Isolation and Structure Elucidation of Two New Alkaloids, Pandamarilactonine-C and -D, from *Pandanus amaryllifolius* and Revision of Relative Stereochemistry of Pandamarilactonine-A and -B by Total Synthesis

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Two new pyrrolidine alkaloids, pandamarilactonine-C and -D, were isolated from *Pandanus amaryllifolius*. Based on the total synthesis of pandamarilactonine-C and its related alkaloid, pandamarilactonine-A, the relative stereochemistry of pandamarilactonine-A and -B, which was previously proposed by spectroscopic analysis, was revised.

Key words pyrrolidine alkaloid; *Pandanus*; pandamarilactonine; total synthesis; structure elucidation; structure revision

The genus Pandanus (Pandanaceae) comprises approximately 600 species that are widely distributed in tropical and subtropical regions. Several Pandanus species are used as a remedy for toothache and rheumatism, and as diuretic, cardiotonic, etc.¹⁾ In a recent pharmacological screening, the hypoglycemic effect of an extract of *P. odorus* was noted.²⁻⁴⁾ In our continuing search for structurally unique and biologically active *Pandanus* alkaloids, new pyrrolidine alkaloids were found in *P. amaryllifolius* ROXB.^{5–7)} Further investigation of the minor constituents in the fresh leaves of this plant resulted in the isolation of two new alkaloids, pandamarilactonine-C and -D (1, 2). In this communication, we describe the structure elucidation of these alkaloids as well as the total synthesis of pandamarilactonine-C (1) and its related alkaloid, pandamarilactonine-A (3), which resulted in the revision of the relative stereochemistry of pandamarilactonine-A and -B(3, 4), which was previously proposed based on the results of spectroscopic analysis.

The crude alkaloid fraction (1.10 g), which was obtained from young leaves of *P. amaryllifolius*,⁵⁾ was initially separated by SiO₂ column chromatography using MeOH/CHCl₃ gradient, and then the 2—5% MeOH/CHCl₃ eluate was subjected to SiO₂ medium pressure liquid chromatography using 2% EtOH/CHCl₃ to give 20 mg of pandamarilactonine-C and 4 mg of pandamarilactonine-D.

The two new compounds, pandamarilactonine-C $(1)^{8}$ and pandamarilactonine-D $(2)^{9}$ gave respectively m/z 318.1691 $[M+H]^+$ and m/z 318.1704 $[M+H]^+$ by high-resolution FAB-MS, which established the molecular formulas of the two compounds as $C_{18}H_{23}NO_4$, and indicated that these are 1303

isomers of coexisting alkaloids, pandamarilactonine-A and B.5) The UV, ¹H- and ¹³C-NMR spectra of pandamarilactonine-C and -D were very similar to those of pandamarilactonine-A and B, which comprised γ -butylidene- α -methyl α . β unsaturated γ -lactone and pyrrolidinyl α,β -unsaturated γ lactone residues, indicating that these four alkaloids are stereoisomers. Unambiguous assignment of all the carbons and protons in $1^{(8)}$ and $2^{(9)}$ was realized using ${}^{1}H^{-1}H$ correlated spectroscopy (COSY), heteronuclear multiple quantum coherence (HMQC), and heteronuclear multiple bond correlation (HMBC) spectra. Comparing the ¹³C-NMR data of 1 and 2 with those of 3 and $4^{(5)}$ a significant difference was observed in the chemical shifts at the C-4 position. The signals of 1 (δ 130.3) and 2 (δ 133.9) appeared in the higher field than those of 3 (δ 137.7) and 4 (δ 137.7), which can be reasonably interpreted in terms of the γ -gauche effect due to the E configuration in the γ -alkylidene- γ -lactone moiety of both 1 and 2. Differential nuclear Overhauser effect (NOE) experiments on 1 and 2, i.e., observation of a clear peak enhancement from H-4 to H-7, also demonstrated the \hat{E} configuration of the γ -alkylidenebutenolide moiety. All of the above findings enabled us to elucidate the molecular structures of the new alkaloids (1, 2) except for the stereochemistry of the vicinal asymmetric center at the C-14 and C-15 positions.

To confirm the structures and the relative stereochemistry at the C-14 and C-15 positions in the new alkaloids, we attempted the total synthesis of pandamarilactonines. The lower part of the alkaloids, *i.e.*, the pyrrolidinyl α,β -unsaturated γ -lactone residue, was synthesized according to the procedure of Martin *et al.*¹⁰ as follows. Compound (7), whose stereochemistry at the vicinal positions was established to be *threo* by X-ray analysis,¹⁰ was prepared by vinylogous Mannich coupling reaction of **5** and **6**. The protecting group on the nitrogen in **7** was removed by treatment with TMSI in CH₃CN to give **8** in 94% yield. The iodide (**13**) corresponding to the upper part of pandamarilactonines was prepared by a three-step operation: i) condensation of the alde-





Chart 1. Total Synthesis of (\pm) -Pandamarilactonine-A and -C

hyde (9) with the siloxyfuran (10) in the presence of $BF_3 \cdot Et_2O$; ii) dehydration of the resultant aldol adduct (11) with trifluoromethanesulfonic anhydride and pyridine¹¹⁾ (48% overall yield); and iii) halogen exchange of (12) with NaI in acetone (47% yield). The thus-obtained C9 unit (13) consisted of Z and E isomers in the ratio of 4.6:1, and was then condensed with secondary amine (8) in acetonitrile in the presence of K₂CO₃ to give two pandamarilactonines in 33% and 7% yields, both of which possessed the threo form at the C-14 and C-15 positions. The major synthetic product having the Z configuration was completely identical with natural pandamarilactonine-A and the minor one having the Econfiguration was identical with natural pandamarilactonine-C, respectively, by direct comparison of their chromatographic behavior, as well as their mass, and ¹H- and ¹³C-NMR spectra. Based on the results of this synthetic study, the structures of pandamarilactonine-C, -D, -A, and -B were unambiguously established to be formulas 1 to 4, respectively. The stereochemistry at the vicinal centers in pandamarilactonine-A and -B was analyzed by spectroscopic methods and the structures having the erythro-form for pandamarilactonine-A and the *threo*-form for B have been proposed.⁵⁾ However, they should be revised as above.

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- Pandamarilactonine-C (1): An amorphous powder, $[\alpha]_D^{23} + 26.2^\circ$ (c= 8) 0.99, CHCl₃), UV (MeOH) λ_{max} nm (log ε): 275 (4.23), 220 (sh), 201 (3.67). IR (neat) $v_{\text{max}} \text{ cm}^{-1}$: 1758 (lactone). FAB-MS (NBA) m/z; 318 [M+H]⁺. HR-FAB-MS (NBA): Calcd for C₁₈H₂₃NO₄: 318.1704, Found 318.1691. ¹H-NMR (500 MHz, CDCl₃) δ: 7.31 (1H, dd, J=0.9, 1.5 Hz, H-4), 7.04 (1H, d-like, J=1.5 Hz, H-16), 5.64 (1H, dd, J=8.2, 8.5 Hz, H-6), 4.80 (1H, ddd, J=1.8, 1.8, 5.5 Hz, H-15), 3.12 (1H, ddlike, J=6.7, 7.3 Hz, H-11), 2.91 (1H, ddd, J=7.9, 8.2, 11.9 Hz, H-9), 2.78 (1H, m, H-14), 2.45 (1H, ddd, J=5.8, 7.0, 11.9 Hz, H-9), 2.35 (1H, ddd, J=7.0, 8.2, 14.7 Hz, H-7), 2.25 (2H, m, H-7 and H-11), 2.02 (3H, d, J=0.9 Hz, H₃-21), 1.94 (3H, d-like, J=1.8 Hz, H₃-20), 1.71-1.82 (2H, m, H-12, H-13), 1.62-1.68 (3H, m, H₂-8, H-13), 1.45 (1H, m, H-12). ¹³C-NMR (125 MHz, CDCl₃) δ : 174.2 (C-18), 171.0 (C-2), 148.7 (C-5), 146.7 (C-16), 131.3 (C-17), 130.3 (C-4), 129.1 (C-3), 113.4 (C-6), 83.8 (C-15), 65.6 (C-14), 55.2 (C-9), 54.2 (C-11), 29.1 (C-8), 26.1 (C-12), 24.2 (C-7), 23.8 (C-13), 10.8 (C-20, C-21).
- 9) Pandamarilactonine-D (2): An amorphous powder, $[\alpha]_D^{25} 0^{\circ} (c=0.21, CHCl_3), UV (MeOH) \lambda_{max} nm (log <math>\varepsilon$): 276 (4.38), 234 (sh), 203 (4.30). IR (neat) v_{max} cm⁻¹: 1760 (lactone). FAB-MS (NBA) m/z; 318 [M+H]⁺. HR-FAB-MS (NBA): Calcd for $C_{18}H_{23}NO_4$: 318.1704, Found 318.1704. ¹H-NMR (500 MHz, CDCl₃) & 7.35 (1H, m, H-4), 6.99 (1H, dd, J=1.5, 1.7 Hz, H-16), 5.60 (1H, dd, J=8.3, 8.8 Hz, H-6), 4.74 (1H, m, H-15), 3.08 (1H, m, H-11), 2.77 (1H, m, H-9), 2.72 (1H, m, H-14), 2.42 (1H, m, H-9), 2.29 (2H, m, H₂-7), 2.25 (1H, m, H-11), 2.01 (3H, m, H₃-21), 1.94 (3H, dd-like, J=1.5, 2.0 Hz, H₃-20), 1.78–1.92 (4H, m, H₂-12, H₂-13), 1.63 (2H, m, H₂-8). ¹³C-NMR (125 MHz, CDCl₃) & 174.3 (C-18), 171.0 (C-2), 149.0 (C-5), 147.1 (C-16), 131.0 (C-17), 133.9 (C-4), 130.3 (C-3), 113.3 (C-6), 83.0 (C-15), 65.9 (C-14), 55.1 (C-9), 54.0 (C-11), 28.7 (C-8), 26.3 (C-12), 24.0 (C-7), 23.9 (C-13), 10.8 (C-20), 10.7 (C-21).
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