New Iridoids from Gelsemium Species

Noriyuki Kogure,^a Naoko Ishii,^a Hiromi Kobayashi,^a Mariko Kitajima,^a Sumphan Wongseripipatana,^b and Hiromitsu Takayama*,^a

^a Graduate School of Pharmaceutical Sciences, Chiba University; 1–33 Yayoi-cho, Inage-ku, Chiba 263–8522, Japan: and

^b Faculty of Pharmaceutical Sciences, Chulalongkorn University; Bangkok 10500, Thailand.

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Four new iridoids structurally related to gelsemide (5) were isolated from two Loganiaceous plants, Gelsemium elegans and G. rankinii. Among them, GEIR-1 (1) has a novel tetracyclic caged structure.

Key words Gelsemium; iridoid; gelsemide

The genus *Gelsemium*, which belongs to Loganiaceae, comprises three species: *G. elegans* Benth, *G. rankinii* Small, and *G. sempervirens* Ait, from which more than seventy indole alkaloids have been isolated. ^{1,2)} We have proved that the origin of "Yakatsu," one of the ancient medicines stored in the Shosoin repository in Japan, is *G. elegans*. ³⁾ Recently, we found that among the structurally diverse *Gelsemium* alkaloids, some gelsedine-type alkaloids showed cytotoxicity against A431 epidermoid carcinoma cells. ⁴⁾ In our continuing chemical studies on *Gelsemium* plants, ^{5—9)} we found new iridoids (1—4) in the leaves of *G. elegans* and *G. rankinii*. In this paper, we describe the structure elucidation of these compounds (Fig. 1).

Results and Discussion

The leaves of G. elegans (3575 g dry weight) were extracted with hot MeOH to yield the extract (941.8 g). The MeOH extract was dissolved in H_2O containing a small amount of MeOH and extracted successively with n-hexane, AcOEt, 5% MeOH/CHCl₃, and n-BuOH. The 5%

Fig. 1. Structures of New Iridoids (1—4) and Gelsemide (5)

Table 1. ¹H- and ¹³C-NMR Data of **1—4**

MeOH/CHCl₃ extract (6.91 g) was separated by SiO₂ flash column chromatography to afford two new iridoids, GEIR-1 (1, 41.2 mg) and GEIR-2 (3, 3.6 mg), along with known alkaloids, gelsenicine, 14,15-dihydroxygelsenicine, ⁵⁾ gelsemoxonine, ⁵⁾ gelsedilam, ⁸⁾ 14-acetoxygelsedilam, ⁸⁾ gelseiridone, ⁸⁾ and gelsefuranidine. ⁸⁾ Further, the third new iridoid, GEIR-3 (4, 119.7 mg), was isolated from the *n*-BuOH fraction. Using a similar procedure, the fourth new iridoid, GRIR-1 (2, 24.4 mg), was isolated from the *n*-BuOH fraction of *G. rank-inii* (see Experimental).

GEIR-1 (1) was obtained as colorless prisms (mp 119— 120 °C). The HR-FAB-MS spectrum gave a protonated molecular ion peak at m/z 213.0763 (M+H⁺) that corresponded to the molecular formula $C_{10}H_{13}O_5$ (m/z 213.0768). The IR spectrum suggested that 1 has a hydroxyl group (3558 cm⁻¹) and a lactone moiety (1766 cm⁻¹). The ¹H-NMR spectrum showed characteristic signals for iridoids, such as methyl protons $[\delta 1.02 (d, H_3-10)]$ and low-field oxygenated protons [δ 5.23 (dd, H-6), δ 5.16 (br s, H-3)] (Table 1). The ¹³C-NMR spectrum showed 10 carbons including one carbonyl carbon [δ 176.2 (C-11)] and one acetal carbon [δ 94.1 (C-3)] (Table 1). ¹H-¹H COSY and HMQC analyses (Fig. 2a) indicated the presence of a seven sp³ carbon chain (-CHCHCHCHCHCHCH₃, C-3, 4, 5, 6, 7, 8, 10), the terminal methine carbon of which was estimated to be an acetal residue from the chemical shift ($\delta_{\rm C}$ 94.1). HMBC crosspeaks of two protons [δ 5.23 (H-6), δ 2.98 (H-4)] and a carbonyl carbon (δ 176.2) indicated the presence of a γ -lactone (Fig. 2a). In addition, the presence of a quaternary carbon (δ 74.9) and an oxymethylene carbon (δ 64.0) was suggested. HMBC

	GEIR-1 (1)		GRIR-1 (2)		GEIR-2 (3)		GEIR-3 (4)	
	$\delta_{ m H}$ (400 MHz)	$\delta_{\rm C}$ (100 MHz)	$\delta_{\mathrm{H}}(500\mathrm{MHz})$	$\delta_{\rm C}$ (125 MHz)	$\delta_{ m H}$ (600 MHz)	$\delta_{\rm C}$ (150 MHz)	$\delta_{ m H}$ (400 MHz)	$\delta_{\rm C}$ (125 MHz)
1	3.61 (d, 9.4)	64.0	3.80 (d, 11.9)	58.9	4.01 (d, 11.7)	59.1	3.94 (dd, 11.0, 2.0)	74.1
	3.57 (d, 9.4)		3.29 (d, 11.9)		3.47 (d, 11.7)		3.75 (d, 11.0)	
3	5.16 (br s)	94.1	5.27 (s)	91.0	5.49 (s)	89.7	7.73 (s)	157.0
4	2.98 (overlapped)	44.8	2.90 (d, 11.2)	44.8	2.99 (overlapped)	44.2		107.1
5	2.98 (overlapped)	49.3	3.12 (dd, 11.2, 6.4)	47.2	2.95 (overlapped)	46.7	2.49 (br s)	46.9
6	5.23 (dd, 7.0, 3.6)	83.0	4.70 (br d, 6.4)	87.3	5.02 (dd, 4.4, 4.4)	83.3	4.24 (dd, 4.0, 4.0)	80.3
7	3.88 (m)	79.5	3.95 (br d, 3.7)	79.0	2.08 (2H, m)	38.2	3.70 (dd, 9.3, 4.0)	73.1
8	2.67 (q, 7.9)	49.8	1.82 (qd, 7.0, 3.7)	40.1	1.85 (m)	35.9	1.62 (m)	44.4
9	\ L ' /	74.9	````	76.0	. /	73.3	. /	74.4
10	1.02 (3H, d, 7.9)	12.7	0.96 (3H, d, 7.0)	6.0	0.99 (3H, d, 6.4)	10.1	1.08 (3H, d, 7.1)	12.9
11		176.2		177.3		175.7		174.0

^{*} To whom correspondence should be addressed. e-mail: takayama@p.chiba-u.ac.jp

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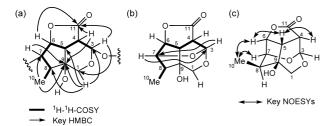


Fig. 2. Key ¹H-¹H COSY, HMBC and NOESY Correlations of GEIR-1

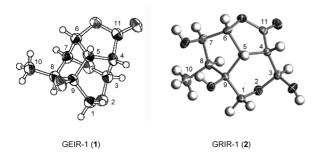


Fig. 3. X-Ray Structures of 1 and 2

correlations between oxymethylene protons (δ 3.61, 3.57, H_2 -1) and an acetal carbon (δ 94.1) indicated that the oxymethylene carbon was attached to one of the acetal oxygens. HMBC correlations between terminal methyl protons (δ 1.02, H₃-10) and H-7 and an oxygenated quaternary carbon (δ 74.9) suggested that the quaternary carbon was attached to C-8. Other cross peaks of three protons (H₂-1, H-4) and the quaternary carbon (δ 74.9) suggested the existence of a 3-oxo-bicyclo[4,3,0]nonane ring (Fig. 2a). HMBC crosspeaks between the protons at δ 3.88 (H-7) and the acetal carbon at δ 94.1 (C-3), and between the protons at δ 5.16 (H-3) and δ 79.5 (C-7) revealed that C-3 and C-7 were connected by an oxygen atom (Fig. 2b). From these analyses, structure 1 having a unique tetracyclic caged structure constructed by an intramolecular acetal function was proposed. NOESY observations illustrated in Fig. 2c supported the stereostructure of 1.

The structure inferred by spectroscopic analysis above was confirmed by X-ray analysis (Fig. 3).¹⁰⁾ This iridoid was considered to be derived from gelsemide ($\mathbf{5}$)¹¹⁾ by Michael attack from 7-OH to C-3 of α,β -unsaturated lactone. When compound $\mathbf{5}$ was treated with PTSA in dioxane, iridoid $\mathbf{1}$ was formed in 12% yield (Chart 1).

GRIR-1 (2) isolated from *G. rankinii* was obtained as colorless prisms (mp 165—169 °C). Its 1 H- and 13 C-NMR spectra were similar to those of GEIR-1 (1), of which molecular formula $C_{10}H_{14}O_6$ had one H_2O molecule more than 1. Coupling constants of the protons at C-6, C-7, and C-8 ($J_{H6-H7}=ca.0$ Hz, $J_{H7-H8}=3.7$ Hz) and NOE between H-7 and H-8 suggested β -orientation of the oxygen function on C-7 (Fig. 4). Taking this finding into account, the molecular formula, and HMBC correlations depicted in Fig. 4, GRIR-1 (2) was considered to be an epimer at C-7 of the dissociated form of the intramolecular acetal function of 1. The structure, including the stereochemistry at C-3 of the hemiacetal function, was finally determined by X-ray crystallographic analysis (Fig. 3). 12

The molecular formula of new iridoid GEIR-2 (3) was es-

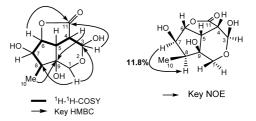


Fig. 4. Key ¹H-¹H COSY, HMBC and NOE Correlations of GRIR-1 (2)

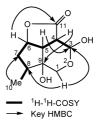


Fig. 5. Key ¹H–¹H COSY and HMBC Correlations of GEIR-2 (3)

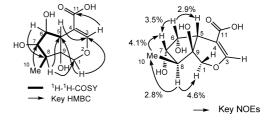


Fig. 6. Key ¹H–¹H COSY, HMBC and NOE Correlations of GEIR-3 (4)

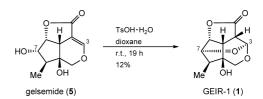


Chart 1. Chemical Conversion of Gelsemide (5) to GEIR-1 (1)

tablished to be $C_{10}H_{14}O_5$ from HR-FAB-MS (m/z 237.0729 [M+Na]⁺), which possessed one oxygen less than **2**. The ¹H-NMR spectrum showed characteristic signals of iridoids, such as methyl protons [δ 0.99 (d, H₃-10)], acetal proton [δ 5.49 (s, H-3)], and low-field oxymethine proton [δ 5.02 (dd, H-6)]. Comparison of the ¹³C-NMR data of **3** with those of **2** (Table 1) particularly of the chemical shift at C-7, indicated that **3** is the 7-deoxyderivative of **2** (Fig. 5). GEIR-2 (**3**) was previously prepared by the enzymatic hydrolysis of 9-hydroxysemperoside (**6**). ¹¹⁾

The HR-EI-MS spectrum of GEIR-3 (4) gave a protonated molecular ion peak at m/z 230.0780 (M⁺) that corresponded to the molecular formula $C_{10}H_{14}O_6$ (m/z 230.0790). UV spectrum and NMR signals at $\delta_{\rm H}$ 7.73 (H-3), $\delta_{\rm C}$ 157.0 (C-3), $\delta_{\rm C}$ 107.1 (C-4), and $\delta_{\rm C}$ 174.0 (C-11) revealed the existence of a β -alkoxyacrylate residue. The ¹³C-NMR spectrum showing one carbonyl carbon [δ 174.0 (C-11)] and four oxygenated carbons [δ 80.3 (C-6), δ 74.4 (C-9), δ 74.1 (C-1), δ 73.1 (C-7)], together with HMBC correlations from H-1 to C-3 and C-5, from H-8 to C-1 and C-5, and from H-3 to C-11, suggested that GEIR-3 (4) has a gelsemide skeleton (Fig. 6).

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Comparing the chemical shift of the proton at C-6 ($\delta_{\rm H}$ 4.24) and the molecular formula of 4 with those of gelsemide (5), it was revealed that compound 4 was a hydrolysis derivative of the lactone moiety of 5. The relative stereochemistry was established by NOE experiments, as shown in Fig. 6.

Experimental

General Procedure 1 H- and 13 C-NMR spectra: JEOL JNM ECP-600, JEOL JNM A-500 or JNM A-400 at 600, 500 or 400 MHz (1 H-NMR) and at 150, 125 or 100 MHz (13 C-NMR), respectively. UV: JASCO V-560. IR: JASCO FT/IR-230. FAB-MS: JEOL JMS-AX500 or AX-505. HR-FAB-MS: JEOL JMS-HX110. EI-MS: JEOL GC-mate. Optical rotation: JASCO P-1020. CD: JASCO J-720WI. TLC: Precoated silica gel 60 F₂₅₄ plates (Merck, 0.25 mm thick). Column chromatography: Silica gel 60 (Merck, 70—230 mesh). Flash column chromatography: Silica gel 60N (Kanto Chemical, 40—50 μm). Medium pressure liquid chromatography (MPLC): C. I. G. prepacked column CPS-HS-221-05 (Kusano Kagakukikai, SiO₂). X-ray crystallography: Rigaku AFC-7 and Bruker APEX II.

Plant Material Gelsemium elegans BENTH. was collected in Phu Laung, Loei Province, Thailand. A voucher specimen was deposited at the Faculty of Pharmaceutical Sciences, Chulalongkorn University, Thailand. Gelsemium rankinii SMALL was harvested from the medicinal plant garden of Chiba University, Japan. A voucher specimen (No. 20051201) was deposited at the Faculty of Pharmaceutical Sciences, Chiba University, Japan.

Extraction and Isolation of 1—5 The leaves of G. elegans Benth. (3575 g dry weight) were extracted with MeOH (20.51, once at room temperature and four times under reflux) to give the extract (941.8 g). The MeOH extract (940.8 g) was dissolved in H₂O (31×2) containing a small amount of MeOH and extracted with n-hexane (1.71×3) to give the nhexane extract (95.91 g). The aqueous layer was successively extracted with AcOEt (1.71×3) , 5% MeOH/CHCl₃ (1.71×4) , and n-BuOH (1.71×4) to give the AcOEt extract (76.91 g), the 5% MeOH/CHCl₃ extract (7.04 g), and the *n*-BuOH extract (278.31 g), respectively. The 5% MeOH–CHCl₃ extract (6.91 g) was separated by SiO₂ flash column chromatography with CHCl₃/MeOH gradient to give 6 fractions: fr. A 0—3% MeOH/CHCl₃ (48.4 mg); fr. B 3—5% MeOH/CHCl₃ (180.4 mg); fr. C 5—10% MeOH/ CHCl₃ (1648 mg); fr. D 10—20% MeOH/CHCl₃ (2037 mg); fr. E 20—50% MeOH/CHCl₃ (1059 mg); and fr. F 50% MeOH/CHCl₃ and MeOH (1572 mg). Fr. C was purified successively by SiO₂ flash column chromatography (MeOH/CHCl3 gradient or MeOH/AcOEt/CHCl3 gradient), MPLC (3% MeOH/CHCl₃), and MPLC (50% AcOEt/CHCl₃) to afford GEIR-1 (1, 41.2 mg). Fr. D was separated by SiO₂ flash column chromatography with MeOH/CHCl₃ gradient to give 5 fractions: fr. DA 5% MeOH/CHCl₃ (174.6 mg); fr. DB 5% MeOH/CHCl₃ (740.9 mg); fr. DC 5% MeOH/CHCl₃ (618.9 mg); fr. DD 5—10% MeOH/CHCl₃ (439.3 mg); and fr. DE MeOH (91.8 mg). Fr. DC was further purified by SiO₂ flash column chromatography (MeOH/AcOEt gradient), SiO₂ flash column chromatography (MeOH/ CHCl₃ gradient), and MPLC (5% MeOH/CHCl₃ and 50% AcOEt/hexane) to afford GEIR-2 (3, 3.6 mg). The *n*-BuOH extract (18.93 g) was separated on a Sephadex LH-20 column with H₂O/MeOH gradient to give 23 fractions. The fraction that was eluted with MeOH (6993 mg) was subjected to SiO₂ flash column chromatography (MeOH/AcOEt gradient or MeOH/CHCl₃ gradient) and purified several times to afford GEIR-3 (4, 119.7 mg).

The aerial part of *G. rankinii* SMALL (1144 g dry weight) was extracted with MeOH (1.81, twice at room temperature and four times under reflux) to give the extract (232.7 g). The MeOH extract was dissolved in $\rm H_2O$ (0.51×2) containing a small amount of MeOH and extracted with *n*-hexane (0.41×3) to give the *n*-hexane extract (29.08 g). The aqueous layer was successively extracted with AcOEt (0.61, 0.51×2), 5% MeOH/CHCl₃ (0.61, 0.51×2), and *n*-BuOH (0.61, 0.51×2) to give the AcOEt extract (14.41 g), the 5% MeOH/CHCl₃ extract (8.92 g), and the *n*-BuOH extract (40.01 g), respectively. The *n*-BuOH extract (40.01 g) was separated on a Sephadex LH-20 column with $\rm H_2O/MeOH$ gradient to give 8 fractions. The fraction that was eluted with $\rm H_2O$ (7749 mg) was purified by $\rm SiO_2$ flash column chromatography (MeOH/CHCl₃ and AcOEt/*n*-hexane gradient) to afford GRIR-1 (2, 24.4 mg).

GEIR-1 (1): Colorless prisms, mp 119—120 °C (CHCl₃). FAB-MS (NBA) m/z: 213 (M+H⁺). HR-FAB-MS (NBA/PEG) m/z: 213.0763 (M+H⁺, Calcd for C₁₀H₁₃O₅: 213.0768). ¹H- and ¹³C-NMR: see Table 1. IR (KBr) cm⁻¹: 3558, 3426, 1766. [α]_D²⁴ +41.9° (c=1.02, MeOH). CD (c=0.450 mmol/l, MeOH, 24 °C) $\Delta\varepsilon$ (nm): 0 (248), -0.34 (215). *Anal*. Calcd for C₁₀H₁₂O₅: C,

56.6; H, 5.7; O, 37.7. Found: C, 56.8; H, 5.8; O, 37.5.

GRIR-1 (2): Colorless prisms, mp 165—169 °C (AcOEt). EI-MS m/z: 212 (M-H $_2$ O $^+$). HR-EI-MS m/z: 212.0668 (M-H $_2$ O $^+$, Calcd for C $_{10}$ H $_{12}$ O $_5$: 212.0684). 1 H- and 13 C-NMR: see Table 1. $[\alpha]_D^{24} + 24.7^\circ$ (c=0.017, MeOH). GEIR-2 (3): White amorphous powder, FAB-MS (NBA+NaCl) m/z: 237 (M+Na $^+$), HR-FAB-MS (Gly+NaCl+H $_2$ O/PEG) m/z: 237.0729 (M+Na $^+$, Calcd for C $_{10}$ H $_1$ 4O $_5$ Na: 237.0739). 1 H- and 13 C-NMR: see Table 1. $[\alpha]_D^{24} + 5.7^\circ$ (c=0.16, MeOH).

GEIR-3 (4): Yellowish amorphous powder, EI-MS m/z: 230 (M⁺), 194, 166, 153. HR-EI-MS m/z: 231.0780 (M⁺, Calcd for $C_{10}H_{14}O_6$: 230.0790). UV $\lambda_{\rm max}$ (MeOH) nm (log ε): 236.5 (4.03). 1H - and ^{13}C -NMR: see Table 1. [ω] $_D^{\rm 1D}$ - 100.1° (c=1.21, MeOH). CD (c=0.202 mmol/l, MeOH, 16 °C) $\Delta\varepsilon$ (nm): 0 (265), -7.91 (231), 0 (207), +1.37 (203).

Chemical Conversion of Gelsemide (5) to GEIR-1 (1) To a stirred solution of gelsemide (5, $5.0\,\mathrm{mg}$, $0.024\,\mathrm{mmol}$) in 1,4-dioxane (0.6 ml), p-toluenesulfonic acid monohydrate ($50.0\,\mathrm{mg}$, $0.263\,\mathrm{mmol}$) was added and the mixture was stirred at room temperature under Ar. After 19 h, the reaction mixture was quenched with 5% aq. NaHCO $_3$ and extracted with 10% MeOH/CHCl $_3$. Then, the organic layer was washed with brine, dried over MgSO $_4$, and evaporated. The residue was purified by SiO $_2$ column chromatography (2% MeOH/CHCl $_3$) to afford 1 ($0.6\,\mathrm{mg}$, 12%), which was identical with the natural product in all respects.

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References and Notes

- Takayama H., Sakai S., "The Alkaloids," Vol. 49, Chap. 1, ed. by Cordell G. A., Academic Press, San Diego, 1997.
- 2) Kitajima M., J. Nat. Med., 61, 14—23 (2007).
- Kitajima M., Arai Y., Takayama H., Aimi N., Proc. Jpn. Acad., Ser. B, 74, 159—163 (1998).
- Kitajima M., Nakamura T., Kogure N., Ogawa M., Mitsuno Y., Ono K., Yano S., Aimi N., Takayama H., *J. Nat. Prod.*, **69**, 715—718 (2006).
- Kitajima M., Kogure N., Yamaguchi K., Takayama H., Aimi N., Org. Lett., 5, 2075—2078 (2003).
- 6) Kogure N., Nishiya C., Kitajima M., Takayama H., *Tetrahedron Lett.*, **46**, 5857—5861 (2005).
- Kitajima M., Urano A., Kogure N., Takayama H., Aimi N., Chem. Pharm. Bull., 51, 1211—1214 (2003).
- Kogure N., Ishii N., Kitajima M., Wongseripipatana S., Takayama H., Org. Lett., 8, 3085—3088 (2006).
- Kogure N., Someya A., Urano A., Kitajima M., Takayama H., J. Nat. Med., 61, 208—212 (2007).
- 10) X-ray crystallographic analysis of 1. All measurements were carried out on a Rigaku AFC7S diffractometer with graphite monochromated $CuK\alpha$ radiation. Crystal data: orthorhombic, $C_{10}H_{12}O_5$ (Mw: 212.2), space group $P2_12_1$ with a=9.231(1)Å, b=11.200(2)Å, c=8.910(2) Å, V=921.3(3)ų, Z=4, and $D_{calc}=1.53$ g/cm³. The structure was solved by direct methods (SIR97) and expanded using Fourier techniques (DIRDIF94). Non-hydrogen atoms were refined anisotropically. Hydrogen atoms were included but not refined. The final cycle of full-matrix least-squares refinement was based on 979 reflections ($I>3.00\sigma(I)$, $2\theta<137.88$) and 138 variable parameters and converged with unweighted and weighted agreement factors of R=0.055, $R_w=0.092$.
- Jensen S. R., Kirk O., Nielsen B. J., Norrestam R., *Phytochemistry*, 26, 1725—1731 (1987).
- 12) X-ray crystallographic analysis of 2. All measurements were carried out on a Bruker APEX II with graphite monochromated MoKα radiation. Crystal data: orthorhombic, C₁₀H₁₄O₆ (Mw: 230.2), space group P2₁2₁2₁ with a=6.3717(4) Å, b=10.9746(7) Å, c=14.6919(9) Å, V=1027.36(11) Å³, Z=4, and D_{calc}=1.488 g/cm³. The structure was solved by direct methods (SHELEX97) and expanded using Fourier techniques (DIRDIF94). Non-hydrogen atoms were refined anisotropically. Hydrogen atoms were included but not refined. The final cycle of full-matrix least-squares refinement was based on 2366 reflections and 150 variable parameters and converged with unweighted and weighted agreement factors of R=0.028, R_w=0.069.