

The Enantioselective Total Synthesis of a β -Carboline Alkaloid, (*S*)-(-)-Dichotomine C

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The first enantioselective synthesis of a β -carboline alkaloid, dichotomine C, possessing antiallergic effects, was achieved by constructing a β -carboline framework based on the microwave-assisted thermal electrocyclic reaction of a 1-azahexatriene system, followed by the Sharpless asymmetric dihydroxylation.

Key words (*S*)-(-)-dichotomine C; first synthesis; enantioselective synthesis; aza-electrocyclic reaction; microwave

Six new β -carboline alkaloids, dichotomines A, B, C and D, and dichotomides I and II, possessing antiallergic effects, were isolated from *Stellaria dichotoma* by Yoshikawa and co-workers in 2004 (Fig. 1).¹⁾ The structures of the new compounds were determined by spectroscopic and chemical analyses. The absolute configuration of the C14-position in dichotomines A–D were determined to be the *S*-form using

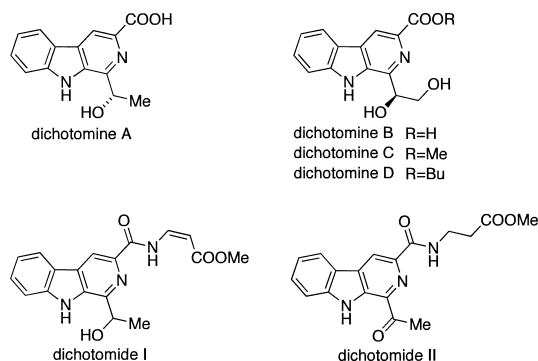


Fig. 1

the modified Mosher's method. Furthermore, examination of the antiallergic effects of the isolated compounds on the release of the β -hexosaminidase in RBL-2H3 cells indicated that dichotomine C had inhibitory activity.

In the course of our studies, we developed a synthesis of biologically active condensed heterocyclic compounds, including natural products, based on a thermal electrocyclic reaction^{2–4)} of either hexatriene^{5–7)} or azahexatriene^{5,6,8)} systems incorporating a principal aromatic or heteroaromatic moiety. Here, we describe the first total synthesis of dichotomine C (**1**) by applying pyrido-annulation and the Sharpless asymmetric dihydroxylation. In a retro-synthetic analysis (Chart 1), we speculated that dichotomine C (**1**) could be derived from 1-chloro- β -carboline **3** through asymmetric dihydroxylation of 1-ethenyl- β -carboline **2**. We also speculated that alkyl β -carboline-3-carboxylate **4** could be obtained by a thermal electrocyclic reaction of 3-alkenylindole-2-carbaldehyde oxime **5** as an application of the synthesis of the β -carboline framework using the 1-azahexatriene system.

The required β -carboline **4** was prepared in three steps starting from *N*-methoxymethyl(MOM)-3-iodoindole-2-carbaldehyde⁹⁾ (**6**) (Chart 2). The Heck reaction¹⁰⁾ between the aldehyde **6** and methyl acrylate in the presence of Pd(OAc)₂ in dimethylformamide gave the 3-alkenylindole-2-carbaldehyde **7** (85%). Subsequent treatment of the alkenylindole **7** with hydroxylamine produced the oxime **5** (75%) as the 1-azahexatriene system, which was subjected to a microwave-assisted thermal electrocyclic reaction in 1,2-dichlorobenzene to yield methyl β -carboline-3-carboxylate **4** (94%).

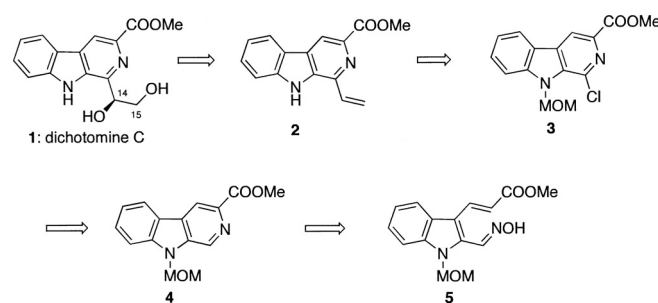


Chart 1

Table 1. Physical and Spectroscopic Data of Dichotomine C (**1**)¹⁵⁾

	Natural product (1) ^{a)}	Synthetic dichotomine C ^{b)} (+)- 1 from AD-mix- α	Synthetic dichotomine C ^{b)} (-)- 1 from AD-mix- β
¹ H-NMR (pyridine- <i>d</i> ₅)	3.97 (3H, s) 4.61 (2H, m) 5.94 (1H, m) 7.39 (1H, m) 7.60 (1H, m) 7.75 (1H, d, <i>J</i> =7.3 Hz) 8.34 (1H, d, <i>J</i> =7.3 Hz) 9.09 (1H, s) 12.60 (1H, br s)	3.96 (3H, s) 4.55–4.72 (2H, m) 5.90–5.96 (1H, m) 7.36–7.43 (1H, m) 7.56–7.63 (1H, m) 7.74 (1H, d, <i>J</i> =7.3 Hz) 8.35 (1H, d, <i>J</i> =7.3 Hz) 9.10 (1H, s) 12.64 (1H, br s)	3.96 (3H, s) 4.54–4.71 (2H, m) 5.91–5.94 (1H, m) 7.37–7.42 (1H, m) 7.56–7.63 (1H, m) 7.72 (1H, d, <i>J</i> =7.3 Hz) 8.35 (1H, d, <i>J</i> =7.3 Hz) 9.12 (1H, s) 12.62 (1H, br s)
MS <i>m/z</i>	286	286	286
mp °C	—	200–202	196–198
IR (cm ⁻¹)	3300, 1723	3267, 1720	3345, 1716
[α] _D ²⁵	-16.6° (<i>c</i> =0.50, MeOH)	+11.9° (<i>c</i> =0.50, MeOH)	-17.9° (<i>c</i> =0.02, MeOH)

a) 500 MHz, b) 300 MHz.

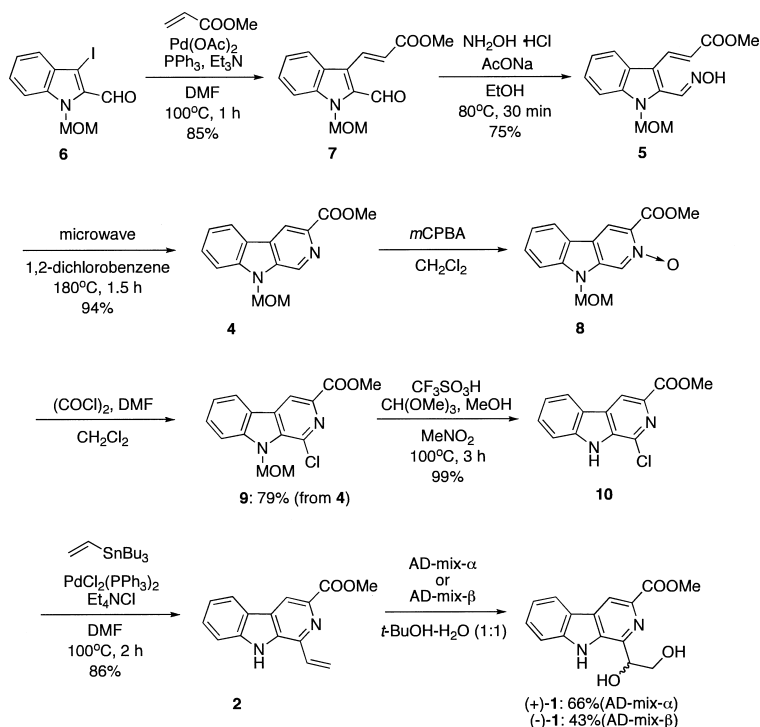


Chart 2

The key compound **2** was synthesized from the *N*-MOM- β -carboline **4** in four steps. Namely, treatment of **4** with *m*-chloroperbenzoic acid (*m*CPBA) followed by chlorination with oxalyl chloride¹¹ in CH_2Cl_2 yielded the *N*-MOM-1-chloro- β -carboline **9** (79% from **4**). Cleavage of the MOM group of **9** with trifluoromethanesulfonic acid in the presence of methanol and trimethyl orthoformate in nitromethane¹² afforded the 1-chloro- β -carboline **10** (99%). The key compound **2** was synthesized in 86% yield from **10** and vinyltributyltin in the presence of $\text{PdCl}_2(\text{PPh}_3)_2$ in dimethylformamide by the Stille reaction. Finally, asymmetric dihydroxylation of the 1-ethenyl- β -carboline **2** with AD-mix- α and - β was examined. Contrary to the Sharpless rule,¹³ (–)-dichotomine C (–)-**1** (43%, $[\alpha]_D^{27} -17.9^\circ$, $c=0.02$ in MeOH, 99.8%ee¹⁴) and its enantiomer (+)-**1** (66%, $[\alpha]_D^{27} +11.9^\circ$, $c=0.5$ in MeOH, 80%ee¹⁴) were obtained by AD-mix- β and AD-mix- α , respectively. The physical and spectroscopic data of synthetic dichotomine C (**1**) were identical to those of natural dichotomine C (**1**), as shown in Table 1.

In conclusion, an enantioselective total synthesis of the 1,3-disubstituted β -carboline alkaloid (*S*)-(–)-dichotomine C (**1**) and its enantiomer (+)-**1** was achieved in an eight-step sequence (17% overall yield from **6** to (–)-**1**) by a microwave-assisted thermal electrocyclic reaction of the 1-azahexatriene system involving the indole 2,3-bond, followed by the Sharpless dihydroxylation. Further studies in this series are in progress.

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 14) The enantiomeric excess of the synthetic (–)-**1** and its enantiomer (+)-**1** were determined by the ¹H-NMR ratio of their (*R*)-M α NP esters, which was synthesized by esterification of the C15-primary hydroxy group in dichotomine C (**1**) with pivaloyl chloride, followed by treatment of the C14-secondary hydroxy group with (*R*)-2-methoxy-2-(1-naphthyl)propionic acid (M α NP acid) in the presence of dicyclohexylcarbodiimide and dimethylaminopyridine.
 15) Although DMSO-*d*₆ as the measurement solvent for the ¹H-NMR spectrum of natural (–)-dichotomine C (**1**) was described in ref. 1, the ¹H-NMR spectrum data of synthetic (–)-**1** by DMSO-*d*₆ were not identical with those of natural (–)-**1**: ¹H-NMR (300 MHz, DMSO-*d*₆) δ : 3.77–3.87 (2H, m), 3.89 (3H, s), 4.83 (1H, t, $J=5.9$ Hz), 5.04–5.12 (1H, m), 5.91 (1H, d, $J=4.4$ Hz), 7.27 (1H, t, $J=7.3$ Hz), 7.56 (1H, t, $J=7.3$ Hz), 7.72–7.76 (1H, m), 8.35 (1H, d, $J=7.3$ Hz), 8.82 (1H, s), 11.61 (1H, s). The ¹H-NMR spectrum of synthetic (–)-**1** was measured again in pyridine *d*₆ by the corrected information from Professor M. Yoshikawa.