

## Re-revision of the Stereo Structure of Piperidine Lactone, an Intermediate in the Synthesis of Febrifugine

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**The stereo structure of piperidine lactone (3), an intermediate of the antimalarial agent febrifugine ((+)-1) prepared by a synthetic method, was re-revised to the *cis*-form from the *trans*-form.**

**Key words** febrifugine; piperidine lactone; structure determination

Febrifugine ((+)-1) and isofebrifugine ((+)-2) are antimalarial agents that were isolated from *Dichroa febrifuga* and *Hydrangea umbellata*, respectively (Fig. 1).<sup>1–4</sup> The plane structure<sup>5</sup> of (+)-1 and (+)-2 was first proposed in 1950. Subsequently, their relative<sup>6</sup> and absolute<sup>7</sup> structures were proposed, based on Baker's synthetic work.<sup>8–10</sup> The relative configuration<sup>11</sup> of (+)-1 was corrected in 1973 and then the absolute structures<sup>12,13</sup> of (+)-1 and (+)-2 were corrected in 1999. Currently, dramatic medical<sup>14–17</sup> and synthetic<sup>18–25</sup> studies of (+)-1 are in progress.

Historically, Barringer *et al.*<sup>11</sup> corrected the configuration of febrifugine to the *trans*-form from the *cis*-form in 1973, based on the <sup>1</sup>H-NMR data of an intermediate, piperidine lactone (3), which Baker *et al.* synthesized.<sup>9</sup> Each proton on the bridge-head of 3 was observed at  $\delta$  4.69 and 5.15 ppm with a coupling constant of 8.5 Hz (Table 2). Recently, Kobayashi *et al.* synthesized dimethyl derivatives of *trans*-3 in which each proton on the bridge-head was observed at  $\delta$  3.09 and 4.09 ppm with a coupling constant of 10.4 Hz.<sup>26,27</sup> From these directly opposing results, we thought that the structure of 3 could not be determined from the coupling constant.

In a molecular calculation, *cis*-3 afforded two optimized conformers with similar heats of formation, but differing chair (*cis*-3a) and boat (*cis*-3b) forms of the piperidine ring

(Table 1). The dihedral angle between H<sub>3a</sub>-C<sub>3a</sub>-C<sub>7a</sub>-H<sub>7a</sub> involving the bridge-head protons of *cis*-3b was smaller than in *cis*-3a. These results showed that even if the configuration of 3 was *cis*, the coupling constant between H<sub>3a</sub>-H<sub>7a</sub> would reasonably be 8.5 Hz. Since this result strongly supports our hypothesis, we discuss the exact structure of 3 synthesized by Baker *et al.* and corrected by Barringer *et al.*

*cis*-Piperidine lactone (*cis*-3), an intermediate of ( $\pm$ )-isofebrifugine ( $\pm$ -2), was prepared from *cis*-2-allyl-3-hydroxypiperidine derivatives (4) using our previously reported method<sup>18,19</sup> (Chart 1). Ozonolysis of 4, followed by PCC oxidation afforded *cis*-piperidine lactone (*cis*-5) with a cbz group. Hydrogenolysis and benzylation of *cis*-5 gave *cis*-3, which has the same <sup>1</sup>H-NMR spectrum as reported by Barringer *et al.*<sup>11</sup> Each proton on the bridge-head of *cis*-3 in the <sup>1</sup>H-NMR (60 MHz) spectrum was observed at  $\delta$  4.69 and 5.15 ppm, with a coupling constant of 7.7 Hz (Table 2).

The configuration of piperidine lactone (3) synthesized by

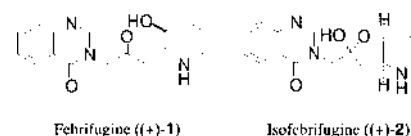
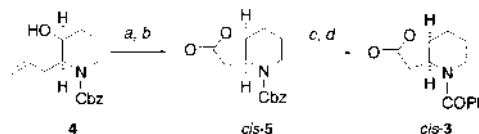


Fig. 1. The Final Corrected Structures of (+)-1 and (+)-2



Reagents and conditions: a, i) O<sub>3</sub>, MeOH, -78 °C, 2 h; ii) Me<sub>2</sub>S, room temperature (rt), 1 h; b, PCC, AcONa, CH<sub>2</sub>Cl<sub>2</sub>, rt, 3 h, 55%; c, H<sub>2</sub>, Pd(OH)<sub>2</sub>/C, acetone, rt, 24 h; d, PhCOCl, Et<sub>3</sub>N, CHCl<sub>3</sub>, rt, 1 h, 32%.

Chart 1

Table 1

Calculation method	Heat of formation (kcal/mol)		Dihedral angle (H <sub>3a</sub> -C <sub>3a</sub> -C <sub>7a</sub> -H <sub>7a</sub> )	
	3a	3b	3a	3b
PM3	-98.19	-97.73	-22.9	-4.2
AM1	-86.63	-86.39	-22.5	-10.6
MNDO	-88.43	-89.06	-15.2	8.3

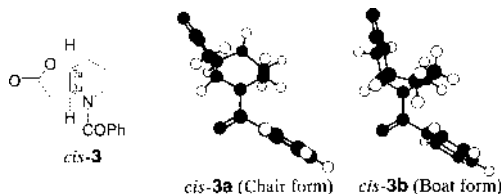
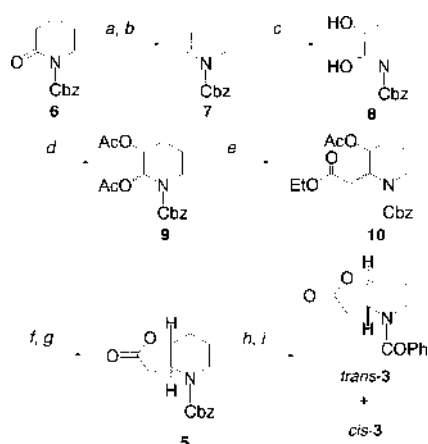


Table 2

Compound	mp (°C)	<sup>1</sup> H-NMR	
		H <sub>3a</sub>	H <sub>7a</sub>
3	99–101	5.15 (q, J=8.5 Hz) <sup>a)</sup>	4.69 (m, J=8.5, 8.5, 5.0 Hz) <sup>a)</sup>
<i>cis</i> -3 <sup>d)</sup>	85–86	5.15 (q, J=7.7 Hz) <sup>a)</sup> 5.09 (q, J=7.8 Hz) <sup>b,c)</sup>	4.69 (dt, J=7.7, 4.9 Hz) <sup>a)</sup> 4.68 (dd, J=12.5, 7.5 Hz) <sup>b,c)</sup>
<i>trans</i> -3 <sup>e)</sup>	133–134	3.24–3.65 (m) <sup>a)</sup> 3.30–3.39 (m) <sup>b)</sup>	3.97–4.08 (m) <sup>a)</sup> 4.06 (ddd, J=11.0, 10.0, 4.0 Hz) <sup>b)</sup>

a) 60 MHz. b) 500 MHz. c) 80 °C. d) NOE between H<sub>3a</sub> and H<sub>7a</sub> in NOESY spectrum was observed. e) NOE between H<sub>3a</sub> and H<sub>7a</sub> in NOESY spectrum was not observed.



Reagents and conditions: a,  $\text{NaBH}_4$ , MeOH,  $0^\circ\text{C}$ , 15 min; b,  $\text{H}_2\text{SO}_4$ , THF, rt, 1 h, 52%; c, Oxone<sup>®</sup>, acetone,  $\text{K}_2\text{CO}_3$ ,  $\text{H}_2\text{O}$ , rt, 2 h, 76% (*trans*:*cis*=1:1); d,  $\text{Ac}_2\text{O}$ ,  $\text{Et}_3\text{N}$ , DMAP, rt, 2 h, 88%; e,  $\text{CH}_2=\text{C}(\text{OEt})\text{OTMS}$ ,  $\text{Sc}(\text{OTf})_3$ ,  $\text{CH}_2\text{Cl}_2$ , rt, 13 h, 62%; f, 10% KOH, MeOH,  $\text{H}_2\text{O}$ , rt, 1.5 h; g, EDC, DMAP,  $\text{CH}_2\text{Cl}_2$ , rt, 3 h, 62% (*trans*:*cis*=41:58); h,  $\text{H}_2$ ,  $\text{Pd}(\text{OH})_2/\text{C}$ , acetone, rt, 18 h; i,  $\text{PhCOCl}$ ,  $\text{Et}_3\text{N}$ ,  $\text{CHCl}_3$ , rt, 1 h, 33% (*trans*-3)+44% (*cis*-3).

Chart 2

Baker *et al.* is *cis* based on the spectrum data. However, the difference in the melting points of **3** and *cis*-**3** was large. Consequently, we attempted to synthesis *trans*-**3** (Chart 2). Considering green chemistry, the key intermediate, piperidine-2,3-diol (**8**), was prepared by our new method using oxidation of 2,3-dehydropiperidine (**7**) with Oxone<sup>®</sup> instead of osmium reagent. *N*-Cbz-protected piperidine lactone (**5**) was prepared from **8** according to Kobayashi's method.<sup>26,27</sup> As they discussed, the reaction of **9** with non-substituted ketene silyl acetal has low diastereoselectivity. Deprotection followed by benzoylation of **5** afforded a separable mixture (3:4) of *trans*-**3** and *cis*-**3**.

The melting point and <sup>1</sup>H-NMR data of **3** reported by Barringer *et al.*, *cis*-**3**, and *trans*-**3** are summarized in Table 2. It is clear that **3** is *cis*-**3**, since the difference in the melting points of **3** and *trans*-**3** was larger than that between **3** and *cis*-**3**. Moreover, Burgess *et al.*, who reported another method of synthesizing febrifugine in 1996, used Barringer's result to determine the syn or anti configuration of a derivative of **4** prepared by the reaction of piperidine oxide with allyl silane.<sup>28</sup> This result should also be corrected.

Now, we know that the configuration of substituents on the piperidine ring in febrifugine is *trans*, without a doubt. Therefore, our result suggests that the isomerization of the *cis*-form to the *trans*-form occurs in the synthetic stage after **3** in Baker's synthesis of febrifugine. We are now in the

process of determining at which stage in Baker's synthesis of febrifugine the isomerization occurs.

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