

## Briaviodiol A, a New Cembranoid from a Soft Coral *Briareum violacea*

Yu-Chia CHANG,<sup>a,b,c</sup> I-Chu HUANG,<sup>c</sup> Michael Yen-Nan CHIANG,<sup>d</sup> Tsong-Long HWANG,<sup>e</sup>  
Ting-Hsuan KUNG,<sup>c</sup> Chan-Shing LIN,<sup>\*a,f,g</sup> Jyh-Horng SHEU,<sup>f,g</sup> and Ping-Jyun SUNG<sup>\*b,c,f,g</sup>

<sup>a</sup> Doctoral Degree Program in Marine Biotechnology, National Sun Yat-sen University and Academia Sinica; <sup>d</sup> Department of Chemistry, National Sun Yat-sen University; <sup>f</sup> Department of Marine Biotechnology and Resources, National Sun Yat-sen University; <sup>g</sup> Asia-Pacific Ocean Research Center, National Sun Yat-sen University; Kaohsiung 804, Taiwan; <sup>b</sup> Graduate Institute of Marine Biotechnology, Department of Life Science and the Institute of Biotechnology, National Dong Hwa University; Pingtung 944, Taiwan; <sup>c</sup> National Museum of Marine Biology and Aquarium; Pingtung 944, Taiwan; and <sup>e</sup> Graduate Institute of Natural Products, Chang Gung University; Taoyuan 333, Taiwan.

Received August 19, 2010; accepted September 15, 2010; published online September 16, 2010

**A new cembrane diterpenoid, briaviodiol A (1), has been isolated from a soft coral *Briareum violacea*. The structure of 1 was determined by spectroscopic methods and further confirmed by a single-crystal X-ray diffraction analysis.**

**Key words** *Briareum violacea*; cembrane; briaviodiol; diterpenoid

Previous studies on the chemical constituents from the octocoral *Briareum violacea* (QUOY and GAIMARD 1883, family Briareidae) (formerly known as *Pachyclavularia violacea*) have yielded a series of interesting natural products including cembrane,<sup>1–5</sup> briarane,<sup>3,6,7</sup> briarellin,<sup>8–10</sup> and secosterol analogues.<sup>10</sup> Recently, our chemical examination on the soft coral *B. violacea* has resulted in the isolation of a new cembranoid designated as briaviodiol A (**1**). The structure of **1** was elucidated by spectroscopic methods and further confirmed by a single-crystal X-ray diffraction analysis.

### Results and Discussion

Briaviodiol A (**1**) was obtained as a white powder and could be recrystallized from methanol to form colorless prisms. The molecular formula of **1** was established as C<sub>21</sub>H<sub>32</sub>O<sub>6</sub> (six degrees of unsaturation) from a sodiated molecule at *m/z* 403 in the electrospray ionization (ESI)-MS spectrum and further supported by high resolution (HR)-ESI-MS (*m/z* 403.2093, Calcd 403.2096, [C<sub>21</sub>H<sub>32</sub>O<sub>6</sub>Na]<sup>+</sup>). The IR spectrum of **1** showed bands at 3411 and 1752 cm<sup>-1</sup>, consistently with the presence of hydroxy and ester groups. From the <sup>13</sup>C-NMR data of **1** (Table 1), a suite of resonances at δ<sub>C</sub> 171.3 (s, C-17), 158.1 (s, C-1), 130.6 (s, C-15), 108.5 (s, C-2), and 10.4 (q, C-16), could be assigned to the α-methyl-γ-butenolide moiety by comparison with the <sup>13</sup>C-NMR data of a known metabolite pachyclavulariolide F.<sup>3</sup> An additional unsaturated functionality was indicated by <sup>13</sup>C-NMR resonances at δ<sub>C</sub> 131.5 (d, C-5) and 129.9 (s, C-4), suggesting the presence of a trisubstituted olefin. On the basis of overall unsaturation data, **1** was concluded to be a cembrane-type diterpenoid molecule possessing three rings.

The <sup>1</sup>H-NMR spectrum of **1** showed the presence of four methyl groups: a doublet (δ<sub>H</sub> 0.86, couple to a methine proton at δ<sub>H</sub> 1.26), a singlet at δ<sub>H</sub> 1.38 representing a methyl group on an oxygenated quaternary carbon, a vinyl methyl (δ<sub>H</sub> 1.76, s), a methyl on the butenolide moiety (δ<sub>H</sub> 2.11, s). The <sup>1</sup>H-NMR coupling information in the <sup>1</sup>H-<sup>1</sup>H correlation spectroscopy (COSY) spectrum of **1** enabled identification of the C-5/-6/-7/-8/-9/-10/-11, C-13/-14, C-8/-19, and C-5/-18 (by allylic coupling) units (Fig. 1), which were assembled the assistance of a heteronuclear multiple bond coherence (HMBC) experiment (Fig. 1). The HMBC correlations be-

Table 1. <sup>1</sup>H- and <sup>13</sup>C-NMR Data for Cembranoid **1**

C/H	δ <sub>H</sub> <sup>a)</sup>	δ <sub>C</sub> <sup>b)</sup>
1		158.1 (s) <sup>d)</sup>
2		108.5 (s)
3a/b	3.02 d (14.0) <sup>c)</sup> ; 2.35 d (14.0)	47.2 (t)
4		129.9 (s)
5	5.53 t (8.0)	131.5 (d)
6a/b	2.14 m; 1.54 m	30.5 (t)
7a/b	1.16 m; 1.80 m	32.5 (t)
8	1.26 m	38.3 (d)
9	3.70 dt (10.0, 6.8)	84.7 (d)
10a/b	1.67 m; 2.00 m	38.2 (t)
11	2.11 m (2H)	24.6 (t)
12		85.4 (s)
13	3.41 dd (10.0, 1.2)	76.2 (d)
14	5.20 dd (4.8, 1.2)	68.4 (d)
15		130.6 (s)
16	2.11 s	10.4 (q)
17		171.3 (s)
18	1.76 s	18.9 (q)
19	0.86 d (6.4)	16.3 (q)
20	1.38 s	23.0 (q)
2-OCH <sub>3</sub>	3.23 s	50.8 (q)
OH-13	2.25 d (10.0)	
OH-14	2.38 d (4.8)	

a) Spectra recorded at 400 MHz in CDCl<sub>3</sub> at 25 °C. b) Spectra recorded at 100 MHz in CDCl<sub>3</sub> at 25 °C. c) *J* values (in Hz) in parentheses. d) Multiplicity deduced by DEPT and indicated by usual symbols.

tween protons and quaternary carbons of **1**, such as H<sub>2</sub>-3, H-14, H<sub>3</sub>-16/C-1; H<sub>2</sub>-3, H-14/C-2; H<sub>2</sub>-3, H<sub>3</sub>-18/C-4; H<sub>2</sub>-11, H-13, H<sub>3</sub>-20/C-12; H-14, H<sub>3</sub>-16/C-15; and H<sub>3</sub>-16/C-17, permitted elucidation of the carbon skeleton. A vinyl methyl at C-4 was confirmed by the allylic coupling between H-5/H<sub>3</sub>-18 in the <sup>1</sup>H-<sup>1</sup>H COSY spectrum and by the HMBC correlations between H<sub>3</sub>-18/C-3, -4, -5; H-5/C-18; and H<sub>2</sub>-3/C-18. Furthermore, the presence of hydroxy groups at C-13 and C-14 were deduced from the <sup>1</sup>H-<sup>1</sup>H COSY correlations between the hydroxy protons (δ<sub>H</sub> 2.25, 1H, d, *J*=10.0 Hz; 2.38, 1H, d, *J*=4.8 Hz) and the oxymethine protons at δ<sub>H</sub> 3.41 (1H, dd, *J*=10.0, 1.2 Hz, H-13) and 5.20 (1H, dd, *J*=4.8, 1.2 Hz, H-14). The 2-methoxy group was indicated by an HMBC correlation between the proton signal of a methoxy group (δ<sub>H</sub> 3.23, 3H, s) and an oxygenated quaternary carbon (δ<sub>C</sub> 108.5,

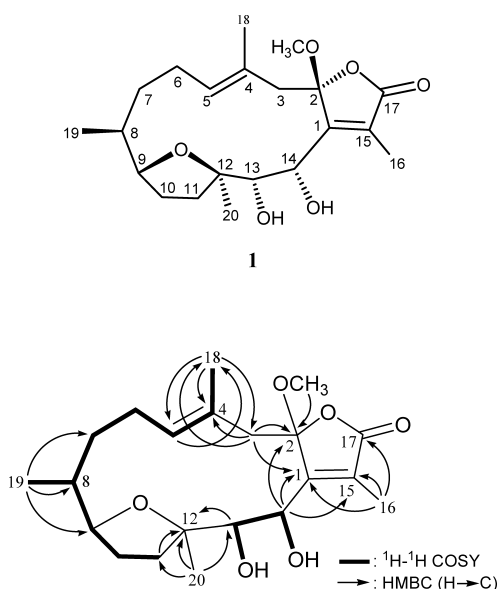


Fig. 1.  $^1\text{H}$ - $^1\text{H}$  COSY and Selective HMBC Correlations of **1**

s, C-2). Based on the consideration of molecular formula, a more oxygen atom had to be placed between C-9 and C-12 to form a tetrahydrofuran ring.

Due to the conformational mobility of the 14-membered ring, the stereochemistry for the chiral centers of **1** could not be fully determined by a nuclear Overhauser effect spectroscopy (NOESY) experiment. Fortunately, the single crystals of **1** could be obtained from the slow evaporation of methanol solution of **1**. A single crystal X-ray diffraction analysis was then carried out and the ORTEP drawing of the structure of **1** is shown in Fig. 2, which unambiguously confirmed the structure of **1**, and the relative configurations of chiral centers of **1** were assigned as  $2S^*$ ,  $8R^*$ ,  $9S^*$ ,  $12R^*$ ,  $13S^*$ , and  $14R^*$ . The geometry of the C-4/5 double bond in **1** was also found to exist in *trans* configuration.

In the cytotoxicity testing, briaviodiol A (**1**) was not active toward the DLD-1 (human colon adenocarcinoma), CCRF-CEM (human T-cell acute lymphoblastic leukemia), HL-60 (human promyelocytic leukemia), and P388D1 (murine macrophage cell) tumor cells ( $\text{ED}_{50} > 40 \text{ mg/ml}$ ). Due to the screening platforms are limited; and lots of material were consumed in physical and spectral experiments. The other possible biological activities for this interesting substance will not be assayed at this stage. The extensive assay platforms for the natural products will be set up by the National Science and Technology Program of Biotechnology and Pharmaceuticals (NSTPBP), Taiwan.<sup>11</sup> The other possible bioactivities for this compound will be studied if we can get enough material in the future.

#### Experimental

**General** Melting points were measured on a FARGO apparatus and were uncorrected. Optical rotation values were measured with a JASCO P-1010 digital polarimeter. IR spectra were obtained on a VARIAN DIGILAB FTS 1000 Fourier transform-infrared (FT-IR) spectrophotometer. NMR spectra were recorded on a VARIAN MERCURY PLUS 400 FT-NMR at 400 MHz for  $^1\text{H}$ -NMR and 100 MHz for  $^{13}\text{C}$ -NMR, respectively, in  $\text{CDCl}_3$ . Proton chemical shifts were referenced to the residual  $\text{CHCl}_3$  signal ( $\delta_{\text{H}}$  7.26 ppm).  $^{13}\text{C}$ -NMR spectra were referenced to the center peak of  $\text{CDCl}_3$  at  $\delta_{\text{C}}$  77.1 ppm. ESI-MS and HR-ESI-MS data were recorded on a BRUKER APEX II mass spectrometer. Gravity column chromatography was per-

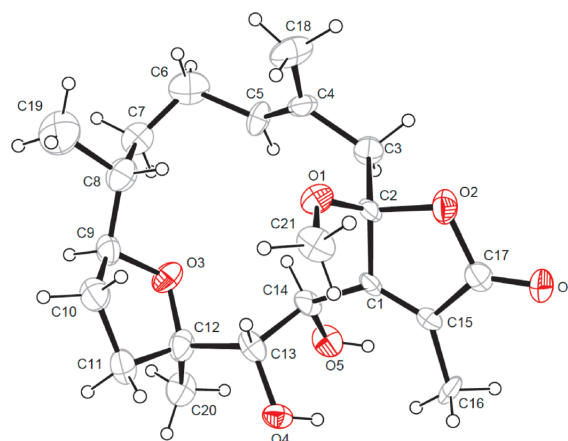


Fig. 2. Computer-Generated ORTEP Plot of **1** Showing the Relative Configuration

formed on silica gel (230–400 mesh, Merck, Darmstadt, Germany). TLC was carried out on precoated Kieselgel 60 F<sub>254</sub> (0.2 mm, Merck) and spots were visualized by spraying with 10%  $\text{H}_2\text{SO}_4$  solution followed by heating. HPLC was performed using a system comprised of a HITACHI L-7100 pump, a HITACHI L-7455 photodiode array detector, and a RHEODYNE 7725 injection port. A normal phase column (Hibar 250×10 mm, LiChrospher Si 60, 5  $\mu\text{m}$ ) was used for HPLC.

**Animal Material** Specimen of the octocoral *Briareum violaceum* was collected off the coast of Pingtung, southern Taiwan, in March 2008, and this organism was identified by comparison with previous descriptions.<sup>12</sup> The voucher specimen was deposited in the National Museum of Marine Biology and Aquarium, Taiwan.

**Extraction and Isolation** The freeze-dried and sliced bodies of *B. violaceum* (wet weight 1015 g, dry weight 395 g) were minced and extracted with a mixture of MeOH and  $\text{CH}_2\text{Cl}_2$  (1 : 1) at room temperature. The extract was partitioned between EtOAc and  $\text{H}_2\text{O}$ . The EtOAc layer (12.2 g) was separated on silica gel and eluted using *n*-hexane/EtOAc (stepwise, 100 : 1–pure EtOAc) to yield 20 fractions. Fraction 7 was separated by silica gel and eluted using *n*-hexane/acetone (stepwise, 6 : 1–pure acetone) to yield the 13 fractions 7A–M. Fraction 7F was repurified by normal phase HPLC using a mixture of *n*-hexane and acetone to yield **1** (2.7 mg, 5/1).

**Briaviodiol A (1)**: Colorless prisms; mp 162–164 °C;  $[\alpha]_{\text{D}}^{23} -31^\circ$  ( $c = 0.14$ ,  $\text{CHCl}_3$ ); IR (neat)  $\nu_{\text{max}}$  3411, 1752  $\text{cm}^{-1}$ ;  $^1\text{H}$ -NMR ( $\text{CDCl}_3$ , 400 MHz) and  $^{13}\text{C}$ -NMR ( $\text{CDCl}_3$ , 100 MHz) data see Table 1; ESI-MS  $m/z$  403 ( $\text{M} + \text{Na}$ )<sup>+</sup>; HR-ESI-MS  $m/z$  403.2093 (Calcd for  $\text{C}_{21}\text{H}_{32}\text{O}_6\text{Na}$ , 403.2096).

**X-Ray Diffraction Analysis of Briaviodiol A (1)** Suitable colorless prisms of **1** were obtained from a solution of MeOH. The crystal (0.50×0.30×0.20 mm) belongs to the monoclinic system, space group  $P2_12_1$  (#19), with  $a = 7.3813(19) \text{ \AA}$ ,  $b = 9.700(3) \text{ \AA}$ ,  $c = 31.818(11) \text{ \AA}$ ,  $V = 2278.1(12) \text{ \AA}^3$ ,  $Z = 4$ ,  $D_{\text{calcd}} = 1.162 \text{ g/cm}^3$ , and  $\lambda (\text{MoK}\alpha) = 0.71073 \text{ \AA}$ . Intensity data were measured on a Rigaku AFC7S diffractometer up to  $2\theta_{\text{max}}$  of 52.00°. All 2571 reflections were collected. The structure was solved using direct methods and refined using a full-matrix least-squares procedure. The refined structural model converged to a final  $R1 = 0.0458$  and  $wR2 = 0.1086$  for 1255 observed reflections [ $I > 2\sigma(I)$ ] and 264 variable parameters.<sup>13</sup>

**Cytotoxicity Testing** The cytotoxicity of tested compound **1** was assayed with a modification of MTT [3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide] colorimetric method. Cytotoxicity assays were carried out according to the procedures described previously.<sup>14,15</sup>

**Acknowledgements** This research work was supported by Grants from the National Museum of Marine Biology and Aquarium (99200321 and 99200322); National Dong Hwa University; Asia-Pacific Ocean Research Center, National Sun Yat-sen University (97C031702); and National Science and Technology Program for Biotechnology and Pharmaceuticals, National Science Council (NSC 98-2323-B-291-001, 99-2323-B-291-001, and 98-2320-B-291-001-MY3), Taiwan, awarded to P.-J.S.

#### References and Notes

- Bowden B. F., Coll J. C., Mitchell S. J., Raston C. L., Stokic G. J., White A. H., *Aust. J. Chem.*, **32**, 2265–2274 (1979).
- Inman W., Crews P., *J. Org. Chem.*, **54**, 2526–2529 (1989).

- 3) Xu L., Patrick B. O., Roberge M., Allen T., van Ofwegen L., Andersen R. J., *Tetrahedron*, **56**, 9031—9037 (2000).
- 4) Sheu J.-H., Wang G.-H., Sung P.-J., Duh C.-Y., Chiang M. Y., *Tetrahedron*, **57**, 7639—7648 (2001).
- 5) Sheu J.-H., Wang G.-H., Duh C.-Y., Soong K., *J. Nat. Prod.*, **66**, 662—666 (2003).
- 6) Iwasaki J., Ito H., Aoyagi M., Sato Y., Iguchi K., *J. Nat. Prod.*, **69**, 2—6 (2006).
- 7) Ito H., Iwasaki J., Sato Y., Aoyagi M., Iguchi K., Yamori T., *Chem. Pharm. Bull.*, **55**, 1671—1676 (2007).
- 8) Wang G.-H., Sheu J.-H., Chiang M. Y., Lee T.-J., *Tetrahedron Lett.*, **42**, 2333—2336 (2001).
- 9) Wang G.-H., Sheu J.-H., Duh C.-Y., Chiang M. Y., *J. Nat. Prod.*, **65**, 1475—1478 (2002).
- 10) Anta C., González N., Rodríguez J., Jiménez C., *J. Nat. Prod.*, **65**, 1357—1359 (2002).
- 11) The National Science and Technology Program for Biotechnology and Pharmaceuticals (NSTPBP) is a national project prompted together by National Science Council (NSC), Ministry of Economic Affairs (MOEA), and the Department of Health (DOH), Taiwan. NSTPBP is designated to advance the research and development of Chinese herbal medicines, new chemical drugs and also protein drugs.
- 12) Fabricus K., Alderslade P., “Soft Corals and Sea Fans—A Comprehensive Guide to the Tropical Shallow-Water Genera of the Central-West Pacific, the Indian Ocean and the Red Sea,” Australian Institute of Marine Science, Queensland, 2001, pp. 55 and 154—157.
- 13) Crystallographic data of the structure of briaviodiol A (**1**) have been deposited with Cambridge Crystallographic Data Center as supplementary publication number CCDC 784285. Copies of the data can be obtained, free of charge, on application the CCDC, 12 Union Road, Cambridge, CB2 1EZ, UK [fax: +44 (0) 1223 336033 or e-mail: deposit@ccdc.cam.ac.uk].
- 14) Alley M. C., Scudiero D. A., Monks A., Hursey M. L., Czerwinski M. J., Fine D. L., Abbott B. J., Mayo J. G., Shoemark R. H., Boyd M. R., *Cancer Res.*, **48**, 589—601 (1988).
- 15) Scudiero D. A., Shoemark R. H., Paull K. D., Monks A., Tierney S., Nofziger T. H., Currens M. J., Seniff D., Boyd M. R., *Cancer Res.*, **48**, 4827—4833 (1988).