# Discovery of *N*-(3-{4-[(3-Fluorobenzyl)oxy]phenoxy}propyl)-2-pyridin-4ylacetamide as a Potent and Selective Reverse NCX Inhibitor

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In the setting of heart failure and myocardial ischemia-reperfusion, the sodium-calcium exchanger (NCX) can lead to calcium overload, which is responsible for contractile dysfunction and arrhythmia. NCX is an attractive target for treatment in heart failure and myocardial ischemia-reperfusion. We have designed and synthesized a series of benzyloxyphenyl derivatives based on compound 3. These derivatives have been evaluated for their inhibitory activity against both the reverse and forward modes of NCX. We have discovered a novel potent and selective reverse NCX inhibitor (12) with an IC<sub>50</sub> value of  $0.085 \,\mu$ M against reverse NCX.

Key words sodium-calcium exchanger (NCX); anti-arrhythmics; transporter

Intracellular Ca<sup>2+</sup> is of primary importance in the pathogenesis of ischemia and reperfusion injury in the myocardium. Recent studies have suggested that a massive Ca<sup>2+</sup> influx may occur as a consequence of Na<sup>+</sup>-Ca<sup>2+</sup> exchange via the sodium-calcium exchanger (NCX) during reperfusion which, in turn, may be caused by an accumulation of Na<sup>+</sup> via the sodium-hydrogen exchanger (NHE) during ischemia.<sup>1,2)</sup> This results in an intracellular Ca<sup>2+</sup> overload, the detrimental effects of which include myocardial contracture, stunning, necrosis, and reperfusion arrhythmia.3-9) NCX functions in both reverse and forward modes, and it is well known that an overactive reverse NCX causes Ca<sup>2+</sup> overload. Inhibition of reverse NCX overactivity would effectively block this overload and prevent damage to the myocardium in ischemiareperfusion. Therefore, reverse NCX inhibitors are currently considered to be beneficial in treating disease states.<sup>10,11</sup> Recently, quinazoline derivatives<sup>12)</sup> and a series of benzyl-oxyphenyl derivatives<sup>13–16)</sup> have been identified as NCX inhibitors. KB-R7943 (1) is one of the most widely known NCX inhibitors, and is used as a tool in heart and renal failure models,<sup>17,18)</sup> and SEA0400 (2) is well known as a potent reverse NCX inhibitor that is efficacious in myocardial is-chemia-reperfusion injury<sup>19-22</sup> (Fig. 1). We have recently discovered reverse NCX inhibitors, such as 3, which we have reported elesewhere.<sup>16,23,24)</sup> To create potent and selective reverse NCX inhibitors, we have now designed a novel class of NCX inhibitors based on 3, and in this paper describe the re-



Fig. 1. Several Inhibitors of Sodium–Calcium Exchanger (1) KB-R7943; (2) SEA0400; (3) YM-252077

sults of our work on the synthesis and structure-activity relationships (SAR) of this novel class of benzyloxyphenyl derivatives.

### Chemistry

Compounds 4a-e were converted into compounds 5a-e via O-alkylation with 1-(bromomethyl)-3-fluorobenzene. Desired compounds 6a-e were prepared from 5a-e by condensation with pyridin-4-ylmethylamine as shown in Chart 1. Intermediate 4e was synthesized from 7 via S-alkylation with the corresponding alkylbromide. Compounds 9a-c were obtained by *O*-alkylation of  $\mathbf{8}^{15}$  with the corresponding alkylbromides followed by hydrolysis of the ester group. Compounds 10a-c were afforded by condensation with 9a-c and pyridin-4-ylmethylamine. Compound 8 was converted into a compound whose amino group was protected with a phthaloyl group, followed by deprotection of the phthaloyl group with hydrazine to give compound 11. Compound 12 was afforded via condensation of 11 with pyridin-4ylacetic acid. Compound 13 was prepared by O-alkylation of 8 with *tert*-butyl(2-bromoethyl)carbamate followed by deprotection of the tert-butoxycarbonyl group. Desired compound 14 was obtained from 8 via condensation with bis(trichloromethyl)carbonate and pyridin-4-ylmethylamine.

### **Results and Discussion**

In order to measure the inhibitory effect of the synthesized compounds on the reverse mode of NCX activity, an Na<sup>+</sup>-dependent Ca<sup>2+</sup> influx assay was performed according to reported protocols, using <sup>45</sup>Ca and CCL39 cells stably expressing NCX1.1.<sup>13,23</sup> The inhibitory effect on the forward mode of NCX activity was assayed by a cell necrosis assay, which also used NCX1.1-expressing CCL39 cells.<sup>23,25</sup> The inhibitory potencies of our novel compounds were thus evaluated in both reverse and forward NCX assays. These compounds were then compared to reference compounds KB-R7943 (1), SEA0400 (2) and compound 3.

The structure–activity relationships of our novel series of NCX inhibitors are summarized in Tables 1 and 2. We have reported a SAR of the nicotinamide part.<sup>16</sup> Compound



Reagents and conditions: (a) 1-(bromomethyl)-3-fluorobenzene,  $K_2CO_3$ , DMF; (b) NaOH, MeOH; (c) pyridin-4-ylmethylamine, WSC · HCl, HOBt, THF; (d) Br(CH<sub>2</sub>)<sub>3</sub>CO<sub>2</sub>Et, K<sub>2</sub>CO<sub>3</sub>, MeCN; (e) Br-(CH<sub>2</sub>)<sub>n</sub>-CO<sub>2</sub>R (n = 3—5, R = Me, or Et), K<sub>2</sub>CO<sub>3</sub>, CH<sub>3</sub>CN; (f) 2-(3-bromopropyl)-1*H*-isoindole-1,3(2*H*)-dione, K<sub>2</sub>CO<sub>3</sub>, CH<sub>3</sub>CN; (g) H<sub>2</sub>NNH<sub>2</sub>·H<sub>2</sub>O, MeOH-CHCl<sub>3</sub>; (h) pyridin-4-ylacetic acid, WSC·HCl, HOBt, THF; (i) *tert*-butyl (2-bromoethyl)carbamate, K<sub>2</sub>CO<sub>3</sub>, CH<sub>3</sub>CN, 80 °C; (j) HCl, AcOEt; (k) bis(trichloromethyl)carbonate, Et<sub>3</sub>N, THF, then pyridin-4-ylmethylamine.

Chart 1. Preparation of Compounds 6a-e, 10a-c, 12 and 14

3, with an N-(pyridin-4-ylmethyl)piperidine-1-carboxamide structure, had a potent inhibitory activity against reverse NCX with an IC<sub>50</sub> value of 0.22  $\mu$ M. To create novel structures in the position of the piperidine-1-carboxamide moiety, we attempted to change the piperidine-1-carboxamide linker into an alkyl chain structure. To examine the effect of the length of the alkyl-linker on reverse NCX inhibitory activity, we evaluated compounds with several different lengths of structure between the phenoxy and pyridine parts (Table 1). Compounds 6a and 6b, with a linker length of 5 atoms between the phenoxy part and the pyridine ring produced an extreme reduction in reverse NCX inhibitory activity compared to compound 3. A compound with a 6-atoms linker (6c) had similar inhibitory activity as 3. When we changed the linker length to 7 atoms (10a), the inhibitory activity was increased. This prompted us to introduce longer linkages, with 8 atoms (10b) and 9 atoms (10c) between the phenoxy part and the pyridine ring. However, the inhibitory activities of **10b** and 10c were reduced with increasing linker length. Compound 10a showed the most potent inhibitory activity, with an  $IC_{50}$ value of  $0.10 \,\mu\text{M}$  against reverse NCX (Table 1). The results indicated that linker length is important for inhibitory activity against reverse NCX, and also prompted us to introduce another linker connecting the phenoxy part with the pyridine

ring as shown in Table 2. Replacements of -O-(10a) with -C-(6d) and -S-(6e) slightly reduced inhibitory activity against reverse NCX. Compound 12, with an inverse amide to compound 10a, also showed slightly increased inhibitory activity over 10a with an IC<sub>50</sub> value of  $0.085 \,\mu\text{M}$  against reverse NCX. Compound 12 was 3-fold more potent than SEA0400 (2), and also showed higher selectivity. Replacement of the amide linkage with a urea linkage (14) resulted in a 2.6-fold decrease in inhibitory potency against reverse NCX compared to 10a. Each of these derivatives with an alkyl chain linker showed higher selectivity than 3.

# Conclusion

A series of benzyloxyphenyl derivatives have been prepared and evaluated for their inhibitory activities against the reverse and forward modes of NCX. By modifying the piperidine-1-carboxamide moiety, we found that a compound containing a *N*-propyl-2-pyridin-4-ylacetamide (12), rather than a *N*-(pyridin-4-ylmethyl)piperidine-1-carboxamide (3), had enhanced reverse NCX inhibitory activity and increased selectivity. Compound 12 (YM-270951) had an IC<sub>50</sub> value of 0.085  $\mu$ M against reverse NCX and higher selectivity. The reverse NCX inhibitory activity of 12 was approximately 3times greater than that of of SEA0400 (2). This study could Table 1. Inhibitory Activity of Compounds 6a—c and 10a—c against the Sodium–Calcium Exchanger



Compd.	L	$^{45}$ Ca influx <sup><i>a</i>)</sup> IC <sub>50</sub> ( $\mu$ M) <sup><i>c</i>)</sup>	Cell necrosis <sup>b)</sup> EC <sub>50</sub> ( $\mu$ M) <sup>c)</sup>	Selectivity <sup>d)</sup>
6a	,0, <u>,</u> 0, <u>,</u> ,	17	NT <sup>e)</sup>	_
6b	∽N~	5.6	NT <sup>e)</sup>	_
6c	∽~~u <sup>k</sup> ~	0.25	50	200
10a	,0, H	0.10	43	430
10b	o~~~~~H~~	0.23	>100	>430
10c	_0N^	0.63	63	100
3	<sup>o</sup> Cnyt	0.22	27	120
SEA0400 (2) KB-R7943 (1)	0	0.29 5.1	98 24	340 4.7

Sodium–Calcium Exchanger

Table 2. Inhibitory Activity of Compounds 6d, 6e, 12 and 14 against the

Compd.	L	<sup>45</sup> Ca influx <sup><i>a</i>)</sup> IC <sub>50</sub> ( $\mu$ м) <sup><i>c</i>)</sup>	Cell necrosis <sup>b)</sup> EC <sub>50</sub> $(\mu M)^{c)}$	Selectivity <sup>d</sup>
6d	∽∽∽ <sup>O</sup> H∧∽	0.17	34	200
6e	,s,, H	0.20	48	240
12	,o~~H	0.085	42	490
14	~°~~N <sup>O</sup> H	0.26	40	150
10a	~°~~~ <sup>O</sup> H	0.10	43	430
3	<sup>o</sup> C <sub>N</sub> H	0.22	27	120
	$\cap$			

For footnotes a-d) refer to Table 1.

*a*) Activity against the NCX1.1 expressed in CCL39 cells. <sup>45</sup>Ca influx reflects NCX inhibitory activity in the reverse mode. *b*) Activity against the NCX1.1 expressed in CCL39 cells. Cell necrosis reflects NCX inhibitory activity in the forward mode. *c*)  $Ic_{50}$  values and  $EC_{50}$  values were determined in a single experimental run in triplicate. *d*) Ratio of  $EC_{50}$  value of cell necrosis and  $IC_{50}$  value of <sup>45</sup>Ca influx. *e*) Not tested.

provide a novel approach to more potent reverse NCX inhibitors with higher selectivity.

# Experimental

**Chemistry** Melting points were determined with a Yanaco MP-500D melting point apparatus or a Buchi B-545 melting point apparatus and are uncorrected. <sup>1</sup>H-NMR spectra were recorded on a JEOL JNM-LA300 or a JNM-EX400 spectrometer and the chemical shifts are expressed in  $\delta$  (ppm) values with tetramethylsilane as an internal standard (in NMR description, s=singlet, d=doublet, t=triplet, m=multiplet, and br=broad peak). Mass spectra were recorded on a Hitachi M-80 or a JEOL JMS-LX2000 spectrometer. The elemental analyses were performed with a Yanaco MT-5 microanalyzer (C, H, N) and were within ±0.4% of theoretical values. Drying of organic solutions during workup was done over anhydrous Na<sub>2</sub>SO<sub>4</sub>.

**Ethyl 4-[(4-Hydroxyphenyl)sulfanyl]butanoate (4e)** To a mixture of 4-sulfanylphenol (7) (1.26 g, 10.0 mmol) in MeCN (30 ml) were added  $K_2CO_3$  (2.07 g, 15 mmol) and ethyl 4-bromobutanoate (1.57 ml, 11.0 mmol) at room temperature. The mixture was stirred for 1 h and partitioned between CHCl<sub>3</sub> and H<sub>2</sub>O. The organic layer was dried and concentrated *in vacuo*. The residue was purified by column chromatography on silica gel (hexane : AcOEt=1 : 0-4 : 1) to give **4e** as a colorless oil (2.22 g, 92%): <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  1.25 (3H, t, *J*=7.1 Hz), 1.83–1.94 (2H, m), 2.44 (2H, t, *J*=7.1 Hz), 2.84 (2H, t, *J*=7.1 Hz), 4.12 (2H, q, *J*=7.1 Hz), 5.17 (1H, s), 6.77 (2H, d, *J*=8.6 Hz), 7.30 (2H, d, *J*=8.7 Hz); MS (FAB) *m/z* 240 (M)<sup>+</sup>.

 $\label{eq:4-[(3-Fluorobenzy])oxy]phenoxy}acetic Acid (5a) A mixture of methyl (4-hydroxyphenoxy)acetate (4a) (500 mg, 2.74 mmol), K_2CO_3 (455 mg, 3.29 mmol) in DMF (5 ml) was added 1-(bromomethyl)-3-fluorobenzene (0.35 ml, 2.85 mmol) at room temperature. The mixture was stirred at room temperature for overnight. Water was added to the mixture and appeared precipitate was collected to give methyl {4-[(3-fluorobenzyl)oxy]phenoxy}acetate as a colorless solid (797 mg, 100%). The mixture of methyl {4-[(3-fluorobenzyl)oxy]phenoxy}acetate (747 mg, 2.57 mmol) in MeOH (15 ml) and 1 M NaOH (3.09 ml) was stirred at 60 °C for 3 h. To the mixture was added 1 M HCl. The precipitate was collected and$ 

washed with H<sub>2</sub>O to afford **5a** as a colorless solid (637 mg, 90%): <sup>1</sup>H-NMR (400 MHz, DMSO- $d_6$ )  $\delta$  4.58 (2H, s), 5.06 (2H, s), 6.84 (2H, d, J=9.3 Hz), 6.93 (2H, d, J=9.3 Hz), 7.11—7.18 (1H, m), 7.23—7.29 (2H, m), 7.39—7.46 (1H, m), 12.93 (1H, br s); MS (FAB) m/z 275 (M–H)<sup>-</sup>.

**3-{4-[(3-Fluorobenzyl)oxy]phenyl}propanoic** Acid (5b) Compound **5b** was prepared from **4b** by a procedure similar to that described for **5a**. Compound **5b** was obtained as a colorless solid (89% in 2 steps): <sup>1</sup>H-NMR (400 MHz, DMSO- $d_6$ )  $\delta$  2.45—2.52 (2H, m), 2.75 (2H, t, *J*=7.5 Hz), 5.09 (2H, s), 6.92 (2H, d, *J*=8.8 Hz), 7.11—7.18 (3H, m), 7.24—7.30 (2H, m), 7.40—7.46 (1H, m), 12.07 (1H, s); MS (FAB) *m/z* 274 (M)<sup>+</sup>.

**4-{4-[(3-Fluorobenzyl)oxy]phenyl}butanoic** Acid (5c) Compound 5c was prepared from 4c by a procedure similar to that described for 5a. Compound 5c was obtained as a colorless solid (89% in 2 steps): <sup>1</sup>H-NMR (400 MHz, DMSO- $d_6$ )  $\delta$  1.72—1.80 (2H, m), 2.19 (2H, t, J=7.4 Hz), 2.49—2.55 (2H, m), 5.10 (2H, s), 6.93 (2H, d, J=8.5 Hz), 7.10 (2H, d, J=8.5 Hz), 7.10—7.18 (1H, m), 7.24—7.30 (2H, m), 7.40—7.47 (1H, m), 12.03 (1H, s); MS (FAB) *m/z* 288 (M)<sup>+</sup>.

**5-{4-[(3-Fluorobenzyl)oxy]phenyl}pentanoic Acid (5d)** Compound **5d** was prepared from **4d** by a procedure similar to that described for **5a**. Compound **5d** was obtained as a colorless powder (94% in 2 steps): <sup>1</sup>H-NMR (300 MHz, DMSO- $d_6$ )  $\delta$  1.45—1.57 (4H, m), 2.15—2.25 (2H, m), 2.48—2.54 (2H, m), 5.09 (2H, s), 6.91 (2H, d, *J*=8.5 Hz), 7.10 (2H, d, *J*=8.5 Hz), 7.13—7.19 (1H, m), 7.23—7.30 (2H, m), 7.39—7.47 (1H, m), 11.98 (1H, br s); MS (FAB) *m/z* 301 (M)<sup>-</sup>.

**4-({4-[(3-Fluorobenzyl)oxy]phenyl}sulfanyl)butanoic Acid (5e)** Compound **5e** was prepared from **4e** by a procedure similar to that described for **5a**. Compound **5e** was obtained as a colorless solid (95% in 2 steps): <sup>1</sup>H-NMR (300 MHz, DMSO- $d_6$ )  $\delta$  1.63—1.75 (2H, m), 2.26 (2H, t, *J*=7.2 Hz), 2.85 (2H, t, *J*=7.2 Hz), 5.11 (2H, s), 6.99 (2H, d, *J*=8.8 Hz), 7.12—7.20 (1H, m), 7.24—7.35 (4H, m), 7.39—7.48 (1H, m); MS (FAB) *m/z* 319 (M)<sup>-</sup>.

**2-{4-[(3-Fluorobenzyl)oxy]phenoxy}-***N***-(pyridin-4-ylmethyl)acetamide** Hydrochloride (6a) To a mixture of pyridin-4-ylmethylamine (108 mg, 1.00 mmol), 1-hydroxybenzotriazole (HOBt) (68 mg, 0.50 mmol), 1-ethyl-3-(3-dimethylaminopropyl)-carbodiimide hydrochloride (WSC·HCl) (211 mg, 1.10 mmol) and THF (5 ml) was added 5a (276 mg, 1.00 mmol) at room temperature. The mixture was stirred for 2 h. The reaction mixture was quenched by saturated NaHCO<sub>3</sub>. The mixture was partitioned between CHCl<sub>3</sub> and aqueous NaOH. The organic layer was dried and concentrated *in vacuo* to give light yellow solid. The residue was purified by column chromatography on silica gel (CHCl<sub>3</sub>: MeOH=98:2) to afford a free base of **6a** (256 mg). The compound was dissolved in AcOEt, CH<sub>3</sub>CN and MeOH, the mixture was treated with  $4 \le 1/4$  AcOEt (0.210 ml, 0.838 mmol) and was recrystallized to give **6a** as a colorless solid (182 mg, 45%): mp 154—157 °C; <sup>1</sup>H-NMR (400 MHz, DMSO- $d_6$ )  $\delta$  4.57 (2H, s), 4.58 (2H, s), 5.09 (2H, s), 6.93—7.01 (4H, m), 7.12—7.19 (1H, m), 7.24—7.32 (2H, m), 7.41—7.47 (1H, m), 7.80 (2H, d, J=6.3 Hz), 8.79 (2H, d, J=6.4 Hz), 8.96 (1H, t, J=6.3 Hz); MS (FAB) m/z 367 (M+H)<sup>+</sup>. *Anal.* Calcd for C<sub>21</sub>H<sub>19</sub>N<sub>2</sub>O<sub>3</sub>F · 0.95HCl: C, 62.90; H, 5.01; N, 6.99; F, 4.74; Cl, 8.40. Found: C, 63.12; H, 5.04; N, 7.16; F, 4.56; Cl, 8.44.

**3-{4-[(3-Fluorobenzyl)oxy]phenyl}-***N***-(pyridin-4-ylmethyl)propanamide Hydrochloride (6b)** Compound **6b** was prepared from **5b** by a procedure similar to that described for **6a**. Compound **6b** was obtained as a colorless solid (70%): mp 176—177 °C; <sup>1</sup>H-NMR (400 MHz, DMSO- $d_6$ )  $\delta$  2.48—2.54 (2H, m), 2.80 (2H, t, J=7.5 Hz), 4.49 (2H, d, J=3.8 Hz), 5.11 (2H, s), 6.94 (2H, d, J=8.8 Hz), 7.11—7.19 (3H, m), 7.25—7.32 (2H, m), 7.41—7.47 (1H, m), 7.70 (2H, d, J=6.4 Hz), 8.68—8.74 (1H, m), 8.76 (2H, d, J=6.3 Hz); MS (FAB) *m*/*z* 365 (M+H)<sup>+</sup>. *Anal.* Calcd for C<sub>22</sub>H<sub>21</sub>N<sub>2</sub>O<sub>2</sub>F · HCl: C, 65.91; H, 5.53; N, 6.99; F, 4.74; Cl, 8.84. Found: C, 65.62; H, 5.55; N, 6.95; F, 4.75; Cl, 8.60.

**4-{4-[(3-Fluorobenzyl)oxy]phenyl}-***N***-(pyridin-4-ylmethyl)butanamide** Hydrochloride (6c) Compound 6c was prepared from 5c by a procedure similar to that described for 6a. Compound 6c was obtained as a colorless solid (42%): mp 109—112 °C; <sup>1</sup>H-NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ 1.76—1.88 (2H, m), 2.23 (2H, t, *J*=7.8 Hz), 2.48—2.55 (2H, m), 4.50 (2H, d, *J*=5.9 Hz), 5.10 (2H, s), 6.93 (2H, d, *J*=8.8 Hz), 7.09—7.19 (3H, m), 7.24—7.30 (2H, m), 7.41—7.47 (1H, m), 7.82 (2H, d, *J*=6.3 Hz), 8.68 (1H, t, *J*=5.9 Hz), 8.82 (2H, d, *J*=6.9 Hz); MS (FAB) *m/z* 379 (M+H)<sup>+</sup>. *Anal.* Calcd for C<sub>23</sub>H<sub>23</sub>N<sub>2</sub>O<sub>2</sub>F·HCl: C, 66.58; H, 5.83; N, 6.75; F, 4.58; Cl, 8.54. Found: C, 66.44; H, 5.91; N, 6.73; F, 4.42; Cl, 8.52.

N-(4-{4-[(3-Fluorobenzyl)oxy]phenyl}butyl)-2-pyridin-4-ylacetamide Oxalate (6d) To a mixture of pyridin-4-ylmethylamine (108 mg, 1.00 mmol), 1-hydroxybenzotriazole (HOBt) (68 mg, 0.50 mmol), 1-ethyl-3-(3-dimethylaminopropyl)-carbodiimide hydrochloride (WSC·HCl) (211 mg, 1.10 mmol) and THF (6 ml) was added 5d (302 mg, 1.00 mmol) at room temperature. The mixture was stirred for 3 h. The reaction mixture was quenched by saturated NaHCO3. The mixture was partitioned between CHCl<sub>3</sub> and aqueous NaOH. The organic layer was dried and concentrated in vacuo to give light yellow solid. The residue was purified by column chromatography on silica gel (CHCl<sub>3</sub>: MeOH=98: 2-96:4) to afford a free base of 6d (371 mg). The compound was dissolved in MeOH. To the mixture was added oxalic acid (94 mg, 1.04 mmol). The mixture was concentrated in vacuo, the residue was recrystallized from AcOEt-CH<sub>3</sub>CN (5:1) to give 6d as a colorless powder (284 mg, 59%): mp 128-130 °C; <sup>1</sup>H-NMR (400 MHz, DMSO- $d_6$ )  $\delta$  1.49–1.47 (4H, m), 2.16–2.22 (2H, m), 2.48– 2.54 (2H, m), 4.29 (2H, d, J=5.9 Hz), 5.09 (2H, s), 6.92 (2H, d, J=8.3 Hz), 7.10 (2H, d, J=8.3 Hz), 7.12-7.18 (1H, m), 7.24-7.30 (4H, m), 7.40-7.49 (1H, m), 8.34-8.44 (1H, m), 8.47-8.53 (2H, m); MS (FAB) m/z 393  $(M+H)^+$ . Anal. Calcd for  $C_{24}H_{25}N_2O_2F \cdot (CO_2H)_2$ : C, 64.72; H, 5.64; N, 5.81; F, 3.94. Found: C, 64.65; H, 5.57; N, 5.75; F, 4.06.

**4-({4-[(3-Fluorobenzy])oxy]pheny}sulfany]**-*N*-(pyridin-4-yImethy])butanamide Hydrochloride (6e) Compound 6e was prepared from 5e by a procedure similar to that described for 6a. Compound 6e was obtained as a colorless powder (57%): mp 101—103 °C; <sup>1</sup>H-NMR (400 MHz, DMSO- $d_6$ )  $\delta$  1.72—1.82 (2H, m), 2.36 (2H, t, J=7.3 Hz), 2.87 (2H, t, J=7.3 Hz), 4.50 (2H, d, J=5.4 Hz), 5.12 (2H, s), 7.00 (2H, d, J=8.8 Hz), 7.13—7.19 (1H, m), 7.25—7.35 (4H, m), 7.41—7.48 (1H, m), 7.82 (2H, d, J=6.9 Hz), 8.70—8.77 (1H, m), 8.81 (2H, d, J=6.3 Hz); MS (FAB) *m*/*z* 411 (M+H)<sup>+</sup>. *Anal.* Calcd for C<sub>23</sub>H<sub>23</sub>N<sub>2</sub>O<sub>2</sub>SF·HCl: C, 60.11; H, 5.57; N, 6.10; F, 4.13; Cl, 7.71; S, 6.98. Found: C, 59.91; H, 5.44; N, 6.07; F, 3.92; Cl, 7.65; S, 6.98.

4-{4-[(3-Fluorobenzyl)oxy]phenoxy}butanoic Acid (9a) To a mixture of 4-[(3-fluorobenzyl)oxy]phenol 8 (436 mg, 2.00 mmol), K<sub>2</sub>CO<sub>3</sub> (332 mg, 2.40 mmol) in CH<sub>3</sub>CN (10 ml) was added ethyl 4-bromobutanoate (0.315 ml, 2.20 mmol) at room temperature. The mixture was stirred at 80 °C for 26 h. The mixture was partitioned between CHCl<sub>3</sub> and aqueous NaOH. The organic layer was dried and concentrated in vacuo. The residue was purified by column chromatography on silica gel (hexane: AcOEt=1:0-4:1) to give ethyl 4-{4-[(3-fluorobenzyl)oxy]phenoxy}butanoate as a colorless oil (630 mg, 95%). The mixture of the intermediate (620 mg, 1.87 mmol) in 1 MNaOH (2.8 ml, 2.8 mmol) and EtOH (5 ml) was stirred at room temperature for 4 h. The mixture was concentrated in vacuo. 1 M HCl was added to the residue. The precipitate was collected to afford 9a as a colorless powder (537 mg, 94%): <sup>1</sup>H-NMR (300 MHz, DMSO- $d_6$ )  $\delta$  1.84—1.95 (2H, m), 2.32-2.40 (2H, m), 3.87-3.94 (2H, m), 5.06 (2H, s), 6.85 (2H, d, J=9.1 Hz), 6.93 (2H, d, J=9.1 Hz), 7.10-7.18 (1H, m), 7.22-7.30 (2H, m), 7.38-7.47 (1H, m), 12.11 (1H, s); MS (FAB) m/z 304 (M)<sup>-</sup>.

**5-{4-[(3-Fluorobenzyl)oxy]phenoxy}pentanoic Acid (9b)** Compound **9b** was prepared from **8** by a procedure similar to that described for **9a**. Compound **9b** was obtained as a colorless powder (92% in 2 steps): <sup>1</sup>H-

**6-{4-[(3-Fluorobenzy])oxy]phenoxy}hexanoic** Acid (9c) Compound 9c was prepared from 8 by a procedure similar to that described for 9a. Compound 9c was obtained as a colorless powder (92% in 2 steps): <sup>1</sup>H-NMR (300 MHz, DMSO- $d_6$ )  $\delta$  1.33—1.46 (2H, m), 1.49—1.60 (2H, m), 1.62—1.73 (2H, m), 2.18—2.26 (2H, m), 3.84—3.91 (2H, m), 5.06 (2H, s), 6.84 (2H, d, J=9.2 Hz), 6.93 (2H, d, J=9.2 Hz), 7.10—7.18 (1H, m), 7.22— 7.30 (2H, m), 7.38—7.47 (1H, m), 11.99 (1H, br s); MS (FAB) *m/z* 331 (M)<sup>-</sup>.

**4-{4-[(3-Fluorobenzy])oxy]phenoxy}-***N*-(**pyridin-4-ylmethyl)butanamide Hydrochloride (10a)** Compound **10a** was prepared from **9a** by a procedure similar to that described for **6a**. Compound **10a** was obtained as a colorless powder (77%): mp 144—146 °C; <sup>1</sup>H-NMR (400 MHz, DMSO- $d_c$ )  $\delta$  1.90—2.00 (2H, m), 2.80—2.41 (2H, m), 3.91 (2H, t, *J*=6.3 Hz), 4.52 (2H, d, *J*=5.9 Hz), 5.07 (2H, s), 6.85 (2H, d, *J*=9.3 Hz), 6.94 (2H, d, *J*=9.3 Hz), 7.11—7.18 (1H, m), 7.23—7.30 (2H, m), 7.39—7.47 (1H, m), 7.83 (2H, d, *J*=6.4 Hz), 8.73—8.83 (3H, m); MS (FAB) *m/z* 395 (M+H)<sup>+</sup>. *Anal.* Calcd for C<sub>23</sub>H<sub>23</sub>N<sub>2</sub>O<sub>3</sub>F·HCl: C, 64.11; H, 5.61; N, 6.50; F, 4.41; Cl, 8.23. Found: C, 63.99; H, 5.63; N, 6.51; F, 4.29; Cl, 8.26.

**5-{4-[(3-Fluorobenzy])oxy]phenoxy}-***N*-(**pyridin-4-ylmethyl)pentanamide Hydrochloride (10b)** Compound **10b** was prepared from **9b** by a procedure similar to that described for **6a**. Compound **10b** was obtained as a beige powder (79%): mp 110—112 °C; <sup>1</sup>H-NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  1.66—1.72 (4H, m), 2.26—2.34 (2H, m), 3.88—3.93 (2H, m), 4.51 (2H, d, J=5.9 Hz), 5.06 (2H, s), 6.85 (2H, d, J=8.8 Hz), 6.93 (2H, d, J=8.8 Hz), 7.11—7.18 (1H, m), 7.23—7.29 (2H, m), 7.39—7.46 (1H, m), 7.83 (2H, d, J=6.8 Hz), 8.69—8.74 (1H, m), 8.81 (2H, d, J=6.9 Hz); MS (FAB) *m*/*z* 409 (M+H)<sup>+</sup>. *Anal.* Calcd for C<sub>24</sub>H<sub>25</sub>N<sub>2</sub>O<sub>3</sub>F·HCl: C, 64.79; H, 5.89; N, 6.30; F, 4.27; Cl, 7.97. Found: C, 64.68; H, 5.83; N, 6.21; F, 4.27; Cl, 8.09.

**6-{4-[(3-Fluorobenzy])oxy]phenoxy}-***N*-(**pyridin-4-yImethyl)hexanamide Hydrochloride (10c)** Compound **10c** was prepared from **9c** by a procedure similar to that described for **6a**. Compound **10c** was obtained as a beige powder (68%): mp 125—127 °C; <sup>1</sup>H-NMR (400 MHz, DMSO- $d_6$ )  $\delta$  1.36—1.46 (2H, m), 1.55—1.73 (4H, m), 2.22—2.28 (2H, m), 3.89 (2H, t, *J*=6.3 Hz), 4.51 (2H, d, *J*=5.8 Hz), 5.06 (2H, s), 6.84 (2H, d, *J*=9.2 Hz), 6.93 (2H, d, *J*=9.2 Hz), 7.11—7.18 (1H, m), 7.22—7.30 (2H, m), 7.39—7.46 (1H, m), 7.82 (2H, d, *J*=6.3 Hz), 8.66—8.72 (1H, m), 8.80 (2H, d, *J*=6.3 Hz); MS (FAB) *m/z* 423 (M+H)<sup>+</sup>. *Anal.* Calcd for C<sub>25</sub>H<sub>27</sub>N<sub>2</sub>O<sub>3</sub>F·HCI: C, 65.42; H, 6.15; N, 6.10; F, 4.14; Cl, 7.72. Found: C, 65.27; H, 6.14; N, 6.09; F, 4.21; Cl, 7.69.

(3-{4-[(3-Fluorobenzyl)oxy]phenoxy}propyl)amine (11) To a mixture of 4-[(3-fluorobenzyl)oxy]phenol 8 (655 mg, 3.0 mmol) and K<sub>2</sub>CO<sub>3</sub> (829 mg, 6.0 mmol) in MeCN (10 ml) was added 2-(3-bromopropyl)-1H-isoindole-1,3(2H)-dione (1.05 g, 3.90 mmol) at room temperature. The mixture was stirred at 80 °C. H<sub>2</sub>O was added to the mixture. The precipitate was collected by filtration, washed with H<sub>2</sub>O and dried in vacuo to give 2-(3-{4-[(3-fluorobenzyl)oxy]phenoxy}propyl)-1H-isoindole-1,3(2H)-dione as a light brown powder (1.21 g, 99%). The mixture of 2-(3-{4-[(3-fluorobenzyl)oxy]phenoxy}propyl)-1H-isoindole-1,3(2H)-dione (1.21 g, 2.98 mmol) and hydrazine hydrate (448 mg, 8.94 mmol) in MeOH (10 ml) and CHCl<sub>3</sub> (20 ml) was stirred at room temperature for overnight. The precipitate was filtered and washed with CHCl<sub>3</sub>. The filtrate was washed with aqueous NaOH and brine and dried, concentrated in vacuo. The residue was purified by column chromatography on silica gel (CHCl<sub>3</sub>: MeOH=88:12) to give 11 as a beige solid (539 mg, 66%): <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  1.85–1.96 (2H, m), 2.87-2.93 (2H, m), 3.97-4.04 (2H, m), 5.01 (2H, s), 6.83 (2H, d, J=9.2 Hz), 6.89 (2H, d, J=9.2 Hz), 6.96-7.04 (1H, m), 7.12-7.21 (2H, m), 7.21–7.28 (1H, m); MS (FAB) m/z 276 (M+H)<sup>+</sup>

*N*-(3-{4-[(3-Fluorobenzy])oxy]phenoxy}propy])-2-pyridin-4-ylacetamide Hydrochloride (12) Compound 12 was prepared from 11 by a procedure similar to that described for 6a. Compound 12 was obtained as a slightly pink powder (93%): mp 143—145 °C; <sup>1</sup>H-NMR (400 MHz, DMSO $d_6$ )  $\delta$  3.33—3.40 (2H, m), 3.91 (2H, t, J=5.6 Hz), 4.47 (2H, d, J=5.3 Hz), 5.07 (2H, s), 6.55—6.65 (1H, m), 6.88 (2H, d, J=9.3 Hz), 6.92—6.97 (3H, m), 7.11—7.18 (1H, m), 7.23—7.29 (2H, m), 7.40—7.47 (1H, m), 7.84 (2H, d, J=6.3 Hz), 8.81 (2H, d, J=6.4 Hz); MS (FAB) m/z 395 (M+H)<sup>+</sup>. Anal. Calcd for C<sub>23</sub>H<sub>23</sub>N<sub>2</sub>O<sub>3</sub>F·HCl: C, 64.11; H, 5.61; N, 6.50; F, 4.41; Cl, 8.23. Found: C, 64.01; H, 5.53; N, 6.47; F, 4.30; Cl, 8.09.

(2-{4-[(3-Fluorobenzyl)oxy]phenoxy}ethyl)amine (13) A mixture of *tert*-butyl(2-bromoethyl)carbamate (1.0 g, 4.5 mmol), 4-[(3-fluorobenzyl)-

oxy]phenol **8** (655 mg, 3.00 mmol) and K<sub>2</sub>CO<sub>3</sub> (1.24 g, 9.00 mmol) in MeCN (20 ml) was stirred at 80 °C for over night. The mixture was partitioned between CHCl<sub>3</sub> and aqueous NaOH. The organic layer was dried and concentrated *in vacuo*. The residue was purified by column chromatography on silica gel (CHCl<sub>3</sub>:MeOH=99:1) to give *tert*-butyl (2-{4-[(3-fluorobenzy])oxy]phenoxy}ethyl)carbamate as a light syrup (604 mg). The mixture of *tert*-butyl (2-{4-[(3-fluorobenzyl)oxy]phenoxy}ethyl)carbamate (604 mg, 1.67 mmol) and AcOEt (5 ml) and 4  $\mbox{M}$  AcOEt solution of HCl (2.1 ml, 8.36 mmol) was stirred at room temperature for 6.5 h. The mixture was concentrated *in vacuo*. The residue was partitioned between CHCl<sub>3</sub> and aqueous NaOH. The organic layer was dried and concentrated *in vacuo* to give **13** as a beige syrup (448 mg, 57% in 2 steps): <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  3.06 (2H, d, J=5.1 Hz), 3.94 (2H, d, J=5.1 Hz), 5.01 (2H, s), 6.84 (2H, d, J=9.2 Hz), 6.89 (2H, d, J=9.3 Hz), 6.96—7.04 (1H, m), 7.12—7.21 (2H, m), 7.29—7.38 (1H, m); MS (FAB) *m*/z 262 (M+H)<sup>+</sup>.

N-(2-{4-[(3-Fluorobenzyl)oxy]phenoxy}ethyl)-N'-(pyridin-4-ylmethyl)urea Hydrochloride (14) To a mixture of bis(trichloromethyl)carbonate (190 mg, 0.64 mmol) in THF (5 ml) was added a mixture of 13 (430 mg, 1.60 mmol), Et<sub>3</sub>N (0.335 ml, 2.40 mmol) in THF (5 ml) at 0 °C. The mixture was stirred at room temperature for 1 h. To the mixture was added a mixture of 4-aminomethylpyridine (208 mg, 1.92 mmol),  $Et_3N$  (0.268 ml, 1.92 mmol) in THF (5 ml) at 6 °C. The mixture was stirred at room temperature for 1 h. The reaction mixture was quenched by saturated NaHCO<sub>3</sub>. The mixture was partitioned between CHCl<sub>2</sub> and aqueous NaOH. The organic layer was dried and concentrated in vacuo to give light yellow solid. The residue was purified by column chromatography on silica gel (CHCl<sub>3</sub>: MeOH=92: 8) to afford a free base of 14 (260 mg). The compound was dissolved in AcOEt (7 ml) and MeOH (5 ml), the mixture was treated with 4 M HCl/AcOEt (0.197 ml, 0.789 mmol) and concentrated in vacuo. The residue was recrystallized from AcOEt-CH<sub>3</sub>CN (5:2) to give 14 as a beige powder (231 mg, 33%): mp 115—116 °C; <sup>1</sup>H-NMR (400 MHz, DMSO-d<sub>6</sub>) δ 3.33—3.40 (2H, m), 3.91 (2H, t, J=5.6 Hz), 4.47 (2H, d, J=5.3 Hz), 5.07 (2H, s), 6.55-6.65 (1H, m), 6.88 (2H, d, J=9.3 Hz), 6.92-6.97 (3H, m), 7.11-7.18 (1H, m), 7.23-7.29 (2H, m), 7.40-7.47 (1H, m), 7.84 (2H, d, J=6.3 Hz), 8.81 (2H, d, J=6.4 Hz); MS (FAB) m/z 396 (M+H)<sup>+</sup>. Anal. Calcd for C<sub>22</sub>H<sub>22</sub>N<sub>3</sub>O<sub>3</sub>F·HCl·H<sub>2</sub>O: C, 58.73; H, 5.60; N, 9.34; F, 4.22; Cl, 7.88. Found: C, 58.79; H, 5.36; N, 9.54; F, 4.14; Cl, 7.85.

**Pharmacology** The methods of <sup>45</sup>Ca influx assay and Cell necrosis assay were described in previous report.<sup>23)</sup>

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