# **Synthesis and Structure–Activity Relationships of 4-Amino-5-chloro-***N***-(1,4-dialkylhexahydro-1,4-diazepin-6-yl)-2-methoxybenzamide Derivatives, Novel and Potent Serotonin 5-HT<sub>3</sub> and Dopamine D<sub>2</sub> Receptors Dual Antagonist**

Yoshimi HIROKAWA,\*,*<sup>a</sup>* Hiroshi HARADA, *<sup>a</sup>* Takashi YOSHIKAWA, *<sup>b</sup>* Naoyuki YOSHIDA, *<sup>b</sup>* and Shiro KATO*<sup>a</sup>*

*<sup>a</sup> Medicinal Chemistry Group, Chemistry Research Laboratories, Dainippon Pharmaceutical Co. Ltd.; and <sup>b</sup> Discovery Pharmacology II Group, Pharmacolgy & Microbiology Research Laboratories, Dainippon Pharmaceutical Co. Ltd.; 33–94 Enoki, Suita, Osaka 564–0053, Japan.* Received February 18, 2002; accepted April 3, 2002

In search of a dopamine D<sub>2</sub> and serotonin 5-HT<sub>3</sub> receptors dual antagonist as a potential broad antiemetic **agent, a number of benzamides were prepared from 4-amino-5-chloro-2-methoxybenzoic acid derivatives and 6** amino-1,4-dialkylhexahydro-1,4-diazepines and evaluated for their binding affinity for the dopamine D<sub>2</sub> and the **serotonin 5-HT3 receptors using rat brain synaptic and rat cortical membranes, respectively. From the results of** both *in vitro* receptor binding and *in vivo* biological assays for the dopamine D<sub>2</sub> receptor, 1-ethyl-4-methylhexahy**dro-1,4-diazepine ring was selected as an optimum amine moiety. Introduction of one methyl group on the nitrogen atom at the 4-position and/or modification of the substituent at the 5-position of the 4-amino-5-chloro-2** methoxybenzoyl moiety caused a marked increase in the dopamine D<sub>2</sub> receptor binding affinity along with a po**tent 5-HT3 receptor binding affinity. Among the compounds, 5-chloro-***N***-(1-ethyl-4-methylhexahydro-1,4-diazepin-6-yl)-2-methoxy-4-methylaminobenzamide (82), 5-bromo (110), and 5-iodo (112) analogues exhibited a** much higher affinity for the dopamine D<sub>2</sub> receptor than that of metoclopramide  $(IC_{50} = 17.5 - 61.0 \text{ nm} \text{ vs. } 483 \text{ nm})$ . **In particular, 82 showed a potent antagonistic activity for both receptors** *in vivo* **tests. Optical resolution of the racemate 82 brought about a dramatic change in the pharmacological profile with the (***R***)-enantiomer exhibiting a** strong affinity for both the dopamine D<sub>2</sub> and the 5-HT<sub>3</sub> receptors, while the corresponding (*S*)-enantiomer had a potent and selective serotonin 5-HT<sub>3</sub> receptor binding affinity.

Key words serotonin 5-HT<sub>3</sub> receptor; dopamine D<sub>2</sub> receptor; hexahydro-1,4-diazepine; antiemetic; 4-amino-5-chloro-2methoxybenzamide; optical resolution

A number of the 2-alkoxy-4-amino-5-chlorobenzamide family possessing a potent serotoninergic or dopaminergic activity has been reported.<sup>1)</sup> The potency and selectivity of the benzamides are dependent upon the structure of their basic moiety. The classic and parent benzamide of this family is metoclopramide, which is used clinically as a stimulant of upper gastrointestinal motility and an antiemetic agent.<sup>2,3)</sup> Effects of metoclopramide are believed to be due to a combination of a relatively weak serotonin (5-hydroxytryptamine) 5-HT<sub>3</sub> (5-HT<sub>3</sub>) and dopamine  $D_2$  receptors antagonism and a serotonin  $5-HT<sub>4</sub>$  receptor agonism. The weak affinity and lack of selectivity of metoclopramide for these receptors can be explained by the large number of permissible conformers due to the flexibility of the 2-(diethylamino)ethyl chain. To date, several benzamide derivatives with potent and selective activity for the dopamine  $D_2$ , and the serotonin 5-HT<sub>4</sub>, and 5-HT<sub>3</sub> receptors have been reported. Thus, zacopride,<sup>4)</sup> BRL 24682,<sup>5)</sup> renzapride,<sup>6)</sup> SC 53116,<sup>7)</sup> and mosapride<sup>8,9)</sup> with a rigid folded framework as the amine moiety showed good affinity for the 5-HT<sub>3</sub> and/or serotonin 5-HT<sub>4</sub> receptors. On the other hand, clebopride, a compound **1**, and BRL 25594 having a benzyl group on the nitrogen atom in the amine moiety had high affinity for the dopamine  $D_2$  and  $D_3$  receptors. $^{10)}$ 

Several potent and selective  $5-HT_3$  receptor antagonists such as ondansetron and granisetron have been used clinically to prevent nausea and emesis induced by cancer chemotherapeutic agents such as cisplatin and radiation treatment.<sup>11—14)</sup> The nausea and emesis are common side effects that can cause patients to refuse subsequent chemotherapeutic sessions.<sup>15)</sup> On the other hand, dopamine  $D_2$  receptor antagonists such as phenothiazines and butyrophenones are known to be effective in the treatment of emesis and vomiting induced by centrally acting emetic stimuli such as antiparkinsonian drugs, loperamide, apomorphine, and morphine.16) In addition, the traditional antiemetic agent domperidone, a peripheral dopamine  $D<sub>2</sub>$  receptor antagonist, has been shown to be effective for the treatment of chronic upper gastrointestinal distress and the prevention of nausea and vomiting resulting from variety of causes.<sup>17,18)</sup> However, dopamine  $D<sub>2</sub>$  receptor antagonists including domperidone are only minimally effective against chemotherapy- or radiationinduced nausea and vomiting.<sup>19)</sup> Previously, we reported that the structurally novel 4-amino-*N*-(1-benzyl-4-methylhexahydro-1,4-diazepin-6-yl)-5-chloro-2-methoxybenzamide (**2x**) and the corresponding 4-amino-5-chloro-2-ethoxybenzamide  $(2y)$  are potent and selective 5-HT<sub>3</sub> receptor antagonists.<sup>20)</sup> In the course of our studies on the structure–activity relationships (SARs) of **2x**, **y**, we found that replacement of the benzyl group in the hexahydro-1,4-diazepine ring by an alkyl group results in a significant increase in the dopamine  $D<sub>2</sub>$  receptor binding affinity along with a potent  $5-HT<sub>3</sub>$  receptor antagonistic activity.<sup>21)</sup> Thus, the combination of a dopamine  $D<sub>2</sub>$  and a 5-HT<sub>3</sub> receptor antagonistic activity was seen as a good strategy for the development of effective therapeutic agents for the treatment of nausea and emesis induced by cancer chemotherapeutic agents, radiation treatment, antiparkinsonian drugs, morphine, and variety of causes. These observations led us to modify the alkyl group in the hexahydro-1,4-diazepine ring of **2x**, **y**, and prepare the optically active *N*-(1-ethyl-4-methylhexahydro-1,4-diazepin-6-yl)benzamides, resulting in the discovery of (*R*)-5-chloro-*N*-(1-ethyl-



4-methylhexahydro-1,4-diazepin-6-yl)-2-methoxy-4-methylaminobenzamide  $[(R)$ -82], a potent dopamine  $D_2$  and  $5-HT_3$ receptors antagonist. The present paper describes the synthesis of 4-amino-5-chloro-*N*-(1,4-dialkylhexahydro-1,4-diazepin-6-yl)-2-methoxybenzamides and other related compounds, and evaluates their SARs as regard to the binding affinity for the dopamine  $D_2$  and the 5-HT<sub>3</sub> receptors.

## **Chemistry**

The requisite 6-aminohexahydro-1,4-diazepine derivatives **8a**—**l** were prepared by the methods shown in Charts 1 and 2. The *N,N'*-dialkylethylenediamines **5a**—**g** were obtained by reaction of the commercially available 2-alkylaminoethanols **3l**—**o** with thionyl chloride, followed by treatment of the resultant  $N-(\beta$ -chloroethyl)alkylamine hydrochlorides **4l**—**o** with an appropriate monoalkylamine. The 6-amino-1,4-dialkylhexahydro-1,4-diazepine derivatives **8a**, **b**, **d**, **e**, **g** were synthesized from the corresponding *N*,*N*<sup> $\prime$ </sup>-dialkylethylenediamines **5a**, **b**, **d**, **e**, **g** and 2-benzyloxycarbonylaminopropenal22) (**6**) *via* the 6-benzyloxycarbonylamino-1,4-dialkylhexahydro-1,4-diazepines **7a**, **b**, **d**, **e**, **g** in a similar manner to that described in our previous paper.<sup>22)</sup> As the isopropylhexahydro-1,4-diazepines **8c**,**f** were not obtained by

reaction of  $5c$ , **f** with **6**, a previously reported method<sup>23)</sup> was used. Reaction of **5c** or **5f** with tris(hydroxymethyl)nitromethane gave the 6-hydroxymethyl-6-nitrohexahydro-1,4 diazepine **9c** or **9f**, which was treated with potassium *tert*-butoxide, followed by careful neutralization with an aqueous hydroxylamine hydrochloride, to produce the 6-nitorohexahydro-1,4-diazepine **10c** or **10f** as an unstable oil, respectively. Hydrogenation of **10c** or **10f** over Raney Ni gave the desired **8c** or **8f**, respectively (Chart 1).

Introduction of the butyl, 2,2,2-trifluoroethyl, 2,2-difluoroethyl, and 2-methoxyethyl groups at the nitrogen atom in the hexahydro-1,4-diazepine ring was performed by reductive alkylation according to the method of Marchini *et al.*24) Treatment of 6-acetylamino-1-ethylhexahydro-1,4-diazepine (**12**) and 6-benzoylamino-1-methylhexahydro-1,4-diazepine (**15**) prepared by debenzylation of the 6-(protected amino) hexahydro-1,4-diazepines **11**20) and **14**22) with the corresponding carboxylic acids and  $N$ a $BH$ <sub>4</sub> directly gave the 6-(protected amino)hexahydro-1,4-diazepines **13** and **16i**—**k** in moderate to good yield. Deprotection of **13** and **16i**—**k** with aqueous HCl produced the desired amines **8h**—**k**. The 6 amino-1-(3-hydroxypropyl)-4-methylhexahydro-1,4-diazepine (**8l**) was prepared as follows. Addition of acrolein to **15**, followed by reduction of the formyl group with NaBH<sub>4</sub> gave the 1-(3-hydroxypropyl)-4-methylhexahydro-1,4-diazepine (**16l**). **16l** was attempted to hydrolyze with aqueous HCl, but the reaction was unsuccessful. Next, acetylation of the hydroxyl group of **16l**, followed by treatment of the resultant **17** with concentrated HCl afforded the amine **8l** in 39% overall yield (Chart 2).

The optically active 6-amino-1-ethyl-4-methylhexahydro-1,4-diazepines [(*R*)- and (*S*)-**8a**] were prepared from the optically pure  $1H$ -indazol-3-carboxamide  $(R)$ -18 and its enantiomer (*S*)-**18**25) as shown in Chart 3. Hydrogenation of (*R*) and  $(S)$ -18 over Pd/C, followed by reductive ethylation<sup>26</sup> of  $(R)$ - and  $(S)$ -19 using acetaldehyde and NaBH<sub>4</sub> gave  $(R)$ - and (*S*)-**20**, which were treated with concentrated HCl to produce the optically pure (*R*)- and (*S*)-**8a** in excellent overall yield.

The preparation of various benzoic acid derivatives is exhibited in Charts 4—6. The 2-alkoxy-5-chlorobenzoic acids **23d**—**f**, **25**, **27**, **28**, **29a**, **b**, **30** were synthesized as follows. Reaction of methyl 4-acetylamino-5-chloro-2-hydroxybenzoate<sup>27)</sup> (21) with ally bromide, benzyl bromide, and butyl iodide, followed by alkaline hydrolysis of the resultants **22d**—**f** gave 2-allyloxy-, 2-benzyloxy-, and 2-butoxy-4-amino-5 chlorobenzoic acids (**23d**—**f**), respectively. The 5-chloro-2 isopropoxy-4-methylaminobenzoic acid (**25**) was obtained by methylation of  $22c^{20}$  with MeI in the presence of NaH, followed by alkaline hydrolysis of **24** in 72% overall yield. On the other hand, a similar reaction of **22a** with EtI produced the 4-(*N*-acetyl-*N*-ethyl)amino-5-chloro-2-methoxybenzoic acid **26** instead of the corresponding methyl ester. Alkaline hydrolysis of **26** furnished the corresponding 4-ethylaminobenzoic acid **27**. The 4-acetylaminobenzoic ester **22a** was treated with 1.2 molar equivalents of NaOH in aqueous MeOH at  $60^{\circ}$ C for 1 h to give the corresponding 4-acetylaminobenzoic acid **28** in excellent yield. The 4-dimethylaminobenzoic acids **29a**, **b** were directly synthesized by reductive alkylation of 23a,  $b^{28,29}$  using HCHO and NaBH<sub>3</sub>CN in moderate yield. Treatment of **23a** with a mixture of  $HCO<sub>2</sub>H$  and  $Ac<sub>2</sub>O$  gave the corresponding 4-formylamino-



Reagents and conditions: i, SOCl<sub>2</sub>, CHCl<sub>3</sub>, reflux; ii, B-NH<sub>2</sub>, EtOH-(H<sub>2</sub>O), 50 °C; iii, 1) 5 °C, MeOH, 2) NaBH<sub>4</sub>, room temperature; iv, HBr in AcOH, room temperature; iv, HBr in AcOH, room temperature; iv, HBr in Chart 1





Reagents and conditions: i, 1)  $CICO<sub>2</sub>(Cl)CHMe$ ,  $CICH<sub>2</sub>CH<sub>2</sub>Cl$ , reflux, 2) MeOH, reflux; ii, MeCH<sub>2</sub>CH<sub>2</sub>CO<sub>2</sub>H, NaBH<sub>4</sub>, toluene, reflux; iii, 10% HCl, reflux; iv, Pd/C, H<sub>2</sub>, EtOH–AcOH; v, CF<sub>3</sub>CO<sub>2</sub>H, (CHF<sub>2</sub>CO<sub>2</sub>H, MeOCH<sub>2</sub>CO<sub>2</sub>H, or CH<sub>2</sub>=CHCHO), NaBH<sub>4</sub>, toluene (MeOH); vi, 35% HCl, reflux; vii, Ac<sub>2</sub>O, pyridine, CHCl<sub>3</sub>, reflux.

#### Chart 2

benzoic acid **30** in 85% yield.

The preparation of the 5-substituted 4-amino-2-ethoxybenzoic acids **33a**, **b** is shown in Chart 5. Reaction of **31**28,29)



Reagents and conditions: i, Pd/C, H<sub>2</sub>, EtOH; ii, 1) MeCHO, MeOH,  $5^{\circ}$ C, 2) NaBH<sub>4</sub>, room temperature; iii, 35% HCl, reflux.

Chart 3

with *N*-bromosuccinimide (NBS) and iodine monochloride (ICl), followed by alkaline hydrolysis of the 5-halogenobenzoic esters **32a**, **b** gave the 5-bromo- and 5-iodo-4-amino-2 ethoxybenzoic acids (**33a**, **b**).

The 5-substituted 2-methoxy-4-methylaminobenzoic acids **37a**—**d**, **43**, and **47** were prepared as shown in Chart 6. The 2-methoxy-4-[*N*-methyl-*N*-(*p*-toluenesulfonyl)]aminobenzoic acid (**35**) was synthesized by modification of the Iwanami *et al.* method.30) The salicylic acid derivative **34**30) was treated with *ca.* 3 molar equivalents of dimethylsulfate in the presence of KOH to produce **35** instead of the corresponding methyl ester in 67% yield. Reaction of **35** with NBS, *N*chlorosuccinimide (NCS), ICl, and nitric acid-concentrated H2SO4 mixture afforded the 5-substituted 4-methylamino benzoic acids **36a**—**d**, respectively. Deprotection of **36a**—**d** with concentrated  $H_2SO_4$  afforded the corresponding 4methylaminobenzoic acids **37a**, **b**, **d**, except for the 5 iodobenzoic acid **37c**, in excellent yield. As the acidic hydrolysis of **36c** eliminates the iodine atom at the 5-position, another route for the preparation of **37c** was examined. Treatment of the 4-methylamino-2-methoxybenzoic ester **38**30) with Ac<sub>2</sub>O provided the 4-(*N*-acetyl-*N*-methyl)aminobenzoic ester **39**, which was halogenated with ICl to give **40** in good yield. Instead of acid hydrolysis, alkaline hydrolysis of **40** smoothly proceeded to afford the desired 5-iodo-2-methoxy-



Reagents and conditions: i,  $CH_2=CHCH_2Br$ , (PhCH<sub>2</sub>Br or BuI),  $K_2CO_3$ Bu<sub>4</sub>N<sup>+</sup>Br<sup>-</sup>, MeCOEt, reflux; ii, NaOH, MeOH-H<sub>2</sub>O, reflux; iii, MeI or EtI, NaH, THF, room temperature; iv, 20% NaOH, reflux; v, NaOH, MeOH-H<sub>2</sub>O, 60 °C; vi, HCHO, NaBH<sub>3</sub>CN, MeCN, room temperature; vii, Ac<sub>2</sub>O, HCO<sub>2</sub>H,  $65^{\circ}$ C.



42b;  $R_1 = CHO$ **NCS** EtS COOH  $E<sub>1</sub>SO<sub>2</sub>$ COOH COOMe EtS COOMe ix xii  $\mathbf{x}$ HN OMe HN `OMe HN `OM HI OM 46 47 44 45 Ńе. Ńе Me Me

4-methylaminobenzoic acid (**37c**) in 84% overall yield. In order to protect the 4-methylamino group of **38**, treatment of **38** with benzyl bromide in the presence of  $K_2CO_3$  gave the 4-(*N*-benzyl-*N*-methyl)aminobenzoic ester **41**. Under Vilsmeier reaction conditions, **41** gave an inseparable mixture of the desired 5-formyl-4-methylaminobenzoic ester **42a** and the 5-formyl-4-(*N*-formyl-*N*-methyl)aminobenzoic ester **42b**. Vilsmeier reaction of the moderately reactive benzoic ester **41** caused not only formylation at the 5-position but also concurrent debenzylation at the 4-amino moiety, thereby producing **42a** and the further *N*-formylated product **42b**. Alkaline hydrolysis of the mixture of **42a**, **b** produced the single product **43** in 30% yield. On the other hand, reaction of **38** with  $Br<sub>2</sub>$  and ammonium thiocyanate, followed by treatment of the resulting 5-thiocyanobenzoic ester **44** with KOH and  $Et<sub>2</sub>SO<sub>4</sub>$  gave the 5-ethylthiobenzoic ester 45, although the yield was poor. Alkaline hydrolysis of **45** and successive oxidation of the resultant benzoic acid **46** with hydrogen peroxide afforded the 5-ethylsulfonylbenzoic acid derivative **47** in good yield.

The benzotriazole- and benzimidazole-5-carboxylic acids **51** and **53** were prepared as shown in Chart 7. Introduction of a nitro group at the 3-position of **22a**, followed by hydrogenation of the resultant methyl 4-acetylamino-5-chloro-2 methoxy-3-nitrobenzoate (**48**) over Raney Ni gave the 3 aminobenzoic ester **49**. Treatment of **49** with sodium nitrite in acidic solution, and successive acidic hydrolysis of the



Reagents and conditions: i, NBS or ICl, DMF, 80 °C; ii, NaOH, MeOH–H<sub>2</sub>O,

Chart 5

COOH

OMe

COOMe

OMe

COOH

OMe

 $37a-d$ 

νń

40

43

Reagents and conditions: i, Me<sub>2</sub>SO<sub>4</sub>, KOH, Me<sub>2</sub>CO, reflux; ii, NBS (NCS, HNO<sub>3</sub>–H<sub>2</sub>SO<sub>4</sub>), DMF, 80 °C; iii, H<sub>2</sub>SO<sub>4</sub>, 5 °C; iv, Ac<sub>2</sub>O, Et<sub>3</sub>N, room temperature; v, ICl, DMF, room temperature; vi, 20% NaOH, reflux; vii, PhCH, Br, K, CO., DMF, reflux; viii, POCl, DMF, 80 °C; ix, NaOH, MeOH–H<sub>2</sub>O, reflux; x, Br<sub>2</sub>, NH<sub>4</sub>SCN, MeOH, 5 °C; xi, Et<sub>2</sub>SO<sub>4</sub>, KOH, MeOH, reflux; xii, H<sub>2</sub>O<sub>2</sub>, AcOH, room temperature.

benzotriazole **50** afforded 7-chloro-4-methoxy-1*H*-benzotriazole-5-carboxylic acid (**51**). Reflux of a solution of **49** in the presence of TsOH in toluene gave the benzimidazole-5-carboxylic ester **52**, and alkaline hydrolysis of **52** provided the benzimidazole-5-carboxylic acid **53**.

The benzamide and carboxamide derivatives **71**, **72**, **80**— **82**, **85**—**97**, **98**—**112**, **114**, **115**, **117**, **118**, **120**—**123**, **125**— **130**, (*R*)- and (*S*)-**81**, (*R*)- and (*S*)-**82**, (*R*)- and (*S*)-**110**, (*R*) and (*S*)-**112** were synthesized by reaction of the appropriate benzoic acid or carboxylic acid with the amines **8a**—**o**, (*R*) and (*S*)-**8a** in the presence of 1-ethyl-3-[3-(dimethylamino) propyl]carbodiimide hydrochloride (EDC) as a coupling reagent (Chart 8).

On the other hand, hydrogenation of **2x** over Pd/C gave the debenzylated benzamide **75**. The 1-(2-hydroxyethyl)benzamide derivative **96** was derived from the 1-(2-methoxyethyl)- 4-methylhexahydro-1,4-diazepine analogue **95** using trimethylsilyl iodide. Treatment of the 6-(*tert*-butoxycarbonylamino)-2,3-dimethoxybenzoic acid<sup>31)</sup> (62) with thionyl chloride followed by condensation of the resultant acid chloride with **8a** afforded directly the 6-amino-2,3-dimethoxybenzamide **124** in 49% yield (Chart 9).

The 5-fluoro-2-methoxy-4-methylaminbenzamide **113** was prepared from 2,4,5-trifluorobenzoic acid (**63**) as shown in Chart 10. Treatment of **63** with *N*-benzylmethylamine afforded the regioselective 4-(*N*-benzyl-*N*-methyl)amino-2,5 difluorobenzoic acid (**64**) in 49% yield. Condensation of **64**

with **8a** produced the 4-(*N*-benzyl-*N*-methyl)aminobenzamide **65**, which was underwent a substitution reaction with NaOMe to provide the 2-methoxy-4-(*N*-benzyl-*N*-methyl) aminobenzamide **66**. Hydrogenation of **66** over Pd/C afforded the desired 5-fluorobenzamide 113. NaBH<sub>4</sub> reduction of **115** gave the corresponding 5-hydroxymethylbenzamide **116**. The 5-sulfamoylbenzamide **119** was prepared as follows. Introduction of a chlorosulfonyl group at the 5-position of the commercially available 4-chloro-2-methoxybezoic acid (**67**) was performed by reaction with chlorosulfuric acid. Reaction of the resultant 4-chloro-5-chlorosulfonyl-2 methoxybenzoic acid (**68**) with 28% aqueous ammonia gave the 5-sulfamoylbenzoic acid **69**, which was condensed with **8a** to provide the 4-chloro-5-sulfamoylbenzamide (**70**). Finally, treatment of 70 with MeNH<sub>2</sub> in a sealed tube afforded the desired **119** in 84% yield (Chart 10).

## **Results and Discussion**

The affinity of the benzamide and carboxamide derivatives for the  $5-\text{HT}_3$  receptor in the central nervous system was determined by inhibition of the specific binding of  $[^{3}H]$ -GR65630 to rat cortical membrane according to the method previously reported for [<sup>3</sup>H]quipazine binding.<sup>20)</sup> The affinity for the dopamine  $D<sub>2</sub>$  receptor was evaluated in binding assays by competition for the binding of the radioligand  $[^3H]$ spiperone, a dopamine  $D<sub>2</sub>$  receptor agonist, to binding sites in rat



Reagents and conditions: i,  $HNO<sub>3</sub>-H<sub>2</sub>SO<sub>4</sub>$ , <10 °C; ii, Raney Ni, H<sub>2</sub>, EtOH–H<sub>2</sub>O; iii, H<sub>2</sub>SO<sub>4</sub>, NaNO<sub>2</sub>, H<sub>2</sub>O, 5 °C; iv, reflux; v, *p*-TsOH, toluene, reflux; vi, NaOH, MeOH-H<sub>2</sub>O, reflux.

Chart 7



Reagents and conditions: i, Pd/C, H<sub>2</sub>, EtOH, 50 °C; ii, TMSI, CHCl<sub>3</sub>, room temperature; iii, SOCl<sub>2</sub>, DMF, toluene, 60 °C; iv, 8a, CHCl<sub>3</sub>, room temperature.

Chart 9



Reagents and conditions: i, EDC,  $CH_2Cl_2$ , room temperature.



Reagents and conditions: i, PhCH<sub>2</sub>NHMe, Et<sub>3</sub>N, pyridine, reflux; ii, 8a, EDC, CH<sub>2</sub>Cl<sub>2</sub>, room temperature; iii, NaOMe, DMF, 120 °C; iv, Pd/C, H<sub>2</sub>, EtOH–AcOH, 50 °C; v, NaBH<sub>4</sub>, MeOH, room temperature; vi, ClSO<sub>3</sub>H, 60—70 °C; vii, aq. NH<sub>3</sub>, 40—50 °C; viii, 8a, EDC, CH<sub>2</sub>Cl<sub>2</sub>, room temperature; ix, NH<sub>2</sub>Me, EtOH, 140 °C.

Chart 10

striatum.<sup>32)</sup> Receptor binding data at the  $5-HT<sub>3</sub>$  and the dopamine  $D_2$  receptors for compounds  $2x, y$ , and  $71-130$ along with those for each enantiomer of **81**, **82**, **110**, and **112** are listed in Table 1. For comparison, data for metoclopramide, the potent and selective  $5-\text{HT}_3$  receptor antagonist, ondansetron, and the potent and selective dopamine  $D<sub>2</sub>$  receptor antagonist, domperidone are included in Table 6.

Metoclopramide was characterized by its weak affinity for the dopamine  $D_2$  and the 5-HT<sub>3</sub> receptors, while domperidone (15.1 nm) showed a potent affinity for only dopamine  $D_2$ receptor, and ondansetron (1.26 nm) was exhibited a potent affinity for only  $5-HT_3$  receptor. First, the effects on affinity of a substitution at the 1- and 4-positions in the hexahydro-1,4-diazepine ring of **2x**, **y** leading to **71**—**97** were examined and the newly prepared benzamides were compared to **2x**, **y**, two potent and selective  $5-HT<sub>3</sub>$  receptor antagonists (serotonin 5-HT<sub>4</sub><sup>33</sup> and dopamine  $D_2$  receptor binding affinities  $>1000$  nm). Benzamides 71 and 72 having 1-benzyl-4methylhexahydro-1,4-diazepine ring in the amine moiety, as well as  $2x$ , y, showed a much strong affinity for the  $5-HT<sub>3</sub>$  receptor regardless of the substituent at the 2-alkoxy group in the benzoyl moiety. However, their affinity for the dopamine  $D_2$  receptor was weak ( $>1000 \text{ nm}$ ). Replacement of the methyl group in **2x** by an ethyl group (giving **73**) resulted in an increase in affinity for the dopamine  $D_2$  receptor (292 nm). On the other hand, the corresponding 2-ethoxybenzamide **74** did not show good affinity for the dopamine  $D<sub>2</sub>$  receptor, in spite of a potent  $5-HT<sub>3</sub>$  receptor binding affinity. An interesting increase in the affinity of  $73$  for the dopamine  $D<sub>2</sub>$  receptor led us to prompt further modifications of the substituent in the hexahydro-1,4-diazepine ring. The debenzylated benzamide (**75**) of **2x** exhibited weak affinity for the dopamine  $D_2$  receptor. Though **76**—**79**, having a 1,4-dimethylhexahydro-1,4-diazepine ring, did not display good binding affinity for the dopamine  $D<sub>2</sub>$  receptor, the 4-methylamino counterpart **80** showed an enhanced affinity for this receptor. Compound **80** exhibited a binding affinity for the dopamine  $D_2$  receptor *ca.* 2.5-fold stronger than that of metoclopramide (198 nm)  $vs.$  483 nm) along with a potent  $5-HT<sub>3</sub>$  receptor affinity (1.86 nM). Replacement of the benzyl group of **2x**, **73**, and **74** by an ethyl, propyl, isopropyl, or cyclopropyl group (yielding **81**, **83**, **84**, **86**—**91**) resulted in an increase in the binding affinity for the dopamine  $D_2$  receptor except for 88 with a 1isopropyl-4-methylhexahydro-1,4-diazepine ring. The affinity of the benzamides prepared for the dopamine  $D<sub>2</sub>$  receptor did not depend upon the size of the substituent and **91** with a 1 cyclopropyl-4-ethylhexahydro-1,4-diazepine ring had the most potent affinity (50.3 nm) for this receptor. The affinity of **91** for the dopamine  $D_2$  receptor was *ca*. 10-fold more potent than that of metoclopramide and *ca.* 3-fold less potent than that of domperidone. The dopamine  $D_2$  receptor binding site of the substituent in the hexahydro-1,4-diazepine ring is thought to have a spatial narrow pocket with an optimal volume for a methyl, ethyl, or cyclopropyl group. This results are in contrast with those previously reported regarding the recognition of the dopamine  $D_2$  receptor<sup>34</sup>); *i.e.* the role of the *N*-benzyl substitution and the extended shape of the molecule were shown to be important for the recognition of the dopamine  $D<sub>2</sub>$  receptor by clebopride, BRL 25594, and other structurally-related dopaminergic antagonists, which displayed high affinity for this receptor. The *N*-methylamino analogous **82** and **85** increased the binding affinity for the

Table 1. Physical Data for 5-Chloro-2-alkoxy-*N*-(hexahydro-1,4-diazepin-6-yl)benzamides (**71**, **72**, **75**, **80**—**82**, **85**—**97**)





*a*) Abbreviations for the solvents used are as follows: M=methanol, E=ethanol, T=toluene, IP=isopropanol, DE=diethyl ether. *b*) All compounds were analyzed for C, H, N, and halogen; analytical results were within ±0.4% for the theoretical values. c) See ref. 20. d) Oxalic acid. e) See Experimental Section. f) Maleic acid. g) Fumaric acid. acid. acid. acid. acid. acid. acid. acid. acid

dopamine  $D_2$  receptor compared with the parents  $81$  and  $83$ , respectively (61.0 nM and 48.4 nM *vs.* 127 nM and 216 nM, respectively). A similar result has been reported in the SARs study of the antipsychotic drug nemonapride (Fig. 1), which showed a very strong affinity for the dopamine  $D_2$  receptor; *i.e.* the 5-chloro-2-methoxy-4-methylaminobenzamide showed a more potent affinity (1 order of magnitude) for the dopamine  $D_2$  receptor than the corresponding 4-aminobenzamide.30) Benzamides with a butyl group (**92**) and alkyl substituents having a fluoro (**93**, **94**), methoxy (**95**), and hydroxy  $(96, 97)$  groups exhibited weak affinity for the dopamine  $D<sub>2</sub>$ receptor along with a potent  $5-HT<sub>3</sub>$  receptor binding affinity. The presence of a large substituent with oxygen atom and fluorine had an unfavorable effect on the affinity for the

Table 2. Physical Data for 5-Chloro-2-alkoxy-*N*-(1-ethyl-4-methylhexahydro-1,4-diazepin-6-yl)benzamides (**98**—**108**)





*a*) Abbreviations for the solvents used are as follows: C=chloroform. See footnote *a* in Table 1. *b*, *c*) See footnote *b*, *c* in Table 1. *d*) The presence of ethanol was confirmed by <sup>1</sup>H-NMR spectra.

dopamine  $D<sub>2</sub>$  receptor. From the results above, benzamides **81—83**, **85—87**, **90**, and **91** with a moderate dopamine  $D_2$ receptor binding affinity were selected and further *in vivo* biological assays involving inhibition of apomorphine-induced emesis in dogs at 1.0 mg/kg, *p.o.* were performed. The inhibition data for metoclopramide and domperidone was included for comparison. Benzamides **81** and **82** having 1-ethyl-4 methylhexahydro-1,4-diazepine ring showed a potent inhibition of apomorphine-induced emesis with  $ED_{50}$  values of 0.20 and 0.22 mg/kg,  $p.o.$ , respectively (Table 7). These  $ED_{50}$ values were *ca*. 2-fold stronger than the  $ED_{50}$  value of metoclopramide (0.45 mg/kg, *p.o.*). On the other hand, **83**, **85**— **87**, **90**, and **91** displayed weak inhibition of apomorphine-induced emesis compared with metoclopramide, although their affinity for the dopamine  $D<sub>2</sub>$  receptor was higher than that of metoclopramide. These compounds were, thus, considered to have low oral bioavailability. Therefore, the optimum substituents at the 1- and 4-positions in the hexahydro-1,4-diazepine ring were found to be ethyl and methyl groups, respectively.

Next, the influence of a substitution at the 4- and 2-positions in the benzoyl moiety of **81** was examined in detail, while keeping the 1-ethyl-4-methylhexahydro-1,4-diazepine ring constant (**98**—**108**). As mentioned above, the 4-methylaminobenzamide derivative **82** showed an increased affinity for the dopamine  $D<sub>2</sub>$  receptor with a maintained potent binding affinity for the  $5-\text{HT}_3$  receptor. Replacement of the

methyl group of **82** by an ethyl group (yielding **98**) and further introduction of a methyl group into the 4-methylamino group of **82** (yielding **99** with 4-dimehylamino group) led to a decrease in the binding affinity for the dopamine  $D<sub>2</sub>$  receptor, although the affinity of 98 for the  $5-HT_3$  receptor was *ca*. 4-fold stronger than that of the parent **81**. The 4-dimethylaminobenzamide derivative **99**, on the other hand, showed a decrease in affinity for both the  $5-HT_3$  and the dopamine  $D_2$ receptors. Introduction of a formyl group (giving **100**) weakened the affinity for the dopamine  $D<sub>2</sub>$  receptor, but maintained a strong  $5-HT<sub>3</sub>$  receptor binding affinity. In contrast, the 4-acetylamino derivative **101** did not show any affinity for both receptors, although the reason for this was unclear. Replacement of the methoxy group at the 2-position of **81**, **82**, and **99** by an ethoxy (**102**, **103**), isopropoxy (**104**, **105**), allyloxy (**106**), benzyloxy (**107**), or butoxy (**108**) group tended to significantly decrease the affinity for the dopamine  $D_2$  receptor. Thus, a small substituent, such as a methoxy group at this position, was found to be essential for recognition of the dopamine  $D_2$  receptor. The 5-HT<sub>3</sub> receptor binding affinity of the compounds described above was potent, particularly that of the 2-isopropoxybenzamides **104** and **105**  $(0.46 \text{ nm}$  and  $0.75 \text{ nm}$ , respectively) which had an affinity much stronger than that of ondansetron  $(1.26 \text{ nm})$ . Thus, for both receptor binding assays, the optimum substituents at the 2- and 4-positions in the benzoyl moiety were found to be methoxy and methylamino groups, respectively.

Table 3. Physical Data for 2-Alkoxy-*N*-(1-ethyl-4-methylhexahydro-1,4-diazepin-6-yl)benzamides (**109**—**119**)





*a*) See footnote *a* in Table 1. *b*) All compounds were analyzed for C, H, N, S, and halogen; analytical results were within±0.4% for the theoretical values. *c*) Fumaric acid. *d*) Oxalic acid. *e*) See Experimental Section. *f*) Maleic acid.

The influence of a substitution at the 5-position in the benzoyl moiety of **82** and **102** was next studied (**109**—**119**). Replacement of the chlorine atom of **102** and **82** by a bromine atom (giving **109** and **110**) caused a remarkable increase in affinity for both receptors. The 5-bromo-2-methoxy-4-methylaminobenzamide **110**, in particular, showed a potent affinity for the dopamine  $D_2$  (35.5 nm) and the 5-HT<sub>3</sub> (2.54 nm) receptors. Replacement of the iodine atom in **102** and **82** (yielding **111** and **112**) resulted in an increase in the dopamine  $D<sub>2</sub>$  receptor binding affinity but no change in the  $5-\text{HT}_3$  receptor binding affinity. Introduction of fluoro, nitro, formyl, hydroxymethyl, ethylthio, ethylsulfonyl, and sulfamoyl groups at the 5-position (**113**—**119**) did not provide any improvements in affinity for both receptors, except for the 5-ethylthiobenzamide  $117$  (dopamine  $D<sub>2</sub>$  receptor binding affinity; 26.1 nm) and the 5-formylbenzamide  $115$  (5-HT<sub>3</sub> receptor binding affinity; 3.00 nM). From these results, the chlorine, bromine and iodine atoms were found to be optimal substituents at the 5-position in the benzoyl moiety.

The influence of several 2-methoxybenzamides and carboxamide derivatives (**120**—**130**) on the affinity for both the dopamine  $D_2$  and the 5-HT<sub>3</sub> receptors was examined. The 5iodo-2-methoxy (**120**), 3-bromo-2,6-dimethoxy (**121**), 3,5-dibromo-2,6-dimethoxy (**122**), 4-chloro-2-methoxy-5-nitro (**123**), 6-amino-2,3-dimethoxy (**124**), 5-bromo-3-chloro-6 hydroxy-2-methoxy (**125**), 5-ethylthio-2-methoxy (**126**), 5 ethylsulfonyl-2-methoxy (**127**), and 5-sulfamoyl-2-methoxy

(**128**) benzamides, and the benzotriazole- (**129**) and the benzimidazole- (**130**) 5-carboxamides were examined. All the compounds showed weak affinity for both receptors except **128** which displayed a moderate affinity for the  $5-HT<sub>3</sub>$  receptor (13.7 nm). From these results, the nitrogen atom at the 4position, the halogeno atom at the 5-position, and the methoxy group at the 2-position in the benzoyl moiety were found to be essential for a potent affinity for both receptors. It is well-known that the pharmacophore for  $5-HT<sub>3</sub>$  receptor antagonists consists of an aromatic ring, a carbonyl group, and a basic nitrogen, and that their location is crucial for the activity. $35)$  Thus, it is convincing that a number of benzamides, regardless of the nature of the  $N^1, N^4$ -disubstituents on the hexahydro-1,4-diazepine ring and the benzoyl moiety, possess nanomolar affinity for the  $5-HT<sub>3</sub>$  receptor and are essentially equipotent to previously described  $5-\text{HT}_3$  receptor antagonists such as ondansetron and granisetron. On the other hand, although no distinct SARs concerning the affinity of benzamides for the dopamine  $D<sub>2</sub>$  receptor were identified, 2-methoxy, 4-methylamino, and 5-bromo or iodo groups in the benzoyl moiety provided some improvement in the affinity for this receptor.

Finally, the binding affinity of the enantiomers of **81**, **82**, **110**, and **112** for both receptors was studied. The binding affinity for the dopamine  $D_2$  receptor of the  $(R)$ -enantiomers of **81**, **82**, **110** [(*R*)-**81**, (*R*)-**82**, (*R*)-**110**] was *ca.* 2-fold higher than that of the corresponding racemates except the 5Table 4. Physical Data for *N*-(1-Ethyl-4-methylhexahydro-1,4-diazepin-6-yl)benzamides (**120**—**130**)





*a*) Abbreviations for the solvents used are as follows: AN=acetonitrile. See footnote *a* in Table 1. *b*) See footnote *b* in Table 3. *c*) See footnote *c* in Table 1. *d*) The presence of ethanol was confirmed by the <sup>1</sup>H-NMR spectra. *e*) See Experimental Section. *f*) Fumaric acid. *g*) Maleic acid. *h*) The presence of isopropanol was confirmed by <sup>1</sup>H-NMR spectra.

Table 5. Physical Data for the Optically Active 2-Methoxy-*N*-(1-ethyl-4-methylhexahydro-1,4-diazepin-6-yl)benzamides [(*R*)-**81**, **82**, **110**, **112** and (*S*)-**81**, **82**, **110**, **112**]





 $a, b$ ) See footnote  $a, b$  in Table 1. *c*) The enantiomeric purity was confirmed to be >99% ee by HPLC analysis [column; CHIRALPAK AS (DAICEL Chemical Industries, Ltd.),  $4.6 \times 250$  mm i.d., eluent; hexane/EtOH/Et<sub>2</sub>NH=94/6/0.2, flow rate; 0.8 ml/min, column temperature; 25 °C, detection; 254 nm]. retention time: (*R*)-81; 24.4 min, (*S*)-81; 28.0 min. *d*) The absolute configuration of the amine moiety is *R*. *e*) The absolute configuration of the amine moiety is *S*. *f*) Maleic acid. *g*) Oxalic acid. *h*) The presence of isopropanol was confirmed by <sup>1</sup>H-NMR spectra.

Table 6. Dopamine  $D_2$  and 5-HT<sub>3</sub> Receptors Binding Affinity for Benzamide and Carboxamide Derivatives

	Dopamine $D_2$ receptor	$5-HT_3$ receptor
Compds.	binding affinity <sup>a)</sup> $IC_{50}$ (nM)	binding affinity <sup>b)</sup> $IC_{50}$ (nM)
2x	>1000	2.07
2y 71	>1000 >1000	0.30 2.35
72	>1000	>2.0
73	292	1.91
74	>1000	0.17
75	758	13.7
76 77	>1000 >1000	9.56 1.41
78	>1000	7.60
79	>1000	2.07
80	198	1.86
81	127	8.50
82 83	61.0 216	4.76 2.27
84	875	4.30
85	48.4	2.21
86	218	24.3
87 88	160 >1000	19.0 51.8
89	468	24.9
90	182	3.16
91	50.3	5.05
92	538	4.50
93 94	>1000 449	3.35 1.80
95	665	9.20
96	371	1.65
97	647	5.51
98	181	1.89
99 100	558 430	35.9 7.08
101	>1000	>100
102	747	5.66
103	>1000	14.6
104	>1000	0.46
105 106	>1000 >1000	0.75 4.75
107	>1000	2.38
108	>1000	4.53
109	492	1.63
110 111	35.5 539	2.54 5.33
112	17.5	5.14
113	186	11.4
114	88.4	11.0
115 116	245 >1000	3.00 >100
117	26.1	70.9
118	382	>100
119	725	>100
120	355	>100
121 122	>1000 >1000	>100 >100
123	>1000	>100
124	>1000	>100
125	>1000	>100
126 127	635 >1000	>100 >100
128	758	13.7
129	>1000	>100
130	>1000	66.3
$(R) - 81$	86.9	11.8
$(S) - 81$ $(R) - 82$	517 34.6	7.00 2.86
$(S) - 82$	320	1.49





a) Determined in rat brain synaptic membrane using [<sup>3</sup>H]spiperone. *b*) Determined in rat brain cortical membrane using [3H]GR65630.

Table 7. Inhibition of Apomorphine-Induced Emesis in Dogs by **81**—**83**, **85**—**87**, **90**, **91**, **110**, (*R*)-**81**, (*R*)-**82**, (*R*)-**110**, and (*R*)-**112**

Compds.	Inhibition of apomorphine-induced emesis $(1.0 \,\text{mg/kg}, p.o.)$ % [ED <sub>50</sub> : mg/kg, p.o.]
81	100 [0.20]
$(R) - 81$	87
82	100 [0.22]
$(R) - 82$	100 [0.13]
83	52
85	49
86	25
87	14
90	40
91	52
110	56
$(R) - 110$	79 [0.40]
$(R) - 112$	90
Metoclopramide	100 [0.45]
Domperidone	100 [0.02]

iodobenzamide (*R*)-**112** which had an affinity for the dopamine  $D_2$  receptor *ca*. 2-fold weaker than that of its racemate **112**. On the other hand, the affinity of all (*R*)-enantiomers for the  $5-\text{HT}_3$  receptor was approximately the same as that of each corresponding racemate. Although the (*S*) enantiomers [(*S*)-**81**, (*S*)-**82**, (*S*)-**110**, (*S*)-**112**] exhibited weak affinity for the dopamine  $D_2$  receptor, a strong 5-HT<sub>3</sub> receptor binding affinity was retained except (*S*)-**110**. Although there were marked differences in affinity for the dopamine  $D<sub>2</sub>$ receptor between the enantiomers examined, (*R*)-**110** showed the most potent affinity for both receptors (dopamine  $D_2$  receptor;  $18.8 \text{ nm}$ ,  $5-\text{HT}_3$  receptor;  $3.40 \text{ nm}$ ). Moreover, each enantiomer's affinity for both receptors was slightly lower than the dopamine  $D<sub>2</sub>$  receptor affinity of domperidone and the  $5-HT<sub>3</sub>$  receptor binding affinity of ondansetron, respectively, and was much stronger than the affinity of metoclopramide for both receptors (Table 6).

As for the inhibition of apomorphine-induced emesis in dogs, the results are presented in Table 7. (*R*)-**110** inhibited the emesis in dogs with an  $ED_{50}$  of 0.40 mg/kg,  $p.o.$  On the other hand, although (*R*)-**112**, at a dose of 1.0 mg/kg, *p.o.*, did not completely inhibit this emesis as compared with **82** (90% *vs.* 100% inhibition), it showed stronger inhibition than (*R*)- **110**. The inhibition percent caused by (*R*)-**81** and (*R*)-**82** at a dose of 1.0 mg/kg, *p.o.* was 87% and 100% ( $ED_{50} = 0.13$ )

mg/kg, *p.o.*), respectively. These results indicate that **82** and its  $(R)$ -enantiomer  $[(R)$ -82] have potent antagonistic activity for the dopamine D<sub>2</sub> receptor *in vivo* biological assay. Next,  $(R)$ -82 was evaluated for its 5-HT<sub>3</sub> receptor antagonistic activity *in vivo* by measuring its ability to inhibit the von Bezold–Jarisch reflex induced 2-methyl-5-HT in rats<sup>20)</sup>; the 5- $HT_3$  receptor antagonistic activity of  $(R)$ -82, 2x, ondanstron, and metoclopramide was characterized by an  $IC_{50} = 1.4$ , 0.86, 2.8, and 181  $\mu$ g/kg, i.v., respectively. From the results above it was, thus, concluded that  $(R)$ -82 exhibits a strong affinity for the dopamine  $D_2$  and the 5-HT<sub>3</sub> receptors and a potent antagonistic activity *in vivo* biological assay for the both receptors. In addition, (*R*)-**82**, as well as **2x** and **2y**, showed a weak binding affinity for the serotonin  $5-HT<sub>4</sub>$  receptor  $(>1000 \text{ nm})^{33}$  and did not display any serotonin 5-HT<sub>4</sub> receptor agonistic activity even at  $10^{-5}$  M in isolated guinea pig ileum preparations.

Benzamide derivatives with good affinity for the serotonin 5-HT<sub>4</sub> and 5-HT<sub>3</sub> or the serotonin 5-HT<sub>4</sub> and the dopamine  $D<sub>2</sub>$  receptors are well-known thus far.<sup>36)</sup> However, there has been no report on benzamides having a potent affinity for both the dopamine  $D_2$  and the 5-HT<sub>3</sub> receptors. In this study, we have shown that benzamide  $82$ , a novel dopamine  $D_2$ and  $5-\text{HT}_3$  receptors dual antagonist, is structurally related to the 1-benzyl-4-methyl and 1,4-dimethylhexahydro-1,4-diazepinylbenzamides **2x** and **76**, which have been described as potent and selective  $5-\text{HT}_3$  receptor antagonists. Moreover, the fact that (*R*)-**82**, (*S*)-**82**, and **2x** had marked structural similarities but different pharmacological profiles, was intriguing. Although the hexahydro-1,4-diazepine ring of **82** is to some extent conformationally restricted compared with the (diethylamino)ethyl side chain of metoclopramide, it still has some degree of conformational freedom. It is speculated that the 1-ethyl-4-methylhexahydro-1,4-diazepine ring of **82** adopts a markedly favorable conformation in the binding site of the dopamine  $D_2$  and the 5-HT<sub>3</sub> receptors where one or two basic nitrogen atoms and the ethyl and/or methyl groups occupy the correct orientation to block both receptors. The role of the substituents in the hexahydro-1,4-diazepine ring in the recognition of these compounds by the dopamine  $D<sub>2</sub>$ and the  $5-\text{HT}_3$  receptors and structural analysis of the differences in the pharmacological profile between each enantiomer are in progress.

In summary, replacement of the 1-benzyl group in the hexahydro-1,4-diazepine ring of **2x**, a potent and selective 5-  $HT_3$  receptor antagonist, by an ethyl group resulted in a remarkable increase in affinity for the dopamine  $D<sub>2</sub>$  receptor, while keeping a potent  $5-HT<sub>3</sub>$  receptor binding affinity. After modification of the 4-amino-5-chloro-2-methoxybenzoyl moiety of the 1-ethyl-4-methylhexahydro-1,4-diazepinylbenzamide **81**, the corresponding 4-methylaminobenzamide **82** was found to have strong antagonistic activity, both *in vitro* (binding affinity for the dopamine  $D_2$  and the 5-HT<sub>3</sub> receptors) and *in vivo* (inhibition of apomorphine-induced emesis and the von Bezold–Jarisch reflex induced 2-methyl-5-HT) tests. Optical resolution of **82** brought about a drastic change in the binding affinity for the dopamine  $D<sub>2</sub>$  receptor, particularly, the  $(R)$ -enantiomer  $[(R)$ -82] exhibited an affinity for the dopamine  $D_2$  receptor *ca*. 2-fold that of **82**, while  $(S)$ -82 was found to have a weaker affinity for this receptor. On the other hand, the binding affinity for the  $5-HT<sub>3</sub>$  receptor of each enantiomer  $[(R)$ -82,  $(S)$ -82] was *ca*. 2 times more potent than that of the racemate **82**. Although **82** and its enantiomers showed a significant difference in the binding affinity for the dopamine  $D_2$  and the 5-HT<sub>3</sub> receptors,  $(R)$ -82, which constitute a novel amine moiety of the 4-amino-5-chloro-2 methoxybenzamide family, exhibited a potent dual antagonistic activity for the dopamine  $D_2$  and the 5-HT<sub>3</sub> receptors and is expected to be a broad antiemetic agent.

#### **Experimental**

All melting points were determined on a Yanagimoto micromelting point apparatus without correction. IR spectra were recorded on a Shimadzu FTIR-8200PC spectrometer with KBr disks unless otherwise stated. Electron ionization and atmospheric pressure chemical ionization mass spectra were obtained on a JEOL JMS D-300 and Hitachi M-1000 spectrometer, respectively. <sup>1</sup>H-NMR spectra were recorded on a Varian Gemini-200 (200 MHz) or a JEOL JNM-LA300 (300 MHz) spectrometer using dilute solution in CDCl<sub>3</sub> unless otherwise stated. Chemical shifts were expressed as  $\delta$ (ppm) value from tetramethylsilane as an internal standard and coupling constants (*J*) are given in Hz. Optical rotations were measured at 589 nm with a Jasco P-1020 digital polarimeter. Organic extracts were dried over anhydrous MgSO<sub>4</sub> or anhydrous Na<sub>2</sub>SO<sub>4</sub>. The solvents were evaporated under reduced pressure. Flash chromatography was carried out on 60  $\mu$ m mesh silica gel (Fuji Silysia FL60D).

The following known 6-aminohexahydro-1,4-diazepines and benzoic acids were prepared according to the cited literature: 6-amino-1-benzyl-4 methylhexahydro-1,4-diazepine23,37) (**8m**), 6-amino-1,4-dimethylhexahydro-1,4-diazepine23) (**8n**), 6-amino-1,4-diethylhexahydro-1,4-diazepine20) (**8o**), 4 amino-5-chloro-2-ethoxybenzoic acid28,29) (**23b**), 4-amino-5-chloro-2-propoxybenzoic acid<sup>20)</sup> (23g), 4-amino-5-chloro-2-(isopropoxy)benzoic acid<sup>20)</sup> (23c), 5-iodo-2-methoxybenzoic acid<sup>38)</sup> (54), 3-bromo-2,6-dimethoxybenzoic acid<sup>39)</sup> (**55**), 3,5-dibromo-2,6-dimethoxybenzoic acid39) (**56**), 4-chloro-2-methoxy-5-nitrobenzoic acid40) (**57**), 3-bromo-5-chloro-2-hydroxy-6-methoxybenzoic acid<sup>41)</sup> (58), 5-ethylthio-2-methoxybenzoic acid<sup>42)</sup> (59), 5-ethylsulfonyl-2methoxybenzoic acid<sup>43)</sup> (60), and 2-methoxy-5-sulfamoylbenzoic acid<sup>44)</sup> (**61**). Methyl 4-acetylamino-5-chloro-2-methoxybenzoate (**22a**), 4-amino-5 chloro-2-methoxybenzoic acid (**23a**), 2,4,5-trifluorobenzoic acid (**63**), and 4 chloro-2-methoxybenzoic acid (**67**) are commercially available.

*N***-Ethyl-***N*<sup> $\prime$ </sup>**-methylethylenediamine (5a)** SOCl<sub>2</sub> (267 g, 2.2 mol) was added dropwise to a solution of 2-(ethylamino)ethanol (**3m**, 100 g, 1.1 mol) in CHCl<sub>3</sub> (845 ml) kept at *ca.*  $0^{\circ}$ C. The mixture was heated to reflux for 6 h and cooled to room temperature. The reaction mixture was poured into  $Et<sub>2</sub>O$ (*ca.* 1000 ml), and the whole was stirred at room temperature for 15 h. In order to decompose the excess of SOCl<sub>2</sub>, excess EtOH was added slowly to the solution at *ca.* 0°C. The resulting powder was collected by filtration, washed with Et<sub>2</sub>O, and dried to give  $150 g (93%)$  of *N*-(2-chloroethyl)-*N*ethylamine hydrochloride (**4m**), which was used in the next step without further purification. **4m** (100 g, 0.69 mol) was added portionwise to a mixture of 40% aqueous MeNH<sub>2</sub> (700 ml), EtOH (150 ml), and H<sub>2</sub>O (600 ml) kept at 10 °C. The mixture was heated at 50 °C for 16 h and cooled to room temperature. After addition of a large amount of solid  $K_2CO_3$ , the upper layer was separated. EtOH was evaporated under atmospheric pressure, and the residue was dissolved in CH<sub>2</sub>Cl<sub>2</sub>. The organic solution was dried over anhydrous  $Na<sub>2</sub>SO<sub>4</sub>$  and evaporated at *ca.* 35 °C to leave a pale brown oil, which was distilled to give 50.0 g (71%) of **5a** as a colorless oil, bp 30— 31 °C/18 mmHg. <sup>1</sup>H-NMR δ: 1.12 (3H, t, J=7), 2.32 (2H, br s, NH×2), 2.43  $(3H, s)$ , 2.66  $(2H, q, J=7)$ , 2.69–2.76  $(4H, m, NCH, CH, NH)$ . MS:  $m/z$ , 102  $(M^+)$ .

*N***-Ethyl-***N*<sup> $\prime$ </sup>**-propylethylenediamine (5e)** In a similar manner to that described above, crude **5e** was obtained by the reaction of **4m** (24.0 g, 0.17 mol) with PrNH<sub>2</sub> (150 ml, 1.8 mol) in EtOH (20 ml)–H<sub>2</sub>O (100 ml) mixture in 35% yield as a pale yellow oil. Crude **5e** was used in the next step without distillation. <sup>1</sup>H-NMR  $\delta$ : 0.92 (3H, t, *J*=7.5), 1.09 (3H, t, *J*=7.5), 1.48 (2H, sex, *J*=7.5), 1.63 (2H, br s), 2.55 (2H, q, *J*=7.5), 2.65 (2H, m), 2.98 (4H, m). MS:  $m/z$ , 130 (M<sup>+</sup>).

*N***-Cyclopropyl-***N*9**-ethylethylenediamine (5g)** In a similar manner to that described above, **5g** was obtained by the reaction of **4m** (50.0 g, 0.35 mol) with cyclopropylamine (240 ml, 3.5 mol) in EtOH (400 ml) in 43% yield as a colorless oil, bp  $38-40$  °C/*ca*. 20 mmHg. <sup>1</sup>H-NMR  $\delta$ : 0.25-0.5 (4H, m), 1.11 (3H, t, J=7), 2.0 (2H, br s), 2.1 (1H, m), 2.65 (2H, q, J=7), 2.65—2.85 (4H, m). MS:  $m/z$ , 128 (M<sup>+</sup>).

*N***-Methyl-***N***<sup>** $\prime$ **</sup>-propylethylenediamine (5b)** SOCl<sub>2</sub> (60 ml, 0.82 mol)

was added dropwise to a solution of 2-(propylamino)ethanol (**3n**, 41.0 g, 0.40 mol) in CHCl<sub>3</sub> (300 ml) kept at *ca*. 0 °C. The mixture was heated to reflux for 6 h and cooled to room temperature. The reaction mixture was concentrated to dryness. A mixture of  ${}^{i}PfOH$  and  $Et<sub>2</sub>O$  was added carefully to the solid residue at *ca.* 0 °C. The resulting powder was collected by filtration, washed with Et<sub>2</sub>O, and dried to give 20.3 g (32%) of  $N$ -(2-chloroethyl)-*N*-propylamine hydrochloride (**4n**) as a white powder, which was used in the next step without further purification. In a similar manner to that described above, crude **5b** was obtained from the reaction of **4n** (20.0 g, 0.13 mol) with 40% aqueous MeNH<sub>2</sub> (130 ml) in EtOH (120 ml) in 45% yield as a colorless oil and used in the next step without distillation. <sup>1</sup>H-NMR  $\delta$ : 0.92 (3H, t, *J*=7), 1.48 (2H, sex, *J*=7), 1.80 (2H, br s), 2.44 (3H, s), 2.58 (2H, q, *J*=7), 2.71 (4H, m). MS:  $m/z$ , 116 (M<sup>+</sup>).

*N***-Ethyl-***N***<sup>** $\prime$ **</sup>-isopropylethylenediamine (5f)** SOCl<sub>2</sub> (182 g, 1.5 mol) was added dropwise to a solution of 2-(isopropylamino)ethanol (**3o**, 75.0 g, 0.73 mol) in CHCl<sub>3</sub> (550 ml) kept at *ca*. 0 °C. The mixture was heated to reflux for 6 h and cooled to room temperature. After addition of  $Et<sub>2</sub>O$ , the mixture was stood overnight at room temperature. The resulting powder was collected by filtration, washed with Et<sub>2</sub>O, and dried to give 114 g (99%) of *N*-(2-chloroethyl)-*N*-isopropylamine hydrochloride (**4o**) as a hygroscopic brown powder, which was used in the next step without further purification. In a similar manner to that described above, **5f** was obtained from the reaction of  $40$  (57.0 g, 0.36 mol) with  $70\%$  aqueous EtNH<sub>2</sub> (340 ml) in EtOH (350 ml) in 26% yield as a colorless oil, bp 41—43 °C/5 mmHg. <sup>1</sup>H-NMR  $\delta$ : 1.06 (6H, d, *J*=6.5), 1.11 (3H, t, *J*=7), 1.37 (2H, br s), 2.62 (2H, q, *J*=7), 2.72 (4H, m), 2.79 (1H, hep,  $J=6.5$ ). MS:  $m/z$ , 130 (M<sup>+</sup>).

 $N$ **-Isopropyl-** $N'$ **-methylethylenediamine (5c)** In a similar manner to that described above, **5c** was obtained by the reaction of **4o** (57.0 g, 0.37 mol) with 40% aqueous MeNH<sub>2</sub> (360 ml) in EtOH (350 ml) in 29% yield as a colorless oil. <sup>1</sup>H-NMR  $\delta$ : 1.17 (6H, d, J=6.5), 2.34 (2H, br s), 2.43 (3H, s), 2.66–2.74 (4H, m), 2.79 (1H, hep,  $J=6.5$ ). MS:  $m/z$ , 116  $(M^+).$ 

*N***-Cyclopropyl-***N*<sup> $\prime$ </sup>**-methylethylenediamine (5d)** SOCl<sub>2</sub> (332.5 g, 2.8) mol) was added dropwise to a solution of 2-(methylamino)ethanol (**3l**, 100 g, 1.3 mol) in CHCl<sub>3</sub> (1000 ml) kept at *ca*. 0 °C. The mixture was heated to reflux for 6h and cooled to room temperature. The reaction mixture was poured into Et<sub>2</sub>O (*ca.* 1500 ml) and stirred at room temperature for 15 h. In order to decompose the excess of SOCl<sub>2</sub>, a large amount of EtOH was added carefully to the solution at *ca.* 0 °C. The resulting powder was collected by filtration, washed with Et<sub>2</sub>O, and dried to give  $165 g$  (95%) of *N*-(2chloroethyl)-*N*-methylamine hydrochloride (**4l**), which was used in the next step without further purification. **4l** (53.0 g, 0.41 mol) was added portionwise to a mixture of cyclopropylamine (282 ml, 4.1 mol) and EtOH (470 ml) kept below 10 °C. The mixture was heated at 50 °C for 16 h and cooled to room temperature. After addition of a large amount of solid  $K<sub>2</sub>CO<sub>3</sub>$ , the upper layer was separated. EtOH was evaporated under atmospheric pressure, and the resultant residue was dissolved in CH<sub>2</sub>Cl<sub>2</sub>. The organic solution was dried over anhydrous  $Na<sub>2</sub>SO<sub>4</sub>$  and evaporated at *ca*. 35 °C to leave a pale brown oil, which was distilled to give 25.4 g (55%) of **5d** as a colorless oil, bp 35—38 °C/*ca*. 20 mmHg. <sup>1</sup>H-NMR δ: 0.25—0.5 (4H, m), 2.0 (2H, br s), 2.1 (1H, m), 2.42 (3H, s), 2.64—2.85 (4H, m). MS:  $m/z$ , 114 (M<sup>+</sup>).

**6-Amino-1-ethyl-4-methylhexahydro-1,4-diazepine (8a)** Our previously described procedure<sup>22)</sup> was adopted. 1) To a solution of 2-benzyloxycarbonylaminopropenal<sup>23)</sup> (6, 3.8 g, 19 mmol) in MeOH (50 ml) was added **5a** (3.8 g, 37 mmol) at *ca.*  $5^{\circ}$ C. After being stirred for 3 h, NaBH<sub>4</sub> (2.8 g, 74 mmol) was added portionwise. The mixture was stirred at room temperature for 16 h and then concentrated to dryness. The residue was taken into  $CHCl<sub>3</sub>$  and  $H<sub>2</sub>O$ . The organic layer was separated, washed with brine, and dried over anhydrous  $MgSO<sub>4</sub>$ . The solvent was evaporated, and the residue was chromatographed on silica gel with  $CHCl<sub>3</sub>/MeOH=10/1$  to give 2.5 g (46% from **6**) of 6-benzyloxycarbonylamino-1-ethyl-4-methylhexahydro-1,4-diazepine (7a) as an oil. <sup>1</sup>H-NMR δ: 1.10 (3H, t, *J*=7), 2.23 (3H, s), 2.48—3.10 (10H, m), 3.97 (1H, m, 6-H), 5.10 (2H, s), 6.12 (1H, m, NH), 7.25—7.42 (5H, m). MS:  $m/z$ , 292 (MH<sup>+</sup>), 113, 91. IR (neat)  $v \text{ cm}^{-1}$ : 3060, 3030, 1725, 1680, 1630. 2) A mixture of **7a** (2.1 g, 7.2 mmol) and 25% HBr in AcOH (15 ml) was stirred at room temperature for 1 h. The reaction mixture was concentrated, and the residue was dissolved in a small volume of H<sub>2</sub>O. The aqueous solution was washed with Et<sub>2</sub>O and basified with  $48\%$ aqueous NaOH. After addition of solid  $K_2CO_3$ , the mixture was extracted with CHCl<sub>3</sub>. The extract was concentrated to dryness to give  $1.0 \text{ g}$  (88%) of **8a** as an oil. <sup>1</sup>H-NMR  $\delta$ : 1.04 (3H, t, *J*=7), 1.46 (2H, br s), 2.37 (3H, s), 2.34—2.9 (10H, m), 3.05 (1H, m, 6-H). MS:  $m/z$ , 158 (MH<sup>+</sup>).

**6-Amino-1-methyl-4-propylhexahydro-1,4-diazepine (8b)** In a similar manner to that described above, **8b** was prepared from **5b** and **6** *via* 6-benzyloxycarbonylamino-1-methyl-4-propylhexahydro-1,4-diazepine (**7b**) in 32% overall yield. **7b**; <sup>1</sup> H-NMR d: 0.87 (3H, t, *J*57), 1.49 (2H, sex, *J*57), 2.24—3.06 (10H, m), 2.42 (3H, s), 3.93 (1H, m, 6-H), 5.10 (2H, s), 6.07 (1H, br, NH), 7.25–7.46 (5H, m). MS:  $m/z$ , 306 (MH<sup>+</sup>), 91.

**6-Amino-1-cyclopropyl-4-methylhexahydro-1,4-diazepine (8d)** In a similar manner to that described above, **8d** was prepared from **5d** and **6** *via* 6-benzyloxycarbonylamino-1-cylcopropyl-4-methylhexahydro-1,4-diazepine (**7d**) in 5% overall yield. **7d**; <sup>1</sup>H-NMR δ: 0.2—0.5 (4H, m), 1.9 (1H, m), 2.43 (3H, s), 2.35—3.1 (8H, m), 3.82 (1H, m), 5.10 (2H, s), 5.73 (1H, br), 7.2—7.4 (5H, m). MS:  $m/z$ , 304 (MH<sup>+</sup>), 91.

**6-Amino-1-ethyl-4-propylhexahydro-1,4-diazepine (8e)** In a similar manner to that described above, **8e** was prepared from **5e** and **6** *via* 6-benzyloxycarbonylamino-1-ethyl-4-propylhexahydro-1,4-diazepine (**7e**) in 30% overall yield. **7e**; <sup>1</sup> H-NMR d: 0.86 (3H, t, *J*57), 1.02 (3H, t, *J*57), 1.30— 1.56 (2H, m), 2.35—2.93 (12H, m), 3.85 (1H, m), 5.09 (2H, s), 5.90 (1H, br),  $7.23 - 7.43$  (5H, m). MS:  $m/z$ , 320 (MH<sup>+</sup>), 91.

**6-Amino-1-cyclopropyl-4-ethylhexahydro-1,4-diazepine (8g)** In a similar manner to that described above, **8g** was prepared from **5g** and **6** *via* 6 benzyloxycarbonylamino-1-cyclopropyl-4-ethylhexahydro-1,4-diazepine (**7g**) in 5% overall yield. **7g**; <sup>1</sup> H-NMR d: 0.2—0.6 (4H, m), 1.02 (3H, t, *J*57), 1.92 (1H, m), 2.35—3.1 (10H, m), 4.05 (1H, m, 6-H), 5.10 (2H, s), 5.67 (1H, br), 7.25—7.4 (5H, m). MS:  $m/z$ , 318 (MH<sup>+</sup>), 91.

**6-Amino-1-ethyl-4-isopropylhexahydro-1,4-diazepine (8f)** 1) **5f** (6.0 g, 46 mmol) was added to a solution of trishydroxymethylnitromethane (7.4 g, 49 mmol) and NaHCO<sub>3</sub> (2.5 g) in H<sub>2</sub>O (50 ml) at room temperature. The mixture was stirred at 40—50 °C for 2 h and cooled to *ca.* 10 °C. The resulting oil was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The extract was washed with brine and evaporated at *ca.* 25 °C. The oily residue containing 1-ethyl-6-hydroxymethyl-4-isopropyl-6-nitrohexahydro-1,4-diazepine (**9f**) was dissolved in MeOH, and potassium *tert*-butoxide (5.8 g, 52 mmol) was added portionwise below 50 °C. The mixture was stirred at the same temperature for 0.5 h and concentrated to dryness at 40—50 °C. After slow addition of a solution of NH<sub>2</sub>OH · HCl (95%, 3.6 g, 49 mmol) in H<sub>2</sub>O (24 ml), the mixture was extracted with  $CH_2Cl_2$ . The extract was washed with brine and evaporated. The oily residue was chromatographed on silica gel with AcOEt to give 4.3 g (43% from **5f**) of 1-ethyl-4-isopropyl-6-nitrohexahydro-1,4-diazepine (**10f**) as a pale yellow oil. <sup>1</sup>H-NMR  $\delta$ : 0.97, 1.00 (each 3H, d, J=6.3), 1.04 (3H, t, *J*=7.0), 2.54—2.74 (6H, m), 2.91 (1H, hep, *J*=6.3), 3.14 (1H, dd, *J*=6.0, 10.4), 3.16 (1H, dd, *J*=5.8, 14.0), 3.27 (1H, dd, *J*=6.5, 14.0), 3.34 (1H, dd, *J*=6.0, 14.0), 4.54 (1H, quin, *J*=6.5, 6-H). MS:  $m/z$ , 216 (MH<sup>+</sup>). 2) A solution of **10f** (4.3 g, 20 mmol) in EtOH (20 ml) was hydrogenated over Raney Ni (*ca.* 1 g) at room temperature for 3 h. The catalyst was removed by Celite filtration, and the filtrate was concentrated. The residue was dissolved in  $CHCl<sub>3</sub>$ , and the solution was dried over anhydrous  $MgSO<sub>4</sub>$ . The solvent was evaporated to give 3.2 g (86%) of **8f** as a pale yellow oil, which was used in the next step without further purification.

**6-Amino-1-isopropyl-4-methylhexahydro-1,4-diazepine (8c)** In a similar manner to that described above, **8c** was prepared from trishydroxymethylnitromethane and **5c** *via* **9c** and 1-isopropyl-4-methyl-6-nitrohexahydro-1,4-diazepine (**10c**) in 7% overall yield. **10c** (pale yellow oil); <sup>1</sup> H-NMR  $\delta$ : 0.97, 1.02 (each 3H, d, J=6.3), 2.44 (3H, s), 2.5–2.75 (4H, m), 2.90  $(1H, \text{ hep}, J=6.3), 2.99 (1H, \text{dd}, J=6.0, 12.0), 3.12 (1H, \text{dd}, J=5.5, 8.5),$ 3.20 (1H, dd, *J*=6.0, 6.0), 3.30 (1H, dd, *J*=6.0, 13.5), 4.58 (1H, quin, *J*=6, 6-H). MS:  $m/z$ , 202 (MH<sup>+</sup>).

**6-Amino-1-butyl-4-ethylhexahydro-1,4-diazepine (8h)** 1) The method of Olofson *et al.*45,46) was applied. A solution of 6-acetylamino-1-benzyl-4 ethylhexahydro-1,4-diazepine<sup>20)</sup> (11, 20.1 g, 73 mmol) and 1-chloroethyl chloroformate (12.6 g, 88 mmol) in 1,2-dichloroethane (200 ml) was heated to reflux for 1 h and cooled to room temperature. After evaporation of the solvent, the residue was dissolved in MeOH (200 ml). The solution was heated to reflux for 1 h and cooled to room temperature. The solvent was evaporated, and the residue was dissolved in a small volume of H2O. The aqueous solution was basified with 48% aqueous NaOH. After addition of solid  $K_2CO_3$ , the mixture was extracted with CHCl<sub>3</sub>. The extract was concentrated to dryness, and the oily residue was chromatographed on silica gel with CHCl<sub>3</sub>/MeOH=15/1 to give 12.3 g (91%) of 6-acetylamino-1-ethylhexahydro-1,4-diazepine (12) as an oil. <sup>1</sup>H-NMR  $\delta$ : 1.05 (3H, t, *J*=7.0), 2.00 (3H, s), 2.40 (1H, br s), 2.45—3.2 (10H, m), 4.09 (1H, m), 6.80 (1H, br). MS:  $m/z$ , 186 (MH<sup>+</sup>). 2) 12 (2.0 g, 11 mmol) and NaBH<sub>4</sub> (2.1 g, 56 mmol) were added successively to a solution of butyric acid (15.7 g, 0.18 mol) in anhydrous toluene (27 ml) below 20 °C. The mixture was heated to reflux for 3 h and cooled to room temperature. The reaction mixture was washed with  $2 \text{ N}$  aqueous NaOH and dried over anhydrous MgSO<sub>4</sub>. The solvent was evaporated, and the residue was chromatographed on silica gel with

 $CHCl<sub>3</sub>/MeOH=10/1$  to give 2.0 g (77%) of 6-acetylamino-1-butyl-4-ethylhexahydro-1,4-diazepine (13) as a pale yellow oil. <sup>1</sup>H-NMR  $\delta$ : 0.96 (3H, t, *J*=7.0), 1.10 (3H, t, *J*=7.0), 1.18—1.55 (4H, m), 2.00 (3H, s), 2.5—3.1  $(12H, m)$ , 4.27 (1H, m), 7.82 (1H, br d,  $J=8.0$ ). MS:  $m/z$ , 242 (MH<sup>+</sup>). 3) A solution of **13** (2.0 g, 8.3 mmol) in 10% aqueous HCl (20 ml) was heated to reflux for 4 h and cooled to room temperature. The reaction mixture was washed with Et<sub>2</sub>O and basified with 48% aqueous NaOH. After addition of solid  $K_2CO_3$ , the mixture was extracted with CHCl<sub>3</sub>. The extract was evaporated to give 1.2 g (73%) of **8h** as a pale yellow oil. <sup>1</sup>H-NMR  $\delta$ : 0.9 (3H, t, *J*=7.0), 1.05 (3H, t, *J*=7.0), 1.2—1.55 (4H, m), 1.75 (2H, s), 2.4—2.9  $(12H, m)$ , 3.03 (1H, m). MS:  $m/z$ , 200 (MH<sup>+</sup>).

**6-Amino-1-methyl-4-(2,2,2-trifluoroethyl)hexahydro-1,4-diazepine (8i)** 1) A solution of **14**22) (24.3 g, 75 mmol) in EtOH (400 ml)–AcOH (40 ml) mixture was hydrogenated over 10% Pd/C (3.5 g) at *ca.* 50 °C under atmospheric pressure. After no further change in the pressure of  $H_2$  was observed, the solution was cooled to room temperature. The catalyst was filtered through Celite, the filtrate was concentrated, and the oily residue was dissolved in CHCl<sub>2</sub>. The solution was washed successively with 10% aqueous NaHCO<sub>3</sub>, brine, and dried over anhydrous MgSO<sub>4</sub>. The solvent was evaporated to give 16.3 g (93%) of *N*-(1-methylhexahydro-1,4-diazepin-6-yl)benzamide (15) as a colorless oil. <sup>1</sup>H-NMR δ: 2.47 (3H, s), 2.3—3.2 (9H, m), 4.34 (1H, m, 6-H), 7.2—7.6 (4H, m), 7.75—7.95 (2H, m). MS: *m*/*z*, 234  $(MH^{+})$ . 2) NaBH<sub>4</sub> (2.1 g, 56 mmol) and **15** (2.2 g, 9.4 mmol) were added successively to a solution of  $CF_3CO_2H$  (21.1 g, 0.19 mol) in toluene (125 ml) below 20 °C. The mixture was heated to reflux for 3 h and cooled to room temperature. The reaction mixture was then washed with 2 N aqueous NaOH and concentrated to dryness. The residue was chromatographed on silica gel with CHCl<sub>3</sub>/MeOH=10/1 to give 2.0 g (67%) of *N*-[1-methyl-4-(2,2,2-trifluoroethyl)hexahydro-1,4-diazepin-6-yl]benzamide (**16i**) as a pale yellow oil. <sup>1</sup>H-NMR  $\delta$ : 2.42 (3H, s), 2.4–3.5 (8H, m), 3.19 (2H, q, J=9.5, CH<sub>2</sub>CF<sub>3</sub>), 4.33 (1H, m, 6-H), 7.35—7.57 (4H, m), 7.77—7.87 (2H, m). MS: *m*/*z*, 316 (MH<sup>+</sup>). 3) A solution of **16i** (1.5 g, 4.8 mmol) in 35% aqueous HCl (10 ml) was heated to reflux for 8 h and cooled to room temperature. The reaction mixture was washed with  $Et<sub>2</sub>O$  and basified with 48% aqueous NaOH. After addition of solid  $K_2CO_3$ , the mixture was extracted with CHCl<sub>3</sub>. The extract was evaporated to give 1.0 g (quantitative yield) of **8i** as a pale yellow oil.

**6-Amino-1-(2,2-difluoroethyl)-4-methylhexahydro-1,4-diazepine (8j)** In a similar manner to that described above, the reaction of **15** with  $CF<sub>2</sub>HCO<sub>2</sub>H$  and NaBH<sub>4</sub> gave *N*-[1-(2,2-difluoroethyl)-4-methylhexahydro-1,4-diazepin-6-yl]benzamide (16j) in 63% yield. <sup>1</sup>H-NMR  $\delta$ : 2.44 (3H, s), 2.4—3.55 (10H, m), 4.35 (1H, m, 6-H), 5.81 (1H, tt, *J*=5.0, 56, CHF<sub>2</sub>), 7.3—7.7 (4H, m), 7.75—7.9 (2H, m). MS:  $m/z$ , 298 (MH<sup>+</sup>). In a similar manner to that described above, **16j** was converted into the corresponding amine **8j** in quantitative yield as an oil.

**6-Amino-1-(2-methoxyethyl)-4-methylhexahydro-1,4-diazepine (8k)** In a similar manner to that described above, the reaction of **15** with 2 methoxyacetic acid and NaBH4 gave *N*-[1-(2-methoxyethyl)-4-methylhexahydro-1,4-diazepin-6-yl]benzamide (16k) in  $60\%$  yield. <sup>1</sup>H-NMR  $\delta$ : 2.42 (3H, s), 2.5—3.1 (10H, m), 3.22 (3H, s), 3.36—3.52 (2H, m), 4.31 (1H, m, 6-H), 7.35—7.55 (3H, m), 7.7—8.0 (3H, m). MS:  $m/z$ , 292 (MH<sup>+</sup>). In a similar manner to that described above, **16k** was converted into the corresponding amine **8k** in quantitative yield as an oil.

**6-Amino-1-(3-hydroxypropyl)-4-methylhexahydro-1,4-diazepine (8l)** 1) Acrolein (90%, 3.6 g, 58 mmol) was added to a solution of **15** (5.0 g, 21 mmol) and Et<sub>3</sub>N (6.5 g, 64 mmol) in MeOH (50 ml) at *ca.* 0 °C. The mixture was stirred at this temperature for 2 h. Then,  $NabH_4$  (4.9 g, 0.13 mol) was added portionwise at *ca*. 0 °C, and the mixture was stirred at this temperature for 1 h and at room temperature overnight. After evaporation of the solvent, the residue was taken into  $H<sub>2</sub>O$  and CHCl<sub>3</sub>. The organic layer was separated, washed with brine, and dried over anhydrous MgSO<sub>4</sub>. The solvent was evaporated, and the oily residue was chromatographed on silica gel with  $CHCl<sub>3</sub>/MeOH = 10/1$  to give 4.2 g (67%) of *N*-[1-(3-hydroxypropyl)-4methylhexahydro-1,4-diazepin-6-yl]benzamide (**16l**) as a pale yellow oil. <sup>1</sup>H-NMR  $\delta$ : 1.5–1.9 (2H, m), 2.40 (3H, s), 2.45–3.1 (10H, m), 3.85 (2H, t, *J*=7), 4.40 (1H, m), 5.48 (1H, br s), 7.35—7.55 (3H, m), 7.63 (1H, d, *J*=8), 7.85—7.95 (2H, m). MS:  $m/z$ , 292 (MH<sup>+</sup>). 2) A mixture of 161 (2.0 g, 6.9 mmol), Ac<sub>2</sub>O (1.4 g, 14 mmol), pyridine (2 drops), and CHCl<sub>3</sub> (20 ml) was heated to reflux for 5 h and cooled to room temperature. The reaction mixture was washed successively with  $H_2O$  and brine and dried over anhydrous MgSO4. After evaporation of the solvent, the oily residue was chromatographed on silica gel with CHCl<sub>3</sub>/MeOH=10/1 to give 1.4 g (61%) of *N*-[1-(3-acetoxypropyl)-4-methylhexahydro-1,4-diazepin-6-yl]benzamide (**17**) as a pale yellow oil. <sup>1</sup>H-NMR  $\delta$ : 1.79 (2H, quin, *J*=7), 2.05 (3H, s), 2.55 (3H, s), 2.64 (2H, t, *J*=7), 2.5–3.02 (8H, m), 4.12 (2H, t, *J*=7), 4.50  $(1H, m)$ , 7.38—7.55 (3H, m), 7.85—8.1 (3H, m). MS:  $m/z$ , 334 (MH<sup>+</sup>), 3) A solution of **17** (1.2 g, 3.6 mmol) in 35% aqueous HCl (10 ml) was heated to reflux for 8 h and cooled to room temperature. Following work-up similar to that described above, **8l** (0.65 g, 96% yield) was obtained as a pale yellow oil. <sup>1</sup>H-NMR δ: 1.68 (2H, quin, *J*=7), 2.37 (3H, s), 2.35—3.0 (12H, m), 3.15 (1H, m), 3.82 (2H, t,  $J=7$ ). MS:  $m/z$ , 188 (MH<sup>+</sup>).

**(***R***)- and (***S***)-6-Amino-1-ethyl-4-methylhexahydro-1,4-diazepines [(***R***) and (***S***)-8a]** 1) A solution of (*R*)-*N*-[1-methyl-4-(3-methyl)benzylhexahydro-1,4-diazepin-6-yl]-1*H*-indazole-3-carboxamide<sup>25)</sup> [(*R*)-18, >99.5% ee, 28.8 g, 76 mmol] in EtOH (500 ml) was hydrogenated over 10% Pd/C (4.3 g) at room temperature under  $H_2$  atmosphere. After no further change in the pressure of  $H<sub>2</sub>$  was observed, the catalyst was filtered through Celite. The filtrate was concentrated to dryness to give 20.9 g (quantitative yield) of  $(R)$ -*N*-<br>(1-methylhexahydro-1.4-diazepin-6-vl)-1*H*-indazole-3-carboxamide [*R*}-(1-methylhexahydro-1,4-diazepin-6-yl)-1*H*-indazole-3-carboxamide [(*R*)- **19**] as a colorless oil. <sup>1</sup>H-NMR  $\delta$ : 2.25 (1H, br s), 2.50 (3H, s), 2.6 (1H, m), 2.8—3.25 (6H, m), 3.45 (1H, d, *J*=7.5, 14.0), 4.62 (1H, m), 7.2—7.45 (4H, m), 8.42 (1H, d, J=9.0), 8.95 (1H, m). MS:  $m/z$ , 274 (MH<sup>+</sup>). An analytical sample of  $(R)$ -19 was obtained by crystallization of its hydrochloride from EtOH, mp 241—245 °C. *Anal*. Calcd for  $C_{14}H_{19}N_5O \cdot 2HCl \cdot 1/4H_2O$ : C, 47.94; H, 6.18; Cl, 20.22; N, 19.97. Found: C, 47.74; H, 6.22; Cl, 20.44; N, 19.88.  $[\alpha]_D^{29}$  –3.0° (*c*=1.0, MeOH). 2) 80% Acetaldehyde (8.4 g, 0.15 mol) was added to a solution of  $(R)$ -19 (20.9 g) and Et<sub>3</sub>N (16.0 g, 0.16 mol) in MeOH (500 ml) under ice-cooling. The mixture was stirred at the same temperature for 2 h. Then,  $NaBH<sub>4</sub>$  (3.0 g, 79 mmol) was added portionwise to the reaction mixture at this temperature. The whole was warmed to room temperature, stirred for 16 h, and concentrated to dryness. The residue was taken into  $H_2O$  and  $CHCl<sub>3</sub>$ , and the organic layer was separated, washed with brine, and dried over anhydrous  $MgSO<sub>4</sub>$ . The solvent was evaporated, and the residue was chromatographed on silica gel with  $CHCl<sub>3</sub>/MeOH=9/1$  to give 22.9 g (quantitative yield) of (*R*)-*N*-[1-ethyl-4-methylhexahydro-1,4-diazepin-6-yl]-1H-indazole-3-carboxamide [(R)-20] as a colorless oil. <sup>1</sup>H-NMR δ: 1.06 (3H, t, *J*=7.5), 2.52 (3H, s), 2.5–3.6 (10H, m), 4.61 (1H, m), 7.2—7.45 (3H, m), 8.44 (1H, d, *J*=8.0), 9.05 (1H, d, *J*=10), 13.20 (1H, s). MS:  $m/z$ , 302 (MH<sup>+</sup>). 3) A solution of (*R*)-20 (21.2 g, 70 mmol) in 35% aqueous HCl (70 ml) was heated to reflux for 8 h and cooled to *ca.* 5 °C. The resulting solid of 1*H*-indazole-3-carboxylic acid was filtered off and the filtrate was basified with 48% aqueous NaOH. After addition of solid  $K_2CO_3$ , the mixture was extracted with CHCl<sub>3</sub>. The extract was evaporated to give *ca.* 11 g (quantitative yield) of (*R*)-**8a** as a pale yellow oil, which was used in the next step without further purification. This compound was identical to the sample obtained in an alternative preparation,<sup>47)</sup> on the basis of <sup>1</sup>H-NMR, MS, and the retention time of chiral HPLC comparison.

In a similar manner to that described for the conversion of (*R*)-**18** to (*R*)- **8a**,  $(S)$ -8a was prepared from  $(S)$ -18<sup>,25)</sup> the hydrochloride of  $(S)$ -19; mp 241—245 °C (EtOH). *Anal*. Calcd for C<sub>14</sub>H<sub>19</sub>N<sub>5</sub>O·2HCl·1/4H<sub>2</sub>O: C, 47.94; H, 6.18; Cl, 20.22; N, 19.97. Found: C, 48.14; H, 6.29; Cl, 19.95; N, 19.92.  $[\alpha]_D^{29}$  +3.3° (*c*=1.0, MeOH).

**2-Allyloxy-4-amino-5-chlorobenzoic Acid (23d)** 1) A mixture of methyl 4-acetylamino-5-chloro-2-hydoxybenzoate<sup>27)</sup> (21, 4.9 g, 20 mmol),  $K_2CO_3$  (4.2 g, 30 mmol), KI (0.5 g), tetrabutylammonium bromide (0.3 g, 0.93 mmol), allyl bromide (2.4 g, 20 mmol), and methyl ethyl ketone (100 ml) was heated to reflux for 9 h and cooled to room temperature. The solvent was evaporated, and the residue was dissolved in  $H_2O$  and  $CHCl<sub>3</sub>$ . The organic layer was separated, washed with brine, and dried over anhydrous  $MgSO<sub>4</sub>$ . The solvent was evaporated, and the oily residue was solidified on standing at room temperature. The solid was triturated with EtOH/Et<sub>2</sub>O to give  $1.9g$  (33%) of methyl 4-acetylamino-2-allyloxy-5chlorobenzoate (22d) as a colorless crystal, mp 110.5—112 °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 2.16 (3H, s), 3.78 (3H, s), 4.58 (2H, d, *J*=3), 5.27 (1H, d, *J*=11), 5.52 (1H, d, *J*=16), 6.02 (1H, m), 7.76 (1H, s), 7.83 (1H, s), 9.57 (1H, s). *Anal*. Calcd for C<sub>13</sub>H<sub>14</sub>ClNO<sub>4</sub>: C, 55.04; H, 4.97; Cl, 12.50; N, 4.94. Found: C, 54.78; H, 5.04; Cl, 12.43; N, 4.83. MS:  $m/z$ , 284 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3323, 1728, 1693, 1678, 1601, 1583, 1404, 1236. 2) A mixture of **22d**  $(1.8 \text{ g}, 6.3 \text{ mmol})$ , NaOH  $(2.0 \text{ g}, 50 \text{ mmol})$ , MeOH  $(10 \text{ ml})$ , and H<sub>2</sub>O  $(15 \text{ ml})$ was heated to reflux for 4 h and cooled to *ca.* 5 °C. After acidification with 35% aqueous HCl, the precipitates were collected by filtration and recrystallized from EtOH/hexane to give 1.0 g (69%) of **23d** as a colorless crystal, mp 142—143 °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 4.50 (2H, s), 5.25 (1H, d, *J*=10.6), 5.51 (1H, d, *J*=17.2), 5.95—6.1 (3H, m), 6.42 (1H, s), 7.60 (1H, s), 11.90 (1H, s). *Anal.* Calcd for C<sub>10</sub>H<sub>10</sub>ClNO<sub>3</sub>: C, 52.76; H, 4.43; Cl, 15.57; N, 6.15. Found: C, 52.71; H, 4.47; Cl, 15.54; N, 6.07. MS:  $m/z$ , 228 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3474, 3327, 3254, 1705, 1622, 1443.

**4-Amino-2-benzyloxy-5-chlorobenzoic Acid (23e)** In a similar manner to that described above, 1) **22e** (colorless crystal) was prepared from **21** and

benzyl bromide in 31% yield, mp  $125-126$  °C (triturated with EtOH/Et<sub>2</sub>O). <sup>1</sup>H-NMR (DMSO- $d_6$ )  $\delta$ : 2.15 (3H, s), 3.79 (3H, s), 5.15 (2H, s), 7.27—7.47, 7.47—7.60 (5H, m), 7.78 (1H, s), 7.92 (1H, s), 9.58 (1H, s). *Anal.* Calcd for  $C_{17}H_{16}CINO_4$ : C, 61.18; H, 4.83; Cl, 10.62; N, 4.20. Found: C, 60.97; H, 4.88; Cl, 10.84; N, 4.12. MS:  $m/z$ , 334 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3352, 3333, 1707, 1684, 1605, 1580, 1411, 1238. 2) **23e** (colorless crystal) was obtained from **22e** in 80% yield, mp 190—191 °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 5.09 (2H, s), 6.08 (2H, s), 6.53 (1H, s), 7.25—7.45, 7.45—7.6 (5H, m), 7.62 (1H, s), 11.94 (1H, s). *Anal.* Calcd for  $C_{14}H_{12}CINO_3 \cdot 1/4H_2O$ : C, 59.58; H, 4.46; Cl, 12.56; N, 4.96. Found: C, 59.78; H, 4.35; Cl, 12.48; N, 4.95. MS: *m*/*z*, 278 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3470, 3323, 3285, 1701, 1626, 1593, 1441, 1400, 1229.

**4-Amino-2-butoxy-5-chlorobenzoic Acid (23f)** In a similar manner to that described above, 1) **22f** (colorless fine needle) was prepared from **21** and BuI in 43% yield, mp 105.5—107 °C (EtOH/hexane). <sup>1</sup>H-NMR  $\delta$ : 0.98  $(3H, t, J=7)$ ,  $1.42-1.63$   $(2H, m)$ ,  $1.74-1.89$   $(2H, m)$ ,  $2.27$   $(3H, s)$ ,  $3.78$ (3H, s), 4.07 (2H, t, *J*=6), 7.75 (1H, br s), 7.88 (1H, s), 8.27 (1H, s). *Anal.* Calcd for  $C_{14}H_{18}CINO_4$ : C, 56.10; H, 6.05; Cl, 11.83; N, 4.67. Found: C, 55.85; H, 6.04; Cl, 11.70; N, 4.66. MS:  $m/z$ , 300 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3283, 1699, 1676, 1574, 1412, 1248. 2) **23f** (colorless fine needle) was obtained from **22f** in 94% yield, mp 124.5—125 °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 0.92 (3H, t, J=7.4), 1.38–1.54 (2H, m), 1.62–1.74 (2H, m), 3.92 (2H, t, *J*56.4), 6.03 (2H, s), 6.44 (1H, s), 7.57 (1H, s), 11.77 (1H, br s). *Anal.* Calcd for  $C_{11}H_{14}CINO$ ; C, 54.22; H, 5.79; Cl, 14.55; N, 5.75. Found: C, 53.86; H, 5.72; Cl, 14.53; N, 5.70. MS:  $m/z$ , 244 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3497, 3381, 3267, 1724, 1620, 1591, 1441, 1410, 1230.

**5-Chloro-2-isopropoxy-4-methylaminobenzoic Acid (25)** 1) NaH (60% dispersion in mineral oil, 5.9 g, 0.15 mol) was added portionwise to a mixture of methyl 4-acetylamino-5-chloro-2-isopropoxybenzoate<sup>20)</sup> (22c, 10.3 g, 36 mmol), MeI (30.6 g, 0.22 mol), and anhydrous THF (60 ml) at *ca.* 5 °C. The reaction mixture was stirred at room temperature for 16 h and concentrated to dryness. After careful addition of  $H<sub>2</sub>O$ , the precipitates were collected by filtration, washed with H<sub>2</sub>O, and dried to give  $8.1 g$  (75%) of methyl 4-(*N*-acetyl-*N*-methyl)amino-5-chloro-2-isopropoxybenzoate (**24**) as a colorless crystal, mp 91—92.5 °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 1.39 (6H, d, *J*=6.1), 1.85 (3H, s), 3.19 (3H, s), 3.91 (3H, s), 4.55 (1H, hep, *J*=6.1), 6.88 (1H, s), 7.88 (1H, s). *Anal.* Calcd for C<sub>14</sub>H<sub>18</sub>ClNO<sub>4</sub>: C, 56.10; H, 6.05; Cl, 11.83; N, 4.67. Found: C, 56.14; H, 6.11; Cl, 11.85; N, 4.55. MS: *m*/*z*, 300 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 1715, 1666, 1599, 1244, 1105. 2) A mixture of 24 (4.5 g, 15 mmol) and 20% aqueous NaOH (60 ml) was heated to reflux for 15 h and cooled to *ca.* 5 °C. After acidification with 10% aqueous HCl, the precipitates were collected by filtration and recrystallized from MeOH to give 3.5 g (96%) of 25 as a colorless crystal, mp 154—155 °C. <sup>1</sup>H-NMR (DMSO- $d_6$ )  $\delta$ : 1.29 (6H, d, *J*=6.1), 2.82 (3H, d, *J*=4.8, NMe), 4.71 (1H, hep, *J*=6.1), 6.19 (1H, q, J=4.8, NH), 6.21 (1H, s), 7.52 (1H, s), 11.73 (1H, br s, CO<sub>2</sub>H). *Anal.* Calcd for C<sub>11</sub>H<sub>14</sub>ClNO<sub>3</sub>: C, 54.22; H, 5.79; Cl, 14.55; N, 5.75. Found: C, 53.97; H, 5.84; Cl, 14.41; N, 5.72. MS:  $m/z$ , 243 (M<sup>+</sup>). IR cm<sup>-1</sup>: 3396, 1713, 1607, 1454, 1389.

**5-Chloro-4-ethylamino-2-methoxybenzoic Acid (27)** 1) Methyl 4 acetylamino-5-chloro-2-methoxybenzoate (**22a**, 10.0 g, 39 mmol) was added portionwise to a suspension of NaH (60% dispersion in mineral oil, 4.7 g, 0.12 mol) in anhydrous tetrahydrofuran (THF) (75 ml) under ice-cooling. The mixture was warmed to room temperature and stirred for 1 h. After the mixture was recooled to *ca.* 5 °C, EtI (31 ml, 0.39 mol) was added, and the whole was stirred at room temperature overnight. The reaction mixture was poured carefully into cold  $H_2O$ , and the volatiles were evaporated. The aqueous solution was washed with CHCl<sub>3</sub>, acidified with 35% aqueous HCl, and extracted with CHCl<sub>3</sub>. The extract was washed with brine and concentrated to dryness to give *ca.* 10 g of 4-(*N*-acetyl-*N*-ethyl)amino-5-chloro-2 methoxybenzoic acid (**26**) as a pale yellow amorphous solid, which was used in the next step without further purification. <sup>1</sup>H-NMR (DMSO- $d_6$ )  $\delta$ : 1.02 (3H, t, J=7.1), 3.11 (3H, s), 3.46 (1H, dq, J=7.1, 13.6), 3.77 (1H, dq, *J*57.1, 13.6), 3.85 (3H, s), 7.26 (1H, s), 7.78 (1H, s), 13.09 (1H, br s). IR cm<sup>-1</sup>: 1717, 1647, 1600, 1227. MS:  $m/z$ , 272 (MH<sup>+</sup>). 2) In a similar manner to that described above, **27** was obtained from **26** in 63% yield from **22a**. An analytical sample of **27** was obtained by recrystallization from acetone/hexane, mp 128—130 °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 1.21 (3H, t, *J*=7.5), 3.28 (2H, q, *J*=7.5), 3.83 (3H, s), 6.00 (1H, br), 6.27 (1H, s), 7.64 (1H, s), 11.98 (1H, br s). *Anal*. Calcd for  $C_{10}H_{12}CINO_3$ : C, 52.30; H, 5.27; Cl, 15.44; N, 6.10. Found: C, 52.11; H, 5.27; Cl, 15.21; N, 5.93. MS: *m*/*z*, 230 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3427, 3281, 1707, 1606, 1425, 1323.

**4-Acetylamino-5-chloro-2-methoxybenzoic Acid (28)** A suspension of **22a** (50.0 g, 0.19 mol) and NaOH (9.0 g, 0.23 mol) in a mixture of MeOH (80 ml) and H<sub>2</sub>O (150 ml) was heated at  $60^{\circ}$ C for 1 h and cooled to room temperature. After evaporation of MeOH, the resulting aqueous solution was

acidified with 35% aqueous HCl. The precipitates were collected by filtration and dissolved in MeOH. The solution was concentrated, and the crystalline precipitates were collected by filtration and dried to give 39.6 g (84%) of **28** as a colorless crystal, mp 210—213 °C. <sup>1</sup>H-NMR (DMSO- $d_6$ )  $\delta$ : 2.15 (3H, s), 3.77 (3H, s), 7.71 (1H, s), 7.76 (1H, s), 9.55 (1H, s). *Anal.* Calcd for  $C_{10}H_{10}CINO_4$ : C, 49.30; H, 4.14; Cl, 14.55; N, 5.75. Found: C, 49.27; H, 4.22; Cl, 14.43; N, 5.66. MS:  $m/z$ , 244 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3369, 3260, 1720, 1693, 1604, 1582, 1522, 1450, 1408, 1337, 1240.

**5-Chloro-4-dimethylamino-2-ethoxybenzoic Acid (29b)** A mixture of 4-amino-5-chloro-2-ethoxybenzoic acid28) (**23b**, 2.6 g, 12 mmol), HCHO  $(37\% \text{ solution in H}_2O, 8 \text{ ml}, 0.11 \text{ mol})$ , NaBH<sub>2</sub>CN  $(95\%, 1.9 \text{ g}, 27 \text{ mmol})$ , and MeCN (40 ml) was stirred at room temperature for 6 h. After addition of HCHO (37% solution in H<sub>2</sub>O, 4 ml, 54 mmol) and NaBH<sub>2</sub>CN (95%, 1.0 g, 14 mmol), the mixture was restirred at room temperature for 16 h. The solvent was evaporated, and the residue was acidified with 10% aqueous HCl. The precipitates were collected by filtration and recrystallized from EtOH to give 1.7 g (58%) of 29b as a white crystal, mp  $144-145$  °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 1.33 (3H, t, *J*=7.0), 2.83 (6H, s), 4.12 (2H, q, *J*=7.0), 6.66 (1H, s), 7.64 (1H, s), 12.29 (1H, br s). *Anal.* Calcd for  $C_{11}H_{14}CINO_3$ : C, 54.22; H, 5.79; Cl, 14.55; N, 5.75. Found: C, 54.00; H, 5.81; Cl, 14.62; N, 5.57. MS:  $m/z$ , 244 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3294, 1717, 1605, 1406, 1238.

**5-Chloro-4-dimethylamino-2-methoxybenzoic Acid (29a)** In a similar manner to that described above, **29a** (colorless crystal) was prepared from 4 amino-5-chloro-2-methoxybenzoic acid (**23a**) in 55% yield, mp 150— 151.5 °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 2.85 (6H, s), 3.83 (3H, s), 6.67 (1H, s), 7.65 (1H, s), 12.39 (1H, br s). *Anal.* Calcd for C<sub>10</sub>H<sub>12</sub>ClNO<sub>3</sub>: C, 52.30; H, 5.27; Cl, 15.44; N, 6.10. Found: C, 52.36; H, 5.26; Cl, 15.36; N, 5.95. MS: *m*/*z*, 230 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 1686, 1597, 1244.

**5-Chloro-4-formylamino-2-methoxybenzoic Acid (30)** A mixture of Ac<sub>2</sub>O (55.7 g, 0.55 mol) and HCO<sub>2</sub>H (99%, 100.6 g, 2.2 mol) was heated at 50 °C for 1 h. After addition of **23a** (11.0 g, 55 mmol), the whole was heated at 65 °C for 2 h and cooled to room temperature. The reaction mixture was poured into cold  $H_2O$ , and the precipitates were collected by filtration, washed with H<sub>2</sub>O, and dried to give 10.6 g (85%) of **30** as a gray powder, mp 207—208 °C. <sup>1</sup>H-NMR (DMSO-*d<sub>6</sub>) δ*: 3.79 (3H, s), 7.75 (1H, s), 8.15 (1H, s), 8.44 (1H, s), 10.08 (1H, s), 12.76 (1H, s). *Anal*. Calcd for C<sub>9</sub>H<sub>8</sub>ClNO<sub>4</sub>: C, 47.08; H, 3.51; Cl, 15.44; N, 6.10. Found: C, 46.99; H, 3.53; Cl, 15.41; N, 6.00. MS:  $m/z$ , 230 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3323, 3275, 1717, 1678, 1609, 1583, 1518, 1447, 1418, 1271.

**4-Amino-5-bromo-2-ethoxybenzoic Acid (33a)** 1) A solution of methyl 4-acetylamino-2-ethoxybenzoate28,29) (**31**, 10.0 g, 42 mmol) and NBS (8.2 g, 46 mmol) in *N*,*N*-dimethylformamide (DMF) (60 ml) was heated at 80 °C for 3.5 h and cooled to room temperature. After addition of  $H_2O$  $(100 \text{ ml})$ , the precipitates were collected by filtration, washed with H<sub>2</sub>O, and dried to give 12.5 g (94%) of methyl 4-acetylamino-5-bromo-2-ethoxybenzoate (32a) as a white fine needle, mp  $172-173$  °C. <sup>1</sup>H-NMR (DMSO- $d_6$ ) d: 1.31 (3H, t, *J*57), 2.13 (3H, s), 3.76 (3H, s), 4.04 (2H, q, *J*57), 7.64 (1H, s), 7.86 (1H, s), 9.44 (1H, s). *Anal*. Calcd for C<sub>12</sub>H<sub>14</sub>BrNO<sub>4</sub>: C, 45.59; H, 4.46; Br, 25.27; N, 4.43. Found: C, 45.57; H, 4.48; Br, 25.38; N, 4.43. MS: m/z, 315 (M<sup>+</sup>). IR cm<sup>-1</sup>: 3231, 2988, 1690, 1665, 1572, 1408, 1389, 1252. 2) A mixture of **32a** (4.1 g, 13 mmol), NaOH (3.5 g, 88 mmol), MeOH (12 ml), and H<sub>2</sub>O (16 ml) was heated to reflux for 4 h and cooled to *ca.*  $5^{\circ}$ C. The reaction mixture was acidified with  $25\%$  aqueous  $H_2SO_4$ , and the precipitates were collected by filtration and recrystallized from MeOH to afford 3.3 g (98%) of **33a** as a colorless crystal, mp  $179-181$  °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 1.33 (3H, t, *J*=7.0), 3.98 (2H, q, *J*=7.0), 5.99 (2H, br s, NH<sub>2</sub>), 6.44 (1H, s), 7.73 (1H, s), 11.79 (1H, br s, CO<sub>2</sub>H). *Anal*. Calcd for C9H10BrNO3: C, 41.56; H, 3.88; Br, 30.72; N, 5.39. Found: C, 41.20; H, 3.87; Br, 30.50; N, 5.23. MS:  $m/z$ , 259 (M<sup>+</sup>). IR cm<sup>-1</sup>: 3489, 3327, 1699, 1628, 1585, 1447, 1402.

**4-Amino-2-ethoxy-5-iodobenzoic Acid (33b)** 1) A solution of **31**  $(5.0 g, 21 mmol)$  and ICl  $(3.8 g, 23 mmol)$  in DMF  $(25 ml)$  was stirred at room temperature for 0.5 h, heated at 80 °C for 2 h, and cooled to room temperature. After addition of  $H<sub>2</sub>O$  (100 ml), the precipitates were collected by filtration and recrystallized from acetone to give 3.6 g (47%) of methyl 4 acetylamino-2-ethoxy-5-iodobenzoate (**32b**) as a colorless fine needle, mp 188—189 °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 1.32 (3H, t, *J*=7), 2.10 (3H, s), 3.77 (3H, s), 4.04 (2H, q, *J*57), 7.42 (1H, s), 8.06 (1H, s), 9.36 (1H, s). *Anal.* Calcd for  $C_{12}H_{14}INO_4$ : C, 39.69; H, 3.89; I, 34.95; N, 3.86. Found: C, 39.88; H, 3.93; I, 34.83; N, 3.67. MS:  $m/z$ , 363 (M<sup>+</sup>). IR cm<sup>-1</sup>: 3221, 2984, 1684, 1665, 1566, 1406, 1387, 1252. 2) In a similar manner to that described above, **32b** was hydrolyzed to give **33b** as a colorless crystal in 91% yield, mp 190—193 °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 1.32 (3H, t, *J*=7.0), 3.98 (2H, q, *J*=7.0), 5.87 (2H, br s, NH<sub>2</sub>), 6.42 (1H, s), 7.93 (1H, s), 11.75 (1H, br s). *Anal.* Calcd for C<sub>0</sub>H<sub>10</sub>INO<sub>3</sub>: C, 35.20; H, 3.28; I, 41.33; N, 4.56. Found: C, 35.18; H, 3.32; I, 41.20; N, 4.40. MS:  $m/z$ , 307 (M<sup>+</sup>). IR cm<sup>-1</sup>: 3468, 3314, 1703, 1622, 1583, 1435, 1400.

**2-Methoxy-4-[***N***-methyl-***N***-(***p***-toluenesulfonyl)]aminobenzoic Acid (35)** A mixture of 4-(*p*-toluenesulfonyl)aminosalicylic acid<sup>30</sup> (34, 250 g, 0.81 mol), acetone (1700 ml), and KOH (239 g, 4.3 mol) was stirred at room temperature for 20 min.  $Me<sub>2</sub>SO<sub>4</sub>$  (339 g, 2.7 mol) was added dropwise to the mixture at a rate that maintained boiling. The whole was heated to reflux for 2 h and cooled to room temperature. After evaporation of the solvent, the residue was dissolved in H<sub>2</sub>O and washed with AcOEt. The aqueous solution was acidified with 35% aqueous HCl, and the precipitates were collected by filtration, washed with H<sub>2</sub>O, and dried to give  $183 g$  (67%) of **35**. An analytic sample of **35** was obtained by recrystallization from acetone as a colorless prism, mp 136—137 °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 2.48 (3H, s), 2.17 (3H, s), 3.72 (3H, s), 6.73 (1H, dd,  $J=2.0$ , 8.5), 6.86 (1H, d,  $J=2.0$ ), 7.35—7.52 (4H, m), 7.59 (1H, d, J=8.5), 12.50 (1H, s, CO<sub>2</sub>H). *Anal.* Calcd for  $C_{16}H_{17}NO_5S$ : C, 57.30; H, 5.11; N, 4.18; S, 9.56. Found: C, 57.04; H, 5.10; N, 4.16; S, 9.52. MS:  $m/z$ , 336 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3287, 1736, 1607, 1578, 1421, 1350.

**5-Bromo-2-methoxy-4-methylaminobenzoic Acid (37a)** 1) A mixture of **35** (2.8 g, 8.4 mmol), *N*-bromosuccinimide (NBS) (1.65 g, 9.3 mmol), and DMF (50 ml) was heated at 80 °C for 4 h and cooled to room temperature. After addition of  $H_2O$ , the precipitates were collected by filtration, washed successively with H<sub>2</sub>O and Et<sub>2</sub>O, and dried to give  $3.4 \text{ g}$  (98%) of 5-bromo-2-methoxy-4-[*N*-methyl-*N*-(*p*-toluenesulfonyl)]aminobenzoic acid (**36a**). An analytic sample of **36a** was obtained by recrystallization from acetone, mp 229—231 °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 2.42 (3H, s), 3.10 (3H, s), 3.60 (3H, s), 6.54 (1H, s), 7.48 (2H, d,  $J=8.0$ ), 7.67 (2H, d,  $J=8.0$ ), 7.87 (1H, s), 13.10 (1H, s). *Anal.* Calcd for C<sub>16</sub>H<sub>16</sub>BrNO<sub>5</sub>S: C, 46.39; H, 3.89; Br, 19.29; N, 3.38; S, 7.74. Found: C, 46.03; H, 3.85; Br, 19.62; N, 3.39; S, 7.68. MS: *m*/*z*, 414 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 1688, 1342, 1153. 2) A mixture of **36a** (5.0 g, 12 mmol) and concentrated  $H_2SO_4$  (30 ml) was stirred for 1.5 h under icecooling. The reaction mixture was poured into cold  $H_2O$ , and the resulting precipitates were collected by filtration, washed successively with H<sub>2</sub>O and Et<sub>2</sub>O, and dried to give 2.9 g (92%) of 37a, mp  $176-178$  °C. <sup>1</sup>H-NMR (DMSO-*d*6) d: 3.03 (3H, s), 3.93 (3H, s), 6.04 (1H, br), 6.17 (1H, s), 7.77 (1H, s), 11.60 (1H, s). *Anal.* Calcd for C<sub>9</sub>H<sub>10</sub>BrNO<sub>3</sub>: C, 41.56; H, 3.88; Br, 30.72; N, 5.39. Found: C, 41.43; H, 3.88; Br, 30.36; N, 5.23. MS: *m*/*z*, 260 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3410, 1668, 1597, 1551, 1474, 1391, 1348, 1265, 1250, 1221.

**5-Chloro-2-methoxy-4-methylaminobenzoic Acid (37b)** In a similar manner to that described above, **37b** was obtained from 5-chloro-2 methoxy-4-[*N*-methyl-*N*-(*p*-toluenesulfonyl)]aminobenzoic acid (**36b**), which was prepared from **35** using *N*-chlorosuccinimide (NCS). **36b**; mp 210.5—212.5 °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 2.42 (3H, s), 3.01 (3H, s), 3.62 (3H, s), 6.62 (1H, s), 7.47 (2H, d, *J*=8.0), 7.67 (2H, d, *J*=8.0), 7.72 (1H, s), 13.13 (1H, s). *Anal.* Calcd for C<sub>16</sub>H<sub>16</sub>ClNO<sub>5</sub>S: C, 51.96; H, 4.43; Cl, 9.72; N, 3.62; S, 8.67. Found: C, 51.66; H, 4.43; Cl, 9.72; N, 3.62; S, 8.53. MS: *m/z*, 370 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 1690, 1344, 1153. **37b**; mp 186—187 °C (EtOH, lit.<sup>30)</sup> mp 188—189 °C). <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 2.84 (3H, d, *J*=5.0, NMe), 3.83 (3H, s), 6.19 (1H, s), 6.23 (1H, br, *J*=5.0, NH), 7.61 (1H, s), 11.84 (1H, s). *Anal.* Calcd for  $C_9H_{10}CINO_3$ : C, 50.13; H, 4.67; Cl, 16.44; N, 6.50. Found: C, 50.04; H, 4.66; Cl, 16.38; N, 6.43. MS:  $m/z$ , 216 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3422, 1666, 1601, 1553, 1252, 1219.

**2-Methoxy-4-methylamino-5-nitrobenzoic Acid (37d)** In a similar manner to that described above, 2-methoxy-4-[*N*-methyl-*N*-(*p*-toluenesulfonyl)]amino-5-nitrobenzoic acid (**36d**) was prepared from **35** using fuming  $HNO<sub>3</sub>$  and concentrated H<sub>2</sub>SO<sub>4</sub>. **36d** was treated with concentrated  $H<sub>2</sub>SO<sub>4</sub>$  to give **37d**. **36d**; mp 184—185 °C. <sup>1</sup> H-NMR (DMSO-*d*6) d: 2.42 (3H, s), 3.03 (3H, s), 3.68 (3H, s), 6.60 (1H, s), 7.48 (2H, d, *J*=8.0), 7.53 (2H, d, *J*=8.0), 8.27 (1H, s), 13.39 (1H, s). *Anal*. Calcd for C<sub>16</sub>H<sub>16</sub>N<sub>2</sub>O<sub>7</sub>S: C, 50.52; H, 4.24; N, 7.36; S, 8.43. Found: C, 50.62; H, 4.26; N, 7.58; S, 8.40. MS: *m*/*z*, 381 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 1690, 1611, 1531, 1441, 1352, 1271, 1171. **37d**; mp 280—283 °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 3.03 (3H, d, *J*=5.0, NMe), 3.94 (3H, s), 6.28 (1H, s), 8.58 (1H, br, *J*=5.0, NH), 8.59 (1H, s), 12.50 (1H, br s). Anal. Calcd for C<sub>9</sub>H<sub>10</sub>N<sub>2</sub>O<sub>5</sub>: C, 47.79; H, 4.46; N, 12.39. Found: C, 47.64; H, 4.43; N, 12.34. MS:  $m/z$ , 227 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3371, 1682, 1620, 1572, 1514, 1248.

**5-Iodo-2-methoxy-4-methylaminobenzoic Acid (37c)** 1) A mixture of methyl 2-methoxy-4-methylaminobenzoate<sup>30)</sup> (38, 4.0 g, 21 mmol), Ac<sub>2</sub>O  $(10 \text{ ml}, 0.11 \text{ mol})$ , and Et<sub>3</sub>N  $(1 \text{ ml})$  was stirred at room temperature for 15 h. The reaction mixture was diluted with  $H_2O$ , and then extracted with AcOEt. The extract was washed with brine and concentrated to dryness. The residue was chromatographed on silica gel with AcOEt to give 4.7 g (97%) of methyl 4-(*N*-acetyl-*N*-methyl)amino-2-methoxybenzoate (**39**) as a colorless oil, which was used in the next step without further purification. <sup>1</sup>H-NMR  $\delta$ : 1.96 (3H, s), 3.28 (3H, s), 3.92 (6H, s), 6.81 (1H, d, *J*=2.0), 6.83 (1H, dd, *J*=2.0, 8.0), 7.84 (1H, d, *J*=8.0). IR cm<sup>-1</sup>: 1728, 1661, 1603. MS: *m*/*z*, 238  $(MH<sup>+</sup>)$ . 2) ICl  $(6.4 g, 39 mmol)$  was added to a solution of **39**  $(4.7 g, 4.7 g)$ 20 mmol) in DMF (20 ml) under ice-cooling. The mixture was warmed to room temperature and stirred for 2 h. The reaction mixture was poured into cold H<sub>2</sub>O, and the precipitates were collected by filtration, washed with H<sub>2</sub>O, and dried to give 6.8 g (94%) of methyl 4-(*N*-acetyl-*N*-methyl)amino-5-iodo-2-methoxybenzoate (**40**) as a pale yellow powder. An analytical sample of 40 was obtained by recrystallization from AcOEt, mp 162-163 °C. <sup>1</sup>H-NMR δ: 1.83 (3H, s), 3.28 (3H, s), 3.90 (3H, s), 3.92 (3H, s), 6.89 (1H, s), 8.27 (1H, s). *Anal.* Calcd for C<sub>12</sub>H<sub>14</sub>INO<sub>4</sub>: C, 39.69; H, 3.89; I, 34.95; N, 3.86. Found: C, 39.39; H, 3.79; I, 34.83; N, 3.78. MS;  $m/z$ , 236 (M-I<sup>+</sup>). IR cm<sup>2</sup><sup>1</sup> : 1722, 1637, 1601. 3) A mixture of **40** (3.8 g, 10 mmol) and 20% aqueous NaOH (40 ml) was heated to reflux for 15 h and cooled to *ca.* 5 °C. After acidification with 35% aqueous HCl, the precipitates were collected by filtration, washed with H<sub>2</sub>O, and dried to give  $3.0 \text{ g}$  (93%) of **37c**, mp 187– 188 °C (dec., MeOH/EtOH). <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>) δ: 2.84 (3H, d, J=5.0, NMe), 3.83 (3H, s), 5.68 (1H, br, *J*=5.0, NH), 6.08 (1H, s), 8.00 (1H, s). *Anal.* Calcd for C<sub>9</sub>H<sub>10</sub>INO<sub>3</sub>: C, 35.20; H, 3.28; I, 41.33; N, 4.56. Found: C, 35.36; H, 3.26; I, 41.15; N, 4.50. MS:  $m/z$ , 308 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3395, 1668, 1589, 1545, 1258, 1225.

**5-Formyl-2-methoxy-4-methylaminobenzoic Acid (43)** 1) A mixture of **38** (4.0 g, 21 mmol), benzyl bromide  $(5.3 \text{ g}, 31 \text{ mmol})$ , K<sub>2</sub>CO<sub>3</sub> (5.7 g, 41 mmol), and DMF (40 ml) was heated to reflux for 4 h and cooled to room temperature. After evaporation of the solvent, the residue was dissolved in H<sub>2</sub>O and CHCl<sub>3</sub>. The organic layer was separated, washed with brine, dried over anhydrous  $MgSO<sub>4</sub>$ , and concentrated to dryness. The residue was chromatographed on silica gel with AcOEt to afford 5.85 g of a white solid. The solid was recrystallized from AcOEt/hexane to give 3.8 g (65%) of methyl 4- (*N*-benzyl-*N*-methyl)amino-2-methoxybezoate (**41**), mp 71.5—72.5 °C. <sup>1</sup> H-NMR δ: 3.12 (3H, s), 3.82 (6H, s), 4.61 (2H, s), 6.21 (1H, d, *J*=2.5), 6.32 (1H, dd, *J*=2.5, 9.0), 7.14—7.4 (5H, m), 7.78 (1H, d, *J*=9.0). *Anal.* Calcd for  $C_{17}H_{19}NO_3$ : C, 71.56; H, 6.71; N, 4.91. Found: C, 71.50; H, 6.75; N, 5.07. MS:  $m/z$ , 286 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 2941, 1676, 1601, 1435, 1387, 1298, 1254. 2) POCl<sub>3</sub> (4.9 g, 32 mmol) was added dropwise to DMF  $(8.2 \text{ ml},$ 0.11 mol) under ice-cooling. The mixture was warmed to room temperature and stirred for 0.5 h. After addition of **41** (2.8 g, 9.8 mmol), the whole was heated at 80 °C for 4 h and cooled to room temperature. The reaction mixture was poured into ice  $H<sub>2</sub>O$  (80 ml). Then, AcONa (25 g) and  $H<sub>2</sub>O$  (62 ml) were added, and the mixture was extracted with AcOEt. The extract was washed with brine and concentrated. The resulting insoluble materials were filtered off, and the filtrate was chromatographed on silica gel with AcOEt to afford 0.75 g of an unseparatable mixture of methyl 5-formyl-2-methoxy-4 methylaminobenzoate (**42a**) and methyl 5-formyl-4-(*N*-formyl-*N*-methyl) amino-2-methoxybenzoate (42b). <sup>1</sup>H-NMR δ: 3.37 (3H, s), 3.41 (3H, s), 3.92 (3H, s), 3.93 (3H, s), 3.99 (3H, s), 4.01 (3H, s), 6.78 (1H, s), 6.83 (1H, s), 8.30 (1H, s), 8.37 (1H, s), 8.41 (1H, s), 8.42 (1H, s), 9.96 (1H, s), 9.88 (1H, s). MS:  $m/z$ , 224 (MH<sup>+</sup> of **42a**), 252 (MH<sup>+</sup> of **42b**). 3) A solution of the mixture of **42a** and **42b** (0.75 g) and NaOH (0.4 g, 10 mmol) in MeOH  $(2.5 \text{ ml})$ –H<sub>2</sub>O (5 ml) mixture was heated to reflux for 1 h and cooled to room temperature. The reaction mixture was concentrated, and the aqueous solution was acidified with 35% aqueous HCl and extracted with CHCl<sub>3</sub>. The extract was concentrated to dryness, and the residue was triturated with  $Et<sub>2</sub>O$ to give 0.62 g (30% yield from **41**) of **43** as a white solid. An analytical sample of 43 was obtained by recrystallization from MeOH/Et<sub>2</sub>O, mp 231-232 °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>)  $\delta$ : 2.94 (3H, d, *J*=5.0, NMe), 3.90 (3H, s), 6.17 (1H, s), 8.09 (1H, s), 8.64 (1H, br, *J*55.0, NH), 9.69 (1H, s, CHO), 12.08 (1H, s). *Anal.* Calcd for C<sub>10</sub>H<sub>11</sub>NO<sub>4</sub>: C, 57.41; H, 5.30; N, 6.70. Found: C, 57.34; H, 5.20; N, 6.79. MS:  $m/z$ , 209 (M<sup>+</sup>). IR cm<sup>-1</sup>: 3329, 1724, 1649, 1614, 1570, 1423, 1304, 1256.

**5-Ethylthio-2-methoxy-4-methylaminobenzoic Acid (46)** 1) A solution of Br<sub>2</sub> (2.8 ml, 55 mmol) in MeOH (7.8 ml) was added to a mixture of **38** (10.0 g, 51 mmol), NH4SCN (6.2 g, 82 mmol), and MeOH (36 ml) under ice-cooling. The whole was stirred at this temperature for 5 h and poured into H<sub>2</sub>O. The solid precipitate was collected by filtration, washed with H<sub>2</sub>O, and dried to give 9.6 g (74%) of methyl 5-cyanothio-2-methoxy-4-methylaminobenzoate (44) as a white solid, mp 162—164 °C. <sup>1</sup>H-NMR (DMSO*d*<sub>6</sub>) δ: 2.90 (3H, d, *J*=5.0, NMe), 3.72 (3H, s), 3.88 (3H, s), 6.21 (1H, s), 6.88 (1H, br,  $J=5.0$ , NH), 7.92 (1H, s). *Anal*. Calcd for  $C_{11}H_{12}N_{2}O_{2}S$ 1/10H2O: C, 52.00; H, 4.84; N, 11.02; S, 12.62. Found: C, 51.87; H, 4.92; N, 10.81; S, 12.42. MS:  $m/z$ , 253 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3366, 2160, 1686, 1601, 1562, 1472, 1248. 2) A mixture of **44** (5.0 g, 20 mmol), KOH (1.3 g, 23 mmol), and MeOH (100 ml) was heated to reflux for 1 h and cooled to room temperature. After addition of  $Et_2SO_4$  (2.3 g, 15 mmol), the whole was heated to reflux for 48 h and cooled to room temperature. The solvent was evaporated, and the residue was diluted with H<sub>2</sub>O and acidified with 35% aqueous HCl. After addition of  $CHCl<sub>3</sub>$ , the insoluble materials were filtered off. The organic layer of the filtrate was separated, washed with brine, and concentrated to dryness. The residue was chromatographed on silica gel with  $CHCl<sub>3</sub>/ACOEt=50/1$  to give 1.7 g (34%) of methyl 5-ethylthio-2methoxy-4-methylaminobenzoate (**45**) as an oil, which was used in the next step without further purification. <sup>1</sup>H-NMR  $\delta$ : 1.18 (3H, t, *J*=7.5), 2.62 (2H, q,  $J=7.5$ ), 2.85 (3H, d,  $J=5.0$ ), 3.82 (3H, s), 3.93 (3H, s), 5.64 (1H, br, *J*=5.0), 6.05 (1H, s), 8.02 (1H, s). MS:  $m/z$ , 256 (MH<sup>+</sup>). 3) In a similar manner to that described above, **45** was hydrolyzed with aqueous NaOH to give 46 as a pale yellow powder in 92% yield. 46; mp  $143-145$  °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 1.07 (3H, t, *J*=7.5), 2.59 (2H, q, *J*=7.5), 2.85 (3H, d, *J*=5.0), 3.84 (3H, s), 6.10 (1H, s), 6.28 (1H, br, *J*=5.0), 7.78 (1H, s), 11.68 (1H, s). *Anal*. Calcd for  $C_{11}H_{15}NO_3S \cdot 1/4H_2O$ : C, 53.75; H, 6.36; N, 5.70; S, 13.04. Found: C, 53.92; H, 6.07; N, 5.89; S, 12.93. MS:  $m/z$ , 242 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3364, 1678, 1595, 1549, 1252.

**5-Ethylsulfonyl-2-methoxy-4-methylaminobenzoic Acid (47)** A mixture of  $46$  (1.5 g, 6.2 mmol),  $30\%$  aqueous H<sub>2</sub>O<sub>2</sub> (1.8 g, 16 mmol), and AcOH (15 ml) was stirred at room temperature overnight. After evaporation of the solvent, the residue was dissolved in CHCl<sub>3</sub>. The organic solution was washed with brine, dried over anhydrous  $MgSO<sub>4</sub>$ , and concentrated to dryness to give 1.2 g (71%) of **47** as a solid. An analytical sample of **47** was obtained by recrystallization from aqueous AcOH, mp 216—218 °C. <sup>1</sup>H-NMR  $(DMSO-d_6)$   $\delta$ : 1.08 (3H, t, *J*=7.3), 2.91 (3H, d, *J*=5.0), 3.17 (2H, q, *J*=7.3), 3.31 (3H, s), 6.22 (1H, s), 6.68 (1H, br, *J*=5.0, NH), 8.04 (1H, s), 12.12 (1H, br s). *Anal*. Calcd for C<sub>11</sub>H<sub>15</sub>NO<sub>5</sub>S: C, 48.34; H, 5.53; N, 5.12; S, 11.73. Found: C, 48.29; H, 5.67; N, 4.83; S, 11.65. MS:  $m/z$ , 274 (M<sup>+</sup>). IR cm<sup>-1</sup>: 3379, 1720, 1609, 1572.

**Methyl 4-Acetylamino-5-chloro-2-methoxy-3-nitrobenzoate (48)** Concentrated H<sub>2</sub>SO<sub>4</sub> (18 ml) was added dropwise to fuming HNO<sub>3</sub> (*d* 1.52, 150 ml, 306 mol) under ice-cooling. **22a** (50.0 g, 0.19 mol) was added portionwise to the mixture below 10 °C. The whole was stirred at the same temperature for 5 min and poured into cold H<sub>2</sub>O. The solid precipitate was collected by filtration, washed with  $H_2O$ , and dried to give  $42.6 g$  (73%) of  $48$ as a pale yellow solid. An analytical sample of **48** was obtained by recrystallization from EtOH, mp 137—138 °C. <sup>1</sup>H-NMR δ: 2.21 (3H, s), 3.96 (3H, s), 4.00 (3H, s), 7.31 (1H, s), 8.10 (1H, s). *Anal.* Calcd for  $C_{11}H_{11}CIN_2O_6$ : C, 43.65; H, 3.66; Cl, 11.71; N, 9.26. Found: C, 43.41; H, 3.68; Cl, 11.62; N, 9.17. MS:  $m/z$ , 303 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3321, 1711, 1541, 1491, 1363, 1240, 1146.

**Methyl 4-Acetylamino-3-amino-5-chloro-2-methoxybenzoate (49)** A mixture of 48 (18.0 g, 60 mmol), EtOH (180 ml), and H<sub>2</sub>O (10 ml) was hydrogenated over Raney Ni (wet, *ca.* 2 g) at room temperature under an initial pressure of  $4.4 \text{ kg/cm}^2$ . After no further change in the pressure of H<sub>2</sub> (3.6) kg/cm2 , *ca.* 1 h) was observed, the catalyst was filtered through Celite. The filtrate was concentrated to dryness to give 15.6 g (96%) of **49** as a solid. An analytical sample of **49** was obtained by recrystallization from EtOH, mp 149—151 °C. <sup>1</sup>H-NMR δ: 2.29 (3H, s), 3.85 (3H, s), 3.90 (3H, s), 4.51 (2H, s), 7.23 (1H, s), 7.32 (1H, s). *Anal*. Calcd for C<sub>11</sub>H<sub>13</sub>ClN<sub>2</sub>O<sub>4</sub>: C, 48.45; H, 4.81; Cl, 13.00; N, 10.27. Found: C, 48.46; H, 4.85; Cl, 12.99; N, 10.06. MS:  $m/z$ , 273 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3319, 3242, 1736, 1655, 1609, 1522, 1475, 1290, 1205.

**7-Chloro-4-methoxy-1***H***-benzotriazole-5-carboxylic Acid (51)** A solution of **49** (8.8 g, 32 mmol) in concentrated  $H_2SO_4$  (22 ml) was added to cold H<sub>2</sub>O (88 ml). A solution of NaNO<sub>2</sub> (3.3 g, 48 mmol) in H<sub>2</sub>O (10 ml) was added dropwise to the mixture under ice-cooling, and the whole was stirred at the same temperature for 1 h. The acidic solution containing methyl 7 chloro-4-methoxy-1*H*-benzotriazole-5-carboxylate (**50**) was heated to reflux for 18 h and cooled to *ca.* 5 °C. The precipitates were collected by filtration, washed with cold H<sub>2</sub>O, and dried to give  $4.7 \text{ g}$  (64%) of **51**, mp 204— 206 °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 4.46 (3H, s), 7.78 (1H, s), 13.02 (1H, s). *Anal.* Calcd for C<sub>8</sub>H<sub>6</sub>ClN<sub>3</sub>O<sub>3</sub>· 7/4H<sub>2</sub>O: C, 37.08; H, 3.70; Cl, 13.68; N, 16.22. Found: C, 37.14; H, 3.59; Cl, 13.47; N, 16.29. MS:  $m/z$ , 228 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 1701, 1612, 1609, 1501, 1364, 1246.

**7-Chloro-4-methoxy-2-methyl-1***H***-benzimidazole-5-carboxylic Acid (53)** 1) A mixture of **49** (3.0 g, 11 mmol), *p*-toluenesulfonic acid monohydrate (0.1 g), and toluene (50 ml) was heated to reflux with an attached Dean–Stark trap for 1 h and cooled to *ca.* 5 °C. The precipitates were collected by filtration, washed with toluene, and dried to give 2.8 g (quantitative yield) of methyl 7-chloro-4-methoxy-2-methyl-1*H*-benzimidazole-5-carboxylate (**52**). An analytic sample of **52** was obtained by recrystallization

from EtOH/hexane, mp 195.5—197.5 °C. <sup>1</sup>H-NMR δ: 2.68 (3H, s), 3.93 (3H, s), 4.16 (3H, s), 6.78 (1H, s), 7.74 (1H, s). *Anal.* Calcd for  $C_{11}H_{11}CIN_2O_3$ : C, 51.88; H, 4.35; Cl, 13.92; N, 11.00. Found: C, 51.89; H, 4.42; Cl, 13.75; N, 10.86. MS:  $m/z$ , 255 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 1728, 1711, 1543, 1327, 1198. 2) A mixture of **52** (1.25 g, 4.9 mmol), MeOH (10 ml), H2O (10 ml), and NaOH (1.0 g, 25 mmol) was heated to reflux for 1.5 h and cooled to room temperature. After evaporation of MeOH, the aqueous solution was acidified with 35% aqueous HCl. The solution was concentrated, and the residue was crystallized from aqueous EtOH to give 1.3 g (96%) of the hydrochloride of **53**, mp 213—215 °C. <sup>1</sup>H-NMR (DMSO- $d_6$ )  $\delta$ : 2.82 (3H, d, *J*=2.5), 4.06 (3H, s), 7.69 (1H, *J*=2.5), 10.2 (br s). *Anal*. Calcd for C10H9ClN2O3 · HCl: C, 43.34; H, 3.64; Cl, 25.59; N, 10.11. Found: C, 43.29; H, 3.64; Cl, 25.77; N, 10.06. MS:  $m/z$ , 241 (MH<sup>+</sup> of free base). IR cm<sup>-1</sup>: 2826, 1726, 1626, 1580, 1373, 1267, 1190.

**4-Amino-5-chloro-2-methoxy-***N***-[1-methylhexahydro-1,4-diazepin-6 yl]benzamide Dimaleate (75)** A solution of **2x** (1.4 g, 3.5 mmol) in EtOH (20 ml) was hydrogenated over 10% Pd/C (0.3 g) at 50 °C for 3 h under  $H_2$ atmosphere. The catalyst was filtered through Celite. The filtrate was concentrated to dryness to give 1.1 g (quantitative yield) of **75** as a pale yellow oil, which was converted into the maleate in the usual manner. <sup>1</sup>H-NMR (DMSO-*d*6) d: 2.53 (3H, s), 2.7—3.45 (9H, m), 3.84 (3H, s), 4.30 (1H, m), 6.04 (2H, s), 6.14 (4H, s, maleic acid), 6.48 (1H, s), 7.80 (1H, s), 8.35 (1H, d,  $J=8.0$ ). MS:  $m/z$ , 313 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3520, 3348, 1646, 1634, 1600, 1578, 1485, 1454, 1358.

**5-Chloro-***N***-[1-(2-hydroxyethyl)-4-methylhexahydro-1,4-diazepin-6 yl]-2-methoxy-4-methylaminobenzamide Dioxalate (96)** Trimethylsilyl iodide (TMSI, 0.4 ml, 2.8 mmol) was added to a solution of the free base of **95** (1.0 g, 2.6 mmol) in CHCl<sub>3</sub> (20 ml) at room temperature under  $N_2$  atmosphere. After being stirred for 6 h, TMSI (0.4 ml, 2.8 mmol) was added. The mixture was stirred at this temperature overnight. MeOH was carefully added to the reaction mixture, and the volatiles were evaporated. The residue was dissolved in CHCl<sub>3</sub>, and the solution was washed successively with saturated aqueous NaHCO<sub>3</sub>, aqueous Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>, and brine and dried over anhydrous MgSO4. After evaporation of the solvent, the residue was chromatographed on silica gel with CHCl<sub>3</sub>/MeOH=10/1 to give 0.6 g (62%) of **96** as a pale yellow oil, which was converted into the oxalate in the usual manner. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 2.72 (3H, s), 2.79 (3H, t, *J*=6.0), 2.84 (3H, d,  $J=5.0$ , NMe),  $2.87-3.4$  (8H, m),  $3.56$  (2H, q,  $J=6.0$ ),  $3.95$  (3H, s),  $4.30$ (1H, m), 6.17 (1H, br d, J=5.0, NH), 6.24 (1H, s), 7.74 (1H, s), 8.34 (1H, d, *J*=7.5). MS: *m*/*z*, 371 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3385, 1688, 1611, 1470.

*N***-(1-Ethyl-4-methylhexahydro-1,4-diazepin-6-yl)-5-fluoro-2-methoxy-4-methylaminobenzamide Dimaleate (113)** 1) A mixture of 2,4,5-trifluorobenzoic acid (63, 10.0 g, 57 mmol), Et<sub>3</sub>N (10 ml), pyridine (50 ml), and *N*benzylmethylamine (20.7 g, 0.17 mol) was heated to reflux for 45 h and cooled to room temperature. After the volatiles were evaporated, the residue was dissolved in  $H<sub>2</sub>O$ . The solution was then acidified with 35% aqueous HCl and extracted with CHCl<sub>3</sub>. The extract was washed successively with H<sub>2</sub>O and brine and concentrated to dryness. The residue was chromatographed on silica gel with CHCl<sub>3</sub>/MeOH=10/1 to give 7.7 g (49%) of 4-(*N*-benzyl-*N*-methyl)amino-2,5-difluorobenzoic acid (**64**) as a white solid. An analytic sample of **64** was obtained by recrystallization from acetone/ hexane, mp 148—149 °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 2.78 (3H, s), 4.49 (2H, s), 6.56 (1H, dd, *J*=7.5, 13.0), 7.2—7.4 (5H, m), 7.55 (1H, dd, *J*=7.0, 14.0), 7.82 (1H, br). *Anal*. Calcd for C<sub>15</sub>H<sub>13</sub>F<sub>2</sub>NO<sub>2</sub>: C, 64.98; H, 4.73; F, 13.70; N, 5.05. Found: C, 64.64; H, 4.66; F, 13.62; N, 5.08. MS:  $m/z$ , 278 (MH<sup>+</sup>). IR cm<sup>2</sup><sup>1</sup> : 1686, 1626, 1537, 1406, 1277. 2) A mixture of **64** (5.0 g, 18 mmol), **8a** (2.8 g, 18 mmol), 1-ethyl-3-[3-(dimethylamino)propyl]carbodiimide hydrochloride (EDC) (3.8 g, 20 mmol), and CH<sub>2</sub>Cl<sub>2</sub> (50 ml) was stirred at room temperature for 4 h. The reaction mixture was washed successively with H<sub>2</sub>O, 10% aqueous NaOH, and brine and concentrated to dryness. The residue was chromatographed on silica gel with  $CHCl<sub>3</sub>/MeOH = 10/1$  to give 4.2 g (56%) of 4-(*N*-benzyl-*N*-methyl)amino-2,5-difluoro-*N*-(1-ethyl-4 methylhexahydro-1,4-diazepin-6-yl)benzamide (65) as a pale yellow oil. <sup>1</sup>H-NMR δ: 1.05 (3H, t, *J*=7.5), 2.38 (3H, s), 2.45—3.00 (10H, m), 2.89 (3H, s), 4.30 (1H, m), 4.47 (2H, s), 6.45 (1H, dd, *J*=7.5, 14.0), 7.2—7.4 (5H, m), 7.72 (1H, dd,  $J=7.5$ , 14.0), 7.82 (1H, m). MS:  $m/z$ , 417 (MH<sup>+</sup>). The oil was converted into the oxalate in the usual manner. An analytical sample of the oxalate was obtained by crystallization from EtOH/PrOH, mp 126—127 °C. *Anal.* Calcd for  $C_{23}H_{30}F_2N_4O \cdot 5/2C_2H_2O_4 \cdot 1/4H_2O$ : C, 51.34; H, 5.62; F, 5.80; N, 8.55. Found: C, 51.23; H, 5.65; F, 5.94; N, 8.53. IR cm<sup>-1</sup>: 3422, 1630, 1508. 3) A mixture of **65** (1.9 g, 4.6 mmol), NaOMe (powder, 0.8 g, 15 mmol), and anhydrous DMF (20 ml) was heated at 120 °C for 15 h and cooled to room temperature. After evaporation of the solvent, the residue was taken into H<sub>2</sub>O and CHCl<sub>3</sub>. The organic layer was separated, washed

with brine, and dried over anhydrous  $MgSO<sub>4</sub>$ . The solvent was evaporated, and the residue was chromatographed on silica gel with  $CHCl<sub>3</sub>/MeOH=$ 10/1 to give 1.5 g (77%) of 4-(*N*-benzyl-*N*-methyl)amino-*N*-(1-ethyl-4 methylhexahydro-1,4-diazepin-6-yl)-5-fluoro-2-methoxybenzamide (**66**) as a pale yellow oil. <sup>1</sup>H-NMR  $\delta$ : 1.07 (3H, t, J=7.5), 2.42 (3H, s), 2.45—3.10 (10H, m), 2.90 (3H, s), 3.85 (3H, s), 4.34 (1H, m), 4.47 (2H, s), 6.29 (1H, d, *J*=7.0), 7.1—7.4 (5H, m), 7.85 (1H, d, *J*=15.0), 8.67 (1H, d, *J*=8.0). MS:  $m/z$ , 429 (MH<sup>+</sup>). The oil was converted into the oxalate in the usual manner. An analytical sample of the oxalate was obtained by crystallization from EtOH/<sup>i</sup>PrOH, mp 146—149 °C. *Anal*. Calcd for  $C_{24}H_{33}FN_4O_2 \cdot 2C_2O_2H_4$ . H2O: C, 53.67; H, 6.27; F, 3.03; N, 8.94. Found: C, 53.41; H, 6.01; F, 3.16; N, 8.90. IR cm<sup>-1</sup>: 3406, 1616, 1514. 4) A mixture of 66 (1.2 g, 2.8 mmol), EtOH (10 ml), and AcOH (10 ml) was hydrogenated over 10% Pd/C (0.12 g) at 50 °C for 2 h under atmospheric pressure. The catalyst was filtered through Celite, and the filtrate was concentrated to dryness. After addition of saturated aqueous NaHCO<sub>3</sub>, the mixture was extracted with CHCl<sub>3</sub>. The extract was washed with brine and concentrated to dryness. The residue was chromatographed on silica gel with CHCl<sub>3</sub>/MeOH=10/1 to give 0.7 g (74%) of **113** as a pale yellow oil, which was converted into the maleate in the usual manner. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 1.11 (3H, t, *J*=7.5), 2.68 (3H, s), 2.7-3.3 (13H, m), 3.94 (3H, s), 4.31 (1H, m), 6.16 (4H, s, maleic acid), 6.28 (1H, d,  $J=7.5$ ), 6.31 (1H, br), 7.48 (1H, d,  $J=13.5$ ), 8.43 (1H, d,  $J=7.5$ ). MS:  $m/z$ , 339 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3364, 1624, 1576, 1528, 1487, 1364.

*N***-(1-Ethyl-4-methylhexahydro-1,4-diazepin-6-yl)-5-hydroxymethyl-2 methoxy-4-methylaminobenzamide (116)** NaBH<sub>4</sub> (0.35 g, 9.3 mmol) was added to a solution of **115** (1.1 g, 3.2 mmol) in MeOH (20 ml) at room temperature. The reaction mixture was stirred at the same temperature for 4 h and concentrated to dryness. The residue was taken into  $H_2O$  and  $CHCl<sub>3</sub>$ , and the organic layer was separated, washed with brine, and dried over anhydrous MgSO4. After evaporation of the solvent, the residue was chromatographed on silica gel with CHCl<sub>3</sub>/MeOH=10/1 to 7/1 to give 0.7 g (63%) of 116 as a pale yellow solid. <sup>1</sup>H-NMR (DMSO- $d_6$ )  $\delta$ : 1.06 (3H, t, *J*=7.0), 2.0 (1H, br), 2.40 (3H, s), 2.5—3.05 (10H, m), 2.92 (3H, d, *J*=5.0), 3.96 (3H, s), 4.33 (1H, m), 4.63 (2H, s), 5.32 (1H, br, *J*=5.0), 6.10 (1H, s), 7.87 (1H, s), 8.56 (1H, d, J=8.0). MS:  $m/z$ , 351 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3312, 1632, 1601, 1522, 1456, 1352, 1286, 1215.

*N***-(1-Ethyl-4-methylhexahydro-1,4-diazepin-6-yl)-2-methoxy-4-methylamino-5-sulfamoylbenzamide (119)** 1) 4-Chloro-2-methoxybenzoic acid (**67**, 10.0 g, 54 mmol) was slowly added in small portions to cold (*ca.* 0 °C)  $CISO<sub>3</sub>H$  (50 ml). The mixture was warmed to room temperature and stirred until **67** dissolved completely. The solution was then slowly warmed to 60— 70 °C, stirred at this temperature for 2 h, and cooled to room temperature. The reaction mixture was carefully poured into crushed ice. (Caution:  $CISO<sub>3</sub>H$  reacts violently with  $H<sub>2</sub>O$ !) The solid precipitate was collected by filtration, washed with H<sub>2</sub>O, and dried to give  $13.4 g$  (88%) of 4-chloro-5chlorosulfonyl-2-methoxybenzoic acid (**68**) as a white powder, mp 190— 192 °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 3.84 (3H, s), 7.12 (1H, s), 8.19 (1H, s), 13.87 (1H, s). *Anal.* Calcd for C<sub>8</sub>H<sub>6</sub>Cl<sub>2</sub>O<sub>5</sub>S: C, 33.70; H, 2.12; Cl, 24.87; S, 11.25. Found: C, 33.62; H, 2.25; Cl, 24.60; S, 11.21. MS:  $m/z$ , 284 (M<sup>+</sup>). IR cm<sup>-1</sup>: 1703, 1684, 1587, 1389, 1254. 2) To 28% aqueous NH<sub>3</sub> (50 ml) was added **68** (12.0 g, 42 mmol) in small portions under ice-cooling. The mixture was warmed to 40—50 °C and stirred for 2 h at this temperature. After being cooled to *ca.* 5 °C, the reaction mixture was acidified with 35% aqueous HCl. The solid was collected by filtration, washed with  $H_2O$ , and dried to give 10.0 g (89%) of 4-chloro-2-methoxy-5-sulfamoylbenzoic acid (**69**) as a white powder, mp 224—226 °C. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 3.83 (3H, s), 7.41 (1H, s), 7.59 (2H, s), 8.26 (1H, s), 13.15 (1H, s). *Anal.* Calcd for C<sub>8</sub>H<sub>8</sub>ClNO<sub>5</sub>S: C, 36.17; H, 3.04; Cl, 13.34; N, 5.27; S, 12.07. Found: C, 35.95; H, 3.11; Cl, 13.31; N, 5.21; S, 11.95. MS:  $m/z$ , 265 (M<sup>+</sup>). IR cm<sup>-1</sup>: 3398, 3290, 1694, 1595, 1556, 1352, 1244. 3) A mixture of **69** (1.7 g, 6.4 mmol), **8a** (1.2 g, 7.6 mmol), EDC (1.4 g, 7.3 mmol), and CH<sub>2</sub>Cl<sub>2</sub> (20 ml) was stirred at room temperature for 4 h. The reaction mixture was concentrated to dryness, and the residue was dissolved in  $H_2O$  and washed with Et<sub>2</sub>O. After addition of 2  $\alpha$  aqueous NaOH and solid K<sub>2</sub>CO<sub>3</sub>, the mixture was extracted with CHCl<sub>3</sub>. The extract was concentrated to dryness, and the residue was chromatographed on silica gel with CHCl<sub>3</sub>/MeOH=9/1 to 7/1 to give 1.6 g (62% from **69**) of 4-chloro-*N*-(1-ethyl-4-methylhexahydro-1,4-diazepin-6-yl)-2-methoxy-5-sulfamoylbenzamide (**70**) as a pale yellow amorphous solid. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>) δ: 0.96 (3H, t, *J*=7.5), 2.32 (3H, s), 2.4—2.9 (10H, m), 4.02 (3H, s), 4.11 (1H, m), 7.45 (1H, s), 7.58 (2H, s), 8.46 (1H, s), 8.74 (1H, d, *J*=7.5). MS:  $m/z$ , 405 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 1697, 1647, 1593, 1529, 1339, 1242, 1169. The amorphous solid was converted into the dioxalate in the usual manner. An analytical sample of the dioxalate was obtained by crystallization from MeOH/EtOH, mp 192—194 °C. *Anal.*

Calcd for  $C_{16}H_{25}CIN_4O_4S \cdot 2C_2O_2H_4 \cdot 3/4H_2O$ : C, 40.14; H, 5.14; Cl, 5.92; N, 9.36; S, 5.36. Found: C, 39.90; H, 5.26; Cl, 5.83; N, 9.27; S, 5.25. 4) A mixture of  $70$  (1.2 g, 3.0 mmol) and  $30\%$  NH<sub>2</sub>Me in EtOH (60 ml) in a sealed tube was heated at 140 °C for 30 h and cooled to room temperature. The volatiles were evaporated, and 1 N aqueous NaOH was added to the residue. The mixture was extracted with AcOEt, and the extract was concentrated to dryness to give 1.0 g (84%) of 119 as a white solid. <sup>1</sup>H-NMR (DMSO- $d_6$ )  $\delta$ : 0.97 (3H, t, *J*=7.5), 2.30 (3H, s), 2.36—2.87 (10H, m), 2.92 (3H, d, *J*=5.0), 3.98 (3H, s), 4.07 (1H, m), 6.24 (1H, s), 6.30 (1H, br, *J*=5.0), 7.24 (2H, s), 8.30 (1H, s), 8.48 (1H, d, *J*=7.05). MS:  $m/z$ , 400 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3420, 3379, 1626, 1597, 1514, 1317.

**6-Amino-2,3-dimethoxy-***N***-(1-ethyl-4-methylhexahydro-1,4-diazepin-6-yl)benzamide 7/4 Oxalate (124)** SOCl<sub>2</sub> (0.36 ml, 4.9 mmol) was added to a suspension of 6-(*tert*-butoxycarbonyl)amino-2,3-dimethoxybenzoic acid (**62**, 0.9 g, 3.0 mmol), DMF (1 drop), and toluene (20 ml) at room temperature. The mixture was heated at 60 °C for 1.5 h and cooled to room temperature. The volatiles were evaporated, and the residue was dissolved in toluene. The solvent was then reevaporated and the oily residue was used in the next step without further purification. A solution of the oil obtained above in CHCl3 (20 ml) was added dropwise to a solution of **8a** (0.48 g, 3.1 mmol) in  $CHCl<sub>3</sub>$  (10 ml) under ice-cooling. The mixture was stirred at room temperature for 3 h, washed successively with  $H_2O$ , saturated aqueous NaHCO<sub>3</sub>, and brine, and dried over anhydrous MgSO<sub>4</sub>. After evaporation of the solvent, the residue was chromatographed on silica gel with  $CHCl<sub>3</sub>/MeOH=10/1$  to afford 0.5 g (49%) of **124** as a brown oil. The oil was converted into the oxalate in the usual manner. <sup>1</sup>H-NMR (DMSO- $d_6$ )  $\delta$ : 1.10 (3H, t, *J*=7), 2.62 (3H, s), 2.86 (2H, q, J=7), 3.04 (2H, s), 2.9–3.25 (8H, m), 3.69 (3H, s), 3.73 (3H, s), 4.48 (1H, m), 6.43 (1H, d,  $J=9$ ), 6.92 (1H, d,  $J=9$ ), 8.47 (1H, d, J=9). MS:  $m/z$ , 337 (MH<sup>+</sup>). IR cm<sup>-1</sup>: 3358, 1607, 1491, 1332.

**General Procedure for the Preparation of the Benzamide and Carboxamide Derivatives [71, 72, 80—82, 85—97, 98—112, 114, 115, 117, 118, 120—123, 125—130, (***R***)- and (***S***)-81, (***R***)- and (***S***)-82, (***R***)- and (***S***)- 110, (***R***)- and (***S***)-112]** A mixture of the benzoic acid derivative or 1*H*benzotriazole- and 1*H*-benzimidazole-5-carboxylic acids (10 mmol), amine (11 mmol), EDC (12 mmol), and CH<sub>2</sub>Cl<sub>2</sub> (80 ml) was stirred at room temperature for  $4-5$  h. The reaction mixture was washed successively with  $H<sub>2</sub>O$ , 10% aqueous NaOH, and brine, and dried over anhydrous MgSO4. The solvent was evaporated, and the residue was chromatographed on silica gel. The free base thus obtained was recrystallized from the solvent shown in Tables 1—5 or converted into the fumarate, the maleate, or the oxalate in the usual manner, followed by recrystallization from the solvent shown in Tables 1— 5. Physical data for the benzamide and the carboxamide derivatives are listed in Tables 1—5.

**Binding Assays for Dopamine**  $D_2$  **and**  $5-HT_3$  **Receptors** The binding assays for the dopamine  $D_2$  and the 5-HT<sub>3</sub> receptors were carried out according to the method described in previous papers.<sup>20,32)</sup>

**Effect on Apomorphine-Induced Emesis in Dogs**48) Male beagle dogs, weighing 10—16 kg, were used. Groups of three to six dogs received a subcutaneous injection of apomorphine hydrochloride (0.3 mg/kg) 2 h after pretreatment with test compounds (1.0 mg/kg, *p.o.*). Emesis inhibition was evaluated by the frequency of emetic episodes 1 h after treatment with apomorphine hydrochloride.

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