Chemoenzymatic Synthesis of Naturally Occurring Benzyl 6-*O***-Glycosyl**b**-D-glucopyranosides**

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Direct β **-glucosidation between benzyl alcohol and D-glucose (5) using the immobilized** β **-glucosidase from almonds with the synthetic prepolymer ENTP-4000 gave a benzyl** β **-p-glucoside (1) in 53% yield. The coupling of** the benzyl β -D-glucopyranoside congener (8) derived from 1 with phenyl 2,3,4-tri-*O*-acetyl-1-thio- β -D-xylopyra**noside (9), ethyl 2,3,4-tri-***O***-acetyl-1-thio-**a**-L-rhamnopyranoside (13), and 2,3,4-tri-***O***-acetyl-**a**-L-arabinopyranosyl bromide (15) afforded 10, 14, and 16, respectively, as coupled products. Deprotection of 10, 14, and 16 provided the synthetic benzyl β-D-xylopyranosyl-(1→6)-β-D-glucopyranoside (2), benzyl α-L-rhamnopyranosyl-(1**→**6)-**b**-D-glucopyranoside (3), and benzyl** a**-L-arabinopyranosyl-(1**→**6)-**b**-D-glucopyranoside (4), respectively.**

Key words β -glucosidase; β -glucosidation; natural product synthesis; benzyl β -D-glucopyranoside

Prunus mume SIEB. *et* ZUCC. (Rosaceae) has been used as medicinal food in Japan for a long time and is reported to have many pharmacological properties, such as the inhibitory effects on bradykinin and prostaglandin $E₂$ production in the abdominal cavities of mice, and the effects on angiotensinconverting enzyme, aldosterone, and corticosterone levels in rat plasma.¹⁾ Moreover, it has been reported that the methanolic extract of *P. mune* exhibited inhibitory effects on rat lens aldose reductose and platelet aggregation.²⁾ Benzyl β -D-glucopyranoside (1) is one of the pharmacologically active constituents of *P. mune*. On the other hand, three kinds of naturally occurring benzyl $6-O$ -glycosyl- β -p-glucopyranoside congeners, benzyl β -D-xylopyranosyl-(1→6)- β -D-glucopyranoside (2) ,^{3,4)} benzyl α -L-rhamnopyranosyl- $(1\rightarrow6)$ - β - D -glucopyranoside $(3)^{5}$ and benzyl α -L-arabinopyranosyl- $(1\rightarrow 6)$ - β -D-glucopyranoside $(4)^{6}$ were isolated from *Alangium chinense*, 4) *Margyricarpus setosus*, 5) and *Lycopersicon esculentum*, 6) respectively. For the purpose of investigation of pharmacological activity of these β -D-glucopyranoside congeners in comparison to **1**, the synthesis of the above-mentioned β -D-glucopyranoside congeners has aroused our interest. In this paper, we describe the synthesis of benzyl β -Dglucopyranoside (**1**) and its naturally occurring benzyl 6-*O*glycosyl- β -D-glucopyranoside congeners 2, 3 and 4, based on the selective β -glycosidation of benzyl alcohol with p-glucose (5) catalyzed by the immobilized β -glucosidase (EC 3.2.1.21) from almonds.

Enzymatic β **-Glycosidation** In the case of the direct β glycosidation of primary alcohols with D-glucose (**5**) using β -glucosidase (EC 3.2.1.21) from almonds under thermodynamic conditions, a high concentration of alcohol or a medium with low water activity is reported to be effective, γ and this method is applied to the synthesis of **1**. Meanwhile, the synthesis of 1 using 4-nitrophenyl β -D-glucopyranoside as a glycosyl donor was reported previously by $us.^{8}$. On the other hand, we reported the effectiveness of immobilization of β -glucosidase (EC 3.2.1.21) from almonds with a photocross-linkable resin prepolymer (ENTP-4000) in the direct β -glucosidation of 1,8-octanediol with p-glucose (5).⁹⁾ Then we examined the direct β -glucosidation of benzyl alcohol

with D-glucose (5) using the reported immobilized β -glucosidase (EC 3.2.1.21)⁹⁾ from almonds. When a large amount of benzyl alcohol (23.4 eq) was used as an acceptor for p-glucose (5) in the presence of the immobilized β -glucosidase, benzyl β -D-glucopyranoside (1) was obtained in 53% yield. Moreover, the same β -glucosidation using the recovered immobilized enzyme afforded **1** in 52% yield.

Synthesis of Benzyl β **-D-Xylopyranosyl-(1→6)-** β **-D-glucopyranoside (2)** Silylation of **1** gave a silyl ether (**6**; 95% yield), which was subjected to acetylation to afford an acetate (**7**) in 99% yield. Deprotection of silyl group of **7** using 1 M IBr solution in CH₂Cl₂ gave a primary alcohol (8) in 76% yield. When 2% I₂ in MeOH solution or *N*-bromosuccinimide (NBS) instead of 1 M IBr solution in CH₂Cl₂ was applied for desilylation of **7**, the yield of **8** was improved to 90% or 98%, respectively. By the following reported procedure,¹⁰⁾ the coupling reaction of benzyl β -D-glucopyranoside congener (8) and phenyl $2,3,4$ -tri-*O*-acetyl-1-thio- β -D-xylopyranoside (**9**) 11) in the presence of *N*-iodosuccinimide (NIS) and trifluoromethanesulfonic acid (TfOH) gave an inseparable mixture $(10:11=3:2)$ of the coupled products 10 and **11** in 40% yield. Finally, treatment of this mixture with K_2CO_3 in MeOH provided the synthetic benzyl β -D-xylopyranosyl- $(1\rightarrow6)$ - β -D-glucopyranoside (2, 55% yield) and benzyl b-D-xylopyranosyl-(1→4)-b-D-glucopyranoside (**12**, 37% yield). The spectral data $(^1H-$ and $^{13}C-_{NMR})$ and specific rotation ($\left[\alpha\right]_D$ -48.3° (MeOH)) of the synthetic **2** were identical with those (¹H- and ¹³C-NMR and $[\alpha]_D$ –48.0° (MeOH)) of the natural product **2**. 4) Where, **11** was formed by glycosylation of benzyl 2,3,6-tri-*O*-acetyl- β -D-glycopyranoside (8a), which was formed from **8** by 4→6 acyl migration in the reaction medium. When **8** was treated with the same coupling reaction medium without **9**, **8a** was partly formed in the reaction mixture. The structure of **12** was determined by the HMBC correlation between H-1" ($\delta_{\rm H}$ 4.33)/C-4' ($\delta_{\rm C}$ 80.8). The low yield of the desired compound (**2**) could be explained by the following reason. In the coupling process between **8** and **9**, partial 4→6 acyl migration in the substrate (**8**) under acidic condition (TfOH) might be occurred to afford the undesired coupled product (**11**). Low yield is pre-

$$
\begin{array}{ccc}\n\mathsf{H}^{\mathsf{O}}\mathsf{H}^{\mathsf{O}}\mathsf{H}^{\mathsf{O}} & \mathsf{H}^{\mathsf{O}}\mathsf{H}^{\mathsf{O}}\mathsf{H}^{\mathsf{O}} \\
\mathsf{H}^{\mathsf{O}}\mathsf{H}^{\mathsf{O}} & \mathsf{H}^{\mathsf{O}}\mathsf{H}^{\mathsf{O}}\mathsf{H}^{\mathsf{O}}\mathsf{H}^{\mathsf{O}} \\
\mathsf{H}^{\mathsf{O}}\mathsf{H}^{\mathsf{O}} & \mathsf{H}^{\mathsf{O}}\mathsf{H}
$$

Benzyl β -D-glucopyranoside (1) $R^1 = H$

Benzyl O-β-D-xylopyranosy-(1→6)-β-D-glucopyranoside (2)

$$
R1 = \frac{HO}{HO}
$$

Benzyl O- α -L-rhamnopyranosy-(1->6)- β -D-glucopyranoside (3)

$$
R^1 = \frac{HO}{Me}
$$

Benzyl O- α -L-arabinopyranosy-(1->6)- β -D-glucopyranoside (4) OH

$$
B_1 = \bigoplus_{H \cup A} B_1
$$

Chart 1

sumably due to low reactivity of **9** as a glycosyl donor. To overcome this low yield process, the usage of 2,3,4-tri-*O* b enzoyl- α -D-xylopyranosyl bromide is presumably effective.

Synthesis of Benzyl α **-L-Rhamnopyranosyl-(1→6)-** β **-D**glucopyranoside (3) Ethyl $2,3,4$ -tri-*O*-acetyl-1-thio- α -Lrhamnopyranoside (**13**) was synthesized by applying the reported method¹²⁾ based on the BF₃ · Et₂O catalyzed reaction of ethane thiol and tetra- O -acetyl- α -L-rhamnopyranoside obtained by acetylation of α -L-rhamnose. By the following reported procedure,¹⁰⁾ the coupling reaction of benzyl β -D-glucopyranoside congener (**8**) and ethyl 2,3,4-tri-*O*-acetyl-1 thio- α -L-rhamnopyranoside (13) in the presence of *N*-iodosuccinimide (NIS) and trifluoromethanesulfonic acid (TfOH) gave the coupled product (**14**) in 76% yield. Finally, treatment of **14** with K_2CO_3 in MeOH provided the synthetic
benzyl β -p-rhamnopyranosyl- $(1 \rightarrow 6)$ - β -p-glucopyranoside β -D-rhamnopyranosyl-(1→6)- β -D-glucopyranoside (3) 85% yield. The spectral data $(^1H-$ and $^{13}C- NMR)$ and specific rotation ($[\alpha]_D$ -52.8° (MeOH)) of the synthetic 3 were identical with those (${}^{1}H$ - and ${}^{13}C$ -NMR and $[\alpha]_{D}$ -40.0° (MeOH)) of the natural product **3**. 5)

Synthesis of Benzyl α **-L-Arabinopyranosyl-(1→6)-** β **-Dglucopyranoside (4)** By following the reported procedure,¹³⁾ the coupling reaction of benzyl β -D-glucopyranoside congener (**8**) and 2,3,4-tri-*O*-acetyl-a-L-arabinopyranosyl bromide (**15**) in the presence of silver triflate (AgOTf) and 4A molecular sieves gave the coupled product (**16**) in 40% yield. Finally, treatment of 16 with K_2CO_3 in MeOH provided quantitatively the synthetic benzyl α -L-arabinopyranosyl- $(1\rightarrow 6)$ - β - D -glucopyranoside (4). The spectral data (¹Hand ¹³C-NMR) and specific rotation ($[\alpha]_D$ -33.6° (MeOH)) of the synthetic 4 were identical with those $(^1H-$ and $^{13}C-$ NMR and $[\alpha]_D$ -39.0° (MeOH)) of the natural product 4.⁶ The low yield of **16** is presumably overcome by the usage of 2,3,4-tri-*O*-benzoyl- α -L-arabinopyranosyl bromide instead of **15**.

Conclusion

Direct β -glucosidation between benzyl alcohol and D-glucose (5) using the immobilized β -glucosidase from almonds with the synthetic prepolymer ENTP-4000 gave a benzyl β -D-glucoside (1) in 53% yield. The coupling of the benzyl β -D-glucopyranoside congener (**8**) derived from **1** and phenyl 2,3,4-tri-*O*-acetyl-1-thio- β -D-xylopyranoside (9), ethyl 2,3,4tri-*O*-acetyl-1-thio- α -L-rhamnopyranoside (13), and 2,3,4-tri-

a; TBDMSCI/4-N, N-dimethylaminopyridine/pyridine b; Ac₂O/4-N,N-dimethylaminopyridine/pyridine c; IBr or I₂ or NBS d; K₂CO₃/MeOH

Chart 2

O-acetyl- α -L-arabinopyranosyl bromide (15) afforded the coupled products **10**, **14**, and **16**, respectively. Deprotection of the coupled products **10**, **14**, and **16** afforded the synthetic benzyl b-D-xylopyranosyl-(1→6)-b-D-glucopyranoside (**2**), benzyl α -L-rhamnopyranosyl-(1→6)- β -D-glucopyranoside (3), and benzyl α -L-arabinopyranosyl- $(1\rightarrow 6)$ - β -D-glucopyranoside (**4**), respectively.

Experimental

¹H- and ¹³C-NMR spectra were recorded on a JEOL JNM-LA 500 spectrometer (Tokyo, Japan). Spectra were recorded with 5—10% (w/v) solution in CDCl₃ with Me₄Si as an internal reference. Melting points were determined on a Yanaco MP-3S micromelting point apparatus and are uncorrected. Optical rotations were measured on a JASCO DIP-370 digital polarimeter. The FAB mass spectra were obtained with a JEOL JMS-AX 500 (matrix; *m*-nitrobenzyl alcohol (NBA)) spectrometer. IR spectra were recorded on a JASCO FT/IR-300 spectrometer. All evaporations were performed under reduced pressure. For column chromatography, silica gel (Kieselgel 60) was employed.

Immobilization of β **-D-Glucosidase Using a Prepolymer** β -D-Glucosidase (EC 3.2.1.21) from almonds was purchased from Sigma Chemical Co. (G-0395, 2.5—3.6 U/mg). Immobilization of β -D-glucosidase from almonds on the photocross-linkable resin prepolymer (ENTP-4000) was carried out using the following procedure. One gram of ENTP-4000 was mixed with 10 mg of a photosensitizer, benzoin ethyl ether, and 110 mg of β -D-glucosidase from almonds (3.4 units/mg). The mixture was layered on a sheet of transparent polyester film (thickness, *ca.* 0.5 mm). The layer was covered with transparent thin film and then illuminated with chemical lamps (wavelength range, 300—400 nm) for 3 min. The gel film thus obtained was cut into small pieces $(0.5 \times 5 \times 5 \text{ mm})$ and used for the bioconversion reaction.

Enzymatic Synthesis of Benzyl β **-D-Glucopyranoside (1) 1) A mix**ture of D-glucose **5** (1.1 g, 6.1 mmol), benzyl alcohol (18.8 g, 173.9 mmol), water (2 ml), and the immobilized β -glucosidase was incubated for 4 d at 50 °C. The reaction mixture was filtered off and the filtrate was directly chromatographed on silica gel (35 g) to give benzyl alcohol (16.9 g, 90% recovery) from the CHCl₃ eluent and β -glucoside (1, 0.875 g, 53% yield) as colorless solid from the CHCl₃/MeOH=10:1 eluent. The NMR (¹H- and ¹³C-NMR) data of β -glucoside **1** were identical with those of the reported β glucoside **1**. 7,8)

2) A mixture of D-glucose **5** (1.1 g, 6.1 mmol), benzyl alcohol (18.8 g, 173.9 mmol), water (2 ml), and the recovered immobilized β -glucosidase was incubated for 4 d at 50 °C. The reaction mixture was filtered off and the filtrate was directly chromatographed on silica gel (35 g) to give benzyl alcohol (17.1 g, 91% recovery) from the CHCl₃ eluent and β -glucoside (1, 0.858 g, 52% yield) as colorless solid from the CHCl₃/MeOH=10 : 1 eluent.

Benzyl 6-*O-tert***-Butyldimethylsilyl-β-D-glucopyranoside (6)** To a solution of **1** (1.00 g, 3.70 mmol) in pyridine (30 ml) was added TBDMSCl (1.10 g, 7.33 mmol) and 4-*N*,*N*-dimethylaminopyridine (DMAP; 10 mg, 0.08 mmol) at 0° C, and the whole mixture was stirred for 12 h at room temperature. The reaction mixture was condensed and the resulting residue diluted with water and extracted with AcOEt. The organic layer was washed with 10% aqueous HCl, 7% aqueous $NAHCO₃$, and brine. The organic layer was dried over $MgSO₄$ and evaporated to give a residue, which was chromatographed on silica gel (50 g, CHCl₃/MeOH (30:1)) to afford 6 (1.35 g, 95%) as a colorless amorphous powder. **6**: $[\alpha]_D^{21}$ -58.8° (*c*=0.873, CHCl₃); IR (KBr): 3423 (br), 3032, 2954, 2927, 2881, 2855 cm⁻¹, ¹H-NMR (CDCl₃): δ 0.00 (6H, s), 0.82 (9H, s), 3.13 (1H, m), 3.25—3.38 (3H, m), 3.66 (1H, dd, *J*=6.0, 11.2 Hz), 3.81 (1H, dd, *J*=2.4, 11.2 Hz), 4.17 (1H, d, *J*=7.2 Hz), 4.25 (1H, br s), 4.44 (1H, br s), 4.47 (1H, d, $J=11.6$ Hz), 4.73 (1H, d, $J=11.6$ Hz), 5.20 (1H, brs), 7.12—7.26 (5H, m); ¹³C-NMR (CDCl₃): δ -4.77, -4.75, 18.7, 26.3 [3C], 63.9, 70.7, 71.2, 73.6, 76.1, 76.7, 101.3, 127.9, 128.4 [2C], 128.5 [2C], 137.3; *Anal.* Calcd for C₁₉H₃₂O₆Si: C, 59.34; H, 8.39%. Found: C, 58.91; H, 8.86%.

Benzyl 2,3,4-Tri*-O***-acetyl-6-***O-tert***-butyldimethylsilyl-**b**-D-glucopyranoside (7)** To a solution of **6** (1.35 g, 3.51 mmol) in pyridine (5 ml) was added Ac₂O (4 ml) and 4-*N*,*N*-dimethylaminopyridine (DMAP; 10 mg, 0.08 mmol) at 0° C, and the whole mixture was stirred for 12 h at room temperature. The reaction mixture was diluted with water and extracted with AcOEt. The organic layer was washed with 0.5 M aqueous HCl, 7% aqueous NaHCO₃, and brine. The organic layer was dried over $MgSO₄$ and evaporated to give a residue, which was chromatographed on silica gel (50 g, *n*hexane/AcOEt $(8:1)$) to afford 7 $(1.79g, 99\%)$ as a colorless amorphous powder. **7**: $[\alpha]_D^{22} - 33.2^{\circ}$ (*c*=1.08, CHCl₃); IR (KBr): 1759 cm⁻¹, ¹H-NMR (CDCl₃): δ 0.00 (6H, s), 0.83 (9H, s), 1.92 (3H, s), 1.93 (3H, s), 1.93 (3H, s), 3.44 (1H, m), 3.64 (2H, m), 4.44 (1H, d, $J=8.0$ Hz), 4.55 (1H, d, *J*=12.4 Hz), 4.80 (1H, d, *J*=12.4 Hz), 4.95 (2H, m), 5.08 (1H, t, *J*=9.2 Hz), 7.18—7.29 (5H, m); ¹³C-NMR (CDCl₃): δ -5.17 [2C], 18.5, 20.8, 20.8, 20.8, 25.9 [3C], 62.5, 69.1, 70.3, 71.5, 73.2, 74.8, 98.9, 127.7 [2C], 127.8, 128.3 [2C], 136.7, 169.2, 169.2, 170.2; *Anal.* Calcd for C₂₅H₃₈O₉Si: C, 58.80; H, 7.50%. Found: C, 58.87; H, 7.56%.

Benzyl 2,3,4-Tri*-O***-acetyl-**b**-D-glucopyranoside (8)** 1) Method 1: To a solution of 7 (2.11 g, 4.13 mmol) in MeOH (35 ml) was added 1 m IBr in CH_2Cl_2 solution (6.20 g) at room temperature, and the whole mixture was stirred for 3 h at the same temperature. The reaction mixture was diluted with saturated sodium thiosulfate solution and extracted with CHCl₃. The organic layer was washed with brine and dried over MgSO₄. Evaporation of the organic solvent gave a residue, which was chromatographed on silica gel (50 g, *n*-hexane/AcOEt (2 : 1)) to afford **8** (1.25 g, 76%) as a colorless amorphous powder. **8**: $[\alpha]_D^{23}$ -44.3° (*c*=0.75, CHCl₃); IR (KBr): 3522, 2962, 1734 cm^{-1} , ¹H-NMR (CDCl₃): δ 1.99 (6H, s), 2.02 (3H, s), 3.51 (1H, m), 3.60 (1H, dd, *J*=11.6, 4.8 Hz), 3.73 (1H, d, *J*=11.6 Hz), 4.59 (1H, d, *J*=8.0 Hz), 4.64 (1H, d, *J*=12.4 Hz), 4.88 (1H, d, *J*=12.4 Hz), 5.04 (2H, m), 5.21 (1H, t, $J=9.6$ Hz), 7.27—7.359 (5H, m); ¹³C-NMR (CDCl₃): δ 20.4 [2C], 20.4, 61.0, 68.5, 70.7, 71.2, 72.6, 73.9, 99.2, 127.3 [2C], 127.6, 128.0 [2C], 136.5, 168.9, 169.5, 169.8; *Anal.* Calcd for C₁₉H₂₄O₉: C, 57.57; H, 6.10%. Found: C, 57.37; H, 6.12%.

2) Method 2: To a solution of **7** (0.79 g, 1.55 mmol) in MeOH (10 ml) was added 2% I₂ in MeOH solution (10 ml) at room temperature, and the whole mixture was stirred for 4 d at the same temperature. The reaction mixture was worked up in the same way as for method 1 to give **8** (0.55 g, 90%) as a colorless amorphous powder.

3) Method 3: To a solution of **7** (0.015 g, 0.03 mmol) in 95% aqueous DMSO (1.0 ml)/THF (0.5 ml)/H2O (28.5 ml) solution was added NBS (0.003 g, 0.033 mmol) at room temperature, and the whole mixture was stirred for 17 h at the same temperature. The reaction mixture was worked up in the same way as for method 1 to give **8** (0.012 g, 98%) as a colorless amorphous powder.

A Mixtur of Benzyl 2,3,4,2,3,4-*O***-Hexaacetyl-**b**-D-xylopyranosyl- (1**→**6)-**b**-D-glucopyranoside (10) and Benzyl 2,3,6,2,3,4-***O***-Hexaacetyl**b**-D-xylopyranosyl-(1**→**4)-**b**-D-glucopyranoside (11)** A mixture of **8** (0.067 g, 0.168 mmol), **9** (0.109 g, 0.451 mmol), and NIS (0.120 g, 0.533 mmol) in CH₂Cl₂ (5 ml) was stirred for 30 min at 0 °C. To the above mentioned mixture was added methane sulfonic acid (0.048 g, 0.50 mmol) at 0 °C and the whole mixture was stirred for 3 h at the same temperature. The reaction mixture was filtered with the aid of celite and the filtrate was condensed to afford a residue, which was chromatographed on silica gel (20 g, *n*-hexane/AcOEt (3:1)) to give a mixture (10:11=3:2, 0.044 g, 40%) as a colorless oil. Major **10**: IR (KBr): 1751 cm⁻¹, ¹H-NMR (CDCl₃): δ 1.59— 2.14 (18H, m), 3.34 (1H, dd, J=8, 12 Hz), 3.64 (1H, m), 3.85 (1H, d, *J*=10 Hz), 4.13 (1H, dd, *J*=4, 12 Hz), 4.51 (1H, d, *J*=8, 12.0 Hz), 4.55 (1H, d, $J=6.4$ Hz), 4.63 (1H, d, $J=12$ Hz), 4.87–5.04 (5H, m), 5.15 (2H, m), 7.25—7.35 (5H, m); ¹³C-NMR (CDCl₃): δ 20.7, 20.7, 20.7, 20.8, 20.8, 20.8, 62.0, 67.7, 68.7, 69.0, 70.4, 70.5, 71.2, 71.3, 72.8, 73.2, 76.7, 99.0, 100.5, 127.4, 127.8, 128.3, 128.3, 136.6, 169.1, 169.1, 169.2, 169.6, 169.8, 170.0; HR-FAB-MS (NBA) m/z : Calcd for C₃₀H₃₈O₁₆Na 677.2031 (M+Na)⁺. Found: 677.2058.

Benzyl β **-D-Xylopyranosyl-(1→6)-** β **-D-glucopyranoside (2) and Benzyl** b**-D-Xylopyranosyl-(1**→**4)-**b**-D-glucopyranoside (12)** A mixture of the above-mentioned mixture $(10, 11, 0.026 \text{ g}, 0.27 \text{ mmol})$ and $K_2CO_3 (0.005 \text{ g})$ in MeOH (2 ml) was stirred for 12 h at room temperature. The reaction mixture was condensed to give a residue, which was chromatographed on silica gel (10 g) to afford **2** (0.009 g, 55%) as a colorless amorphous from CHCl₃/MeOH $(4:1)$ eluent and by-product 12 $(0.006 \text{ g}, 37\%)$ as a colorless amorphous from CHCl₃/MeOH (4:1). **2**: $[\alpha]_D^{19} - 48.3^{\circ}$ (*c*=0.76, MeOH); IR (KBr): 3348 (br), 2925, 2872 cm⁻¹, ¹H-NMR (CD₃OD): δ 3.09–3.41 (8H, m), 3.68 (1H, d, $J=12$ Hz), 3.77 (1H, dd, $J=5$, 12 Hz), 4.02 (1H, d, $J=10$ Hz), 4.26 (2H, d, J=8 Hz), 4.56 (1H, d, J=12 Hz), 4.82 (1H, d, J=12 Hz), 7.16—7.34 (5H, m); ¹³C-NMR (CD₃OD): δ 66.9, 69.8, 71.1, 71.5, 71.9, 74.8, 75.0, 77.0, 77.7, 77.9, 103.2, 105.5, 128.5, 129.1 [2C], 129.1 [2C], 138.8; HR-FAB-MS (NBA) m/z : Calcd for C₁₈H₂₇O₁₀: 403.1604 (M+1)⁺. Found: 403.1579. By-product **12**: $[\alpha]_D^{19} - 38.6^\circ$ (*c*=0.79, MeOH); IR (KBr): 3348 (br), 2925, 2872 cm⁻¹, ¹H-NMR (CDCl₃): δ 3.17–3.56 (8H, m), 3.82—3.93 (3H, m), 4.33 (1H, d, *J*=8 Hz), 4.37 (1H, d, *J*=7.6 Hz), 4.65 (1H, d, $J=12$ Hz), 4.90 (1H, d, $J=12$ Hz), 7.24—7.41 (5H, m); ¹³C-NMR (CD₃OD): δ 61.8, 67.1, 70.9, 71.8, 74.8, 74.9, 76.1, 76.4, 77.8, 80.8, 103.0, 105.3, 128.5, 129.0 [2C], 129.1 [2C], 138.8; HR-FAB-MS (NBA) *m*/*z*: Calcd for $C_{18}H_{27}O_{10}$: 403.1604 (M+1)⁺. Found: 403.1558.

Ethyl 2,3,4-*O***-Triacetyl-1-thio-**a**-L-rhamnopyranoside (13)** To a solution of $1,2,3,4$ -*O*-tetraacetyl- α -L-rhamnopyranose (10.0 g, 30.1 mmol) in CHCl₃ (50 ml) was added ethane thiol (2.35 g, 39.1 mmol) and 47% BF_3 · Et₂O complex (22.4 g, 74 mmol) at 0 °C and the whole mixture was stirred for 6 h at room temperature. The reaction mixture was diluted with water and the organic layer was washed with 7% aqueous NaHCO₃, and brine. The organic layer was dried over $MgSO₄$ and evaporated to give a residue, which was chromatographed on silica gel (50 g, *n*-hexane/AcOEt (5 : 1)) to afford a mixture of α - and β -isomers **13** (8.82 g, 88%). Crystallization of this mixture from ether gave a isomers **13** (4.05 g, 40%) as a colorless needles. **13**: mp 135 °C; $[\alpha]_D^{23} + 69.1$ ° ($c = 1.06$, CHCl₃); IR (KBr): 1749 cm^{-1} , ¹H-NMR (CDCl₃): δ 1.29 (3H, d, J=6 Hz), 1.99 (3H, s), 2.06 (3H, s), 2.16 (3H, s), 2.58—2.69 (2H, m), 4.24 (1H, m), 5.10 (1H, t, *J*10 Hz), 5.20 (1H, d, *J*1.6 Hz), 5.24 (1H, dd, *J*10, 3.2 Hz), 5.34 (1H, dd, $J=1.6$, 3.2 Hz); ¹³C-NMR (CDCl₃): δ 14.9, 17.7, 20.6, 20.6, 20.8, 25.6, 70.4, 70.8, 71.9, 74.9, 81.9, 169.8, 170.2, 170.3; *Anal*. Calcd for C₁₄H₂₂O₇S; C, 50.29; H, 6.63%. Found: C, 50.43; H, 6.64%.

Benzyl 2,3,4,2,3,4-*O***-Hexaacetyl-**a**-L-rhamnopyranosyl-(1**→**6)-**b**-Dglucopyranoside (14)** A mixture of **8** (0.155 g, 0.39 mmol), **13** (0.13 g, 0.39 mmol), molecular sieves 4A (0.5 g) and NIS (0.187 g, 0.83 mmol) in CH₂Cl₂ (5 ml) was stirred for 30 min at 0 °C. To the above mentioned mixture was added methane sulfonic acid $(0.098 \text{ g}, 1.02 \text{ mmol})$ at 0° C and the whole mixture was stirred for 3 h at the same temperature. The reaction mixture was filtered with the aid of celite and the filtrate was condensed to afford a residue, which was chromatographed on silica gel (20 g, *n*hexane/AcOEt $(3:1)$) to give **14** $(0.20 \text{ g}, 76\%)$ as a colorless amorphous powder. **14**: $[\alpha]_D^{21}$ – 71.5° (*c*=1.08, CHCl₃); ¹H-NMR (CDCl₃): δ 1.22 (3H, d, $J=4.4$ Hz), 2.00 (3H, s), 2.00 (3H, s), 2.01 (3H, s), 2.05 (3H, s), 2.05 (3H, s), 2.12 (3H, s), 3.65—3.75 (3H, m), 3.85—3.93 (1H, m), 4.56 (2H, d, *J*=8 Hz), 4.63 (2H, d, *J*=12 Hz), 4.86 (1H, d, *J*=0.8 Hz), 4.87 (2H, d, *J*=12 Hz), 4.96 (1H, t, *J*=9.6 Hz), 5.04 (1H, dd, *J*=9.6, 8.0 Hz), 5.05–5.10 (1H, m), 5.28–5.31 (2H, m), 5.78 (1H, t, *J*=9.6 Hz), 7.26–7.37 (5H, m); ¹³C-NMR (CDCl₃): δ 17.2, 20.4, 20.4, 20.45, 20.5, 20.6, 20.65, 66.4, 66.7, 68.8, 69.2, 69.4, 70.4, 70.8, 71.1, 72.6, 73.3, 90.0, 99.0, 127.5 [2C], 127.7,

128.3 [2C], 136.6, 169.1, 169.3, 169.7, 169.75, 169.8, 170.0; *Anal.* Calcd for $C_{31}H_{40}O_{16}$; C, 55.68; H, 6.03%. Found: C, 55.44; H, 5.92%.

Benzyl a**-L-Rhamnopyranosyl-(1**→**6)-**b**-D-glucopyranoside (3)** A mixture of **13** (0.493 g, 0.73 mmol) and K_2CO_3 (0.005 g) in MeOH (2 ml) was stirred for 12 h at room temperature. The reaction mixture was condensed to give a residue, which was chromatographed on silica gel (10 g, CHCl₃/MeOH (4:1)) to afford **3** (0.26 g, 85%) as a colorless oil **3**: $[\alpha]_D^{19}$ -52.8° (*c*=1.37, MeOH); IR (KBr): 3375 (br), 2926 cm⁻¹, ¹H-NMR (CD₃OD): δ 1.24 (3H, d, *J*=8 Hz), 3.22—3.40 (3H, m), 3.55—3.71 (3H, m), 3.87 (1H, dd, *J*=1, 3 Hz), 3.94 (1H, d, *J*=7.6 Hz), 4.29 (1H, d, *J*=8 Hz), 4.58 (1H, d, $J=12$ Hz), 4.83 (1H, d, $J=12$ Hz), 7.20—7.38 (5H, m); ¹³C-NMR (CD₃OD): δ 18.11, 67.95, 69.60, 71.42, 71.61, 71.95, 72.17, 73.76, 74.78, 76.59, 77.08, 101.92, 102.81, 128.50, 129.03 [2C], 129.04 [2C], 138.41; HR-FAB-MS (NBA) m/z : Calcd for C₁₉H₂₈O₁₀Na; 439.1580 $(M+Na)^+$. Found: 439.1601.

Benzyl $2,3,4,2',3',4'-O$ -Hexaacetyl- α -L-arabinopyranosyl- $(1\rightarrow 6)$ - β -D**glucopyranoside (16)** To a solution of $\mathbf{8}$ (0.6 g, 1.52 mmol) in CH₂Cl₂ (10 ml) was added 2,3,4-tri-*O*-acetyl-a-L-arabinopyranosyl bromide (**15**, 1.03 g, 3.03 mmol) and tetramethylurea (TMU, 0.704 g, 6.6 mmol) and AgOTf (0.778 g, 3.03 mmol) at 0 °C under argon atmosphere. The whole was covered with aluminum foil and stirred for 3.5 h at room temperature. The reaction mixture was cooled at 0° C and quenched with AcOEt (100 ml) and 7% aqueous NaHCO₃ solution (20 ml). The organic layer was washed with brine and dried over $Na₃SO₄$. Evaporation of the organic solvent gave a residue, which was chromatographed on silicagel (20 g, *n*-hexane/AcOEt $(4:1)$) to afford **16** (0.4 g, 40% yield) as a colorless amorphous. **16**: $[\alpha]_D^{26}$ -32.1° (*c*=1.01, CHCl₃); IR (KBr): 1739 cm⁻¹, ¹H-NMR (CDCl₃): δ 1.99 (3H, s), 2.01 (3H, s), 2.02 (3H, s), 2.03—2.04 (6H, m), 2.14 (3H, s), 3.57— 3.62 (2H, m), 3.65–3.71 (1H, m), 3.90 (1H, dd, J=1.5, 10.6 Hz), 4.31 (1H, dd, $J=3.5$, 12.1 Hz), 4.48 (1H, d, $J=7.0$ Hz), 4.52 (1H, d, $J=8.0$ Hz), 4.61 (1H, d, $J=12$ Hz), 4.89 (1H, d, $J=12$ Hz), 4.95 (1H, dd, $J=9.6$, 9.6 Hz), 5.03 (1H, dd, $J=9.6$, 9.6 Hz), 5.02–5.05 (1H, m), 5.15 (1H, d, $J=9.6$ Hz), 5.17—5.26 (2H, m), 7.28—7.38 (5H, m); ¹³C-NMR (CDCl₃): δ 20.69, 20.73 [2C], 20.75, 20.89, 21.01, 63.18, 67.61, 67.99, 69.13, 69.22, 70.12, 70.54, 71.43, 72.97, 73.28, 99.17, 100.91, 127.71 [2C], 128.05, 128.56 [2C], 136.85, 169.45, 169.53, 169.68, 170.21, 170.32, 170.36; HR-FAB-MS (NBA) m/z : Calcd for C₃₀H₃₉O₁₆ 655.2238 (M+1)⁺. Found: 655.2241.

Benzyl b**-D-Arabinopyranosyl-(1**→**6)-**b**-D-glucopyranoside (4)** A mixture of 16 (0.15 g, 0.23 mmol) and K_2CO_3 (0.032 g) in a mixed solvent $(MeOH : CH₂Cl₂=5:1; 12 ml)$ was stirred for 15 min at room temperature. The reaction mixture was condensed to give a residue, which was chromatographed on silica gel (10 g, $CH_2Cl_2/EtOH$ (5 : 1)) to afford 4 (0.062 g, 69%). **4**: $[\alpha]_D^{24}$ -33.6° (c =0.025, H₂O); IR (KBr): 3600—3100 cm⁻¹, ¹H-NMR (CD₃OD+D₂O): δ 3.17–3.22 (1H, m), 3.30–3.39 (2H, m), 3.40– 3.52 (4H, m), 3.72 (1H, dd, *J*=5.0, 11.6 Hz), 3.76—3.84 (2H, m), 4.04 (1H, dd, $J=2.0$, 11.6 Hz), 4.26 (1H, d, $J=7.0$ Hz), 4.35 (1H, d, $J=7.6$ Hz), 4.60 (1H, d, J=12 Hz), 4.83 (1H, d, J=12 Hz), 7.21–7.37 (5H, m); ¹³C-NMR (CD_3OD+D_2O) : δ 66.31, 66.99, 69.38, 70.83, 72.01, 72.23, 73.70, 74.52, 76.36, 77.24, 102.88, 104.88, 129.02, 129.32 [2C], 129.46 [2C], 138.26; HR-FAB-MS (NBA) m/z : Calcd for C₁₈H₂₇O₁₀ 403.1604 (M+1)⁺. Found: 403.1594.

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