Synthesis and Antispasmodic Activity Evaluation of Bis-(Papaverine) Analogues

Jaskiran KAUR,^a Narendra Nath GHOSH,^b and Ramesh CHANDRA^{*,b}

^a Department of Chemistry, University of Pennsylvania; Philadelphia, Pennsylvania, 19104, U.S.A.: and ^bDr. B. R. Ambedkar Centre for Biomedical Research, University of Delhi; 110007, Delhi, India. Received October 10, 2003; accepted December 22, 2003

A new series of N-substituted bis-(tetrahydropapaverine) ring systems have been synthesised in expectation of better antispasmodic activity in comparison with papaverine. The synthesis of the targeted heterocycles is described along with a discussion of their structure activity relationship. The general synthetic methods of bis-(tetrahydropapaverine) analogues involve tetrahydropapaverine, various piperazines, diisocyanates and diisothiocyanates as starting materials. Pharmacological evaluation involves the *in vitro* antispasmodic activity on a freshly removed guinea pig ileum using a force displacement transducer amplifier connected to a physiograph. Among the analogues synthesized in the present study, N,N'-bis-[2-carbamoyl-1-(3,4-dimethoxybenzyl)-6,7dimethoxy-1,2,3,4-tetrahydroisoquinolinyl]piperazine (22), was found to be the most potent muscle relaxant (IC₅₀: 0.31 μ M).

Key words papaverine; tetrahydropapaverine; dimer; antispasmodic activity; piperazine; urea

Compounds containing symmetrical dimers of a number of natural molecules are of increasing importance in biochemical as well as in medicinal chemistry research.¹⁾ The bis(naphthalimide) DMP-840 (Fig. 1) shows antineoplastic activity against a variety of human solid tumor cell lines in culture^{2,3)} and as xenografts in nude mice,^{4,5)} and is in clinical trial.^{6,7)} Other examples under study include bis-(benzonapthalimides),⁸⁾ bis-(imidazoacridiones),^{9,10)} bis-(triazoloacridinones),¹¹⁾ bis-(anthracyclines),^{12,13)} bis-(acridine-4carboxamides),¹⁴⁾ bis-(phenazine-1-carboxamides).^{15,16)} and bis-(indeno[1,2-*b*]quinoline-6-carboxamides).¹⁷⁾ A comparative study of a variety of bis analogues with their monomeric counterparts has revealed that there are variable but significant gains in potency for the dimeric species.¹⁸⁾ Due to interesting biological activities and unique chemical structures, bis analogues of various naturally occurring alkaloids serve as lead compounds for discovery of new drugs.¹⁹⁾

Dimeric analogues of a number of classes of chromophores joined by various cyclic and acyclic linkers are also of current interest as potential antispasmodic agents.²⁰⁾ Natural bis-(isoquinoline) alkaloids have emerged as an important structural class based upon their high degree of bio-



logical activity.²¹⁾ A typical example is tetrandrine,²²⁾ a potent antispasmodic agent. Papaverine, a benzylisoquinoline alkaloid, is a well known smooth muscle relaxant and a vasodilator.^{23,24)} Considering these factors and in continuation to our earlier efforts^{25,26)} of finding a potent antispasmodic agent, we envisaged that dimers of papaverine with suitable linkers at nitrogen can turn out to be more potent antispasmodic agents in comparison to papaverine.

Herein, we report the synthesis and antispasmodic activity of a series of bis-(tetrahydropapaverine) analogues with urea and acetyl linkers.

Results and Discussion

For the purpose of this study, numerous bis-(tetrahydropapverine) analogues 6-13 were readily prepared by direct reaction of tetrahydropapaverine (5) (2 molar equivalent) with the corresponding (i) substituted bis-(isocyanates) and (ii) substituted bis-(isothiocyanates) in acetonitrile (Chart 1). In a typical reaction amide 3 required for the synthesis of tetrahydropapaverine (5) was prepared by the condensation of homoveratryl amine (1) and homoveratric acid (2) in xylene with azeotropic removal of water. Cyclization of amide 3 to 3,4-dihydroisoquinoline (4) was achieved in Bischler-Napierlaski fashion, in refluxing toluene with phosphorous oxychloride. Reduction of 3,4-dihydroisoquinoline with sodium borohydride in methanol resulted in the formation of tetrahydropapaverine 5 (Chart 1).²⁶⁾ Products 6-13 were obtained in 80-85% yield and were purified by column chromatography.

Reaction of tetrahydropapaverine (5) with chloroacetylchloride and triethylamine in dichloromethane lead to the formation of *N*-(chloroacetyl)tetrahydropapaverine (14). The reaction of tetrahydropapaverine (5) and *N*-(chloroacetyl)tetrahydropapaverine (14) in acetone in the presence of potassium carbonate and tetrabutylammonium iodide at room temperature yielded 15 (Chart 2). The direct reaction of *N*-(chloroacetyl)tetrahydropapaverine (14) (2 molar equivalent) with piperazine or *cis*-2,6-dimethylpiperazine or homopiperazine (1 molar equivalent) in acetone in the presence of







potassium carbonate and tetrabutylammonium iodide yielded **16—18** respectively (Chart 3). Products **16—18** were purified by column chromatography eluting with ethylacetate and hexane and obtained in 73—76% yields.

Tetrahydropapaverine efficiently displaced one of the imidazole rings of carbonyldiimidazole²⁶⁾ to give tetrahydropapaverine carbamoyl imidazole **19** on refluxing it with carbonyldiimidazole (CDI) in tetrahyrofuran (THF) for 24 h. The reaction of methyl iodide with carbamoyl imidazole **19** (4 molar equivalents) in acetonitrile for one day at room temperature produced the imidazolium salt quantitatively. This imidazolium salt required no additional purification for final conversion to ureas. Addition of tetrahydropapaverine to a solution of imidazolium salt in dichloromethane with triethylamine at room temperature afforded tetra substituted urea derivative **21** (Chart 4). It was purified by column chromatography and obtained in 79% yield. The activation of the leaving imidazole ring was found to be necessary for the formation of desired ureas since the carbamoyl imidazoles were unreactive towards tetrahydropapaverine even under refluxing conditions for prolonged periods. Addition of triethylamine (1 molar equivalent) resulted in improved yields of ureas.

The reaction of imidazolium salt **20** (2 molar equivalent) with piperazine or *cis*-2,6-dimethylpiperazine or homopiperazine (1 molar equivalent) in dichloromethane and triethylamine (Chart 5) afforded compounds **22**—**24** respectively. These compounds were purified by column chromatography and obtained in 75—80% yield.

Antispasmodic activity studies of compounds 6-13, 15-18, and 21-24 were performed on guinea pig ileum. Acetylcholine (10^{-3} M) was used to induce sustained contraction in guinea pig ileum. When the contraction response to acetylcholine reached a steady level, papaverine was added cumulatively. Papaverine inhibited the acetylcholine induced contraction in a concentration dependent manner. Similarly, acetylcholine induced contraction inhibition by tetrahydropapaverine derivatives 6-13, 15-18, and 21-24 were studied. Dose-response curves were obtained for papaverine and all these derivatives. The results revealed that most of the test compounds 6-13, 15-18, and 21-24 more potently inhibited the acetylcholine induced contractions than papaverine. Table 1 summarizes the concentration of papaverine and test compounds producing 50% relaxation (IC₅₀) of acetylcholine induced contraction for compounds 6-13 and Table 2 summarizes the IC₅₀ values of compounds 15–18, and 21–24. IC50 values were determined graphically from dose-response curves obtained by measuring antispasmodic activity at six concentrations (0.1, 0.5, 1.0, 5.0, 10, $20 \,\mu\text{M}$) of the compounds in duplicate of triplicates.

The structure activity relationship of these compounds has





been focused on alterations in the linker chain. Three types of linkers have been introduced. First category of linkers consists of two urea moieties linked by spacers like benzene, substituted benzene, cyclohexane and butyl chain (6–13). Second category has piperazinyl, substituted piperazinyl ring placed between two acetyl groups, as linkers (15–18). Third category consists of piperazinyl, substituted piperazinyl ring placed between two carbamoyl groups, as linkers (21–24).

Among the three types of linkers, the compounds with third category of linkers (21-24) were found to be most potent. The corresponding compounds with CH₂ spacers (15-18), exhibited a decrease in antispasmodic activity. The compounds with trisubstituted urea spacers (6-13) were found to be least active among the bis-(tetrahydropapaverine) derivatives reported.

The bis-(tetrahydropapaverine) derivatives with trisubsti-

tuted urea spacers 6—13 exhibited antispasmodic activities not significantly different from papaverine. Compound 6, with a benzene ring as a spacer, introduced between the two amide groups, gave IC_{50} value of 6.17 μ M. Its thio analogue 9 exhibited a small decrease in antispasmodic activity. Substitution of a cyclohexane ring 8 in place of benzene ring decreased the antispasmodic activity (IC_{50} : 9.31 μ M). Compound 10 with an aliphatic chain as the spacer between the two thiomide groups exhibited IC_{50} value further lower as compared to compound 8. Compounds 11 and 12 having methyl group substituted in benzene ring were not as good antispasmodic agents as papaverine. They exhibited IC_{50} values similar to that exhibited by compounds 8 and 10. Substitution of a chloro group in addition to a methyl group in the benzene ring (13) enhanced the antispasmodic activity.

The bis-(tetrahydropapaverine) derivatives with acetyl







Table 2. Concentration of Bis-(tetrahydropapaverine) Derivatives 15-18 and 21-24 Producing 50% Relaxation (IC₅₀) of Acetylcholine Induced Contractions





a) IC_{50} of papaverine=7.31 μ M. IC	C_{50s} were calculated	using a mean of at lea	ist 3
measurements (all duplicates) for 6 con	ncentrations in the ra	ange 0.1—20 µм.	

linkers **15**—**18** exhibited significantly higher antispasmodic activities in comparison to papaverine. Compound **15** exhibited an IC₅₀ value of 0.61 μ M. Introduction of a piperazine ring **16** between the two acetyl groups increased the antispasmodic activity (IC₅₀: 0.43 μ M). Methyl substitution (**17**) in the piperazine ring at positions 2 and 6, lead to a fall in the activity (IC₅₀: 1.07 μ M). Substitution of homopiperazine ring (**18**) in place of piperazine ring slightly increased the activity (IC₅₀: 0.95 μ M).

Among the tetra substituted urea derivatives **21**—**24**, the IC₅₀ value of **21** was found to be $0.42 \,\mu$ M. Addition of a piperazine spacer between the two carbamoyl groups further increased the antispasmodic activity (IC₅₀: $0.31 \,\mu$ M). Methyl substitution (**23**) on the piperazine ring at positions 2 and 6, lead to a fall in the activity (IC₅₀: $0.75 \,\mu$ M). Substitution of homopiperazine ring (**24**) in place of piperazine ring slightly increased the activity (IC₅₀: $70 \,\mu$ M).

Compound **10** is the least potent and compound **22** is the most active bis-(tetrahydropapaverine) analogue of the series.

Conclusion

N-Substituted bis-(tetrahydropapaverine) derivatives showed better antispasmodic activity in comparison with papaverine in guinea pigs in preliminary screenings. Compounds with tetra substituted urea linkers were found to be more active than compounds with trisubstituted urea linkers and compounds with acetyl linkers. Compound **22** was found

a) IC_{50} of papaverine=7.31 μ M. $IC_{50\delta}$ were calculated using a mean of at least 3 measurements (all duplicates) for 6 concentrations in the range 0.1—20 μ M.

to be the most potent compound among the series. However, contrary to our predictions, none of the bis analogues were able to beat the best compound of our previously reported²⁶⁾ piperazinyl based tetra substituted urea derivative of papaverine.

Experimental

Melting points are recorded in open capillary tubes on Buchi melting point B-540 instrument and are uncorrected. Solvent system used throughout the experimental work for running TLC plates was ethylacetate–hexane. The ¹H-NMR spectra were recorded in CDCl₃ as solvent (using TMS as internal standard) on a Bruker Avance Spectrospin 300 instrument at 300 MHz. Mass spectra were run on a Perkin Elmer LCMS API-2000 mass spectrometer. IR spectra were recorded using KBr discs on a Shimadzu FTIR-8300 spectrophotometer. Elemental analysis was carried out on Heraeus Rapid CHN analyser.

N,*N*'-[1,4-Benzene-{diyl-bis-(iminocarbonyl)}-bis-{1-(3,4-dimethoxybenzyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline}] (6) A solution of 1,4-phenylenediisocyanate (0.192 g, 1.2 mmol) in dry acetonitrile (5 ml) was added slowly to a solution of tetrahydropapaverine **5** (0.686 g, 2 mmol) in dry acetonitrile (5 ml). The mixture was stirred at room temperature for 2 h. After the completion (TLC) of the reaction, evaporated off the solvent, diluted the residue with water (25 ml) and extracted with ethylacetate (3×25 ml). The collective organic portion was washed with brine and dried (Na_2SO_4). It was concentrated under reduced pressure and the residue was chromatographed on silica gel using ethylacetate–hexane as eluent. A final recrystallization from ethylacetate–hexane (15:85) afforded pure **6** as a viscous solid (0.685 g, 81%). ¹H-NMR (CDCl₃, 300 MHz) δ : 2.60 (4H, m), 3.10 (4H, m), 2.97 (4H, m), 3.29 (4H, m), 3.73 (2×3H, s), 3.78 (2×3H, s), 3.81 (2×3H, s), 3.85 (2×3H, s), 5.01 (2H, m), 5.56 (2H, s), 6.41 (2H, s), 6.56 (2H, s), 6.79 (4H, m), 6.97 (6H, m). IR (KBr) $v \text{ cm}^{-1}$: 1639. MS (*m*/*z*): 847 (M+H)⁺. *Anal.* Calcd for C₄₈H₅₄N₄O₁₀: C, 68.06; H, 6.42; N, 6.61. Found: C, 68.15; H, 6.31; N, 6.52.

N,*N*'-[4,4'-Methylene-{diyl-bis-(phenyliminocarbonyl)}-bis-{1-(3,4-dimethoxybenzyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline}] (7) Synthesized, from tetrahydropapaverine (5) and 4,4'-methylene-bis-(phenylisocyanate), in a manner similar to that described for 6. The product was obtained as a light yellow solid (80%) mp 122—123 °C. ¹H-NMR (CDCl₃, 300 MHz) δ : 2.67 (4H, m), 3.14 (4H, m), 3.43 (4H, m), 3.79 (2×3H, s), 3.82 (2×3H, s), 3.83 (2×3H, s), 3.87 (2×3H, s), 4.11 (2H, m), 5.02 (2H, s), 5.66 (2H, s), 6.48 (2H, s), 6.64 (2H, s), 6.74 (4H, m), 6.83 (4H, m), 6.99 (6H, m). IR (KBr) v cm⁻¹: 1640.5. MS (*m*/*z*): 937 (M+H)⁺. *Anal.* Calcd for C₅₅H₆₀N₄O₁₀: C, 70.49; H, 6.45; N, 5.97. Found: C, 70.35; H, 6.39; N, 5.82.

N,*N*'-[*trans*-Cyclohexane-{diyl-bis-(iminothiocarbonyl)}-bis-{1-(3,4-dimethoxybenzyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline}] (8) Synthesized, from 5 and *trans*-1,4-cyclohexanediisothiocyanate, in a manner similar to that described for 6. The product was obtained as a light yellow solid (85%) mp 118—120 °C. ¹H-NMR (CDCl₃, 300 MHz) δ : 1.47 (8H, m), 2.32 (4H, m), 2.78 (4H, m), 2.93 (4H, s), 3.23 (2H, s), 3.71 (2×3H, s), 3.77 (2×3H, s), 3.83 (4×3H, s), 4.88 (2H, m), 6.33 (2H, m), 6.58 (6H, m), 6.74 (2H, m). IR (KBr) vcm⁻¹: 1232. MS (*m*/*z*): 885 (M+H)⁺. *Anal.* Calcd for (C4₈H₆₀N₄O₈S₂: C, 65.13; H, 6.82; N, 6.32. Found: C, 65.51; H, 6.65; N, 6.41.

N,*N*'-[1,4-Benzene-{diyl-bis-(iminothiocarbonyl)}-bis-{1-(3,4dimethoxybenzyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline}] (9) Synthesized, from 5 and 1,4-phenylenediisothiocyanate, in a manner similar to that described for 6. The product was obtained as a white solid (85%) mp 126—128 °C. ¹H-NMR (CDCl₃, 300 MHz) δ : 2.80 (4H, m), 3.24 (8H, m), 3.75 (2×3H, s), 3.83 (2×3H, s), 3.87 (4×3H, s), 5.29 (2H, m), 6.48 (2H, m), 6.60 (6H, m), 6.73 (10H, m). IR (KBr) v cm⁻¹: 1260. MS (*m*/*z*): 879 (M+H)⁺. *Anal.* Calcd for C₄₈H₅₄N₄O₈S₂: C, 65.58; H, 6.19; N, 6.37. Found: C, 65.51; H, 6.15; N, 6.42.

N,*N*'-[1,4-Butane-{diyl-bis-(iminothiocarbonyl)}-bis-{1-(3,4dimethoxybenzyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline}] (10) Synthesized, from 5 and 1,4-butanediisothiocyanate, in a manner similar to that described for 6. The product was obtained as a white solid (83%) mp 56—58 °C. ¹H-NMR (CDCl₃, 300 MHz) δ: 1.64 (4H, m), 2.90 (4H, m), 2.98 (4H, m), 3.27 (4H, m), 3.58 (4H, m), 3.72 (2×3H, s), 3.84 (2×3H, s), 3.86 (4×3H, s), 5.12 (2H, m), 5.61 (2H, m), 6.30 (2H, s), 6.67 (2H, m), 6.79 (6H, m). IR (KBr) v cm⁻¹: 1254. MS (*m*/*z*): 859 (M+H)⁺. *Anal.* Calcd for C₄₆H₅₈N₄O₈S₂: C, 64.31; H, 6.80; N, 6.52. Found: C, 64.49; H, 6.75; N, 6.42.

N,*N*'-[1,3-Benzene-{diyl-bis-(iminocarbonyl)-2-methyl}-bis-{1-(3,4-dimethoxybenzyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline}] (11) Synthesized, from 5 and 2-methyl-*m*-phenylenediisocyanate, in a manner similar to that described for 6. The product was obtained as a white solid (81%) mp 171—172 °C. ¹H-NMR (CDCl₃, 300 MHz) δ : 2.70 (4H, m), 2.97 (4H, m), 3.14 (4H, m), 3.44 (3H, s), 3.76 (2×3H, s), 3.79 (2×3H, s), 3.88 (4×3H, s), 5.12 (2H, m), 5.70(2H, m), 6.30 (2H, m), 6.67 (6H, m), 6.79 (3H, m), 6.91 (2H, m). IR (KBr) vcm⁻¹: 1618. MS (*m/z*): 861 (M+H)⁺. *Anal.* Calcd for C₄₉H₅₆N₄O₁₀: C, 68.35; H, 6.55; N, 6.50. Found: C, 68.27; H, 6.65; N, 6.41.

N,*N*'-[1,3-Benzene-{diyl-bis-(iminocarbonyl)-4-methyl}-bis-{1-(3,4-dimethoxybenzyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline}] (12) Synthesized, from 5 and 4-methyl-*m*-phenylenediisocyanate, in a manner similar to that described for 6. The product was obtained as a viscous solid (82%). ¹H-NMR (CDCl₃, 300 MHz) δ : 2.71 (4H, m), 2.95 (4H, m), 3.11 (4H, m), 3.31 (3H, s), 3.73 (2×3H, s), 3.79 (2×3H, s), 3.86 (4×3H, s), 5.10 (2H, m), 5.70 (2H, m), 6.30 (2H, m), 6.71 (6H, m), 6.80 (3H, m), 7.01 (2H, m). IR (KBr) v cm⁻¹: 1620. MS (*m*/*z*): 861 (M+H)⁺. *Anal.* Calcd for C₄₉H₅₆N₄O₁₀: C, 68.35; H, 6.55; N, 6.50. Found: C, 68.37; H, 6.63; N, 6.42.

 $N_{\rm e}N'$ -[1,3-Benzene-{diyl-bis-(iminocarbonyl)-4-chloro-6-methyl}-bis-{1-(3,4-dimethoxybenzyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline}] (13) Synthesized, from 5 and 4-chloro-6-methyl-*m*-phenylenediisocyanate, in a manner similar to that described for 51. The product was obtained as a viscous solid (84%). ¹H-NMR (CDCl₃, 300 MHz) & 2.68 (4H, m), 2.98 (4H, m), 3.24 (4H, m), 3.61 (3H, s), 3.72 (2×3H, s), 3.79 (2×3H, s), 3.87 (4×3H, s), 5.13 (2H, m), 5.85 (2H, m), 6.30 (2H, m), 6.67 (6H, m), 7.05 (2H, m), 7.99 (2H, s), 8.05 (1H, s). IR v(KBr) cm⁻¹: 1654. MS (*m/z*): 895.5 (M+H)⁺. *Anal.* Calcd for C₄₉H₅₅N₄O₁₀Cl: C, 65.72; H, 6.19; N, 6.25. Found: C, 65.57; H, 6.24; N, 6.41.

N,*N*'-Bis-[2-acetyl-1-(3,4-dimethoxybenzyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline] (15) To a solution of 1-(3,4-dimethoxybenzyl)-

2-chloroacetyl-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline (14) (0.85 g, 2 mmol) and tetrahydropapaverine (5) (0.68 g, 2 mmol) in acetone (10 ml) was added activated K₂CO₃ (1 g, 7.2 mmol) and tetrabutylammonium iodide (0.18 g, 0.5 mmol). The mixture was stirred for 12 h at room temperature. After the completion (TLC) of reaction, acetone was evaporated and diluted the reaction mixture with water (25 ml) and extracted with ethylacetate (3×25 ml). The collective organic portion was washed with brine and dried over Na₂SO₄. It was concentrated under reduced pressure and the residue was chromatographed on silica gel using ethylacetate–hexane as eluent to obtain pure **47** as a white solid (1.08 g, 75%) mp 74—75 °C. ¹H-NMR (CDCl₃, 300 MHz) δ : 2.45 (4H, m), 3.04 (4H, m), 3.39 (4H, m), 3.50 (2H, s), 3.74 (2×3H, s), 3.81 (2×3H, s), 3.87 (4×3H, s), 5.07 (2H, m), 6.31 (2H, s), 6.61 (2H, s), 6.80 (6H, m). IR (KBr) v cm⁻¹: 1635. MS (*m/z*): 727 (M+H)⁺. *Anal.* Calcd for C₄₂H₅₀N₂O₉: C, 69.40; H, 6.93; N, 3.85. Found: C, 69.51; H, 6.71; N, 3.81.

N,N'-Bis-[2-acetyl-1-(3,4-dimethoxybenzyl)-6,7-dimethoxy-1,2,3,4tetrahydroisoquinolinyl]piperazine (16) To a solution of 1-(3,4dimethoxybenzyl)-2-chloroacetyl-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline (14) (0.85 g, 2 mmol) and piperazine (0.086 g, 1 mmol) in acetone (10 ml) was added activated K2CO3 (1 g, 7.2 mmol) and tetrabutylammonium iodide (0.18 g, 0.5 mmol). The mixture was stirred for 12 h at room temperature. After the completion (tlc) of reaction, acetone was evaporated and diluted the reaction mixture with water (25 ml) and extracted with ethylacetate $(3 \times 25 \text{ ml})$. The collective organic portion was washed with brine and dried over Na2SO4. It was concentrated under reduced pressure and the residue was chromatographed on silica gel using ethylacetate-hexane as eluent to obtain pure 16 as a pale yellow solid (0.64 g, 76%) mp 80-82 °C. ¹H-NMR (CDCl₃, 300 MHz) δ: 2.84 (8H, m), 2.90 (4H, m), 2.93 (4H, m), 3.09 (4H, m), 3.62 (4H, s), 3.69 (2×3H, s), 3.73 (2×3H, s), 3.79 (4×3H, s), 5.07 (2H, m), 6.25 (1H, s), 6.27 (1H, s), 6.60 (2H, m), 6.75 (6H, m). IR (KBr) $v \text{ cm}^{-1}$: 1638. MS (m/z): 853 (M+H)⁺. Anal. Calcd for C₄₈H₆₀N₄O₁₀: C, 67.58; H, 7.08; N, 6.79. Found: C, 67.69; H, 7.17; N, 6.84.

 $N_{\rm s}N'$ -Bis-[2-acetyl-1-(3,4-dimethoxybenzyl)-6,7-dimethoxy-1,2,3,4tetrahydroisoquinolinyl]-2,6-dimethylpiperazine (17) Synthesized, from a solution of 14 and 2,6-dimethylpiperazine, in a manner similar to that described for 16. The product was obtained as a pale yellow solid (73%) mp 63—64 °C. ¹H-NMR (CDCl₃, 300 MHz) δ : 1.85 (6H, s), 2.76 (8H, m), 2.87 (4H, m), 2.92 (4H, m), 3.11 (4H, m), 3.55 (2H, s), 3.61 (2H, s), 3.71 (2×3H, s), 3.78 (2×3H, s), 3.89 (4×3H, s), 5.13 (2H, m), 6.24 (1H, s), 6.55 (1H, s), 6.65 (2H, m), 6.79 (6H, m). IR v (KBr) cm⁻¹: 1639. MS (*m*/*z*): 882 (M+H)⁺. *Anal.* Calcd for C₅₀H₆₄N₄O₁₀: C, 68.16; H, 7.32; N, 6.35. Found: C, 68.21; H, 7.21; N, 6.41.

N,*N*'-**Bis-[2-acetyl-1-(3,4-dimethoxybenzyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinolinyl]homopiperazine (18)** Synthesized, from a solution of **14** and homopiperazine, in a manner similar to that described for **16**. The product was obtained as a light yellow solid (75%) mp 100—102 °C. ¹H-NMR (CDCl₃, 300 MHz) δ : 2.53 (10H, m), 2.81 (8H, m), 2.97 (4H, m), 3.31 (4H, m), 3.72 (2×3H, s), 3.77 (2×3H, s), 3.83 (4×3H, s), 5.57 (2H, m), 6.15 (1H, s), 6.18 (1H, s), 6.69 (8H, m). IR (KBr) ν cm⁻¹: 1635. MS (*m/z*): 867 (M+H)⁺. *Anal.* Calcd for C₄₉H₆₂N₄O₁₀: C, 67.87; H, 6.04; N, 7.20. Found: C, 67.81; H, 6.11; N, 7.31.

N,*N*'-Bis-[1-(3,4-dimethoxybenzyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline]urea (21) To a solution of cationic carbamoyl imidazolium salt of 19 (0.88 g, 2 mmol) and tetrahydropapaverine (0.68 g, 2 mmol) in anhydrous dichloromethane was added triethylamine (0.27 ml, 2.0 mmol). The mixture was stirred at room temperature for 12 h, then washed with 1.0 m HCl (10 ml), the organic layer dried over anhydrous Na₂SO₄, filtered and concentrated under vacuum to yield the required tetra substituted urea as an oil which was column chromatographed (ethylacetate-hexane) to obtain pure 21 as a white solid (1.12 g, 79%) mp 68— 69 °C. ¹H-NMR (CDCl₃, 300 MHz) δ : 2.43 (4H, m), 2.92 (4H, m), 3.35 (4H, m), 3.72 (2×3H, s), 3.79 (2×3H, s), 3.88 (4×3H, s), 5.05 (2H, m), 6.30 (2H, s), 6.55 (2H, s), 6.67 (6H, m). IR (KBr) v cm⁻¹: 1633.5. MS (*m*/2): 713 (M+H)⁺. *Anal.* Calcd for C₄₁H₄₈N₂O₉: C, 69.08; H, 6.78; N, 3.92. Found: C, 69.18; H, 6.71; N, 3.81.

N,*N*'-**Bis-[2-carbamoyl-1-(3,4-dimethoxybenzyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinolinyl]piperazine (22)** To a solution of cationic carbamoyl imidazolium salt of **19** (0.88 g, 2 mmol) and piperazine (0.086 g, 1 mmol) in dichloromethane was added triethylamine (0.27 ml, 2.0 mmol). The mixture was stirred at room temperature for 12 h, then washed with 1.0 M HCl (10 ml), the organic layer dried over anhydrous Na₂SO₄, filtered and concentrated under vacuum to yield the required tetra substituted urea as an oil which was column chromatographed (ethylacetate–hexane) to obtain pure **22** as a light yellow solid (0.61 g, 75%) mp 73—74 °C. ¹H-NMR (CDCl₃, 300 MHz) δ : 2.73 (8H, m), 2.88 (4H, m), 2.92 (4H, m), 3.03 (4H, m), 3.67 (2×3H, s), 3.71 (2×3H, s), 3.77 (4×3H, s), 5.05 (2H, m), 6.23 (1H, s), 6.26 (1H, s), 6.58 (2H, m), 6.72 (6H, m). IR ν (KBr) cm⁻¹: 1637. MS (*m*/*z*): 825 (M+H)⁺. *Anal.* Calcd for C₄₆H₅₆N₄O₁₀: C, 66.97; H, 6.84; N, 6.79. Found: C, 66.58; H, 6.71; N, 6.81.

N,*N*'-**Bis-[2-carbamoyl-1-(3,4-dimethoxybenzyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinolinyl]-2,6-dimethylpiperazine (23)** Synthesized, from cationic carbamoyl imidazolium salt of **19** and 2,6-dimethylpiperazine, in a manner similar to that described for **22**. The product was obtained as a viscous solid (80%). ¹H-NMR (CDCl₃, 300 MHz) δ : 1.80 (6H, s), 2.70 (8H, m), 2.90 (4H, m), 2.93 (4H, m), 3.08 (4H, m), 3.69 (2×3H, s), 3.75 (2×3H, s), 3.87 (4×3H, s), 5.15 (2H, m), 6.27 (1H, s), 6.29 (1H, s), 6.64 (2H, m), 6.74 (6H, m). IR (KBr) v cm⁻¹: 1635. MS (*m/z*): 853 (M+H)⁺. *Anal.* Calcd for C₄₈H₆₀N₄O₁₀: C, 67.58; H, 7.08; N, 6.79. Found: C, 67.51; H, 7.11; N, 6.81.

N,*N*'-**Bis-[2-carbamoyl-1-(3,4-dimethoxybenzyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinolinyl]homopiperazine (24)** Synthesized, from cationic carbamoyl imidazolium salt of **19** and homopiperazine, in a manner similar to that described for **44**. The product was obtained as a white solid (76%) mp 82—83 °C. ¹H-NMR (CDCl₃, 300 MHz) δ: 2.73 (10H, m), 3.07 (4H, m), 3.15 (4H, m), 3.29 (4H, m), 3.64 (2×3H, s), 3.82 (2×3H, s), 3.89 (4×3H, s), 4.95 (2H, m), 6.17 (1H, s), 6.24 (1H, s), 6.56 (2H, s), 6.70 (6H, m). IR (KBr) v cm⁻¹: 1633. MS (*m*/*z*): 839 (M+H)⁺. *Anal.* Calcd for C₄₇H₅₈N₄O₁₀: C, 67.28; H, 6.85; N, 6.67. Found: C, 67.51; H, 6.79; N, 6.81.

Pharmacological Methods Antispasmodic activity: isolated guinea pig ileum Guinea pigs of both sexes (300—500 g) were sacrificed by cervical dislocation and exsanguinated. The abdomen was opened and the terminal ileum carefully dissected, repeatedly washed and the connective tissue removed. Intestinal segments of 1.0 to 1.5 cm length were set up under 1 g resting tension in a 10 ml organ bath with Tyrode's solution: (mm: NaCl 137, KCl 2.7, CaCl₂ 1.8, MgCl₂ 1.05, NaHCO₃ 11.9, NaH₂PO₄ 0.42, glucose 5.6). The bath temperature was maintained at 37 °C and aerated with compressed air. Tension changes in the tissues were monitored using force displacement transducer amplifier connected to physiograph (Polyrite, Recorders and Medicare System).

After 30 min of equilibration period, concentration-response curves to the cumulative addition of acetylcholine (ACh) (0.1 nm-0.1 mm) were constructed. After constant responses had been obtained, concentration-response was repeated in the presence of increasing concentrations of the standard antagonist papaverine (0.1–20 μ M). After a further control experiment (concentration-response curve of ACh) the test compounds 6-13, 15-18 and 21-24 were measured. All substances were incubated 3 min prior to the cumulative addition of ACh; after each experiment all substances were carefully washed out. For each test substance a new ileum preparation was used. Test compounds were dissolved in 46% EtOH to yield solutions (0.1- $20\,\mu\text{M}$). The antispasmodic activity was assessed as a percent reduction over the initial value. The IC_{50} values (concentration required to inhibit the 100%) $[E_{max}]$ Ach response to 50%) were obtained from individual experiments with 3 to 5 different concentrations of test compounds. As EtOH in the concentration used also inhibits ACh-induced contractions the antispasmodic activity of EtOH was measured and subtracted from the values obtained with test compounds.

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