Anti-inflammatory Alkaloids from the Stems of *Picrasma quassioides* BENNET

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During further chemical and biological investigations of *Picrasma quassioides* BENNET, four new bis- β -carboline alkaloids, quassidines E—H (1—4), and three new β -carboline alkaloids, canthin-16-one-14-butyric acid (5), 3-(1,1-dimethoxylmethyl)- β -carboline (6), and 6,12-dimethoxy-3-formyl- β -carboline (7), were isolated from its anti-inflammatory CHCl₃-soluble fraction. Structures of new compounds were elucidated and characterized by MS and NMR analysis. A plausible biogenetic pathway for quassidine E (1), the first bis- β -carboline alkaloid in which a canthin-6-one moiety and a β -carboline moiety were connected together by a single carbon–carbon bond from the nature, was proposed. Quassidines E—G (1—3) showed potent inhibitory activity on the production of nitric oxide (NO), tumor necrosis factor α (TNF- α), or interleukin 6 (IL-6) in mouse monocyte-macrophage RAW264.7 cells stimulated by lipopolysaccharide (LPS). Analysis of anti-inflammatory activity of all β -carboline and bis- β -carboline alkaloids from *P. quassioides* showed that the carbonyl groups or double carbon–carbon bonds at C-14 for β -carbolines and C-14' for bis- β -carbolines were bioactive groups for their *in vitro* anti-inflammatory activity. Structure–activity relationship of these compounds on inhibitory activity of the three inflammatory cytokines was discussed.

Key words *Picrasma quassioides*; alkaloid; anti-inflammatory; bis- β -carboline; β -carboline

Chronic inflammation has been found to mediate a wide variety of diseases, including cardiovascular diseases, cancer, Alzheimer's disease, and autoimmune diseases.¹⁾ How is inflammation diagnosed and its biomarkers is not fully understood, however, the role of inflammatory cytokines, such as nitric oxide (NO), tumor necrosis factor α (TNF- α), and interleukin 6 (IL-6), have been linked with chronic inflammation.²⁾ During our screening studies on NO inhibitors in traditional medicine, the CHCl₃-soluble fraction of 95% EtOH extract of Picrasma quassioides (Simaroubaceae) stems showed potent NO inhibitory activity.³⁾ Previous chemical and biological investigations on the bioactive fraction have resulted in the isolation of four new bis- β -carbolines, quassidines A—D,³⁾ and five new β -carbolines, together with 19 known alkaloids.4,5) Quassidine A was the first reported bis- β -carboline alkaloid possessing a novel cyclobutane unit.^{3,6)} Our further investigation of the CHCl₂-soluble fraction led to the isolation and characterization of four new bis- β -carbolines, quassidines E—H (1—4), and three new β -carbolines, canthin-16-one-14-butyric acid (5), 3-(1,1-dimethoxylmethyl)- β -carboline (6), and 6,12-dimethoxy-3-formyl- β carboline (7). Quassidine E (1) was a novel bis- β -carboline alkaloid in which a canthin-6-one moiety and a β -carboline moiety were connected together by a single carbon-carbon bond. In this paper, the isolation, structure elucidation, and in vitro anti-inflammatory activity of new compounds were reported. In addition, the plausible biogenetic pathway of quassidine E (1) was proposed and the structure-activity relationship of bis- β -carboline and β -carboline alkaloids was also discussed.

Results and Discussion

By a series of column chromatography using silica gel, Sephadex LH-20, and octadecylsilyl (ODS) in combination with reversed phase HPLC, the CHCl₃-soluble fraction of *P. quassioides* stems resulted in the purification of four new bis- β -carbolines (1—4) and three new β -carbolines (5—7), as shown in Fig. 1.

Quassidine E (1) was obtained as a yellowish powder and had a molecular formula of $C_{27}H_{18}N_4O_2$ established by the HR-ESI-MS ion peak at *m/z* 429.1356 [M–H][–] (Calcd for $C_{27}H_{17}N_4O_2$, 429.1357). The UV spectrum of 1 displayed absorption bonds at 243, 282, 332, 367, and 383 nm, suggesting the presence of β -carboline chromophores.⁷⁾ The absorption

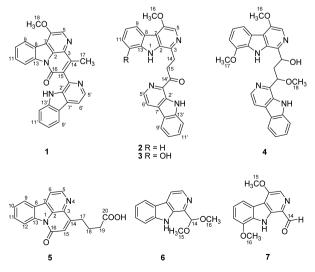


Fig. 1. Chemical Structures of 1-7 from P. quassioides

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No.	1	2	3	4
1		11.71 (br s)	11.46 (br s)	11.03 (br s)
5	8.79 (s)	7.90 (s)	7.83 (s)	7.93 (s)
9	8.52 (d, 8.0)	8.19 (d, 8.0)	7.64 (d, 7.6)	7.77 (d, 7.8)
10	7.75 (td, 8.0,1.0)	7.24 (td, 8.0, 1.0)	7.03 (t, 7.6)	7.18 (t, 7.8)
11	7.62 (td, 8.0, 1.0)	7.51 (td, 8.0, 1.0)	6.93 (d, 7.6)	7.11 (d, 7.8)
12	8.27 (d, 8.0)	7.62 (d, 8.0)		
14		3.55 (t, 7.0)	3.56 (t, 7.0)	5.08 (m)
15		4.00 (t, 7.0)	3.91 (t, 7.0)	2.76 (m)
				2.63 (m)
1'	11.40 (br s)	11.88 (br s)	11.90 (br s)	11.40 (br s)
5'	8.48 (d, 5.2)	8.52 (d, 5.0)	8.53 (d, 5.0)	8.35 (d, 5.2)
6'	8.19 (d, 5.2)	8.43 (d, 5.0)	8.43 (d, 5.0)	8.08 (d, 5.2)
9'	8.29 (d, 8.0)	8.29 (d, 7.9)	8.28 (d, 8.0)	8.22 (d, 8.1)
10'	7.26 (td, 8.0, 1.0)	7.29 (td, 7.9, 1.0)	7.29 (td, 8.0, 1.0)	7.23 (td, 8.1, 1.0
11'	7.52 (td, 8.0, 1.0)	7.57 (td, 7.9, 1.0)	7.58 (td, 8.0, 1.0)	7.54 (td, 8.1, 1.0
12'	7.45 (d, 8.0)	7.78 (d, 7.9)	7.77 (d, 8.0)	7.69 (d, 8.1)
14'				5.00 (m)
16		4.04 (s)	4.01 (s)	4.06 (s)
17	2.36 (s)			4.03 (s)
18	4.33 (s)			3.18 (s)
12-OH			9.93 (s)	
14-OH				5.59 (d, 5.9)

Table 2. ¹³C-NMR Spectroscopic Data of Compounds 1—4 in DMSO- d_6

No.	1	2	3	4
2	131.9	134.9	134.8	133.5
3	130.5	137.7	138.4	140.8
5	130.6	119.84	119.5	119.1
6	152.1	149.9	149.8	150.4
7	115.6	116.2	116.8	117.5
8	137.8	120.3	122.0	121.1
9	116.2	123.2	114.0	115.4
10	129.6	119.3	120.11	120.5
11	125.7	126.7	110.9	107.3
12	124.1	111.6	143.6	145.8
13	123.1	139.6	129.6	129.5
14	147.8	26.7	27.2	69.0
15	131.1	35.1	38.5	39.7
2'	134.8	133.9	134.0	140.7
3'	139.1	135.8	135.8	144.3
5'	137.9	137.3	137.4	137.0
6'	114.4	119.2	119.2	114.2
7′	128.0	130.9	130.9	120.5
8′	120.7	119.81	119.8	128.5
9′	121.7	121.7	121.7	121.4
10'	119.2	120.0	120.05	119.2
11'	128.1	128.7	128.8	128.0
12'	111.7	112.9	113.0	112.3
13'	140.4	141.8	141.8	133.3
14'		202.3	202.3	80.2
16	158.7	55.9	55.8	55.96
17	13.7			55.5
18	57.2			55.95

bonds at 1665 and 1626 cm⁻¹ in the IR spectrum of **1** indicated the presence of an α,β -unsaturated amide fragment. Analysis of the ¹H-NMR data in conjunction with the HSQC spectrum revealed the presence of one NH, one methoxyl, one methyl, eleven protonated sp^2 carbons (Table 1). The ¹³C-NMR spectrum further showed the presence of one carbonyl, fourteen nonprotonated sp^2 carbons (including two oxygenated quaternary carbons) (Table 2). Although **1** was

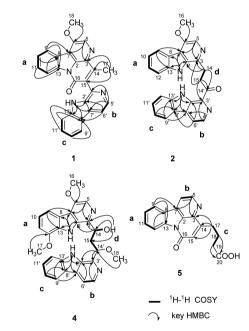


Fig. 2. Key Correlations and Partial Structures of 1, 2, 4 and 5

suggested to be related to bis- β -carboline alkaloids (such as quassidines A and B) by comparing the ¹H- and ¹³C-NMR data,³⁾ the gross structure of **1** was quite different on the basis of a detailed analysis of 2D-NMR data including COSY, HSQC, HMBC spectra. Three partial structures **a**—**c** were deduced from COSY analysis of **1** (Fig. 2). HMBC correlations for H-5/C-3,7, H-9/C-7, H-10/C-8, and H-11/C-13 established partial structure of canthin-6-one moiety; the long-ranged correlations H₃-17/C-3,14,15 and H₃-18/C-6 determined the position of the methyl and methoxyl groups, respectively. Furthermore, HMBC correlations from H-5' to C-7' as well as correlations from H-6' and H-10' to C-8' could connect two partial structural units **b** and **c**. The connection of β -carboline moiety was unambiguously established by

N	$5^{a)}$	5 ^{<i>a</i>)}		6 ^{b)}		$7^{b)}$	
No.	$\delta_{ m H}$ (J in Hz)	$\delta_{ m C}$	$\delta_{ m H} (J ext{ in Hz})$	$\delta_{ m c}$	$\delta_{ m H}$ (<i>J</i> in Hz)	$\delta_{ m c}$	
1			9.16 (br s)		10.13 (br s)		
2		131.3		133.5		136.4	
3		135.8		140.2		131.8	
5	8.84 (d, 4.9)	145.1	8.45 (d, 5.2)	138.2	8.26 (s)	123.2	
6	8.32 (d, 4.9)	117.3	7.95 (d, 5.2)	114.9		154.9	
7		129.5		130.1		119.0	
8		124.2		121.1		121.0	
9	8.37 (d, 8.0)	123.5	8.13 (d, 8.0)	121.6	7.88 (d, 8.0)	116.1	
10	7.58 (td, 8.0, 1.0)	125.4	7.28 (td, 8.0, 1.0)	119.9	7.28 (t, 8.0)	121.8	
11	7.75 (td, 8.0, 1.0)	130.7	7.56 (td, 8.0, 1.0)	128.6	7.03 (d, 8.0)	108.1	
12	8.49 (d, 8.0)	116.1	7.54 (d, 8.0)	111.6		146.2	
13		138.5		140.3		130.6	
14		153.0	5.76 (s)	106.6	10.23 (s)	193.6	
15	6.79 (s)	125.1	3.53 (s)	54.2	4.28 (s)	56.6	
16		158.8	3.53 (s)	54.2	4.05 (s)	55.7	
17	3.05 (t, 7.4)	28.5	~ /				
18	2.03 (quai, 7.4)	23.8					
19	2.35 (t, 7.4)	33.1					
20		174.1					

Table 3. ¹H- and ¹³C-NMR Spectroscopic Data of Compounds 5-7

a) Measured in DMSO-d₆. b) Measured in CDCl₃.

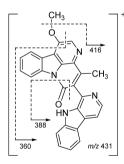


Fig. 3. Cleavage Pathway of Positive $\mathrm{ESI}\text{-}\mathrm{MS}^{4)}$ Fragmentation of Quassidine E (1)

HMBC data: correlations H-5'/C-3', H-6'/C-2', and H-11'/C-13' determined the position of the three quaternary carbons; cross peaks observed for NH/C-2',7',8',13' suggested the placement of the NH. Finally, the 6-methoxy-14methylcanthin-16-one moiety was deduced to be connected to the β -carboline moiety *via* a single carbon–carbon bond between C-15 and C-3'. With all the information, the structure of **1** was determined as shown in Fig. 1. It was the first bis- β -carboline alkaloid obtained from the nature, in which the canthin-6-one moiety and the β -carboline moiety were connected together by a single carbon–carbon bond, named as quassidine E (**1**). The complete structure of **1** was further confirmed by its positive ESI-MS⁴ analysis (Fig. 3).

Quassidine F (2) had a molecular formula of $C_{26}H_{20}N_4O_2$ established by the HR-ESI-MS ion peak at m/z 421.1681 $[M+H]^+$ (Calcd for $C_{26}H_{21}N_4O_2$, 421.1665). The UV spectrum of **2** showed absorption bonds at 244 and 285 nm. Four partial structures **a**—**d** deduced from COSY spectra were connected by key HMBC correlations as follows: (1) correlations for H-5/C-3,7, H-9/C-7,13, NH-1/C-2,7,8,13 and H₃-16/C-6 established one 6-methoxy- β -carboline moiety; (2) correlations for H-5'/C-3',7', H-6'/C-2',8', H-9'/C-7',13', and NH-1'/C-2',7',8',13' established the other β -carboline moiety; (3) correlations for H₂-14/C-2,3,14' and H₂-15/C-

3,3',14' determined the two β -carboline moieties were connected together by partial structure **d** and the carbonyl at C-14' *via* the carbons C-3 and C-3'. Thus, the structure of **2** was determined as shown in Fig. 1, named as quassidine F.

Quassidine G (3) had a molecular formula of $C_{26}H_{20}N_4O_3$ established by the HR-ESI-MS ion peak at m/z 435.1480 $[M-H]^-$ (Calcd for $C_{26}H_{19}N_4O_3$, 435.1457), suggesting that 3 was a hydroxyl-derivative of 2. Oxidization of C-12 aromatic methine was straightforward from ¹H- and ¹³C-NMR data (δ_{OH} 9.93, δ_C 143.6) assigned as 12-hydroxylquassidine F, which was confirmed by the HMBC correlations of 3 from 12-OH to C-11, C-12, and C-13.

Quassidine H (4) had a molecular formula of $C_{28}H_{26}N_4O_4$ established by the HR-ESI-MS ion peak at m/z 483.2047 $[M+H]^+$ (Calcd for C₂₈H₂₇N₄O₄, 483.2032). The UV spectrum of 4 showed absorption bonds at 245 and 289 nm. The ¹H- and ¹³C-NMR data of **4** were similar to those of **2**, and the main difference was the carbonyl C-14', the methylene C-14, and the aromatic methine C-12 were correspondingly replaced by an oxygenated methine at $\delta_{\rm C}$ 80.2, an oxygenated methine at $\delta_{\rm C}$ 69.0, and an aromatic quaternary carbon at $\delta_{\rm C}$ 145.8. The HMBC correlations for H₃-18/C-14', 14-OH/C-14,3,15, and H₂-17/C-12 determined the placement of two methoxyl groups and the hydroxyl group (Fig. 2). Thus, the structure of 4 was assigned as shown in Fig. 1, named as quassidine H. The relative and absolute configuration of 4 remained to be determined in account of the scarcity of samples and the presence of 14'-OCH₃.

Compound **5** was obtained as a yellowish powder and had a molecular formula of $C_{18}H_{14}N_2O_3$ established by the HR-ESI-MS ion peak at m/z 305.0930 $[M-H]^-$ (Calcd for $C_{18}H_{13}N_2O_3$, 305.0926). The UV spectrum of **5** displayed absorption bonds at 243 and 281 nm, indicating the presence of a β -carboline chromophore. No NH was observed in the ¹H-NMR spectrum and 15 aromatic carbons were displayed in the ¹³C-NMR spectrum, suggesting the presence of a canthin-6-one moiety. Three partial structures **a**—**c** were deduced from COSY analysis of **5** (Fig. 1). The HMBC correlations H_2 -17/C-3,14,15 and H_2 -19/C-20 determined the butyric acid moiety was located at position C-14. Thus, the structure of **5** was determined as canthin-16-one-14-butyric acid.

Compound **6** had a molecular formula of $C_{14}H_{14}N_2O_2$ which was established by the HR-ESI-MS ion peak at m/z241.0976 [M-H]⁻. It was induced to be a monomeric β -carboline by comparing the ¹H- and ¹³C-NMR data of **6** to those of **2**. In the HMBC spectrum, H-14 showed correlations to C-2 and C-3, while H₃-15 and H₃-16 both showed correlations to C-14, which determined the dimethoxylmethyl group was located at C-3 position. Thus, the structure of **6** was finally determined as 3-(1,1-dimethoxylmethyl)- β -carboline, which might be an artifact arising from 3-formyl- β -carboline (**10**). The aldehyde group of **10** could carry out a nucleophilic addition reaction with CH₃OH, the solvent widely used in the experimental procedures, resulting in the production of **6**.

The molecular formula of 7 was established as $C_{14}H_{12}N_2O_3$ by its HR-ESI-MS. ¹H- and ¹³C-NMR data of 7 were similar to those of **6** except for the formyl group (δ_H 10.23, δ_C 193.6) and two methoxyl groups (δ_C 55.7, δ_C 56.6). The HMBC correlations for H-14/C-2,3, H₃-15/C-6, and H₃-16/C-12 determined the formyl group and two methoxyl groups were attached at C-3, C-6, and C-12, respectively. Thus, the structure of 7 was determined as 6,12-dimethoxy-3-formyl- β -carboline, as shown in Fig. 1.

A plausible biogenetic pathway for quassidine E (1) was proposed as shown in Chart 1. The two precursors, picrasidine X (1a) and 3-acetyl- β -carboline (1b),^{8,9)} were known β carboline alkaloids obtained from *P. quassioides*. Compound 1a underwent decarboxylation and oxidation at C-15 to produce 1c. When it lost a proton at CH₂-14, compound 1c could convert to a carbanion, which attacked the carbonyl at C-14' in 1b to form 1d. By intramolecular acylation and dehydration, compound 1d was transformed to compound 1f, which led to the final formation of quasidine E (1) following its oxidation and methylation at C-6.

The *in vitro* anti-inflammatory activity of seven new compounds (1—7) and 23 β -carbolines (8—30) was evaluated by monitoring the inhibition activity on the production of NO, TNF- α , or IL-6 in mouse monocyte-macrophage RAW 264.7 cells stimulated by lipopolysaccharide (LPS). As shown in Table 4, quassidine E (1), especially quassidines F (2) and G (3) showed potent inhibitory activity, whereas quassidines $B-D^{3}$ and H (4) showed potent toxicity on normal RAW 264.7 cells on the concentration of 100 μ g/ml. The major difference of quassidines F (2) and G (3) with quassidines B— D and H (4) was the oxidation level of C-14', ketone groups for the former and hydroxyl groups for the latter. Thus, the ketone groups at C-14' might be the bioactive groups at C-14' might be the bioactive groups at C-14' might be the toxic groups at C-14' might be the toxic groups to normal RAW 264.7 cells for quassidines B—D and H (4) on test concentration of 100 μ g/ml.

The similar conclusion was deduced by analyzing the inhibition activity of β -carbolines (7–30)^{4,5)} shown in Table 5. Compounds 7 and 10–15 showed much stronger inhibitory activity on NO, TNF- α , or IL-6 than other β -carbolines. The carbons C-14 of 7 and 10–15 were all sp^2 hybridization, such as aldehyde groups, ester groups, and double carbon–carbon bonds. As a consequence, the carbonyl groups or double carbon–carbon bonds at C-14 were crucial for *in vitro* anti-inflammatory activity of β -carbolines.

In this work, four new bis- β -carboline alkaloids, quassidines E—H (1—4), and three new β -carboline alkaloids (5—7) were isolated from the stems of *P. quassioides*. Their structures were fully elucidated by 2D-NMR analysis. A plausible biogenetic pathway for the novel quassidine E (1) was proposed. Analysis of the anti-inflammatory activity of bis- β -carboline and β -carboline alkaloids (7—30) suggested that the carbonyl groups or double carbon–carbon bonds at C-14 for β -carbolines and C-14' for bis- β -carbolines were bioactive groups for these compounds.

Table 4. Inhibitory Effects of Bis- β -carbolines from *P. quassioides* on NO, TNF- α , and IL-6 Production in LPS-Stimulated RAW 264.7 Cells (*n*=4)

	Inhibition activity (IC ₅₀ μ M)		
	NO	TNF- α	IL-6
Quassidine E (1) Quassidine F (2)	20.51 ± 1.67 9.85 ± 0.76	25.64 ± 2.32 >100	45.35±3.29 24.32±2.17
Quassidine G (3)	13.09 ± 1.15	12.27±1.39	17.14±1.64
Quassidine H (4) Hydrocortisone ^{a)}	$+++^{b)}$ 64.34±6.17	$+++85.64\pm6.23$	$+++63.86\pm5.25$

a) Hydrocortisone, positive control. b) Toxicity to normal RAW 264.7 cells on the concentration of $100 \mu g/ml$.

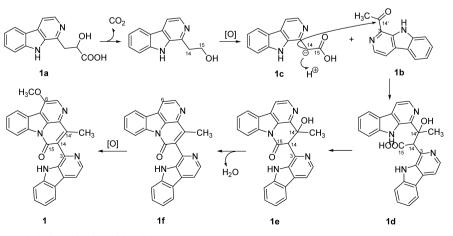


Chart 1. A Plausible Biogenetic Pathway for Quassidine E (1)

D	P R,	R_2	D	In	nhibition activity (IC ₅₀ μ M)	
R ₁	(^a 10-OH)	R ₃	NO	TNF- α	IL-6	
	_	_	47.90±4.13	25.54±3.31	66.96±6.58	
Н	Н	$CH(OCH_3)_2$	26.94 ± 1.96	35.02 ± 2.88	50.43 ± 4.7	
OCH ₃	OCH ₃	CHO	17.42 ± 0.98	>100	>100	
OCH ₃	OH	COOCH ₃	54.79 ± 3.45	58.64 ± 4.78	57.20±4.26	
OH	Н	COOCH ₃	65.82 ± 6.02	35.09 ± 2.48	92.01±6.33	
Н	Н	СНО	19.04 ± 1.85	9.98 ± 1.14	14.84 ± 1.35	
Н	Н	COOCH ₃	7.68 ± 0.83	29.23 ± 2.08	3.87±0.58	
Н	Н	COOCH ₂ CH ₂	23.33 ± 2.56	21.87 ± 1.66	30.54±2.93	
OCH ₃		Vinyl	4.07 ± 0.55	9.00 ± 1.17	6.54 ± 0.61	
OCH ₃	OCH ₃	Vinyl	3.96 ± 0.63	7.42 ± 0.94	11.30 ± 0.99	
OCH ₂	Н	CHO	55.16 ± 3.52	>100	13.17 ± 1.11	
OCH ₃	OCH ₃	CH ₂ CH ₂ OH	89.81±6.22	>100	>100	
OCH ₃	OCH ₃	CH(OH)CH ₃	$+++^{b)}$	+++	+++	
OCH ₂	OCH ₃		43.17 ± 4.02	23.02 ± 1.93	60.35 ± 4.69	
OCH ₃	Н		63.62 ± 4.31	+++	+++	
OCH ₃	Н		77.85 ± 0.83	>100	>100	
	Н		>100	>100	>100	
Н	Н		18.13 ± 1.36	>100	33.40 ± 0.46	
OCH ₂	Н		+++	>100	26.84 ± 2.92	
	OCH ₂		+++	+++	+++	
	Н	2 5	>100	>100	>100	
Н	Н		>100	>100	>100	
Н	Н	CH ₂ CH(OH)COOH	>100	>100	>100	
Н	OH^{a}	OH	>100	>100	>100	
Н	Н	OH	74.87 ± 6.88	32.39 ± 3.06	41.65 ± 3.94	
Н	Н	OCH ₂	63.46 ± 7.21	64.41 ± 5.23	>100	
	OCH ₃ OH H H OCH ₃ OCH ₃ H H H	$\begin{array}{cccc} {\rm OCH}_3 & {\rm OCH}_3 & {\rm OH} \\ {\rm OH} & {\rm H} \\ {\rm H} & {\rm H} \\ {\rm OCH}_3 & {\rm OCH}_3 \\ {\rm OCH}_3 & {\rm H} \\ {\rm H} \\ {\rm H} & {\rm H} \\ {\rm H} & {\rm H} \\ {\rm H} \end{array} $	OCH_3 OCH_3 CHO OCH_3 OH $COOCH_3$ OH H $COOCH_3$ H H $COOCH_3$ H H $COOCH_3$ H H $COOCH_2CH_3$ OCH_3 OCH_3 $Vinyl$ OCH_3 OCH_3 $Vinyl$ OCH_3 OCH_3 CHO OCH_3 OCH_3 CHO OCH_3 OCH_3 $CHOO$ OCH_3 OCH_3 $CH(OH)CH_3$ OCH_3 OCH_3 $CH(OH)CH_2OH$ OCH_3 OCH_3 $CH(OH)CH_2OH$ OCH_3 H CH_2CH_2OH OCH_3 H CH_2CH_2OH OCH_3 H CH_2OH OCH_3 H CH_2OH OCH_3 H CH_2CH_3 OCH_3 H CH_2CH_3 OCH_3 H CH_2CH_3 OCH_3 H CH_3 <	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	

a) Hydrocortisone, positive control. b) Toxicity to normal RAW 264.7 cells on the concentration of 100 µg/ml.

Experimental

General Procedures Column chromatography was performed with silica gel (Haiyang; 200—300 mesh) and Sephadex LH-20 (Mitsubishi Kasei, Tokyo, Japan), HPLC purification was carried out on a Gilson apparatus equipped with a 306 pump and a UV/VIS-512 detector by using reversed-phase preparative column (21.2×250 mm, 5 μ m, Welch XB-C18, v=8 ml/min). Optical rotations were measured on a Jasco P-1020 digital polartmeter (*l*=1 cm). UV spectra were acquired in DMSO-*d*₆ and CDCl₃ (chemical shifts were referenced to the solvent signals) on a Bruker Avance 400 operating at 400 MHz, using an inverse probe fitted with a gradient along the Z-axis. ESI-MS were measured on a Finnigan LCQ Advantage MAX spectrometer, and HR-ESI-MS were measured on a Agilent 6210 LC/MSD TOF mass spectrometer.

Plant Material The stems of *P. quassioides* were collected from Guangxi Autonomy, China in June 2005, and authenticated at Guangzhou University of Traditional Chinese Medicine. A voucher specimen has been deposited at Institute of Traditional Chinese Medicine and Natural Products, Jinan University, Guangzhou (510632), China.

Extraction and Isolation The dried stems of *P. quassioides* (100 kg) were extracted with 95% ethanol, and the combined solvent extracts were concentrated under reduced pressure. The extract (200 g) was suspended in water and successively extracted with CHCl₃, EtOAc, and *n*-BuOH to yield CHCl₃-soluble fraction (128 g), EtOAc-soluble fraction (8.12 g), and *n*-BuOH fraction (20.8 g). The CHCl₃-soluble fraction was subjected to a silica gel column eluted with cyclohexane, cyclohexane/EtOAc, EtOAc, and MeOH to afford ten fractions (PQC1—PQC10). PQC3 (4.5 g, cyclohexane/EtOAc 50:50) was isolated by a silica gel chromatography column using gradient elution with CHCl₃/CH₃OH (98:2—95:5) to yield four fractions. Purification of fraction PQC3-3 (1.1 g, CHCl₃/CH₃OH 98:2) on a further silica gel chromatography column using an isocratic elution with CHCl₃/CH₃OH 98:2 to afford compound 7 (26.3 mg). The fraction PQC7 (19.2 g, cyclohexane/EtOAc 50:50) was purified by silica gel chromatograph

phy column using gradient elution with CHCl₃/CH₃OH (100:0-0:100) to afford seven fractions. The fourth fraction PQC7-4 (6.0 g, CHCl₃/CH₃OH 98:2) was isolated on ODS-MPLC using a gradient elution with CH₃OH/H₂O (90:10-100:0) to yield ten subfractions, the eight subfraction PQC7-4a-8 (0.41 g) was purified on RP-HPLC (75% CH₃OH) to afford compounds 2 (28 mg) and 4 (5.4 mg). PQC7-5 (3.12 g, CHCl₃/CH₃OH 94:6) was purified on RP-HPLC (75% CH₃OH) to yield compound 3 (10.2 mg), and PQC7-6 (0.51 g, CHCl₃/CH₃OH 94:6) was purified on RP-HPLC (50% CH₃OH) to yield compound 6 (6.2 mg). PQC8 (6.8 g, cyclohexane/EtOAc 50:50) was isolated on silica gel chromatography column using a step gradient of CHCl₃/CH₃OH to afford nine fractions, the third fraction PQC8-3 (0.37 g, CHCl₃/CH₃OH 98:2) was purified on RP-HPLC (70% CH₃OH) to yield compound 1 (19.4 mg). In addition, the fourth fraction PQC8-4 (0.89 g, CHCl₃/CH₃OH 98:2) was isolated on ODS-MPLC (70% CH₃OH) to afford PQC8-4-3 (0.24 g), which was further purified by Sphadex LH-20 (CHCl₃/CH₃OH 50:50) to yield compound 5 (27.8 mg). The isolation procedures of compounds 8-30 were reported in our previous papers.3,4)

Quassidine E (1): Yellowish powders; ¹H- and ¹³C-NMR data, see Tables 1 and 2; IR (KBr) v_{max} 3320, 1665, 1626, 1604, 1492, 1339, 1317, 982, 742 cm⁻¹; UV (CHCl₃) λ_{max} (log ε) 243 (4.69), 282 (4.40), 332 (4.16), 367 (4.37), 382 (4.40) nm; HR-ESI-MS m/z 429.1356 [M-H]⁻ (Calcd for C₂₇H₁₇N₄O₂, 429.1357); ESI-MS (positive) m/z 861.1 [2M+H]⁺, 431.2 [M+H]⁺; ESI-MS (negative) m/z 429.1 [M-H]⁻; ESI-MS (positive) m/z861 [2M+H]⁺, 431 [M+H]⁺; ESI-MS²) (positive) m/z 416 [M+H-CH₃]⁺; ESI-MS³) (positive) m/z 388 [M+H-CH₃-CO]⁺; ESI-MS⁴) (positive) m/z360 [M+H-CH₃-CO-CO]⁺.

Quassidine F (2): Yellowish needles; ¹H- and ¹³C-NMR data, see Tables 1 and 2; IR (KBr) v_{max} 3160, 1673, 1624, 1592, 1516, 1493, 1455, 1428, 1347, 1253, 1119, 741 cm⁻¹; UV (CHCl₃) λ_{max} (log ε) 244 (4.66), 285 (4.36), 348 (3.98) nm; HR-ESI-MS m/z 421.1681 [M+H]⁺ (Calcd for C₂₆H₂₁N₄O₂, 421.1665); ESI-MS (positive) m/z 421 [M+H]⁺; ESI-MS (neg-

ative) m/z 419 [M-H]⁻.

Quassidine G (3): Yellowish powders; ¹H- and ¹³C-NMR data, see Tables 1 and 2; IR (KBr) v_{max} 3420, 2925, 1657, 1626, 1560, 1542, 1496, 1432, 1397, 1284, 1207, 746 cm⁻¹; UV (CHCl₃) λ_{max} (log ε) 244 (4.37), 288 (3.88), 349 (3.59) nm; HR-ESI-MS m/z 435.1480 [M-H]⁻ (Calcd for C₂₆H₁₉N₄O₃, 435.1457); ESI-MS (positive) m/z 437 [M+H]⁺; ESI-MS (negative) m/z 435 [M-H]⁻.

Quassidine H (4): Yellowish powders; ¹H- and ¹³C-NMR data, Tables 1 and 2; IR (KBr) v_{max} 3423, 2929, 1630, 1579, 1505, 1426, 1319, 1077 cm⁻¹; UV (CHCl₃) λ_{max} (log ε) 245 (4.39), 289 (3.91), 348 (3.56) nm; HR-ESI-MS *m/z* 483.2047 [M+H]⁺ (Calcd for C₂₈H₂₇N₄O₄, 483.2032); [α]₂₅^D 0 (*c*=0.5, CHCl₃); ESI-MS (positive) *m/z* 987.3 [2M+Na]⁺, 483.2 [M+H]⁺; ESI-MS (negative) *m/z* 481.5 [M-H]⁻.

Canthin-16-one-14-butyric Acid (**5**): Yellowish powders; ¹H- and ¹³C-NMR data, see Table 3; IR (KBr) v_{max} 3433, 2935, 1660, 1634, 1594, 1481, 1452, 1422, 1272, 1125 cm⁻¹; UV (CHCl₃) λ_{max} (log ε) 243 (4.23), 281 (3.95) nm; IR (KBr) v_{max} 3433, 2935, 1660, 1634, 1594, 1481, 1452, 1422, 1272, 1125 cm⁻¹; HR-ESI-MS *m/z* 305.0930 [M-H]⁻ (Calcd for C₁₈H₁₃N₂O₃, 305.0926); ESI-MS (negative) *m/z* 305.2 [M-H]⁻.

3-(1,1-Dimethoxylmethyl)- β -carboline (6): Yellowish powders; ¹H- and ¹³C-NMR data, see Table 3; ESI-MS (positive) *m/z* 243 [M+H]⁺; IR (KBr) v_{max} 3447, 2932, 1627, 1495, 1430, 1320, 1273, 1127, 748 cm⁻¹; UV (CHCl₃) λ_{max} (log ε) 244 (4.69), 290 (4.50), 335 (3.95), 349 (3.96) nm; HR-ESI-MS *m/z* 241.0976 [M-H]⁻ (Calcd for C₁₄H₁₃N₂O₂, 241.0983); ESI-MS (negative) *m/z* 241 [M-H]⁻.

6,12-Dimethoxy-3-formyl-β-carboline (7): Yellowish powders; ¹H- and ¹³C-NMR data, see Table 3; IR (KBr) v_{max} 3309, 1671, 1578, 1491, 1416, 1344, 1306, 1259, 1060, 957, 724 cm⁻¹; UV (CHCl₃) λ_{max} (log ε) 244 (4.40), 284 (4.31), 376 (3.86) nm; HR-ESI-MS *m/z* 279.0733 [M+Na]⁺ (Calcd for C₁₄H₁₂N₂O₃Na, 279.0746); ESI-MS (positive) *m/z* 279 [M+Na]⁺, 257 [M+H]⁺; ESI-MS (negative) *m/z* 255 [M-H]⁻.

Bioassay for NO, TNF-\alpha, and IL-6 Production Anti-inflammatory activity was evaluated according to the methods described in our previous paper.³⁾ Briefly, The RAW 264.7 cells were cultured in 96-well plates at the initial density of 5×10⁵ cells/ml in RPMI 1640 medium (200 μ l/well) for 1 h. Then the test compound dissolved in dimethyl sulfoxide (DMSO) at various concentrations was added (0.4 μl/well) and LPS (Sigma) was also added with a final concentration of 1 μg/ml. LPS groups only received LPS, and the control groups only received 0.4 μl DMSO per well. The Cells were cultured 24 h for NO analysis. NO was determined by measuring the amount of nitrite in the cell culture supernatant, using Griess reagent. 100 μl of the supernatant from incubates was mixed with an equal volume of Griess reagent, the absorbance at 540 nm were measured and the inhibitory rate was calculated.¹⁰ Cytotoxicity was determined by the MTT (Sigma) colorimetric assay, after 24 h incubation with test compound. TNF-α and IL-6 were detected in the cell culture supernatant, using enzyme-linked immunosorbent assay kits, mouse TNF-α ELISA kit and mouse IL-6 ELISA kit (R & D), according to the manufacturer's recommendations, respectively.

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