

## Highly Oxygenated Lanostane Triterpenoids from the Fungus *Ganoderma applanatum*

Fei WANG and Ji-Kai LIU\*

State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy of Sciences, Kunming 650204, China.

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**Two new highly oxygenated lanostane triterpenoids, ganoderic acid AP2 (1) and ganoderic acid AP3 (2), were isolated from the fruiting bodies of the fungus *Ganoderma applanatum* (Ganodermataceae), along with four known analogues, ganoderenic acids A, B, D and G (3—6). The structures of the new compounds were elucidated on the basis of extensive spectroscopic analysis.**

**Key words** *Ganoderma applanatum*; ganoderic acid; lanostane triterpenoid

The fungus *Ganoderma lucidum* is a well-known Chinese crude drug that has been used clinically in China, Japan and Korea for a long time. More than 140 highly oxygenated lanostane-type triterpenoids have been isolated from the fruiting bodies, mycelia and spores of *G. lucidum*, some of them exhibiting a very broad spectrum of biological activities and pharmacological functions.<sup>1)</sup> Other *Ganoderma* spp. have also been used in traditional medicines for the treatment of cancer, hypertension, chronic bronchitis, diabetes, and arteriosclerosis and as a tonic or sedative. In the case of *G. applanatum* (= *Elfvigia applanata*), some highly oxygenated lanostane triterpenoids,<sup>2–6)</sup> such as ganoderenic acids, ganoderic acids, applanoxidic acids and elfvingic acids, were also isolated in addition to several meroterpenoids.<sup>7,8)</sup> In a continuation of the studies on the bioactive principles of higher fungi from China, we have conducted a further chemical study on *G. applanatum*. Two new highly oxygenated lanostane triterpenoids, ganoderic acid AP2 (1) and ganoderic acid AP3 (2), were isolated from the fruiting bodies along with four known analogues, ganoderenic acids A, B, D and G (3—6). We describe here the isolation and structure elucidation of the new triterpene acids.

Compound 1, obtained as colorless, amorphous powder, has a molecule formula of C<sub>34</sub>H<sub>50</sub>O<sub>8</sub> based on the positive-ion HR-ESI-MS, showing a quasi-molecular ion peak at *m/z* 587.3573 [M+H]<sup>+</sup> (Calcd for C<sub>34</sub>H<sub>51</sub>O<sub>8</sub>, 587.3583), and the <sup>13</sup>C-NMR (DEPT) spectrum. The IR spectrum showed the

absorption bands of hydroxyl (3434 cm<sup>-1</sup>), ester or/and carboxylic carbonyl (1742, 1726 cm<sup>-1</sup>),  $\alpha,\beta$ -unsaturated ketone carbonyl (1678 cm<sup>-1</sup>) as well as double bonds (1643 cm<sup>-1</sup>) groups. The <sup>13</sup>C-NMR spectrum (Table 1) exhibited 34 carbon signals, containing  $\alpha,\beta$ -unsaturated carboxylic carbonyl and ketone carbonyl resonances at  $\delta$  172.7 (s), 191.6 (s), two double bonds at  $\delta$  127.2 (s), 140.1 (s), 144.2 (d), 161.5 (s), three oxygen-bearing methine carbons at  $\delta$  74.6 (d), 78.5 (d), 80.1 (d), as well as characteristic signals at  $\delta$  170.5 (s), 21.0 (q); 170.6 (s), 21.3 (q) due to two acetoxy groups. The following signals in the <sup>1</sup>H-NMR spectrum (Table 1) were easily distinguishable: an olefinic triplet at  $\delta$  6.84 (br t, *J*=7.0 Hz), three oxygenated methine protons at  $\delta$  5.61 (s), 5.20 (dd, *J*=8.8, 6.6 Hz), 3.24 (dd, *J*=11.0, 5.1 Hz), two acetoxy methyl singlets at  $\delta$  2.19 (s), 2.11 (s), a vinyl methyl broad singlet at  $\delta$  1.83 (br s), together with five tertiary methyls and two secondary ones at the upfield zone. The above-mentioned NMR data suggested that compound 1 was an acetylated lanostane-type triterpene acid, and this conclusion was also in accordance with the previous study in which the fungus was considered a rich source of lanostane derivatives.<sup>2–6)</sup>

The signal at  $\delta$ <sub>H</sub> 3.24 (dd, *J*=11.0, 5.1 Hz) was typical for the H-3 $\alpha$  in the overwhelming triterpenoids, and the observed HMBC correlations (Fig. 2) from this proton to two tertiary methyl carbons  $\delta$ <sub>C</sub> 15.7 (q, C-29), 28.2 (q, C-28) further indicated the presence of a hydroxyl at C-3. The olefinic triplet at  $\delta$ <sub>H</sub> 6.84 and the vinyl methyl broad singlet at  $\delta$ <sub>H</sub> 1.83 all correlated with the carboxylic carbon  $\delta$ <sub>C</sub> 172.7 in the HMBC spectrum, suggesting that  $\alpha,\beta$ -unsaturated carboxyl group must be located at the bottom of the side chain. The important HMBC correlations: from  $\delta$ <sub>H</sub> 1.14 (s, Me-19) to  $\delta$ <sub>C</sub> 140.1 (s, C-9), from  $\delta$ <sub>H</sub> 1.41 (s, Me-30) to the downfield olefinic carbon  $\delta$ <sub>C</sub> 161.5 (s, C-8) were detected, consequently the other double bond and ketone carbonyl were doubtless emplaced at C-8 and C-11 (not at C-8 and C-7), respectively. The downfield oxygenated methine singlet proton at  $\delta$  5.61 was necessarily attached with an acetoxy and assigned to H-12, considering the appearance of the HMBC correlations from the proton to  $\delta$ <sub>C</sub> 140.1 (s, C-9) and ketone carbon resonance  $\delta$ <sub>C</sub> 191.6 (s, C-11). Likewise, the remaining acetoxy group was easily located at C-15 by the same means. The stereochemistry of 1 was deduced as 3 $\beta$ -hydroxy, 12 $\beta$ -acetoxy and 15 $\alpha$ -acetoxy by the careful analysis of the ROESY spectrum (Fig. 3). The configuration of the double bond at C-

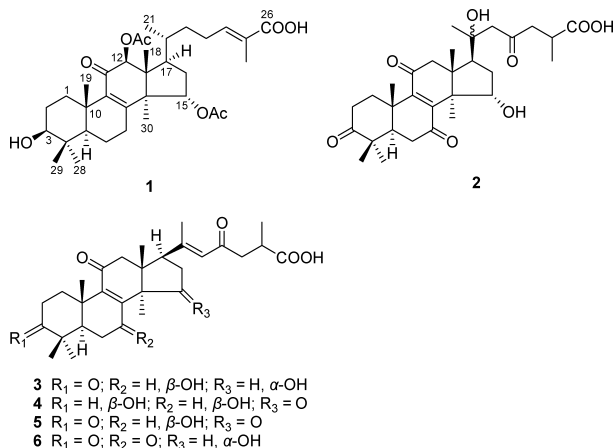


Fig. 1. Structures of Compounds 1—6 from *G. applanatum*

\* To whom correspondence should be addressed. e-mail: jkliu@mail.kib.ac.cn

Table 1. NMR Spectral Data for Compounds **1** and **2** in CDCl<sub>3</sub>

No.	Ganoderic acid AP2 ( <b>1</b> )		Ganoderic acid AP3 ( <b>2</b> )	
	$\delta_C$	$\delta_H$	$\delta_C$	$\delta_H$
1	33.8 (t)	1.02 (m, H $_{\alpha}$ ) 2.80 (br d, 13.6, H $_{\beta}$ )	35.0 (t)	1.81 (ddd, 14.2, 9.3, 6.6, H $_{\alpha}$ ) 2.97 (m, H $_{\beta}$ )
2	27.5 (t)	1.61 (m); 1.66 (m)	33.9 (t)	2.52 (m, H $_{\beta}$ ); 2.62 (m, H $_{\alpha}$ )
3	78.5 (d)	3.24 (dd, 11.0, 5.1)	215.4 (s)	
4	38.8 (s)		46.4 (s)	
5	51.3 (d)	0.88 (br d, 11.7)	49.0 (d)	2.27 (dd, 15.2, 2.4)
6	17.3 (t)	1.46 (m, H $_{\beta}$ ); 1.74 (m, H $_{\alpha}$ )	36.8 (t)	2.47 (m, H $_{\alpha}$ ); 2.63 (m, H $_{\beta}$ )
7	29.3 (t)	2.19 (m, H $_{\beta}$ ); 2.29 (m, H $_{\alpha}$ )	204.6 (s)	
8	161.5 (s)		150.4 (s)	
9	140.1 (s)		152.3 (s)	
10	37.5 (s)		39.1 (s)	
11	191.6 (s)		201.2 (s)	
12	80.1 (d)	5.61 (s)	52.1 (t)	2.69 (d, 17.0, H $_{\beta}$ ) 2.86 (d, 17.0, H $_{\alpha}$ )
13	51.5 (s)		47.8 (s)	
14	53.9 (s)		52.8 (s)	
15	74.6 (d)	5.20 (dd, 8.8, 6.6)	71.8 (d)	4.35 (dd, 10.0, 5.6)
16	33.6 (t)	1.74 (m); 2.26 (m)	30.2 (t)	1.67 (ddd, 14.2, 9.8, 5.6, H $_{\alpha}$ ) 2.55 (m, H $_{\beta}$ )
17	48.6 (d)	2.20 (m)	50.9 (d)	2.11 (dd, 9.8, 9.8)
18	12.3 (q)	0.92 (s)	18.9 (q)	1.05 (s)
19	19.0 (q)	1.14 (s)	17.5 (q)	1.26 (s)
20	34.1 (d)	1.49 (m)	73.3 (s)	4.81 (br s, OH)
21	19.7 (q)	0.94 (d, 6.6)	26.6 (q)	1.30 (s)
22	33.7 (t)	1.17 (m); 1.58 (m)	52.5 (t)	2.56 (d, 16.3); 2.64 (d, 16.3)
23	26.3 (t)	2.11 (m); 2.24 (m)	211.2 (s)	
24	144.2 (d)	6.84 (br t, 7.0)	47.7 (t)	2.46 (m); 2.89 (m)
25	127.2 (s)		34.5 (d)	2.96 (m)
26	172.7 (s)		177.8 (s)	
27	12.0 (q)	1.83 (br s)	17.0 (q)	1.23 (d, 6.8)
28	28.2 (q)	1.03 (s)	27.2 (q)	1.15 (s)
29	15.7 (q)	0.82 (s)	20.2 (q)	1.12 (s)
30	19.8 (q)	1.41 (s)	20.6 (q)	1.18 (s)
COCH <sub>3</sub> (12)	170.5 (s)			
COCH <sub>3</sub> (12)	21.0 (q)	2.19 (s)		
COCH <sub>3</sub> (15)	170.6 (s)			
COCH <sub>3</sub> (15)	21.3 (q)	2.11 (s)		

The unambiguous assignments were made on the basis of HSQC, HMBC and ROESY experiments.

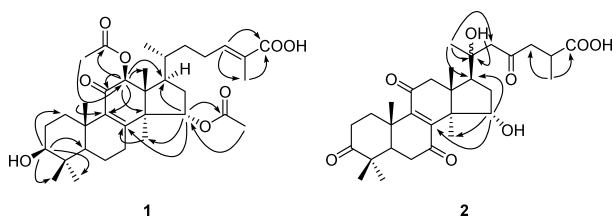


Fig. 2. Selected HMBC Correlations of Compounds **1** and **2**

24 was assigned to be *E*-orientation because of Me-27 characteristic upfield resonance at  $\delta_C$  12.0 (q). Therefore, the structure of **1** was elucidated as 12 $\beta$ ,15 $\alpha$ -diacetoxy-3 $\beta$ -hydroxy-11-oxolanost-8,24(*E*)-dien-26-oic acid, named ganoderic acid AP2.

Compound **2**, also obtained as colorless, amorphous powder, has a molecule formula of C<sub>30</sub>H<sub>42</sub>O<sub>8</sub> based on the negative-ion HR-ESI-MS, showing a quasi-molecular ion peak at *m/z* 529.2786 [M-H]<sup>-</sup> (Calcd for C<sub>30</sub>H<sub>41</sub>O<sub>8</sub>, 529.2801), and the <sup>13</sup>C-NMR (DEPT) spectrum. The IR spectrum showed the absorption bands of hydroxyl (3444 cm<sup>-1</sup>) and unsaturated carbonyl (1709, 1670 cm<sup>-1</sup>) groups. The <sup>1</sup>H- and <sup>13</sup>C-NMR spectroscopic character (Table 1) suggested that com-

pound **2** possessed a skeleton of lanostane triterpene acid, and the NMR signals were similar to those of ganoderic acid G (**6**),<sup>2)</sup> also isolated from this fungus. Nevertheless, there was a remarkable difference as follows: the signals at  $\delta_C$  155.6 (s, C-20), 124.5 (d, C-22), 197.4 (s, C-23) due to the unsaturated ketone group at the side chain of **6** were absent and replaced by a set of newly arisen signals at  $\delta_C$  73.3 (s, C-20), 52.5 (t, C-22), 211.2 (s, C-23) in the <sup>13</sup>C-NMR spectra of **2**, indicating that this double bond was hydrogenated and substituted by hydroxyl at C-20, which was further supported by the fact that the HMBC correlations from the broad hydroxyl singlet  $\delta_H$  4.81 to  $\delta_C$  73.3 (s, C-20), from the methyl singlet  $\delta_H$  1.30 (s, Me-20) to  $\delta_C$  52.5 (t, C-22) were clearly detectable. The stereochemistry of 15-hydroxy was deduced as  $\alpha$ -orientation by the observable ROESY correlation (Fig. 3) between H-15 and Me-18. Accordingly the structure of **2** was characterized as 15 $\alpha$ ,20 $\xi$ -dihydroxy-3,7,11,23-tetraoxolanost-8-en-26-oic acid, *i.e.*, ganoderic acid AP3. Complete assignments (Table 1) of the <sup>1</sup>H-NMR signals were unambiguously performed by careful analysis of HSQC, HMBC and ROESY experiments.

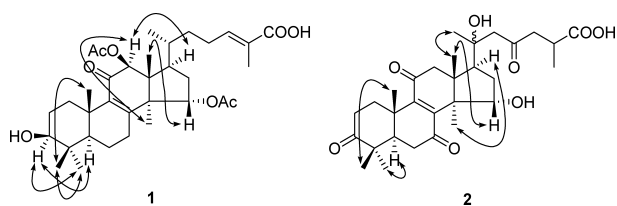


Fig. 3. Significant ROESY Correlations of Compounds 1 and 2

## Experimental

**General Experimental Procedures** Melting points were measured on a PHMK 79/2289 micro-melting point apparatus and uncorrected. Optical rotations were measured on a Horiba SEPA-300 polarimeter. UV spectra were recorded on a Shimadzu UV-2401PC spectrophotometer. IR spectra were obtained using a Bruker Tensor 27 FT-IR spectrometer with KBr pellets. NMR spectra were acquired with a Bruker DRX-500 spectrometer in  $\text{CDCl}_3$  at room temperature, and chemical shifts were referred to TMS as internal standard. EI-MS were taken on a Finnigan-MAT 90 instrument, and ESI-MS and HR-ESI-MS were recorded with an API QSTAR Pulsar i spectrometer.

Column chromatography was performed using silica gel (200–300 mesh, Qingdao Marine Chemical Inc., China) and Chromatorex C-18 (40–75  $\mu\text{m}$ , Fuji Silysia Chemical Ltd., Japan). Fractions were monitored by Agilent 1100 reversed-phase HPLC (Zorbax SB-C-18 column, 5  $\mu\text{m}$ , 4.6  $\times$  150 mm, 30–100%  $\text{CH}_3\text{CN}$  in  $\text{H}_2\text{O}$  over 15 min, 1 ml/min).

**Fungus Material** The fresh fruiting bodies of *G. applanatum* were collected at the Gaoligong Mountains in Yunnan Province, China, in August 2006. The voucher specimen was deposited in the Herbarium of Kunming Institute of Botany, Chinese Academy of Sciences.

**Extraction and Isolation** The fresh fruiting bodies (3.0 kg) were immersed in 12 l  $\text{CHCl}_3$ -MeOH (1/1, v/v) and left at room temperature for two weeks. Then the extraction was concentrated *in vacuo* to give a brown gum (70.0 g), which was fractionated by silica gel column using  $\text{CHCl}_3$ -MeOH gradient elution.

The fractions (5.0 g) eluted with 1–2% MeOH, mainly containing the triterpene acids, were further repeatedly separated by low pressure Chromatorex C-18 column chromatography. Purification of these fractions using different gradients afforded the new triterpenes, ganoderic acid AP2 (1; 38 mg;

80% MeOH in  $\text{H}_2\text{O}$ ;  $t_R$  10.3 min) and ganoderic acid AP3 (2; 25 mg; 50% MeOH;  $t_R$  6.2 min), and four known analogues, ganoderic acid A (3; 45 mg; 45% MeOH;  $t_R$  5.9 min), ganoderic acid B (4; 120 mg; 40% MeOH;  $t_R$  5.5 min), ganoderic acid D (5; 350 mg; 58% MeOH;  $t_R$  7.1 min) and ganoderic acid G (6; 65 mg; 60% MeOH;  $t_R$  7.7 min).

**Ganoderic Acid AP2 (1):** Colorless, amorphous powder. mp 121–122 °C.  $[\alpha]_D^{20} +84.3^\circ$  ( $c=0.18$ ,  $\text{CHCl}_3$ ). UV  $\lambda_{\text{max}}$  ( $\text{CHCl}_3$ ) nm (log  $\epsilon$ ): 253 (3.90). IR (KBr)  $\text{cm}^{-1}$ : 3434, 2956, 2932, 2874, 1742, 1726, 1678, 1643, 1380, 1237, 1035.  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR data: see Table 1. EI-MS  $m/z$ : 586 [ $\text{M}$ ] $^+$  (9), 544 (4), 527 (9), 526 (10), 508 (4), 484 (11), 466 (17), 448 (13), 362 (13), 291 (43), 277 (100), 274 (37), 264 (35), 213 (217), 193 (80); HR-ESI-MS (pos.): 587.3573 [ $\text{M}+\text{H}$ ] $^+$  (Calcd for  $\text{C}_{34}\text{H}_{51}\text{O}_8$ : 587.3583).

**Ganoderic Acid AP3 (2):** Colorless, amorphous powder. mp 118–119 °C.  $[\alpha]_D^{15} +141.9^\circ$  ( $c=0.35$ ,  $\text{CHCl}_3$ ). UV  $\lambda_{\text{max}}$  ( $\text{CHCl}_3$ ) nm (log  $\epsilon$ ): 263 (3.76). IR (KBr)  $\text{cm}^{-1}$ : 3444, 2956, 2922, 2852, 1709, 1670, 1464, 1389, 1233, 1177, 1054.  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR data: see Table 1. EI-MS  $m/z$ : 400 (12), 382 (43), 367 (10), 364 (17), 340 (70), 339 (100), 321 (13), 311 (13), 300 (57), 301 (57); ESI-MS (neg.): 529 [ $\text{M}-\text{H}$ ] $^-$ , 399, 129; HR-ESI-MS (neg.): 529.2786 [ $\text{M}-\text{H}$ ] $^-$  (Calcd for  $\text{C}_{30}\text{H}_{41}\text{O}_8$ : 529.2801).

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