Preferential Intramolecular Ring Closure of Aminoalcohols with Diethyl Carbonate to Oxazolidinones

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The closure by cyclization with diethyl carbonate $(EtO)_2CO$ from aminoalcohols 1 as starting material can lead to the oxazolidinones 2a, b and 2c, respectively. In the reaction of *trans*-isomer (6) and $(EtO)_2CO$, isolated products were also only 5-membered oxazolidinone derivative (7), containing its dehydrated derivative 8. The preferential formation of the 5-membered oxazolidinone ring system apparently indicated that this process (5-Exo-Trig ring closure) is more favorable than that of 6- or 7-membered ring derivative (3 or 9) by 6- or 7-Exo-Trig ring closure.

Kew words ring closure; aminoalcohol; diethyl carbonate; β -aminoalanines; oxazolidinone; 5-Exo-Trig

In connection with our interest in the chemistry of β -aminoalanines,¹⁻⁴) we have reported the synthetic application of these materials as a synthon for the preparation of a 5-membered oxazolidinone ring system.²⁾ Some of the oxazolidinone derivatives have attracted much attention not only as a synthetic target but also because of the importance of the related compounds in the biological activities.^{5,6)}

This paper deals with the relative rates of ring closure by cyclization with diethyl carbonate $(EtO)_2CO$ from aminoalcohols as starting material. We report here two typical experimental results for the cyclization that has two possibilities as a function of the ring intramolecular cyclization by $(EtO)_2CO.1$

Results and Discussions

The objective of this report is to explore the orientation of ring closure of aminoalcohol by $(EtO)_2CO$. The compounds $(1^{7)}$ and $6^{8)}$ are selected as the starting aminoalcohols, which have two hydroxy groups and a basic secondary amine in the molecules, so that the cyclization of these compounds employing $(EtO)_2CO$ can lead to the corresponding heterocycles (*e.g.* compounds **2** and **3**, respectively, in the case of aminoalcohol **1**).

Since the experiment using compound 1 under similar reaction conditions reported previously²) resulted in a quite complex reaction mixture by TLC analysis, it is not used to determine conditions suitable for cyclization of this aminoalcohol (1). Then, we tried the cyclization of (1) under the conditions of this aminoalcohol (1) and $(EtO)_2CO$ in a 1/1.25 ratio. In this experiment, we isolated the majority of cyclized products (2) in 81% total yield [*cis* (2a, 2b): *trans* (2c)=*ca*. 3/1 determined by ¹H-NMR]. The relative ratio of *cis/trans* was based upon the signal at δ 3.67 and δ 3.72 for *cis* isomers (2a, 2b). Because a diastereomeric mixture of aminoalcohol (1) (*cis*: *trans*=*ca*. 3/1) was used as the starting material (see Experimental), the 5-membered ring products (oxazolidinones) may generate the products 2 as a mixture of four diastereoisomers theoretically.

We were able to isolate two *cis*-isomers (**2a**, **b**, α or β stereoisomers, respectively), and our repeated trials for separation of two pure *trans*-isomers were unsuccessful. However, NMR spectroscopic analysis showed the isolated sample **2c** was a mixture of two stereoisomers (see Experimental). In this case, the preferential formation of a 5-membered oxazolidinone ring system apparently indicated that this process (5-Exo-Trig⁹⁾ ring closure) is more favorable than that of 6-membered ring derivative **3** by 6-Exo-Trig ring closure. The fact that the 6-membered ring system can be isolated easily in the reaction of a noncompetitive cyclization system, *i.e.*, the formation of compound (**5**)¹⁰ from the start-



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Chart 2

ing compound **4** is one piece of evidence in support of the above consideration (see Experimental).

In the reaction of *trans*-isomer (6)⁸⁾ and (EtO)₂CO, isolated products were also only 5-membered oxazolidinone derivatives (**7**), containing its dehydrated derivative **8**, in 53% total yield, and we could isolate none of the expectable 7membered compound such as the structure **9** by 7-Exo-Trig ring closure. In addition to the above result, the reaction of aminoalcohol **10** with (EtO)₂CO under similar conditions gave no expectable cyclized product **11**, and resulted in the recovery of the starting material.¹¹

We are considering that the stereoelectronic requirement for the ring closure to 6 or 7-membered heterocyclic compounds may give rise to higher energies of both transition states rather than that of ring closure for 5-membered oxazolidinones (5-Exo-Trig ring closure).^{9,12)}

So far as we know, no previous report has dealt with these competitive ring closure tests as described above. We may emphasize that the reaction of β -aminoalcohol with (EtO)₂CO under above limited reaction conditions at least, even in the presence of competitive hydroxy groups at the γ or δ -position, brings about cyclization to form preferentially the 5-membered oxazolidinones. Further synthetic applications of the above-mentioned information for related derivatives, particularly the quest for the biologically active lead compounds, are in progress.

Experimental

Melting points are uncorrected. IR spectra were measured with a Shimadzu FT/IR-8100 spectrometer. ¹H- and ¹³C-NMR spectra were obtained on a JEOL JNM A-500 (500 MHz for ¹H, 125 MHz for ¹³C) at 35 °C unless otherwise noted. Chemical shifts are expressed as δ ppm downfield from an internal tetramethylsilane (TMS). The signal assignments were confirmed with ¹H-¹H two-dimensional (2D) correlation spectroscopy (COSY), ¹H-¹³C heteronuclear multiple-quantum coherence (HMQC), ¹H-¹³C heteronuclear multiple-bond connectivity (HMBC) spectra. The stereochemistry of the starting materials and products were also supported by NOEs experiment. FAB-MS were obtained with a JEOL JMS-HX110 mass spectrometer. The preparation of racemic β -aminoalanines as starting materials has already been described in our previous paper.¹⁻⁴⁾ The following abbreviations in the brackets were used for the cyclohexane ring (Cyhx), the piperidine ring (Ppd), and for the oxazolidinone ring (Oxaz), respectively.

Reaction of Aminoalcohol (1) with (Et₂O)CO [Preparation of Oxazolidin-2-one (2)] A mixture of 1^{7} (0.34 g, 1.259 mmol) and freshly distillated diethyl carbonate (0.149 g, 1.590 mmol) in anhydrous benzene was refluxed for 10 min in the presence of a catalytic amount of sodium methoxide. After concentration of the solvent under reduced pressure, the residue was dissolved in Et₂O, dried over anhydrous Na₂SO₄ and evaporated under reduced pressure to give an oily residue. The residue was simply chromatographed on silica gel with EtOH to afford a total amount of 0.30 g (81% yield) of an isomeric mixture of the cyclized oxazolidinone derivatives (2a : 2b : 2c=ca. 5/5/3 determinable by ¹H-NMR). This material was carefully chromatographed on silica gel (AcOEt \rightarrow EtOH \rightarrow MeOH as eluants) to afford pure isomers 2a and 2b, and an isomeric mixture of 2c. In spite of careful analysis of 2D NMR data, we found no confirmable evidence regarding α - and β -isomers). The data of the products are shown below.

3-[*cis*-(**2-**Hydroxycyclohexyl)methyl]-4-(piperidin-1-ylmethyl)oxazolidin-2-one (Compound 2a) mp 101—103 °C. IR (KBr) cm⁻¹: 1728. ¹H-NMR (DMSO- d_6) δ : 1.16—1.78 (15H, m, Ppd H-3—H-5 and Cyhx H-2—H-6), 2.28 (1H, dd, *J*=13.0, 6.0 Hz, C<u>H</u>H-1-Ppd), 2.23—2.37 (4H, m, Ppd H-2, H-6), 2.54 (1H, dd, *J*=13.0, 6.5 Hz, CH<u>H</u>-1-Ppd), 3.06 (1H, dd, *J*=13.0, 7.0 Hz, Cyhx-C<u>H</u>H-Oxaz), 3.20 (1H, dd, *J*=13.0, 8.0 Hz, Cyhx-CH<u>H</u>-Oxaz), 3.67 (1H, br s, Cyhx H-1), 3.89 (1H, dd, *J*=8.0, 6.0 Hz, Cxaz-H_A-5), 3.93—3.96 (1H, m, Oxaz H-4), 4.14 (1H, br, OH), 4.30 (1H, dd, *J*=8.0, 8.0 Hz, Oxaz H_B-5). ¹³C-NMR (DMSO- d_6) δ : 19.8, 24.1, 24.4, 24.9, and 25.5 (×2) (Cyhx C-3—C-5 and Ppd C-3—C-5), 32.5 (Cyhx C-6), 39.4 (Cyhx C-2), 45.1 (Cyhx-<u>C</u>H₂-Oxaz), 52.6 (Oxaz C-4), 54.4 (×2) (Ppd C-2 and C-6), 61.1 (<u>C</u>H₂-1-Ppd), 64.7 (Cyhx C-1), 66.0 (Oxaz C-5), 158.2 (CO). FAB-MS (positive) *m/z*: 197 (C₁₀H₁₇N₂O⁺₂), 297 (M+H)⁺. HR-FAB-MS(+) *m/z*: 297.2173 (Calcd for C₁₆H₂₉N₂O₃: 297.2178).

3-[*cis*-(**2-**Hydroxycyclohexyl)methyl]-4-(piperidin-1-ylmethyl)oxazolidin-2-one (Compound 2b) mp 111—112 °C. IR (KBr) cm⁻¹: 1717. ¹H-NMR (DMSO- d_6) δ : 1.18—1.68 (15H, m, Ppd H-3—H-5 and Cyhx H-2— H-6), 2.28 (1H, dd, *J*=13.0, 6.0 Hz, C<u>H</u>H-1-Ppd), 2.35—2.36 (4H, m, Ppd H-2, H-6), 2.49—2.52 (1H, m, CH<u>H</u>-1-Ppd), 3.08 (1H, dd, *J*=14.0, 6.5 Hz, Cyhx-C<u>H</u>H-Oxaz), 3.20—3.28 (1H, Cyhx-CH<u>H</u>-Oxaz), 3.72 (1H, br s, Cyhx H-1), 3.91 (1H, dd, *J*=8.0, 6.0 Hz, Oxaz H_A-5), 3.94—3.97 (1H, m, Oxaz H-4), 4.30 (1H, dd, *J*=8.0, 8.0 Hz, Oxaz H_B-5), 4.36 (1H, br, OH). ¹³C-NMR (DMSO- d_6) δ : 20.0, 23.7, 23.8, 24.5, and 25.5 (×2) (Cyhx C-3—C-5 and Ppd C-3—C-5), 32.7 (Cyhx C-6), 39.0 (Cyhx C-2), 44.4 (Cyhx-<u>C</u>H₂-Oxaz), 51.8 (Oxaz C-4), 54.5 (×2) (Ppd C-2 and C-6), 60.8 (<u>C</u>H₂-1-Ppd), 65.5 (Cyhx C-1), 66.0 (Oxaz C-5), 157.9 (CO). FAB-MS (positive) *m/z*: 197 (C₁₀H₁₇N₂O⁺₂), 297 (M+H)⁺. HR-FAB-MS(+) *m/z*: 297.2178 (Calcd for C₁₆H₂₉N₂O₃: 297.2178).

3-[trans-(2-Hydroxycyclohexyl)methyl]-4-(piperidin-1-ylmethyl)oxa-

zolidin-2-one (Compound 2c) Obtained as oil. IR (KBr) cm⁻¹: 1744. ¹H-NMR (DMSO- d_6) δ: 0.82—1.82 (15H, m, Ppd H-3—H-5 and Cyhx H-2—H-6), 2.28 (1H, dd, J=12.5, 7.0 Hz, C<u>H</u>H-1-Ppd), 2.49—2.50 (4H, m, Ppd H-2, H-6), 2.53 (1H, dd, J=12.5, 5.5 Hz, CH<u>H</u>-1-Ppd), 3.04—3.28 (2H, m, Cyhx-C<u>H</u>H-Oxaz+Cyhx H-1), 3.44—3.50 (1H, m, Cyhx-C<u>H</u>H-1-Oxaz), 3.91 (1H, dd, J=8.0, 5.5 Hz, Oxaz H_a-5), 3.94—3.98 (1H, m, Oxaz H-4), 4.28 (1H, dd, J=8.0, 8.0 Hz, Oxaz H_B-5), 4.52 (1H, br, OH). ¹³C-NMR (DMSO- d_6) δ: 23.8, 24.1, 24.2, 24.7, 24.8, 25.4 (×2), 25.5 (×2), 25.6, 28.0, and 28.4 (Cyhx C-3—C-5 and Ppd C-3—C-5), 35.2 (Cyhx C-6 for *α*- and *β*-isomers), 42.3 and 44.0 (Cyhx C-2), 44.5 and 45.6 (Cyhx-<u>C</u>H₂-Oxaz), 51.0 and 53.8 (Oxaz C-4), 54.0 and 54.6 (Ppd C-2 and C-6), 60.0 and 60.8 (CH₂-1-Ppd), 66.0 and 66.1 (Oxaz C-5), 71.2 and 73.9 (Cyhx C-1), 157.9 and 158.0 (CO). FAB-MS (positive) *m*/*z*: 197 (C₁₀H₁₇N₂O⁺₂), 297 (M+H)⁺. HR-FAB-MS(+) *m*/*z*: 297.2171 (Calcd for C₁₆H₂₀N₂O₃: 297.2178).

Reaction of Aminoalcohol (6) with (Et₂O)CO [Preparation of oxazolidin-2-ones (7, 8)] Freshly distillated diethyl carbonate (0.108 g, 0.915 mmol) and catalytic amounts of sodium methoxide were added to a solution of **6** (0.3 g, 0.833 mmol) in anhydrous benzene and the mixture was heated at 110°C for 10 min. The resulting mixture was dissolved in Et₂O and extracted with 1 N-HCl. The acidic aqueous layer was washed with Et₂O, neutralized with K₂CO₃, and then extracted with Et₂O. The ethereal extract was dried over anhydrous Na₂SO₄. Evaporation and purification by chromatography on silica gel (AcOEt) gave the free product 7 (170 mg, 53%), in which dehydrated compound **8** (*ca.* 18%) was detected by ¹H-NMR analysis. This material was carefully chromatographed on silica gel (AcOEt) to afford pure products 7 and **8** respectively. Both compounds α - and β -isomers and stereochemistry of the hydroxy group in compound 7 could not be specified. *E* and *Z* isomers for the compound **8** were not also determinable by careful spectroscopic analysis. The data of the products are shown below.

3-[trans-(2-(Hydroxy(phenyl)methyl)cyclohexyl)methyl]-4-(piperidin-1-ylmethyl)oxazolidin-2-one (Compound 7) Colorless very hygroscopic viscous oil. IR (KBr) cm⁻¹: 3453, 1752, 1743, 1727. ¹H-NMR (DMSO-d₆) δ: 1.13-1.62 (15H, m, Cyhx H-2-H-6 and Ppd H-3-H-5), 1.90-1.98 (1H, m, Cyhx H-1), 2.29-2.40 (5H, m, Ppd H-2, H-6 and CHH-Ppd), 2.55—2.59 (1H, m, CHH-Ppd), 3.21 (1H, dd, J=14.0, 11.0 Hz, CHH-Cyhx), 3.44 (1H, J=14.0, 4.5 Hz, CHH-Cyhx), 3.86-3.87 (1H, m, Oxaz H-4), 3.96—4.00 (1H, m, Oxaz H_A-5), 4.31 (1H, dd, J=8.5, 8.5 Hz, Oxaz H_B-5), 4.68 (1H, d, J=6.0 Hz, Ph-CHOH), 5.14 (1H, br s, OH), 7.21-7.34 (5H, m, ArH). ¹³C-NMR (DMSO- d_6) δ : 23.2, 23.8 (×2), 25.5 (×2), 25.8 and 26.6 (Ppd C-3-C-5 and Cyhx C-3-C-6), 34.0 (Cyhx C-1), 44.9 (CH2-Cyhx), 45.8 (Cyhx C-2), 51.1 (Oxaz C-4), 54.7 (×2) (Ppd C-2 and C-6), 60.1 (CH2-1-Ppd), 66.2 (Oxaz C-5), 73.2 (Ph-CH-OH), 126.5 (×2), 126.6, and 127.7 (×2) (Ar C-2-C-6), 144.8 (Ar C-1), 157.8 (CO). HR-FAB-MS(-) m/z: 385.2491 (C23H33N2O3: 385.2475). Anal. Calcd for C23H34N2O3 · HCl· 2.6H2O: C, 64.83; H, 9.52; N, 9.45. Found: C, 64.93; H, 9.56; N, 9.31.

3-[(2-Benzylidenecyclohexyl)methyl]-4-(piperidin-1-ylmethyl)oxazolidin-2-one (Compound 8) Colorless hygroscopic viscous oil. IR (KBr) cm⁻¹: 1752. ¹H-NMR (DMSO- d_6) δ : 1.16—2.56 (21H, Cyhx H-1, H-3—H-6, Ppd C-2—C-6, and CH₂-1-Ppd), 3.20—3.42 (2H, m, CH₂-Cyhx), 3.77—3.99 (2H, m, Oxaz H-4, H_A-5), 4.20—4.32 (1H, m, Oxaz H_B-5), 5.63—5.72 (1H, m, PhCH=), 7.20—7.36 (5H, m, ArH). ¹³C-NMR (DMSO- d_6) δ : 22.1, 22.6, 23.0, 23.1, 23.5, 24.5, 25.1, 25.3 (×4), 25.5, 25.8 and 26.2 (Cyhx C-3—C-6 and Ppd C-3—C-5), 34.0 and 42.3 (Cyhx C-1), 44.9 and 45.6 (CH₂-Cyhx), 51.2 and 52.7 (Oxaz C-4), 54.3 (×2) and 54.4 (×2) (Ppd C-2 and C6), 59.2 and 60.4 (CH₂-1-Ppd), 65.7 and 65.9 (Oxaz C-5), 80.2 and 80.3 (Ph-CH=), 126.1 (×2), 126.2 (×2), 126.4, 127.3, 127.8 (×2) and 127.9 (×2) (Ar C-2—C-6), 138.1 and 138.3 (Ar C-1), 153.0 and 153.1 (Cyhx C-2), 157.5 and 157.6 (CO). FAB-MS (positive) *m/z*: 369 (M+H)⁺. *Anal.* Calcd for C₂₃H₃₂N₂O₂·1.3H₂O: C, 70.48; H, 8.90; N, 7.15. Found: C, 70.51; H, 8.63; N, 6.99.

References and Notes

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- 7) This compound was prepared from 2-hydroxycyclohexanecarboxylic acid and β-piperidinoalanine according to the procedure reported previously.^{1,2)} For the preparation of 2-hydroxycyclohexanecarboxylic acid, see de Raadt A., Griengl H., Petsch M., Plachota P., Schoo N., Weber H., *Tetrahedron: Asymmetry*, **7**, 473–490 (1996). Compound (1): IR (KBr) cm⁻¹: 3410. ¹H-NMR (DMSO-*d*₆) δ: 1.05–1.64 (15H, m, Cyhx H-2–H-6 and Ppd H-3–H-5), 2.18–2.38 (6H, m, CH₂-1-Ppd and Ppd H-2, H-6), 2.42–2.52 (1H, m, Cyhx-CHH), 2.56–2.62 (1H, m, NHCHCH₂N=), 2.64–2.71 (1H, m, Cyhx-CHH), 3.0–3.3 (3H, br, OH ×2, and NH), 3.31–3.34 (2H, m, CH₂OH), 3.79–3.83 (1H, m, Cyhx H-1). FAB-MS (positive) *m/z*: 271 (M+H)⁺. *Anal.* Calcd for C₁₅H₃₀N₂O₂·0.4H₂O: C, 64.90; H, 11.18; N, 10.09. Found: C, 64.93; H, 11.17; N, 10.01.
- 8) This compound was prepared from *trans* 2-benzoylcyclohexanecarboxylic acid and β-piperidinoalanine by the same method described previously.^{1,2)} For the preparation of *trans* 2-benzoylcyclohexanecarboxylic acid, see Miyano S., Abe N., Fujisaki F., Sumoto K., *Heterocycles*, **26**, 1813—1826 (1987). Compound (**6**): IR (KBr) cm⁻¹: 3410, 3303. ¹H-NMR (DMSO-*d*₆, 70 °C) δ: 1.04—1.75 (16H, m, Cyhx H-1—H-6, and Ppd H-3—H-5), 2.24—2.82 (9H, m, Ppd H-2, H-6, CH₂-Ppd, Cyhx-CH₂NHCHCH₂ and CH₂NHCHCH₂N=), 3.37 (2H, dd, *J*=5.5, 4.0 Hz, CH₂OH), 3.8—4.8 (2H, br, OH and OH or NH), 4.59 (1H, dd, *J*=13.0, 6.5 Hz, PhCHOH), 5.0—6.0 (1H, br NH or OH), 7.25—7.28 (5H, m, ArH). FAB-MS (positive) *m/z*: 361 (M+H)⁺. *Anal.* Calcd for C₂₂H₃₆N₂O₂·0.15H₂O: C, 72.74; H, 10.07; N, 7.71. Found: C, 72.76; H, 10.16; N, 7.43.
- 9) Baldwin J. E., J. Chem. Soc., Chem. Commun., 1976, 734 (1976).
- 10) This compound was obtained from the reaction in the same manner described for (2). Starting 2-hydroxycyclohexanecarboxylic acid (*cis*: *trans*=*ca*. 3/1) was obtained by the method reported previously.⁷⁾ The product **5** obtained in this experiment was an oily material. IR (KBr) cm⁻¹: 3453 1691. ¹H-NMR (DMSO-*d*₆) &: 1.21—1.81 (6H, m, Cyhx H-5, H_A-6, H-7 and H_A-8), 1.61—1.68 (1H, m, Cyhx H_B-6), 1.74—1.81 (1H, m, Cyhx H_B-8), 1.93—1.99 (1H, m, H-4a), 2.81 (3H, s, CH₃), 2.90 (1H, dd, *J*=12.0, 2.5 Hz, H-4), 3.44 (1H, dd *J*=12.0, 5.5 Hz, H-4), 4.39—4.41 (1H, m, H-8a). ¹³C-NMR (DMSO-*d*₆) &: 19.5, and 23.2 (C-7), 23.5, 23.9, 24.4, and 27.0 (C-5 and C-6), 29.2 and 30.8 (C-8), 31.4 and 35.8 (C-4a), 35.9 and 36.5 (CH₃), 51.3 and 51.9 (C-4), 74.1 and 78.2 (C-8a), 152.5 (×2) (CO). FAB-MS (positive) *m*/*z*: 170 (M+H)⁺. *Anal.* Calcd for C₉H₁₅NO₂·0.15H₂O: C, 62.88; H, 8.97; N, 8.15. Found: C, 62.85; H, 9.07; N, 7.87.
- 11) Other reaction conditions (including post-treatments of the reaction mixture) with the compound (10) provide a complex mixture including a few unknown unstable compounds, and our attempts to isolate the target 7-membered heterocycle (11) under a few different reaction conditions were unsuccessful. However, the observations of the formation of a product showed an IR absorption band at 1686 cm⁻¹ and an ion peak of 258 in negative FAB-MS, though the product is still as a labile mixture, are strongly indicating the formation of the target 7-membered compound (11).
- Both types of noncompetitive ring closures by 6-Exo-Trig and 7-Exo-Trig process are classified as allowed ring closure reactions in Baldwin's report.⁹⁾