Polystachyne F, a 5,10-seco-Neoclerodane from Salvia polystachya¹⁾

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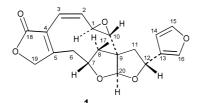
A new 5,10-*seco*-neoclerodane diterpene, polystachyne F (1), was isolated from the aerial parts of *Salvia polystachya*. Its structure was established on the basis of its spectroscopic properties and X-ray crystallographic analysis. Some correlations in the NOESY spectrum of 1 suggested the existence of a conformational equilibrium.

Key words Salvia polystachya; Lamiaceae; 5,10-seco-neoclerodane; polystachyne F

Salvia polystachya ORT. (Lamiaceae) is a small shrub native from México where its seeds, known as "chía", are used as food and beverage ingredients.^{2,3)} In Mexican traditional medicine, this plant is mainly used as purgative and antigastralgic, although other uses such as antipyretic, antipaludic, antihemorragic and emollient, have also been described.²⁾ In an earlier work, we isolated eight clerodane type diterpenes (salvifaricin, linearolactone, dehydrokerlin and polystachynes A—E) from this plant.⁴⁾ Now, we described the isolation of a further clerodane derivative from this species.

Results and Discussion

Polystachyne F (1) was obtained as colorless plates. Its molecular formula, $C_{20}H_{20}O_6$, deduced from the quasimolecular ion peak at m/z 357 (ESI-MS-positive mode), together with the data of the ¹³C-NMR DEPT spectrum, revealed its diterpenic nature. The IR spectrum exhibited absorption bands at 1752 (α,β -unsaturated- γ -lactone), 1645 (double bonds), 1502 and 874 (β -substituted furan ring) cm⁻¹. The ¹³C-NMR spectrum confirmed the presence of the γ -lactone by the signals of a carboxylic and a oxygenated methylene carbons at $\delta_{\rm C}$ 172.3 (C-18) and 72.4 (C-19). The latter correlated in the HSQC spectrum with the proton signals of an AB system (J=17.5 Hz) at $\delta_{\rm H}$ 5.07 (H-19a) and 4.70 (H-19b). ²J, ³J H–C-couplings of these protons with two non protonated vinylic carbons at $\delta_{\rm C}$ 126.3 (C-4) and 161.0 (C-5) led to propose a C-4 double bond, whose presence was supported by the allylic couplings between the CH₂-19 protons and an olefinic proton at $\delta_{\rm H}$ 6.29 (1H, d, J=11 Hz, H-3), observed in the ¹H-¹H COSY spectrum. In this spectrum, the observed couplings between the H-3 signal and the signal at $\delta_{\rm H}$ 5.90 (1H, brt, J=8.8 Hz, H-2), and of the latter with that at $\delta_{\rm H}$ 3.42 (1H, br d, J=5.5 Hz, H-1), which in turn was coupled with the singlet signal at $\delta_{\rm H}$ 3.01 (1H, H-10), allowed us to established a C-2 double bond and a 1(10) epoxy group. The presence of these functionalities was corroborated by the ¹³C-NMR signals at $\delta_{\rm C}$ 132.9 (C-2), 122.5 (C-3), 57.9 (C-1) and 58.3 (C-10).



Clerodane diterpenes usually posses one secondary and one tertiary methyl groups at C-8 and C-9 respectively. In the ¹H-NMR spectrum of **1** only one methyl signal at $\delta_{\rm H}$ 1.46 (3H, d, J=7.5 Hz) was observed. This signal was assigned to the secondary methyl group (CH₃-17) at C-8, since the C-8 proton $\delta_{\rm H}$ 2.11 (1H, q, J=7.5 Hz, H-8) correlated, in the HMBC spectrum with C-17, C-10, and with the signals of a quaternary carbon at $\delta_{\rm C}$ 56.8 (C-9), a CH₂ carbon at $\delta_{\rm C}$ 36.1 (C-6), and an acetalic carbon at $\delta_{\rm C}$ 108.2. The last carbon signal was attributed to C-20 by its ${}^{2}J$, ${}^{3}J$ H–C-couplings with the signals of H-8, H-11a and with two CH protons whose chemical shifts at δ 5.16 and 4.66, indicated be geminal to oxygenated functions. The former was assigned to H-12 by its ¹H–¹H couplings with both C-11 protons, while the latter was attributed to H-7 due to its couplings with the C-6 protons. The above mentioned and the observed HMBC correlations of C-20 with H-7 and H-12 implies the presence of ethereal C-7-C-20 and C-12-C-20 linkages.

The ¹H- and ¹³C-NMR spectra of **1** showed signals for a β substituted furan at $\delta_{\rm C}$ 128.2 (C-13), $\delta_{\rm C}$ 108.5, $\delta_{\rm H}$ 6.34 (CH-14), $\delta_{\rm C}$ 143.7, $\delta_{\rm H}$ 7.37 (CH-15) and $\delta_{\rm C}$ 138.9, $\delta_{\rm H}$ 7.35 (CH-16). The C-13, C-14 and C-15 carbon signals correlated in the HMBC spectrum with the H-12 proton signal, thus establishing the position of the furan ring and the structure **1** for polystachyne F.

Analysis of both, Dreiding models and NOESY spectrum of **1** led us to establish its relative configuration as follows: NOE interactions of the equatorial CH₃-17 with H-7, H-11a and H-12 indicated a 7*R*, 8*S*, 9*R* configurations, if a 12*R* configuration, as those assigned to the neoclerodanes previously isolated from *S. polystachya*,⁴⁾ is assumed. On the other hand, the β axial H-8 showed strong interactions with H-6, H-7 and H-10, thus establishing a β -orientation of H-10 and, as consequence, a 10*R* configuration. The NOE between H-2 and H-3 indicated a *Z* configuration of the C-2 double bond and the NOEs between H-2 and H-10 and between H-1 and H-20 established a *trans* epoxy group, and therefore *R* configurations at C-1 and C-20. These assignments were confirmed by the X-ray crystallographic analysis of **1** (Fig. 1).

Conformation A (Fig. 2) agrees with the above mentioned NOEs exhibited by compound **1**, nevertheless, the observed NOEs of H-1 with H-2 and H-6 and that of H-8 with H-19b (weak), can only be explained if compound **1** adopts conformation B (Fig. 2). The broadening of the signals for C-1, C-2 and C-6 protons, supports the presence of both conformers, but they should be in a rapid equilibrium since the ¹H- and

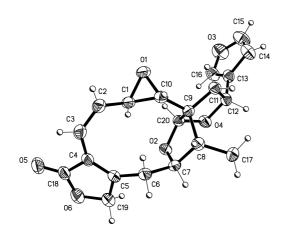
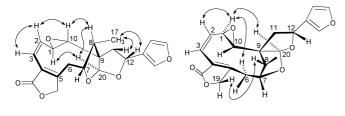


Fig. 1. ORTEP Diagram of Polystachyne F (1)



 Conformer A
 Conformer B

 Fig. 2.
 NOE Interactions for Conformers A and B of Polystachyne F

¹³C-NMR signals did not appear in duplicate. A similar conformational behavior was observed in other 5,10-*seco*-neoclerodanes like tonalensin, a quite related compound isolated from *Salvia tonalensis*,⁵⁾ and 7α -hydroxystrictic acid methyl ester isolated from *Coniza hypoleuca*.⁶⁾

Experimental

General Experimental Procedures The melting point (uncorrected) was determined in a Fisher–Johns apparatus. Column chromatographies were performed on silica gel 60 (Merck G), while silica gel 230–400 mesh (Macherey–Nagel) was used for flash chromatography. TLC was carried out on precoated Macherey–Nagel Sil G/UV₂₅₄ plates of 0.25 thickness, and spots were visualized by spraying with 3% CeSO₄ in H₂SO₄ 2 N followed by heating. The optical rotation was measured on a Perkin Elmer 343 polarimeter. The UV and IR spectra were recorded on a Shimadzu UV 160U and a Bruker Tensor 27 spectrometers, respectively. ¹H- (500 MHz) and ¹³C-(125 MHz) NMR spectra were recorded on a Varian Unity Plus 500 spectrometer with TMS as internal standard. Mass spectrum was recorded on a Bruker Daltonics Esquire 6000 spectrometer.

Plant Material Aerial parts of *Salvia polystachya* ORT. were collected, as in the previous study, in Huitzilac, State of Morelos, México, in June 2005. A voucher specimen (MEXU-573762) is deposited at the Herbarium of the Instituto de Biología, UNAM.

Extraction and Isolation The air-dried powered leaves of *S. poly-stachya* (1.2 kg) were packed in a glass column and extracted with Me₂CO (101). The extracts were combined and concentrated *in vacuo* to dryness. The dried residue (86.9 g) was chromatographed on a column of silica gel eluted with CH₂Cl₂ (fr. A, 21, 32.4 g; fr. B, 41, 14.67 g); CH₂Cl₂-EtOAc 19:1 (fr. C, 1.51, 14.2 g); CH₂Cl₂-EtOAc 7:3 (fr. D, 1.51, 21.37 g); EtOAc (fr. F, 21, 3.04 g). Fraction B was chromatographed over silica gel using mixtures of hexane–EtOAc–Me₂CO (80:10:4, frs. 1—36; 70:15:9, frs. 37—82; 200 ml each). Fractions 41—66 were combined and further purified by column chromatography eluted with hexane–EtOAc 7:3. 37 fractions of 50 ml each were obtained, from which, fractions 20—27 (210.9 mg) were combined and submitted to flash chromatography on silica gel 230—400-mesh eluted with mixtures of hexane–EtOAc (4:1, frs. 1—30; 4:2, frs.

Table 1. 1 H- and 13 C-NMR Data of Polystachyne F (1) (500, 125 MHz in CDCl₃, TMS)

Position	$\delta_{_{ m H}}~(J{ m in}{ m Hz})$	$\delta_{ m c}$	DEPT	HMBC (H–C) couplings
1	3.42 br d (5.5)	57.9	СН	C-2, C-10
2	5.90 br t (8.8)	132.9	CH	C-4
3	6.29 d (11)	122.5	CH	C-1, C-5
4		126.3	С	
5		161.0	С	
6,6′	2.80 br s (w/2=17)	36.1	CH_2	C-5, C-7, C-8, C-19
7	4.66 t (9)	85.0	CH	C-9, C-17, C-20
8	2.11 q (7.5)	47.9	CH	C-6, C-9, C-10, C-17, C-20
9		56.8	С	
10	3.01 s	58.3	CH	C-9, C-20
11	2.58 br dd (13.5, 8.5)	34.3	CH_2	C-8, C-9, C-13, C-20
	2.28 dd (13.5, 7.5)			C-8, C-9, C-10, C-12, C-13
12	5.16 t (7.5)	73.4	CH	C-11, C-13, C-14, C-16, C-20
13		128.2	С	
14	6.34 s	108.5	CH	C-13, C-15, C-16
15	7.37 t (2.5)	143.7	CH	C-13, C-14, C-16
16	7.35 br s	138.9	CH	C-14, C-15
17	1.46 d (7.5)	17.1	CH_3	C-7, C-8, C-9
18		172.3	С	
19	5.07 d (17.5)	72.4	CH_2	C-4, C-5
	4.70 d (17.5)			C-4, C-5
20	5.05 s	108.2	СН	C-7, C-8, C-9, C-10

31—85; 25 ml each). Compound 1 (32.1 mg) was purified from fractions 50—64 by crystallization from Me_2CO -hexane.

Polystachyne F (1) was obtained as colorless plates (Me₂CO–hexane); mp 135–138 °C. ¹H- and ¹³C-NMR spectroscopic data and HMBC correlations, see Table 1. IR (film) cm⁻¹: 1752, 1645, 1502, 874. UV λ_{max} (MeOH) nm (ε): 210 (14820). ESI-MS (positive mode) *m/z*: 379 [M+Na]⁺, 357 [M+H]⁺. [α]_D²² – 166.8° (*c*=0.180, CHCl₃).

X-Ray Crystallographic Analysis of Polystachyne F (1) Crystals data: $C_{20}H_{20}O_6 \cdot 0.5C_3H_6O$, formula weight 385.40, crystal size $0.446 \times 0.322 \times 0.058$ mm, monoclinic, space group C2, Z=4 with a=23.772(2) Å, b=8.406(1) Å, c=9.898(1) Å, $\beta=104.034(2)^\circ$, V=1918.9(3) Å³, $D_{calc}=1.334$ mg/m³, F(000)=816 and GOF=1.039. Unit cell and intensity data were collected on a Bruker Smart Apex CCD diffractometer with MoK α radiation ($\lambda=0.71073$ Å), $\mu=0.099$ mm⁻¹. The structure was solved by direct methods and refined by full-matrix least square on F^2 method.⁷¹ The final data for *R* and *wR* factors with $I>2\sigma(I)$ were 0.0367 and 0.0884, respectively. Crystallographic data for the structure reported in this paper have been deposited with the Cambridge Crystallographic Data Centre (CCDC 612712).

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References and Notes

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