

Chemical Constituents of Nepalese Propolis (II)

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A novel flavanone (1), three new isoflavones (2–4) and a new flavan-3-ol (5) were isolated along with ten other known flavonoids (6–15) from the methanolic extract of propolis collected from Chitwan, Nepal. Their structures were determined on the basis of spectral analysis.

Key words propolis; dihydroflavonol; isoflavone; flavan-3-ol; stopped-flow LC-CD spectrum

Propolis, a complex resinous material collected by honeybees from buds and exudates of certain plant sources neighboring its hive, is considered to possess broad spectrum of biological activities and has historical utilization in folk medicine. Thus, it is extensively being used in health food, pharmaceutical preparations *etc.* The chemical consistency of propolis is highly dependent on the flora of the region from where it is collected.^{1–5)}

As a part of our ongoing study on propolis of different geographical locations of Nepal, previously we have reported three new and nine known flavonoids from propolis of Chitwan.⁶⁾ In this paper, we will report on further isolation of new compounds from the remaining fractions and sub fractions of the same propolis.

Results and Discussion

The separation of untouched fraction 2 by column of MCI gel CHP-20P followed by Lobar RP-8 column, normal-phase PTLC and ODS HPLC, and continued chromatography of fraction 3 by normal-phase PTLC and ODS HPLC afforded five new compounds (1–5) together with ten previously reported compounds: Odoratin (6),⁷⁾ (+)-Medicarpin (7),⁸⁾ (+)-Vesticarpan (8),⁹⁾ Plathymenin (9),¹⁰⁾ (2*S*)-7-Methoxyflavanone (10),¹¹⁾ (2*S*)-7-Hydroxyflavanone (11),¹²⁾ *S*-4-Methoxydalbergiquinol (12),¹³⁾ (4*S*,6*S*)-4-Hydroxy-3-methoxy-6-(1-phenyl-2-propenyl)-2-cyclohexene-1-one (13),¹⁴⁾ Cearoin (14)¹⁵⁾ and (1*R*,2*S*,4*S*,5*S*)-2-methoxy-5-

[(1*R*)-1-phenyl-2-propenyl]-1,4-cyclohexanediol (15).¹⁴⁾

Compound 1 was obtained as yellow crystals with molecular formula C₁₅H₁₂O₅, as established by the molecular ion peak at *m/z*: 272.0655 [M]⁺ in its HR-EI-MS. Its ¹H-NMR spectrum (Table 1) displayed two *trans*-diaxially related oxygenated methine doublets at δ_H 5.04 (1H, d, *J*=11.7 Hz, H-2) and 4.48 (1H, d, *J*=11.7 Hz, H-3), signals due to a mono-substituted benzene ring at δ_H 7.54 (2H, dd, *J*=8.0, 1.4 Hz, H-2',6'), 7.41 (1H, dt, *J*=8.0, 1.4 Hz, H-3',5') and 7.38 (1H,

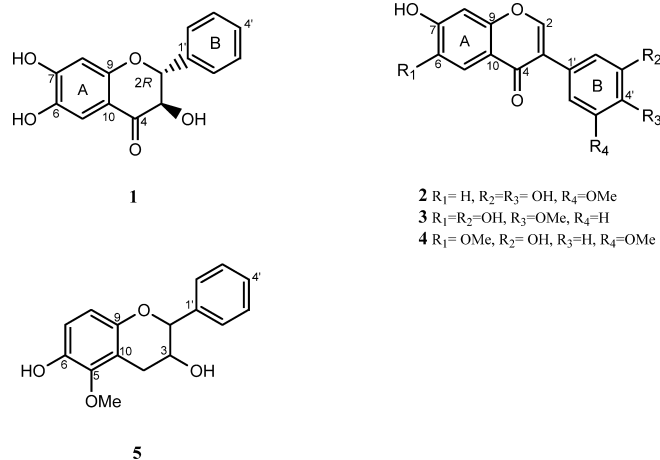


Fig. 1. Structures of New Compounds (1–5)

Table 1. ¹H-NMR Data (δ, *J* in Hz) of Compounds 1–5 (500 MHz)

| Proton | 1 ^{a)} | 2 ^{a)} | 3 ^{b)} | 4 ^{a)} | 5 ^{a)} |
|--------|--------------------|-----------------|--------------------|-----------------|---------------------------|
| 2 | 5.04 d (11.7) | 8.16 s | 8.18 s | 8.12 s | 4.61 d (5.0) |
| 3 | 4.48 d (11.7) | | | | 3.96 ddd (10.0, 5.0, 2.5) |
| 4 | | | | | 2.85 dd (14.5, 2.5) |
| 5 | 7.19 s | 7.58 d (8.5) | 8.27 s | 7.55 s | 2.49 dd (14.5, 10.0) |
| 6 | | 6.96 br d (8.5) | | | |
| 7 | | | | | 6.52 d (8.5) |
| 8 | 6.39 s | 6.95 br s | 7.28 s | 6.92 s | 6.47 d (8.5) |
| 2' | 7.54 dd (8.0, 1.4) | 6.95 br s | 7.80 d (2.0) | 6.97 br s | 7.42 dd (7.3, 1.5) |
| 3' | 7.41 dt (8.0, 1.4) | | | | 7.33 dt (7.3, 1.5) |
| 4' | 7.38 tt (8.0, 1.4) | | | 6.97 br s | 7.25 tt (7.3, 1.5) |
| 5' | 7.41 dt (8.0, 1.4) | | 7.03 d (8.4) | | 7.33 dt (7.3, 1.5) |
| 6' | 7.54 dd (8.0, 1.4) | 7.06 br s | 7.33 dd (8.4, 2.0) | 7.05 br s | 7.42 dd (7.3, 1.5) |
| 5 OMe | | | | 3.95 s | 3.62 s |
| 6 OMe | | | | | |
| 4' OMe | | | 3.77 s | | |
| 5' OMe | | 3.87 s | | 3.88 s | |

a) Measured in CD₃OD. b) Measured in C₅D₅N.

tt, $J=8.0, 1.4\text{ Hz}$, H-4'), and two aromatic singlet protons at δ_{H} 7.19 (1H, s, H-5) and δ_{H} 6.39 (1H, s, H-8). Consistent with the $^1\text{H-NMR}$ spectral analysis, the $^{13}\text{C-NMR}$ spectrum of **1** also displayed two oxygenated methine type resonances at δ_{C} 85.9 (C-2) and δ_{C} 74.8 (C-3), the signals due to two aromatic rings (with one mono-substituted) (Table 2), and a conjugated ketone at δ_{C} 194.4 (C-4). Accordingly, this suggested that compound **1** could be a flavanone with an unsubstituted B ring.^{16–18} Thus, presence of three hydroxyl groups in **1** could be inferred from the molecular formula $\text{C}_{15}\text{H}_{12}\text{O}_5$. Also, by comparing with literature values of similar compounds, the position of hydroxyl group attached to carbons at δ_{C} 138.9 and δ_{C} 156.1 in $^{13}\text{C-NMR}$ spectrum could be unambiguously assigned to position 6 and 7, respectively.¹⁹ This hypothesized structure was confirmed by HMBC correlations as summarized in Fig. 2a. The stopped-flow LC-CD spectrum of compound **1** displayed positive and negative Cotton effects at 338 nm and 286 nm, respectively, which suggests the absolute stereochemistry of C-2 to be *R*.²⁰ Thus,

Table 2. $^{13}\text{C-NMR}$ Data (125 MHz) of Compounds **1**–**5**

| Position | 1 ^{a)} | 2 ^{a)} | 3 ^{b)} | 4 ^{a)} | 5 ^{a)} |
|----------|------------------------|------------------------|------------------------|------------------------|------------------------|
| 2 | 85.9 | 154.6 | 152.5 | 154.6 | 78.4 |
| 3 | 74.8 | 125.3 | 124.2 | 126.5 | 77.2 |
| 4 | 194.4 | 178.5 | 175.7 | 177.7 | 33.3 |
| 5 | 111.7 | 117.4 | 109.5 | 105.5 | 148.2 |
| 6 | 138.9 | 115.4 | 146.5 | 148.8 | 139.4 |
| 7 | 156.1 | 151.7 | 154.4 | 155.3 | 121.8 |
| 8 | 104.2 | 112.7 | 103.7 | 104.0 | 111.8 |
| 9 | 158.8 | 148.4 | 152.2 | 154.4 | 146.0 |
| 10 | 112.6 | 118.8 | 118.2 | 117.8 | 124.8 |
| 1' | 138.9 | 126.3 | 126.7 | 125.3 | 143.5 |
| 2' | 128.9 | 121.7 | 117.9 | 112.7 | 128.3 |
| 3' | 129.4 | 134.2 | 147.9 | 147.5 | 129.0 |
| 4' | 129.7 | 147.4 | 148.6 | 121.7 | 129.0 |
| 5' | 129.4 | 149.2 | 112.4 | 149.2 | 129.0 |
| 6' | 128.9 | 117.5 | 120.5 | 117.5 | 128.3 |
| 5 OMe | | | | | 60.7 |
| 6 OMe | | | | 56.7 | |
| 4' OMe | | | 56.0 | | |
| 5' OMe | | 56.5 | | 56.5 | |

a) Measured in CD_3OD . b) Measured in $\text{C}_2\text{D}_5\text{N}$.

the new compound **1** was structurally characterized as (2*R*,3*R*)-3,6,7-trihydroxyflavanone.

Compound **2** was isolated as colorless crystals and gave a molecular ion peak at m/z : 301.0695 $[\text{M}+\text{H}]^+$ in its HR-FAB-MS corresponding to molecular formula $\text{C}_{16}\text{H}_{12}\text{O}_6$. The absorption band at 260 nm in the UV spectrum in combination with a singlet resonance in $^1\text{H-NMR}$ spectrum (Table 1) at 8.16 ppm and corresponding olefinic oxymethine signal in $^{13}\text{C-NMR}$ spectrum (Table 2) at 154.6 ppm due to H-2 and C-2, respectively, suggested this compound to have an isoflavone skeleton.^{16,18,21} The substitution pattern of ring A and ring B was determined by analysis of coupling constants, and $^1\text{H-}^1\text{H}$ COSY, HMQC and HMBC experiments. A pair of AB proton doublets at δ_{H} 7.58 (1H, d, $J=8.5\text{ Hz}$) and δ_{H} 6.96 (1H, br d, $J=8.5\text{ Hz}$) along with a broad singlet at δ_{H} 6.95 observed in $^1\text{H-NMR}$ spectrum were assigned to aromatic protons H-5, H-6 and H-8, respectively, while hydroxyl group attached to carbon (δ_{C} 151.7) was placed at position 7 on the basis HMBC cross-peaks observed for H-5 with C-4, C-7 and C-9, and for H-8 with C-6, C-7, C-9 and C-10. Moreover, HMBC cross-peaks observed for the broad singlets at δ_{H} 6.95 (H-2') and δ_{H} 7.06 (H-6') with C-3 and C-4' (δ_{C} 147.4), for H-2' with C-3' (δ_{C} 134.2), and for methoxyl protons and H-6' with C-5' (δ_{C} 149.2) implied the B ring substitution pattern for **2** to be as shown in Fig. 2b. The position of methoxyl group was further confirmed by the NOESY interaction observed between H-6' and the methoxyl protons. Thus, the structure of **2** was determined to be 5'-methoxy-7,3',4'-trihydroxyisoflavone.

Compound **3** was isolated as colorless crystals, and has molecular formula $\text{C}_{16}\text{H}_{12}\text{O}_6$ as determined by the molecular ion peak at m/z : 301.0695 $[\text{M}+\text{H}]^+$ in its HR-FAB-MS. Like in compound **2**, a singlet at 8.18 ppm (H-2) in its $^1\text{H-NMR}$ spectrum in combination with resonances at 152.5 ppm (C-2) and 175.7 ppm (C-4) in $^{13}\text{C-NMR}$ spectrum concluded this compound to have an isoflavone skeleton. A sharp singlet at δ_{H} 8.27 (1H, s) which showed HMBC correlation with the signal due to the carbonyl carbon (C-4) was assigned H-5, while another singlet at δ_{H} 7.28 (1H, s) was attributed to H-8 and the hydroxyl groups were placed at position 6 and 7 on the basis of HMBC cross-peaks observed for H-5 and H-8

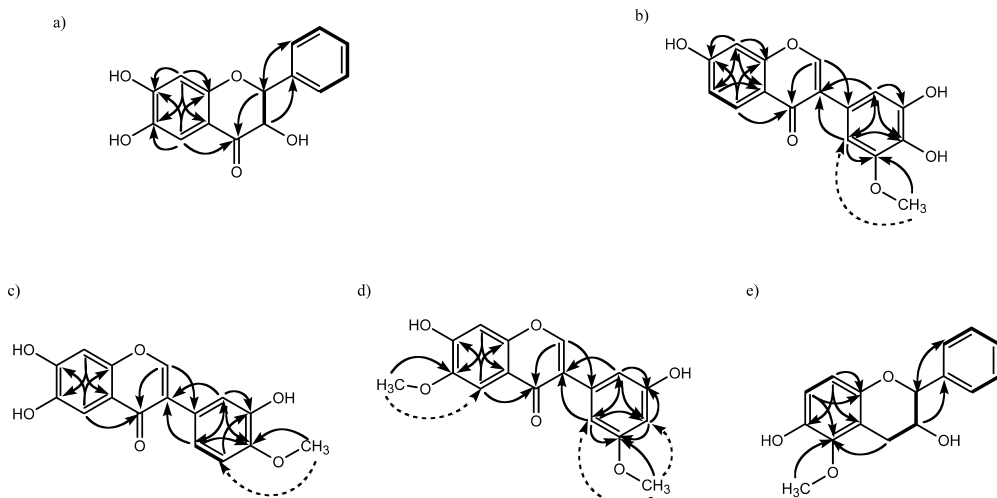


Fig. 2. (a), (b), (c), (d) and (e) Connectivities (Bold Lines) Deduced by the COSY Spectrum and Significant HMBC Correlations (Arrow) Observed for **1**, **2**, **3**, **4** and **5**, Respectively. Selected NOESY Correlations (Dashed Arrow) Observed for **2**, **3** and **4**.

with C-6 (δ_c 146.5), C-7 (δ_c 154.4), C-9 (δ_c 152.2) and C-10 (δ_c 118.2). The decision concerning the relative position of hydroxyl group at 6 and 7 was made by comparing the ^{13}C -NMR data with literature values.¹⁹⁾ The signals due to an ABX spin system at δ_H 7.03 (1H, d, $J=8.4$ Hz), δ_H 7.33 (1H, dd, $J=8.4, 2.0$ Hz) and δ_H 7.80 (1H, d, $J=2.0$ Hz) observed in ^1H -NMR spectrum were assigned to aromatic protons H-5', H-6' and H-2', respectively, on the basis of HMBC correlations from H-2', H-6' to C-3. The HMBC correlations from H-2', H-6' and methoxyl protons to C-4' (δ_c 148.6), from H-2' and H-5' to C-3' (δ_c 147.9) together with the NOESY correlation between H-5' and the methoxyl protons confirmed the position of methoxyl group at 4' and hydroxyl group at 3'.

The HR-FAB-MS of compound **4** displayed molecular ion at m/z 315.0890 $[\text{M}+\text{H}]^+$, in agreement with the molecular formula $\text{C}_{17}\text{H}_{14}\text{O}_6$. Its ^1H -NMR spectrum also showed the characteristic singlet at 8.12 ppm for isoflavone along with five singlets in aromatic region and two signals for methoxyl group at δ_H 3.95 and δ_H 3.88, while its ^{13}C -NMR spectrum resembled to that of **3** except in having one additional signal for methoxyl group at δ_c 56.7 (OCH₃-6), downfield shift of signal due to 5' and upfield shift of signal due to 4', indicating the presence of hydroxyl or methoxyl function at position 6, 7, 3' and 5' in **4** instead of 6, 7, 3' and 4' in **3**. Observation of HMBC correlation from OCH₃-6 to C-6, from OCH₃-5' to C-5' implied the hydroxyl group to be linked to C-7 and C-3', and methoxyl group to be linked to C-6 and C-5', with other HMBC correlations in **4** being the same as those of **3**. In addition, the NOESY cross-peak between H-5 and OCH₃-6, and H-6' and OCH₃-5' also confirmed positions of methoxyl groups at position 6 and 5'.

Compound **5** was isolated as yellow solid, and has molecular formula $\text{C}_{16}\text{H}_{16}\text{O}_4$ as determined by the molecular ion peak at m/z : 273.1125 $[\text{M}+\text{H}]^+$ in its HR-FAB-MS. Its ^1H -NMR spectrum showed four signals due to two methine protons at 4.61 (1H, d, $J=5.0$ Hz), 3.96 (1H, ddd, $J=10.0, 5.0, 2.5$ Hz) and methylene protons at 2.85 (1H, dd, $J=14.5, 2.5$ Hz), 2.49 (1H, dd, $J=14.5, 10.0$ Hz) suggesting this compound could be a flavan-3-ol.²²⁾ This suggestion was further supported by the signals observed in its ^{13}C -NMR spectrum at δ_c 78.4, 77.2 and 33.3 for C-2, C-3 and C-4, respectively.^{18,22)} Moreover, the signals at δ_H 7.42 (2H, dd, $J=7.3, 1.5$ Hz, H-2',6'), 7.33 (2H, dt, $J=7.3, 1.5$ Hz, H-3',5') and 7.25 (1H, tt, $J=7.3, 1.5$ Hz, H-4') suggested ring B to be unsubstituted. The two *ortho*-coupled doublets at δ_H 6.52 (1H, d, $J=8.5$ Hz) and 6.47 (1H, d, $J=8.5$ Hz) were assigned to aromatic protons H-7 and H-8, respectively, on the basis of HMBC cross-peaks observed for H-8 with C-9 and C-10. The HMBC cross-peaks observed for a singlet due to methoxyl group at δ_H 3.62 and protons at position 4 with C-5 concluded the position of methoxyl group at position 5. Full assignment of the ^1H - and ^{13}C -NMR chemical shifts were established by the analysis of H-H COSY, HMQC and HMBC spectral data as shown in Fig. 2e furnishing compound **5** as 3,6-diol-5-methoxyflavan.

Experimental

General Optical rotations were measured on a JASCO DIP-1000 polarimeter. ^1H -, ^{13}C -NMR spectra were recorded on a JEOL A-500 FT-NMR spectrometer, and the chemical shifts were expressed in the δ (ppm) scale with the TMS as an internal standard. FAB-MS, HR-FAB-MS, EI-MS and

HR-EI-MS were recorded on a JEOL JMS-700 spectrometer. Analytical and preparative TLC were performed on precoated silica gel 60 F₂₅₄ or RP-18 F₂₅₄S plates (Merck, 0.25 or 0.50 mm thickness), and detection was carried out by spraying EtOH-H₂SO₄ reagent followed by heating. Column chromatography was carried out on a silica gel (Silica gel 60, Merck), Lobar LiChroprep RP-18 (Merck), Diaion HP-20 (Mitsubishi Chemical Corporation) and MCI gel CHP-20P (Mitsubishi Chemical Corporation). HPLC-CD spectra were measured in stopped-flow mode using Capcell Pak C₁₈ column (5 μm , 4.6 mm i.d. \times 250 mm, Shiseido Fine Chemicals) at room temperature, eluted with MeOH-H₂O (2:3) at flow rate of 0.5 ml min⁻¹ on a Jasco CD-2095_{Plus} Chiral detector equipped with two Shimadzu LC-10AD_{VP} pumps, CTO-10A_{VP} column oven and SCL-10A_{VP} system controller.

Biological Material Propolis was collected in Chitwan, Nepal, in 2005. A Voucher specimen has been deposited at the Division of Natural medicines, Kyoritsu University of Pharmacy.

Extraction and Isolation Propolis (500 g) collected from Chitwan (Nepal), was successively extracted with refluxing MeOH and water to give methanol (303 g) and water (35.3 g) extracts, respectively. Part of the MeOH extract (70.7 g) was passed through Diaion HP-20 and eluted successively with H₂O, 50% MeOH, MeOH and CHCl₃ to get water eluate (0.61 g), 50% MeOH eluate (19.0 g), MeOH eluate (43.2 g) and CHCl₃ eluate (5.40 g), respectively. The 50% MeOH eluate was further subjected to a column of MCI gel CHP-20P eluted with CH₃CN-MeOH-H₂O (1:1:4→1:1:0) to afford five subfractions [fr. 1: CH₃CN-MeOH-H₂O (1:1:4) eluate, 1.15 g; fr. 2: CH₃CN-MeOH-H₂O (1:1:3) eluate, 3.32 g; fr. 3: CH₃CN-MeOH-H₂O (1:1:2) eluate, 5.43 g; fr. 4: CH₃CN-MeOH-H₂O (1:1:1) eluate, 6.09 g; fr. 5: CH₃CN-MeOH-H₂O (1:1:0) eluate, 1.33 g].

Fraction 2 (3.32 g) was fractionated on a MCI gel CHP-20P column eluted with CH₃CN-H₂O (1:4→1:0) to get six fractions (fr. 2a-1, 0.02 g; fr. 2a-2, 1.02 g; fr. 2a-3, 1.09 g; fr. 2a-4, 0.62 g; fr. 2a-5, 0.47 g; fr. 2a-6, 0.10 g). Subfraction 2a-1 to 2a-4 were chromatographed on Lobar RP-8 column with H₂O-MeOH (40%→0% H₂O), followed by normal-phase preparative TLC with CH₃CN-C₆H₆ (1:8) and ODS gel preparative HPLC (CH₃CN:H₂O, 1:3→4:1) to get **4** (21.5 mg), **6** (9.70 mg), **7** (208.4 mg), **8** (17.6 mg), **9** (40.1 mg), **10** (7.41 mg), **11** (27.6 mg) and **15** (19.3 mg).

Fraction 3 (5.43 g) was rechromatographed on MCI gel CHP-20P column with CH₃CN-MeOH-H₂O (1:1:4→1:1:0) to afford six subfractions (fr. 2-1, 0.06 g; fr. 2-2, 1.23 g; fr. 2-3, 1.01 g; fr. 2-4, 1.60 g; fr. 2-5, 0.74 g; fr. 2-6, 0.12 g). Subfractions 2-2 to 2-6 were further chromatographed on Lobar RP-18 column eluted with H₂O-MeOH (40%→0% H₂O), followed by normal-phase preparative TLC with CH₃CN-C₆H₆ (1:5) and ODS gel preparative HPLC (CH₃CN:MeOH:H₂O, 1:1:3→1:1:0), and recrystallization gave **1** (8.3 mg), **2** (33.6 mg), **3** (29.1 mg), **5** (2.45 mg), **12** (22.5 mg), **13** (52.2 mg) and **14** (108.1 mg).

Compound **1**: Yellow crystals; $[\alpha]_D^{26} +28.2^\circ$ ($c=0.03$, CHCl₃); HR-EI-MS m/z : 272.0655 [Calcd for C₁₅H₁₂O₅, 272.0685]; CD (40% MeOH-H₂O) $[\theta]_{338} +3380$; $[\theta]_{286} -15,990$; ^1H - and ^{13}C -NMR data, see Tables 1 and 2.

Compound **2**: Colorless crystals; HR-FAB-MS m/z : 301.0695 (Calcd for C₁₆H₁₃O₆ $[\text{M}+\text{H}]^+$, 301.0712); UV (MeOH) λ_{max} nm: 260.2, 297; ^1H - and ^{13}C -NMR, see Tables 1 and 2.

Compound **3**: Colorless crystals; HR-FAB-MS m/z : 301.0695 (Calcd for C₁₆H₁₃O₆ $[\text{M}+\text{H}]^+$, 301.0712); ^1H - and ^{13}C -NMR, see Tables 1 and 2.

Compound **4**: White solid; HR-FAB-MS m/z : 315.0890 (Calcd for C₁₇H₁₅O₆ $[\text{M}+\text{H}]^+$, 315.0869); ^1H - and ^{13}C -NMR, see Tables 1 and 2.

Compound **5**: Yellow solid; $[\alpha]_D^{26} +8.23^\circ$ ($c=0.01$, CHCl₃); HR-FAB-MS m/z : 273.1125 (Calcd for C₁₇H₁₅O₅ $[\text{M}+\text{H}]^+$, 273.1127); ^1H - and ^{13}C -NMR data, see Tables 1 and 2.

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