Structure of Three New Carotenoids with a 3-Methoxy-5-keto-5,6-seco-4,6-cyclo- β End Group from the Seeds of *Pittosporum tobira*

Yasuhiro Fujiwara,* Hisashi Maruwaka, Fujio Toki, Keiji Hashimoto, and Takashi Maoka

Kyoto Pharmaceutical University, Yamashina-ku, Kyoto 607–8414, Japan. Received February 23, 2001; accepted April 13, 2001

Three new carotenoids with a 3-methoxy-5-keto-5,6-seco-4,6-cyclo- β end group (1—3) have been isolated from the seeds of *Pittosporum tobira*. Their structures were elucidated by detailed analyses of nuclear magnetic resonance and UV data.

Key words carotenoid; *Pittosporum tobira*; 3-methoxy-carotenoid; structure elucidation; NMR; ¹H-¹H nuclear Overhauser effect

In the course of studies on new carotenoids in natural products, we have reported the isolation and structure elucidation of the oyster carotenoids with unique end groups: crassostreaxanthins^{1*a*,*b*}; the retrocarotenoid anhydroeschscholtz-xanthin^{1*c*}; the di-*Z*-carotenoids cucumariaxanthins^{1*d*}; and the purple carotenoid rhodobacterioxanthin.^{1*e*} In a previous paper, we reported the isolation and structure elucidation of the C₆₉ carotenoids pittosporumxanthins A1 and A2 from the seeds of *Pittosporum tobira* of Pittosporaceae.²⁾ Recently, we have isolated three new carotenoids (1—3) with the new 3-methoxy-5-keto-5,6-seco-4,6-cyclo- β end group (3-methoxy-5-acetyl-4,18-dinor- β end group³⁾) from the red seeds of *P tobira* collected from the bank of the Kamogawa River, Kyoto, in December 1998. This paper reports the isolation and structural elucidation of the new carotenoids.

Results and Discussions

The seeds (20 kg) of *P. tobira* were extracted with methanol. The methanol extract was saponified with KOH– methanol and extracted with Et_2O –*n*-hexane. The extracted solution was washed with H_2O , dried, and evaporated under reduced pressure. The residual crude carotenoids were isolated by column chromatography on silica gel with increasing percentages of ether in *n*-hexane. The ether fractions purified by high-performance liquid chromatography (HPLC) on silica gel and octadecyl silica (ODS) furnished three carotenoids (1, 10 mg; 2, 2 mg; 3, 1 mg).

Carotenoids 1—3 were all obtained as a red amorphous powder. Acetylation of 1—3 with acetic anhydride in pyridine gave the mono-acetate of 1—3, respectively. The molecular formulae of 1—3 were established to be $C_{41}H_{56}O_4$ using high-resolution electron-impact mass spectrometry (HR-EI-MS).

The ¹H-NMR spectra of **1**—**3** in CDCl_3 showed 11 singlet methyl signals, respectively. Among those, five methyl signals had common chemical shifts of δ 1.29, 1.42, 1.99, 2.31, and 3.35 in **1**—**3**. The ¹H-NMR data suggested that the carotenoids **1**—**3** have a common structural unit of a hitherto unknown type in the molecules. Thus the structural analysis of the chief product **1** was undertaken in detail.

Carotenoid **1** in Et₂O showed an absorption maximum at 457 nm, suggesting the existence of a conjugated decaene structural chromophore.⁴⁾ Both the ¹³C- and the ¹H-NMR signals of **1** in CDCl₃ were assigned by distortionless enhancement by polarization transfer (DEPT), ¹³C–¹H correlation

spectroscopy (COSY), double-quantum filtered (DQF)-COSY, nuclear Overhauser effect (NOE), and ¹H-homodecoupling (including decoupling difference) experiments, and by comparison with those of related compounds.^{5,6)} Their assigned data are shown in Table 1. The ¹³C-NMR and DEPT experiments with **1** in CDCl₃ established the presence of 41 carbons and 55 carbon-bonded protons (11 methyls, 3 methylenes, 16 methines). The ¹³C-¹H COSY spectra of **1** established all the one-bond ¹³C-¹H connectivities. The chemical shifts of carbons and protons of the half-partial structure (C1' to C 20') in **1** were identical to those of (3*S*,5*R*,8*R*)-auroxanthin.^{5a)}

The structure elucidation of the unassigned remaining half (C1 to C20 and a methoxyl group) in 1 was as follows. In the DQF-COSY spectra of 1, the cross-peaks of ¹H spin-couplings were observed between following proton-proton pairs: H2–H3, H7–H8, H10–H11–H12, and H14–H15. The magnitude of their spin-coupling constants is shown in Table 1. In the low-power ¹H-homodecoupling difference experiments with 1, irradiation of the methyl proton signals at δ 1.97 (H20) and 1.99 (H19) revealed the existence of long-range couplings between H20–H14 and between H19–H10. The NOE results in ¹H–¹H NOE difference experiments with 1 are summarized in Fig. 2.

As can be seen from Fig. 2, NOEs were observed between H20–H15, H11 and between H19–H11, H7 in the protons



(9Z)-Violaxanthin

Fig. 1. Structures of 1-3, and (9Z)-Violaxanthin

Carbon No.	1		2	3	(9Z)-Violaxanthin
	${\pmb \delta}_{\rm C}{}^{a)}$ mult.	$\delta_{\mathrm{H}}^{\ b)}$ mult. (<i>J</i> , Hz)	$\delta_{ m H}^{\ \ b)}$ mult. (<i>J</i> , Hz)	$\delta_{\mathrm{H}}^{\ b)}$ mult. (<i>J</i> , Hz)	$\delta_{ m H}$ mult. (<i>J</i> , Hz)
1	46.5 s	_	_	_	_
2	45.8 t	1.83 dd (13.5, 4)	1.84 dd (13.5, 4)	1.83 dd (13.5, 4)	1.26 dd (14.5, 10)
2	8404	1.9/dd(13.5,7)	1.98 dd (13.5, 7)	1.9/dd(13.5,7)	1.63 m
3	84.0 d	4.58 dd (7, 4)	4.58 dd (7, 4)	4.58 dd (7, 4)	3.92 m
4	130.88	—	—	—	2.41 ddd (14.5, 5.3, 1.5)
5	200.0 s	—	—	—	—
6	159.7 s	—	—	—	—
7	121.3 d	7.16 d (16)	7.17 d (16)	7.17 d (16)	5.93 d (15.5)
8	142.0 d	6.86 d (16)	6.86 d (16)	6.86 d (16)	6.84 d (15.5)
9	135.9 s	—	_		—
10	136.0 d	6.33 d (11)	6.33 d (11)	6.33 d (11)	6.08 d (11.5)
11	124.6 d	6.61 dd (15, 11)	6.63 dd (15, 11)	6.63 dd (15, 11)	6.76 dd (15, 11.5)
12	139.7 d	6.43 d (15)	6.43 d (15)	6.43 d (15)	6.29 d (15)
13	136.0 s	—	—	_	—
14	133.9 d	6.27 dm (10)	<i>ca</i> . 6.30 m	<i>ca</i> . 6.30 m	6.25 dm (12)
15	130.8 d	6.62 m	<i>ca</i> . 6.65 m	<i>ca</i> . 6.65 m	<i>ca.</i> 6.63 m
16	29.5 q	1.29 s	1.29 s	1.29 s	1.01 s
17	29.0 q	1.42 s	1.42 s	1.42 s	1.17 s
18	30.0 q	2.31 s	2.31 s	2.31 s	1.21 s
19	12.5 q	1.99 s	1.99 s	1.99 s	1.93 s
20	12.8 q	1.97 s	1.97 s	1.97 s	1.96 s
O <u>CH</u> ₃	56.3 q	3.35 s	3.35 s	3.35 s	—
1'	33.7 s	—	_		—
2'	46.7 t	1.51 dd (14, 3.5)	1.27 dd (14.5, 10)	1.24 dd (14.5,10)	1.24 dd (14.5, 10)
		1.77 dd (14, 4)	1.63 ddd (14.5, 3.5, 1.5)	1.63 ddd (14.5, 3.5, 1.5)	1.63 ddd (14.5, 3.5, 1.5)
3'	67.7 d	4,23 m	3.92 m	3.90 m	3.91 m
4'	47.4 t	1.99 dd (13.5, 4)	1.65 dd (14.5, 9)	1.63 dd (14.5,9)	1.63 dd (14.5, 9)
		2.13 dd (13.5, 4)	2.41 ddd (14.5, 5.5, 1.5)	2.38 ddd (14.5, 5, 1.5)	2.39 ddd (14.5, 5, 1.5)
5'	86.9 s	—	—	_	—
6'	154.1 s	—	—	_	—
7'	119.9 d	5.26 d (1)	5.94 d (15.5)	5.88 d (15.5)	5.88 d (15.5)
8'	87.7 d	5.17 d (1)	6.84 d (15.5)	6.29 d (15.5)	6.29 d (15.5)
9'	138.2 s	—	—	—	—
10'	127.2 d	6.20 d (11)	6.07 d (11.5)	6.19 d (11.5)	6.20 d (11.5)
11'	124.7 d	6.51 dd (15, 11)	6.76 dd (15, 11.5)	6.62 dd (15, 11.5)	6.60 dd (15, 11.5)
12'	137.5 d	6.32 d (15)	6.30 d (15)	6.38 d (15)	6.37 d (15)
13'	136.3 s	—	—	_	_
14'	132.2 d	6.23 dm (12)	6.25 dm (12)	6.25 dm (12)	6.26 dm (12)
15'	129.8 d	6.62 m	<i>ca</i> . 6.65 m	<i>ca</i> . 6.65 m	<i>ca</i> . 6.63 m
16'	31.4 q	1.17 s	1.01 s	0.98 s	0.98 s
17'	28.9 q	1.33 s	1.17 s	1.15 s	1.15 s
18'	29.0 q	1.62 s	1.22 s	1.19 s	1.19 s
19'	12.7 q	1.72 s	1.93 s	1.93 s	1.93 s
20'	12.9 q	1.95 s	1.97 s	1.97 s	1.97 s

Table 1. NMR Data of Carotenoids 1-3, and (9Z)-Violaxanthin^{6a)} in CDCl₃

 1 H-(300 MHz) and 13 C-NMR (75 MHz) chemical shifts are reported downfield from internal tetramethylsilane (TMS) (=0.00). *a*) Assignments are based on DEPT and 13 C- 1 H COSY, and on comparison with related carotenoids.⁵ *b*) Assignments are based on DQF-COSY, 1 H- 1 H NOE, 1 H-homodecoupling difference experiments, and comparison with those of the related carotenoids.^{5.6}

from H7 to H15, H19, and H20. Based on the NMR experimental results mentioned above, the connections of C7 to C15, C19, and C20, and all-*E* geometries of Δ^7 , Δ^9 , Δ^{11} , and Δ^{13} were clarified. The connections of =C7-C6(=)-C1(C16,17)-C2-C3(O-)- in 1 were deduced on the basis of the observed NOEs between H7–H16, H17, and between H2–H17, and the COSY correlations between H2–H3. In addition, the carbon (C4) was connected to the C6 (ene structure) and acetyl group to satisfy the conjugated decaene structure in 1 estimated by the UV–visible data. The remaining connections of C3H (at $\delta_{\rm H} 4.58$)–C4 and C3–OCH₃ (at $\delta_{\rm H} 3.35$) in 1 were based on the NOEs between acetyl-methyl protons ($\delta_{\rm H} 2.31$) and H3, and between methoxy-methyl protons and H2, H3, respectively. Then the assigned structure of



Fig. 2. Structure and NOE Data Summary of 1

the end group and the geometry of $\Delta^{5,6}$ in 1 were straightforward for their ¹H- and ¹³C-NMR data, and NOE data. Therefore the unique structure of the carotenoid 1 was assigned to be 3-methoxy-3'-hydroxy-5',8'-epoxy-5',8'-dihyodro-5,6seco-4,6-cyclo- β , β -caroten-5-one as shown in Fig. 1.

Carotenoids 2 and 3 in Et₂O showed absorption maxima at

467 and 470 nm, respectively, suggesting the presence of a conjugated hendecaene structure in the molecule.⁴⁾ The ¹H-NMR data of 2 and 3 in CDCl₃ suggested the existence of (9Z)- and (all-E)-violaxanthin partial structures in addition to the partial structure found in 1, respectively. Thus the NMR data of 2 and 3 were compared with those of (9Z)- and (all-E)-violaxanthin, respectively.^{6a)} The structure and ¹H-NMR data of (9Z)-violaxanthin are shown in Fig. 1 and Table 1, respectively. The ¹H data of H2,2' to H20,20' in (all-E)-violaxanthin are nearly the same as those of H2' to H20' in (9Z)-violaxanthin.⁶⁾ As shown in Table 1, the ¹H chemical shifts and their spin-couplings of H2' to H20' in 2 and 3 were almost identical with those of H2 to H20 and of H2' to H20' in (9Z)-violaxanthin, respectively. That is, the data indicated the inclusion of the (9Z)- and (all-E)-violaxanthin partial structure in the molecule of 2 and 3. Thus the structures of 2 and 3 were determined to be (9'Z)- and (all-E)-3-methoxy-3'hydroxy-5',6'-epoxy-5',6'-dihyodro-5,6-seco-4,6-cyclo- β , β caroten-5-one, respectively, as shown in Fig. 1.

The elucidated structures of 1, 2, and 3 are unique 3methoxy carotenoids that have not been reported previously in natural products to our knowledge. Therefore these are the first examples of carotenoids including the 3-methoxy-5keto-5,6-seco-4,6-cyclo- β end group.

Experimental

General Experimental Procedures The UV-visible spectra were recorded on a Shimadzu UV-240 spectrophotometer in Et₂O. The EI-MS spectra were recorded using a JEOL JMS-SX 102AQQ mass spectrometer with ionization energy of 20 eV. The ¹³C- (75.4 MHz) and ¹H- (300 MHz) NMR spectra were recorded on a Varian XL-300 NMR spectrometer in CDCl₃ with TMS as an internal standard. Two-dimensional (2D) experiments were acquired using the standard Varian pulse sequence programs and the software used to obtain 2D spectra was from Varian, version 6.1E or A. Circular dichroism (CD) spectra were recorded in Et₂O at room temperature with a JASCO J-500 spectropolarimeter. HPLC was performed on a Shimadzu LC-6AD instrument with a Shimadzu SPD-6AV spectrophotometer set at 450 nm. The columns used were a Shim-Pack PREP-SIL (Shimadzu, $20\,\text{mm} \times 250\,\text{mm}$, $5\,\mu\text{m}$) and a Lichrospher 100 RP-18 (Cica Merck, $20 \times$ 250 mm, $10 \mu \text{m}$). Analytical thin-layer chromatography (TLC) was carried out on Silica gel 60 F254 plates (Merck) with the developing solvent of 30% acetone in *n*-hexane.

Plant Material The red colored seeds of *P. tobira* AITON (Pittosporaceae) were collected in December 1998, on the banks of the Kamogawa River, Kyoto, Japan. Voucher specimens have been deposited at Kyoto Pharmaceutical University.

Extraction and Isolation of Carotenoids The seeds (20 kg) of *P. tobira* were washed with *n*-hexane and extracted with methanol. The methanol extract was transferred to Et_2O –*n*-hexane (1 : 1). The organic layer was washed with H₂O, dried, and evaporated under reduced pressure. The residue was saponified with 5% KOH–methanol for 12 h at 30 °C and extracted with Et_2O –*n*-hexane (1 : 1). The extracted solution was washed with H₂O, dried, and evaporated under reduced pressure. The residue carotenoids (1.0 g) were separated by column chromatography on silica gel with increasing percentages of ether in *n*-hexane. The ether fractions further separated by HPLC on silica gel with acetone–*n*-hexane (2 : 8) and purified by HPLC on ODS with CHCl₃–CH₃CN (2 : 8) furnished carotenoids **1** (10 mg), **2** (2 mg),

and 3 (1 mg).

In addition to the new carotenoids **1**, **2**, and **3**, five known carotenoids were isolated and identified by UV–visible, EI-MS, ¹H-NMR, and CD spectral data^{4–7}: antheraxanthin (10 mg); (9*Z*)-violaxanthin (30 mg); violaxanthin (10 mg); (3*S*,5*R*,8*R*)-auroxanthin (20 mg); neoxanthin (20 mg).

Carotenoid 1: Red amorphous powder. TLC: *Rf* 0.45. HPLC: retention time, min: 13 (silica gel, acetone/*n*-hexane=2/8, 10 ml/min). UV–visible λ_{max} (Et₂O) nm: 457. ¹³C- and ¹H-NMR (CDCl₃): Table 1. HR-EI-MS *m/z*,: 612.4178 (M⁺) (Calcd for C₄₁H₅₆O₄: 612.4177).

Carotenoid **2**: Red amorphous powder. TLC: *Rf* 0.45. HPLC: retention time, min: 11 (silica gel, acetone/*n*-hexane=2/8, 10 ml/min). UV–visible λ_{max} (Et₂O) nm: 467. ¹H-NMR (CDCl₃): Table 1. HR-EI-MS *m/z*: 612.4164 (M⁺) (Calcd for C₄₁H₅₆O₄: 612. 4177).

Carotenoid 3: Red amorphous powder. TLC: *Rf* 0.45. HPLC: retention time, min: 12 (silica gel, acetone/*n*-hexane=2/8, 10 ml/min). UV–visible λ_{max} (Et₂O) nm: 470. ¹H-NMR (CDCl₃): Table 1. HR-EI-MS *m*/*z*,: 612.4170 (M⁺) (Calcd for C₄₁H₅₆O₄: 612.4177).

Acetylation of 1—3 To the carotenoid 1 (*ca.* 50 μ g, TLC: *Rf* 0.45) dissolved in dry piridine 1 ml was added acetic anhydride 0.5 ml. The reaction was carried out at room temperature, and monitored by TLC at regular intervals. After 1 h, a less polar product (monoacetate, UV–visible λ_{max} (Et₂O) 457 nm) with *Rf* 0.65 was formed by acetylation.

Acetylation of 2 and 3 was carried out in a similar manner to give less polar products (TLC: Rf 0.65 from 2 and Rf 0.65 from 3), respectively.

As a comparative experiment, acetylation reaction of violaxanthin (TLC: Rf 0.3) was performed. The reaction was run in a similar manner to that above. Two less polar acetylation products (TLC: Rf 0.5, monoacetate and Rf 0.7, diacetate) were given.

Acknowledgments The authors are grateful to Miss K. Oda of Kyoto Pharmaceutical University for MS measurements.

References

- a) Fujiwara Y., Maoka T., Ookubo M., Matsuno T., *Tetrahedron Lett.*, 33, 4941—4944 (1992); b) Maoka T., Hashimoto K., Akimoto N., Fujiwara Y., *J. Nat. Prod.*, 64, 578—581 (2001); c) Ida K., Masamoto K., Maoka T., Fujiwara Y., Takeda S., Hasegawa E., *J. Plant Res.*, 108, 369—376 (1995); d) Tsushima M., Fujiwara Y., Matsuno T., *J. Nat. Prod.*, 59, 30—34 (1996); e) Maoka T., Mochida K., Okuda Y., Ito Y., Fujiwara Y., *Chem. Pharm. Bull.*, 45, 1225—1227 (1997).
- 2) Fujiwara Y., Maoka T., Tetrahedron Lett., 42, 2693-2696 (2001).
- cf. a) Robert A. B., Sporn M. B., "The Retinoids," Vol. 2, ed. by Sporn M. B., Roberts A. B., Goodman D. S., Academic Press, Inc. Orlando, 1984, p. 420 (Appendix D8); b) Straub O., "List of Natural Carotenoids in Key to Carotenoids," ed. by Pfander H., Birkhäuser Verlag, Basel, 1987.
- Britton G., "UV/Visible Spectroscopy in Carotenoids," Vol. 1B, ed. by Britton, G., Liaaen-Jensen S., Pfander H., Birkhäuser Verlag, Basel, 1995, pp. 13—62.
- 5) a) Märki-Fisher E., Buchecker R., Eugster C. H., *Helv. Chim. Acta*, 67, 2143—2154 (1984); b) cf. Molnár P., Deli J., Matus Z., Tóth G., Steck A., Pfander H., *Helv. Chim. Acta*, 82, 1994—2002 (1999); c) cf. Englert G., "NMR Spectroscopy in Carotenoids," Vol. 1B, ed. by Britton G., Liaaen-Jensen S., Pfander H., Birkhäuser Verlag, Basel, 1995, pp. 147—260.
- a) Acemogle M., Uebelhart P., Ray M., Eugster C. H., *Helv. Chim. Acta*, **71**, 931–956 (1988); *b*) *cf*. Märki-Fisher E., Buchecker R., Eugster C. H., Englert G., Noack K., Vecchi M., *ibid.*, **65**, 2198–2211 (1982).
- Duchecker R., Neack K., "Circular Dichroism in Carotenoids," Vol. 1B, ed. by Britton G., Liaaen-Jensen S., Pfander H., Birkhäuser Verlag, Basel, 1995, Chapter 3, pp. 63—116.