## **Three New Monoterpenoids from the Fruit of** *Genipa americana*

Masateru Ono,<sup>\*,*a*</sup> Naoki Ishimatsu,<sup>*a*</sup> Chikako Masuoka,<sup>*a*</sup> Hitoshi Yoshimitsu,<sup>b</sup> Ryota Tsuchihashi,<sup>c</sup> Masafumi OKAWA, *<sup>c</sup>* Junei KINJO, *<sup>c</sup>* Tsuyoshi IKEDA, *<sup>d</sup>* and Toshihiro NOHARA*<sup>d</sup>*

*aKyushu Tokai University School of Agriculture; 5435 Minamiaso, Aso, Kumamoto 869–1404, Japan: b Faculty of Pharmaceutical Sciences, Sojo University; 4–22–2 Ikeda, Kumamoto 860–0082, Japan: <sup>c</sup> Faculty of Pharmaceutical Sciences, Fukuoka University; 8–19–1 Nanakuma, Jonan-ku, Fukuoka 814–0180, Japan: and <sup>d</sup> Faculty of Medical and Pharmaceutical Sciences, Kumamoto University; 5–1 Oe-honmachi, Kumamoto 862–0973, Japan.* Received August 2, 2006; accepted January 17, 2007

**Three new monoterpenoids, called genipacetal, genipamide, and genipaol, were isolated from the fruit of** *Genipa americana* **L. (Rubiaceae), along with the four known iridoids genipin, gardendiol, deacetyl asperulosidic acid methyl ester, and shanzhiside. Their chemical structures were determined on the basis of spectroscopic data.**

**Key words** *Genipa americana*; monoterpenoid; iridoid; alkaloid; Rubiaceae

*Genipa americana* L. (Rubiaceae) is widely distributed in tropical Central and South America. The fruit of this plant is edible and popular as source of beverages. The white flesh turns yellow to bluish-purple and finally to jet-black on exposure to the  $air<sup>1</sup>$ .

We earlier reported the isolation and structural elucidation of eight iridoid glucosides from the fruit of *G. americana*. 2) As part of the continuing study of the constituents of this fruit, we now describe the isolation and structural elucidation of three new monoterpenoids (**1**—**3**), along with four known iridoids (**4**—**7**) from this fruit.

Compounds  $4-7$  were identified as genipin  $(4)$ ,<sup>3)</sup> gardendiol  $(5)$ ,<sup>4)</sup> deacetyl asperulosidic acid methyl ester  $(6)$ ,<sup>5)</sup> and shanzhiside (**7**) 6) based on their physical and spectral data, although detailed <sup>1</sup> H- and 13C-NMR spectral data of **6** and **7** have not been reported in the literature.

Compound **1**, called genipacetal, was obtained as a colorless syrup and exhibited an  $[M+Na]^+$  ion peak at  $m/z$  249 in the positive FAB-MS; the high-resolution (HR) FAB-MS indicated the molecular formula of 1 to be  $C_{11}H_{14}O_5$ . The <sup>1</sup>H-NMR spectrum of **1** revealed the presence of three oxygenated methine protons  $[\delta 5.74 \, (d, J=5.0 \, Hz), 5.30 \, (s), 4.64]$ (d,  $J=5.0$  Hz)], one methoxyl group ( $\delta$  3.68), five methine protons [d 3.20 (ddd, *J*-5.0, 5.0, 9.5 Hz), 2.83 (dd, *J*-5.0, 5.0 Hz), 2.69 (d, *J*-1.5 Hz), 2.65 (d, *J*-4.5 Hz), 2.54 (dd,  $J=4.5$ , 9.5 Hz)], and one methylene group [ $\delta$  1.74 (d, *J*=11.0 Hz), 1.67 (dd, *J*=1.5, 11.0 Hz)]. The <sup>1</sup>H<sup>-1</sup>H correlation spectroscopy (COSY) spectrum of **1** indicated the presence of the partial structure as shown in Fig. 1. The  $^{13}$ C-NMR spectrum of **1** gave signals due to one carboxyl carbon ( $\delta$  174.9), two acetal carbons ( $\delta$  110.8, 100.6), one oxygenated methine carbon ( $\delta$  84.4), one methoxyl carbon ( $\delta$ 52.5), five methine carbons ( $\delta$  54.7, 50.3, 50.2, 49.7, 42.1), and one methylene carbon ( $\delta$  38.3). The <sup>1</sup>H- and <sup>13</sup>C-NMR signals were assigned with the aid of heteronuclear multiplequantum coherence (HMQC) and heteronuclear multiplebond correlation (HMBC) spectra (Tables 1, 2). From the cross-peaks observed in the HMBC spectrum (H-1/C-5, C-8, C-9, and C-10; H-3/C-10 and C-11; H-4/C-3, C-5, C-6, C-9, and C-11; H-5/C-3; H-6/C-3, C-7, C-8, and C-9; H-7/C-5; H- $10/C-1$  and C-3; and H<sub>3</sub>CO/C-11) the planar structure of 1, which is a monoterpenoid possessing a new skeleton with one methoxycarbonyl group, one five-membered cyclic acetal group, and one five-membered cyclic hemiacetal group, could be suggested (Fig. 1). Ordinary acetylation of **1** with acetic anhydride and pyridine afforded the corresponding monoacetate (**1a**) rather than triacetate. Comparing the chemical shifts of the signals between **1** and **1a**, the signal due to H-1 in **1** was shifted downfield by 0.85 ppm. Thus the presence of a hemiacetal group at C-1 and an epoxy bridge between C-3 and C-10 was established. The relative configurations at the C-5, C-7, C-8, C-9, and C-10 chiral centers were automatically established by forming the epoxy bridge. In the nuclear Overhauser and exchange spectroscopy (NOESY) spectrum of **1**, a key NOE was observed between H-1 and H-4; this indicated that the relative configurations at C-1 and C-4 were as shown in Fig. 2. Furthermore, the coupling constant values for the signals in the <sup>1</sup>H-NMR spectrum were analogous to those calculated for the dihedral angles, which was simulated using CAChe CONFLEX, $6$ ) and the Karplus equation<sup>7)</sup> (Table 3). Consequently, the structure of **1** was concluded to be as shown in Fig. 3.



Fig. 1. Partial Structure Elucidated by the <sup>1</sup>H-<sup>1</sup>H COSY Spectrum (Bold Lines) and <sup>1</sup> H–13C Long-Range Correlations (Arrow) Observed for **1** and **3** in the HMBC Spectra (in CD<sub>3</sub>OD, 500 MHz)





 $\delta$  in ppm from tetramethylsilane (TMS) (coupling constants (*J*) in Hz are given in parentheses).

Table 2. <sup>13</sup>C-NMR Data for  $1 - 3$ , 6 and 7 (in CD<sub>3</sub>OD, 125 MHz)

	1	$\mathbf{2}$	3	6	7
1	100.6	171.4	97.3	101.6	94.9
3	84.4	139.2	153.9	155.4	152.3
$\overline{4}$	50.3	112.0	110.4	108.3	112.1
5	42.1	38.9	46.5	42.7	41.6
6	38.3	40.9	38.3	75.4	77.8
7	50.2	128.1	128.6	129.9	49.1
8	49.7	144.0	147.6	151.5	79.0
9	54.7	50.2	81.0	45.9	51.8
10	110.8	61.7	59.8	61.7	24.7
11	174.9	168.7	169.8	169.5	171.6
$11\text{-OCH}_3$	52.5	52.0	51.7	51.8	
$4'$ -OCH <sub>3</sub>		52.1			
1'		47.9		100.5	99.8
2'		24.9		75.0	74.7
3'		31.7		77.9	78.0
4'		175.0		71.7	71.7
5'				78.5	78.3
$6^{\prime}$				62.9	62.9



Compound **2**, called genipamide, was obtained as a colorless syrup, and its positive FAB-MS showed an  $[M+H]$ <sup>+</sup> ion peak at *m*/*z* 324. The molecular formula of **2** was determined to be  $C_{16}H_{21}NO_6$  using HR positive FAB-MS. The <sup>1</sup>H-NMR spectrum of 2 indicated signals due to two olefinic protons  $\delta$ 7.34 (s), 5.83 (d-like,  $J=1.0$  Hz)], one oxygenated methylene group  $\begin{bmatrix} \delta & 4.38 \\ \end{bmatrix}$  (dd,  $J=1.0$ , 11.5 Hz), 4.33 (dd,  $J=1.0$ , 11.5 Hz)], and two methoxyl groups ( $\delta$  3.75, 3.65). The <sup>13</sup>C-NMR spectrum of **2** showed 16 carbon signals, which were composed of signals due to three carboxyl carbons ( $\delta$  175.0, 171.4, 168.7), four olefinic carbons ( $\delta$  144.0, 139.2, 128.1, 112.0), one oxygenated methylene carbon ( $\delta$  61.7), two methoxyl carbons ( $\delta$  52.1, 52.0), two methine carbons ( $\delta$ 50.2, 38.9), and four methylene carbons ( $\delta$  47.9, 40.9, 31.7, 24.9). In the same manner as for 1, these  ${}^{1}H$ - and  ${}^{13}C$ -NMR



Fig. 2. CAChe Drawings and Key NOE Correlation (Arrow) Observed for 1 in the NOESY Spectrum (in CD<sub>3</sub>OD, 500 MHz)



Fig. 3. Structures of **1**—**7** and **1a**

signals were examined in detail, and the planar structure of **2**, an iridoidal alkaloid derivative possessing a  $\delta$  lactam ring and methyl 4-amino butylate group, was characterized as illustrated in Fig. 1. Since the chemical shifts of the signals

Table 3. Calculated and Observed <sup>1</sup>H-<sup>1</sup>H Coupling Constants of 1

	Dihedral angle $(°)$	Calculated (Hz)	Observed (Hz)
$H-1/H-9$	$-88.5$	$-0.3$	0.0
$H - 3/H - 4$	$-106.6$	0.6	0.0
$H - 3/H - 7$	32.4	5.8	5.0
$H-4/H-5$	77.9	0.1	0.0
$H-5/Ha-6$	63.9	1.4	0.0
$H-5/Hb-6$	$-62.6$	1.5	0.0
$H-5/H-9$	44.8	4.0	4.5
Ha-6/H-7	$-61.6$	1.6	0.0
$Hb-6/H-7$	63.6	1.4	0.0
$H-7/H-8$	$-33.0$	5.7	5.0
$H - 8/H - 9$	$-16.5$	7.5	9.5
$H - 8/H - 10$	16.9	7.5	5.0

due to H-5 and H-9 were similar to each other, the configurations at C-5 and C-9 could not be determined using NOESY spectral data. However, the coupling constant value between H-5 and H-9 and chemical shift values of the signals due to C-5 and C-9 were imposable on those of gardemide A  $(9)^{8}$ and 2-hydroxyethyl gardenamide A  $(10)$ , <sup>9)</sup> and therefore the relative configurations at C-5 and C-9 were concluded to be identical with those of **9** and **10**. The structure of **2** was therefore defined as 2-(3-methoxycarbonyl-propyl) gardenamide A.

Compound **3**, called genipaol, was obtained as a colorless syrup and exhibited an  $[M-H]$ <sup>-</sup> ion peak at  $m/z$  241 in the negative FAB-MS. The molecular formula of **3** was determined to be  $C_{11}H_{14}O_6$  using HR negative FAB-MS. The <sup>1</sup>Hand 13C-NMR spectra of **3** were similar to those of **4**, apart from the appearance of the signal due to one oxygenated quaternary carbon ( $\delta$  81.0) and lack of the signals due to one methine group. From these data, **3** was considered to be 5- or 9-hydroxy genipin. In the HMBC spectrum, a key correlation was observed between the signal due to H-3 and the signal due to methine carbon ( $\delta$  46.5, C-5). The NOESY spectrum gave correlations between  $H-1$  and  $H<sub>2</sub>-10$ . Although the configuration of **3** has not been confirmed, taking account of a biogenetic point of view, it is probably the same as that of **4**. 10) Thus **3** was considered to be 9-hydroxygenipin.

To the best of our knowledge, **1**—**3** are new monoterpenoids, and **5**—**7** are the first examples of isolation from the fruit of *G. americana*.

## **Experimental**

All the instruments and materials used were the same as those cited in a previous report, $^{11)}$  unless otherwise specified.

**Plant Material** The fruit of *G. americana* was purchased in October 1993 from the Fundation pala la Investigation Technologica del Recurso Agrobiologico Andio, a research institute of Andes agricultural bioresources in Peru and identified by Sokurates Shiota, Executive Director, Fundation pala la Investigation Technologica del Recurso Agrobiologico Andio.

**Extraction and Isolation** The dried and powdered fruit of *G. americana* (284 g) was extracted with MeOH (450 ml) five times under reflux for 1 h, and the solvent was removed under reduced pressure to give a syrup (97.5 g). The MeOH extract was chromatographed on a Diaion HP20 column, eluted with H<sub>2</sub>O, 60% MeOH, 80% MeOH, MeOH, and acetone, to give fractions (frs.) 1—7. Fraction 4 (23.6 g) was subjected to silica gel column chromatography (CC) eluted with a gradient of mixtures of  $CHCl<sub>3</sub>-MeOH-H<sub>2</sub>O$  (14:2:0.1, 10:2:0.2, 8:2:0.2, 7:3:0.5, 6:4:1,  $0:1:0$ ) to give frs. 4.1—4.10. Fractions 4.2 (498 mg) and 4.3 (581 mg) were each subjected to Chromatorex ODS CC eluted with a gradient of mixtures of MeOH–H2O (starting from 30% MeOH to 100% MeOH) to furnish **4** (16 mg) from fr. 4.2, and **3** (9 mg) and frs. 4.3.1—4.3.4 from fr. 4.3. HPLC (COSMOSIL 5C18 AR-II, Nacalai Tesque Inc., 20 mm i.d.250 mm) of fr.

4.3.1 (100 mg) eluted with 30% MeOH gave **5** (10 mg). Fraction 4.8 (2803 mg) was subjected to silica gel CC using a gradient of mixtures of  $CHCl<sub>3</sub>-MeOH-H<sub>2</sub>O$  (14:2:0.1, 10:2:0.2, 8:2:0.2, 7:3:0.5, 6:4:1, 0 : 1 : 0) as eluents to furnish frs. 4.8.1—4.8.7. A part (320 mg) of fr. 4.8.6 (1823 mg) was subjected to HPLC (COSMOSIL SIL-06, Nacalai Tesque Inc., 20 mm i.d. $\times$ 250 mm) eluted with CHCl<sub>3</sub>–MeOH–H<sub>2</sub>O (8 : 2 : 0.2) to afford frs. 4.8.6.1—4.8.6.11. HPLC (COSMOSIL 5C18 AR-II) of 4.8.6.10 (43 mg), using 25% MeOH as eluent, gave **6** (11 mg). Fraction 4.10 (9142 mg) was chromatographed on a silica gel column using a gradient of mixtures of CHCl<sub>3</sub>–MeOH–H<sub>2</sub>O (14:2:0.1, 10:2:0.2, 8:2:0.2, 7:3:0.5,  $6:4:1, 0:1:0$  as eluents to afford frs.  $4.10.1 - 4.10.8$ . A part (79 mg) of fr. 4.10.8 (4101 mg) was subjected to HPLC (COSMOSIL SIL-06) using CHCl<sub>3</sub>–MeOH–H<sub>2</sub>O  $(6:4:1)$  as eluent to afford 7  $(34 \text{ mg})$ . Fraction 5 (5.0 g) was subjected to Chromatorex ODS CC eluted with a gradient of mixtures of MeOH–H<sub>2</sub>O (30% MeOH, 40% MeOH, 50% MeOH, MeOH) to afford frs. 5.1—5.5. Fraction 5.1 (4545 mg) was chromatographed on a silica gel column using a gradient of mixtures of CHCl<sub>3</sub>-MeOH-H<sub>2</sub>O (14:2:0.1,  $10:2:0.2$ ,  $8:2:0.2$ ,  $7:3:0.5$ ,  $6:4:1$ ) as eluents to give frs. 5.1.1–5.1.9. Fraction 5.1.3 (151 mg) was subjected to HPLC (COSMOSIL 5C18 AR-II) using 35% MeOH as eluent to furnish **1** (20 mg). Chromatography of fr. 5.1.4 (179 mg) on a Chromatorex ODS column eluted with a gradient of mixtures of MeOH–H<sub>2</sub>O (30% MeOH, 40% MeOH, 50% MeOH, MeOH) gave **2** (15 mg).

Genipacetal (1): Colorless syrup.  $[\alpha]_D^{32} + 100.3^\circ$  (*c*=2.0, MeOH). Positive FAB-MS  $m/z$ : 249 [M+Na]<sup>+</sup>, negative FAB-MS  $m/z$ : 225 [M-H]<sup>-</sup>, HR positive FAB-MS  $m/z$ : 249.0737  $[M+Na]^+$  (Calcd for C<sub>11</sub>H<sub>14</sub>O<sub>5</sub>Na: 249.0739).<sup>1</sup>H-NMR spectral data: see Table 1. <sup>13</sup>C-NMR spectral data: see Table 2.

Genipamide (2): Colorless syrup.  $[\alpha]_D^{32} + 207.7^\circ$  (*c*=1.4, MeOH). Positive FAB-MS  $m/z$ : 324 [M+H]<sup>+</sup>, HR positive FAB-MS  $m/z$ : 324.1440 [M+H]<sup>+</sup> (Calcd for  $C_{16}H_{22}NO_6$ : 324.1447). <sup>1</sup>H-NMR spectral data: see Table 1. <sup>13</sup>C-NMR spectral data: see Table 2.

Genipaol (3): Colorless syrup.  $[\alpha]_D^{27} + 101.4^{\circ}$  ( $c = 1.1$ , MeOH). Negative FAB-MS  $m/z$ : 241 [M-H]<sup>-</sup>, HR negaitive FAB-MS  $m/z$ : 241.0724 [M-H]<sup>-</sup> (Calcd for  $C_{11}H_{13}O_6$ : 241.0710). <sup>1</sup>H-NMR spectral data: see Table 1. <sup>13</sup>C-NMR spectral data: see Table 2.

**Acetylation of 1** Compound  $1(5 \text{ mg})$  in Ac<sub>2</sub>O–pyridine  $(1:1, 1 \text{ ml})$  was left to stand at room temperature overnight. After removal of the reagent under a stream of  $N_2$ , the residue was partitioned between ether and  $H_2O$ . The ether layer was concentrated to give **1a** (5 mg).

1a: Syrup. <sup>1</sup>H-NMR spectral data: see Table 1.

**Computional Methods** Calculations were performed using CAChe (Version 4.1.1) with extended MM2 parameters<sup>12)</sup> (Fujitsu Co., Japan), which was run on a Macintosh Powerbook G3/400. Convergence was obtained when the difference in the energies between two successive interactions was less than 0.00001 kcal/mol. Drawing was performed using the Chem3D program (Cambridge Scientifics Computing Inc., Cambridge, MA, U.S.A.).

**Acknowledgments** We express our appreciation to Mr. K. Takeda of Kumamoto University for measurement of the NMR spectra. We thank Mr. H. Harazono of Fukuoka University for measurement of the FAB-MS.

## **References**

- 1) Ueda S., Iwahashi Y., Tokuda Y., *J. Nat. Prod.*, **54**, 1677—1680 (1991).
- 2) Ono M., Ueno M., Masuoka C., Ikeda T., Nohara T., *Chem. Pharm. Bull.*, **53**, 1342—1344 (2005).
- 3) Jensen S. R., *Phytochemistry*, **22**, 1761—1765 (1983).
- 4) Zhao W. M., Xu J. P., Qin G. W., *Phytochemistry*, **37**, 1079—1082 (1994).
- 5) Inouye H., Takeda Y., Saito S., Nishimura H., Sakuragi R., *Yakugaku Zasshi*, **94**, 577—586 (1974).
- 6) Goto H., Osawa E., *Tetrahedron Lett.*, **33**, 1343—1346 (1992).
- 7) Karplus M., *J. Chem. Phys.*, **30**, 11—15 (1959).
- 8) Machida K., Onodera R., Furuta K., Kikuchi M., *Chem. Pharm. Bull.*, **46**, 1295—1300 (1998).
- 9) Machida K., Kazumi O., Ishii M., Kakuda R., Yaoita Y., Kikuchi M., *Chem. Pharm. Bull.*, **48**, 746—748 (2000).
- 10) Horeau A., Nouaille A., *Tetrahedron Lett.*, **22**, 1939—1942 (1971).
- 11) Ono M., Nishida Y., Masuoka C., Li J.-C., Okawa M., Ikeda T., Nohara T., *J. Nat. Prod.*, **67**, 2073—2075 (2005).
- 12) Allinger N. L., *J. Am. Chem. Soc.*, **99**, 8127—8134 (1977).