by distillation to afford (*R*)-**14** (83.9 mg, 52%) as an oil. bp 130—140 °C (1 mmHg) (bulb-to-bulb distillation). 92% ee; HPLC [chiralcel OD; hexane :2-propanol=50:1; flow rate 0.5 ml/min]. $[\alpha]_D^{27}$ -16.5° (*c*=1.13, acetone). ¹H-NMR and mass spectra were identical to those for the product obtained from **2a**.

References and Notes

- For reviews on the creation of quaternary carbon centers, see: a) Martin S. F., *Tetrahedron*, **36**, 419–460 (1980); b) Tomioka K., Koga K., *Yuki Gosei Kagaku Kyokai Shi*, **44**, 545–557 (1986); c) Fuji K., *Chem. Rev.*, **93**, 2037–2066 (1993); d) Corey E. J., Guzman-Perez A., *Angew. Chem. Int. Ed. Engl.*, **37**, 388–401 (1998). For recent reports of the formation of stereogenic quaternary carbon centers, see: e) Ishibashi N., Miyazawa M., Miyashita M., *Tetrahedron Lett.*, **39**, 3775–3778 (1998); f) Akeboshi T., Ohtsuka Y., Sugai T., Ohta H., *Tetrahedron*, **54**, 7387–7394 (1998); g) Ruble J. C., Fu G. C., *J. Am. Chem. Soc.*, **120**, 11532–11533 (1998) and references cited therein.
- a) Murakata M., Jono T., Mizuno Y., Hoshino O., J. Am. Chem. Soc., 119, 11713—11714 (1997). b) Murakata M., Jono T., Hoshino O., Tetrahedron: Asymmetry, 9, 2087—2092 (1998).
- For recent examples of enantioselective radical-mediated reactions, 3) see: a) Murakata M., Tsutsui H., Hoshino O., J. Chem. Soc., Chem. Commun., 1995, 481-482; b) Urabe H., Yamashita K., Suzuki K., Kobayashi K., Sato F., J. Org. Chem., 60, 3576-3577 (1995); c) Wu J. H., Radinov R., Porter N. A., J. Am. Chem. Soc., 117, 11029-11030 (1995); d) Nishida M., Hayashi H., Nishida A., Kawahara N., Chem. Commun., 1996, 579-580; e) Sibi M. P., Ji J., Wu J. H., Gürtler S., Porter N. A., J. Am. Chem. Soc., 118, 9200-9201 (1996); f) Nanni D., Curran D. P., Tetrahedron: Asymmetry, 7, 2417-2422 (1996); g) Haque M. B., Roberts B. P., Tetrahedron Lett., 37, 9123-9126 (1996); h) Wu J. H., Zhang G., Porter N. A., ibid., 38, 2067-2070 (1997); i) Fhal A.-R., Renaud P., ibid., 38, 2661-2664 (1997); j) Sibi M. P., Ji J., J. Org. Chem., 62, 3800-3801 (1997); k) Sibi M. P., Shay J. J., Ji J., Tetrahedron Lett., 38, 5955-5958 (1997); l) Porter N. A., Wu J. H., Zhang G., Reed A. D., J. Org. Chem., 62, 6702-6703 (1997); m) Blumenstein M., Schwarzkopf K., Metzger J. O., Angew. Chem. Int. Ed.

Engl., **36**, 235—236 (1997); Schwarzkopf K., Blumenstein M., Hayen A., Metzger J. O., *Eur. J. Org. Chem.*, **1998**, 177—181 and references cited therein. For a recent review of Lewis acids in free radical reactions including enantioselective reactions, see: Renaud P., Gerster M., *Angew. Chem. Int. Ed. Engl.*, **37**, 2562—2579 (1998).

- Preparation of (S)-3 and cyclization have been reported, see: a) Wynberg H., Helder R., *Tetrahedron Lett.*, 1975, 4057–4060; b) Hermann K., Wynberg H., J. Org. Chem., 44, 2238–2244 (1979). Selected examples of other asymmetric syntheses of 3, see: c) Kobayashi N., Iwai K., J. Am. Chem. Soc., 100, 7071–7072 (1978); d) Sera A., Takagi K., Katayama H., Yamada H., J. Org. Chem., 53, 1157–1161 (1988); e) Cram D. J., Sogah G. D. Y., J. Chem. Soc., Chem. Commun., 1981, 625–628; f) Brunner H., Hammer B., Angew. Chem. Int. Ed. Engl., 23, 312–313 (1984); g) Desimoni G., Quadrelli P., Righetti P. P., Tetrahedron, 46, 2927–2934 (1990); h) Desimoni G., Dusi G., Faita G., Quadrelli P., Righetti P. P., *ibid.*, 51, 4131–4144 (1995) and references cited therein.
- 5) The specific rotation of (S)-3 in our study showed [α]^{tr}₅₇₇ -70.4° (c= 2.16, benzene). [α]^{tr}₅₇₈ -77° (c=2, benzene) for (S)-3 has been reported, see: ref. 4a and b. It has also been reported in ref. 4b that enone 4 was obtained as two crops of crystals {major crop: [α]^{tr}₅₇₈ +41° (c=2.1, benzene), minor crop: [α]^{tr}₅₇₈ +313° (c=2.0, benzene)} by the cyclization of (S)-3 under basic conditions (using NaOMe) followed by fractional recrystallization.
- 6) Huet F., Lechevallier A., Pellet M., Conia J. M., Synthesis, **1978**, 63–65.
- For selected examples of the synthesis of dihydrocoumarins from indanones by the Baeyer–Villiger reaction, see: a) Hawthorne M. F., Emmons W. D., J. Am. Chem. Soc., 80, 6398–6404 (1958); b) Mateos J. L., Menchaca H., J. Org. Chem., 29, 2026–2028 (1964); c) Chatterjee A., Bhattacharya S., Banerji J., Ghosh P. C., Indian J. Chem., Sect. B, 15B, 214–216 (1997); d) Sonnet P. E., Oliver J. E., J. Heterocycl. Chem., 11, 263–265 (1974); e) Murakata M., Tsutsui H., Hoshino O., Heterocycles, 46, 517–522 (1997).
- For a review of the Baeyer–Villiger reaction, see: *a*) Krow G. R., Org. React., 43, 251 (1993). Fe₂O₃-Catalyzed Baeyer–Villiger oxidation has been reported, see: *b*) Murahashi S., Oda Y., Naota T., Tetrahedron Lett., 33, 7557–7560 (1992).
- 9) For reports of the rate of the Baeyer–Villiger reaction, see: refs 7a and b.