

## Diastereomeric Diterpenes from *Coleus blumei*

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**The chloroform extract of the air-dried leaves of *Coleus blumei* afforded a mixture of diastereomers of a new abietane type diterpene whose structures were elucidated by extensive one and two dimensional (1D, 2D) NMR and mass spectrometry. Acetylation of the mixture afforded a single compound. Antimicrobial tests on the diterpene indicate that it is active against *Bacillus subtilis*, *Pseudomonas aeruginosa* and *Candida albicans*.**

**Key words** *Coleus blumei*; Labiatae; abietane diterpene; antimicrobial

*Coleus blumei* is an ornamental medicinal plant found throughout the Philippines. The pounded leaves of the plant are used as cure for headaches, and for the healing of bruises. A decoction of the plant is taken internally for dyspepsia, and dropped into the eyes for ophthalmia.<sup>1)</sup> Previous studies on *C. blumei* reported the isolation of flavonoids<sup>2)</sup> and coleon O,<sup>3)</sup> an abietane type diterpenoid, from the plant. Similar abietane type diterpenes have been commonly isolated from several other *Coleus* species, and have been shown to have biological activity. Coleon O from *C. blumei* and *C. scutellarioides* was found to stimulate the rooting of mung bean cuttings by more than 400%.<sup>3)</sup> Barbatusol from *C. barbatus* induced at 3 mg/kg (i.v. rats) potent lowering of blood pressure.<sup>4)</sup> Forskolin from *C. forskohlii* is a potent inotropic, anti-hypertensive, platelet aggregation inhibitor and adenylate cyclase stimulant.<sup>5)</sup>

We now report the isolation, structural elucidation and antimicrobial test results of a new diterpene (**1**) from *C. blumei*. Analysis of the NMR data from **1** and its tetraacetate (**2**) indicate **1** is an approximately 1 : 1 mixture of diastereomers which differ only by their stereochemistry at C-15.

### Results and Discussion

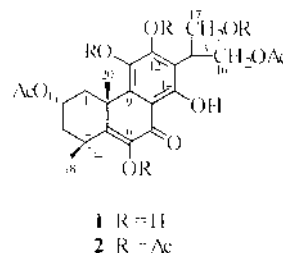
The <sup>1</sup>H-NMR (Table 1) and COSY spectral data of **1** indicated resonances for chelated hydroxyl doublet at δ 13.081 (br s) and 13.077 (br s)<sup>6)</sup> and four other hydroxyl groups at δ 6.94 (1H), 5.05 (1H, br) and two closely spaced doublets at δ 11.646 (br s) and 11.667 (br s) and δ 6.081 (s) and 6.061 (s). Methylene protons at δ 2.35 (H-3', dd, *J*=15.9, 6.6 Hz) and δ 1.70 (H-3, dd, *J*=15.9, 2.7 Hz) were coupled to the methine proton at δ 5.41 (H-2, dq), which was in turn coupled to the methylene hydrogens at δ 3.86 (H-1', m) and δ 1.45 (H-1, m). The methylene proton at δ 3.95 (H-16, m) was geminally coupled to the hydrogen at δ 4.79 (H-16', m), which was in turn coupled to the proton at δ 3.90 (H-15, m). The latter proton was coupled to the methylene proton at δ 3.99 (H-17', m) which was geminally coupled to the hydrogen at δ 3.92 (H-17, m). Three methyl groups were indicated by the resonances at δ 1.47 (3H, s), 1.53 (3H, s) and the closely spaced methyl doublet at δ 1.632 (s) and 1.622 (s). The relatively deshielded methyl resonances indicated that they are neighbors to oxygenated groups and olefins. The resonances at δ 2.01 (3H, s) and the closely spaced doublet at δ 2.184 (s) and 2.179 (s) were attributed to two acetate methyls. The signals at δ 1.632, 1.622, 2.184, 2.179, 6.081, 6.061, 11.667, 11.646, 13.077 and 13.081 appeared “dou-

plets” in the 300 MHz spectrum, suggesting the possibility of a mixture of diastereomers. However in no case did the chemical shift difference in the “doublets” exceed 0.02 ppm.

The <sup>13</sup>C- and DEPT NMR spectral data of **1** (Table 1) indicated 24 carbon resonances with the following functionalities: the carbonyl of a conjugated ketone at δ 182.7; the carbonyls of two acetates at δ 170.7 and 173.8; five methyl groups at δ 28.5, 29.5, 21.0, 21.4 and closely spaced doublet at δ 24.69 and 24.75; two methylenes at δ 42.4 and the closely spaced doublet at δ 34.02 and 34.07 and two oxygenated methylenes at δ 61.3 and closely spaced doublet at δ 61.69 and 61.62; one methine which appeared as closely spaced doublet at δ 36.62 and 36.57 and one oxygenated methine (δ 68.2); two aliphatic quaternary carbons (δ 40.4, 35.9); and eight olefinic quaternary carbons (δ 105–152). The deshielded resonances (δ 141–152) were assigned to oxygenated carbons, while the shielded resonances (δ 105–133.5) were attributed to non-oxygenated carbons, except that the deshielded resonance at δ 142.1 was assigned to C-5. The deshielding was attributed to the steric effect on C-5 of neighboring groups. This assignment was confirmed by gHMBC. As with the proton NMR, a few of the carbon signals (those at δ 24.69, 24.75, 34.07, 34.02, 36.57, 36.62, 61.62, 61.69) appeared as “doublets” in the 75 MHz <sup>13</sup>C spectra, but with less than 0.07 ppm difference between the signals.

The high resolution mass spectrum (HR-MS) of **1** indicated a molecular ion at *m/z* 478.1850 corresponding to a molecular formula of C<sub>24</sub>H<sub>30</sub>O<sub>10</sub>. Based on the molecular formula, the index of hydrogen deficiency was ten. With four double bonds and three carbonyl groups in **1**, the rest of the hydrogen deficiency was accounted for by a tricyclic system.

The <sup>1</sup>H and <sup>13</sup>C assignments for **1** (Table 1) were verified by gHMQC and connectivity was verified by an inverse long-range heteronuclear experiment gHMBC optimized for *J*=10 Hz (Table 1). The carbon at δ 142.1 (C-5) was long-



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Table 1.  $^1\text{H}$ -,  $^{13}\text{C}$ -NMR and HMBC Spectral Data of **1** in  $\text{CDCl}_3$ 

Carbon No.	$^{13}\text{C}$ , $\delta$	$^1\text{H}$ , $\delta$	HMBC
C-1	34.07 <sup>a)</sup> , 34.02 <sup>a)</sup>	1.45 (1H, m, H-1), 3.86 (1H, m, H-1')	H-3, H-20
C-2	68.2	5.41 (1H, dq, $J=6.5, 2.3$ , H-2)	H-1'
C-3	42.4	1.70 (1H, dd, $J=15.9, 2.7$ Hz, H-3), 2.35 (1H, dd, $J=15.9, 6.6$ Hz, H-3')	H-1
C-4	35.9	—	H-2, H-3, H-3', H-18, H-19
C-5	142.1	—	H-1, H-18, H-19, H-20
C-6	141.0	—	6-OH
C-7	182.7	—	6-OH
C-8	105.4	—	14-OH
C-9	133.5	—	H-20, 11-OH
C-10	40.4	—	H-1, H-1', H-20
C-11	135.8	—	11-OH
C-12	150.5	—	11-OH
C-13	109.8	—	14-OH
C-14	152.3	—	14-OH
C-15	36.57 <sup>a)</sup> , 36.62 <sup>a)</sup>	3.90 (1H, m)	—
C-16 <sup>c)</sup>	61.3	3.95 (1H, m, H-16), 4.79 (1H, m, H-16')	
C-17 <sup>c)</sup>	61.62 <sup>a)</sup> , 61.69 <sup>a)</sup>	3.92 (1H, m, H-17), 3.99 (1H, m, H-17')	H-16'
C-18	28.5	1.47 (3H, s, H-18)	H-3, H-19
C-19	29.5	1.53 (3H, s, H-19)	H-3, H-18
C-20	24.69 <sup>a)</sup> , 24.75 <sup>a)</sup>	1.632 <sup>b)</sup> (3H, s, H-20), 1.622 <sup>b)</sup> (3H, s, H-20)	
OAc (C-2)	170.7 21.0	— 2.01 (3H, s)	2-CO <sub>2</sub> CH <sub>3</sub>
OAc (C-16)	173.8 21.4	— 2.184 <sup>b)</sup> (3H, s), 2.179 <sup>b)</sup> (3H, s)	16-CO <sub>2</sub> CH <sub>3</sub> , H-16
OH (C-6)	—	6.94 (1H)	
OH (C-11)	—	6.081 <sup>b)</sup> (1H), 6.061 <sup>b)</sup> (1H)	
OH (C-12)	—	11.667 <sup>b)</sup> (1H, br s), 11.646 <sup>b)</sup> (1H, br s)	
OH (C-14)	—	13.081 <sup>b)</sup> (1H, br s), 13.077 <sup>b)</sup> (1H, br s)	
OH (C-17)	—	5.05 (1H, br s)	

a) Carbon doublets due to diastereomerism. b) Proton doublets due to diastereomerism. c) Maybe interchanged in the diastereomers.

range correlated to the methyl protons at  $\delta$  1.47 (H-18), 1.53 (H-19) and 1.632 and 1.622 (H-20). This confirms the assignment of the resonance at  $\delta$  142.1 to C-5. The carbonyl carbon at  $\delta$  173.8 was long-range correlated to the methylene proton at  $\delta$  4.79 (H-16) and the acetate methyl at  $\delta$  2.184 and 2.179. This confirms the attachment to C-16 of the acetate carbonyl at  $\delta$  173.8. The positions of the aromatic hydroxyls were determined by gHMBC. The hydroxyl proton at  $\delta$  6.94 (6-OH) was long-range correlated to the carbons at  $\delta$  141.0 (C-6) and 182.7 (C-7), while the hydroxyl proton at  $\delta$  6.081 and 6.061 (11-OH) was long-range correlated to the carbons at  $\delta$  135.8 (C-11), 150.5 (C-12) and 133.5 (C-9). The hydroxyl proton at  $\delta$  13.081 and 13.077 (14-OH) was long-range correlated to the carbons at  $\delta$  105.4 (C-8), 109.8 (C-13) and 152.3 (C-14).

Although most coleons have an oxygenated carbon at C-3 rather than C-2 as proposed for **1**, C-2 oxygenation is known, for example xanthanthusin A,<sup>7)</sup> which was isolated from *Coleus xanthanthus*.

The relative stereochemistry of **1** in ring A was determined by NOESY which indicated correlation through space of the  $^1\text{H}$  nuclei in the molecule (Table 2). Thus, the methine proton at  $\delta$  5.41 (H-2) shows correlation in space with the methyl protons at  $\delta$  1.632 and 1.622 (H-20), indicating that H-2 is in the axial position. This makes the acetate attached to C-2 in the equatorial position. The methyl protons at  $\delta$

Table 2. NOESY Correlation Data of **1** in  $\text{CDCl}_3$ 

$^1\text{H}$	NOESY correlation
H-1'	H-1, H-2
H-2	H-1', H-3', H-20
H-3'	H-2, H-3, H-18, H-20
H-18	H-3'
H-16	H-16'

1.632 and 1.622 (H-20) are close in space to the methyl protons at  $\delta$  1.47 (H-18). Thus, these methyl groups are in the axial positions and the remaining methyl group at  $\delta$  1.53 is in the equatorial position.

As noted above, the observation that several of the  $^1\text{H}$  and  $^{13}\text{C}$  signals appeared as closely spaced "doublets" rather than the expected singlets suggested that **1** was an approximate 1:1 mixture of diastereomers, rather than a single compound. Of the three chiral centres in **1**, it did not appear reasonable that inversion of stereochemistry at either C-2 or C-10 would produce diastereomeric compounds which differed in  $^{13}\text{C}$  chemical shift by less than 0.07 ppm. We therefore propose that **1** is approximately a 1:1 mixture of the two diastereomers formed by inversion of the stereochemistry at C-15. Such a mixture is likely if the acetylating enzyme cannot distinguish the diastereotopic hydroxyls at C-16 and C-17.

Table 3. Antimicrobial Test Results on **1**

Sample	Concn. ( $\mu\text{g}$ )	<i>Staphylococcus aureus</i>		<i>Escherichia coli</i>		<i>Pseudomonas aeruginosa</i>		<i>Bacillus subtilis</i>		<i>Candida albicans</i>		<i>Aspergillus niger</i>		<i>Trichophyton mentagrophytes</i>	
		C.Z. (mm)	A.I.	C.Z. (mm)	A.I.	C.Z. (mm)	A.I.	C.Z. (mm)	A.I.	C.Z. (mm)	A.I.	C.Z. (mm)	A.I.	C.Z.	A.I.
<b>1</b>	60	11, 11	0.1	—	0	13, 13	0.3	14, 14	0.4	15, 15	0.5	12, 12	0.2	—	0
Standard		20	1.0	25	1.5	25	1.5	20	1.0	30	2.0	30	2.0	30	2.0
Antibiotic		Chloramphenicol		Tetracycline		Tetracycline		Chloramphenicol		Clotrimazole		Clotrimazole		Clotrimazole	

C.Z.: clearing zone, A.I.: antimicrobial index.

Given that **1** is almost planar, except for the area around the C-1 and C-2 carbons, this would not be surprising. Hydrogen-bonding interactions from the C-16/C-17 oxygens through the C-11 and C-12 hydroxyls also provides a pathway by which the difference in C-15 stereochemistry can be "transmitted" to C-1 and C-20.

To test this proposal, **1** was acetylated to the tetraacetate (**2**) in acetic anhydride/pyridine overnight, thereby removing the C-15 stereocentre.  $^1\text{H}$  and  $^{13}\text{C}$  spectra of **2** showed considerable simplification relative to **1**, and no appearance of "doublets" in either  $^1\text{H}$ - or  $^{13}\text{C}$ -NMR. We therefore conclude that **1** is diastereomeric at C-15.

Compound **1** was tested for its antimicrobial potential, and the results of the study are presented in Table 3. Among the seven microorganisms tested, **1** was found to be active against *Bacillus subtilis*, *Pseudomonas aeruginosa* and *Candida albicans* with an average antimicrobial index of 0.4, 0.3 and 0.5, respectively, at a concentration of 60  $\mu\text{g}$ . For the antibiotics chloramphenicol for *B. subtilis*, tetracycline for *P. aeruginosa* and clotrimazole for *C. albicans*, the average antimicrobial index are 1.0, 1.5 and 2.0, respectively, at a concentration of 30  $\mu\text{g}$ . Thus, **1** has a low antimicrobial activity.

## Experimental

**General Experimental Procedures** IR spectra were recorded on a Perkin-Elmer 1600 Fourier Transform IR spectrometer and UV spectra on a HP 8452A diode array spectrometer. NMR spectra were recorded on a Bruker AMX Fourier Transform 300 in  $\text{CDCl}_3$  at 300 MHz for  $^1\text{H}$  and 75 MHz for  $^{13}\text{C}$ . The high- and low-resolution EI-MS were recorded on a Micromass AutoSpec mass spectrometer. Column chromatography was performed with Silica gel 60 (70–230 mesh); TLC was performed with plastic backed plates coated with Silica gel  $\text{F}_{254}$  TLC; plates were visualized by spraying with vanillin– $\text{H}_2\text{SO}_4$  and warming.

**Plant Material** *Coleus blumei* was collected from Ilo-ilo, Philippines in November 1996. It was identified as *Coleus blumei* BENTH (Labiatae) at the Philippine National Museum, and a voucher specimen DLSUCD #028 is kept at the Chemistry Department of De La Salle University.

**Extraction and Isolation** Air-dried leaves of *Coleus blumei* (600 g) were ground in an osterizer, then extracted with  $\text{CHCl}_3$  (2.5 l) at room temperature for 2 d. The mixture was filtered and the filtrate concentrated *in vacuo* to afford a crude extract (40 g). This extract was dissolved in EtOH (750 ml), then placed in an ice bath. To the solution was added 4% aqueous  $\text{Pb}(\text{OAc})_2$  to precipitate the more polar components.<sup>8)</sup> The mixture was then filtered and the filtrate concentrated *in vacuo* until a mixture of water and oily residue remained. The concentrate was extracted with  $\text{CHCl}_3$ , and the extract was dried with anhydrous  $\text{Na}_2\text{SO}_4$ , then filtered. The filtrate was concentrated *in vacuo* to afford the treated extract (11.7 g), which was fractionated in a column of silica gel (70–230 mesh) using increasing proportions of  $\text{Me}_2\text{CO}$  in  $\text{CHCl}_3$  (10% increments, 100 ml) as eluents. The  $\text{CHCl}_3$  fraction was rechromatographed in 10%  $\text{Me}_2\text{CO}$  in  $\text{CHCl}_3$  (2 $\times$ ), followed by rechromatography in  $\text{CH}_2\text{Cl}_2$ : $\text{CH}_3\text{CN}$ : $\text{Et}_2\text{O}$  (18:1:1) to afford **1** (yellow-

ish oil, 15 mg). IR  $\nu_{\text{max}}$  ( $\text{CHCl}_3$ )  $\text{cm}^{-1}$ : 3394 (OH), 2957, 2957, 2928, 2858, 1735 (C=O), 1623 (C=O), 1598 (C=C), 1463, 1420, 1386, 1369, 1327, 1287 (C–O–C), 1268 (C–O–C), 1251 (C–O–C), 1178 (C–O), 1162 (C–O), 1145 (C–O), 1038 (C–O), 968, 757; UV  $\lambda_{\text{max}}$  ( $\text{CHCl}_3$ ) nm: 271, 294 (sh), 324, 391;  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR data are listed in Table 1; EI-MS  $m/z$  478 [ $\text{M}^+$ ] (27), 403 (15), 374 (14), 358 (32), 344 (35), 343 (100); HR-EI-MS  $m/z$  478.1850 [ $\text{M}^+$ ] ( $\text{C}_{24}\text{H}_{30}\text{O}_{10}$ ).

**Acetylation** **1** (2 mg) was acetylated in acetic anhydride/pyridine overnight to afford the tetraacetate **2** (oil, 1 mg).  $^1\text{H}$ -NMR ( $\text{CDCl}_3$ ):  $\delta$  1.26 (3H, s), 1.58 (2 $\times$ 3H, s), 1.66 (1H, m, H1), 1.78 (1H, m, H3), 1.99 (3H, s), 2.00 (3H, s), 2.02 (3H, s) 2.29 (1H, m, H3), 2.34 (3H, s), 2.36 (3H, s), 2.37 (1H, m, H1) 2.39 (3H, s), 3.68 (1H, m, H15), 4.36–4.58 (4H, m, H16, H17), 5.34 (1H, m, H2), 13.5 (1H, br s, OH).  $^{13}\text{C}$ -NMR ( $\text{CDCl}_3$ ):  $\delta$  20.3, 20.8, 20.9, 21.3, 22.7, 29.3, 29.3, 31.2, 31.9, 36.5, 42.1, 42.5, 62.8, 62.8, 67.0, 115.6, 119.2, 148.7, 160.6, 167.2, 168.5, 170.6, 171.0, 185.6. HR-EI-MS  $m/z$  646.2256 (Calcd for  $\text{C}_{32}\text{H}_{38}\text{O}_{14}$ , 646.2262).

**Antimicrobial Test** Microbial suspension containing approximately  $6 \times 10^8$  cells/ml was prepared for each test organism for 24-h culture of *S. aureus*, *E. coli*, *P. aeruginosa*, *B. subtilis* and *C. albicans* and from a 5 d old *A. niger* and *T. mentagrophytes*. The suspending medium used for each microbial suspension was 0.1% peptone water.

One-tenth milliliter of *S. aureus*, *E. coli*, *P. aeruginosa*, *B. subtilis* and *C. albicans* were transferred into pre-poured nutrient agar (NA) plates. While *A. niger* and *T. mentagrophytes* were transferred into pre-poured potato dextrose agar (PDA) plates. About 5 ml of corresponding medium, melted and cooled to about 45  $^\circ\text{C}$  was poured into the plate. The plate was swirled to distribute the microbial cells evenly on the plate and the agar overlay was allowed to solidify. One-centimeter wells were cut from equidistant points of the seeded agar plates using sterile cork borer. Sixty micrograms of samples and 30  $\mu\text{g}$  of standard agent was used. Antimicrobial index (AI) was computed by subtracting the diameter of the well from the clearing zone divided by the diameter of the well.

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