

## Myrsiniosides A—E: Megastigmane Glycosides from the Leaves of *Myrsine seguinii* LEV.

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Eight megastigmane glycosides were isolated from the leaves of *Myrsine seguinii* collected in Okinawa. Three of them were found to be known compounds, *i.e.*, ampelopsiosionoside, alangionside J, and linarionoside A. The structures of the new megastigmane glycosides were elucidated from the spectroscopic data and their absolute stereochemistries were determined in detail using a modified Mosher's method.

**Key words** *Myrsine seguinii*; Myrsinaceae; myrsinioside; megastigmane glycoside; modified Mosher's method

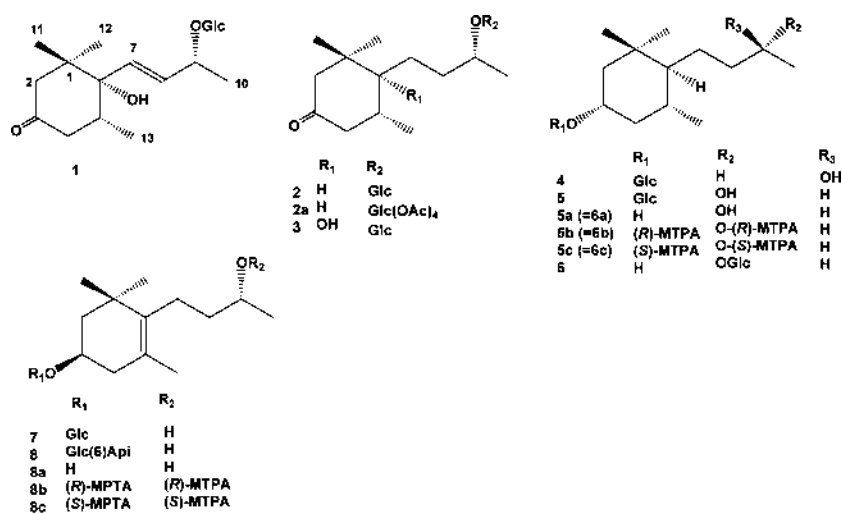
More than 10 species belonging to the Myrsinaceae are distributed in temperate and subtropical areas of Japan.<sup>1)</sup> Plants in this family are generally used as anthelmintics,<sup>2)</sup> and in China *Ardisia japonica* is used as an antitussive, expectorant, antidote and diuretic.<sup>3)</sup> *Myrsine seguinii* (syn. *Rapanea neriifolia*) is known to contain rapanone, which is isolated from stem bark, and also used as an anthelmintic for cattle. Antiviral activity has been found in crude extracts of *M. australis* and *M. salicina* growing in New Zealand.<sup>4)</sup>

From the leaves of *M. seguinii*, collected in Okinawa, flavonol glycosides, lignan sulfate and arbutin derivatives have been isolated.<sup>5)</sup> Although quantitative analysis was not performed, the content of arbutin, which is currently used in cosmetics as an antioxidant, seemed to be fairly high.<sup>5c)</sup> On further extensive phytochemical investigation, several megastigmane glycoside derivatives, which comprise a currently expanding family, were isolated. This paper deals with their structural elucidation.

### Results and Discussion

A *n*-BuOH-soluble fraction obtained by solvent partitioning of a MeOH extract of the leaves of *Myrsine seguinii* was separated by column chromatography on a highly porous synthetic resin (HP-20), and normal and reversed-phase silica gels, and by liquid–liquid partition chromatography to afford compounds **1**–**8** in the yields shown in the experimental section.

Compound **1**,  $[\alpha]_D$  ca. 0°, was isolated as an amorphous powder and was identified as ampelopsiosionoside, which has been isolated from *Ampelopsis brevipedunculata*,<sup>6)</sup> by NMR, high-resolution (HR) FAB-MS, and circular dichroism (CD) spectroscopic analyses. However, a relatively large specific optical rotation value ( $[\alpha]_D$  –35.1°) was reported for **1**. An analogous compound, 6'-*O*- $\beta$ -D-xylopyranosyl ampelopsiosionoside (platanionoside C), isolated from *Alangium platanifolium* var. *platanifolium*, showed an  $[\alpha]_D$  value of –26.0°. The  $[\alpha]_D$  value for **1** was calculated from that of 6'-*O*- $\beta$ -D-xylopyranosyl ampelopsiosionoside to be nearly –7°, using the Klyne rule,<sup>7)</sup> when the  $[M]_D$  value of methyl



MTPA:  $\alpha$ -Methoxy- $\alpha$ -trifluoromethylphenylacetic acid

Fig. 1. Structures

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orientation, and the CD spectrum also showed a positive Cotton effect at 281 nm ( $[\theta] +334$ ). Therefore the structure of **3** was elucidated to be (5*R*,6*R*,9*R*)-megastigman-3-on-6,9-diol 9-*O*- $\beta$ -D-glucopyranoside, namely dihydroampelopsinioside.

Compounds **4** (myrsinioside C, minor) and **5** (major) were separated after exhaustive isolation work by repeated preparative HPLC. Although they exhibit different retention times on HPLC, their NMR spectroscopic data were essentially superimposable (see Table 1), and they were not distinguishable from those of alangionoside J [(3*S*,5*R*,6*S*,9*R*)-megastigman-3,9-diol 3-*O*- $\beta$ -D-glucopyranoside], isolated from *Alangium premnifolium*.<sup>9</sup> Since compounds **4** and **5**, and alangionoside J commonly bear a sugar unit on the hydroxyl group at the 3-position, the absolute configurations of their ring systems must be the same. This evidence indicated that compounds **4** and **5** could be different as to the absolute configuration of the 9-position. To determine the absolute configuration of the 9-position, compound **5** was enzymatically hydrolyzed and that of the aglycone was analyzed by a modified Mosher's method (Fig. 3).<sup>10</sup> As a result, compound **5** was identified as alangionoside J, while the structure of compound **4** was then elucidated to be (3*S*,5*R*,6*S*,9*S*)-megastigman-3,9-diol 3-*O*- $\beta$ -D-glucopyranoside, and the compound was named myrsinioside C.

Myrsinioside D (**6**),  $[\alpha]_D -28.6^\circ$ , was isolated as an amorphous powder and its elemental composition, C<sub>19</sub>H<sub>36</sub>O<sub>7</sub>, was found to be the same as those of alangionoside J (**5**) and myrsinioside C (**4**). The <sup>1</sup>H- and <sup>13</sup>C-NMR spectra, including <sup>1</sup>H-<sup>1</sup>H COSY, HSQC, and HMBC spectra, of the aglycone portion were essentially superimposable with those of dihydroalangionoside I [(3*S*,5*R*,6*S*,9*R*)-megastigman-3,9-diol 9-*O*- $\beta$ -D-(6'-*O*- $\beta$ -D-apiofuranosyl)glucopyranoside] catalytically derived from the corresponding 7-ene (alangionoside I).<sup>9</sup> Therefore, **6** was expected to be a positional isomer of **5** in the position of the sugar linkage. The absolute stereochemistry of the ring system is most probably the same as that of **4** and **5**. However, in this class of compounds, the stereochemistry at the side chain does not affect the NMR spectral behavior of the ring system, even when the aglycones exhibit diastereogenic and enantiomeric relations. Thus the stereochemistry of the ring system was similarly determined using the modified Mosher's method (Fig. 3;  $\Delta\delta S-\delta R$  values are in parentheses) to be the same as that of **4** and **5**, as shown in Fig. 1.

Compounds **7** and **8** were isolated as amorphous powders and spectroscopic analyses revealed that both have the common megastigmane-type aglycone, megastigman-5-en-3,9-diol. The sugar moiety of **7** was assigned to be  $\beta$ -glucopyranose, whereas that of **8** was expected to be composed of two sugar units, such as terminal  $\beta$ -D-apiofuranose and 6-*O*-glycosylated  $\beta$ -D-glucopyranose,<sup>9</sup> from the negative-ion HR-FAB-MS (C<sub>24</sub>H<sub>42</sub>O<sub>11</sub>) spectrum, and the two anomeric protons and carbons [ $\delta_H$  4.40 ( $J=8$  Hz) on  $\delta_C$  102.6 and  $\delta_H$  4.98 ( $J=2$  Hz) on  $\delta_C$  110.9] in the one- and two-dimensional NMR spectra. Therefore compound **7** was identified as linariioside A, which has been isolated from *Linaria japonica*.<sup>11</sup> Thus compound **8** was 6'-*O*- $\beta$ -D-apiofuranosyl linariioside A, and we named it myrsinioside E. To determine the absolute configuration of the 9-position of **8**, the same procedure as that used for the aforementioned compounds

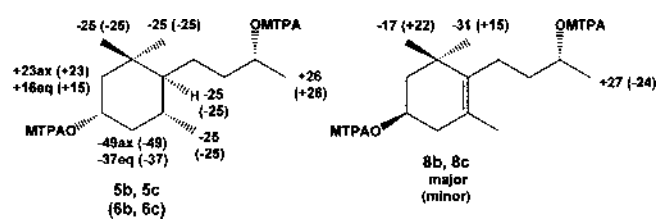


Fig. 3. Results of the Modified Mosher's Method for **5b** and **5c**, **6b** and **6c**, and **8b** and **8c**

The  $\Delta\delta$  values are in Hz ( $\delta S-\delta R$ , 400 MHz). The figures in parentheses in the structure on the left are the results for **6b** and **6c**. The figures in parentheses for **8b** and **8c** are for the minor stereoisomer.

was applied. However, the <sup>1</sup>H-NMR spectra of the (*R*)- and (*S*)-MTPA diesters (**8b,c**) of the aglycone (**8a**) of **8** showed highly complex signals. From the results obtained with the modified Mosher's method applied to both the major peaks and minor peaks, it was concluded that myrsinioside E was a mixture of 9*S*- and 9*R*-derivatives in a ratio of nearly 1:2 (Fig. 3). Finally, the structure of myrsinioside E (**8**) was characterized as a mixture of (3*S*,9*R*)- and (3*S*,9*S*)-megastigman-5-en-3,9-diol 3-*O*- $\beta$ -D-(6'-*O*- $\beta$ -D-apiofuranosyl)glucopyranoside, as shown in Fig. 1. For this reason, the same assumption must be applicable to compound **7**. Linariioside A (**7**) isolated from this plant was most probably a mixture of 9*R*- and 9*S*-diastereomers.

#### Experimental

The melting point was determined with a Yanagimoto micro melting point apparatus and is uncorrected. Optical rotations were measured on a Union Giken PM-101 digital polarimeter. FT-IR spectra were recorded on a Horiba FT-710 spectrophotometer. <sup>1</sup>H- and <sup>13</sup>C-NMR spectra were recorded on a JEOL  $\alpha$ -400 spectrometer (400 and 100 MHz, respectively) with tetramethylsilane (TMS) as internal standard. HR-FAB-MS analyses were carried out on a JEOL SX-102 mass spectrometer with PEG-400 as the calibration matrix. CD spectra were measured on a JASCO J-720 spectropolarimeter. Silica gel and reversed-phase octadecyl silica (ODS) gel open column chromatographies (RPCC) were performed on silica gel 60 (Merck, 70–230 mesh) and Cosmosil 75C<sub>18</sub>-OPN (Nacalai Tesque Co., Ltd., Kyoto, Japan) [ $\Phi=50$  mm, L=25 cm, linear gradient: MeOH-H<sub>2</sub>O (1:9, 1.5 l)→(7:3, 1.5 l), fractions of 10 g being collected. The droplet counter-current chromatograph (DCCC) (Tokyo Rikakikai, Tokyo, Japan) was equipped with 500 glass columns ( $\Phi=2$  mm, L=40 cm), and the lower and upper layers of the solvent, CHCl<sub>3</sub>-MeOH-H<sub>2</sub>O-*n*-PrOH (9:12:8:2), were used for the stationary and mobile phases, respectively. Five-gram fractions were collected and numbered according to the order of elution of the mobile phase. HPLC was performed on an ODS column [ $\Phi=20$  or 6 mm, L=250 mm; Inertsil, GL Science Co., Ltd. (Tokyo, Japan)] with UV at 254 or 210 nm and refractive index monitors. Precoated Silica gel 60 F254 TLC plates (Merck, 0.25 mm in thickness) were used for identification and preparative purification. Emulsin, hesperidinase and methyl  $\beta$ -D-xylopyranoside were purchased from Sigma Chem. Co. (St. Louis, MO, U.S.A.), and (*R*)-(+)- and (*S*)-(-)- $\alpha$ -methoxy- $\alpha$ -trifluoromethylphenylacetic acids (MPTA) were from Nacalai Tesque Co., Ltd.

**Plant Material** Leaves of *Myrsine segunii* were collected in Okinawa prefecture in 1992 and the plant material was identified by Anki Takushi of the Okinawa Prefectural Experimental Station of Forestry, to whom the authors are very grateful. A voucher specimen was deposited in the Herbarium of the Institute of Pharmaceutical Sciences, Hiroshima University Faculty of Medicine (No. 92-MS-Okinawa-0727).

**Extraction and Isolation** Dried leaves (5.95 kg) of *M. segunii* were extracted with MeOH (40 l  $\times$  3) and then the MeOH extract was concentrated to 2 l. The concentrated MeOH layer was washed with *n*-hexane (1 l  $\times$  2, *n*-hexane-soluble fraction, 61.1 g), and the MeOH was concentrated to yield a viscous gummy material. The latter was suspended in H<sub>2</sub>O (3 l), and then extracted with EtOAc (3 l) and *n*-BuOH (3 l) successively to give EtOAc- and *n*-BuOH-soluble fractions (195 g and 200 g, respectively). The remaining H<sub>2</sub>O layer was concentrated to furnish an H<sub>2</sub>O-soluble fraction (380 g).

A portion (50.0 g) of the *n*-BuOH-soluble fraction was separated first by column chromatography (CC) on a highly porous synthetic resin, Diaion HP-20 ( $\Phi=5.0$  cm,  $L=60$  cm) (Mitsubishi Chemical Co., Ltd., Tokyo, Japan), with MeOH–H<sub>2</sub>O [(1:4, 3.5:1), (2:3, 3:1), (3:2, 3:1) and (4:1, 3:1), and MeOH (3:1)], and 500 ml fractions were collected. The residue (14.8 g in fractions 9–15) of the 40% MeOH eluate obtained on HP-20 CC was subjected to silica gel (200 g) CC with CHCl<sub>3</sub> (2:1) and CHCl<sub>3</sub>–MeOH (99:1, 3:1), (97:3, 3:1), (19:1, 3:1), (37:3, 3:1), (9:1, 3:1), (17:3, 3:1), and (4:1, 3:1), fractions of 500 ml being collected. The residue (761 mg in fractions 41–54) of the 7.5% MeOH eluate was then subjected to RPCC. The residue (183 mg in fractions 99–111) was separated by DCCC to give 101 mg of ampelopsinonioside (1). The residue (66 mg in fractions 112–120) was separated by DCCC and its residue (18 mg in fractions 75–85) was purified by prep. HPLC (MeOH–H<sub>2</sub>O, 35:65) to give 8 mg of myrsinonioside B (3). The residue (86 mg in fractions 141–150) was subjected to DCCC and then to prep. HPLC (MeOH–H<sub>2</sub>O, 2:3) to give 8 mg of linarioside A (7). From the residue (64 mg in fractions 151–161), myrsinonioside C (4, 3 mg) and alangionoside J (5, 15 mg) were isolated by exhaustive HPLC separation [MeOH–H<sub>2</sub>O, 2:3; Inertsil ( $\Phi=6$  mm,  $L=250$  mm); flow rate, 1.6 ml/min;  $t_R$ : 20.0 min and 21.0 min, respectively]. From the residue (39 mg in fractions 162–170), myrsinonioside D (6, 24 mg) was isolated by DCCC. From the residue (1.18 g in fractions 70–80) of the 15–20% MeOH eluate obtained on silica gel CC, 12 mg of myrsinonioside E (8) was isolated by similar chromatographic separation.

From the residue (3.25 g in fractions 16–19) of the 60% MeOH eluate obtained on HP-20 CC, myrsinonioside A (2, 23 mg) was isolated similarly by silica gel CC, RPCC and DCCC.

**Known Compounds Isolated** Compound 1: Amorphous powder;  $[\alpha]_D^{24}$  ca. 0° ( $c=0.71$ , MeOH); CD  $[\theta]$  (nm): +792 (277) ( $c=5.51 \times 10^{-3}$  M),<sup>6a)</sup> <sup>13</sup>C-NMR (CD<sub>3</sub>OD): Table 1. Compound 5:  $[\alpha]_D^{18}$  –41.6° ( $c=0.67$ , MeOH),<sup>9)</sup> <sup>13</sup>C-NMR (CD<sub>3</sub>OD): Table 1. Compound 7: Amorphous powder;  $[\alpha]_D^{18}$  –60.8° ( $c=0.49$ , MeOH),<sup>11)</sup> <sup>13</sup>C-NMR (CD<sub>3</sub>OD): Table 1.

Compound 2: Amorphous powder;  $[\alpha]_D^{24}$  –13.8° ( $c=1.30$ , MeOH); IR  $\nu_{max}$  (KBr) cm<sup>-1</sup>: 3429, 2963, 2928, 2878, 1702, 1377, 1161, 1078, 1035; <sup>1</sup>H-NMR (CD<sub>3</sub>OD)  $\delta$ : 0.761 (3H, s, H<sub>3</sub>-11ax), 1.06 (3H, s, H<sub>3</sub>-12eq), 1.09 (3H, d,  $J=6$  Hz, H<sub>3</sub>-13), 1.20 (3H, d,  $J=6$  Hz, H<sub>3</sub>-10), 1.14 (1H, m, H-6), 1.15 (1H, m, H-7a), 1.67 (1H, m, H-7b), 1.68 (2H, m, H<sub>2</sub>-8), 1.78 (1H, m, H-5), 1.96 (1H, dd,  $J=13$ , 2 Hz, H-2eq), 2.14 (1H, t,  $J=14$  Hz, H-4ax), 2.22 (1H, ddd,  $J=14$ , 5, 2 Hz, H-4eq), 2.38 (1H, d,  $J=13$  Hz, H-2ax), 3.15 (1H, dd,  $J=9$ , 8 Hz, H-2'), 3.26 (1H, m, H-5'), 3.28 (1H, t,  $J=9$  Hz, H-4'), 3.36 (1H, t,  $J=9$  Hz, H-3'), 3.66 (1H, dd,  $J=12$ , 6 Hz, H-6'a), 3.86 (1H, dd,  $J=12$ , 2 Hz, H-6'b), 3.87 (1H, sextet,  $J=6$  Hz, H-9), 4.34 (1H, d,  $J=8$  Hz, H-1'); <sup>13</sup>C-NMR (CD<sub>3</sub>OD): Table 1; CD  $[\theta]$  (nm): +871 (282) ( $c=5.21 \times 10^{-3}$  M, MeOH); HR-FAB-MS (negative-ion mode)  $m/z$ : 373.2228 [M–H]<sup>–</sup> (Calcd for C<sub>19</sub>H<sub>33</sub>O<sub>7</sub>: 373.2226).

Compound 3: Amorphous powder,  $[\alpha]_D^{24}$  –3.0° ( $c=0.41$ , MeOH); IR  $\nu_{max}$  (KBr) cm<sup>-1</sup>: 3429, 2967, 2931, 2881, 1698, 1379, 1078, 1034; <sup>1</sup>H-NMR (CD<sub>3</sub>OD)  $\delta$ : 0.927 (3H, s, H<sub>3</sub>-11ax), 1.08 (3H, s, H<sub>3</sub>-12eq), 1.02 (3H, d,  $J=6$  Hz, H<sub>3</sub>-13), 1.20 (3H, d,  $J=6$  Hz, H<sub>3</sub>-10), 1.66 (2H, m, H<sub>2</sub>-7), 1.74 (1H, dd,  $J=13$ , 2 Hz, H-2eq), 1.75 (1H, ddd,  $J=14$ , 10, 6 Hz, H-7a), 1.88 (1H, ddd,  $J=14$ , 11, 4 Hz, H-7b), 2.03 (1H, ddd,  $J=13$ , 5, 2 Hz, H-4eq), 2.23 (1H, ddd,  $J=13$ , 6, 5 Hz, H-5), 2.46 (1H, t,  $J=13$  Hz, H-4ax), 2.88 (1H, d,  $J=13$  Hz, H-2ax), 3.15 (1H, dd,  $J=9$ , 8 Hz, H-2'), 3.57 (1H, sextet,  $J=6$  Hz, H-9), 3.66 (1H, dd,  $J=12$ , 6 Hz, H-6'a), 3.86 (1H, dd,  $J=12$ , 2 Hz, H-6'b), 4.33 (1H, d,  $J=8$  Hz, H-1'); <sup>13</sup>C-NMR (CD<sub>3</sub>OD): Table 1; CD  $[\theta]$  (nm): +439 (246), +334 (281) ( $c=3.13 \times 10^{-3}$  M); HR-FAB-MS (negative-ion mode)  $m/z$ : 389.2182 [M–H]<sup>–</sup> (Calcd for C<sub>19</sub>H<sub>33</sub>O<sub>8</sub>: 389.2175).

Compound 4: Amorphous powder, <sup>1</sup>H-NMR (CD<sub>3</sub>OD)  $\delta$ : 0.548 (1H, ddd,  $J=11$ , 5, 2 Hz, H-6), 0.831 (3H, s, H<sub>3</sub>-11ax), 0.962 (3H, s, H<sub>3</sub>-12eq), 0.981 (3H, d,  $J=6$  Hz, H<sub>3</sub>-13), 1.06 (1H, q,  $J=12$  Hz, H-4ax), 1.14 (3H, d,  $J=6$  Hz, H<sub>3</sub>-10), 1.15 (1H, t,  $J=14$  Hz, H-2ax), 1.32–1.62 (4H, m, H-5, 7b, 8a, 8b), 1.80 (1H, ddd,  $J=14$ , 4, 2 Hz, H-2eq), 2.02 (1H, m, H-4eq), 3.11 (1H, dd,  $J=9$ , 8 Hz, H-2'), 3.65 (1H, dd,  $J=12$ , 6 Hz, H-6'a), 3.66 (1H, m, H-3), 3.84 (1H, sextet,  $J=6$  Hz, H-9), 3.85 (1H, dd,  $J=12$ , 2 Hz, H-6'b), 4.33 (1H, d,  $J=8$  Hz, H-1'); <sup>13</sup>C-NMR (CD<sub>3</sub>OD): see Table 1; HR-FAB-MS (negative-ion mode)  $m/z$ : 375.2399 [M–H]<sup>–</sup> (Calcd for C<sub>19</sub>H<sub>33</sub>O<sub>7</sub>: 375.2383).

Compound 6: Amorphous powder,  $[\alpha]_D^{24}$  –28.6° ( $c=1.40$ , MeOH); IR  $\nu_{max}$  (KBr) cm<sup>-1</sup>: 3396, 2966, 2924, 2875, 1471, 1372, 1161, 1078, 1024; <sup>1</sup>H-NMR (CD<sub>3</sub>OD)  $\delta$ : 0.523 (1H, ddd,  $J=1$ , 4, 10 Hz, H-6), 0.823 (3H, s, H<sub>3</sub>-11ax), 0.900 (1H, q,  $J=12$  Hz, H-4ax), 0.945 (3H, s, H<sub>3</sub>-12eq), 0.975 (3H, d,  $J=6$  Hz, H<sub>3</sub>-13), 1.08 (1H, t,  $J=12$  Hz, H-2ax), ca. 1.08 (1H, m, H-7a), 1.17 (3H, d,  $J=6$  Hz, H<sub>3</sub>-10), 1.44 (1H, m, H-5), 1.52–1.60 (3H, m, H-7b, -8a, -8b), 1.63 (1H, ddd,  $J=13$ , 4, 2 Hz, H-2eq), 1.88 (1H, m, H-4eq), 3.14 (1H, d,  $J=9$ , 8 Hz, H-2'), 3.66 (1H, dd,  $J=12$ , 6 Hz, H-6'a), 3.69 (1H,

tt,  $J=12$ , 4 Hz, H-3), 3.84 (1H, sextet,  $J=6$  Hz, H-9), 3.85 (1H, dd,  $J=12$ , 2 Hz, H-6'b), 4.33 (1H, d,  $J=8$  Hz, H-1'); <sup>13</sup>C-NMR (CD<sub>3</sub>OD): Table 1; HR-FAB-MS (negative-ion mode)  $m/z$ : 375.2407 [M–H]<sup>–</sup> (Calcd for C<sub>19</sub>H<sub>33</sub>O<sub>7</sub>: 375.2383).

Compound 8: Amorphous powder,  $[\alpha]_D^{18}$  –76.4° ( $c=0.72$ , MeOH); IR  $\nu_{max}$  (KBr) cm<sup>-1</sup>: 3399, 2962, 2928, 2879, 1457, 1368, 1165, 1074, 1051, 1016; <sup>1</sup>H-NMR (CD<sub>3</sub>OD)  $\delta$ : 1.06 (6H, s, H<sub>3</sub>-12eq), 1.07 (3H, s, H<sub>3</sub>-11ax), 1.16 (3H, d,  $J=6$  Hz, H<sub>3</sub>-10), 1.45–1.51 (2H, m, H-8a, -8b), 1.48 (1H, t,  $J=12$  Hz, H-2ax), 1.64 (3H, s, H<sub>3</sub>-13), 1.85 (1H, ddd,  $J=12$ , 4, 2 Hz, H-2eq), 1.90 (1H, dt,  $J=11$ , 6 Hz, H-7a), 2.01 (1H, br dd,  $J=16$ , 9 Hz, H-4ax), 2.20 (1H, dt,  $J=11$ , 6 Hz, H-7b), 2.32 (1H, br dd,  $J=16$ , 4 Hz, H-4eq), 3.14 (1H, dd,  $J=9$ , 8 Hz, H-2'), 3.34 (1H, t,  $J=9$  Hz, H-3'), 3.39 (1H, ddd,  $J=9$ , 6, 2 Hz, H-5'), 3.57 (2H, s, H<sub>2</sub>-5''), 3.60 (1H, dd,  $J=12$ , 6 Hz, H-6'a), 3.70 (1H, sextet,  $J=6$  Hz, H-9), 3.76 (1H, d,  $J=10$  Hz, H-4'a), 3.88 (1H, d,  $J=2$  Hz, H-2''), 3.96 (1H, d,  $J=10$  Hz, H-4'b), 3.97 (1H, dd,  $J=12$ , 2 Hz, H-6'b), 4.00 (1H, m, H-3), 4.40 (1H, d,  $J=8$  Hz, H-1'), 4.98 (1H, d,  $J=2$  Hz, H-1''); <sup>13</sup>C-NMR (CD<sub>3</sub>OD): Table 1; HR-FAB-MS (negative-ion mode)  $m/z$ : 505.2654 [M–H]<sup>–</sup> (Calcd for C<sub>24</sub>H<sub>41</sub>O<sub>11</sub>: 505.2649).

**Acetylation of 2 to 2a** About 2 mg of 2 was acetylated with 100  $\mu$ l each of acetic anhydride and pyridine. After 4 h at 25 °C, the reagents were evaporated to give the tetraacetate (2a). Myrsinonioside tetraacetate (2a): Colorless crystals (EtOH); mp 112–115 °C; <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 0.764 (3H, s, H<sub>3</sub>-11ax), 1.035 (3H, s, H<sub>3</sub>-12eq), 1.038 (3H, d,  $J=7$  Hz, H<sub>3</sub>-13), 1.13 (3H, d,  $J=6$  Hz, H<sub>3</sub>-10), 1.50–1.65 (3H, m, H-7b, -8a, -8b), 1.76 (1H, m, H-5), 2.25 (1H, d,  $J=14$  Hz, H-2ax), 2.29 (1H, ddd,  $J=14$ , 4, 2 Hz, H-4eq), 3.67 (1H, ddd,  $J=9$ , 5, 3 Hz, H-5'), 3.72 (1H, m, H-9), 4.13 (1H, dd,  $J=12$ , 3 Hz, H-6'a), 4.23 (1H, dd,  $J=12$ , 5 Hz, H-6'b), 4.55 (1H, d,  $J=8$  Hz, H-1'), 4.94 (1H, dd,  $J=9$ , 8 Hz, H-2'), 5.08 (1H, t,  $J=9$  Hz, H-4'), 5.21 (1H, t,  $J=9$  Hz, H-3'), other signals were overlapped by the methyl signals of the aglycone and acetyls; <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$ : 19.6 (C-10), 20.59, 20.62, 20.64, 20.67, 20.73, 21.05 (C-11, -13, CH<sub>3</sub>CO $\times$ 4), 24.6 (C-7), 29.9 (C-12), 36.2 (C-5), 39.0 (C-8), 39.4 (C-1), 50.2 (C-4), 52.6 (C-6), 56.5 (C-2), 62.2 (C-6'), 68.7 (C-4'), 71.68, 71.72, 73.0 (C-2', 3', 5'), 76.1 (C-9), 99.2 (C-1'), 169.2, 169.4, 170.3, 170.5 (CH<sub>3</sub>CO $\times$ 4), 211.1 (C-3); HR-FAB-MS (positive-ion mode, *m*-nitrobenzyl alcohol as the matrix)  $m/z$ : 565.2641 [M+Na]<sup>+</sup> (+Na) (Calcd for C<sub>27</sub>H<sub>42</sub>O<sub>11</sub>Na: 565.2625).

**Enzymatic Hydrolysis of 2** Compound 2 (ca. 1 mg) was treated with 2 mg of emulsin at 37 °C in 200  $\mu$ l of H<sub>2</sub>O. Liberation of glucose was traced by TLC analysis (CHCl<sub>3</sub>–MeOH–H<sub>2</sub>O, 15:6:1, *R<sub>f</sub>* values, 2: 0.51, aglycone: 0.81, and *D*-glucose: 0.12).

**Enzymatic Hydrolysis of 5 to 5a** Compound 5 (15.0 mg) was hydrolyzed with emulsin (10 mg) in 2 ml of H<sub>2</sub>O at 37 °C for 12 h. The reaction mixture was concentrated and then subjected to silica gel column (20 g,  $\Phi=15$  mm,  $L=20$  cm) chromatography with CHCl<sub>3</sub> (100 ml), CHCl<sub>3</sub>–MeOH (19:1, 100 ml, 9:1 100 ml, 17:3, 100 ml, 7:3, 300 ml), and 10 ml fractions were collected. The aglycone (5a) and *D*-glucose were recovered in fractions 40–46 (6.2 mg, 73%) and 41–49 (3.5 mg, 49%), respectively. Aglycone (5a): Amorphous powder;  $[\alpha]_D^{21}$  –15.0° ( $c=0.41$ , MeOH); <sup>1</sup>H-NMR (CD<sub>3</sub>OD)  $\delta$ : 0.529 (1H, ddd,  $J=11$ , 5, 3 Hz, H-6), 0.831 (3H, s, H<sub>3</sub>-11ax), 0.905 (1H, q,  $J=12$  Hz, H-4ax), 0.950 (3H, s, H<sub>3</sub>-12eq), 0.972 (3H, d,  $J=6$  Hz, H<sub>3</sub>-13), 1.09 (1H, t,  $J=12$  Hz, H-2ax), 1.15 (3H, d,  $J=6$  Hz, H<sub>3</sub>-10), 1.38–1.61 (5H, m, H-5, 7a, 7b, 8a, 8b), 1.64 (1H, ddd,  $J=12$ , 4, 2 Hz, H-2eq), 1.88 (1H, dddd,  $J=12$ , 11, 4, 2 Hz, H-4eq), 3.65 (1H, sextet,  $J=6$  Hz, H-10), 3.69 (1H, tt,  $J=12$ , 4 Hz, H-3); <sup>13</sup>C-NMR (CD<sub>3</sub>OD): see Table 1; HR-FAB-MS (negative-ion mode)  $m/z$ : 213.1870 [M–H]<sup>–</sup> (Calcd for C<sub>13</sub>H<sub>25</sub>O<sub>2</sub>: 213.1855). *D*-Glucose,  $[\alpha]_D^{21}$  +42.9° ( $c=0.23$ , H<sub>2</sub>O, 24 h after being dissolved in the solvent).

**Preparation of the (R)- and (S)-MTPA Esters (5b,c) from 5a** A solution of 5a (3.1 mg) in 1 ml of dehydrated CH<sub>2</sub>Cl<sub>2</sub> was reacted with (R)-MTPA (42 mg) in the presence of *N,N'*-dicyclohexylcarbodiimide (DCC) (31 mg) and 4-*N,N'*-dimethylaminopyridine (DMAP) (15 mg), and the mixture was occasionally stirred at 25 °C for 30 min. After the addition of 1 ml each of H<sub>2</sub>O and CH<sub>2</sub>Cl<sub>2</sub>, the solution was washed with 5% HCl (1 ml), NaHCO<sub>3</sub>-saturated H<sub>2</sub>O (1 ml), and brine (1 ml) successively. The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub> and then evaporated under reduced pressure. The residue was purified by prep. TLC [silica gel (0.25 mm thickness, applied for 18 cm and developed with CHCl<sub>3</sub>–(CH<sub>3</sub>)<sub>2</sub>CO (9:1) for 9 cm and eluted with CHCl<sub>3</sub>–MeOH (9:1) to furnish the ester, 5b (7.6 mg, 81%)]. Through a similar procedure, 5c (7.1 mg, 76%) was prepared from 5a (3.1 mg) using of (S)-MTPA (43 mg), DCC (29 mg) and 4-DMAP (15 mg).

(3*R*,5*R*,6*S*,9*R*)-Megastigman-3,9-di-ol 3,9-di-(*R*)-MTPA Ester (5b): Amorphous powder; <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 0.559 (1H, ddd,  $J=11$ , 5, 3 Hz, H-6), 0.836 (3H, s, H<sub>3</sub>-11ax), 0.883 (3H, s, H<sub>3</sub>-12eq), 0.951 (3H, d,  $J=6$  Hz, H<sub>3</sub>-13), 1.15 (1H, q,  $J=12$  Hz, H-4ax), 1.23 (1H, t,  $J=12$  Hz, H-2ax), 1.27 (3H,

d,  $J=6$  Hz, H<sub>3</sub>-10), 1.44—1.75 (5H, m, H-5, 7a, 7b, 8a, 8b), 1.72 (1H, ddd,  $J=12, 4, 2$  Hz, H-2eq), 2.05 (1H, dddd,  $J=12, 11, 4, 2$  Hz, H-4eq), 3.52 (3H, q,  $J=1$  Hz, -OCH<sub>3</sub>), 3.54 (3H, q,  $J=1$  Hz, -OCH<sub>3</sub>), 5.07 (1H, sextet,  $J=6$  Hz, H-10), 5.12 (1H, tt  $J=12, 4$  Hz, H-3), 7.38—7.51 (6H, m, aromatic protons), 7.52—7.55 (4H, m, aromatic protons); HR-FAB-MS (positive-ion mode, *m*-nitrobenzyl alcohol as a matrix)  $m/z$ : 669.2647 [M+Na]<sup>+</sup> (+NaI) (Calcd for C<sub>33</sub>H<sub>40</sub>O<sub>6</sub>F<sub>6</sub>Na: 669.2627).

(3*R*,5*R*,6*S*,9*R*)-Megastigman-3,9-diol 3,9-di-(*S*)-MTPA Ester (**5c**): Amorphous powder; <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 0.497 (1H, ddd,  $J=11, 5, 3$  Hz, H-6), 0.774 (3H, s, H<sub>3</sub>-11ax), 0.821 (3H, s, H<sub>3</sub>-12eq), 0.887 (3H, d,  $J=6$  Hz, H<sub>3</sub>-13), 1.02 (1H, q,  $J=12$  Hz, H-4ax), 1.29 (1H, t,  $J=12$  Hz, H-2ax), 1.34 (3H, d,  $J=6$  Hz, H<sub>3</sub>-10), 1.43—1.75 (5H, m, H-5, 7a, 7b, 8a, 8b), 1.75 (1H, ddd,  $J=12, 4, 2$  Hz, H-2eq), 1.94 (1H, dddd,  $J=12, 11, 4, 2$  Hz, H-4eq), 3.53 (3H, q,  $J=1$  Hz, -OCH<sub>3</sub>), 3.56 (3H, q,  $J=1$  Hz, -OCH<sub>3</sub>), 5.08 (1H, sextet,  $J=6$  Hz, H-9), 5.11 (1H, tt,  $J=4, 12$  Hz, H-3), 7.37—7.43 (6H, m, aromatic protons), 7.50—7.56 (4H, m, aromatic protons); HR-FAB-MS (positive-ion mode, *m*-nitrobenzyl alcohol as a matrix)  $m/z$ : 669.2617 [M+Na]<sup>+</sup> (+NaI) (Calcd for C<sub>33</sub>H<sub>40</sub>O<sub>6</sub>F<sub>6</sub>Na: 669.2627).

**Enzymatic Hydrolysis of 6 to 6a** Compound **6** (11.0 mg) was hydrolyzed by emulsin (10 mg) in 2 ml of H<sub>2</sub>O at 37 °C for 12 h. The reaction mixture was concentrated and then subjected to silica gel column (20 g, Φ=15 mm, L=20 cm) chromatography with CHCl<sub>3</sub> (150 ml), CHCl<sub>3</sub>-MeOH (19:1, 150 ml, 9:1, 100 ml, 7:3, 300 ml), 10 ml fractions being collected. The aglycone (**6a**) and D-glucose were recovered in fractions 18—24 (5.7 mg, 91%) and 41—49 (3.4 mg, 65%), respectively. Aglycone (**6a**): Amorphous powder; [α]<sub>D</sub><sup>21</sup> -13.2° (*c*=0.38, MeOH); <sup>1</sup>H- and <sup>13</sup>C-NMR spectral data are essentially the same as those of **5a**. HR-FAB-MS (negative-ion mode)  $m/z$ : 213.1840 [M-H]<sup>-</sup> (Calcd for C<sub>13</sub>H<sub>25</sub>O<sub>2</sub>: 213.1855). D-Glucose, [α]<sub>D</sub><sup>21</sup> +44.1° (*c*=0.23, H<sub>2</sub>O, 24 h after being dissolved in the solvent).

**Preparation of the (R)- and (S)-MTPA Esters (6b,c) from 6a** In a similar manner as from **5a** to **5b** and **5c**, **6b** and **6c** were prepared from **6a** (2.8 mg each) with the respective amounts of the reagents, (*R*)- and (*S*)-MTPA (39, 37 mg), 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride (29, 33 mg) and DMAP (14, 15 mg). A usual workup gave 7.5 mg (87%), and 7.6 mg (88%) of the diesters, respectively.

(3*R*,5*R*,6*S*,9*R*)-Megastigman-3,9-diol 3,9-di-(*R*) and (*S*)-MTPA Esters (**6b,c**): Amorphous powder; <sup>1</sup>H-NMR data of **6b** and **6c** were essentially the same as those of **5b** and **5c**, respectively. 3,9-Di-(*R*)-MTPA ester (**6b**): HR-FAB-MS (positive-ion mode, *m*-nitrobenzyl alcohol as the matrix)  $m/z$ : 669.2621 [M+Na]<sup>+</sup> (+NaI) (Calcd for C<sub>33</sub>H<sub>40</sub>F<sub>6</sub>NaO<sub>6</sub>: 669.2627). 3,9-Di-(*S*)-MTPA ester (**6c**): HR-FAB-MS (positive-ion mode, *m*-nitrobenzyl alcohol as the matrix)  $m/z$ : 669.2634 [M+Na]<sup>+</sup> (+NaI) (Calcd for C<sub>33</sub>H<sub>40</sub>O<sub>6</sub>F<sub>6</sub>Na: 669.2627).

**Enzymatic Hydrolysis of 8 to 8a** Compound **8** (4.8 mg) was hydrolyzed with crude hesperidinase (20 mg) in 2 ml of H<sub>2</sub>O at 37 °C. The reaction was monitored by TLC. At 12 h and 36 h, further amounts of crude hesperidinase (40 mg) were added. At 60 h after the initiation of hydrolysis, the reaction mixture was partitioned with EtOAc (2 ml×4). The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub> and then evaporated to give an aglycone (**8a**) (1.2 mg, 60%). Aglycone (**8a**): Amorphous powder; <sup>1</sup>H-NMR (CD<sub>3</sub>OD) δ: 1.04 (3H, s, H<sub>3</sub>-12eq), 1.05 (3H, s, H<sub>3</sub>-11ax), 1.17 (3H, d,  $J=6$  Hz, H<sub>3</sub>-10), 1.38 (1H, t,  $J=12$  Hz, H-2ax), 1.43—1.53 (3H, m, H-7a, 8a, 8b), 1.63 (3H, s, H<sub>3</sub>-13), 1.68 (1H, ddd,  $J=12, 5, 2$  Hz, H-4eq), 1.71 (1H, m, H-7b), 1.93 (1H, br dd,  $J=17, 10$  Hz, H-4ax), 2.19 (1H, br dd,  $J=17, 6$  Hz, H-4eq), 3.71 (1H, sextet,  $J=6$  Hz, H-9), 3.85 (1H, m, H-3); <sup>13</sup>C-NMR (CD<sub>3</sub>OD): see Table 1; HR-FAB-MS (negative-ion mode)  $m/z$ : 211.1671 [M-H]<sup>-</sup> (Calcd for C<sub>13</sub>H<sub>23</sub>O<sub>2</sub>: 211.1698).

**Preparation of the (R)- and (S)-MTPA Esters (8b,c) from 8a** In a similar manner to as from **5a** to **5b** and **5c**, **8b** and **8c** were prepared from **8a** (0.6 mg each) with the respective amounts of the reagents, (*R*)- and (*S*)-

MTPA (28 and 27 mg), DCC (20 and 22 mg), and DMAP (12, 10 mg). The usual workup gave 1.3 mg of **8b** (72%) and 1.2 mg of **8c** (67%), respectively.

(3*S*,9*S*)-Megastigman-5-en-3,9-diol 3,9-di-(*R*)-MTPA Ester (**8b**): Amorphous powder; <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 0.941 (1/3H, s, H<sub>3</sub>-12, minor), 0.978 (1/3H, s, H<sub>3</sub>-11, minor), 1.01 (2/3H, s, H<sub>3</sub>-12, major), 1.06 (2/3H, s, H<sub>3</sub>-11, major), 1.29 (2/3H, d,  $J=6$  Hz, H<sub>3</sub>-10, major), 1.36 (1/3H, d,  $J=6$  Hz, H<sub>3</sub>-10, minor), 3.53 (2/3H, q,  $J=1$  Hz, -OCH<sub>3</sub>, major), 3.55 (3H, q,  $J=1$  Hz, -OCH<sub>3</sub>), 3.57 (1/3H, q,  $J=1$  Hz, -OCH<sub>3</sub>, minor), 5.11 (1H, sextet, H-9), 5.23 (1H, m, H-3), 7.38—7.41 (6H, m, aromatic protons), 7.51—7.54 (4H, m, aromatic protons), due to overlapping of signals, olefinic methyl and methylene protons could not be assigned; HR-FAB-MS (positive-ion mode, *m*-nitrobenzyl alcohol as the matrix)  $m/z$ : 667.2484 [M+Na]<sup>+</sup> (+NaI) (Calcd for C<sub>32</sub>H<sub>38</sub>O<sub>7</sub>F<sub>6</sub>Na: 667.2470).

(3*S*,9*R*)-Megastigman-5-en-3,9-diol 3,9-di-(*S*)-MTPA Ester (**8c**): Amorphous powder; <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 0.931 (2/3H, s, H<sub>3</sub>-12, major), 0.978 (1/3H, s, H<sub>3</sub>-12, minor), 1.02 (2/3H, s, H<sub>3</sub>-11, major), 1.03 (1/3H, s, H<sub>3</sub>-11, minor), 1.30 (1/3H, d,  $J=6$  Hz, H<sub>3</sub>-10, minor), 1.36 (2/3H, d,  $J=6$  Hz, H<sub>3</sub>-10, major), 1.58 (2/3H, s, H<sub>3</sub>-13, major), 3.52 (1/3H, q,  $J=1$  Hz, -OCH<sub>3</sub>, minor), 3.54 (3H, q,  $J=1$  Hz, -OCH<sub>3</sub>), 3.57 (2/3H, q,  $J=1$  Hz, -OCH<sub>3</sub>, major), 5.13 (1H, sextet,  $J=6$  Hz, H-9), 5.23 (1H, m, H-3), 7.37—7.40 (6H, m, aromatic protons), 7.52—7.54 (4H, m, aromatic protons), due to overlapping of signals, olefinic methyl and methylene protons could not be assigned; HR-FAB-MS (positive-ion mode, *m*-nitrobenzyl alcohol as the matrix)  $m/z$ : 667.2488 [M+Na]<sup>+</sup> (+NaI) (Calcd for C<sub>32</sub>H<sub>38</sub>O<sub>7</sub>F<sub>6</sub>Na: 667.2470).

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