

Flavonoid Constituents from *Spiranthes australis* LINDL.

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Chemical investigation of the whole plant of *Spiranthes australis* (R. BROWN) LINDL. resulted in the isolation and characterization of three new flavonoid constituents, 5-hydroxy-4'-[(2-isopentenyl)oxy]-3,7,3'-trimethoxyflavone (1), 3-O-[[O-β-D-xylopyranosyl]-(1→2)-β-D-glucopyranosyl]-8-(p-hydroxy-benzyl)-kaempferol (2) and 3-O-[[O-2-O-(E)-p-coumaroyl-β-D-xylopyranosyl]-(1→2)-β-D-glucopyranosyl]-8-(p-hydroxy-benzyl)-kaempferol (3), together with six known flavonoid compounds. The structures of new compounds were determined on the basis of spectroscopic analysis, including HR-ESI-MS, 1D- and 2D-NMR techniques and chemical methods.

Key words Orchidaceae; *Spiranthes australis* LINDL.; flavonoid

The traditional Chinese medicine “Panlongcen” is derived from the dried root or whole plant of *Spiranthes australis* (R. BROWN) LINDL., which is an Orchidaceae plant distributed in south China. The “Panlongcen” is used as a folk medicine for the treatment of bacterial and inflammatory diseases, such as cancer, blood and chest disorders.^{1,2} Previous investigations of *Spiranthes australis* (R. BROWN) LINDL. have yielded flavanoids,^{2,3} homocyclotriucallanes and dihydrophenanthrenes.^{4,5}

In a previous paper, we had described the characterization of a new flavonoid compound from *Spiranthes australis* (R. BROWN) LINDL.³ Our continuing chemical studies on this plant have resulted in the isolation of three new flavonoid compounds, 5-hydroxy-4'-[(2-isopentenyl)oxy]-3,7,3'-trimethoxyflavone (1), 3-O-[[O-β-D-xylopyranosyl]-(1→2)-β-D-glucopyranosyl]-8-(p-hydroxy-benzyl)-kaempferol (2) and 3-O-[[O-2-O-(E)-p-coumaroyl-β-D-xylopyranosyl]-(1→2)-β-D-glucopyranosyl]-8-(p-hydroxy-benzyl)-kaempferol (3), together with six known flavonoid compounds, 3-O-[[β-D-xylopyranosyl]-(1→2)-β-D-glucopyranosyl]-kaempferol (4),⁶ 7-O-β-D-glucopyranosyl kaempferol (5),⁷ paeonoside (6),⁸ quercimeritrin (7),⁹ retusine (8),¹⁰ 3,5-dihydroxy-4',7-dimethoxy flavone (9).¹¹ This paper reports the isolation and the structural elucidation of three new flavonoid compounds 1–3.

Results and Discussion

The whole plant of *Spiranthes australis* (R. BROWN) LINDL. was extracted with 70% EtOH and the EtOH extract was then concentrated. The residue was dissolved in water and partitioned with petroleum ether, CHCl₃, EtOAc, and *n*-BuOH, successively. The *n*-BuOH extract was then chromatographed by silica gel, Sephadex LH-20, HPLC to afford compounds 1–3.

Compound 1 was obtained as yellow needles, whose molecular formula was deduced as C₂₃H₂₄O₇ by high resolution ESI-MS (HR-ESI-MS) (*m/z* 413.1588 [M+H]⁺). In the ¹H-NMR spectrum of compound 1 (Table 1), a phenol proton appeared at δ_H 12.66 (1H, s), which was characteristic of 5-OH. The signals at δ_H 7.72 (1H, dd, *J*=8.2, 2.0 Hz), 7.70 (1H, d, *J*=2.0 Hz), 7.00 (1H, d, *J*=8.2 Hz), showed an ABX spin coupling system, indicating the 3'- and 4'-substitute structure of B-ring. The signals at δ_H 6.46 (1H, d, *J*=2.2 Hz) and 6.37 (1H, d, *J*=2.2 Hz) could be assigned to 8- and 6-protons of A-ring. In addition, an olefinic proton signal at δ_H 5.55 (1H, t, *J*=6.5 Hz), –CH₂O– signal at δ_H 4.69 (2H, d, *J*=6.5 Hz), two methyl signals at δ_H 1.81 (3H, s) and 1.78 (3H, s) were also noted in the ¹H-NMR spectrum, together with the signals of δ_C 138.3, 119.4, 65.8, 25.8 and 18.3 in the ¹³C-NMR spectrum, an isopentenyl group can be deduced in the structure. Moreover, the signals at δ_H 3.96 (3H, s), 3.89 (3H, s), 3.87 (3H, s) in ¹H-NMR spectrum and δ_C 60.2, 56.1, 55.8 in ¹³C-NMR spectrum suggested three methoxy groups

Table 1. The ¹H- and ¹³C-NMR Data of Compound 1 in CDCl₃^{a)}

Position	δ _H	δ _C	Position	δ _H	δ _C
2	—	155.9 (s)	4'	—	150.8 (s)
3	—	139.0 (s)	5'	7.00 (1H, d, <i>J</i> =8.2)	112.4 (d)
4	—	178.8 (s)	6'	7.72 (1H, dd, <i>J</i> =8.2, 2.0)	122.1 (d)
5	—	162.1 (s)	1''	4.69 (2H, d, <i>J</i> =6.5)	65.8 (t)
6	6.37 (1H, d, <i>J</i> =2.2)	97.8 (d)	2''	5.55 (1H, t, <i>J</i> =6.5)	119.4 (d)
7	—	165.5 (s)	3''	—	138.3 (s)
8	6.46 (1H, d, <i>J</i> =2.2)	92.2 (d)	4''	1.78 (3H, s)	18.3 (q)
9	—	156.8 (s)	5''	1.81 (3H, s)	25.8 (q)
10	—	106.1 (s)	5-OH	12.7 (1H, s)	—
1'	—	122.8 (s)	3-CH ₃ O	3.87 (3H, s)	60.2 (q)
2'	7.70 (1H, d, <i>J</i> =2.0)	111.6 (d)	7-CH ₃ O	3.89 (3H, s)	56.1 (q)
3'	—	149.2 (s)	4'-CH ₃ O	3.96 (3H, s)	55.8 (q)

a) δ in ppm and *J* in Hz. Run at 300 MHz for ¹H-NMR and 75 MHz for ¹³C-NMR.

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in compound **1**. In the heteronuclear multiple bond correlation (HMBC) experiment, long-range correlation from the signal δ_{H} 4.69 (H-1'') to δ_{C} 150.8 (C-4') placed the isopentenyl group to C-4'. Meanwhile, C-5 hydroxyl group was detected from the HMBC cross-peaks between 5-OH with δ_{C} 97.8 (C-6), 106.1 (C-10) and 162.1 (C-5). On the basis of correlations between δ_{H} 3.89 (MeO-7) with δ_{C} 55.8 (C-7) in the heteronuclear multiple quantum coherence (HMQC) analysis and cross peaks between δ_{H} 3.89 (MeO-7) with δ_{C}

165.5 (C-7), δ_{H} 6.46 (H-8), 6.37 (H-6) with δ_{C} 165.5 (C-7) in the HMBC spectrum, a methoxy group at C-7 position can be deduced. In addition, correlation signals based on the combined analysis of the HMQC and HMBC spectra provided good support for a C-3 methoxy group. A further nuclear Overhauser enhancement and exchange spectroscopy (NOESY) experiment was also carried out, since there was no correlation between H-5' with another methoxy group, this methoxy group should locate at C-3' position. Thus, the structure of compound **1** was established to be 5-hydroxy-4'-[(2-isopentenyl)oxy]-3,7,3'-trimethoxyflavone (Fig. 1).

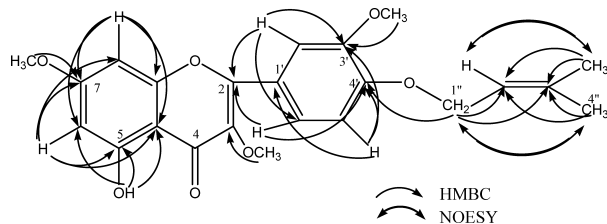


Fig. 1. Structure, Key HMBC and NOESY Correlations of Compound **1**

Compound **2**, isolated as yellow amorphous powder and gave positive reactions with FeCl_3 and HCl-Mg reagents. Its HR-ESI-MS showed a molecular-ion peak at m/z 709.1736 ($[\text{M}+\text{Na}]^+$) indicating the molecular formula $\text{C}_{33}\text{H}_{34}\text{O}_{16}$. Acid hydrolysis of **2** gave D-glucose and D-xylose identified by high-performance TLC (HP-TLC) and comparison with authentic samples. No methoxy signal was observed in the ^1H - and ^{13}C -NMR spectra (Table 2), indicating the hydroxyl

Table 2. The ^1H - and ^{13}C -NMR Data of Compounds **2** and **3** in $\text{DMSO-}d_6^a$

Position	2		3	
	δ_{H}	δ_{C}	δ_{H}	δ_{C}
2	—	155.1 (s)	—	154.9 (s)
3	—	132.8 (s)	—	133.0 (s)
4	—	177.7 (s)	—	177.8 (s)
5	—	159.2 (s)	—	159.2 (s)
6	6.34 (1H, s)	98.4 (d)	6.33 (1H, s)	98.3 (d)
7	—	161.9 (s)	—	161.7 (s)
8	—	106.1 (s)	—	106.0 (s)
9	—	153.7 (s)	—	153.5 (s)
10	—	104.0 (s)	—	104.0 (s)
1'	—	121.2 (s)	—	121.2 (s)
2', 6'	8.00 (2H, d, $J=8.8$)	130.9 (d)	7.98 (2H, d, $J=8.5$)	130.9 (d)
3', 5'	6.87 (2H, d, $J=8.8$)	115.3 (d)	6.86 (2H, d, $J=8.5$)	115.2 (d)
4'	—	160.0 (s)	—	160.0 (s)
1''	—	130.7 (s)	—	130.7 (s)
2'', 6''	7.00 (2H, d, $J=8.3$)	128.9 (d)	6.99 (2H, d, $J=8.0$)	128.9 (d)
3'', 5''	6.60 (2H, d, $J=8.3$)	115.1 (d)	6.60 (2H, d, $J=8.0$)	115.1 (d)
4''	—	155.5 (s)	—	155.5 (s)
7''	3.95 (2H, s)	27.0 (t)	3.94 (2H, s)	26.9 (t)
5-OH	12.6 (1H, s)	—	12.6 (1H, s)	—
Glc				
1	5.76 (1H, d, $J=7.2$)	98.0 (d)	5.74 (1H, d, $J=7.5$)	97.7 (d)
2	3.45 (1H, m)	81.9 (d)	—	79.3 (d)
3	3.10 (1H, m)	77.6 (d)	—	77.0 (d)
4	3.29 (1H, m)	69.5 (d)	—	69.8 (d)
5	3.48 (1H, m)	76.9 (d)	—	76.9 (d)
6	3.53 (1H, m) 3.29 (1H, m)	60.5 (t)	—	60.5 (t)
Xyl				
1	4.59 (1H, d, $J=7.1$)	104.7 (d)	4.99 (1H, d, $J=7.6$)	100.9 (d)
2	3.06 (1H, m)	74.0 (d)	4.66 (1H, dd, $J=7.6, 8.0$)	73.6 (d)
3	3.09 (1H, m)	76.2 (d)	—	74.3 (d)
4	3.09 (1H, m)	69.5 (d)	—	70.4 (d)
5	3.70 (1H, m)	65.8 (t)	—	65.8 (t)
6	3.06 (1H, m)	—	—	—
(<i>E</i>)-Coumaroyl				
1'''	—	—	—	125.4 (s)
2''', 6'''	—	—	7.50 (2H, d, $J=8.4$)	130.2 (d)
3''', 5'''	—	—	6.79 (2H, d, $J=8.4$)	115.8 (d)
4'''	—	—	—	159.7 (s)
7'''	—	—	7.55 (1H, d, $J=16.4$)	144.5 (d)
8'''	—	—	6.34 (1H, d, $J=16.4$)	115.2 (d)
9'''	—	—	—	165.9 (s)

a) δ in ppm and J in Hz. Run at 300 MHz for ^1H -NMR and 75 MHz for ^{13}C -NMR.

substitute structure of compound **2**. The $^1\text{H-NMR}$ spectrum of **2** showed resonances for two AA'BB' spin coupling systems at δ_{H} 8.00 (2H, d, $J=8.8$ Hz), 6.87 (2H, d, $J=8.8$ Hz) and 7.00 (2H, d, $J=8.3$ Hz), 6.60 (2H, d, $J=8.3$ Hz), which suggested two *p*-substitute benzenes in the structure. The phenol proton δ_{H} 12.63 (HO-5) showed correlations with δ_{C} 159.2 (C-5), 104.0 (C-10), 98.4 (C-6) in HMBC experiment, since the signal of δ_{C} 98.4 (C-6) correlated with the signal of δ_{H} 6.34 (H-6) in HMQC spectrum, which also correlated with δ_{C} 161.9 (C-7), 159.2 (C-5), 106.1 (C-8), 104.0 (C-10) in HMBC experiment, a 5,7-dihydroxy structure of A-ring could be deduced. The signal of δ_{H} 3.95 (H-7'') showed correlation with δ_{C} 27.0 (C-7'') in HMQC analysis, which also correlated with δ_{C} 130.7 (C-1''), 128.9 (C-2'', 6''), 106.1 (C-8) in HMBC experiment, the signal δ_{H} of 7.00 (H-2'', 6'') correlated with δ_{C} 128.9 (C-2'', 6'') in HMQC and showed correlations with δ_{C} 155.5 (C-4''), 130.7 (C-1''), 128.9 (C-2'', 6''), 27.0 (C-7'') in HMBC experiment. On the basis of above evidence, a *p*-hydroxy benzyl group was attributed to C-8 position of A-ring. Thus the other *p*-substitute benzene fragment could be assigned to the B-ring of compound **2**, and the 4'-hydroxy substitute structure of B-ring was then elucidated by the information of $^1\text{H-NMR}$, $^{13}\text{C-NMR}$, HMQC and HMBC spectra. Moreover, in the HMQC experiment, the anomeric protons at δ_{H} 5.76 (1H, d, $J=7.1$ Hz) and 4.59 (1H, d, $J=7.1$ Hz) were correlated with carbon signals at δ_{C} 98.0 and 104.7, respectively, which indicated two sugar units in the structure, combining with the acid hydrolysis result and their carbon shifts, the two sugar units were identified as β -D-glucopyranoside and β -D-xylopyranoside. In the HMBC spectrum, the glucose anomeric proton correlated with δ_{C} 132.8 (C-3), the xylose anomeric proton correlated with δ_{C} 81.9 (Glc C-2), which subsequently determined the sugar sequences of oligosaccharide chains. On the basis of aforementioned spectroscopic data, compound **2** was elucidated as 3-*O*-[*O*- β -D-xylopyranosyl]-(1 \rightarrow 2)- β -D-glucopyranosyl]-8-(*p*-hydroxy-benzyl)-kaempferol (Fig. 2).

Compound **3** was obtained as yellow amorphous powder, whose molecular formula was determined as $\text{C}_{42}\text{H}_{40}\text{O}_{18}$ by the quasimolecular ion peak at m/z 855.2110 ($[\text{M}+\text{Na}]^+$) in HR-ESI-MS spectrum. The acid hydrolysis of compound **3** also gave D-glucose and D-xylose identified by HP-TLC and comparison with authentic samples. Meanwhile, from the CHCl_3 extract of acid hydrolysis solution, the same aglycone with compound **2** was harvested. The alkaline hydrolysis test was further performed, which gave compound **2** and *p*-coumaric acid established by HP-TLC, these indicated a similar structure of **3** with **2** and the *p*-coumaroyl group should be located at oligosaccharide chains. This result could also be inferred by comparing the $^1\text{H-}$ and $^{13}\text{C-NMR}$ data (Table 2) of compound **2** with **3**. In the $^1\text{H-}$ and $^{13}\text{C-NMR}$ spectra of **3**, the signals at δ_{H} 7.50 (2H, d, $J=8.4$ Hz), 6.79 (2H, d, $J=8.4$ Hz), 7.55 (1H, d, $J=16.4$ Hz), 6.34 (1H, d, $J=16.4$ Hz) and δ_{C} 125.4, 130.2, 115.8, 159.7, 144.5, 115.2, 165.9 indicated a *p*-coumaroyl substituent in the structure. In the HMBC experiment of compound **3**, the correlation peak was observed between the signals of δ_{H} 4.66 (Xyl H-2) with δ_{C} 165.9 (C-9''), from which the *p*-coumaroyl group could be placed to C-2 position of the xylose. Thus the structure of compound **3** was determined as 3-*O*-[*O*-[*O*-[2-*O*-(*E*)-*p*-coumaroyl]- β -D-xylopyranosyl]-(1 \rightarrow 2)- β -D-glucopyranosyl]-

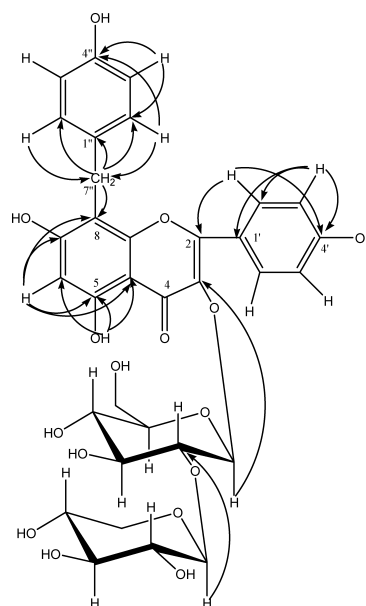


Fig. 2. Structure and Key HMBC Correlations of Compound **2**

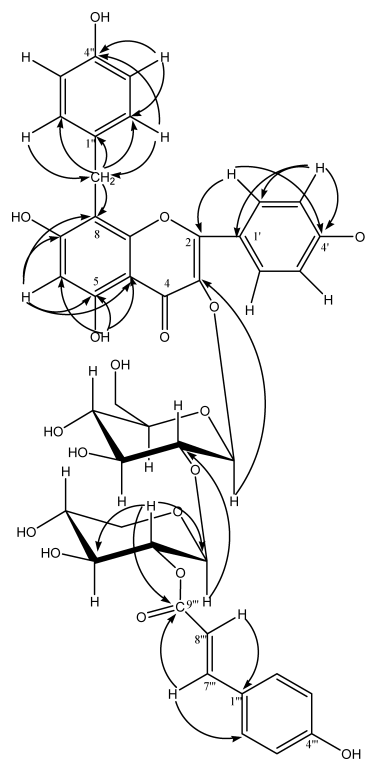


Fig. 3. Structure and Key HMBC Correlations of Compound **3**

8-(*p*-hydroxy-benzyl)-kaempferol (Fig. 3).

The six known compounds 3-*O*-[β -D-xylopyranosyl-(1 \rightarrow 2)- β -D-glucopyranosyl]kaempferol (**4**), 7-*O*- β -D-glucopyranosyl kaempferol (**5**), paeonoside (**6**), quercimeritrin (**7**), retusine (**8**), 3,5-dihydroxy-4',7-dimethoxyflavone (**9**), were also isolated and identified by comparing their NMR and TLC data with authentic samples.

The cytotoxic activity of compounds **1**—**3** against human A549, Hep G2 and HT-29 cell lines were also examined, but there was no cytotoxic effect detected.

Experimental

General Experiment Procedures TLC was conducted on silica gel plates (60F₂₅₄, Merck). HPLC was performed on Waters-600 prep. HPLC instrument equipped with a Amersham Pharmacia Biotech-ODS (250×20 mm) column. Column chromatography (CC) was performed on silica gel (200–300 mesh; Qingdao Haiyang, Co., China), or Sephadex LH-20 (Pharmacia Biotech AB, S-Uppsala). Melting point was measured on a Yanaco MP-S3 micromelting point apparatus. Optical rotations were determined on a Perkin-Elmer 341 polarimeter. UV spectra were obtained on a UV-1201 Shimadzu spectrometer. IR spectra were recorded on a Perkin-Elmer Spectrum GX spectrometer. NMR spectra were obtained on a Bruker ARX-300 instrument. Chemical shifts are expressed as δ (ppm) with trimethylsilane (TMS) as the internal standard. ESI- and HR-ESI-MS were recorded on Bruker APEX II FT-ICR mass spectrometers.

Plant Material Dried whole plants of *Spiranthes australis* (R. BROWN) LINDL. were purchased from the General Corporation of Medicinal Materials of Jiang-Su Province, P. R. China in March 2003. A voucher specimen (SAD20030310) was identified by Professor Y.-Z. Guo and has been deposited in the Department of Chinese Traditional Medicine Analysis, Shenyang Pharmaceutical University.

Extraction and Isolation The whole plants of *Spiranthes australis* (R. BROWN) LINDL. (15 kg) was cut into small pieces and extracted successively with 70% EtOH under reflux to give an EtOH extract (800 g). The extract was concentrated *in vacuo* then suspended in 4 l of H₂O and partitioned successively with 4 l of petroleum ether, 4 l of EtOAc, and 4 l of *n*-BuOH. The *n*-BuOH extract (80 g) was adsorbed on silica gel (100 g) and subjected to CC (silica gel (2000 g) 9×100 cm column; CHCl₃/MeOH gradient) to yield 9 fractions (1–9). Fraction 1 was applied to CC (Sephadex LH-20 (55 g), 3×100 cm column using 55% MeOH). The collected fractions were purified by HPLC (MeOH/H₂O 75:25): **1** (32.5 mg). Fraction 2 was subjected to Sephadex LH-20 chromatography (60 g, 3×100 cm column using 55% MeOH): **8** (12.5 mg) and **9** (10.1 mg). Fraction 4 was also subjected to Sephadex LH-20 chromatography (50 g, 3×100 cm column using 55% MeOH): **5** (21.1 mg) and **7** (18.8 mg). Fraction 7 was applied to CC (Sephadex LH-20 (60 g), 3×100 cm column using 60% MeOH). The collected fractions were purified by HPLC (MeOH/H₂O 65:35): **2** (69.8 mg) and **3** (50.1 mg). Fraction 8 was applied to CC (Sephadex LH-20 (55 g), 3×100 cm column using 50% and 65% MeOH): **4** (18.1 mg) and **6** (21.4 mg).

5-Hydroxy-4'-[(2-isopentenyl)oxy]-3,7,3'-trimethoxyflavone (**1**): Yellow needles (MeOH), melting point (mp) 120–121 °C. UV λ_{\max} (MeOH) nm (log ϵ): 254 (4.40), 354 (3.97); IR (KBr) cm⁻¹: 3430, 1668, 1601, 1495; ¹H-NMR (300 MHz) and ¹³C-NMR (75 MHz) data see Table 1; ESI-MS (pos.) *m/z*: 413 (100, [M+H]⁺); HR-ESI-MS: 413.1588 ([M+H]⁺, C₂₃H₂₅O₇⁺; Calcd 413.1594).

3-*O*-{[*O*- β -D-xylopyranosyl]-(1→2)- β -D-glucopyranosyl}-8-(*p*-hydroxybenzyl)-kaempferol (**2**): Yellow amorphous powder; [α]_D²⁴ -40.5° (*c*=0.16, MeOH); UV λ_{\max} (MeOH) nm (log ϵ): 225 (4.42), 271 (4.10); IR (KBr) cm⁻¹: 3388, 1653, 1611, 1511; ¹H-NMR (300 MHz) and ¹³C-NMR (75 MHz) data see Table 2; ESI-MS (pos.; neg.) *m/z*: 709 (100, [M+Na]⁺); 685 (65, [M-H]⁻); HR-ESI-MS: 709.1736 ([M+Na]⁺, C₃₃H₃₄NaO₁₆⁺;

Calcd 709.1739).

3-*O*-{*O*-[2-*O*-(*E*)-*p*-Coumaroyl- β -D-xylopyranosyl]-(1→2)- β -D-glucopyranosyl}-8-(*p*-hydroxybenzyl)-kaempferol (**3**): Yellow amorphous powder; [α]_D²⁴ -63.7° (*c*=0.12, MeOH); UV λ_{\max} (MeOH) nm (log ϵ): 231 (4.36), 282 (4.02); IR (KBr) cm⁻¹: 3392, 1662, 1615, 1498; ¹H-NMR (300 MHz) and ¹³C-NMR (75 MHz) data see Table 2; ESI-MS (pos.) *m/z*: 855 (100, [M+Na]⁺); HR-ESI-MS: 855.2110 ([M+Na]⁺, C₄₂H₄₀NaO₁₈⁺; Calcd 855.2106).

Acid Hydrolysis of 2 and 3 Compounds **2** and **3** (8 mg each) were refluxed with 15% HCl/MeOH (10 ml) at 80 °C for 4 h. After cooling, the mixture was concentrated and the residue partitioned with CHCl₃/H₂O. The presence of glucose and xylose in this mixture was established by comparison with authentic samples. The HP-TLC in the solvent system MeCOEt/^{*i*}PrOH/Me₂CO/H₂O (20:10:7:6) resulted in the *R*_f 0.26 and 0.39 respectively. The aglycone of compound **3** was identified as same to compound **2** by polyamide TLC (CHCl₃/MeOH 15:1) with the *R*_f 0.43.

Alkaline Hydrolysis of 3 Compound **3** was dissolved in 20 ml solution (MeOH/H₂O, 1:1) containing appropriate NaOH and hydrolyzed under reflux (2 h) at 60 °C. Then the basic solution was evaporated to dryness. The HP-TLC of hydrolysis product in the solvent system CHCl₃/MeOH (2:1) resulted in the *R*_f 0.31 and 0.40 respectively. Compound **2** and *p*-coumaric acid were identified by comparison with compound **2** and authentic sample.

Cytotoxic Activity Compounds **1–3** were examined for their *in vitro* cytotoxic activity against human cell lines A549, Hep G2 and HT-29. The inhibition of growth was determined by *in vitro* treatment of the respective cell lines using MTT assay.¹²⁾

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