Synthesis and Biological Evaluation of 4-Substituted Vitamin D and 14-Epi-Previtamin D Analogs

Daisuke Sawada,^{*a*} Yuya Tsukuda,^{*a*} Hiroshi Sarto,^{*b*} Ken-ichiro Takagi,^{*b*} Kyouhei Horie,^{*b*} Eiji Ochiai,^{*b*} Kazuya Takenouchi,^{*b*} and Atsushi Kittaka^{*,*a*}

^a Faculty of Pharmaceutical Sciences, Teikyo University; Sagamiko, Kanagawa 229–0195, Japan: and ^b Teijin Institute for Bio-medical Research, Teijin Pharma Ltd.; Tokyo 191–8512, Japan. Received September 4, 2009; accepted September 30, 2009; published online October 2, 2009

We synthesized the 4-hydroxy and 4-methoxy analogs of active vitamin D_3 (1 α ,25(OH)₂ D_3 , 1) and its C14epimer with the previtamin D_3 form of 14-epi-1 α ,25(OH)₂pre D_3 (14-epi-pre1). Their vitamin D receptor (VDR) binding affinity and osteocalcin promoter transactivation activity in HOS cells were evaluated, and had lower activity than the natural hormone (1) and 14-epi-pre1, respectively.

Key words vitamin D₃; previtamin D₃; C4-modified seco-steroid; vitamin D receptor; osteocalcin

It is well-known that 1α ,25-dihydroxyvitamin D₃ (1α ,25(OH)₂D₃: 1), which is the biologically most active metabolite of vitamin D₃, plays an important role in the body as a hormone, such as in calcium and phosphorus homeostasis, cellular differentiation, and immune responses, and inevitably, is involved in several diseases.¹⁻⁸ Therefore, its analogs could be good candidates for therapeutic agents, and so far, a large number of derivatives have been synthesized. Vitamin D analogs, such as alphacalcidol, doxercalciferol, paricalcitol, calcipotriol, tacalcitol, and maxacalcitol, are clinically used for osteoporosis, renal failure, secondary hyperparathyroidism, and psoriasis patients.^{1,9}

One of our previous research projects investigated previtamin D₃ analogs¹⁰⁾ and we became interested in modification at the 4-position of 14-epi-previtamin D₃. Among the huge library of vitamin D₃ derivatives, few compounds possess substituents at the 4-position, and their biological effects remain unclear, except 4-fluorovitamin D analogs.^{11,12} We expected that 4-oxy-substitution of the previtamin D form, especially, could construct a new hydrogen bond in vitamin D receptor (VDR), and also 4-substitution would help us to understand the genomic activity of the pre-form compounds.¹⁰ At the beginning of the study of 4-substitution in vitamin D_{2} , we report the synthesis and biological evaluation of 4-hydroxy and 4-methoxy analogs of 14-epi-1 α ,25-dihydroxyprevitamin D_3 (14-epi-1 α ,25(OH)₂preD₃, 14-epi-pre1) and also of 1α , 25(OH)₂D₃ (1), that is, compounds 3 and 2, respectively (Fig. 1).

The target compounds were divided into two fragments, A-ring and CD-ring parts, which were individually synthesized and coupled in a later step. The synthesis of the A-ring fragment is shown in Chart 1. The hydroxy group of the known compound 4 was protected as a tert-butyldimethylsilvl (TBS) ether, and transformed into a bromide using Nbromosuccinimide (NBS) and BaCO₃.¹³⁾ Then, it reacted with activated zinc powder in 1-PrOH/H2O to give aldehyde 5. The addition reaction of lithium (trimethylsilyl)acetylide to 5 proceeded smoothly to afford alcohol 6a, a mixture of diastereomers (major/minor 3/1). After the addition reaction, subsequent in situ reaction of the resultant lithium alkoxide with MeI afforded methoxy compound 6b as a mixture of diastereomers (major/minor 3/1). These hydroxy or methoxy groups of 6a and 6b corresponded to the 4-position in the Aring (steroid numbering), that is, we were able to construct the 4-substitution. Next, deprotection of both trimethylsilyl (TMS) and Bz groups in 6a and 6b was conducted by K₂CO₃ in MeOH, and the following protection of the resultant hydroxyls by TBS groups gave the A-ring fragments 7a and 7b, respectively.



Fig. 1. Structures of Vitamin D₃ and Previtamin D₃ Analogs



* To whom correspondence should be addressed. e-mail: akittaka@pharm.teikyo-u.ac.jp

Stereochemistry at the 4-position of 6a and 6b was determined by using Mosher esters as follows¹⁴: Diastereomers of compound 6a were separated by HPLC, and both reacted with (R)- and (S)- α -methoxy- α -trifluoromethylphenylacetyl chloride (MTPA chloride) to give the corresponding (S)and (R)-MTPA esters (6a-MTPAs), respectively (Fig. 2). Using the four esters, the values of $\Delta \delta$ (= $\delta_{(S)-MTPA ester}$ - $\delta_{(R)-\text{MTPA ester}}$ in the 400 MHz ¹H-NMR spectra were calculated, and the configurations at C4 were determined by applying the modified Mosher's method¹⁴; therefore, we decided the absolute configurations at C4 of the major isomer of 6a and **6b** were *R* configuration.



Fig. 2. Determination of the Absolute Configuration of 6a-MTPAs (Major and Minor)

As in our preceding paper,¹⁰⁾ we transformed $7a^{15)}$ to 9aand $7b^{15}$ to 9b, to try to connect with the CD-ring fragment (8), which was prepared from vitamin D_3 , to obtain the 14epi-1 α ,25(OH)₂preD₃ analogs (10a, b) under basic conditions (Chart 2); however, we could not obtain any coupled products 10a, b due to steric hindrance of the substituent at the 4-position of 9. Therefore, we used Trost's procedure for the coupling reaction.¹⁶⁾



TBSC



The desired CD-ring fragment 11 was synthesized using 8^{10} by the Wittig reaction (Chart 3)^{16,17)} and the coupling reaction with 7a or 7b was conducted under Trost's coupling conditions (Chart 4). Fortunately, the reaction was straightforward, and the desired coupled compounds 12a and 12b were obtained, respectively. Subsequent treatment with tetrabutylammonium fluoride (TBAF) afforded the deprotected compounds, which were then heated at 80 °C in benzene to promote [1,7]-sigmatropic rearrangement. Finally, we could obtain the 4-hydroxy (3a) and 4-methoxy (3b) analogs of 14epi-1 α ,25(OH)₂preD₃.¹⁸⁾ The major diastereomers at C4 of compounds **3a** and **3b** were purified by HPLC to give $3aR^{19}$ and 3bR,²⁰⁾ respectively, and were used for biological evaluation.



The similar coupling reaction between 7a and the known CD-ring fragment 14 was applied to synthesize compound 2a (Chart 5).¹⁶⁾ The reaction gave the coupled product, whose silvl groups were removed by TBAF to afford the desired compound 2a. The major diastereomer at C4 of this compound was purified by HPLC to give $1\alpha, 4\beta, 25$ -trihydroxyvitamin D_3 (**2a** R^{21}), which was used for biological testing.

VDR binding affinity and osteocalcin promoter transactivation activity of the new compounds were evaluated using the chick intestinal or human VDR, and human osteoblast cell line (HOS) cells, respectively. The results are summarized in Table 1 in comparison with the natural hormone 1 and 14-epi-pre1, which was prepared in our laboratory.¹⁰ The new compounds had lower activity than 1, and in terms of the binding affinity for VDR, compounds 3aR and 3bR showed almost the same activity as 14-epi-pre1. As above, the 4-substitution of vitamin D₃ and 14-epi-previtamin D₃ was thought to be ineffective for VDR binding affinity and osteocalcin promoter transactivation activity.

In conclusion, we synthesized the 4-hydroxy and 4methoxy analogs of 1α , 25(OH)₂D₃ (1) and 14-epi-

`он



Chart 5

юн

Table 1. Relative Binding Affinity for VDR and Osteocalcin Promoter Transactivation Activity in HOS Cells of 1, 14-epi-pre1, and New Compounds

Compound	VDR ^{<i>a,b,c</i>)}	Osteocalcin transactivation activity (ED ₅₀ (nM))
1	$100^{b,c)}$	0.03
2aR	2.9^{c}	0.22
14-epi-pre1	$0.5^{b)}$	0.46
3aR	$0.1^{b)}$	20.5
3b <i>R</i>	$0.95^{b)}$	40.5

a) The potency of **1** is normalized to 100. b) Chick intesinal VDR was used. c) Human VDR was used.



 1α ,25(OH)₂preD₃ (**14-epi-pre1**) for the first time. We evaluated VDR binding affinity and osteocalcin promoter transactivation activity in HOS cells. Unfortunately, the biological activities were not improved by 4-substitution; however, the information on new compounds will be useful to create novel potential ligands for VDR.

Acknowledgements We are grateful to Ms. Junko Shimode and Ms. Ayako Kawaji (Teikyo University) for the spectroscopic measurements. This work was supported in part by a Grant-in-Aid from the Ministry of Education, Culture, Sports, Science and Technology, Japan (to D.S.) and by a Grant-in-Aid from the Japan Society for the Promotion of Science (to A.K.).

References and Notes

- For new vitamin D analogs (drugs and drug candidates), see: Posner G. H., Kahraman M., "Vitamin D," 2nd ed., Elsevier Academic Press, New York, 2005, pp. 1405—1422.
- For regulation of immune responses, see: Adorini L., "Vitamin D," 2nd ed., Elsevier Academic Press, New York, 2005, pp. 631—648.
- For mechanism of action, see: Kato S., Fujiki R., Kitagawa H., "Vitamin D," 2nd ed., Elsevier Academic Press, New York, 2005, pp. 305– 312.
- 4) DeLuca H. F., *Nutrition Rev.*, **66** (Suppl. 2), S73—S87 (2008).
- Brown A. J., Slatopolsky E., *Mol. Aspects Med.*, **29**, 433–452 (2008).
 Bouillon R., Okamura W. H., Norman A. W., *Endocr. Rev.*, **16**, 200–
- 257 (1995).
 7) Zhu G. D. Okamura W. H., *Chem. Rev.* 95, 1877—1952 (1995).
- Zhu G. D., Okamura W. H., *Chem. Rev.*, **95**, 1877–1952 (1995).
 Ettinger R. A., DeLuca H. F., *Adv. Drug Res.*, **28**, 269–312 (1996)
- Winger R. A., Deluca H. I., Auv. Drug Res., 26, 209—512 (1990).
 Kubodera N., *Heterocycles*, in press, DOI: 10.3987/REV-09-SR(S)3.

- Sawada D., Katayama T., Tsukuda Y., Saito N., Takano M., Saito H., Takagi K., Ochiai E., Ishizuka S., Takenouchi K., Kittaka A., *Bioorg. Med. Chem. Lett.*, **19**, 5397–5400 (2009), and references cited therein.
- Shimizu M., Iwasaki Y., Yamada S., *Tetrahedron Lett.*, 40, 1697– 1700 (1999).
- Ohno A., Shimizu M., Yamada S., Chem. Pharm. Bull., 50, 475–483 (2002).
- Saito N., Masuda M., Saito H., Takenouchi K., Ishizuka S., Namekawa J., Takimoto-Kamimura M., Kittaka A., *Synthesis*, 2005, 2533–2543 (2005).
- 14) Ohtani I., Kusumi T., Kashman Y., Kakisawa H., J. Am. Chem. Soc., 113, 4092—4096 (1991).
- 15) Enynes **7a** and **7b** were used as diastereomixtures, since it was difficult to separate each diastereomer at this stage without HPLC, respectively.
- 16) Trost B. M., Dumas J., Villa M., J. Am. Chem. Soc., 114, 9836—9845 (1992).
- Deprotection of the TES group was essential for easy separation of the coupled compound from the starting materials in the next reaction (Chart 5).
- Spectroscopic data of 14-epi-1α,25(OH)₂preD₃, see: Maynard D. F., Trankle W. G., Norman A. W., Okamura W. H., *J. Med. Chem.*, 37, 2387–2393 (1994).
- In ¹H-NMR analyses, there were no 6-s-trans signals. Therefore, we estimated the ratio of the two isomers 6-cis/6-s-trans as 95/<5. Spectroscopic data for **3aR**: [α]_D¹⁹ -7.8 (c=0.023, CHCl₃); UV (EtOH) λ_{max} 250.0 nm, λ_{min} 228.0 nm; IR (neat) 3379, 1466, 1377, 1215 cm⁻¹; ¹H-NMR (400 MHz, CDCl₃) δ: 0.86–1.56 (m, 33H), 3.30–3.34 (m, 3H), 3.62–3.66 (m, 1H), 4.03–4.10 (m, 1H), 4.22 (d, J=5.6 Hz, 2H), 4.49 (d, J=2.4 Hz, 1H), 5.73 (t, J=3.7 Hz, 1H), 5.92 (d, J=5.0 Hz, 2H); ¹³C-NMR (100 MHz, CDCl₃) δ: 13.9, 19.5, 19.8, 20.5, 21.7, 29.0, 29.3, 29.4, 29.4, 29.8, 34.1, 44.3, 44.4, 45.6, 51.6, 53.4, 68.3, 69.5, 71.1, 71.6, 77.1, 77.2, 77.5, 100.5, 119.5, 126.0, 132.3; EI-HR-MS Calcd for C₂₇H₄₄O₄ [M++Na]⁺ 455.3132, Found 455.3139.
- 20) In ¹H-NMR analyses, there were no 6-*s*-*trans* signals. Therefore, we estimated the ratio of the two isomers 6-*cis/*6-*s*-*trans* as 95/<5. Spectroscopic data for **3b***R*: [α]_D¹⁹ -142.5 (*c*=0.07, CHCl₃); UV (EtOH) λ_{max} 250.5 nm, λ_{min} 228.0 nm; IR (neat) 3387, 1466, 1377, 1240, 1213 cm⁻¹; ¹H-NMR (400 MHz, CDCl₃) δ: 0.89 (s, 3H), 0.93 (d, *J*=6.4 Hz, 3H), 1.04—1.08 (m, 1H), 1.18—2.08 (m, 28H), 2.16 (s, 2H), 3.47 (s, 3H), 3.67 (d, *J*=6.4 Hz, 1H), 4.03—4.12 (m, 2H), 5.69 (s, 1H), 5.86 (d, *J*=12.5 Hz, 1H), 5.98 (d, *J*=12.5 Hz, 1H); ¹³C-NMR (100 MHz, CDCl₃) δ: 17.2, 19.6, 19.8, 21.6, 21.9, 23.1, 29.1, 29.6, 33.9, 34.1, 35.8, 40.9, 44.4, 45.8, 52.0, 54.8, 56.4, 59.9, 65.6, 66.5, 69.1, 69.9, 71.0, 126.1, 127.4, 133.2, 137.1, 138.0; E1-HR-MS Calcd for C₂₇H₄Q₄ [M+Na]⁺ 469.3288, Found 469.3274.
- 21) Spectroscopic data for **2a***R*: $[\alpha]_D^{21} 87.3$ (c=0.01, CHCl₃); UV (EtOH) λ_{max} 266.0 nm, λ_{min} 212.5 nm; IR (neat) 3383, 1215 cm⁻¹; ¹H-NMR (400 MHz, CDCl₃) δ : 0.93 (d, J=6.4 Hz, 3H), 1.19—2.17 (m, 31H), 2.87 (d, J=12.0 Hz, 1H), 3.72 (ddd, J=7.1, 7.1, 14.2 Hz, 2H), 4.19 (s, 1H), 4.27 (s, 1H), 4.42 (d, J=6.1 Hz, 1H), 5.0 (d, J=2.0 Hz, 1H), 5.4 (s, 1H), 6.02 (d, J=11.7 Hz, 1H), 6.66 (d, J=11.7 Hz, 1H); ¹³C-NMR (100 MHz, CDCl₃) δ : 17.5, 19.8, 20.0, 20.1, 20.8, 22.0, 23.3, 29.1, 29.3, 29.6, 29.7, 30.1, 34.3, 36.1, 38.2, 41.5, 44.7, 45.9, 51.8, 51.9, 52.6, 69.9, 71.5, 77.5, 126.0, 126.3, 134.7, 138.7; EI-HR-MS Calcd for C₂₇H₄₄O₄ [M+Na]⁺ 455.3132, Found 455.3113.