A Facile Method for the Stereoselective Horner–Wadsworth–Emmons Reaction of Aryl Alkyl Ketones

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Excellent Z or E selectivity was observed in the Horner–Wadsworth–Emmons (HWE) reactions of methyl bis(2,2,2-trifluoroethyl)phosphonoacetate or ethyl 2-fluoro-2-diethylphosphonoacetate with aryl alkyl ketones bearing substituents on an aromatic moiety employing Sn(OSO₂CF₃)₂ in the presence of N-ethylpiperidine.

Key words Horner–Wadsworth–Emmons reaction; α,β -unsaturated ester; aryl alkyl ketone; methyl bis(2,2,2-trifluoroethyl)-phosphonoacetate; ethyl 2-fluoro-2-diethylphosphonoacetate; Wittig reaction

The Horner–Wadsworth–Emmons (HWE) reaction is an important method for syntheses of α,β -unsaturated esters, α,β -unsaturated ketones, and other conjugated systems.¹⁾ We have previously reported a new reaction mode for the stereoselective HWE reaction that can be used to prepare α -hydro- and α -fluoro- α,β -unsaturated esters, as shown in Chart 1.^{2–5)} As an extension of the stereoselective HWE reaction, we describe herein highly stereoselective *E*- or *Z*-alkene formation in the HWE reaction of aryl alkyl ketones bearing substituents on the aromatic moiety.

An excellent Z-selectivity was observed in the HWE reactions of methyl bis(2,2,2-trifluoroethyl)phosphonoacetate $(1)^{6}$ with phenyl or naphthyl alkyl ketones using $Sn(OSO_2CF_3)_2$ and N-ethylpiperidine. The mode of the reactions, however, was remarkably different from that of HWE reactions using NaH.²⁾ To our knowledge, detailed information regarding the substituent effects of the aromatic moiety on HWE reactions of aryl alkyl ketones under specific conditions with $Sn(OSO_2CF_3)_2$ has not been reported. We therefore carried out HWE reactions of phosphonate 1 with aryl alkyl ketones 3a—g bearing various substituents on the aromatic moiety, as shown in Chart 2. Results of the HWE reactions of these ketones employing $Sn(OSO_2CF_3)_2$ and N-ethylpiperidine are summarized in Table 1. In the cases of aromatic compounds **3a**, **b**, **d**, all having electron-donating methyl or methoxy group(s), the E: Z stereoselectivity was excellent, with ratios of 2:98, 4:96, and 2:98, respectively (entries 1, 2, and 4 in Table 1). No reaction occurred in the case of ketone 3c with a bulky mesityl group (entry 3). In contrast, the similar reactions of ketones 3e, f, both having an electron-withdrawing nitro or trifluoromethyl group, resulted in a slightly lower stereoselectivity (E:Z=17:83 or 9:91) (entries 5 and 6). The reaction of 3g (entry 7) was carried out as an application of this HWE reaction to the Z-selective synthesis of 4g, a

compound related to ozagrel. Ozagrel is an inhibitor of thromboxane A₂ synthetase and has already been launched on the market as an antithrombic and antiasthmatic drug.^{7–9)} The stereoselective outcome with excellent Z-selectivity in the Sn(OSO₂CF₃)₂-mediated HWE reactions can be rationalized in terms of a six-membered transition state proposed by us.²⁾ The HWE reaction employing NaH afforded a mixture of geometric isomers in a lower stereoselective manner, as shown in Table 2. It is worth noting that the HWE reaction of ketone 3b with the o-tolyl group has not been observed to occur under NaH conditions, whereas the reaction readily proceeds under $Sn(OSO_2CF_3)_2$ conditions (each entry 2 in Tables 1 and 2). The geometry of 4a, b, d—g was confirmed by ¹H–¹H nuclear Overhauser effect (NOE) (400 MHz, $CDCl_3$) experiments. The E: Z ratios of 4a, b, d-g were determined by utilizing ¹H-NMR analysis (400 MHz, CDCl₂).

Ethyl 2-fluoro-2-diethylphosphonoacetate (2),¹⁰⁾ an efficient fluorinating building-block,^{11,12} also provided the excellent stereoselectivity in the HWE reaction under the $Sn(OSO_2CF_3)_2$ conditions, as shown in Chart 3 and Table 3. (E)- α -Fluoro- α , β -unsaturated esters **5a**, **b**, **d**—**f** were obtained as major products in each corresponding reaction (entries 1, 2, 4, 5, and 6 in Table 3). In other words, the aromatic and ester moieties are on the same side of the double bond of (E)-5 as well as in the cases of (Z)-4. The HWE reaction of aryl alkyl ketone 3g with α -fluorophosphonate 2 using $Sn(OSO_2CF_3)_2$ in the presence of N-ethylpiperidine produced 5g with high E selectivity (entry 7 in Table 3). In contrast, similar HWE reactions of 2 with 3a—g by the conventional methods using NaH gave *E*-selective products **5a**, **b**, **d**—**g** in a range of E: Z ratios, 80: 20 - 88: 12, as listed in Table 4. The geometries of 5a, b, d, f, g were confirmed on the basis of ¹H–¹H NOE experiments (400 MHz, CDCl₃) of the corresponding primary alcohols derived by reduction of



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$$\begin{array}{cccc} CF_{3}CH_{2}O_{\downarrow}H_{\downarrow} \\ CF_{3}CH_{2}O_{\downarrow} \\ CF_{3}CH_{3}O_{\downarrow} \\ CF_{3}CH_{3}O_{\downarrow} \\ CF_{3}CH_{3}O_{\downarrow} \\ CF_{3}$$

 $\begin{array}{l} \textbf{a} \colon \textbf{R} = 3.5 \cdot xylyl, \ \textbf{b} \colon \textbf{R} = o \text{-tolyl}, \ \textbf{c} \colon \textbf{R} = mesityl, \ \textbf{d} \colon \textbf{R} = p \text{-methoxyphanyl}, \\ \textbf{e} \colon \textbf{R} = p \text{-nitrophenyl}, \ \textbf{f} \colon \textbf{R} = p \text{-(trifluoromethyl)phenyl}, \ \textbf{g} \colon \textbf{R} \neq \end{array}$

p-(1*H*-imidazol-1-ylmethyl)phenyl

a) Sn(OSO₂CF₃)₂ - A-ethylpiperidine or NaH , b) RCOMe (3a-g)

Chart 2

Table 1. Sn(OSO₂CF₃)₂-Mediated Horner–Wadsworth–Emmons Reactions of 1 with Ketones $3a - g^{\alpha_1}$

Entry	Ketone	Time (h)	Yield $(\%)^{b)}$	Alkene $(E/Z)^{c}$
1	3a	19	55	4a (2:98)
2	3b	21	84	4b (4:96)
3	3c	60 ^d	e)	_
4	3d	21	95	4d (2:98)
5	3e	14	100	4e (17:83)
6	3f	19	97	4f (9:91)
7	3g	20	65	4g (8:92)

a) Conditions: CH₂Cl₂, 0 °C, 1/Sn(OSO₂CF₃)₂/N-ethylpiperidine/**3** (1.4: 1.68: 1.54: 1). b) Isolated yields. c) ¹H-NMR (400 MHz, CDCl₃) analysis. d) 0 °C, 20 h \rightarrow room temperature (rt), 40 h. e) No reaction.



ozagrel

Table 2. NaH-Mediated Horner–Wadsworth–Emmons Reactions of 1 with Ketones $3\mathbf{a}-\mathbf{g}^{a}$

Entry	Ketone	Time (h)	Yield $(\%)^{b}$	Alkene $(E/Z)^{c}$
1	3a	45 ^{<i>d</i>})	54	4a (43 : 57)
2	3b	24	f)	
3	3c	23 ^{e)}	f)	_
4	3d	19	33	4d (51:49)
5	3e	19	99	4e (35:65)
6	3f	18	92	4f (39:61)
7	3g	16	58	4g (43 : 57)

a) Conditions: THF, 0 °C to rt, 1/NaH/3 (1.7:1.5:1). b) Isolated yields. c) ¹H-NMR (400 MHz, CDCl₃) analysis. d) rt. e) 0 °C, 16 h \rightarrow reflux, 7h. f) No reaction.



 $\label{eq:a: R = 3,5-xylyl, b: R = p-tolyl, c: R = mesityl, d: R = p-methoxyphenyl, e: R = p-nitrophenyl, f: R = p-(trifluoromethyl)phenyl, g: R =$

p-(1H-imidazol-1-ylmethyl)phenyl

a) Sn(OSO₂CF₃)₂ - N-ethylpiperidine or NaH , b) RCOMe (3a-g)

Chart 3

5a, **b**, **d**, **f**, **g**. In the case of **5e**, the geometry of the major product was established to be *E* by X-ray crystallographic analysis.¹³⁾ The diastereomer ratios of **5a**, **b**, **d**—**g** were determined by ¹H-NMR (400 MHz, CDCl₃) analysis.

In conclusion, we have demonstrated that the

Table 3. $Sn(OSO_2CF_3)_2$ -Mediated Horner–Wadsworth–Emmons Reactions of 2 with Ketones $3a - g^{\alpha_1}$

Entry	Ketone	Time (h)	Yield (%) ^{b)}	Alkene $(E/Z)^{c)}$
1	3a	18	84	5a (>99 : <1)
2	3b	18	100	5b (98:2)
3	3c	60^{d}	e)	_
4	3d	18	69	5d (98:2)
5	3e	17	96	5e (97:3)
6	3f	18	100	5f (97:3)
7	3g	22	70	5g (96 : 4)

a) Conditions: CH₂Cl₂, 0 °C, **2**/Sn(OSO₂CF₃)₂/*N*-ethylpiperidine/**3** (1.4 : 1.68 : 1.54 : 1). *b*) Isolated yields. *c*) ¹H-NMR (400 MHz, CDCl₃) analysis. *d*) 0 °C, 20 h \rightarrow rt, 40 h. *e*) No reaction.

Table 4. NaH-Mediated Horner–Wadsworth–Emmons Reactions of 2 with Ketones $3a-g^{a}$

Entry	Ketone	Time (h)	Yield $(\%)^{b}$	Alkene $(E/Z)^{c}$
1	3a	20	100	5a (87:13)
2	3b	19	100	5b (88:12)
3	3c	30 ^{<i>d</i>})	e)	_
4	3d	14	100	5d (87:13)
5	3e	6	97	5e (81:19)
6	3f	6	97	5f (83:17)
7	3g	20	73	5g (80:20)

a) Conditions: THF, 0 °C to rt, 2/NaH/3 (1.7:1.5:1). b) Isolated yields. c) ¹H-NMR (400 MHz, CDCl₃) analysis. d) 0 °C, 20 h \rightarrow reflux, 10 h. e) No reaction.



Fig. 1. Computer-Generated Drawing Derived from the X-Ray Coordinates of (E)-5e

 $Sn(OSO_2CF_3)_2$ -mediated HWE reactions of several aryl alkyl ketones bearing various substituents on the aromatic moiety proceed in a highly stereoselective manner and involve generality.

Experimental

All melting points were determined on a Yanaco micro melting point apparatus and are uncorrected. IR spectra were obtained on a Perkin-Elmer 1720 IR Fourier transform spectrometer. ¹H-NMR (200 and 400 MHz) spectra were recorded on a JEOL JNM-FX 200, JEOL JNM-GSX400, or Bruker ARX-400 spectrometer. Chemical shifts are given in δ values (ppm) using tetramethylsilane (TMS) as an internal standard. Electron impact (EI)-MS were recorded on a JEOL JMS SX-102A spectrometer. Elementary combustion analyses were performed by a Yanaco CHN CORDER MT-5. All reactions were monitored by TLC employing 0.25-mm silica gel plates (Merck 5715; 60 F₂₅₄). Preparative TLC (PTLC) was performed on 0.5-mm silica gel plates (Merck 5744; 60 F₂₅₄). Column chromatography was carried out on silica gel (Katayama Chemical K070; 70—300 mesh, Merck 9385; 230—400 mesh). Tetrahydrofuran (THF) was distilled from sodium benzophenone ketyl under N₂. *N*-Ethylpiperidine and CH₂Cl₂ were distilled

from CaH₂. All other solvents were distilled prior to use. All reagents were used as purchased.

Typical Procedure for the HWE Reaction with Sn(OSO₂CF₃)₂ To a suspension of Sn(OSO₂CF₃)₂ (350 mg, 0.84 mmol) and methyl bis(2,2,2-tri-fluoroethyl)phosphonoacetate (1) (150 μ l, 0.70 mmol) in anhydrous CH₂Cl₂ (5 ml) was added *N*-ethylpiperidine (106 μ l, 0.77 mmol) at 0 °C. The mixture was stirred at 0 °C for 1 h under argon, and *p*-methoxyphenyl ethyl ketone (3d) (75 mg, 0.50 mmol) was added. After being stirred at 0 °C for 21 h under argon, the reaction mixture was poured into H₂O (15 ml) and then extracted with CHCl₃ (50 ml×3). To the CHCl₃ extract was added *n*-hexane (180 ml), and the mixture was submitted to filtration through a silica gel short column. The filtrate was purified by PTLC [*n*-hexane/AcOEt (6:1)] to obtain *E*-4d (2 mg, 2%) and *Z*-4d (96 mg, 93%) as a colorless oil, respectively.

Typical Procedure for the HWE Reaction with NaH To a suspension of NaH (abs. 60% in oil, 52 mg, 1.35 mmol) in anhydrous THF (5 ml) was added methyl bis(2,2,2-trifluoroethyl)phosphonoacetate (1) ($320 \ \mu$ l, 1.53 mmol) at 0 °C. The mixture was stirred at 0 °C for 30 min under argon, and *p*-methoxyphenyl ethyl ketone (**3d**) (136 mg, 0.90 mmol) was added. After being stirred at 0 °C for 19 h under argon, 5% HCl (15 ml) was added and then extracted with ether ($40 \ ml \times 3$). The extract was washed with brine, dried over anhydrous MgSO₄. Evaporation *in vacuo* afforded a crude product **4d** (E: Z=51:49), which was purified by PTLC [*n*-hexane/AcOEt (6:1)] to obtain *E*-**4d** ($34 \ mg$, 18%) and *Z*-**4d** ($28 \ mg$, 15%) as a colorless oil, respectively.

Methyl (*E*)-3-(3,5-Xylyl)-2-butenoate [(*E*)-**4a**]: Colorless oil; ¹H-NMR (200 MHz, CDCl₃) δ : 2.33 (6H, s), 2.55 (3H, s), 3.75 (3H, s), 6.11 (1H, s), 7.01—7.08 (3H, m); IR (neat) 2950, 1718, 1628, 1377, 1316, 1218, 1165, 1107 cm⁻¹; EI-MS Calcd for C₁₃H₁₆O₂ MW 204.1150, Found *m*/*z* 204.1150 (M⁺).

Methyl (*Z*)-3-(3,5-Xylyl)-2-butenoate [(*Z*)-4a]: Colorless oil; ¹H-NMR (400 MHz, CDCl₃) δ : 2.15 (3H, d, *J*=1.4 Hz), 2.31 (6H, s), 3.56 (3H, s), 5.88 (1H, q, *J*=1.4 Hz), 6.81 (2H, s), 6.94 (1H, s); IR (neat) 2949, 1732, 1642, 1601, 1435, 1284, 1160, 1106 cm⁻¹; EI-MS Calcd for C₁₃H₁₆O₂ MW 204.1150, Found *m/z* 204.1175 (M⁺).

Methyl (*E*)-3-(*o*-Tolyl)-2-butenoate [(*E*)-**4b**]: Colorless oil; ¹H-NMR (400 MHz, CDCl₃) δ : 2.29 (3H, s), 2.45 (3H, d, *J*=0.9 Hz), 3.75 (3H, s), 5.77 (1H, q, *J*=0.9 Hz), 7.06—7.08 (1H, m), 7.15—7.21 (3H, m); IR (neat) 2950, 1719, 1642, 1375, 1265, 1170, 765 cm⁻¹; EI-MS Calcd for C₁₂H₁₄O₂ MW 190.0994, Found *m*/*z* 190.0991 (M⁺).

Methyl (*Z*)-3-(*o*-Tolyl)-2-butenoate [(*Z*)-**4b**]: Colorless oil; ¹H-NMR (400 MHz, CDCl₃) δ : 2.12 (3H, d, *J*=0.9 Hz), 2.20 (3H, s), 3.53 (3H, s), 5.99 (1H, q, *J*=0.9 Hz), 6.94—6.96 (1H, m), 7.17—7.19 (3H, m); IR (neat) 2950, 1732, 1647, 1373, 1202, 1122, 768 cm⁻¹; EI-MS Calcd for C₁₂H₁₄O₂ MW 190.0994, Found *m/z* 190.0996 (M⁺).

Methyl (*E*)-3-(*p*-Methoxyphenyl)-2-butenoate [(*E*)-**4d**]: Colorless oil; ¹H-NMR (200 MHz, CDCl₃) δ : 2.57 (3H, d, *J*=1.0 Hz), 3.74 (3H, s), 3.83 (3H, s), 6.11 (1H, q, *J*=1.0 Hz), 6.87—6.92 (2H, m), 7.43—7.47 (2H, m); IR (neat) 3015, 1709, 1605, 1514, 1438, 1253, 1213, 833 cm⁻¹; EI-MS Calcd for C₁₂H₁₄O₃ MW 206.0943, Found *m/z* 206.0939 (M⁺); *Anal.* Calcd for C₁₂H₁₄O₃: C, 69.89; H, 6.84. Found: C, 70.00; H, 6.96.

Methyl (*Z*)-3-(*p*-Methoxyphenyl)-2-butenoate [(*Z*)-**4d**]: Colorless oil; ¹H-NMR (200 MHz, CDCl₃) δ : 2.17 (3H, d, *J*=1.2 Hz), 3.58 (3H, s), 3.81 (3H, s), 5.88 (1H, q, *J*=1.2 Hz), 6.86—6.90 (2H, m), 7.17—7.22 (2H, m); IR (neat) 2950, 1725, 1608, 1511, 1377, 1251, 833 cm⁻¹; EI-MS Calcd for C₁₂H₁₄O₃ MW 206.0943, Found *m/z* 206.0934 (M⁺); *Anal.* Calcd for C₁₂H₁₄O₃: C, 69.89; H, 6.84. Found: C, 69.55; H, 6.91.

Methyl (*E*)-3-(*p*-Nitrophenyl)-2-butenoate [(*E*)-**4e**]: Colorless needles (EtOH): mp 206—208.5 °C; ¹H-NMR (200 MHz, CDCl₃) δ: 2.60 (3H, d, J=1.0 Hz), 3.79 (3H, s), 6.19 (1H, q, J=1.0 Hz), 7.60—7.64 (2H, m), 8.22—8.26 (2H, m); IR (KBr) 3079, 2958, 1743, 1729, 1515, 1345, 1292, 1147, 1010, 853 cm⁻¹; EI-MS Calcd for C₁₁H₁₁NO₄ MW 221.0688, Found *m*/*z* 221.0701 (M⁺); *Anal.* Calcd for C₁₁H₁₁NO₄: C, 59.73; H, 5.01; N, 6.33. Found: C, 59.65; H, 5.10; N, 6.23.

Methyl (*Z*)-3-(*p*-Nitrophenyl)-2-butenoate [(*Z*)-4e]: Colorless needles (EtOH): mp 117.5—120 °C; ¹H-NMR (200 MHz, CDCl₃) δ: 2.20 (3H, d, *J*=1.2 Hz), 3.57 (3H, s), 6.01 (1H, q, *J*=1.2 Hz), 7.33—7.37 (2H, m), 8.21—8.25 (2H, m); IR (KBr) 2953, 1743, 1728, 1632, 1515, 1345, 1010, 852 cm⁻¹; EI-MS Calcd for C₁₁H₁₁NO₄ MW 221.0688, Found *m/z* 221.0707 (M⁺); *Anal.* Calcd for C₁₁H₁₁NO₄: C, 59.73; H, 5.01; N, 6.33. Found: C, 59.77; H, 5.09; N, 6.26.

Methyl (*E*)-3-[*p*-(Trifluoromethyl)phenyl]-2-butenoate [(*E*)-**4f**]: Colorless oil; ¹H-NMR (200 MHz, CDCl₃) δ : 2.58 (3H, d, *J*=0.7 Hz), 3.77 (3H,

s), 6.15 (1H, q, J=0.7 Hz), 7.58—7.62 (4H, m); IR (neat) 2953, 1723, 1635, 1437, 1327, 1275, 1083, 723 cm⁻¹; EI-MS Calcd for $C_{12}H_{11}O_2F_3$ MW 244.0711, Found *m/z* 244.0733 (M⁺).

Methyl (*Z*)-3-[*p*-(Trifluoromethyl)phenyl]-2-butenoate [(*Z*)-**4f**]: Colorless oil; ¹H-NMR (200 MHz, CDCl₃) δ : 2.18 (3H, d, *J*=1.2 Hz), 3.57 (3H, s), 5.98 (1H, q, *J*=1.2 Hz), 7.29—7.33 (2H, m), 7.60—7.64 (2H, m); IR (neat) 2953, 1729, 1651, 1445, 1327, 1239, 1086, 729 cm⁻¹; EI-MS Calcd for C₁₂H₁₁O₂F₃ MW 244.0711, Found *m*/*z* 244.0735 (M⁺); *Anal.* Calcd for C₁₂H₁₁O₂F₃: C, 59.02; H, 4.54. Found: C, 58.59; H, 4.62.

An Inseparable Mixture of Methyl (*E*)- and (*Z*)-3-[*p*-(1*H*-Imidazol-1-yl-methyl)phenyl]-2-butenoate [(*E*)- and (*Z*)-4g]: Pale yellow oil; ¹H-NMR (400 MHz, CDCl₃) δ : 2.16 (3H, d, *J*=1.2 Hz, *Z*-isomer), 2.56 (3H, d, *J*=1.0 Hz, *E*-isomer), 3.56 (3H, s, *Z*-isomer), 3.76 (3H, s, *E*-isomer), 5.14 (2H, s, *E*- and *Z*-isomer), 5.93 (1H, d, *J*=1.2 Hz, *Z*-isomer), 6.12 (1H, d, *J*=1.0 Hz, *E*-isomer), 6.91—7.65 (7H, m, *E*- and *Z*-isomer); EI-MS Calcd for C₁₅H₁₆N₂O₂ MW 256.1212, Found *m*/z 256.1194 (M⁺).

Ethyl (*E*)-2-Fluoro-3-(3,5-xylyl)-2-butenoate [(*E*)-**5a**]: Colorless oil; ¹H-NMR (400 MHz, CDCl₃) δ : 1.03 (3H, t, *J*=7.3 Hz), 2.10 (3H, d, ⁴*J*_{H,F}=4.6 Hz), 2.29 (6H, s), 4.04 (2H, q, *J*=7.3 Hz), 6.77 (2H, s), 6.92 (1H, s); IR (neat) 2983, 1730, 1658, 1445, 1376, 1333, 851, 722 cm⁻¹; EI-MS Calcd for C₁₄H₁₇O₂F MW 236.1213, Found *m*/*z* 236.1209 (M⁺); *Anal.* Calcd for C₁₄H₁₇O₂F: C, 71.17; H, 7.25. Found: C, 70.68; H, 7.28.

Ethyl (*Z*)-2-Fluoro-3-(3,5-xylyl)-2-butenoate [(*Z*)-**5a**]: Colorless oil; ¹H-NMR (400 MHz, CDCl₃) δ : 1.36 (3H, t, *J*=7.3 Hz), 2.32 (6H, s), 2.39 (3H, d, ⁴*J*_{H,F}=3.7 Hz), 4.32 (2H, q, *J*=7.3 Hz), 6.95—6.97 (3H, m); IR (neat) 2984, 1724, 1644, 1446, 1369, 1289, 1062 cm⁻¹; EI-MS Calcd for C₁₄H₁₇O₂F MW 236.1213, Found *m*/*z* 236.1203 (M⁺); *Anal.* Calcd for C₁₄H₁₇O₂F: C, 71.17; H, 7.25. Found: C, 70.93; H, 7.37.

Ethyl (*E*)-2-Fluoro-3-(*o*-tolyl)-2-butenoate [(*E*)-**5b**]: Colorless oil; ¹H-NMR (400 MHz, CDCl₃) δ : 0.97 (3H, t, *J*=6.8 Hz), 2.09 (3H, d, ⁴*J*_{H,F}= 4.1 Hz), 2.21 (3H, s), 4.00 (2H, q, *J*=6.8 Hz), 6.98—6.99 (1H, m), 7.14—7.21 (3H, m); IR (neat) 2985, 1732, 1664, 1445, 1376, 1284, 1053, 864 cm⁻¹; EI-MS Calcd for C₁₃H₁₅O₂F MW 222.1056, Found *m/z* 222.1029 (M⁺).

Ethyl (*Z*)-2-Fluoro-3-(*o*-tolyl)-2-butenoate [(*Z*)-**5b**]: Colorless oil; ¹H-NMR (400 MHz, CDCl₃) δ : 1.38 (3H, t, *J*=7.3 Hz), 2.25 (3H, s), 2.35 (3H, d, ⁴*J*_{H,F}=3.2 Hz), 4.34 (2H, q, *J*=7.3 Hz), 7.06—7.07 (1H, m), 7.19—7.23 (3H, m); IR (neat) 2984, 1724, 1657, 1489, 1370, 1263, 1058, 861 cm⁻¹; EI-MS Calcd for C₁₃H₁₅O₂F MW 222.1056, Found *m/z* 222.1052 (M⁺); *Anal.* Calcd for C₁₃H₁₅O₂F: C, 70.25; H, 6.80. Found: C, 70.06; H, 6.87.

Ethyl (*E*)-2-Fluoro-3-(*p*-methoxyphenyl)-2-butenoate [(*E*)-**5d**]: Colorless oil; ¹H-NMR (200 MHz, CDCl₃) δ: 1.10 (3H, t, *J*=7.1 Hz), 2.13 (3H, d, ⁴J_{H,F}=4.4 Hz), 3.82 (3H, s), 4.08 (2H, q, *J*=7.1 Hz), 6.86—6.90 (2H, m), 7.10—7.15 (2H, m); IR (neat) 2984, 1729, 1654, 1577, 1466, 1377, 1151, 833 cm⁻¹; EI-MS Calcd for C₁₃H₁₅O₃F MW 238.1005, Found *m/z* 238.1012 (M⁺); *Anal.* Calcd for C₁₃H₁₅O₃F C, 65.54; H, 6.35. Found: C, 65.16; H, 6.35.

Ethyl (*Z*)-2-Fluoro-3-(*p*-methoxyphenyl)-2-butenoate [(*Z*)-**5d**]: Colorless oil; ¹H-NMR (200 MHz, CDCl₃) δ: 1.37 (3H, t, *J*=7.1 Hz), 2.43 (3H, d, ${}^{4}J_{\rm H,F}$ =3.7 Hz), 3.83 (3H, s), 4.33 (2H, q, *J*=7.1 Hz), 6.89—6.94 (2H, m), 7.36—7.40 (2H, m); IR (neat) 2984, 1719, 1607, 1460, 1369, 1304, 1182, 834 cm⁻¹; EI-MS Calcd for C₁₃H₁₅O₃F MW 238.1005, Found *m/z* 238.1006 (M⁺); *Anal.* Calcd for C₁₃H₁₅O₃F: C, 65.54; H, 6.35. Found: C, 65.65; H, 6.38.

Ethyl (*E*)-2-Fluoro-3-(*p*-nitrophenyl)-2-butenoate [(*E*)-**5e**]: Colorless plates (EtOH): mp 97—98 °C; ¹H-NMR (400 MHz, CDCl₃) δ : 1.09 (3H, t, *J*=7.3 Hz), 2.17 (3H, d, ⁴*J*_{H,F}=4.6 Hz), 4.09 (2H, q, *J*=7.3 Hz), 7.35—7.37 (2H, m), 8.22—8.24 (2H, m); IR (KBr) 3011, 2966, 1725, 1662, 1515, 1349, 1271, 1156, 1046, 852 cm⁻¹; EI-MS Calcd for C₁₂H₁₂NO₄F MW 253.0750, Found *m/z* 253.0748 (M⁺); *Anal.* Calcd for C₁₂H₁₂NO₄F: C, 56.92; H, 4.78; N, 5.53. Found: C, 56.98; H, 4.86; N, 5.52.

Ethyl (*Z*)-2-Fluoro-3-(*p*-nitrophenyl)-2-butenoate [(*Z*)-**5e**]: Colorless needles (EtOH): mp 70.5—72 °C; ¹H-NMR (400 MHz, CDCl₃) δ: 1.39 (3H, t, *J*=7.3 Hz), 2.46 (3H, d, ⁴*J*_{H,F}=3.7 Hz), 4.36 (2H, q, *J*=7.3 Hz), 7.54—7.56 (2H, m), 8.24—8.26 (2H, m); IR (KBr) 3001, 1719, 1646, 1519, 1352, 1278, 1152, 1052, 855 cm⁻¹; EI-MS Calcd for C₁₂H₁₂NO₄F MW 253.0750, Found *m*/*z* 253.0737 (M⁺); *Anal.* Calcd for C₁₂H₁₂NO₄F: C, 56.92; H, 4.78; N, 5.53. Found: C, 56.73; H, 4.77; N, 5.41.

Ethyl (*E*)-2-Fluoro-3-[*p*-(trifluoromethyl)phenyl]-2-butenoate [(*E*)-**5f**]: Colorless oil; ¹H-NMR (400 MHz, CDCl₃) δ: 1.04 (3H, t, *J*=7.3 Hz), 2.15 (3H, d, ⁴*J*_{H,F}=4.1 Hz), 4.06 (2H, q, *J*=7.3 Hz), 7.29—7.31 (2H, m), 7.60—7.63 (2H, m); IR (neat) 1730, 1665, 1618, 1446, 1327, 1269, 1166, 1048, 775 cm⁻¹; EI-MS Calcd for $C_{13}H_{12}O_2F_4$ MW 276.0773, Found *m/z* 276.0775 (M⁺); *Anal.* Calcd for $C_{13}H_{12}O_2F_4$: C, 56.53; H, 4.38. Found: C, 56.36; H,

4.44.

Ethyl (*Z*)-2-Fluoro-3-[*p*-(trifluoromethyl)phenyl]-2-butenoate [(*Z*)-**5f**]: Colorless oil; ¹H-NMR (400 MHz, CDCl₃) δ : 1.38 (3H, t, *J*=7.3 Hz), 2.44 (3H, d, ⁴*J*_{H,F}=3.7 Hz), 4.35 (2H, q, *J*=7.3 Hz), 7.48—7.50 (2H, m), 7.64—7.66 (2H, m); IR (neat) 1727, 1619, 1326, 1307, 1269, 1128, 733 cm⁻¹; EI-MS Calcd for C₁₃H₁₂O₂F₄ MW 276.0773, Found *m*/*z* 276.0770 (M⁺); *Anal.* Calcd for C₁₃H₁₂O₂F₄: C, 56.53; H, 4.38. Found: C, 56.70; H, 4.49.

An Inseparable Mixture of Ethyl (*E*)- and (*Z*)-2-Fluoro-3-[*p*-(1*H*-imidazol-1-ylmethyl)phenyl]-2-butenoate [(*E*)- and (*Z*)-**5g**]: Pale yellow oil; ¹H-NMR (400 MHz, CDCl₃) δ : 1.06 (3H, t, *J*=7.1 Hz, *E*-isomer), 1.38 (3H, t, *J*=7.3 Hz, *Z*-isomer), 2.12 (3H, d, ⁴*J*_{H,F}=4.4 Hz, *E*-isomer), 2.42 (3H, d, ⁴*J*_{H,F}=2.2 Hz, *Z*-isomer), 4.07 (2H, q, *J*=7.1 Hz, *E*-isomer), 4.34 (2H, q, *J*=7.3 Hz, *Z*-isomer), 5.14 (2H, s, *E*- and *Z*-isomer), 6.92—7.56 (7H, m, *E*- and *Z*-isomer); EI-MS Calcd for C₁₆H₁₇N₂O₂ MW 288.1274, Found *m*/*z* 288.1287 (M⁺).

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- 13) The crystallographic data of (*E*)-**5e** are as follows. (*E*)-**5e**: $C_{12}H_{12}NO_4F$, M=253.23, monoclinic, $P2_1/n$, a=12.485(4) Å, b= 6.989(1) Å, c=13.984(4) Å, $\beta=103.47(1)^\circ$, V=1186.6(5) Å³, Z=4, $D_{Calcd}=1.417$ g/cm³.