## First Total Synthesis of the Phenolic 7,8-Dihydro-8-oxoprotoberberine Alkaloid, Cerasonine

## Thanh Nguyen LE and Won-Jea CHO\*

College of Pharmacy and Research Institute of Drug Development, Chonnam National University; Yong-Bong Dong, Bukgu, Kwangju 500–757, South Korea. Received February 19, 2008; accepted April 22, 2008; published online April 25, 2008

First total synthesis of the phenolic protoberberine, cerasonine, was accomplished through a coupling reaction between o-toluamide and benzonitrile. This key step provided the 3-arylisoquinoline which was then successfully converted to 7,8-dihydro-8-oxoprotoberberine through an intramolecular  $S_N 2$  reaction.

Key words cerasonine; protoberberine; 3-arylisoquinoline; coupling reaction

Among the natural alkaloids, protoberberines have been one of the main synthetic target molecules due to their diverse pharmacological properties,<sup>1)</sup> such as antitumor,<sup>2)</sup> antifungal,<sup>3)</sup> and antimicrobial<sup>4,5)</sup> activities. Benzo[c]phenanthridine alkaloids have been biosynthesized from the corresponding protoberberine alkaloids presumably via a 3-arylisoquinoline intermediate.<sup>6)</sup>

Cerasonine 1,<sup>7)</sup> a phenolic protoberberine, was isolated from Polyalthia cerasoides (ROXB.) BEDD. (Annonaceae) in 1997 and has not been synthesized yet (Fig. 1).

As a part of our continuous efforts to synthesize all of the different substitution patterns of protoberberines,<sup>8)</sup> we applied our developed synthetic method to the synthesis of



Fig. 1. Structure of the Protoberberine Alkaloid, Cerasonine

cerasonine. This strategy was based on preparation of a synthetic intermediate that retains all appropriate substituents on the aromatic rings of cerasonine. For this, a coupling reaction between N.N-diethyl-o-toluamide 3 with benzonitrile 4 was carried out to yield the 3-arylisoquinoline 2, which could be converted to protoberberine via an intramolecular  $S_{N2}$  reaction as depicted in Chart 1. Recently, we efficiently synthesized isoquinolines based on this synthetic route and reported the total synthesis of protoberberine and benzo[c]phenanthridine alkaloids by ring closure of the two carbon chain either on position 2 (NH) or 4 of the 3-arylisoquinolinone intermediates.<sup>9–12)</sup> We also succeeded in preparing diversely substituted benzo[c]phenanthridines as well as protoberberines.<sup>13)</sup> The advantages of our methodology are easy accessibility to the starting materials and a one-pot procedure for construction of all carbon atoms in the alkaloids.

## **Results and Discussion**

For the coupling reaction, toluamide and benzonitrile were synthesized by the conventional methods described in Charts 2 and 3. Benzyloxybromobenzaldehyde 5 was prepared from



Chart 1. Retrosynthetic Analysis of Cerasonine



Chart 2. Synthesis of Benzonitrile 4

\* To whom correspondence should be addressed. e-mail: wjcho@jnu.ac.kr

vanillin in two steps.<sup>14)</sup> The aldehyde group of **1** was protected with ethylene glycol and the bromide **6** was converted to benzonitrile by treatment with CuCN in DMF in 85% yield. After hydrolysis of the acetal group with 5% HCl, the resulting aldehyde **8** was reacted with  $Ph_3P^+CH_2OMeI^-/n$ -BuLi to yield styrene **9** as a *cis/trans* (2 : 1) mixture. Without separation, the isomers were hydrolyzed to produce the homobenzaldehyde, which was then reduced with NaBH<sub>4</sub> followed by protection with methoxymethyl chloride to yield the MOM-protected benzonitrile **4** in 89% yield.

The *o*-toluamide **3** was prepared from 3,4-dimethoxytoluene **12** in three steps as shown in Chart 3. Vilsmeier reaction of 3,4-dimethoxytoluene **12** gave the corresponding benzaldehyde **13**,<sup>15</sup> which was then oxidized with NaClO<sub>2</sub><sup>16</sup> to afford the carboxylic acid **14** in 91% yield. Treating the benzoic acid **14** with oxalyl chloride and diethylamine gave *N*,*N*-diethyl *o*-toluamide **3** in 90% yield.<sup>17</sup>

Once we had the starting materials for the coupling reaction, *N*,*N*-diethyl-*o*-toluamide **3** was treated with *n*-BuLi and benzonitrile **4** at -70 °C in THF to afford 3-arylisoquinoline-1(2*H*)-one **15**. The MOM protective group was removed with 10% HCl to give the alcohol **16**, which was then reacted with *p*-TsCl in DMF in the presence of K<sub>2</sub>CO<sub>3</sub> to provide the protoberberine **17**. Cerasonine **1** was easily obtained in 63% yield by hydrogenolysis of **17** in 50 psi H<sub>2</sub> atmosphere with 5% Pd/C catalyst. By comparison of <sup>1</sup>H-NMR data of natural and synthetic compounds as shown in Table 1 and Fig. 2, we synthetically confirmed the structure of cerasonine **1**.

In conclusion, we report here the first total synthesis of phenolic 7,8-dihydro-8-oxoprotoberberine cerasonine in four steps from the toluamide **3** and benzonitrile **4**. Our synthesis illustrates a versatile way to prepare diversely substituted protoberberines, nonphenolic as well as phenolic alkaloids.



Melting points were determined by the capillary method on Electrothermal IA9200 digital melting point apparatus and were uncorrected. Nuclear magnetic resonance (NMR) data for <sup>1</sup>H-NMR were taken on Varian Unity 300 Plus spectrometer and were reported in ppm, downfield from the peak of the internal standard, tetramethylsilane. The data are reported as follows: chemical shift, number of protons, multiplicity (s: singlet, d: doublet, t: triplet, q: quartet, m: multiplet, b: broadened). IR spectra were recorded on JASCO-FT IR spectrometer using CHCl<sub>3</sub> and KBr pellets. Mass spectra were obtained on JEOL JNS-DX 303 applying the electron-impact (EI) method. Column chromatography was performed on Merck silica gel 60 (70—230 mesh). TLC was performed using plates coated with silica gel 60 F254 that were purchased from Merck.

Chemical reagents were purchased from Aldrich Chemical Co. and used without further purification. Solvents were distilled prior to use: THF and

Table 1. <sup>1</sup>H-NMR [J (Hz)] Comparison of Natural and Synthetic Cerasonine



Position	Synthetic cerasonine	Natural cerasonine <sup>7)</sup>
1	7.35	7.23
4	6.83	6.81
5	2.92 (J=6.1  Hz)	2.90 (J=5.9 Hz)
6	4.36 (J=6.1  Hz)	4.34 ( <i>J</i> =5.9 Hz)
9	7.80	7.80
12	6.91	6.93
13	6.72	6.81
3, 10, 11	3.94, 4.01, 4.01	3.99, 4.00, 4.01



Chart 3. Synthesis of Toluamide 3

COOL

MeC

MeC

MeC

MeC

12

14



Fig. 2. <sup>1</sup>H-NMR Spectra of Synthetic Cerasonine



Chart 4. Synthesis of Cerasonine

**2-(4-Benzyloxy-2-bromo-5-methoxyphenyl)-[1,3]dioxolane** (6) A mixture of compound 5 (22.47 g, 70 mmol), ethylene glycol (8.68 g, 140 mmol), and *p*-TsOH (100 mg) was refluxed for 3 h with a Dean–Stark apparatus. The mixture was cooled to 0 °C and NaHCO<sub>3</sub> (300 mg) was added. After filtering, the filtrate was concentrated *in vacuo* to give compound **6** as a yellow solid (23.0 g, 90%). mp: 85–88 °C. <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 7.44–7.29 (m, 5H), 7.13 (s, 3H), 7.05 (s, 1H); 5.97 (s, 1H), 5.12 (s, 2H), 4.20–4.04 (m, 4H), 3.88 (s, 3H). EI-MS: *m/z* 365 (M<sup>+</sup>, 46). HR-MS-EI (calcd for C<sub>17</sub>H<sub>17</sub>BrO<sub>4</sub>): 365.2376, found 365.2381.

**5-Benzyloxy-2-[1,3]dioxolan-2-yl-4-methoxybenzonitrile (7)** A mixture of acetal **6** (21.9 g, 60 mmol) and CuCN (6.3 g, 70 mmol) in DMF (20 ml) was refluxed for 2 h. The hot and dark reaction mixture was poured into a warm solution of sodium cyanide (14.7 g, 0.3 mol) in water. The mixture was shaken well and then extracted with benzene. The combined extract was concentrated and column-purified to give benzonitrile **7** as a yellow solid (15.86 g, 85%). mp: 75–78 °C. IR (cm<sup>-1</sup>): 2220 (CN). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 7.43–7.32 (m, 5H), 7.11 (s, 3H), 7.11 (s, 1H); 5.93 (s, 1H), 5.15 (s, 2H), 4.25–4.06 (m, 4H), 3.95 (s, 3H). EI-MS: *m/z* 311 (M<sup>+</sup>, 100). HR-MS-EI (calcd for C<sub>18</sub>H<sub>17</sub>NO<sub>4</sub>): 311.3468, found 311.3467.

**5-Benzyloxy-2-formyl-4-methoxybenzonitrile (8)** The cyano acetal **7** (15.55 g, 50 mmol) in 5% HCl (100 ml) was warmed to 50—60 °C for 15 min. The solid was collected, washed with water and dried *in vacuo* to give aldehyde **8** as a pale yellow solid (12.03 g, 90%). mp: 116—118 °C. IR (cm<sup>-1</sup>): 2230 (CN). <sup>1</sup>H-NMR (CDCl<sub>3</sub>):  $\delta$  10.24 (s, 1H), 7.50 (s, 1H), 7.43—7.36 (m, 5H), 7.20 (s, 1H); 5.24 (s, 2H), 3.99 (s, 3H). EI-MS: *m/z* 267 (M<sup>+</sup>, 57). HR-MS-EI (calcd for C<sub>16</sub>H<sub>13</sub>NO<sub>3</sub>): 267.2925, found 267.2929.

5-Benzyloxy-4-methoxy-2-(2-methoxyvinyl)benzonitrile (9) To a solution of (methoxymethyl)triphenylphosphonium chloride (20.52 g, 60 mmol) in dry THF (30 ml), 1.6 M n-butyl lithium (38 ml, 60 mmol) was added at 0 °C and the solution was stirred at 0 °C for 1 h. The reaction mixture was then added to aldehyde 8 (10.68 g, 40 mmol) in THF (30 ml). The reaction mixture was stirred at room temperature for 30 min. The reaction was quenched with water and extracted with ethyl acetate. The organic layers were washed with water and brine and dried over sodium sulfate. After removing the solvent, the residue was purified by column chromatography with n-hexane-ethyl acetate (3:1) to afford a mixture of cis/trans isomer (2:1 ratio) as a yellow solid (8.97 g, 76%).<sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ (cis): 7.71 (s, 1 H), 7.43-7.30 (m, 5H), 7.00 (s, 1 H), 6.26 (d, J=7.2 Hz, 1H), 5.55 (d, J=7.2 Hz, 1H), 5.12 (s, 2H), 3.93 (s, 3H), 3.83 (s, 3H).  $\delta$ (trans): 7.43-7.30 (m, 5H), 7.13 (d, J=12.9 Hz, 1H), 6.99 (s, 1H), 6.84 (s, 1H), 6.02 (s, 2H), 6.04. (d, J=12.9 Hz, 1H), 5.11 (s, 2H), 3.93 (s, 3H), 3.73 (s, 3H).

**5-Benzyloxy-4-methoxy-2-(2-oxo-ethyl)benzonitrile (10)** The reaction mixture of *cis/trans* isomer **9** (8.85 g, 30 mmol) in acetone (50 ml) and 10% HCl (20 ml) was refluxed for 3 h. The acetone was removed *in vacuo*, and the reaction mixture was poured into water and extracted with ethyl acetate. The ethyl acetate extracts were washed with water and brine and dried over anhydrous sodium sulfate. After removing the solvent, the residue was purified by column chromatography on silica gel with *n*-hexane–ethyl acetate (1 : 1) to give compound **10** as a solid (7.59 g, 90%). mp: 100–103 °C. IR (cm<sup>-1</sup>): 2220 (CN), 1720 (C=O), 1300–1000 (C–O). <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>) & 9.79 (s, 1 H), 7.43–7.30 (m, 5H), 7.11 (s, 1H), 6.75 (s, 1H), 5.14 (s, 2H), 3.92 (s, 3H). EI-MS: *m/z* 281 (M<sup>+</sup>, 100). HR-MS-EI (calcd for C<sub>17</sub>H<sub>15</sub>NO<sub>3</sub>): 281.2599, found 281.2897.

**5-Benzyloxy-2-(2-hydroxyethyl)-4-methoxybenzonitrile** (11) NaBH<sub>4</sub> (1.52 g, 40 mmol) was added to a mixture of aldehyde **10** (5.62 g, 20 mmol) in acetic acid (40 ml). After the reaction was over, acetic acid was removed *in vacuo* and the resulting mixture was poured into water and extracted with ethyl acetate. The organic solvent was evaporated off and the residue was purified by column chromatography with *n*-hexane–ethyl acetate to give al-cohol **11** as a yellow solid (4.59 g, 81%). mp: 116.5—118.5 °C. IR (cm<sup>-1</sup>): 3360 (OH), 2220 (CN), 1720 (C=O), 1300—1000 (C–O). <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.43—7.32 (m, 5H), 7.06 (s, 1H), 6.85 (s, 1H), 5.12 (s, 2H), 3.92 (s, 3H), 3.90 (m, 2H), 3.01 (t, *J*=6.6 Hz, 2H). EI-MS: *m/z* 283 (M<sup>+</sup>, 79). HR-MS-EI (calcd for C<sub>17</sub>H<sub>17</sub>NO<sub>3</sub>): 283.3358, found 283.3357.

**5-Benzyloxy-4-methoxy-2-(2-methoxymethoxyethyl)benzonitrile** (4) To a mixture of alcohol **11** (4.53 g, 16 mmol) in  $\text{CH}_2\text{Cl}_2$  at 0 °C, diisopropylethylamine (3.87 g, 32 mmol) and chloromethylmethyl ether (2.5 g, 32 mmol) were added. After the reaction was over,  $\text{CH}_2\text{Cl}_2$  was removed *in vacuo* and the residue was purified by column chromatography with *n*hexane–ethyl acetate (3 : 1) to give benzonitrile **4** as a yellow solid (4.64 g, 89%). mp: 72—74 °C. IR (cm<sup>-1</sup>): 2219 (CN). <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.42—7.33 (m, 5H), 7.05 (s, 1H), 6.86 (s, 1H), 5.11 (s, 2H), 4.60 (s, 2H), C<sub>19</sub>H<sub>21</sub>NO<sub>4</sub>): 327.3902, found 327.3901. **6-Methylveratraldehyde (13)** Phosphorus oxychloride (24.5 g, 160 mmol) was added to 6.08 g (40 mmol) of 3,4-dimethoxytoluene under nitrogen with stirring. The mixture was heated to 80 °C, and 11.7 g (160 mmol) of DMF was added while the reaction temperature was maintained at 90— 95 °C. The mixture was stirred at 95 °C for 4 h. The dark brown syrup was cooled to 40 °C, poured cautiously onto crushed ice, and extracted with ether. The combined ether extracts were washed with brine, dried, and evaporated *in vacuo* to give 6-methylveratraldehyde **13** as a brown oil, which crystallized on standing overnight (6.73 g, 94%). mp: 73—74 °C. <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 10.32 (s, 1H), 7.45 (s, 1H), 6.82 (s, 1H), 4.06 (s, 3H), 4.03 (s, 3H), 2.76 (s, 3H). EI-MS: *m/z* 180 (M<sup>+</sup>, 100).

**2-Methyl-4,5-dimethoxybenzoic acid (14)** To a stirred mixture of aldehyde **13** (6.73 g, 37 mmol) in DMSO (50 ml) and NaH<sub>2</sub>PO<sub>4</sub> (1.8 g) in water (20 ml), a solution of 80% NaClO<sub>2</sub> (9 g, 80 mmol) in water (60 ml) was added at room temperature. The reaction mixture stood overnight and then 5% NaHCO<sub>3</sub> solution was added. The aqueous layer was extracted twice with ether and then acidified with c-HCl. The precipitated carboxylic acid was taken up with CH<sub>2</sub>Cl<sub>2</sub>. The extracts were combined, washed with brine, dried, and evaporated to give the benzoic acid **14** (6.94 g, 95%). mp: 139—141 °C. <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.64 (s, 1H), 6.73 (s, 1H), 3.95 (s, 1H), 3.95 (s, 6H), 2.63 (s, 3H). EI-MS: *m/z* 196 (M<sup>+</sup>, 65).

**2-Methyl-4,5-dimethoxy-***N*,*N*-**diethylbenzamide (3)** To a suspension of 2-methyl-4,5-dimethoxybenzoic acid 14 (5 g) in CH<sub>2</sub>Cl<sub>2</sub> (50 ml) containing pyridine (3.2 g), oxalyl chloride (17.8 ml) was added slowly with stirring. After an additional 2 h of stirring, the excess oxalyl chloride was removed *in vacuo* and the last trace of oxalyl chloride was removed by co-distillation with benzene. The obtained acid chloride was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (60 ml) and carefully treated with diethyl amine (18.65 g) at 0 °C. The reaction mixture was diluted with water, the organic layer was separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organic portions were washed with water and brine, dried, and then evaporated. The residue was purified by column chromatography to give 2-methyl-4,5-dimethoxy-*N*,*N*-diethylbenzamide **3** as an oil (5.7 g, 90%). IR (cm<sup>-1</sup>): 1625 (CO). <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 6.68 (s, 2H), 3.97 (s, 3H), 3.94 (s, 3H), 3.75–2.90 (m, 4H), 2.23 (s, 3H), 1.50–0.85 (m, 6H). EI-MS: *m/z* 251 (M<sup>+</sup>, 100).

**3-[5-Benzyloxy-4-methoxy-2-(2-methoxymethoxyethyl)phenyl]-6,7-dimethoxy-2H-isoquinolin-1-one (15)** A solution of *N*,*N*-diethyltoluamide **3** (1.04 g, 4 mmol) and benzonitrile **4** (1.01 g, 3 mmol) in dry THF (40 ml) were added dropwise to a solution of *n*-BuLi (5 ml of 1.6 m in hexane, 8 mmol) in THF (30 ml) at -70 °C and the reaction mixture was stirred at the same temperature for 6 h. The reaction was quenched with water and extracted with ethyl acetate and dried over sodium sulfate. After removing the solvent, the residue was purified by column chromatography with *n*-hexane–ethyl acetate (1:1) to afford compound **15** as an orange oil (507 mg, 33%). IR (cm<sup>-1</sup>): 3400 (NH), 1645 (C=O). <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>) & 10.4 (b, 1H), 7.78 (s, 1H), 7.42—7.33 (m, 5H), 6.97 (s, 1H), 6.89 (s, 1H), 6.81 (s, 1H), 6.33 (s, 1H), 5.19 (s, 2H), 4.72 (s, 2H), 4.00 (s, 3H), 3.99 (s, 3H), 3.92 (s, 3H), 3.88 (m, 2H), 3.31 (s, 3 H), 2.88 (m, 2H). EI-MS m/z (%): 505 (M<sup>+</sup>, 100), 414 (61), 91 (53). EI-MS m/z (%): 505 (M<sup>+</sup>, 35). HR-MS-EI (calcd for C<sub>29</sub>H<sub>31</sub>NO<sub>7</sub>): 505.5831, found 505.5837.

**3-[5-Benzyloxy-2-(2-hydroxy-ethyl)-4-methoxy-phenyl]-6,7-dimethoxy-2***H***-isoquinolin-1-one (16) To the mixture of compound 15 (300 mg, 0.6 mmol) in THF (15 ml), 10% HCl (5 ml) was added and the reaction was refluxed for 2 h. The reaction mixture was poured into water and extracted with ethyl acetate. The ethyl acetate extracts were washed with water and brine and dried over anhydrous sodium sulfate. After removing the solvent, the residue was purified by column chromatography on silica gel with CH<sub>2</sub>Cl<sub>2</sub>: MeOH (20:1) to give the alcohol 16 as a yellow solid (205 mg, 75%). mp: 177–179 °C. IR (cm<sup>-1</sup>): 3400 (NH, OH), 1642 (C=O). <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>) \delta: 7.71 (s, 1H), 7.42–7.33 (m, 5H), 6.92 (s, 1H), 6.88 (s, 1H), 6.82 (s, 1H), 6.36 (s, 1H), 5.11 (s, 2H), 4.13 (t,** *J***=5.4 Hz, 2H), 4.06 (s, 3H), 3.99 (s, 3H), 2.81 (t,** *J***=5.4 Hz, 2H). EI-MS** *m/z* **(%): 461 (M<sup>+</sup>, 38). HR-MS-EI (calcd for C<sub>27</sub>H<sub>27</sub>NO<sub>6</sub>): 461.5288, found 461.5291.** *Anal.* **Calcd for C<sub>27</sub>H<sub>27</sub>NO<sub>6</sub>: C, 70.27; H, 5.90; N, 3.04. Found: C, 70.56; H, 5.79; N, 3.15.** 

**2-Benzyloxy-3,10,11-trimethoxy-5,6-dihydro-isoquino**[**3,2-***a*]**isoquinolin-8-one (17)** The mixture of compound **16** (170 mg, 0.36 mmol), tosyl chloride (133 mg, 0.7 mmol) and K<sub>2</sub>CO<sub>3</sub> (290 mg, 2.1 mmol) in DMF (10 ml) was stirred at 100 °C for 4 h. Water was added and the reaction mixture was extracted with ethyl acetate. The ethyl acetate extracts were washed with water and brine and dried over anhydrous sodium sulfate. After removing the solvent, the residue was purified by column chromatography on silica gel with *n*-hexane–ethyl acetate (1:2) to give 8-oxyprotoberberine **17** as a yellow solid (100 mg, 63%). mp: 173–175 °C. IR (cm<sup>-1</sup>): 1636. <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>) & 7.79 (s, 1H), 7.52–7.41 (m, 5H), 7.30 (s, 1H), 6.89 (s, 1H), 6.76 (s, 1H), 6.70 (s, 1H), 5.22 (s, 2H), 4.36 (t, J=6.0 Hz, 2H), 4.01 (s, 3H), 3.94 (s, 3H), 2.88 (t, J=6.0 Hz, 2H). EI-MS *m/z* (%): 443 (M, 100). HR-MS-EI (calcd for C<sub>27</sub>H<sub>25</sub>NO<sub>5</sub>): 443.5135, found 443.5134. *Anal.* Calcd for C<sub>27</sub>H<sub>25</sub>NO<sub>5</sub>: C, 73.12; H, 5.68; N, 3.16. Found: C, 73.26; H, 5.76; N, 3.17.

**2-Hydroxy-3,10,11-trimethoxy-5,6-dihydro-isoquino[3,2-***a***]isoquinolin-8-one (1) Cerasonine The mixture of compound 17 (80 mg, 0.18 mmol) in EtOH (10 ml) in the presence of 5% Pd/C (20 mg) was treated with 50 psi H<sub>2</sub> for 4 h using Parr apparatus. After the reacted catalyst was filtered off, the filtrate was washed with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic layer was evaporated off to give a residue that was purified by column chromatography with** *n***-hexane–ethyl acetate (1: 2) to yield cerasodine as a white solid (40 mg, 63%). mp: >200 °C (ilt.<sup>7)</sup> oil). IR (cm<sup>-1</sup>): 1649 (CO). UV (EtOH) \lambda\_{max} (log \varepsilon): 230 (3.56), 259 (3.46), 334 (3.30). <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>) \delta: 7.80 (s, 1H), 7.35 (s, 1H), 6.91 (s, 1H), 6.83 (s, 1H), 6.72 (s, 1H), 5.22 (s, 2H), 4.36 (t,** *J***=6.1 Hz, 2H), 4.01 (s, 3H), 4.01 (s, 3H), 3.94 (s, 3H), 2.92 (t,** *J***=6.1 Hz, 2H). EI-MS** *m/z* **(%): 353 (M, 100). HR-MS-EI (calcd for C<sub>20</sub>H<sub>19</sub>NO<sub>5</sub>): 353.3851, found 353.3852.** 

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## References

- Simanek V., "The Alkaloids," Vol. 26, ed. by Brossi A., Academic Press, Orlando, 1985, pp. 185—240.
- 2) Min Y. D., Yang M. C., Lee K. H., Kim K. R., Choi S. U., Lee K. R.,

Arch. Pharm. Res., 29, 757-761 (2006).

- Vollekova A., Kost'alova D., Kettmann V., Toth J., *Phytother. Res.*, 17, 834–837 (2003).
- Hwang J. M., Kuo H. C., Tseng T. H., Liu J. Y., Chu C. Y., Arch. Toxicol., 80, 62–67 (2006).
- Faizi S., Khan R. A., Azher S., Khan S. A., Tauseef S., Ahmad A., *Planta Med.*, 69, 350–355 (2003).
- 6) Takao N., Kamigauchi M., Okada M., *Helv. Chim. Acta*, **66**, 473–484 (1983).
- Gonzalez M. C., Zafra-Polo M. C., Amparo-Blazquez M., Serrano A., Cortes D., J. Nat. Prod., 60, 108–110 (1997).
- Le T. N., Gang S. G., Cho W. J., *Tetrahedron Lett.*, 45, 2763—2766 (2004).
- 9) Le T. N., Cho W. J., Bull. Korean Chem. Soc., 27, 2093-2096 (2006).
- 10) Cho W. J., Le Q. M., Van H. T. M., Lee K. Y., Kang B. Y., Lee E. S., Lee S. K., Kwon Y., *Bioorg. Med. Chem. Lett.*, **17**, 3531–3534 (2007).
- Van H. T. M., Le Q. M., Lee K. Y., Lee E. S., Kwon Y., Kim T. S., Le T. N., Lee S. H., Cho W. J., *Bioorg. Med. Chem. Lett.*, **17**, 3531–3534 (2007).
- 12) Le T. N., Cho W. J., Chem. Pharm. Bull., 54, 476-480 (2006).
- 13) Le T. N., Gang S. G., Cho W. J., J. Org. Chem., 69, 2768–2772 (2004).
- 14) Fukuyama Y., Yaso H., Mori T., Takahashi H., Minami H., Kodama M., *Heterocycles*, 54, 259–274 (2001).
- Charlton J. L., Alauddin M. M., J. Org. Chem., 51, 3490–3493 (1986).
- 16) Dalcanale E., Montanari F., J. Org. Chem., 51, 567-569 (1986).
- 17) Clark R. D., Jahangir, J. Org. Chem., 53, 2378–2381 (1988).