

## Studies on the Constituents of *Viburnum* Species. XIX.<sup>1)</sup> Six New Triterpenoids from *Viburnum dilatatum* THUNB.

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Six new triterpenoids, viburnudienone B<sub>1</sub> methyl ester (**1**), viburnudienone B<sub>2</sub> methyl ester (**2**), viburnudienone H<sub>1</sub> (**3**), viburnudienone H<sub>2</sub> (**4**), viburnenone B<sub>1</sub> methyl ester (**5**) and viburnenone B<sub>2</sub> methyl ester (**6**), were isolated from the leaves of *Viburnum dilatatum* (Caprifoliaceae). Their structures were established by extensive spectroscopic studies.

**Key words** *Viburnum dilatatum*; Caprifoliaceae; triterpene; dammarane; viburnudienone; viburnenone

We have recently reported the isolation of eleven, new class, modified dammarane-type triterpenoids, viburnols A–K, from the CHCl<sub>3</sub> extract of the leaves of *Viburnum dilatatum*.<sup>1,2)</sup> In the present study, six new dammarane-type triterpenoids, viburnudienone B<sub>1</sub> methyl ester (**1**), viburnudienone B<sub>2</sub> methyl ester (**2**), viburnudienone H<sub>1</sub> (**3**), viburnudienone H<sub>2</sub> (**4**), viburnenone B<sub>1</sub> methyl ester (**5**) and viburnenone B<sub>2</sub> methyl ester (**6**), were isolated. This paper deals with the structural elucidation of these compounds.

### Results and Discussion

Viburnudienone B<sub>1</sub> methyl ester (**1**) was obtained as an amorphous powder,  $[\alpha]_D^{25} +26.4^\circ$ . The molecular formula was determined to be C<sub>31</sub>H<sub>46</sub>O<sub>5</sub> by high-resolution (HR)-MS. The <sup>1</sup>H- and <sup>13</sup>C-NMR spectra of **1** were very similar to those of viburnol B methyl ester,<sup>2)</sup> except for the signals due to the side chain part (C<sub>20</sub>–C<sub>27</sub>). As regarding the side chain part, the spectral data revealed the presence of a dienone [269 nm; 1668, 1621 cm<sup>-1</sup>;  $\delta_H$  6.05 (1H, m, H-24), 6.01 (1H, br s, H-22),  $\delta_C$  191.4 (s, C-23), 158.5 (s, C-20), 154.7 (s, C-25), 126.3 (d, C-22), 126.0 (d, C-24)] and three olefinic methyl

groups [ $\delta_H$  2.17 (3H, d,  $J=1.2$  Hz, H-27), 2.10 (1H, d,  $J=1.0$  Hz, H-21), 1.89 (1H, d,  $J=1.2$  Hz, H-26)].

Viburnudienone B<sub>2</sub> methyl ester (**2**) was also obtained as an amorphous powder,  $[\alpha]_D^{25} +17.6^\circ$ . The molecular formula was determined to be C<sub>31</sub>H<sub>46</sub>O<sub>5</sub> by HR-MS. The <sup>1</sup>H- and <sup>13</sup>C-NMR spectra of **2** were very similar to those of **1**, except for the side chain moiety. These spectral data indicated that **1** and **2** were dammarane-type triterpenoids having a 20,24-diene-23-one type side chain, respectively, and differed only at the stereochemistry of the double bond at C-20. The stereochemistry of the C-20, 22 double bond in **1** and **2** was determined to be *E* and *Z* configurations by the nuclear Overhauser effect (NOE) difference spectra (NOE: **1**; H-22/H-17, **2**; H-22/CH<sub>3</sub>-21), respectively. On the basis of the above evidence, **1** and **2** are clearly established as depicted in the formulas.

Viburnudienones H<sub>1</sub> (**3**) and H<sub>2</sub> (**4**) were isolated as an amorphous powder,  $[\alpha]_D^{25} +50.0^\circ$  and  $+44.4^\circ$ , respectively. The <sup>1</sup>H- and <sup>13</sup>C-NMR spectra of **3** and **4** were very similar to those of viburnol H,<sup>1)</sup> except for the signals due to the side chain. As regarding the side chain moieties of **3** and **4**, the

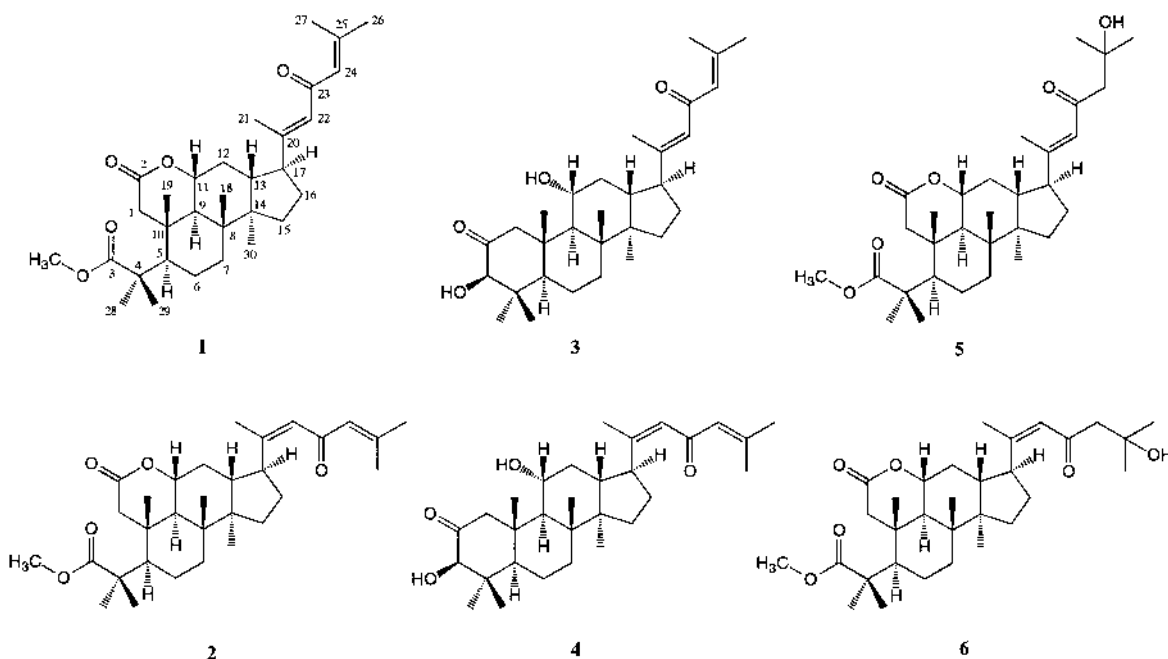


Chart 1

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Table 1.  $^{13}\text{C}$ -NMR Spectral Data (67.8 MHz,  $\text{CDCl}_3$ )

C	1	2	3	4	5	6
1	48.2	48.2	55.9	55.9	48.2	48.1
2	169.7	169.8	212.1	212.2	169.6	169.7
3	179.0	178.9	82.4	82.4	179.0	179.0
4	45.4	45.4	45.3	45.3	45.4	45.4
5	55.3	55.2	55.4	55.4	55.3	55.2
6	19.9	19.9	18.3	18.3	19.9	19.9
7	35.2	35.2	35.7	35.8	35.2	35.2
8	39.7	39.7	41.4	41.3	39.7	39.7
9	47.6	47.5	55.1	55.3	47.6	47.5
10	38.2	38.2	44.9	44.9	38.2	38.2
11	77.2	77.3	70.4	70.6	76.8	77.1
12	32.5	32.5	37.2	37.2	32.5	32.6
13	43.8	44.0	43.6	43.9	43.9	44.1
14	49.3	49.3	49.5	49.6	49.4	49.4
15	31.4	31.4	31.5	31.7	31.4	31.4
16	27.7	27.0	27.9	27.2	27.7	27.1
17	50.5	40.5	51.2	41.1	50.6	40.9
18	15.4	15.4	16.4	16.4	15.4	15.5
19	17.9	17.8	17.7	17.7	17.9	17.9
20	158.5	157.7	158.4	158.6	161.4	161.0
21	16.5	20.2	16.5	20.2	16.9	20.4
22	126.3	128.8	126.3	128.6	124.3	126.9
23	191.4	191.0	191.4	191.2	202.2	202.0
24	126.0	126.6	125.7	126.4	54.2	54.2
25	154.7	154.5	154.5	154.2	70.0	69.9
26	27.8	27.7	27.8	27.7	29.5	29.5
27	20.6	20.6	20.6	20.6	29.4	29.4
28	22.7 <sup>a)</sup>	22.7 <sup>b)</sup>	29.5	29.5	22.7 <sup>c)</sup>	22.7 <sup>d)</sup>
29	27.9 <sup>a)</sup>	27.9 <sup>b)</sup>	16.6	16.6	27.9 <sup>c)</sup>	28.0 <sup>d)</sup>
30	15.8	16.1	15.9	16.2	15.8	16.0
$\text{COOCH}_3$	52.0	52.0	—	—	52.0	52.0

a—d) Signals may be interchanged in each column.

spectral data were almost the same as those of **1** and **2**, respectively. The gross structures of **3** and **4** were proved by heteronuclear multiple bond correlation (HMBC) and nuclear Overhauser enhancement and exchange spectroscopy (NOESY) spectra. Thus, viburnudienones  $\text{H}_1$  (**3**) and  $\text{H}_2$  (**4**) are clearly established as depicted in the formulas.

Viburnenones  $\text{B}_1$  methyl ester (**5**) and  $\text{B}_2$  methyl ester (**6**) were isolated as an amorphous powder,  $[\alpha]_{\text{D}} +26.0^\circ$  and  $+12.5^\circ$ , respectively. Both of their molecular formulas were determined to be  $\text{C}_{31}\text{H}_{48}\text{O}_6$  by HR-MS. The  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectra of **5** and **6** lacked the signals due to a trisubstituted double bond at C-24, 25. Instead, they showed the characteristic signals of a hydroxy-bearing quaternary carbon [**5**:  $\delta_{\text{H}}$  4.17 (1H, s, HO-25),  $\delta_{\text{C}}$  70.0 (s, C-25), **6**:  $\delta_{\text{H}}$  4.14 (1H, s, HO-25),  $\delta_{\text{C}}$  69.9 (s, C-25)] and a methylene group linked to a carbonyl group [**5**:  $\delta_{\text{H}}$  2.60 (2H, s, H-24),  $\delta_{\text{C}}$  54.2 (t, C-24), **6**:  $\delta_{\text{H}}$  2.57 (2H, s, H-24),  $\delta_{\text{C}}$  54.2 (t, C-24)], respectively. These findings suggested that **5** and **6** possessed a 25-hydroxy-20-en-23-one type side chain, respectively. The stereochemistry of the C-20, 22 double bond in **5** and **6** was determined to be *E* and *Z* configurations by comparison of  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectra with **1** and **2**, respectively. Thus, the structures of **5** and **6** were established to be as depicted in the formulas.

#### Experimental

$^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectra were recorded with a JEOL JNM-EX 270 spectrometer (270 and 67.8 MHz, respectively, TMS as int. standard). UV spectra were recorded with a Beckman DU-64 spectrophotometer. IR spectra were recorded with a Perkin-Elmer FT-IR 1725X infrared spectrophotome-

ter. MS were recorded with a JEOL JMS-DX 303 mass spectrometer. Column chromatography was carried out on Kieselgel 60 (Merck; 230–400 mesh), Cosmosil 75C<sub>18</sub>-OPN (Nacalai Tesque). Preparative HPLC was carried out on Tosoh HPLC system using Cosmosil 5C<sub>18</sub>-AR (Nacalai Tesque, 10 mm i.d.×25 cm) and TSK gel OH-120 (Tosoh Co., 7.8 mm i.d.×30 cm) columns with UV detector.

**Extraction and Isolation** Fresh leaves of *V. dilatatum* (4.4 kg) were extracted with MeOH. The MeOH extract was concentrated under reduced pressure and the residue was suspended in a small excess of water. This suspension was successively extracted with  $\text{CHCl}_3$ ,  $\text{Et}_2\text{O}$ , AcOEt and *n*-BuOH. The  $\text{CHCl}_3$  soluble fraction was concentrated under reduced pressure to produce a residue (92.5 g). This residue was chromatographed on a silica gel column using *n*-hexane– $\text{Me}_2\text{CO}$  (4:1) and eluate was separated into four fractions (frs. 1–4). Fraction 3 was rechromatographed on a 75C<sub>18</sub> open column using MeOH– $\text{H}_2\text{O}$  (3:1) and the eluate was separated into six fractions (frs. 3-1–3-6). Fraction 3-2 was subjected to prep. HPLC [5C<sub>18</sub>-AR column; MeOH– $\text{H}_2\text{O}$  (5:1), flow rate: 1.5 ml/min, detection: 270 nm] to give compounds **1** (5.0 mg), **2** (5.0 mg), **3** (1.2 mg) and **4** (1.0 mg). Fraction 3-3 was subjected to prep. HPLC [OH-120 column; *n*-hexane–acetone (5:1), *n*-hexane–AcOEt (3:1), flow rate: 0.75, 1.5 ml/min, detection: 240 nm] to give compounds **5** (2.0 mg) and **6** (0.6 mg).

**Viburnudienone B<sub>1</sub> Methyl Ester (1)** An amorphous powder,  $[\alpha]_{\text{D}} +26.4^\circ$  ( $c=0.5$ ,  $\text{CHCl}_3$ ). EI-MS  $m/z$ : 498  $[\text{M}]^+$ . HR-MS  $m/z$ : 498.3363 ( $\text{M}^+$ , Calcd for  $\text{C}_{31}\text{H}_{46}\text{O}_5$ ; 498.3345). UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm (log  $\epsilon$ ): 269.0 (4.21). IR  $\nu$  ( $\text{CHCl}_3$ )  $\text{cm}^{-1}$ : 2954, 2880, 1719, 1668, 1621.  $^1\text{H}$ -NMR (270 MHz,  $\text{CDCl}_3$ )  $\delta$ : 6.05 (1H, m, 24-H), 6.01 (1H, br s, 22-H), 4.37 (1H, ddd,  $J=11.9, 11.4, 5.0$  Hz, 11-H), 3.70 (3H, s,  $\text{COOCH}_3$ ), 2.50 (1H, d,  $J=16.7$  Hz, 1-H $_{\beta}$ ), 2.38 (1H, m, 17-H), 2.28 (1H, d,  $J=16.7$  Hz, 1-H $_{\alpha}$ ), 2.17 (3H, d,  $J=1.2$  Hz, 27- $\text{CH}_3$ ), 2.10 (3H, d,  $J=1.0$  Hz, 21- $\text{CH}_3$ ), 1.89 (3H, d,  $J=1.2$  Hz, 26- $\text{CH}_3$ ), 1.78 (1H, d,  $J=11.9$  Hz, 9-H), 1.24, 1.22 (3H each, s, 28, 29- $\text{CH}_3$ ), 1.19 (3H, s, 19- $\text{CH}_3$ ), 1.04 (3H, s, 18- $\text{CH}_3$ ), 0.97 (3H, s, 30- $\text{CH}_3$ ).  $^{13}\text{C}$ -NMR (67.8 MHz,  $\text{CDCl}_3$ ): Table 1.

**Viburnudienone B<sub>2</sub> Methyl Ester (2)** An amorphous powder,  $[\alpha]_{\text{D}} +17.6^\circ$  ( $c=0.5$ ,  $\text{CHCl}_3$ ). EI-MS  $m/z$ : 498  $[\text{M}]^+$ . HR-MS  $m/z$ : 498.3331 ( $\text{M}^+$ , Calcd for  $\text{C}_{31}\text{H}_{46}\text{O}_5$ ; 498.3345). UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm (log  $\epsilon$ ): 269.0 (4.07). IR  $\nu$  ( $\text{CHCl}_3$ )  $\text{cm}^{-1}$ : 2954, 2880, 1719, 1668, 1620.  $^1\text{H}$ -NMR (270 MHz,  $\text{CDCl}_3$ )  $\delta$ : 6.08 (1H, t,  $J=1.2$  Hz, 22-H), 6.02 (1H, t,  $J=1.2$  Hz, 24-H), 4.33 (1H, ddd,  $J=11.8, 11.4, 5.0$  Hz, 11-H), 4.13 (1H, ddd,  $J=11.1, 10.5, 6.3$  Hz, 17-H), 3.70 (3H, s,  $\text{COOCH}_3$ ), 2.48 (1H, d,  $J=17.0$  Hz, 1-H $_{\beta}$ ), 2.25 (1H, d,  $J=17.0$  Hz, 1-H $_{\alpha}$ ), 2.14 (3H, d,  $J=1.2$  Hz, 27- $\text{CH}_3$ ), 1.88 (3H, d,  $J=1.2$  Hz, 26- $\text{CH}_3$ ), 1.81 (3H, d,  $J=1.2$  Hz, 21- $\text{CH}_3$ ), 1.77 (1H, d,  $J=11.8$  Hz, 9-H), 1.24, 1.22 (3H each, s, 28, 29- $\text{CH}_3$ ), 1.18 (3H, s, 19- $\text{CH}_3$ ), 1.03 (3H, s, 18- $\text{CH}_3$ ), 1.02 (3H, s, 30- $\text{CH}_3$ ).  $^{13}\text{C}$ -NMR (67.8 MHz,  $\text{CDCl}_3$ ): Table 1.

**Viburnudienone H<sub>1</sub> (3)** An amorphous powder,  $[\alpha]_{\text{D}} +50.0^\circ$  ( $c=0.1$ ,  $\text{CHCl}_3$ ). EI-MS  $m/z$ : 470  $[\text{M}]^+$ . HR-MS  $m/z$ : 470.3437 ( $\text{M}^+$ , Calcd for  $\text{C}_{30}\text{H}_{46}\text{O}_4$ ; 470.3396). UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm (log  $\epsilon$ ): 269.0 (4.03). IR  $\nu$  ( $\text{CHCl}_3$ )  $\text{cm}^{-1}$ : 3472, 2956, 1707, 1669, 1620.  $^1\text{H}$ -NMR (270 MHz,  $\text{CDCl}_3$ )  $\delta$ : 6.06 (1H, t,  $J=1.2$  Hz, 24-H), 6.02 (1H, br s, 22-H), 3.99 (1H, m, 11-H), 3.90 (1H, s, 3-H), 3.50 (1H, d,  $J=13.0$  Hz, 1-H $_{\beta}$ ), 2.32 (1H, m, 17-H), 2.25 (1H, d,  $J=13.0$  Hz, 1-H $_{\alpha}$ ), 2.17 (3H, d,  $J=1.2$  Hz, 27- $\text{CH}_3$ ), 2.10 (3H, d,  $J=1.2$  Hz, 21- $\text{CH}_3$ ), 1.89 (3H, d,  $J=1.2$  Hz, 26- $\text{CH}_3$ ), 1.71 (1H, d,  $J=10.4$  Hz, 9-H), 1.19 (3H, s, 28- $\text{CH}_3$ ), 1.00 (9H, s, 18, 19, 30- $\text{CH}_3$ ), 0.71 (3H, s, 29- $\text{CH}_3$ ).  $^{13}\text{C}$ -NMR (67.8 MHz,  $\text{CDCl}_3$ ): Table 1.

**Viburnudienone H<sub>2</sub> (4)** An amorphous powder,  $[\alpha]_{\text{D}} +44.4^\circ$  ( $c=0.1$ ,  $\text{CHCl}_3$ ). EI-MS  $m/z$ : 470  $[\text{M}]^+$ . HR-MS  $m/z$ : 470.3359 ( $\text{M}^+$ , Calcd for  $\text{C}_{30}\text{H}_{46}\text{O}_4$ ; 470.3396). UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm (log  $\epsilon$ ): 269.0 (3.99). IR  $\nu$  ( $\text{CHCl}_3$ )  $\text{cm}^{-1}$ : 3475, 2955, 1707, 1670, 1620.  $^1\text{H}$ -NMR (270 MHz,  $\text{CDCl}_3$ )  $\delta$ : 6.06 (1H, t,  $J=1.3$  Hz, 22-H), 6.03 (1H, d,  $J=1.3$  Hz, 24-H), 4.06 (1H, m, 17-H), 3.98 (1H, m, 11-H), 3.90 (1H, s, 3-H), 3.51 (1H, d,  $J=12.9$  Hz, 1-H $_{\beta}$ ), 2.23 (1H, d,  $J=12.9$  Hz, 1-H $_{\alpha}$ ), 2.14 (3H, d,  $J=1.3$  Hz, 27- $\text{CH}_3$ ), 1.88 (3H, d,  $J=1.3$  Hz, 26- $\text{CH}_3$ ), 1.81 (3H, d,  $J=1.3$  Hz, 21- $\text{CH}_3$ ), 1.70 (1H, d,  $J=10.6$  Hz, 9-H), 1.19 (3H, s, 28- $\text{CH}_3$ ), 1.04, 0.996, 0.989 (9H, s, 18, 19, 30- $\text{CH}_3$ ), 0.70 (3H, s, 29- $\text{CH}_3$ ).  $^{13}\text{C}$ -NMR (67.8 MHz,  $\text{CDCl}_3$ ): Table 1.

**Viburnenone B<sub>1</sub> Methyl Ester (5)** An amorphous powder,  $[\alpha]_{\text{D}} +26.1^\circ$  ( $c=0.2$ ,  $\text{CHCl}_3$ ). EI-MS  $m/z$ : 516  $[\text{M}]^+$ . HR-MS  $m/z$ : 516.3448 ( $\text{M}^+$ , Calcd for  $\text{C}_{31}\text{H}_{48}\text{O}_6$ ; 516.3450). UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm (log  $\epsilon$ ): 246.0 (4.06). IR  $\nu$  ( $\text{CHCl}_3$ )  $\text{cm}^{-1}$ : 3424, 2963, 1719, 1660, 1602.  $^1\text{H}$ -NMR (270 MHz,  $\text{CDCl}_3$ )  $\delta$ : 5.99 (1H, s, 22-H), 4.37 (1H, ddd,  $J=11.7, 11.4, 5.0$  Hz, 11-H), 4.17 (1H, s, 25-OH), 3.71 (3H, s,  $\text{COOCH}_3$ ), 2.60 (2H, s, 24-H<sub>2</sub>), 2.51 (1H, d,  $J=17.3$  Hz, 1-H $_{\beta}$ ), 2.38 (1H, m, 17-H), 2.29 (1H, d,  $J=17.3$  Hz, 1-H $_{\alpha}$ ), 2.10 (3H, s, 21- $\text{CH}_3$ ), 1.98 (1H, m, 12-H $_{\beta}$ ), 1.78 (1H, d,  $J=11.7$  Hz, 9-H), 1.26 (6H, s, 26, 27- $\text{CH}_3$ ), 1.24, 1.22 (3H each, s, 28, 29- $\text{CH}_3$ ), 1.20 (3H, s, 19- $\text{CH}_3$ ), 1.05 (3H, s, 18- $\text{CH}_3$ ), 0.97 (3H, s, 30- $\text{CH}_3$ ).  $^{13}\text{C}$ -NMR (67.8 MHz,  $\text{CDCl}_3$ ): Table 1.

**Viburnenone B<sub>2</sub> Methyl Ester (6)** An amorphous powder,  $[\alpha]_{\text{D}} +12.5^\circ$

( $c=0.06$ ,  $\text{CHCl}_3$ ). EI-MS  $m/z$ : 516  $[\text{M}]^+$ . HR-MS  $m/z$ : 516.3486 ( $\text{M}^+$ , Calcd for  $\text{C}_{31}\text{H}_{48}\text{O}_6$ ; 516.3450). UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm ( $\log \epsilon$ ): 247.0 (3.96). IR  $\nu$  ( $\text{CHCl}_3$ )  $\text{cm}^{-1}$ : 3441, 2963, 1719, 1658, 1602.  $^1\text{H-NMR}$  (270 MHz,  $\text{CDCl}_3$ )  $\delta$ : 6.06 (1H, d,  $J=1.2$  Hz, 22-H), 4.33 (1H, ddd,  $J=11.5$ , 11.3, 4.8 Hz, 11-H), 4.14 (1H, s, 25-HO), 4.06 (1H, ddd,  $J=11.2$ , 10.5, 6.7 Hz, 17-H), 3.70 (3H, s,  $\text{COOCH}_3$ ), 2.57 (2H, s, 24- $\text{H}_2$ ), 2.49 (1H, d,  $J=17.2$  Hz, 1- $\text{H}_\beta$ ), 2.26 (1H, d,  $J=17.2$  Hz, 1- $\text{H}_\alpha$ ), 1.83 (3H, d,  $J=1.2$  Hz, 21- $\text{CH}_3$ ), 1.78 (1H, d,  $J=11.5$  Hz, 9-H), 1.24 (6H, s, 26, 27- $\text{CH}_3$ ), 1.23, 1.22 (3H each, s, 28, 29- $\text{CH}_3$ ), 1.18 (3H, s, 19- $\text{CH}_3$ ), 1.04 (3H, s, 18- $\text{CH}_3$ ), 1.03 (3H, s, 30- $\text{CH}_3$ ).  $^{13}\text{C-NMR}$  (67.8 MHz,  $\text{CDCl}_3$ ): Table 1.

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