Novel Diterpenes from the Heartwood of *Chamaecyparis obtusa* var. *formosana*

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> Two novel diterpenes, obtusanal B (1) and obtusadione (2), along with obtusanal A (3), obtunone (4), 12-hydroxy-6,7-secoabieta-8,11,13-triene-6,7-dial, 8,12-dihydroxydielmentha-5,9-diene-7,11-dione and myrcene, isolated from the heartwood of *Chamaecyparis obtusa* var. *formosana*, were characterized by spectroscopic means, including 2D-NMR techniques. Compounds 1 and 2 are 7($6\rightarrow$ 2)abeoabietane and 14($8\rightarrow$ 9)abeoabietane type diterpenes, respectively. Their biosyntheses were proposed.

> Key words Chamaecyparis obtusa var. formorsana; Cupressaceae; $7(6\rightarrow 2)$ abeoabietane; $14(8\rightarrow 9)$ abeoabietane; obtusanal; obtusadione

Only two endemic species, Chamaecyparis formosana and C. obtusa var. formosana, among seven species of genus Chamaecyparis are found in the central mountains of Taiwan. They are all huge trees and can live for over a thousand years. The heartwood of C. obtusa var. formosana, with distinguished purple-pink coloring and strong resistance against termites and fungi, caused the heartwood of this plant to have greater economic value than that of C. formosana. In previous papers on the chemical studies of the heartwood of C. obtusa var. formosana, we reported the structural elucidation of novel diterpenes, lignans, monoterpenes and sesquiterpenes.¹⁻⁷⁾ Further detailed reinvestigation of the same extract yielded two novel diterpenes, obtusanal B (1), obtusadione (2), together with obtusanal A (3) (reported in the communication as obtusanal),³⁾ obtunone (4),¹⁾ 12-hydroxy-6,7-secoabieta-8,11,13-triene-6,7-dial (5),8 8,12-dihydroxydielmentha-5,9-diene-7,11-dione (6),9) and myrcene (7).10) The structures of these novel diterpenes were elucidated on the basis of spectral evidence, and compounds 1 and 2 are



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 $7(6\rightarrow 2)$ abeoabietane and $14(8\rightarrow 9)$ abeoabietane diterpenes. Obtusanal B (1) was isolated as colorless needles; its mol-

ecular formula, C₂₀H₂₄O₄, was established through ¹³C-NMR (Table 1) and high-resolution electron impact mass spectroscopy (HR-EI-MS) data and represents nine indices of hydrogen deficiency (IHD). The IR spectrum of 1 showed absorptions for hydroxy (3356 cm^{-1}) , aldehyde (1735 cm^{-1}) cm^{-1}), cyclohexanone (1711 cm^{-1}), conjugated carbonyl (1662 cm^{-1}) , and aromatic groups $(1601, 1576 \text{ cm}^{-1})$. The UV spectrum indicated a benzoyl functionality (λ_{max} 236, 290, and 316 nm). The ¹H-NMR spectrum exhibited signals for three singlet methyl groups [δ 0.59 (H-18), 1.06 (H-19), 1.41 (H-20)], one isopropyl group linked to phenyl [δ 1.24, 1.26 (3H each, d, J=6.9 Hz, H-16, -17), 3.14 (1H, sep, J=6.9 Hz, H-15)], an aldehyde [δ 9.81 (d, J=6.0 Hz, H-6)], two methine protons [δ 2.62 (dd, J=6.0, 2.3 Hz, H-5), 3.77 (dd, J=3.4, 2.3 Hz, H-2)], two methylene protons [δ 2.52 (ddd, J=14.0, 2.3, 2.3 Hz, H_{α}-1), 2.76 (dd, J=14.0, 3.4 Hz, H_{β} -1)], two singlet *para*-phenyl protons [δ 6.81 (H-11), 7.93, (H-14)], and an exchangeable phenolic proton (δ 5.91). The signal of H-14 appeared at a lower field (δ 7.93) due to deshielding by a carbonyl group. H_{α} -1 (2.52) exhibited a larger geminal coupling (J=14.0 Hz), a vicinal coupling (J=2.3 Hz), and a W-form coupling to H_a-5 (δ 2.62, dd, J=6.0, 2.3 Hz). Additionally, H_a-5 expressed W-form coupling. These spectral data suggested the presence of a cyclohexanone moiety. Three carbonyl ¹³C-NMR signals (Table 1) at δ 189.6, 201.7, and 206.3 suggested the presence of conjugated ketone, cyclohexanone, and aldehyde groups, respectively. The correlated spectroscopy (COSY) coincided with the above mentioned structural units. Except for three carbonyl ¹³C-NMR signals (Table 1), the remaining 17 carbon signals included three methyl carbons, three isopropyl carbons, six phenyl carbons with four substitutents including one oxygenated function (δ 159.3), and 5 carbons of cyclohexanone. Analysis of the heteronuclear multiple-bond correlation spectroscopy (HMBC) spectrum and comparison of the spectral data with those of compound 3 indicated that compound 1 possessed the same skeleton as 3, having one carbonyl and one hydroxyl group in the place of two acetoxyl groups existing in compound 3. The relative stereochemistry was described as structure 1 by nuclear Overhauser enhancement and exchange spectroscopy (NOESY) (Fig. 1).

Table 1. ¹³C-NMR Data (CDCl₃, 100 MHz) of Compounds 1–2

No.	1	2
1	36.9 t	33.7 t
2	61.6 d	18.3 t
3	206.3 s	41.7 t
4	44.9 s	33.6 s
5	68.7 d	47.1 d
6	201.7 d	23.4 t
7	189.6 s	39.0 t
8	125.2 s	208.8 s
9	150.3 s	67.8 s
10	35.8 s	44.0 s
11	112.3 d	37.9 t
12	159.3 s	207.3 s
13	135.4 s	153.3 s
14	126.9 d	152.3 d
15	26.9 d	24.9 d
16	22.1 q	21.1 q
17	22.3 q	21.4 q
18	31.3 q	33.8 q
19	25.5 q	22.0 q
20	27.3 q	17.6 q







Fig. 2. HMBC of 2



Fig. 3. NOESY of 2



Chart 1. Proposed Mechanism for Biosynthesis of 1 and 3

Fig. 1. NOESY of 1

Obtusadione (2) had the molecular formula $C_{20}H_{30}O_2$, as determined by HR-EI-MS. No hydroxy absorption was observed in its IR spectrum, but one carbonyl (1712 cm^{-1}) and one olefinic (1610 cm⁻¹) absorption presented. The UV absorptions at λ_{max} 223, 245, 299 nm indicated the presence of a conjugated carbonyl system. Compound 2 had twenty ¹³C-NMR signals (Table 1), including two carbonyl signals (δ 207.3, 208.8) and two olefinic signals [δ 153.3 (C-13), 152.3 (C-14)]. Two carbonyl ¹³C-NMR signals, conjugated UV absorptions, and only one carbonyl absorption band (1712 cm^{-1}) in the IR spectrum indicated that it has one cyclohexanone and one conjugated cyclopentenone. A singlet of vinyl proton with a lowfield at δ 7.26 suggested that it was a β -olefinic proton of conjugated cyclopentenone, unambiguously. An isopropyl group attached at the α -position of conjugated cyclopentenone is attributable to its ¹H-NMR signals $[\delta 1.05, 1.09 \text{ (3H each, d, } J=6.9 \text{ Hz}, \text{ H-16, -17}) \text{ and } 2.60$ (1H, sep, J=6.9 Hz, H-15)]. Based on three singlet of methyl groups [δ 0.87, 0.89, and 1.02 (H-20, H-19, and H-18 respectively)] together with an isopropyl group, we propose that the skeleton of **2** is similar to abietane-type diterpene. Two methylene protons at δ 2.57 (1H, ddd, J=14.6, 7.4, 6.4 Hz, H_{α} -7), and 2.49 (1H, ddd, J=14.6, 5.3, 2.1 Hz, H_{β} -7) were assigned as H₂-7 due to their chemical shifts and having HMBC correlation to C-8 carbonyl carbon ($\delta_{\rm C}$ 208.8). The methylene appeared to be a pair of doublets at δ 3.21 (1H, d, $J=18.7 \text{ Hz}, \text{ H}_{\alpha}-11)$ and 2.13 (1H, d, $J=18.7 \text{ Hz}, \text{ H}_{\beta}-11)$ in ¹H-NMR spectrum, and was considered to be in a position between a quaternary carbon (δ 67.8) and a carbonyl (δ



Chart 2. Proposed Mechanism for Biosynthesis of 2

207.3, C-12) of cyclopentenone due to having HMBC correlation with δ 207.3 and 67.8. This quaternary carbon was assigned to C-9 (spirocarbon) due to showing HMBC with H-20, H-14, H-7. From the HMBC analysis (Fig. 2), the structure of 2 was judged to be $14(8\rightarrow 9)$ abeoabieta-13-ene-8,12dione. Its relative stereochemistry was clear from NOESY (Fig. 3). The NOESY correlations, such as H-5/H-14 and H-5/H-18, confirmed the double bond positioned at an α -axial orientation, accounting for H_{α} -11 and H_{β} -11 having a larger different chemical shift due to H_{α} -11 receiving a deshielding effect from the carbonyl of cyclohexanone. $14(8 \rightarrow 9)$ Abeoabietane is a novel skeleton in naturally occurring products, although in the photoirradiation of podocarpane, derivatives 8 and 9 yielded the two spirans 10 and 11, respectively.¹¹)

The proposed biosynthesis of 1 and 3 from 5 is sketched in Chart 1. Compound 12 is an oxidative product from 5. The enol form 14 from 13 underwent bio-Aldol condensation to yield 15, which was subsequently acetylated and partially oxidized to produce 3 and 1, respectively. Compound 2 was proposed to derive from 16 (Chart 2), which is an unobserved compound occurring naturally but was prepared from ferruginol by oxidization with phenyl seleninic anhydride or benzoyl peroxide.^{12,13} Intermediate **17**, obtained from **16** by acidic treatment, tautomerized to **2**.

Experimental

General Experimental Procedures Extracts were chromatographed on silica gel (Merck 70–230 mesh, 230–400 mesh, ASTM) and purified with a semi-prep. normal phase HPLC column (250×10 mm, 7 μ m, LiChrosorb Si 60) taken on LDC Analytical-III; mp: Yanagimoto micro-melting apparatus; uncorrected. Specific rotation: Jasco DIP-180 digital polarimeter. IR spectra: Perkin-Elmer 983 G spectrophotometer. ¹H- and ¹³C-NMR spectra: Bruker DMX-400 spectrophotometer. EI-MS: JEOL JMS-HX 300 mass spectrometer.

Plant Material The dried heartwood of *C. obtusa* var. *formosana* was collected from Taichung, Taiwan, in 1996. Mr. Muh-Tsuen Kao, formerly a technician with the Department of Botany, National Taiwan University, identified the plant. A voucher specimen has been deposited at the Herbarium of the Department of Botany, National Taiwan University, Taipei, Taiwan.

Extraction and Isolation The dried heartwood of *C. obtusa* var. *formosana* (11 kg) was extracted with Me₂CO (120 l) at r.t. (7d×2). To the evaporated Me₂CO extract, H₂O was added to bring the total volume to 11. This phase was extracted with AcOEt (11×3). The combined AcOEt layers afforded, after evaporation, a black syrup (680 g), which was purified by silica-gel chromatography and HPLC (normal phase on Lichrosorb Si 60), repeatedly, with hexane/AcOEt. Compound 2 (6.3 mg), 4 (8.8 mg), 5 (8.2 mg), 3 (6.7 mg), 1 (8.0 mg) and 6 (10.5 mg) were eluted with 10, 20, 20, 30, 30, and 30% AcOEt in hexane, respectively. The essential oil from steam distillation of the heartwood was subjected to analysis with GC-MS; myrcene (7) was identified by GC-MS computer base data (it was unpublished data).

Obtusanal B (1): mp 192—194 °C. ¹H-NMR: see text. ¹³C-NMR: Table 1. IR (KBr) cm⁻¹: 3356, 1735, 1711, 1662, 1601, 1576, 1319, 1265. UV λ_{max} (MeOH) nm (log ε): 236 (3.78), 290 (3.72), 316 (3.61). HR-EI-MS *m/z*: 328.1665 (M⁺, Calcd for C₂₀H₂₄O₄: 328.1674). EI-MS (70 eV) (rel. int. %) *m/z*: 328 (38, M⁺), 299 (9), 244 (10), 216 (99), 201 (100), 149 (22), 83 (32). $[\alpha]_{D}^{28}$ –195.5° (*c*=0.41, CHCl₃).

Obtusadione (2): mp 178—179 °C. ¹H-NMR (CDCl₃) δ : 0.87, 0.88, 1.02 (3H each, s, H-20, -19, -18), 1.05, 1.09 (3H each, d, J=6.9 Hz, H-16, -17), 1.16, 1.32 (1H each, m, H-1), 1.23, 1.45 (1H each, m, H-3), 1.46—1.53 (2H, m, H-2), 1.75, 2.10 (1H each, m, H-6), 1.85 (1H, dd, J=13.0, 2.7 Hz, H-5),

2.13, 3.21 (1H each, d, J=18.7 Hz, H-11), 2.60 (1H, Sep, J=6.9 Hz, H-15), 2.49 (1H, ddd, J=14.6, 5.3, 2.1 Hz, H_g-7), 2.57 (1H, ddd, J=14.6, 7.4, 6.8 Hz, H_a-7), 7.26 (1H, s, H-14). ¹³C-NMR: Table 1. IR (KBr) cm⁻¹: 1712, 1610, 1390, 1368, 1257, 1195, 1106. UV λ_{max} (MeOH) nm (log ε): 223 (3.89), 245 (3.93), 299 (3.33). HR-EI-MS m/z: 302.2243 (M⁺, Calcd for C₂₀H₃₀O₂: 302.2247). EI-MS (70 eV) (rel. int. %) m/z: 302 (42, M⁺), 241 (13), 166 (31), 137 (18), 88 (32), 73 (36), 70 (76), 61 (100). $[\alpha]_{\rm D}^{19} - 79.1^{\circ}$ (c=0.30, CHCl₃).

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