Lupane Glycosides from the Leaves of Acanthopanax koreanum

Hyun-Sik Choi,^{*a*} Hyoung Ja Kim,^{*b*} Sang-Gil NAM,^{*a*} In-Seon Kim,^{*a*} Kyung-Tae Lee,^{*a*} Chang-Soo Yook,^{*a*} and Yong Sup Lee^{*,*a*}

^a Kyung Hee East-West Pharmaceutical Research Institute, College of Pharmacy and Department of Life and Nanopharmaceutical Science, Kyung Hee University; 1 Hoegi-dong, Dongdaemun-ku, Seoul 130–701, Korea: and ^bLife Sciences Division, Korea Institute of Science and Technology; P. O. Box 131, Cheongryang, Seoul 130–650, Korea. Received July 11, 2008; accepted August 26, 2008

Three new lupane-type saponins, acankoreosides F—H (1—3) were isolated from the methanol extract of the leaves of Acanthopanax koreanum NAKAI. The structures of these three saponins were established by chemical and spectroscopic analysis as 3α , 30-dihydroxylup-20(29)-en-23, 28-dioic acid 28-O-[α -L-rhamnopyranosyl-(1 \rightarrow 4)- β -D-glucopyranosyl-(1 \rightarrow 6)- β -D-glucopyranosyl] ester (1), 3α , 30-dihydroxylup-23-al-20(29)-en-28-oic acid 28-O-[α -L-rhamnopyranosyl-(1 \rightarrow 4)- β -D-glucopyranosyl-(1 \rightarrow 6)- β -D-glucopyranosyl] ester (2), and (20S) 3α -hydroxy-lup-23-al-28, 29-dioic acid 28-O-[α -L-rhamnopyranosyl-(1 \rightarrow 6)- β -D-glucopyranosyl-(1 \rightarrow 6)- β -D-glucopyranosyl-($1\rightarrow$ 6)- β -D-glucopyranosyl-(

Key words Acanthopanax koreanum; Araliaceae; lupane glycoside; acankoreoside

Acanthopanax koreanum is a shrub that belongs to the Araliaceae family and is found in Northeast Asia. Acanthopanax species, such as, A. senticosus, A. gracilistylus, A. obovatus, and A. giraldii have been used to treat rheumatism, allergies, and diabetes,¹⁻⁴⁾ and the root and stem bark of A. koreanum are used as a tonic, as a prophylactic, and to treat rheumatism, paralysis, hepatitis, and diabetes by practitioners of oriental medicine.^{5,6)} The biological activities of lignans and diterpenes from the roots and stems of A. koreanum have been well established.⁷⁻¹¹⁾ In addition, our group has previously reported on lupane-triterpene glycosides, such as, acankoreosides A—E, obtained from the leaves of A. koreanum.¹²⁻¹⁴⁾

During our on-going investigation of the bioactive principles of *A. koreanum*, we isolated and determined the structures of three lupane-type saponins, 3α , 30-dihydroxylup-20(29)-en-23, 28-dioic acid 28-O- $[\alpha$ -L-rhamnopyranosyl- $(1\rightarrow 4)$ - β -D-glucopyranosyl- $(1\rightarrow 6)$ - β -D-glucopyranosyl] ester (1), 3α , 30-dihydroxylup-23-al-20(29)-en-28-oic acid 28-O- $[\alpha$ -L-rhamnopyranosyl- $(1\rightarrow 4)$ - β -D-glucopyranosyl- $(1\rightarrow 6)$ - β -D-glucopyranosyl] ester (2), and (20S) 3α -hydroxylup-23-al-28, 29-dioic acid 28-O- $[\alpha$ -L-rhamnopyranosyl- $(1\rightarrow 4)$ - β -D-glucopyranosyl- $(1\rightarrow 6)$ - β -D-glucopy

Results and Discussion

Compound 1 was obtained as a white amorphous powder of molecular formula $C_{48}H_{75}O_{20}$, as determined by HR-FAB-MS (molecular ion at m/z 971.4863 attributed to $[M-H]^-$). The structure of 1 was established by comparing its chemical shifts to those of acankoreoside A 4 in ¹H-NMR spectra,¹² 1 differed from acankoreoside A due to a hydroxylated methylene group at δ 4.42 and 4.47 (each d, J=15.3 Hz) rather than the secondary methyl group at C-30 in acankoreoside A. The position of this methylene group (C-30) was elucidated due to correlations in the HMBC spectrum between C-20 at δ 156.5 and H-18 (δ 1.91), H-19 (δ 3.33), H-29 (δ 5.08) and H-30 (δ 4.42, 4.47) of 1. The ¹H-NMR spectrum of 1

* To whom correspondence should be addressed. e-mail: kyslee@khu.ac.kr

showed signals due to four tertiary methyl groups at δ 0.89, 0.92, 1.17 and 1.44 (each 3H, s), one secondary methyl group at δ 1.69 (3H, d, J=6.0 Hz), which was assigned to H-6 of rhamnose, three anomeric protons due to two hexosyl residues at δ 6.30 (1H, d, J=8.0 Hz) and 4.96 (1H, d, J=7.9 Hz), and one 6-deoxyhexosyl residue at δ 5.82 (1H, brs); as detailed in Tables 1 and 2. Furthermore, the ¹H-NMR spectrum of 1 showed characteristic signals for H-19 at δ 3.33 (1H, dt, J=3.6, 11.5 Hz), H-13 at δ 2.61 (1H, m), H-5 at δ 2.48 (1H, br d, J=11.0 Hz), and H-3 at δ 4.21 (1H, overlapped), which suggested an aglycone with a 3α -hydroxylupan-28-oic acid skeleton. The ¹³C-NMR spectrum of 1 showed signals at δ 175.1 and 181.2, indicating the presence of an ester carboxyl group at C-28 and a free carboxylic acid at C-23 on the triterpene moiety. The assignment of the α hydroxyl group at C-3 was performed by comparing spectral data to literature values.^{14–16)} Observed chemical shifts of C-1 (δ 33.2), C-5 (δ 45.4) and C-24 (δ 18.2) in the aglycone of 1 confirmed the axial α -position of the 3-hydroxyl group by comparing the corresponding signals of the 3β -epimer derivative [δ value for 39.4 (C-1), 52.4 (C-5) and 11.3 (C-24)].¹⁶

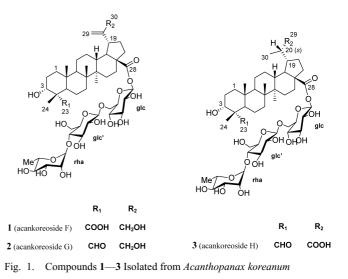


Table 1. NMR Data of Acankoreosides F—H (1—3) (125, 500 MHz, Pyridine- d_5)

No.	1		2		3	
	$\delta_{ m H}$ (mult., Hz)	$\delta_{ m c}$	$\delta_{ m H}$ (mult., Hz)	$\delta_{ m c}$	$\delta_{ m H}$ (mult., Hz)	$\delta_{ m c}$
1	1.44, 1.81 (m)	33.2	1.45, 1.94 (m)	33.1	1.49, 1.80 (m)	33.1
2	1.93, 1.98 (m)	26.1	1.75, 1.94 (m)	26.7	1.76, 1.93 (m)	26.7
3	4.21 (overlapped)	72.8	4.01 (br s)	73.0	4.02 (br s)	73.0
4		51.8		52.5		52.5
5	2.48 (br d, 11.0)	45.4	2.40 (br d, 12.1)	44.0	2.42 (br d, 13.3)	44.0
6	1.58, 1.81 (m)	21.7	1.16, 1.45 (m)	20.9	1.10, 2.15 (m)	21.1
7	1.31, 1.69 (m)	34.5	1.23, 1.57 (m)	34.1	1.25, 1.59 (m)	34.1
8		41.8		41.8		41.8
9	1.61 (m)	51.0	1.57 (m)	50.6	1.66 (m)	50.2
10		37.4		36.9		36.9
11	1.31, 1.45 (m)	21.0	1.07, 1.33 (m)	21.0	1.19, 1.48 (m)	20.8
12	1.17, 1.68 (m)	27.1	1.16, 1.33 (m)	27.0	1.21, 1.86 (m)	26.9
13	2.61 (m)	38.3	2.64 (m)	38.3	2.70 (m)	38.2
14		42.8		42.8		43.0
15	1.17, 2.01 (m)	30.2	1.16, 1.98 (m)	30.1	1.18, 2.01 (m)	30.0
16	1.44, 2.61 (m)	32.2	1.45, 2.64 (m)	32.1	1.46, 2.65 (m)	32.1
17		57.0		56.9		57.3
18	1.91 (m)	50.2	1.95 (m)	50.2	1.55 (t, 11.0)	48.9
19	3.33 (dt, 3.6, 11.5)	43.2	3.34 (dt, 4.0, 11.5)	43.2	3.53 (m)	40.6
20		156.5		156.5	2.98 (dq, 2.7, 6.9)	42.1
21	1.58, 2.30 (m)	32.7	1.57, 2.30 (m)	32.7	1.88, 2.11 (m)	25.0
22	1.58, 2.19 (m)	36.8	1.57, 2.18 (m)	36.7	1.41, 2.19 (m)	37.3
23		181.2	9.93 (s)	209.8	10.0 (s)	209.9
24	1.44 (s)	18.2	1.09 (s)	14.6	1.10 (s)	14.6
25	0.92 (s)	16.8	0.85 (s)	16.4	0.87 (s)	16.3
26	1.17 (s)	16.7	1.14 (s)	16.5	1.16 (s)	16.5
27	0.89 (s)	14.9	0.94 (s)	14.8	0.94 (s)	14.9
28	× *	175.1	· ·	175.0		174.9
29	5.08, 5.49 (br s)	106.1	5.10, 5.50 (br s)	106.1		180.0
30	4.42, 4.47 (each d, 15.3)	64.3	4.42, 4.47 (each d, 15.3)	64.3	1.28 (d, 6.9)	10.0

Table 2. The Sugar Moieties of Acankoreosides F—H (1—3) (125, 500 MHz, Pyridine-d₅)

No.	1		2		3	
	$\delta_{ m H}$ (mult., Hz)	$\delta_{ m c}$	$\delta_{ m H}$ (mult., Hz)	$\delta_{ m C}$	$\delta_{ m H}$ (mult., Hz)	$\delta_{ m c}$
28-glc						
1	6.30 (d, 8.0)	95.2	6.32 (d, 8.2)	95.2	6.33 (d, 8.0)	95.2
2	4.10 (dd, 8.2, 8.5)	74.0	4.10 (br t, 8.9)	74.0	4.08 (br t, 8.6) ^{<i>a</i>})	74.0
3	4.23 (dd, 8.1, 9.0)	78.6	4.22 (t, 9.0)	78.6	4.23 (t, 9.0)	78.6
4	4.29 (t, 9.2)	70.8	4.31 (t, 9.3)	70.7	4.38 (t, 9.5)	70.5
5	4.08^{a}	78.1	4.08 ^{<i>a</i>})	78.1	$4.07^{a)}$	77.8
6	4.28 (dd, 5.3, 12.7)	69.3	$4.29^{a)}$	69.3	4.28 (dd, 4.3, 11.2)	69.3
	4.67 ^{<i>a</i>})		4.67 ^{<i>a</i>})		4.61 (dd, 2.0, 11.4)	
6-glc						
1'	4.96 (d, 7.9)	104.9	4.95 (d, 7.6)	104.9	4.90 (d, 7.8)	105.1
2'	3.93 (dd, 8.2, 8.6)	75.3	3.93 (br t, 8.5)	75.3	3.92 (dd, 8.0, 8.7)	75.2
3'	4.14 (t, 9.0)	76.4	4.14 (t, 9.0)	76.4	4.10 (t, 9.0)	76.4
4'	4.38 (t, 9.3)	78.3	4.38 (t, 9.3)	78.3	4.38 (t, 9.1)	78.2
5'	3.63 (m)	77.1	3.62 (m)	77.1	3.60 (m)	77.1
6'	4.08 (dd, 2.6, 11.4)	61.3	4.08 ^{<i>a</i>})	61.3	4.05 (dd, 3.5, 12.1)	61.2
	4.18 ^{<i>a</i>})		4.19^{a}		4.18 (dd, 1.7, 12.5)	
rha						
1	5.82 (br s)	102.7	5.82 (br s)	102.7	5.81 (br s)	102.6
2	4.68 (br s)	72.5	4.67 (dd, 1.4, 3.0)	72.5	4.67 (dd, 1.5, 3.2)	72.5
3	4.55 (dd, 3.3, 9.2)	72.7	4.53 (dd, 3.3, 9.2)	72.7	4.55 (dd, 3.3, 9.3)	72.6
4	4.34 (t, 9.4)	73.9	4.33 (t, 9.4)	73.9	4.34 (t, 9.6)	73.9
5	4.95 (dq, 6.3, 10.2)	70.3	4.94 (dq, 6.2, 9.5)	70.3	4.93 (m)	70.3
6	1.69 (d, 6.0)	18.5	1.69 (d, 6.2)	18.5	1.69 (d, 6.2)	18.5

a) Overlapping signals.

In the NOESY spectrum, the presence of cross-peaks between H-24 and H-25 as well as H-3 indicated that the methyl group (H-24) was axial, which in turn suggested that the carboxyl group at C-4 was α -positioned. In the HMBC spectrum, the carbonyl carbon signal at δ 175.1 (C-28) showed a ${}^{1}H{-}{}^{13}C$ long range correlation with signals at δ 1.91 (H-18), δ 1.58, and 2.19 (H-22). The other carbonyl signal at δ 181.2 (C-23) was correlated with signals at δ 1.44 (H-24) and 2.48 (H-5). These data indicate that the carbonyl groups of 1 are located at C-17 and C-4. In addition, HMBC correlations of 1 were observed between glc H-1' (δ 6.30) and the C-28 of the triterpene moiety (δ 175.1), between glc H-1" (δ 4.96) and glc C-6' (δ 69.3), and between rha H-1" (δ 5.82) and glc C-4" (δ 78.3), which supported glycoside sequence. Two anomeric proton signals at δ 6.30 (d, J=8.0 Hz) and 4.96 (d, J=7.9 Hz) indicated a β -configuration based on coupling constants, and the α -anomeric configuration of rhamnose was assigned based on the 13C-NMR chemical shift of rhamnosyl C-3 (δ 72.7) and C-5 (δ 70.3).¹⁷ Acid hydrolysis of 1 provided the monosaccharide components of Lrhamnose and D-glucose (identified using authentic samples). Accordingly, the structure of 1 was elucidated to be 3α , 30dihydroxylup-20(29)-en-23.28-dioic acid 28-O-[\alpha-L-rhamnopyranosyl- $(1\rightarrow 4)$ - β -D-glucopyranosyl- $(1\rightarrow 6)$ - β -D-glucopyranosyl] ester, which we named acankoreoside F.

Compound 2 was obtained as a white amorphous powder of molecular formula C48H75O19, as determined by HR-FAB-MS (molecular ion m/z 955.4899 attributed to $[M-H]^{-}$). The ¹H- and ¹³C-NMR spectra of **2** indicated structural features similar to those of acankoreoside F (1) except for the presence of an aldehyde group at C-23. In the ¹H-NMR spectrum of 2, a singlet signal at δ 9.99 corresponded to an aldehyde proton and signals at δ 4.42 and 4.47 (each d, J=15.3 Hz) were assigned to hydroxylated methylene protons of H-30. The ¹³C-NMR spectrum of **2** showed 48 signals including 18 sugar moiety signals, four tertiary methyl groups at δ 14.6 (C-24), 16.4 (C-25), 16.5 (C-26) and 14.8 (C-27), olefinic carbons at δ 156.5 (C-20) and 106.1 (C-29), and an aldehyde signal at δ 209.8 (C-23). Two anomeric proton signals at δ 6.32 (d, J=8.2 Hz) and 4.95 (d, J=7.6 Hz) were characteristic of a β -configuration based on their coupling constants, and the α -anomeric configuration of rhamnose was assigned based on the ¹³C-NMR chemical shift of rhamnosyl C-3 (δ 72.7) and C-5 (δ 70.3).¹⁷⁾ In the HMBC spectrum, the aldehyde signal at δ 209.8 was correlated with the proton signal of a methyl group (H-24) at δ 1.09. In addition, HMBC correlations of 2 were observed between glc H-1' (δ 6.32) and the C-28 of the triterpene moiety (δ 175.0), between glc H-1" (δ 4.95) and glc C-6' (δ 69.3), and between rha H-1''' (δ 5.82) and glc C-4" (δ 78.3), which the glycoside sequence. In the NOESY spectrum, the presence of cross-peaks between H-24 and H-25 as well as H-3 indicated that the methyl group (H-24) was axially orientated, which in turn suggested that the carboxyl group at C-4 was α -positioned. According to 2D NMR analysis, the structure of 2 was elucidated to be 3a,30-dihydroxylup-23-al-20(29)-en-28-oic acid 28-O-a-Lrhamnopyranosyl- $(1\rightarrow 4)$ - β -D-glucopyranosyl- $(1\rightarrow 6)$ - β -Dglucopyranosyl ester, which was named acankoreoside G.

Compound **3** was obtained as a white amorphous powder of molecular formula $C_{48}H_{75}O_{20}$, as determined by HR-FAB-MS (molecular ion at m/z 971.4863 attributed to $[M-H]^-$).

The ¹H- and ¹³C-NMR spectra of **3** indicated a structure similar to that of acankoreoside E.14) Its 13C-NMR spectrum exhibited five tertiary methyl groups at δ 14.6 (C-24), 16.3 (C-25), 16.5 (C-26), 14.7 (C-27) and 10.0 (C-30). The C-23 signal (δ 209.9) of **3** was significantly shifted by 5.3 ppm downfield as compared with that of acankoreoside E and the its carboxylic acid (C-29) at δ 180.0 was shifted by 1.8 ppm upfield, indicating the change in the functional groups at C-23 and C-29. Two anomeric proton signals at δ 6.33 (d, J=8.0 Hz) and 4.90 (d, J=7.8 Hz) were assigned to the β configuration based on their coupling constants, and the α anomeric configuration of rhamnose was assigned based on the ¹³C-NMR chemical shifts of rhamnosyl C-3 (δ 72.6) and C-5 (δ 70.3).¹⁷ In the NOESY spectrum, the presence of cross-peaks between H-24 and H-25 as well as H-3 indicated that the methyl group (H-24) was axially orientated, which in turn suggested an α -position for the carboxyl group at C-4. In the HMBC spectrum, the aldehydic carbonyl signal at δ 209.9 (C-23) was correlated with signals at δ 1.10 (s, H-24) and 2.42 (br d, J=13.3 Hz, H-5), and the carbonyl signal at δ 180.0 (C-29) with proton signals at δ 2.98 (dq, J=2.7, 7.0 Hz, C-20) and 1.28 (d, J=6.9 Hz, H-30), indicating that the aldehyde and carboxylic acid groups were present at C-4 and C-20 of the aglycone, respectively. In addition, HMBC correlations of **3** were observed between glc H-1' (δ 6.33) and the C-28 of the triterpene moiety (δ 174.9), between glc H-1" (δ 4.90) and glc C-6' (δ 69.3), and between rha H-1" (δ 5.81) and glc C-4" (δ 78.2), which supported the glycoside sequence. The downfield shift of the signal at δ 10.0 (assigned to C-30) in comparison with that of acankoreoside E $(\delta_{\text{C-30}}$: 7.0) also supported the above assignments. The absolute stereochemistry at C-20 was established using reported data for a similar compound.^{18,19} The S configuration at C-20 was established based on literature values for the chemical shifts of the carbonyl carbon at C-29 in (20S) 3β acetoxylupan-29-oic acid (δ_{C-30} : 9.6) and (20*R*) 3 β -acetoxylupan-29-oic acid (δ_{C-30} : 17.2). The carbon chemical shift difference between C-30 of (20S) and (20R) 3β -acetoxylupan-29-oic acid was large enough to allow the differentiation. The chemical shift of C-30 (δ_{C-30} : 10.0) in **3** was similar to that of (20S) 3β -acetoxylupan-29-oic acid. Thus, compound **3** was found to be (20S) 3α -hydroxylup-23-al-28,29-dioic acid 28-*O*- α -L-rhamnopyranosyl-(1 \rightarrow 4)- β -D-glucopyranosyl- $(1\rightarrow 6)$ - β -D-glucopyranosyl ester, which we named acankoreoside H.

The three isolates (1-3) were assessed for anti-inflammatory activity by examining their effects on nitric oxide (NO) and prostaglandin E₂ (PGE₂) by lipopolysaccharide (LPS)-induced in RAW 264.7 cells.²⁰⁾ L-N⁶-(1-iminoethyl) lysine (L-NIL, 10 μ M) and NS-398 were used as positive NO and PGE₂ production inhibitor controls, respectively. Acankoreoside F (1) at 200 μ M most potently inhibited PGE₂ (59%) and NO (42%) production (data not shown).

Experimental

General Experimental Procedures Optical rotations were recorded on a JASCO D-1010 spectropolarimeter and an Autopol III Automatic polarimeter (Rudolph Research Flanders, NJ, U.S.A.). NMR spectra were measured in pyridine- d_5 on a Varian UI-500 spectrometer and chemical shifts are quoted relative to tetramethylsilane (TMS). HR-FAB-MS spectra were recorded on a JEOL JMS-700 instrument using glycerol as a matrix in negative ion mode. Diaion HP-20 (Mitsubishi Chem. Co.), Sephadex LH-20 (Pharmacia Biotech), silica gel 60 (0.04—0.063 mm, Merck) and LiChroprep RP-18 (40—63 μ m, Merck) were used for open column chromatography. TLC was performed using silica gel 60 F₂₅₄ plate (Merck).

Plant Material The leaves of *A. koreanum* NAKAI were collected from the Medical Plant Garden at Kyung Hee University during October 2003. Emeritus Professor Chang-Soo Yook (Department of Pharmacognosy at Kyung Hee University), one of the authors, identified the species. A voucher specimen (3-002-0067) was deposited at the Museum of Oriental Medicine in Kyung Hee University.

Extraction and Isolation Dried leaves (1.0 kg) were extracted twice with 1.51 of hot MeOH to provide 152.1 g of dried extract, which was chromatographed on a Diaion HP-20 column using H₂O/30%, 50%, 70%, and 90% MeOH mixes. The 70% MeOH fraction was subsequently chromatographed on a silica gel column using CHCl₂-MeOH-H₂O (8:2:0.2 \rightarrow 7:3:0.5) as eluent to give 8 fractions (Frs. 1-8). Fraction 4 (6.46 g) was then subjected to gel filtration chromatography on a Sephadex LH-20 column using MeOH as eluent. The saponin fraction (5.95 g) so obtained was chromatographed on a silica gel column using CHCl3-MeOH-H2O (7:3:0.2) as eluent and then by gradient chromatography on a LiChroprep RP-18 column using $H_2O/50\%$ to 90% MeOH as eluent to yield compounds 2 (acankoreoside G, 63.8 mg, yield 0.006%) and 3 (acankoreoside H, 44 mg, yield 0.004%). Fraction 7 (2.73 g) was subjected to Sephadex LH-20 column chromatography to give two fractions (Fr. 7-1 and Fr. 7-2). Fraction 7-2 (1.08 g) was purified by silica gel using CHCl₃-MeOH-H₂O (8:2:0.2 \rightarrow 6:4:0.5) as an eluent and finally LiChroprep RP-18 column chromatographed using a gradient solvent system (50→90% MeOH) to yield compound 1 (acankoreoside F, 30 mg, yield 0.003%).

Acankoreoside F (1): White powder; $[\alpha]_D^{25} - 43.0 \ (c=0.50, \text{ MeOH})$; ¹Hand ¹³C-NMR data, see Tables 1 and 2; HR-FAB-MS $m/z \ [M-H]^- 971.4863$ (Calcd for C₄₈H₇₅O₂₀ 971.4852).

Acankoreoside G (2): White powder; $[\alpha]_{D}^{25}$ -58.0 (*c*=0.50, MeOH); ¹Hand ¹³C-NMR data, see Tables 1 and 2; HR-FAB-MS *m*/*z* [M-H]⁻ 955.4899 (Calcd for C₄₈H₇₅O₁₉ 955.4903).

Acankoreoside H (3): White powder; $[\alpha]_{D}^{25} - 37.4$ (*c*=0.50, MeOH); ¹Hand ¹³C-NMR data, see Tables 1 and 2; HR-FAB-MS *m*/*z* [M-H]⁻ 971.4863 (Calcd for C₄₈H₇₅O₂₀ 971.4852).

Acid Hydrolysis of 1 Compound 1 (25 mg) was hydrolyzed in 5% H_2SO_4 under reflux for 3 h. After neutralization with NH₄OH followed by extraction with CHCl₃, the aqueous layer was evaporated *in vacuo* to give a residue (12 mg), which was subjected to silica gel column chromatography (CHCl₃-MeOH-H₂O=6:4:1) to yield D-glucose and L-rhamnose. These sugars were compared with authentic samples on TLC and by measuring optical rotations (+52.8, +12.5, respectively).

Acknowledgements We thank the Korean Basic Science Institute at Korea University for providing NMR and HR-FAB-MS data. This research was supported by the Seoul Research and Business Development Program (10524). In addition, we thank the Korean Ministry of Education & Human Resources Development for support through the Post-BK21 program. We dedicate this work to the memory of Professor Sang-Yong Park our valued mentor and co-worker.

References

- Kang J. S., Linh P. T., Cai X. F., Kim H. S., Lee J. J., Kim Y. H., Arch. Pharm. Res., 24, 407–411 (2001).
- Park S. Y., Chang S. Y., Yook C. S., Nohara T., J. Nat. Prod., 63, 1630—1633 (2000).
- Kang H. S., Kim Y. H., Lee C. S., Lee J. J., Choi I., Pyun K. H., Cell Immunol., 170, 212–221 (1996).
- Hibasami H., Fujikawa T., Takeda H., Nishibe S., Satoh T., Fujisawa T., Nakashima K., Oncol. Rep., 7, 1213—1216 (2000).
- 5) Perry L. M., Metzger J., "Medicinal Plants of East and Southeast Asia," MIT Press, Cambridge, 1980, p. 41.
- Bae K., "The Medicinal Plants of Korea," Kyo-Hak Publishing, Seoul, 2000, p. 361.
- Cai X. F., Shen G., Dat N. T., Kang O. H., Kim J. A., Lee Y. M., Lee J. J., Kim Y. H., *Chem. Pharm. Bull.*, **51**, 605–607 (2003).
- Cai X. F., Shen G., Dat N. T., Kang O. H., Lee Y. M., Lee J. J., Kim Y. H., Arch. Pharm. Res., 26, 731–734 (2003).
- Cai X. F., Lee I. S., Dat N. T., Shen G., Kang J. S., Kim D. H., Kim Y. H., Arch. Pharm. Res., 27, 738—741 (2004).
- 10) Cai X. F., Lee I. S., Shen G., Dat N. T., Lee J. J., Kim Y. H., Arch. Pharm. Res., 27, 825–828 (2004).
- 11) Cai X. F., Lee I. S., Dat N. T., Shen G., Kim Y. H., *Phytother. Res.*, **18**, 677–680 (2004).
- 12) Chang S. Y., Yook C. S., Nohara T., *Chem. Pharm. Bull.*, **46**, 163–165 (1998).
- Chang S. Y., Yook C. S., Nohara T., *Phytochemistry*, **50**, 1369–1374 (1999).
- 14) Park S. Y., Choi H. S., Yook C. S., Nohara T., *Chem. Pharm. Bull.*, 53, 97–99 (2005).
- Sung T. V., Steglich W., Adam G., *Phytochemistry*, **30**, 2349–2356 (1991).
- 16) Adam G., Lischewski M., Phiet H. V., Preiss A., Schmidt J., Sung T. V., *Phytochemistry*, **21**, 1385–1387 (1982).
- 17) Kasai R., Okihara M., Asakawa J., Mizutani K., Tanaka O., *Tetrahedron*, 35, 1427—1432 (1979).
- 18) Corbett R. E., Cong A. N. T., J. Chem. Soc. Perkin Trans. 1, 1985, 2051—2056 (1985).
- 19) Corbett R. E., Cong A. N. T., Holland P. T., Wilkins A. L., Aust. J. Chem., 40, 461–468 (1987).
- Hevel J. M., Marletta M. A., Methods Enzymol., 233, 250–258 (1994).