

Synthetic Studies of the Lichen Macrolide Lepranthin. Stereoselective Synthesis of the Diolide Framework Based on Regioselective Epoxide-Opening Reactions

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Stereoselective synthesis of the 16-membered diolide **27**, a fully functionalized congener of lepranthin (**1**), is described. The requisite five asymmetric carbon centers in monomer **23** were constructed in a highly stereoselective manner by using different epoxide-opening reactions of α,β -unsaturated γ,δ -epoxy esters and epoxy alcohol derivatives as the key steps. The monomer **23** was successfully transformed into the MOM protected diolide **27** by Yamaguchi macrolactonization.

Key words lepranthin; regioselective epoxide-opening reaction; Yamaguchi macrolactonization; dimeric macrolide

Macrolide antibiotics have provided us with good opportunities discovering drugs that exhibit a wide range of biological activities. Bacteria, fungi and algae produce a large number of macrolides, which are classified as polyketide macrolides in their biosynthetic pathways. Interestingly, a few polyketide macrolides have been isolated from lichens, which may represent a symbiotic relationship between fungi and algae.¹⁾ Lepranthin (**1**) was isolated from the crustaceous lichen *Arthonia impolita* (Ehrh.) BORRER by Zopf in 1904.²⁾ Almost a century later, Huneck and colleagues determined the structure of **1** using NMR techniques and finally X-ray crystallographic analysis, which disclosed the unique 16-membered dimeric macrolide structure containing two secondary hydroxyl groups and four secondary acetates.³⁾ In spite of its characteristic macrolide structure, biological activity and synthetic studies of **1** have not been reported so far. We thought that a sufficient supply of **1** by total synthesis should advance the study of biological properties and set about synthetic studies. We report herein the stereoselective synthesis of methoxymethyl group (MOM) protected diolide **27**, a fully functionalized congener of lepranthin (**1**), based on regioselective epoxide-ring opening strategies and subsequent Yamaguchi macrolactonization.

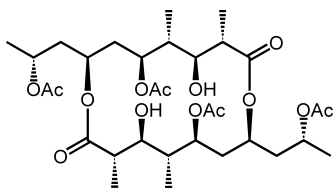
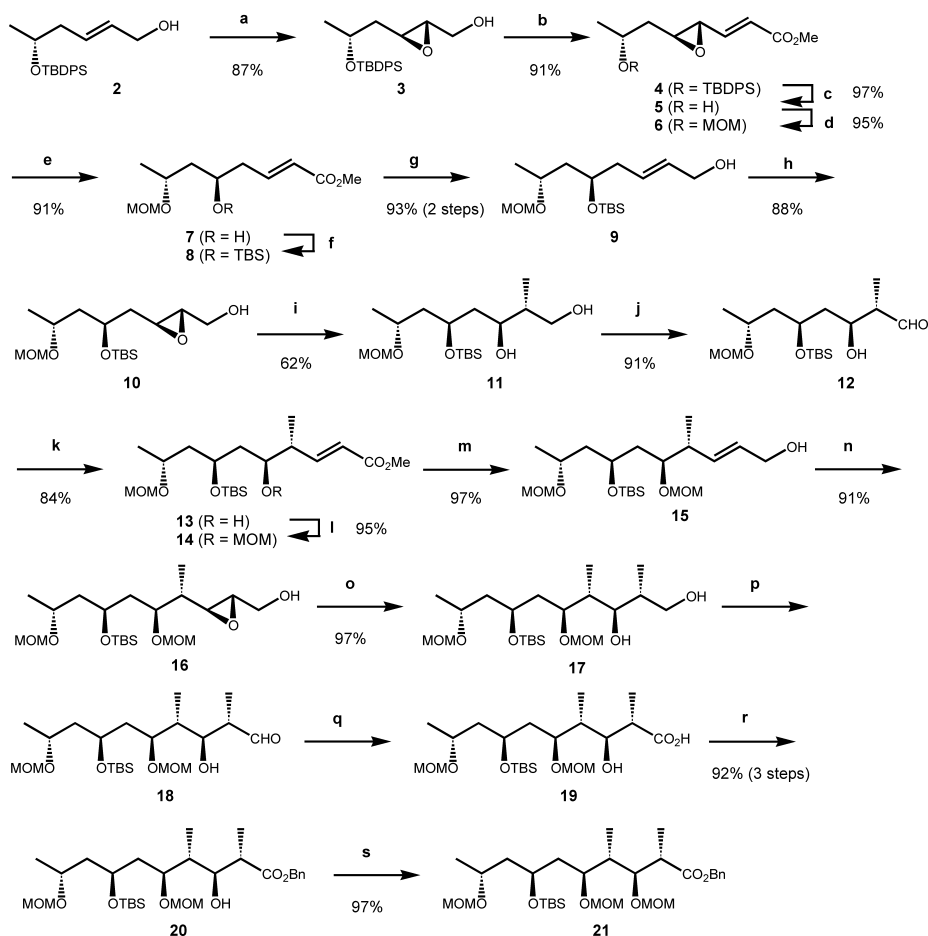


Fig 1. Lepranthin (**1**)

Our synthesis started with allyl alcohol **2**⁴⁾ which was derived from commercially available methyl (*R*)-3-hydroxybutylate in four steps. First, **2** was converted to α,β -unsaturated γ,δ -epoxy ester **6** by a four-step reaction sequence: (1) Katsuki–Sharpless epoxidation⁵⁾ with L-(+)-diethyl tartrate (DET), Ti(O^{*i*}Pr)₄, and *tert*-butyl hydroperoxide (TBHP) in CH₂Cl₂ at –30 °C, leading to epoxy alcohol **3** (87%); (2) Dess–Martin oxidation⁶⁾ followed by a Wittig reaction (91% yield); (3) desilylation (97%); (4) protection of the resulting alcohol with a MOM group (95%). Reductive cleavage of the epoxide **6** with HCOOH and tris(dibenzylideneacetone)-dipalladium(*o*)-chloroform adduct (Pd₂(dba)₃·CHCl₃)⁷⁾ smoothly occurred to give alcohol **7** regioselectively in 91% yield, which was transformed into epoxy alcohol **10** through the sequence of protection of the secondary alcohol with a silyl group, diisobutylaluminum hydride (DIBAH) reduction, and the Katsuki–Sharpless epoxidation. The substitution reaction of **10** with Me₂CuCNLi₂^{8–10)} in Et₂O furnished a mixture of **11** and its regioisomer in a ratio of *ca.* 3 : 1, which was treated with NaIO₄ in aqueous tetrahydrofuran (THF) to afford the pure 1,3-diol **11** in 62% isolated yield. The diol **11** was then converted to α,β -unsaturated ester **14** in three steps: (1) selective oxidation of the primary alcohol to aldehyde with 2,2,6,6-tetramethyl-1-piperidinyloxy free radical (TEMPO) (91%)¹¹⁾; (2) Horner–Emmons olefination (84%); (3) protection of the secondary alcohol with a MOM group (95%). Reduction of the ester **14** with DIBAH followed by the Katsuki–Sharpless epoxidation produced epoxy alcohol **16** in 91% yield. Upon treatment of **16** with Me₂CuCNLi₂⁸⁾ in Et₂O, the regioselective substitution reaction smoothly occurred to give diol **17** as a single product in 97% yield, which was then converted to benzyl ester **20** by a three-step reaction sequence: (1) TEMPO oxidation¹¹⁾; (2) sodium chlorite oxidation^{12–15)}; (3) esterification with benzyl bromide in the presence of Cs₂CO₃ and TBAI in *N,N*-dimethylformamide (DMF) (92%, three steps). The resulting secondary alcohol **20** was protected with a MOM group to provide **21** in 97% yield.

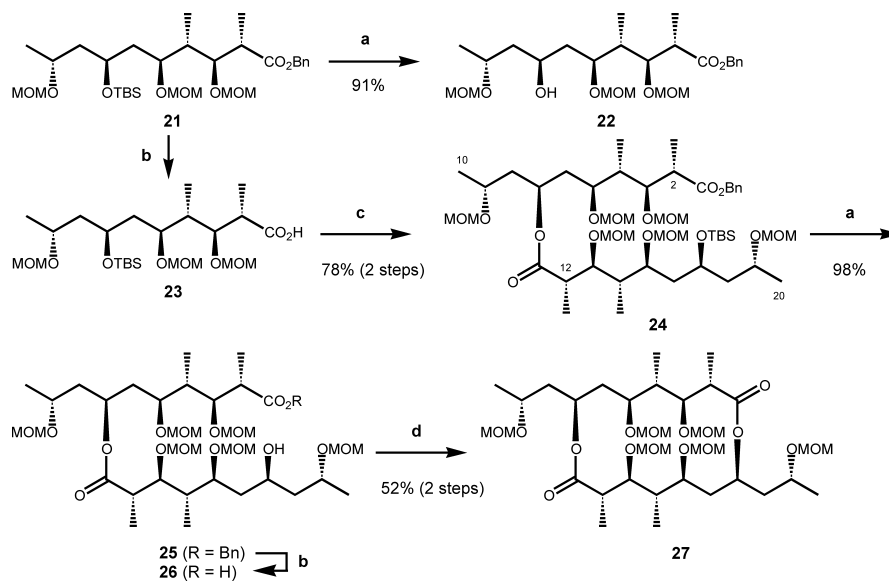
Thus, the contiguous five asymmetric carbon centers in the monomer counterpart **21** were constructed in a highly stereoselective manner by using different epoxide-opening reactions of the γ,δ -epoxy unsaturated ester **6** and two epoxy alcohols **10** and **16**. The remaining task for the synthesis of lepranthin (**1**) is construction of dimeric structure by macrolactonization. For this end, when **21** was treated with DDQ in aqueous THF, the silyl group was successfully removed to give rise to alcohol **22** in 91% yield.¹⁶⁾ On the other hand, hydrogenolysis of **21** over a Pd/C catalyst in AcOEt provided carboxylic acid **23**, which was immediately subjected to esterification with **22** using the Yamaguchi reagent¹⁷⁾ in the presence of *N,N*-diisopropylethylamine (DIPEA) and 4-(dimethylamino)pyridine (DMAP) leading to dimeric ester **24**¹⁸⁾ (78%). The diester **24** was transformed into *seco*-acid **26** via the same reaction sequence for **21**, that is, removal of the silyl group with 2,3-dichloro-5,6-dicyano-1,4-benzoquinone (DDQ) followed by removal of the benzyl ester under catalytic hydrogenation. Finally, the *seco*-acid **26** was transformed into the targeted diolide **27**¹⁹⁾ by macrolactonization¹⁷⁾ with 2,4,6-trichlorobenzoyl chloride in the presence of DIPEA and DMAP in 52% yield.

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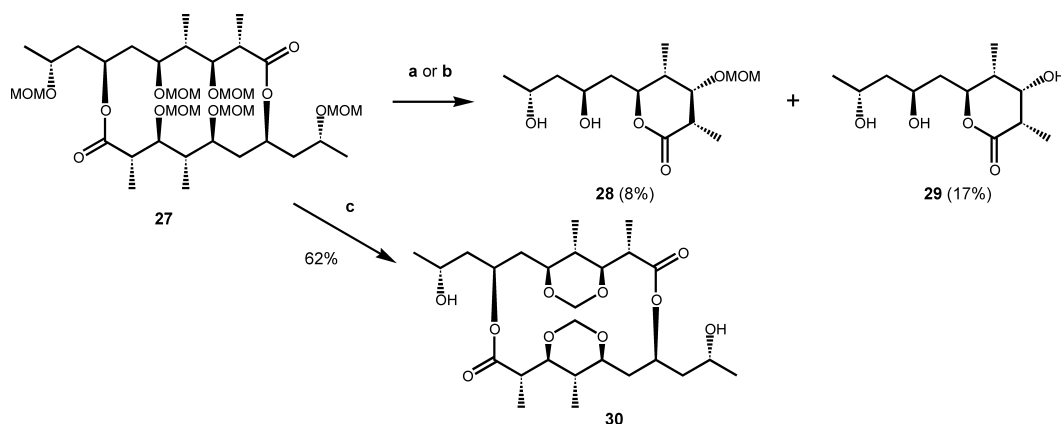
Reagents and Conditions: (a) L-(+)-DET, Ti(OⁱPr)₄, TBHP, CH₂Cl₂, -30 °C; (b) Dess–Martin periodinane, CH₂Cl₂, then Ph₃P=CHCO₂Me; (c) *tert*-*n*-butylammonium fluoride (TBAF), THF; (d) MOMCl, DIPEA, CH₂Cl₂; (e) Pd₂(dba)₃, CHCl₃, HCOOH, Et₃N, Bu₃P; (f) *tert*-butyldimethylsilyl chloride (TBSCl), imidazole, DMF; (g) DIBAH, THF, -25 °C; (h) L-(+)-diisopropyl tartrate (DIPT), Ti(OⁱPr)₄, TBHP, CH₂Cl₂, -40 °C; (i) MeLi, CuCN, Et₂O, -55 °C, then NaIO₄, THF, H₂O; (j) TEMPO, NaOCl, tetra-*n*-butylammonium bromide (TBAB), CH₂Cl₂, NaHCO₃ aq., 0 °C; (k) (MeO)₂P(O)CH₂CO₂Me, NaH, THF, 0 °C; (l) MOMCl, DIPEA, tetra-*n*-butylammonium iodide (TBAI), (CH₂Cl)₂, 50 °C; (m) DIBAH, THF, -30 °C; (n) L-(+)-DIPT, Ti(OⁱPr)₄, TBHP, CH₂Cl₂, -40 °C; (o) MeLi, CuCN, Et₂O, -50 to -10 °C, then NaIO₄, THF, H₂O; (p) TEMPO, NaOCl, TBAB, CH₂Cl₂, NaHCO₃ aq., 0 °C; (q) NaClO₂, NaH₂PO₄·2H₂O, 2-methyl-2-butene, THF, H₂O, 0 °C; (r) benzyl bromide (BnBr), Cs₂CO₃, TBAI, DMF; (s) MOMCl, DIPEA, TBAB, (CH₂Cl)₂, 50 °C.

Chart 1



Reagents and Conditions: (a) DDQ, THF, H₂O, 60 °C; (b) Pd/C, H₂, AcOEt; (c) 22, 2,4,6-trichlorobenzoyl chloride, DIPEA, DMAP, THF; (d) 22, 2,4,6-trichlorobenzoyl chloride, DIPEA, DMAP, THF, benzene.

Chart 2



Reagents and Conditions: (a) TsOH·H₂O, ^tBuOH, H₂O, 80 °C; (b) HCl, THF, 0 °C; (c) CF₃SO₃H, CH₂Cl₂, -20 °C.

Chart 3

Next, we carried out preliminary experiments for removal of the MOM groups in **27**. However, surprisingly, treatment of **27** with a catalytic amount of TsOH·H₂O in ^tBuOH/H₂O at 80 °C or HCl in THF at 0 °C produced a mixture of two fragment lactones **28** and **29**. On the other hand, upon treatment of **27** with CF₃SO₃H in CH₂Cl₂ at -20 °C, symmetric 16-membered diolide **30** containing two dioxane rings was formed.

In summary, we succeeded in the stereoselective synthesis of the diolide congener **27** of lepranthin (**1**) based on regio- and stereospecific epoxide-opening reactions. Namely, reductive cleavage of the epoxide **6** with HCOOH and Pd₂(dba)₃·CHCl₃ was used for construction of the C7 asymmetric center. On the other hand, four contiguous asymmetric carbon centers at the C2—C5 positions were constructed by the epoxide-opening reactions with the Lipshutz reagent (**10**→**11** and/or **16**→**17**). The subsequent key macrolactonization was successfully performed using the Yamaguchi reagent. Further studies toward the total synthesis of lepranthin (**1**) from the crucial intermediate **27** is now in progress in our laboratory.

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- For data of **24**: [α]_D²⁷ -25.11 (*c*=1.02, CHCl₃); HR-ESI-MS *m/z* 1001.5831 (Calcd for C₄₉H₉₀O₁₇Na: 1001.5845); IR (ATR) 1731 cm⁻¹; ¹H-NMR (400 MHz, CDCl₃) δ : 7.36—7.29 (5H, m), 5.25—5.19 (1H, br m), 5.12 (2H, s), 4.66—4.54 (12H, m), 4.04 (1H, tt, *J*=9.6, 2.8 Hz), 3.86—3.81 (1H, m), 3.80—3.74 (2H, br m), 3.73—3.65 (3H, m), 3.38 (6H, s), 3.36 (3H, s), 3.35 (3H, s), 3.34 (3H, s), 3.32 (3H, s), 2.96 (1H, qd, *J*=6.8, 6.0 Hz), 2.87 (1H, qd, *J*=7.2, 5.6 Hz), 2.17—2.11 (2H, m), 1.94—1.53 (7H, m), 1.41—1.33 (1H, m), 1.22—1.18 (12H, m), 0.89 (3H, d, *J*=7.2 Hz), 0.89 (9H, s), 0.83 (3H, d, *J*=7.2 Hz), 0.09 (3H, s), 0.07 (3H, s); ¹³C-NMR (100 MHz, CDCl₃) δ : 174.03 (C), 173.66 (C), 135.88 (C), 128.46 (2C, CH), 128.22 (2C, CH), 128.14 (CH), 98.35 (CH₂), 98.13 (CH₂), 98.87 (CH₂), 95.81 (CH₂), 95.73 (CH₂), 95.51 (CH₂), 83.44 (CH), 83.34 (CH), 74.97 (CH), 74.92 (CH), 71.69 (CH), 70.15 (CH), 69.72 (CH), 67.15 (CH), 66.21 (CH₂), 56.04 (2C, CH₃), 55.88 (CH₃), 55.74 (CH₃), 55.40 (CH₃), 55.22 (CH₃), 45.04 (CH₂), 43.07 (2C, CH), 41.65 (CH₂), 38.86 (CH₂), 38.01 (CH), 37.98 (CH), 35.52 (CH₂), 25.92 (3C, CH₃), 21.83 (CH₃), 21.20 (CH₃), 17.99 (C), 12.27 (2C, CH₃), 10.94 (CH₃), 10.74 (CH₃), -3.80 (CH₃), -4.62 (CH₃).
- For data of **27**: [α]_D³⁰ -40.87 (*c*=1.01, CHCl₃); HR-ESI-MS *m/z* 779.4412 (Calcd for C₃₆H₆₈O₁₆Na: 779.4405); IR (ATR) 1728 cm⁻¹; ¹H-NMR (400 MHz, CDCl₃) δ : 5.21 (2H, m), 4.70—4.53 (12H, m), 3.79 (2H, br d, *J*=5.6 Hz), 3.75—3.66 (4H, m), 3.40 (6H, s), 3.35 (6H, s), 3.32 (6H, s), 2.84 (2H, quintet, *J*=6.7 Hz), 2.15 (2H, m), 1.96—1.75 (4H, m), 1.75—1.57 (4H, m), 1.21 (6H, d, *J*=6.4 Hz), 1.20 (6H, d, *J*=6.4 Hz), 0.86 (6H, d, *J*=6.8 Hz); ¹³C-NMR (100 MHz, CDCl₃) δ : 173.77 (C), 98.23 (CH₂), 95.76 (CH₂), 95.47 (CH₂), 82.98 (CH), 74.42 (CH), 70.13 (CH), 70.03 (CH), 56.09 (CH₃), 55.92 (CH₃), 55.46 (CH₃), 43.11 (CH), 41.37 (CH₂), 37.58 (CH₃), 35.06 (CH₂), 21.21 (CH₃), 12.11 (CH₃), 10.67 (CH₃).